

Tris(pyrazolyl)phosphines with Copper(I): From Monomers to Polymers

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Crystallographic Details

Details for $[C^H\text{Cu}][\text{PF}_6] \cdot \text{DCM}$

In crystal structure $[C^H\text{Cu}][\text{PF}_6] \cdot \text{DCM}$ the molecules of $[C^H\text{Cu}]$ form a 1-dimensional polymeric chain in the crystallographic $uvw=[1,-1,1]$ direction (Fig. S1). The length of $uvw=[1,-1,1]$ in the unit cell is 21.1191(12) Å. Because there are four independent monomeric units, the average Cu...Cu distance is consequently 5.2798(3) Å. This Cu coordination polymer shows “whole molecule” disorder with one disorder component running in “up” direction and the other disorder component in “down” direction (Fig. S2). The crystal structure of $[C^H\text{Cu}][\text{PF}_6] \cdot \text{DCM}$ has triclinic P1 (no. 1) symmetry. In the packing of the crystal the polymeric chains form a pseudo-hexagonal arrangement (Fig. S3).

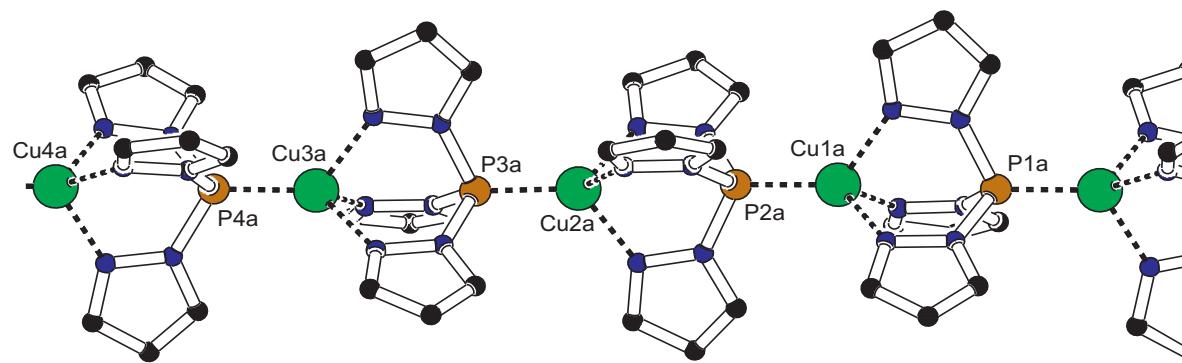


Figure S1. Polymeric coordination chain of $([C^H\text{Cu}][\text{PF}_6])_n$ in crystal structure $[C^H\text{Cu}][\text{PF}_6] \cdot \text{DCM}$. Only the major disorder component is shown. Hydrogen atoms, disordered PF_6^- anions and severely disordered CH_2Cl_2 solvent molecules are omitted for clarity.

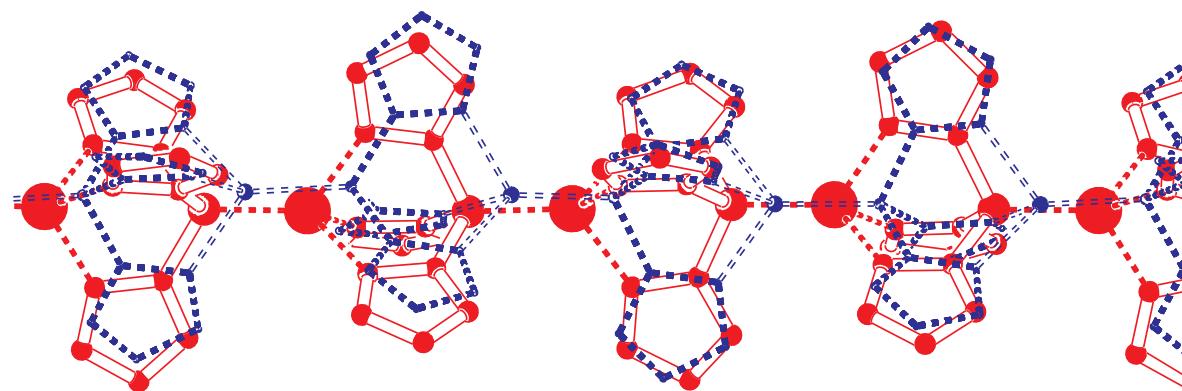


Figure S2. Disorder of the Cu coordination polymer in crystal structure $[C^H\text{Cu}][\text{PF}_6] \cdot \text{DCM}$. The major disorder component is drawn in red, the minor component in blue. The ratio between major and minor component is estimated as 89:11%. Hydrogen atoms, disordered PF_6^- anions and severely disordered CH_2Cl_2 solvent molecules are omitted for clarity.

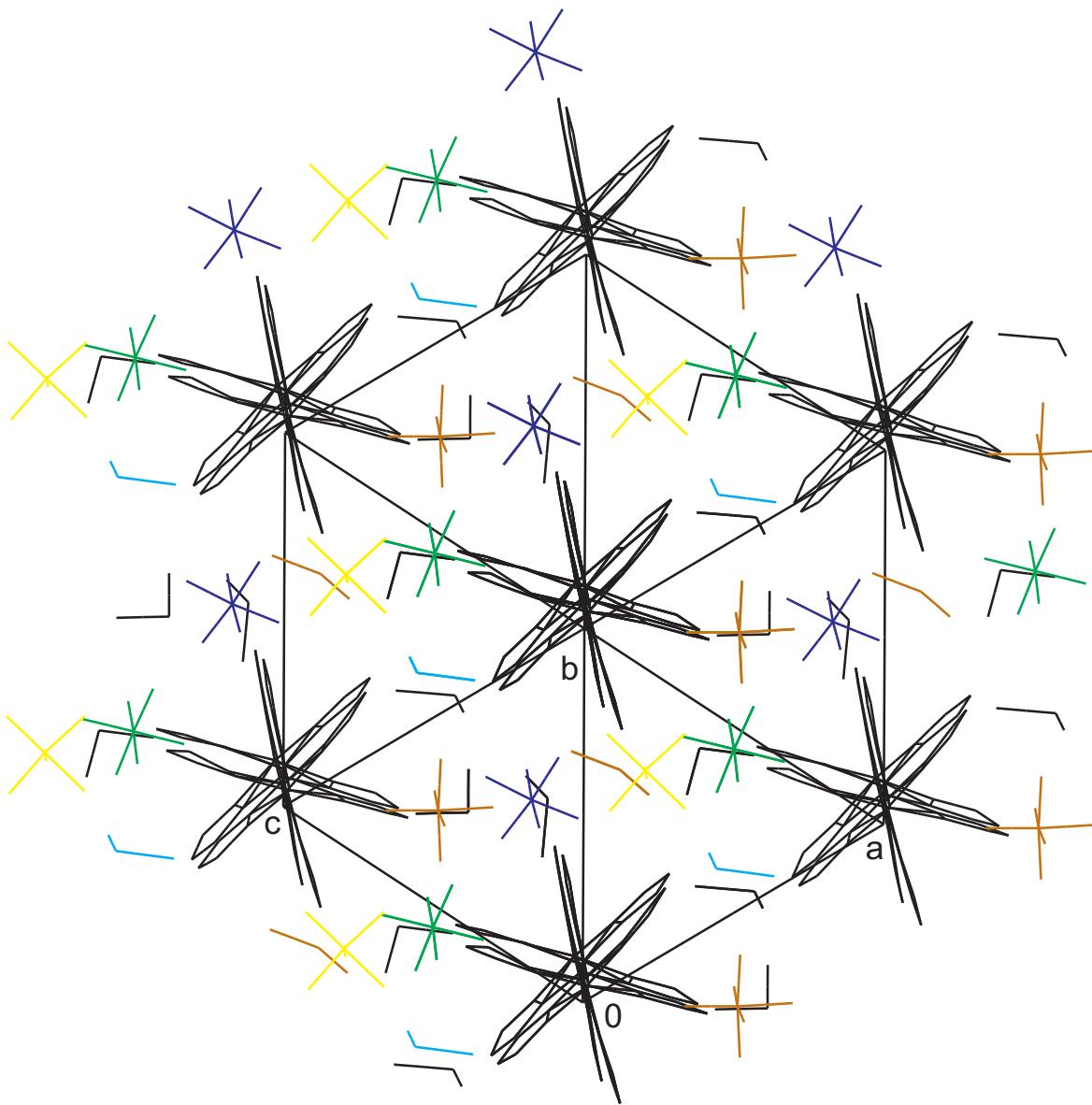


Figure S3. Packing of $[C^H\text{Cu}][\text{PF}_6]\cdot\text{DCM}$ in the crystal. View along the $uvw = [1,-1,1]$ direction. Hydrogen atoms are omitted for clarity. Only the major forms of the disordered molecules are shown. In the final least-squares refinements, the CH_2Cl_2 molecules have been treated with the Squeeze routine.

Details for $[C^H\text{Cu}][\text{PF}_6]\cdot\text{DCE1}$ (crystal 1) and $[C^H\text{Cu}][\text{PF}_6]\cdot\text{DCE2}$ (crystal 2)

The crystal structure of $[C^H\text{Cu}][\text{PF}_6]\cdot\text{DCE1}$ was merohedrally twinned. The twinning can be detected by comparing the merging R_{int} of the trigonal Laue group -31m with the higher symmetric hexagonal 6/mmm (Table S2). For both symmetries the R_{int} values are similar. As expected, in the case of the non-twinned $[C^H\text{Cu}][\text{PF}_6]\cdot\text{DCE2}$ there is a significant difference between Laue groups -31m and 6/mmm. The twin operation for $[C^H\text{Cu}][\text{PF}_6]\cdot\text{DCE1}$ could be found by a coset decomposition (see Table S1) of point group 6/mmm (symmetry of the lattice) with respect to 31m (symmetry of the crystal).¹ The twin operation used in the refinement was a mirror plane perpendicular to the c -axis.

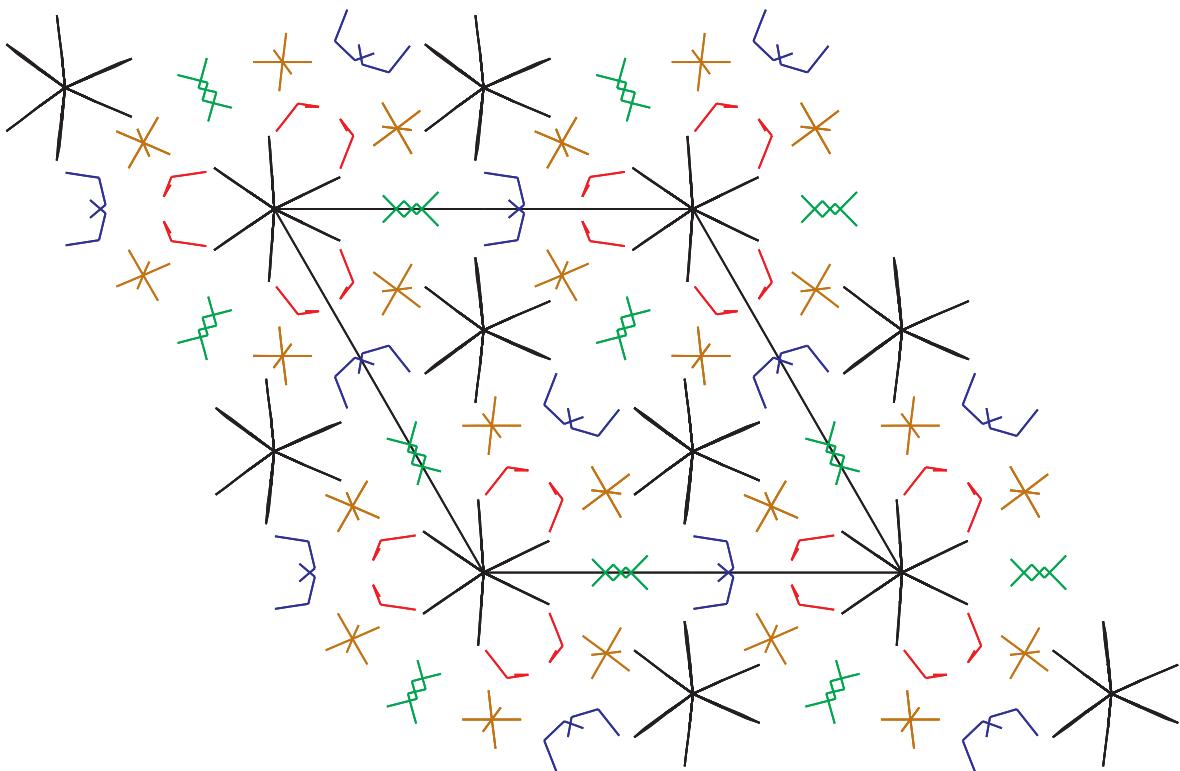


Figure S4. Packing of $[C^H Cu][PF_6] \cdot DCE2$ in the crystal. View along the c -direction. Hydrogen atoms are omitted for clarity. Only the major forms of the disordered molecules are shown.

