# **Supporting Information**

# Unidirectional thermal expansion in KZnB<sub>3</sub>O<sub>6</sub>: role of alkali metals

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1

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### 1. Experimental methods

## 2. Computational methods

3. **Table S1.** The lattice constants for  $K_{0.5}M_{0.5}ZnB_3O_6$  ( $M = Na, K^5$  and Rb) at 298K.

4. **Figure S1.** The variation of lattice constants with temperatures for  $K_{0.5}M_{0.5}ZnB_3O_6$  (*M* = Na, K<sup>5</sup> and Rb).

5. **Table S2.** The calculated bondlengths of *M*-O in  $MZnB_3O_6$  ( $M = Na, K^5$  and Rb) at ground states.

6. **Figure S2.** Specific heat of  $K_{0.5}M_{0.5}ZnB_3O_6$  (*M* = Na, K and Rb) and their fitting results.

7. **Figure S3.** The phonon dispersion of  $MZnB_3O_6$  ( $M = Na, K^5$  and Rb).

8. References

#### **Experimental methods**

**Samples.** Polycrystalline samples of K<sub>1-x</sub>*M*<sub>x</sub>ZnB<sub>3</sub>O<sub>6</sub> (*M* = Na and Rb) (x ≤ 0.8) were synthesized by traditional solid-state reaction techniques. The stoichiometric mixtures of A.R. Na<sub>2</sub>CO<sub>3</sub>/K<sub>2</sub>CO<sub>3</sub>/Rb<sub>2</sub>CO<sub>3</sub>, H<sub>3</sub>BO<sub>3</sub> and ZnO were ground well and packed into Al<sub>2</sub>O<sub>3</sub> crucible. In order to compensate for volatilization of alkali metals, 15% excess of Na<sub>2</sub>CO<sub>3</sub>/K<sub>2</sub>CO<sub>3</sub>/Rb<sub>2</sub>CO<sub>3</sub> (A.R.) is required. They were preheated at 500 °C for 12 h to decompose the carbonate and eliminate the water. Then they were intermediate grindings adequately, and calcined at 700 °C for 48 h. The single-phase powder of them was obtained. The purity of the samples was confirmed by powder X-ray diffraction (XRD) analysis. The powder XRD data were carried out with a Panalytical diffractometer (X'Pert PRO MRD) operating in Bragg–Brentano geometry with CuKα (λ = 1.5418 Å) radiation and a graphite monochromator. The diffraction patterns were taken from 10° to 80° (2θ) with a scan step width of 0.017° at room temperature.

High temperature XRD measurements. Temperature-dependent *in situ* X-ray diffractometry was performed on an XPERT-PRO powder diffractometer system (CoK $\alpha$ 1; 1.78901 Å) equipped with an Anton Paar HTK-1200N Oven Sample stage. The room-temperature diffraction pattern in the angular range from 10 ° to 80 ° with a scanning step width of 0.017 ° was firstly obtained as a standard, and then the sample was heated from 373 K to 973 K at intervals of 100 K. Each diffraction pattern was obtained 30 min after the required

temperature was reached. Unit cell parameters were then calculated by using the pattern indexing software Dicvol06.<sup>1</sup>

#### Computational methods

The first-principles calculations presented in our work were performed with the CASTEP program code with the plane-wave pseudopotential method.<sup>2</sup> We adopted the generalized gradient approximation (GGA) in the form of the Perdew-Burke-Ernzerhof for the exchange-correlation potentials.<sup>3</sup> The ultrasoft pseudopotential with a plane-wave energy cutoff of 410 eV and a  $4\times4\times4$  Monkhorst Pack k-point mesh in the reciprocal space were used for all the calculations.<sup>4</sup> The self-consistent field was set as  $5\times10^{-7}$  eV/atom. Based on the experimental lattice parameters, all independent internal atomic coordinates were optimized (Broyden Fletcher Goldfarb Shanno algorithm) with the convergence standard given as follows: energy change less than  $5\times10^{-6}$  eV/atom, residual force less than 0.01 eV/Å, stress less than 0.02 GPa, and displacement of atom less than  $5\times10^{-4}$  Å. The phonon frequencies and phonon density of states (PHDOS) were obtained with the finite displacement method based on the optimized structures.

	$K_{0.5}Na_{0.5}ZnB_3O_6$	$KZnB_{3}O_{6}^{5}$	$K_{0.5}Rb_{0.5}ZnB_3O_6$
a (Å)	6.716(1)	6.742(1)	6.803(2)
b (Å)	6.785(1)	6.921(1)	6.949(1)
c (Å)	6.986(1)	7.068(1)	7.106(1)
α (°)	63.40(1)	63.13(1)	63.18(1)
β (°)	71.73(1)	72.40(1)	73.19(1)
γ (°)	69.02(1)	69.08(1)	69.08(1)
V (Å <sup>3</sup> )	261.26	270.89	276.73

**Table S1.** The lattice constants for  $K_{0.5}M_{0.5}ZnB_3O_6$  ( $M = Na, K^5$  and Rb) at



**Figure S1.** The variation of lattice constants with temperatures for  $K_{0.5}M_{0.5}ZnB_3O_6$  (*M* = Na, K<sup>5</sup> and Rb).

a) K<sub>0.5</sub>Na<sub>0.5</sub>ZnB<sub>3</sub>O<sub>6</sub>; b) KZnB<sub>3</sub>O<sub>6</sub>;<sup>5</sup> c) K<sub>0.5</sub>Rb<sub>0.5</sub>ZnB<sub>3</sub>O<sub>6</sub>.

at ground states.				
Bonds	M = Na	$M = K^5$	M = Rb	
M-01	2.493	2.863	3.062	
M-O2	3.433	3.228	3.255	
M-O2	2.340	2.665	2.820	
M-O3	2.997	3.115	3.286	
M-O4	2.655	2.852	2.943	
M-O5	3.307	3.276	3.268	
M-O5	2.595	2.871	3.016	
M-06	2.631	2.884	2.992	
M-06	2.882	2.885	2.958	
$\Delta$ %*	31.8%	18.7%	14.2%	

**Table S2.** The calculated bondlengths of *M*-O in  $MZnB_3O_6$  (*M* = Na, K<sup>5</sup> and Rb)

\* $\Delta$ %: Nonuniformity of the bond lengths

 $\Delta$ % = [(bond length) max - (bond length) min] / (bond length) max



**Figure S2.** Specific heat of  $K_{0.5}M_{0.5}ZnB_3O_6$  (*M* = Na, K and Rb) and their fitting results. a)  $K_{0.5}Na_{0.5}ZnB_3O_6$ ; b)  $KZnB_3O_6$ ; c)  $K_{0.5}Rb_{0.5}ZnB_3O_6$ . The symbols stand for experimental data; the red lines are the fitting results of the fraction of the Debye model; the blue short dashes are the fitting results of the fraction of the Einstein model; the black lines are the sum fitting results of Debye and Einstein model.



**Figure S3.** The phonon dispersion of  $MZnB_3O_6$  ( $M = Na, K^5$  and Rb). a) NaZnB<sub>3</sub>O<sub>6</sub>; b) KZnB<sub>3</sub>O<sub>6</sub>;<sup>5</sup> c) RbZnB<sub>3</sub>O<sub>6</sub>. It shows that the phonon dispersions above 20meV are similar, but downshifting on going from Na to K, and then to Rb below 20meV.

#### References

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