Supporting Information

2-(2'-Hydroxyphenyl)-benzothiazole (HBT)-quinoline conjugate: Highly specific fluorescent probe for Hg²⁺based on ESIPT and its application in bioimaging

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Fig.S1 ¹H NMR spectrum of A



Fig.S2 ¹³C NMR spectrum of A



Fig. S3 ESI MS spectrum of the A



Fig. S4 ¹H NMR spectrum of B



Fig. S5 ¹³C NMR spectrum of B



Fig. S6 ESI MS spectrum of the B



Fig. S7 ¹H NMR spectrum of L



Fig. S8 ¹³C NMR spectrum of L



Fig. S9 ESI MS spectrum of the L



Fig. S10 ESI MS spectrum of the $L-{\rm Hg}^{2+}$ complex



Fig. S11 Absorbance titration of L with increasing Hg²⁺ ion concentration in MeCN: H₂O (3:2, v/v, 10 mM HEPES Buffer, pH = 7). λ_{exc} = 340 nm,. Arrow indicates the increasing trend in Hg²⁺ ion concentration.



Fig. S12 Job's plot for determination of binding stoichiometry between L and Hg^{2+}



Fig.S13 Linear response curve of L at 590 nm depending on the Hg^{2+} ion concentration for determination of lowest detection limit

Entry	Probe	Solvent system	Detection limit	Binding constant	Stoichiometry (Ligand:Hg ²⁺)	Imaging applied on	References
						cells	
1	2	CH ₃ CN/H ₂ O	-	1.04×10^{5}	2:1	-	1
		(1:1; V/V)		M ⁻²			
2	1	PBS buffer	20 ppb	-	2:1	-	2
		(~1% CH ₃ CN)					
3	1	PBS buffer	5.1 nM	-	1:1	HeLa cells	3
		(0.5% CH ₃ CN)					
4	2	PBS buffer	3.8 nM	-	1:1	HeLa cells	3
		(0.5% CH ₃ CN					
5	PDP	CH ₃ CN/ H ₂ O (1	4.9 μM	7.5×10^{3}	1:1	lung cancer	4
		: 1;V/V)		M-1		cell line (NCI-	
						H460)	
6	Pvi	PBS buffer (1%	7.8 nM	-	-	HeLa cells	5
		CH ₃ CN)					
7	L	CH ₃ CN: H ₂ O	0.11 μM	1.24×10^{4}	1:1	HeLa cells	Present Study
		(3:2; V/V)		M ⁻¹			

 Table 1: Comparison of some ESIPT based Hg²⁺ sensors

References

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