Table S1. Coset decomposition of point group 6/mmm to 31m as determined by the Bilbao Crystallographic Server.² All symmetry operations of coset 3 are suitable twin operations for the crystal structure of $[C^H Cu][PF_6] \cdot DCE1$.

Coset 1	$\{1, 3^+_{001}, 3^-_{001}, m_{1-10}, m_{120}, m_{210}\}$
Coset 2	$\{2_{001}, 6^-_{001}, 6^+_{001}, m_{110}, m_{100}, m_{010}\}$
Coset 3	$\{2_{110}, 2_{100}, 2_{010}, m_{001}, -6^-_{001}, -6^+_{001}\}$
Coset 4	$\{2_{1-10}, 2_{120}, 2_{210}, -1, -3^+_{001}, -3^-_{001}\}$

Table S2. Merging R values R_{int} for the twinned crystal structure $[C^H Cu][PF_6] \cdot DCE1$ and the non-twinned $[C^H Cu][PF_6] \cdot DCE2$, as determined with the Eval15 software suite.³

Laue group	$[C^H Cu][PF_6] \cdot DCE1$	$[C^H Cu][PF_6] \cdot DCE2$
-3	0.034	0.053
-3m1	0.037	0.256
-31m	0.035	0.055
6/m	0.037	0.256
6/mmm	0.037	0.257

Table S3. Experimental Details of the Crystal Structure Determinations.

	[C ^{Me²} Cu(NCMe)][PF ₆]	[C ^{Me²} Cu(PPh ₃)][PF ₆]	[(C ^H) ₂ Cu][PF ₆]	[C ^H Cu][PF ₆]·DCM	[C ^H Cu][PF ₆]·DCE1	[C ^H Cu][PF ₆]·DCE2
formula	[C ₁₇ H ₂₄ CuN ₇ P](PF ₆)	[C ₃₃ H ₃₆ CuN ₆ P ₂](PF ₆) · CH ₂ Cl ₂	[C ₁₈ H ₁₈ CuN ₁₂ P ₂](PF ₆) + disordered solvent	[C ₉ H ₉ CuN ₆ P](PF ₆) + disordered solvent	[C ₉ H ₉ CuN ₆ P](PF ₆) · 3C ₂ H ₄ Cl ₂	[C ₉ H ₉ CuN ₆ P](PF ₆) · 3C ₂ H ₄ Cl ₂
fw	565.91	872.05	672.89 ⁴	440.70 ⁴	737.56	737.56
crystal size [mm ³]	0.41 × 0.14 × 0.06	0.51 × 0.36 × 0.12	0.41 × 0.17 × 0.15	0.58 × 0.07 × 0.05	0.68 × 0.11 × 0.06	0.40 × 0.13 × 0.06
crystal color	colorless	colorless	colorless	colorless	colorless	colorless
T [K]	150(2)	150(2)	110(2)	150(2)	150(2)	150(2)
crystal system	monoclinic	monoclinic	trigonal	triclinic	trigonal	trigonal
space group	C2/c (no. 15)	P2 ₁ /n (no. 14)	P31c (no. 159)	P1 (no. 1)	P31c (no. 159)	P31c (no. 159)
a [Å]	26.0651(16)	14.3558(1)	19.5288(14)	11.9578(8)	21.2193(6)	21.2337(5)
b [Å]	7.7824(3)	31.5939(2)	-	12.7183(14)	-	-
c [Å]	25.9327(13)	16.7621(1)	18.0311(14)	13.2817(8)	10.6362(5)	10.6155(4)
α [°]	-	-	-	87.521(3)	-	-
β [°]	116.124(3)	90.6316(2)	-	88.058(3)	-	-
γ [°]	-	-	-	84.097(4)	-	-
V [Å ³]	4723.0(4)	7602.08(8)	5955.3(10)	2006.5(3)	4147.4(3)	4145.0(3)
Z	8	8	8	4	6	6
d _{calc} [g/cm ³]	1.592	1.524	1.501 ⁴	1.459 ⁴	1.772	1.773
μ [mm ⁻¹]	1.13	0.91	0.96 ⁴	1.30 ⁴	1.55	1.55
abs. corr. type	multiscan ⁵	multiscan ⁵	multiscan ⁵	multiscan ⁵	numerical ⁵	numerical ⁵
abs. corr. range	0.57-0.75	0.54-0.89	0.67-0.75	0.52-0.75	0.60-0.98	0.61-0.94
(sin θ/λ) _{max} [Å ⁻¹]	0.65	0.65	0.65	0.65	0.65	0.65
refl. measured / unique / observed	33686 / 5582 / 4261	102481 / 17382 / 12960	83529 / 9095 / 8301	38040 / 17156 / 11100	41758 / 6026 / 5335	46288 / 6330 / 5148
R _{int}	0.057	0.067	0.027	0.064	0.035	0.048
parameters / restraints	306 / 0	1014 / 934	481 / 1	1053 / 9151	381 / 250	394 / 208
R1 / wR2 [I>2σ(I)]	0.0484 / 0.1198	0.0481 / 0.1134	0.0317 / 0.0799	0.0697 / 0.1768	0.0307 / 0.0646	0.0462 / 0.1100
R1/wR2 [all refl.]	0.0704 / 0.1328	0.0729 / 0.1250	0.0377 / 0.0830	0.1133 / 0.1987	0.0390 / 0.0677	0.0627 / 0.1176
S	1.023	1.026	1.050	1.064	0.944	1.030
Flack ⁶ x	-	-	0.003(3)	twin	twin	0.03(2)
ρ _{min/max} [e/Å ³]	-0.87 / 1.24	-1.31 / 1.39	-1.04 / 0.99	-0.61 / 0.86	-0.57 / 0.30	-0.57 / 2.36

Structure for [C^HCu][PF₆]·DCM

[C₉H₉CuN₆P](PF₆) + disordered solvent, Fw = 440.70⁴, colorless needle, 0.58 × 0.07 × 0.05 mm³, triclinic, P1 (no. 1), a = 11.9578(8), b = 12.7183(14), c = 13.2817(8) Å, α = 87.521(3), β = 88.058(3), γ = 84.097(4) °, V = 2006.5(3) Å³, Z = 4, D_x = 1.459 g/cm³, μ = 1.30 mm⁻¹. * 38040 Reflections were measured on a Bruker Kappa ApexII diffractometer with sealed tube and Triumph monochromator (λ = 0.71073 Å) at a temperature of 150(2) K up to a resolution of (sin θ/λ)_{max} = 0.65 Å⁻¹. The crystal appeared to be cracked into two fragments related by a 5.2 ° rotation about an arbitrary vector. Two orientation matrices were used for the integration of X-ray intensities with the Eval15 software.³ Multiscan absorption correction and scaling was performed with TWINABS⁵ (correction range 0.52-0.75). 17156 reflections were unique (R_{int} = 0.064), of which 11100 were observed [I>2σ(I)]. The structure was solved with Patterson superposition methods using SHELXT.⁷ Least-squares refinement was performed with SHELXL-2014⁸ against F² of all reflections based on a HKLF-5 file⁹ of four components (inclusive Friedel pairs). The crystal structure contains large voids (509 Å³ / unit cell) filled with disordered CH₂Cl₂ solvent molecules. Their contribution to the structure factors was secured by back-Fourier transformation with the Squeeze routine¹⁰ resulting in 255 electrons / unit cell. The Cu coordination polymer and the PF₆ anions were refined with disorder models. The major component of the coordination polymer and the P atoms of PF₆ were refined with anisotropic displacement parameters. The minor disorder component of the coordination polymer was refined with fixed isotropic displacement parameters of U = 0.05. The fluorine atoms of PF₆ were refined with free isotropic displacement parameters. All hydrogen atoms were introduced in calculated positions and refined with a riding model. 1053 parameters were refined with 9151 restraints (floating origin, distances and angles of the disordered groups). R1/wR2 [I > 2σ(I)]: 0.0697 / 0.1768. R1/wR2 [all refl.]: 0.1133 / 0.1987. S = 1.064. Batch scale factors of the four crystal components: 0.10(3), 0.405(17), 0.05(3). Residual electron density between -0.61 and 0.86 e/Å³. Geometry calculations and checking for higher symmetry was performed with the PLATON program.¹¹

Structure for [C^HCu][PF₆]·DCE1

[C₉H₉CuN₆P](PF₆) · 3C₂H₄Cl₂, Fw = 737.56, colorless needle, 0.68 × 0.11 × 0.06 mm³, trigonal, P31c (no. 159), a = b = 21.2193(6), c = 10.6362(5) Å, V = 4147.4(3) Å³, Z = 6, D_x = 1.772 g/cm³, μ = 1.55 mm⁻¹. 41758 reflections were measured on a Bruker Kappa ApexII diffractometer with sealed tube and Triumph monochromator (λ = 0.71073 Å) at a temperature of 150(2) K up to a resolution of (sin θ/λ)_{max} = 0.65 Å⁻¹. X-ray intensities were integrated with the Eval15 software.³ Numerical absorption correction and scaling was performed with SADABS⁵ (correction range 0.60-0.98). 6026 reflections were unique (R_{int} = 0.031), of which 5335 were observed [I>2σ(I)]. The structure was solved with

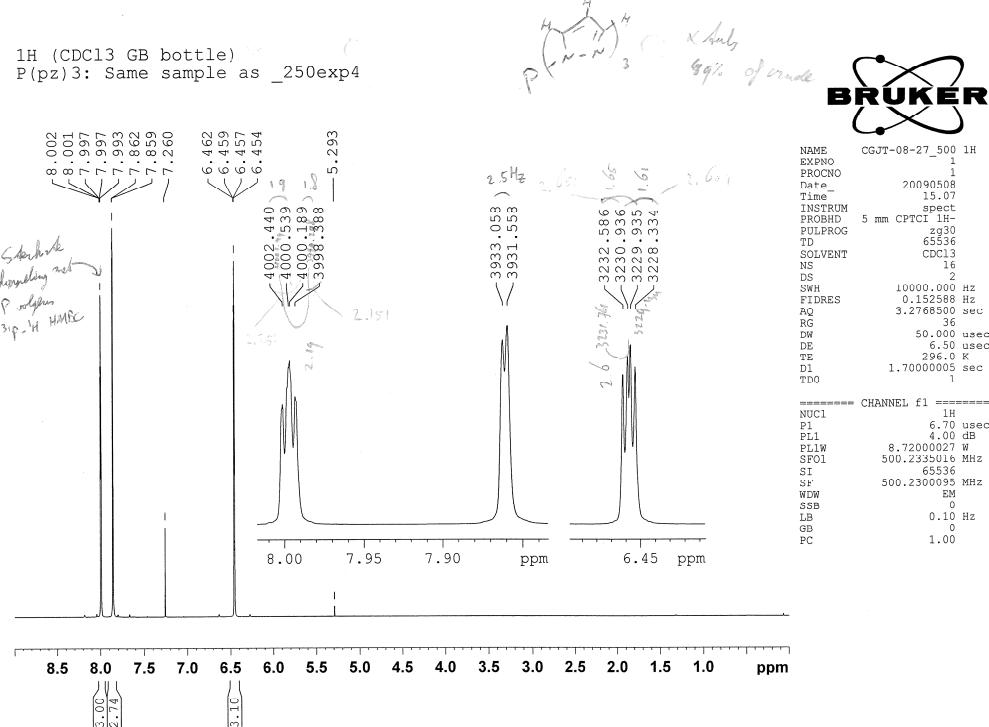
Patterson superposition methods using SHELXT.⁷ Least-squares refinement was performed with SHELXL-2014⁸ against F^2 of all reflections. The crystal was merohedrally twinned. The twin matrix (1,0,0 / 0,1,0 / 0,0,-1) was used for the refinement, resulting in a twin fraction BASF = 0.4815(14). Additional components for inversion twinning resulted in BASF values of zero and were omitted subsequently.

One Cu coordination polymer was refined with a disorder model. The estimated ratio between major and minor disorder component is 85:15%. The slight disorder in the other independent coordination chain has been ignored. All non-hydrogen atoms were refined with anisotropic displacement parameters with the exception of the minor disorder component (isotropic, constrained to the same value with EADP instruction). One of the C₂H₄Cl₂ solvent molecules was also refined with a disorder model. All hydrogen atoms were introduced in calculated positions and refined with a riding model. 381 parameters were refined with 250 restraints (floating origin, distances and angles of the disordered groups). R1/wR2 [I > 2σ(I)]: 0.0307 / 0.0646. R1/wR2 [all refl.]: 0.0390 / 0.0677. S = 0.944. Residual electron density between -0.57 and 0.30 e/Å³. Geometry calculations and checking for higher symmetry was performed with the PLATON program.¹¹

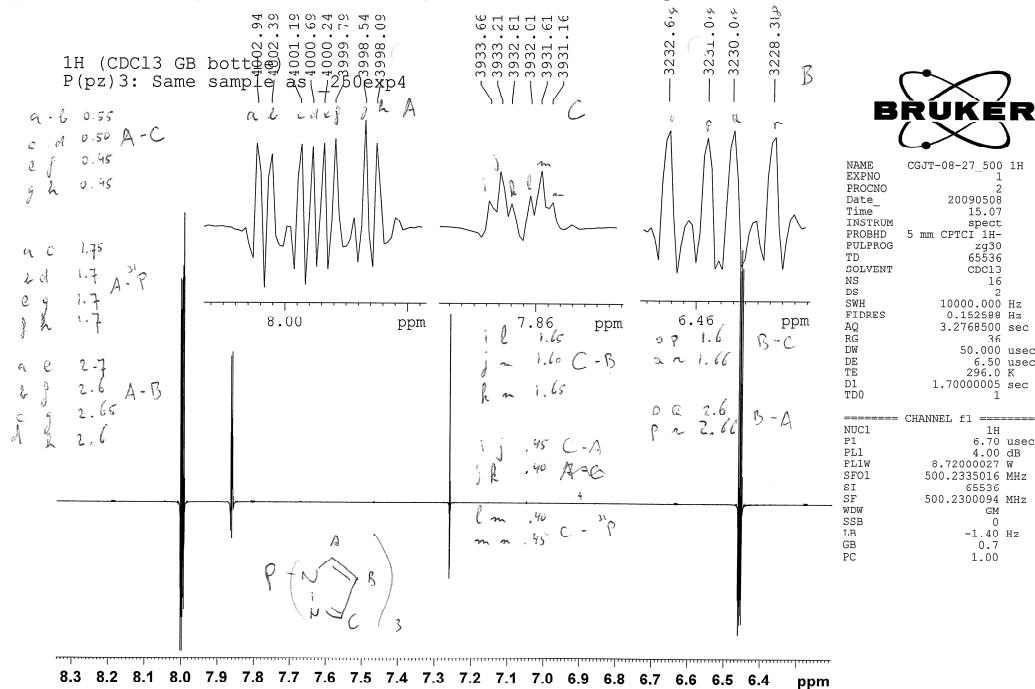
NMR spectra

NMR spectra of tris(pyrazolyl)phosphine ($\mathbf{C^H}$)

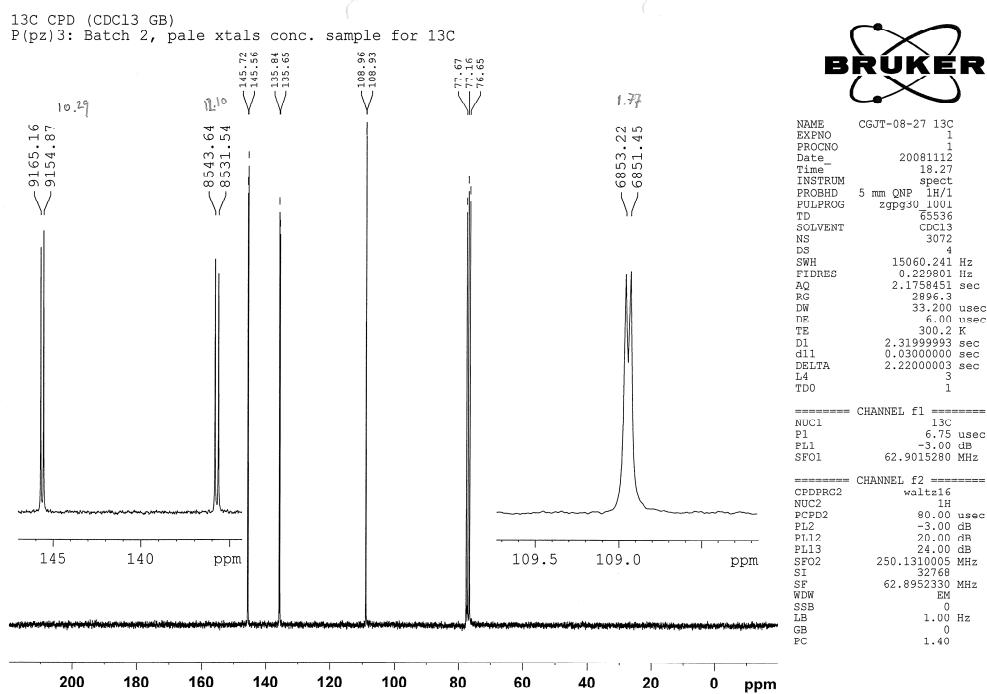
^1H NMR of $\mathbf{C^H}$ (500.2 MHz, CDCl_3)



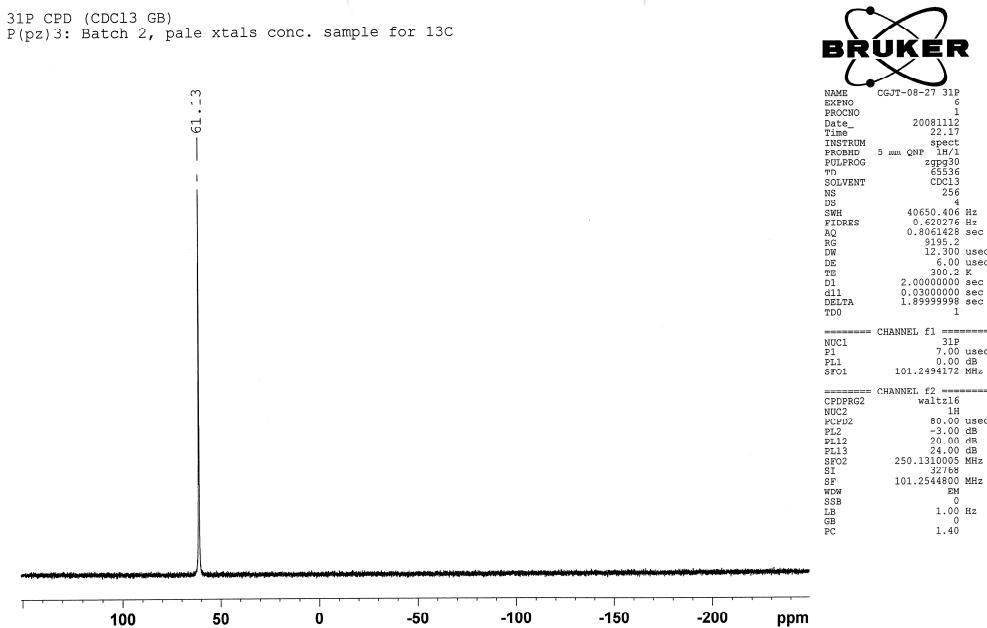
^1H NMR of $\mathbf{C^H}$ (500.2 MHz, CDCl_3) after line narrowing



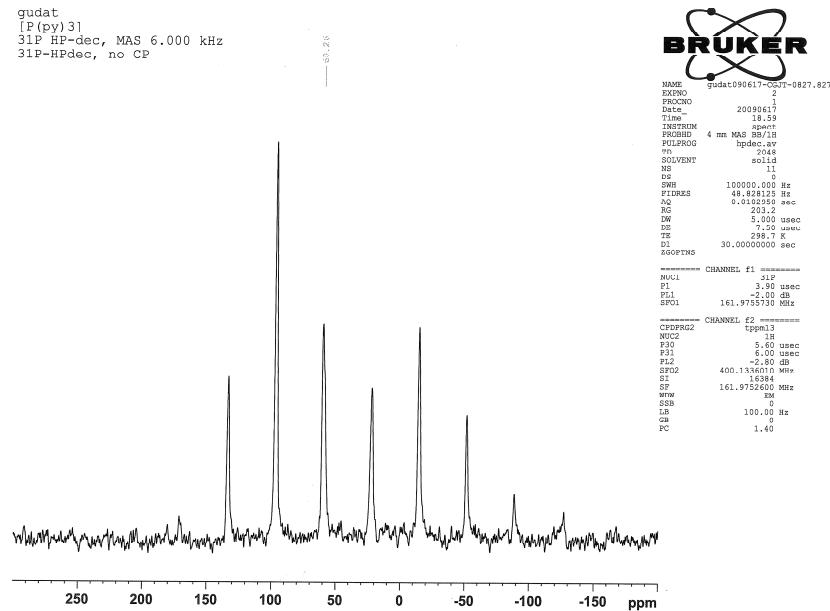
$^{13}\text{C}\{\text{H}\}$ NMR of C^{H} (62.9 MHz, CDCl_3)



$^{31}\text{P}\{\text{H}\}$ NMR of C^{H} (101.3 MHz, CDCl_3)

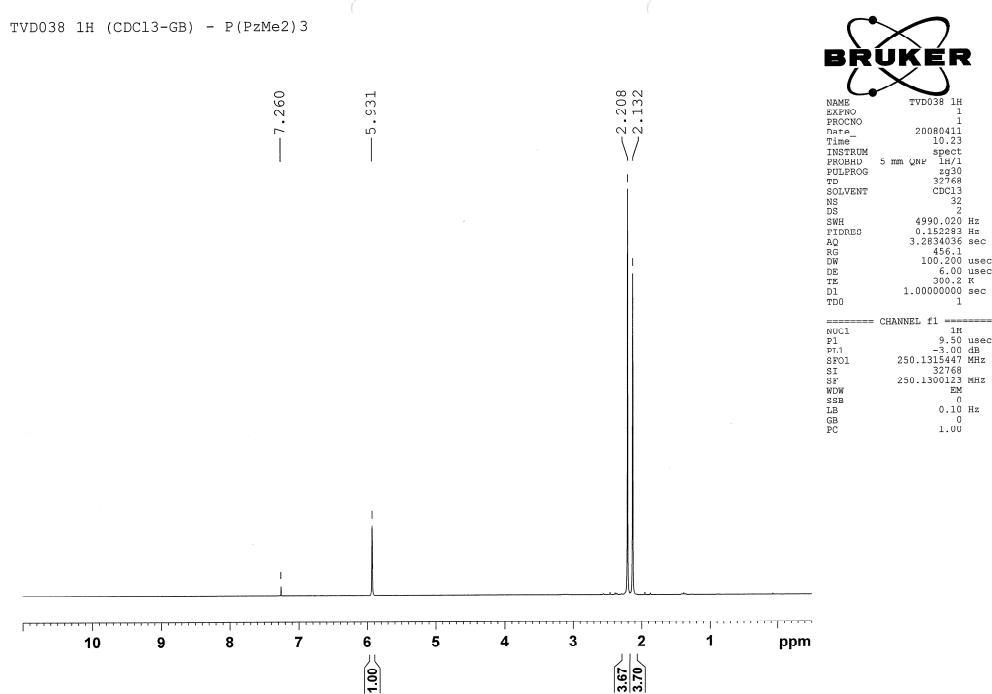


$^{31}\text{P}\{\text{H}\}$ MAS NMR of C^{H} (162.0 MHz)

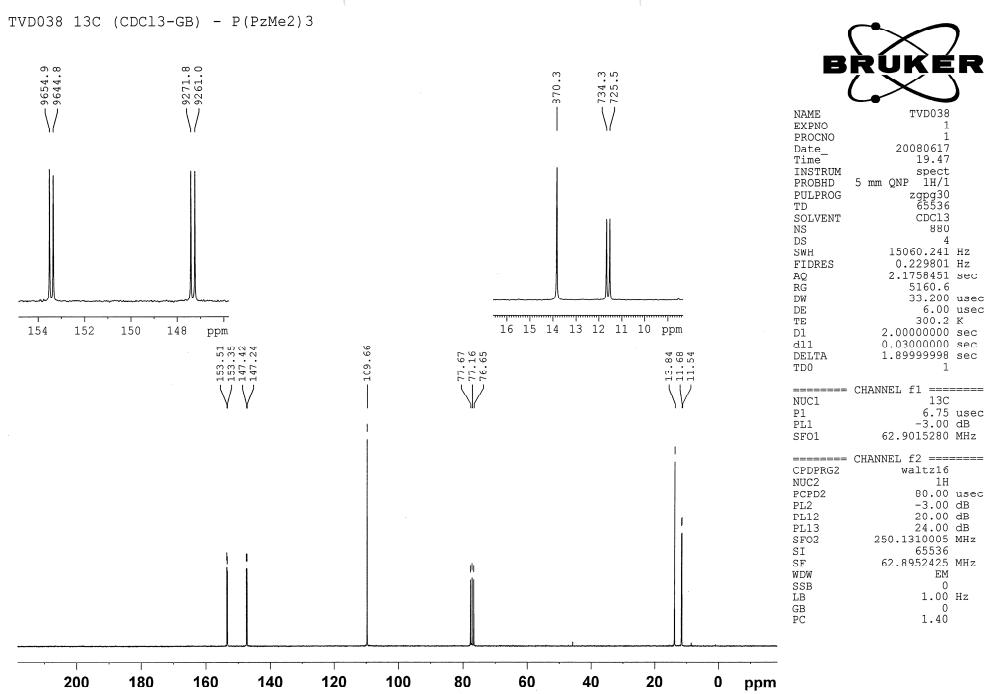


NMR spectra of tris(3,5-dimethylpyrazolyl)phosphine (C^{Me_2})

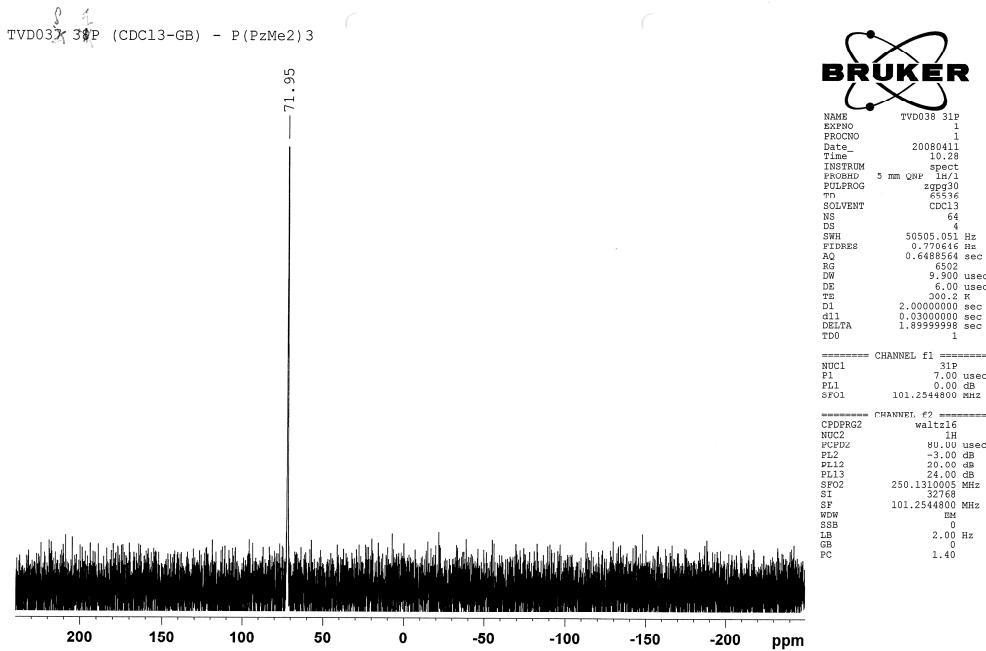
^1H NMR of C^{Me_2} (250.1 MHz, CDCl_3)



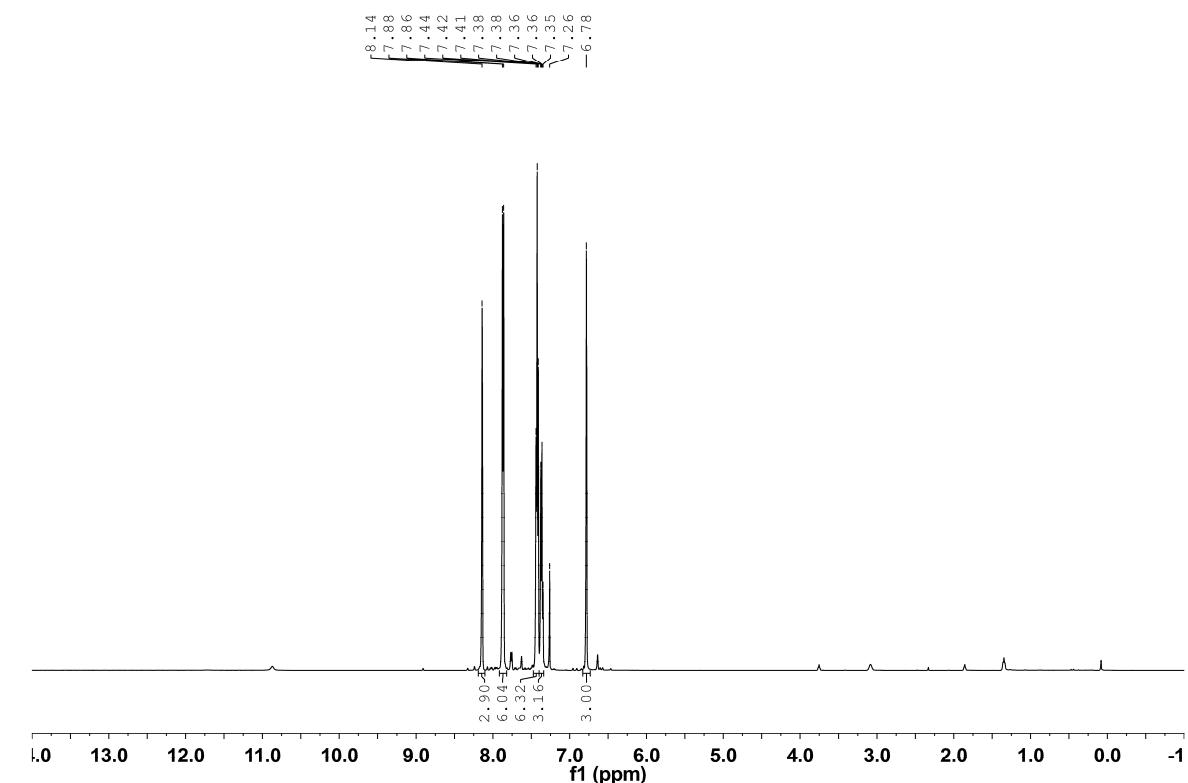
$^{13}\text{C}\{\text{H}\}$ NMR of C^{Me_2} (62.9 MHz, CDCl_3)



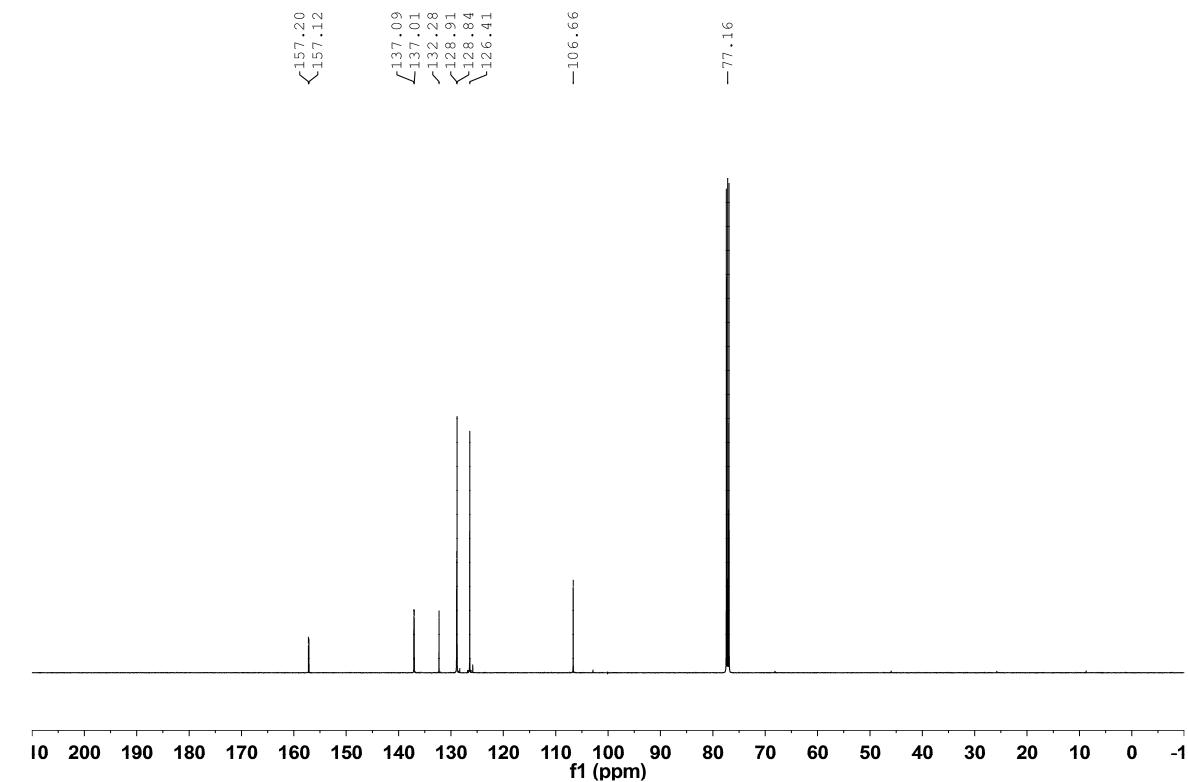
^{31}P NMR of C^{Me_2} (101.3 MHz, CDCl_3)



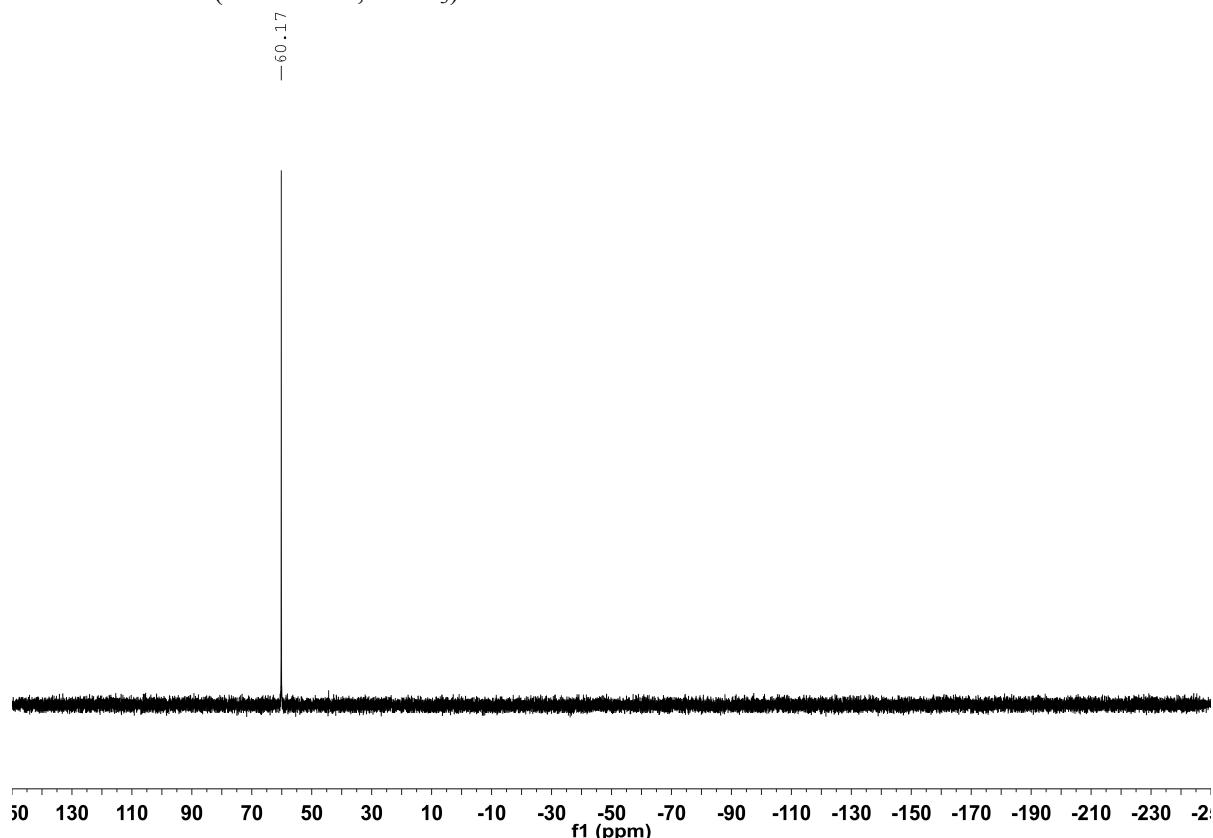
NMR spectra of tris(3-phenylpyrazol-1-yl)phosphine (C^{Ph})
 1H -NMR of C^{Ph} (500.2 MHz, $CDCl_3$)



^{13}C -NMR of C^{Ph} (125.8 MHz, $CDCl_3$)

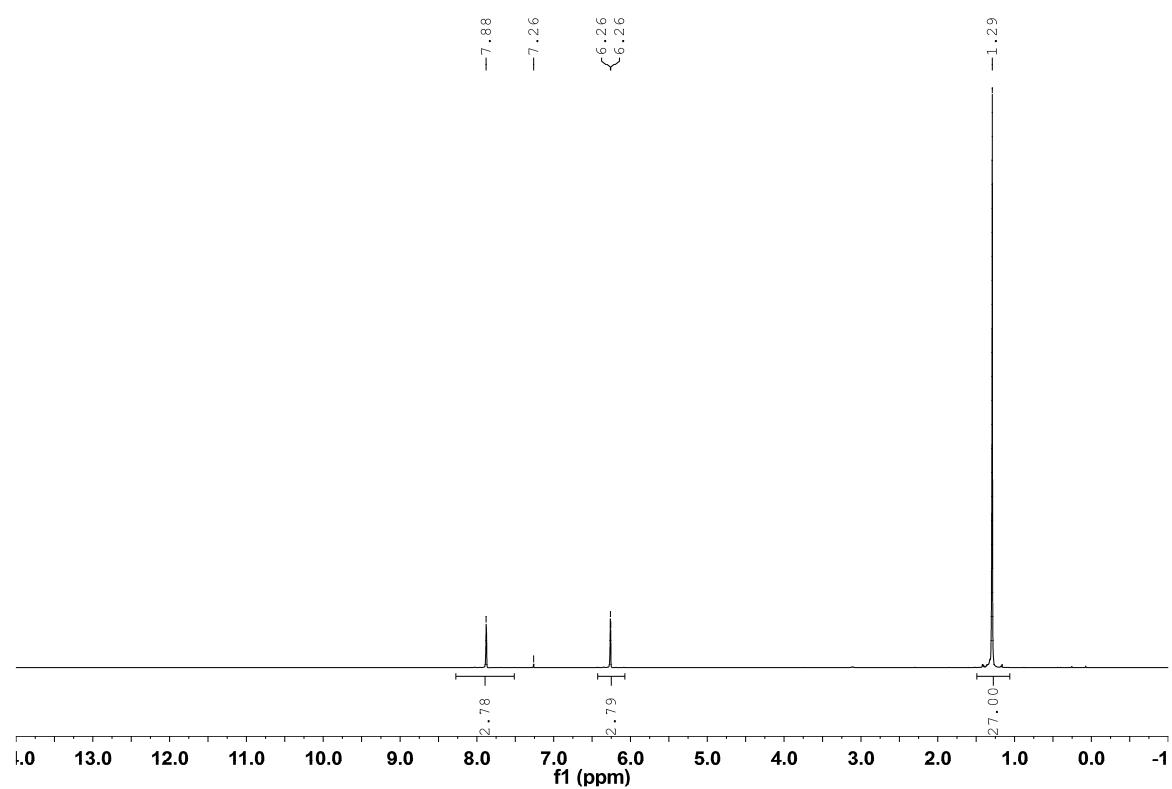


^{31}P -NMR of \mathbf{C}^{Ph} (162.0 MHz, CDCl_3)

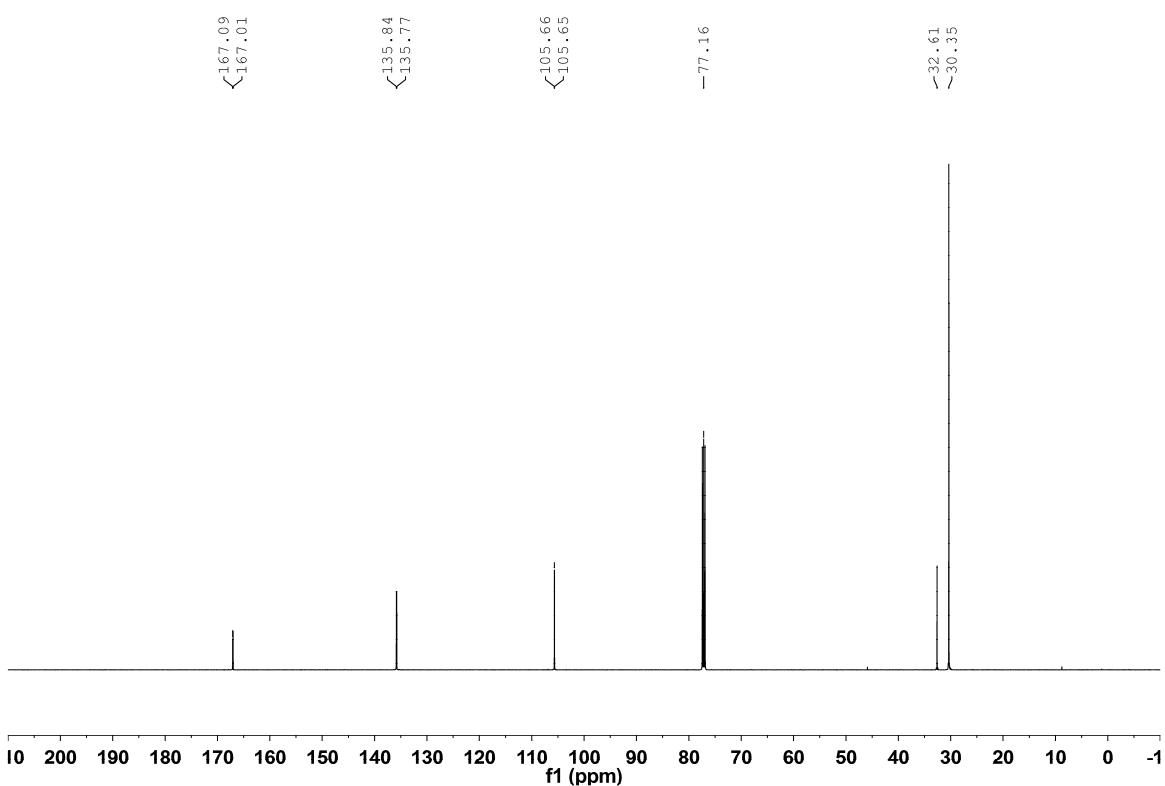


NMR spectra of tris(3-*tert*-butylpyrazol-1-yl)phosphine ($\mathbf{C}^{t\text{-Bu}}$)

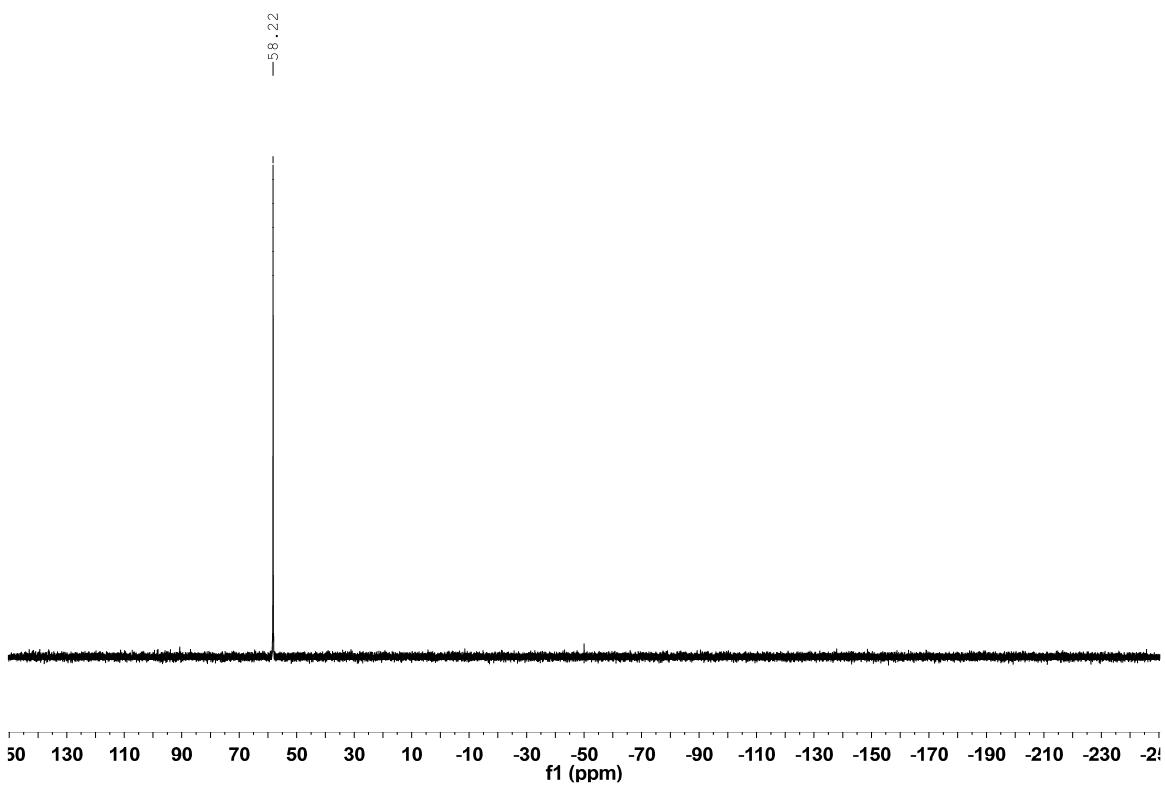
^1H -NMR of $\mathbf{C}^{t\text{-Bu}}$ (500.2 MHz, CDCl_3)



^{13}C -NMR of $\mathbf{C}^{t\text{-Bu}}$ (125.8 MHz, CDCl_3)

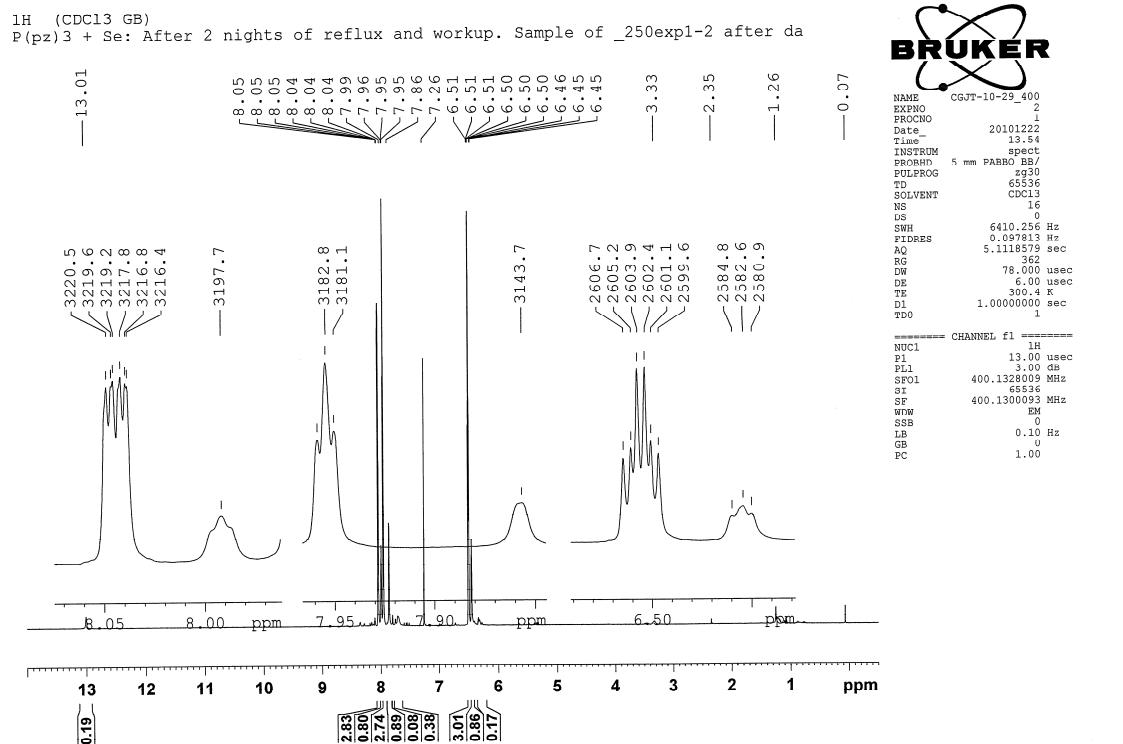


^{31}P -NMR of $\mathbf{C}^{t\text{-Bu}}$ (162.0 MHz, CDCl_3)

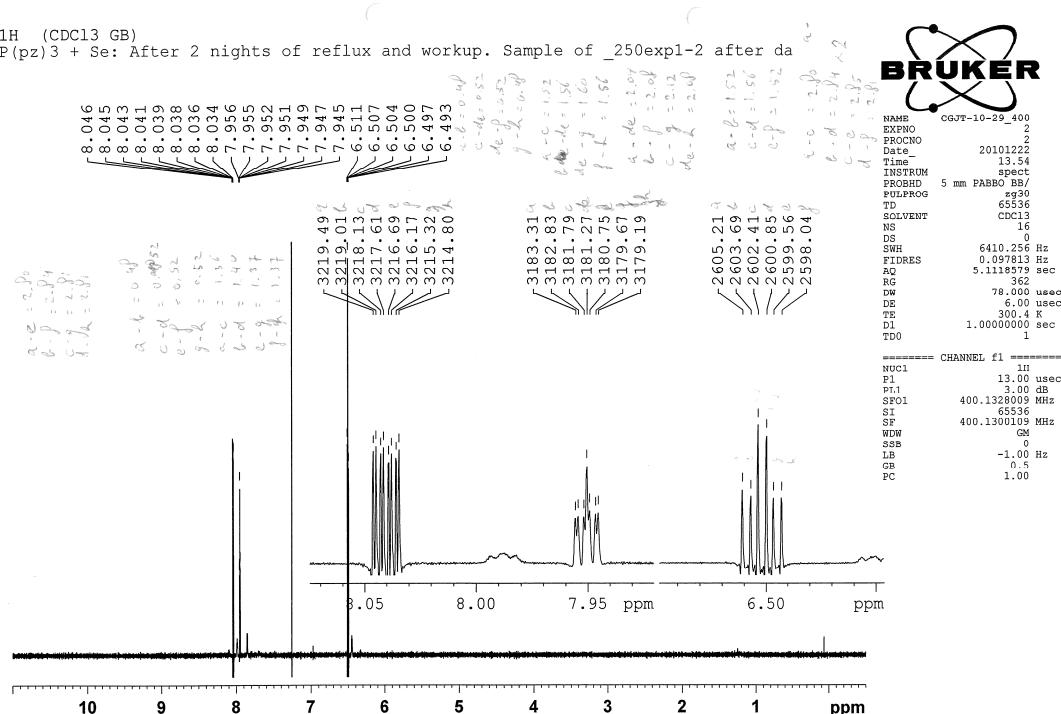


NMR spectra of tris(pyrazolyl)phosphine selenide (SeC^H)

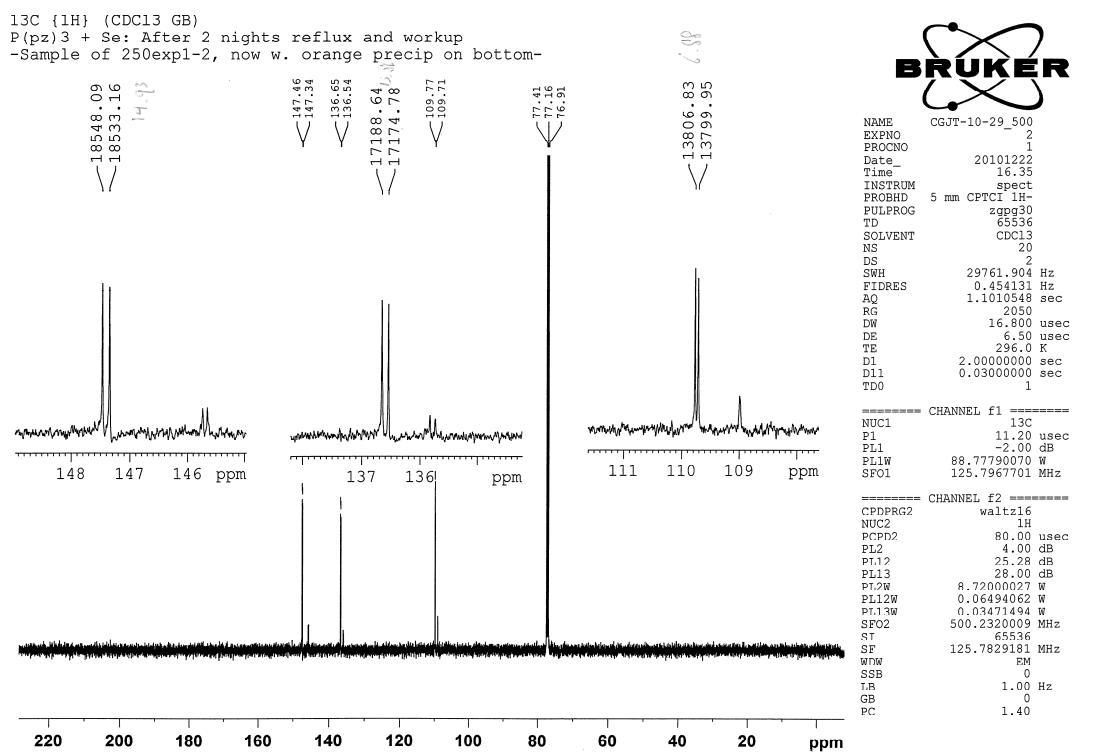
¹H NMR of SeC^H (400.1 MHz, CDCl₃)



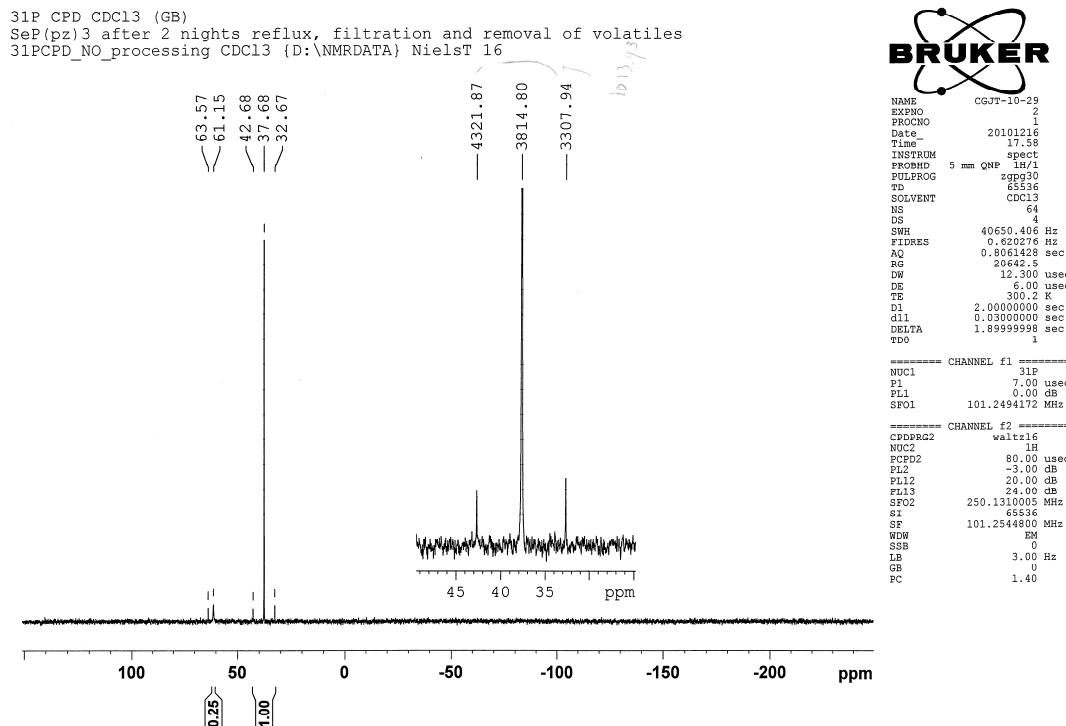
¹H NMR of SeC^H (400.1 MHz, CDCl₃) after line narrowing



$^{13}\text{C}\{\text{H}\}$ NMR of SeC^H (125.8 MHz, CDCl₃)

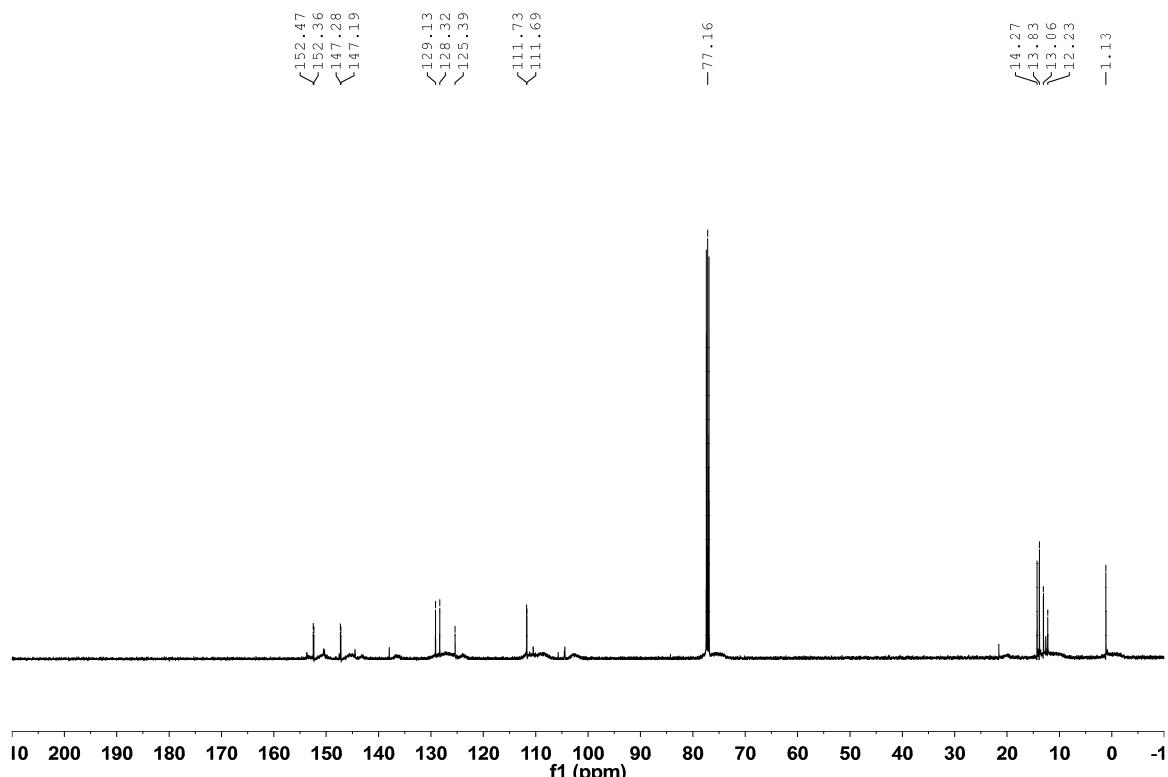


$^{31}\text{P}\{\text{H}\}$ NMR of SeC^H (101.3 MHz, CDCl₃)

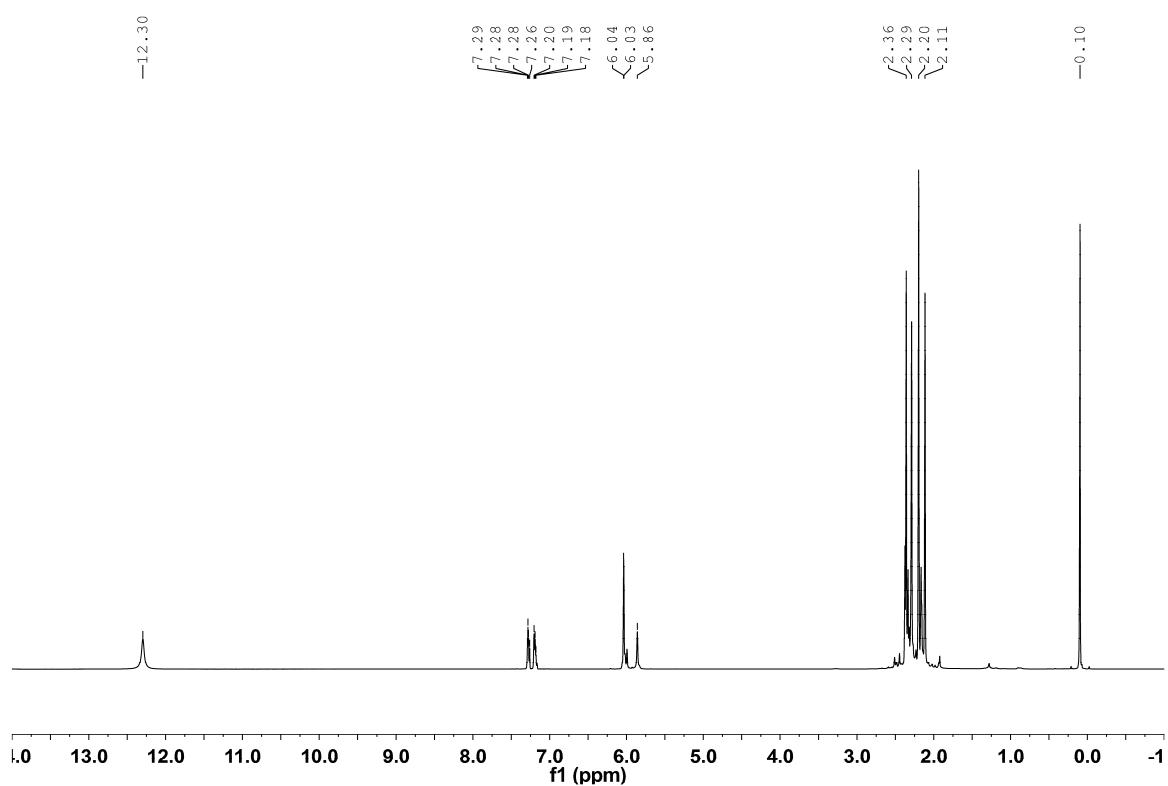


NMR spectra of tris(3,5-dimethylpyrazolyl)phosphine selenide ($\text{SeC}^{\text{Me}2}$)

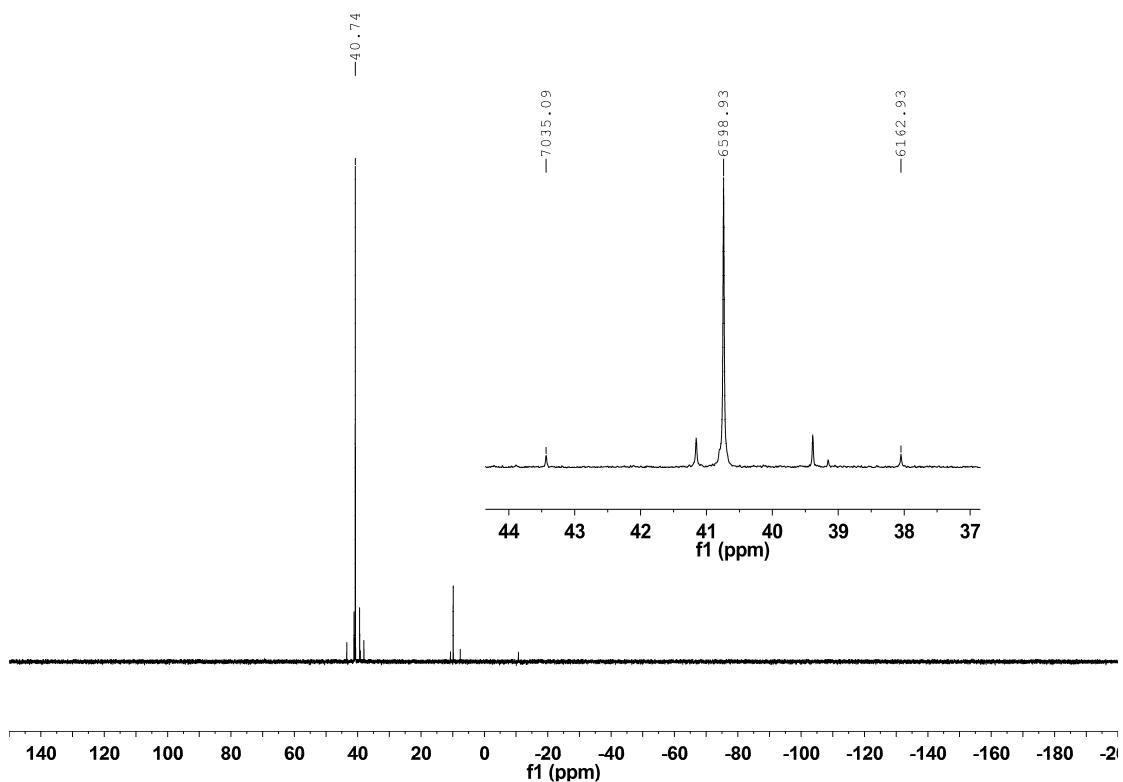
^1H NMR of $\text{SeC}^{\text{Me}2}$ (500.2 MHz, CDCl_3)



^{13}C NMR of $\text{SeC}^{\text{Me}2}$ (125.8 MHz, CDCl_3)

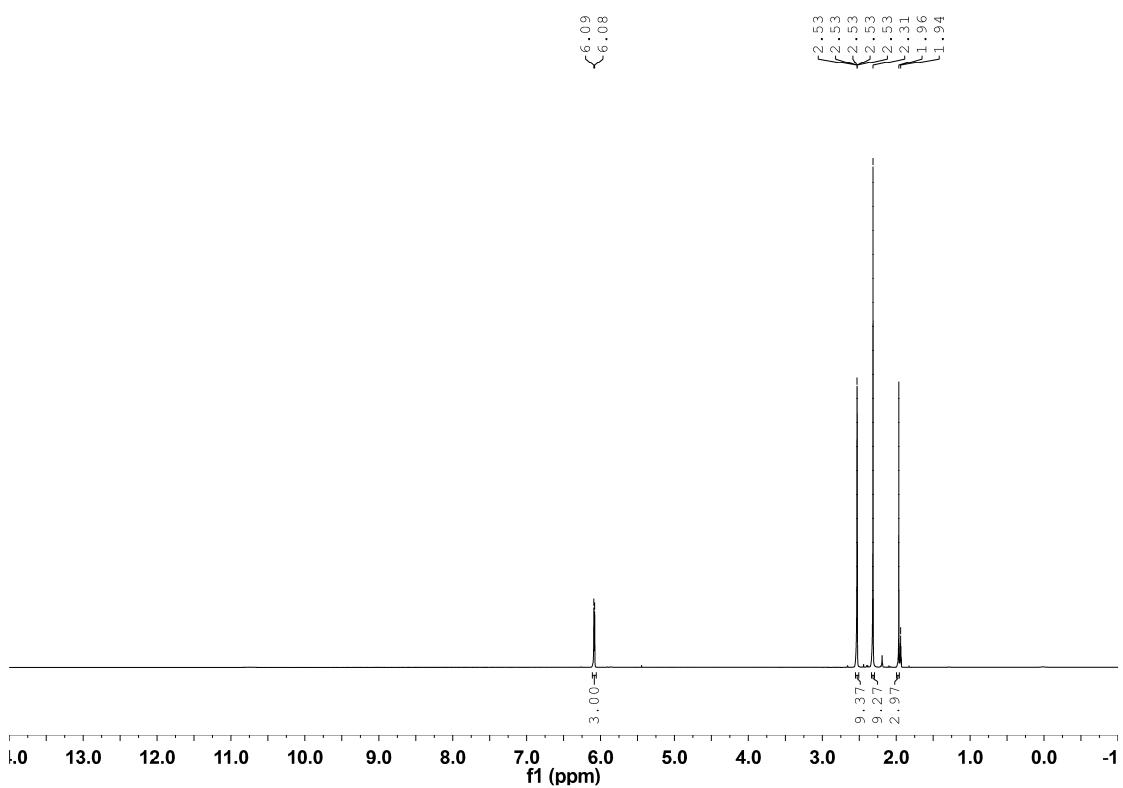


$^{31}\text{P}\{\text{H}\}$ NMR of $\text{SeC}^{\text{Me}2}$ (162.0 MHz, CDCl_3)

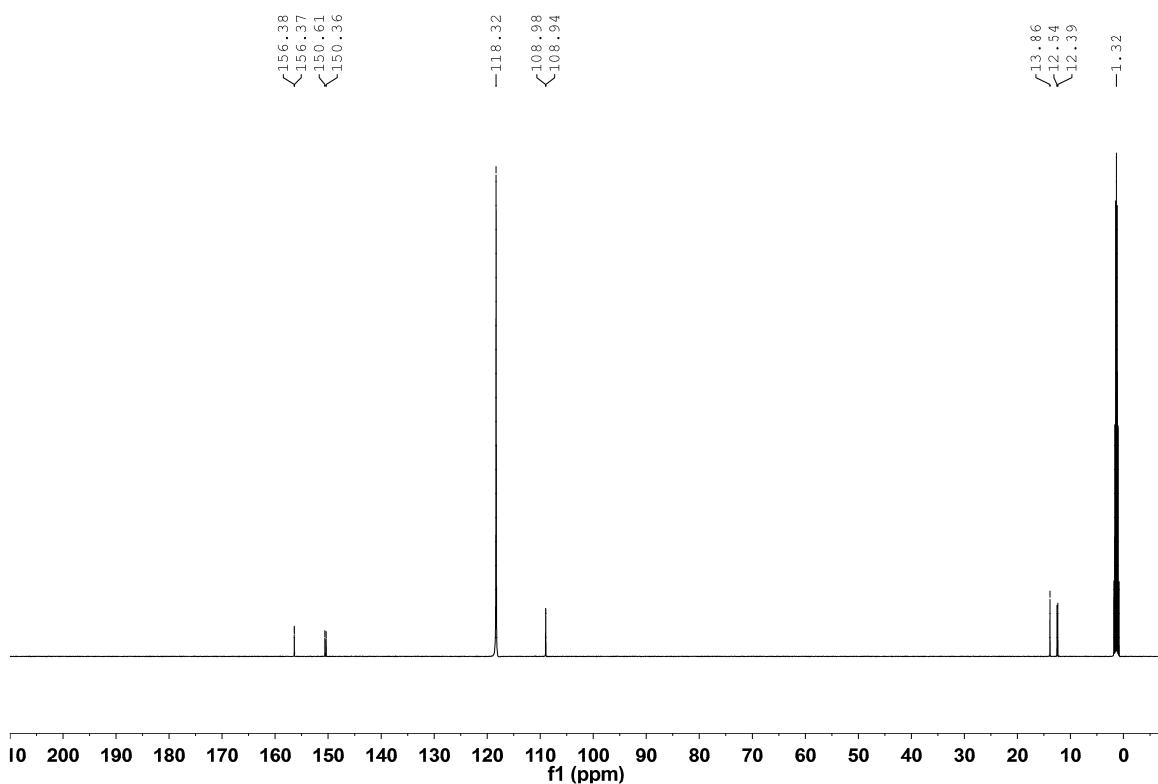


NMR spectra of acetonitrile tris(3,5-dimethylpyrazolyl)phosphine copper(I) hexafluorophosphate ($[\text{C}^{\text{Me}2}\text{Cu}(\text{NCMe})][\text{PF}_6]$)

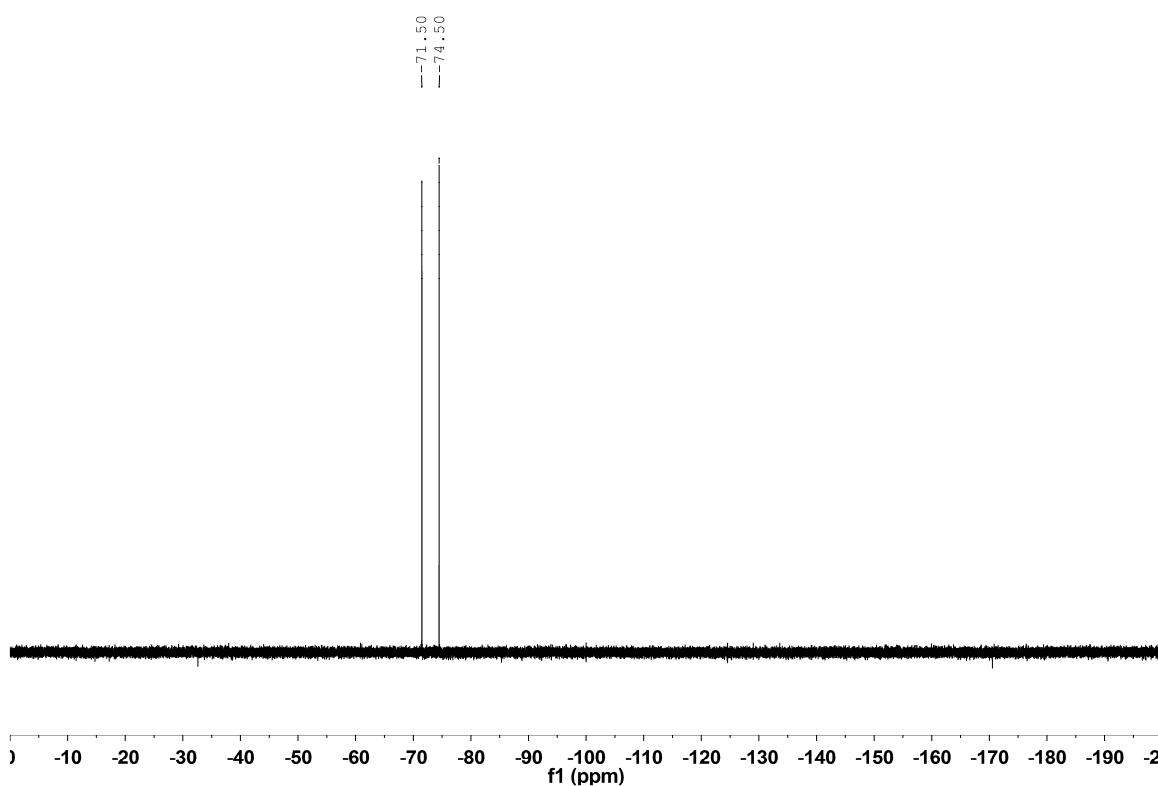
^1H NMR of $[\text{C}^{\text{Me}2}\text{Cu}(\text{NCMe})][\text{PF}_6]$ (500.2 MHz, CD_3CN)



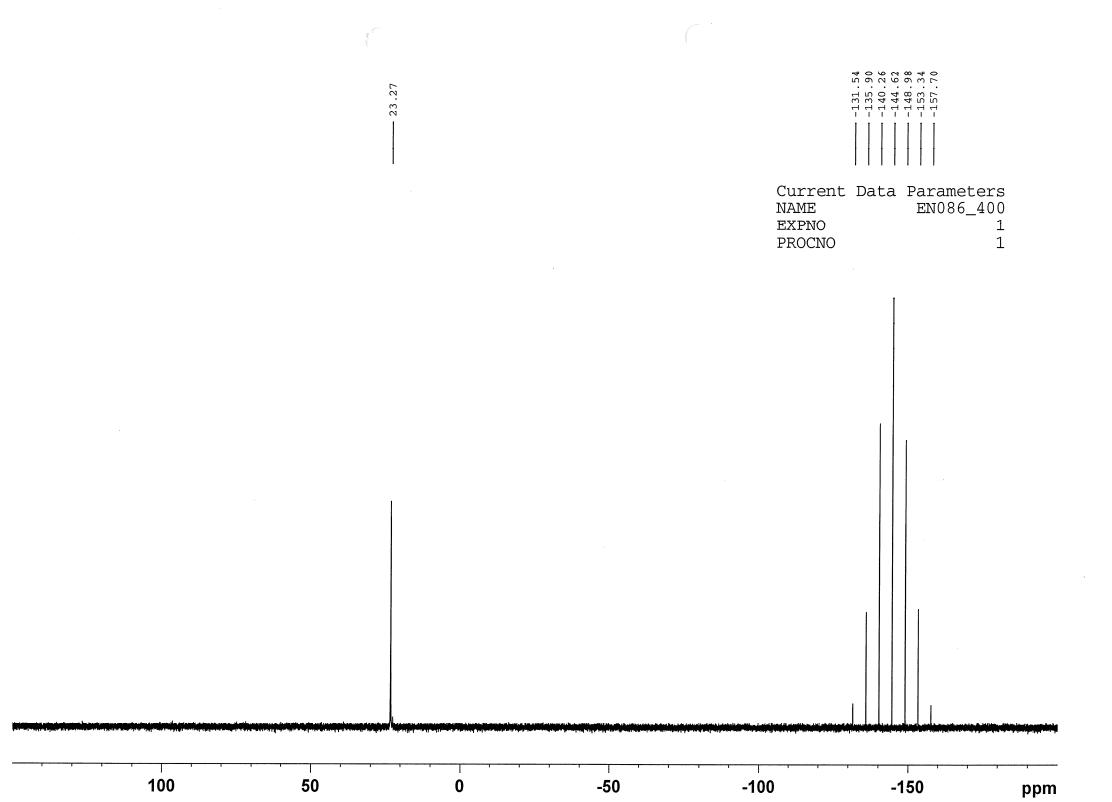
^{13}C NMR of $[\text{C}^{\text{Me}_2}\text{Cu}(\text{NCMe})]\text{[PF}_6]$ (125.8 MHz, CD_3CN)



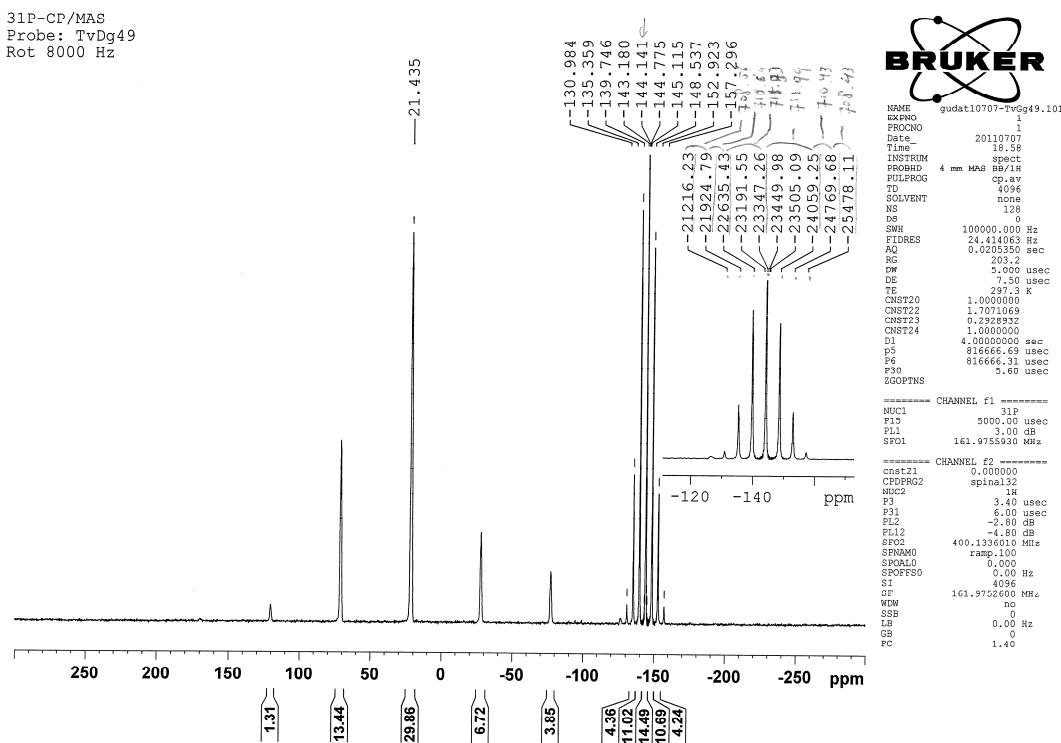
^{19}F NMR of $[\text{C}^{\text{Me}_2}\text{Cu}(\text{NCMe})]\text{[PF}_6]$ (235.3 MHz, CD_3CN)



^{31}P NMR of $[\text{C}^{\text{Me}_2}\text{Cu}(\text{NCMe})][\text{PF}_6]$ (162.0 MHz, CD_3CN)

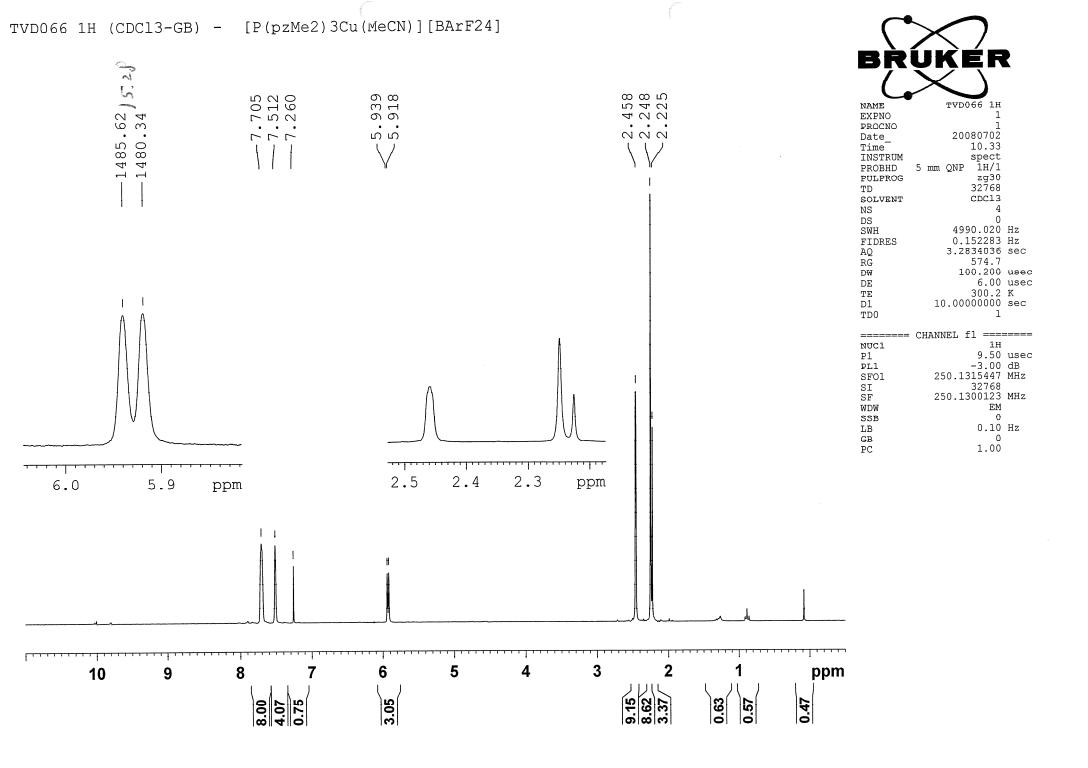


$^{31}\text{P}\{^1\text{H}\}$ CP/MAS NMR of $[\text{C}^{\text{Me}_2}\text{Cu}(\text{NCMe})][\text{PF}_6]$ (162.0 MHz)

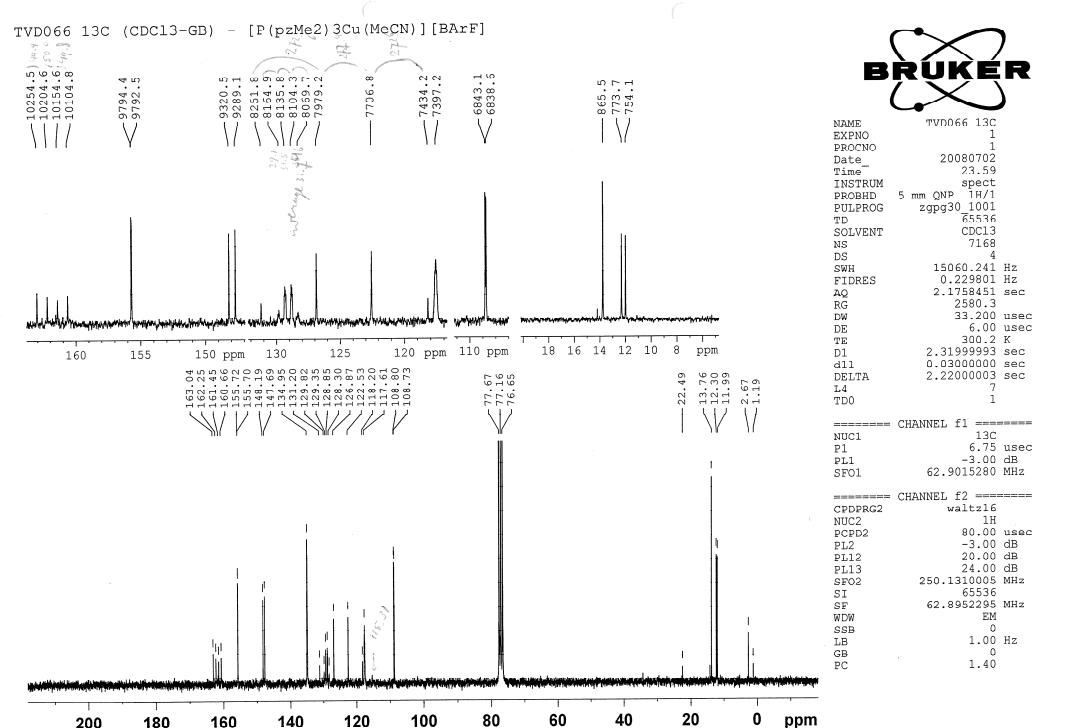


NMR spectra of acetonitrile tris(3,5-dimethylpyrazolyl)phosphine copper(I) tetrakis(3,5-bis(trifluoromethyl)-phenyl)borate ([C^{Me2}Cu(NCMe)][BArF₂₄])

¹H NMR of [C^{Me2}Cu(NCMe)][BArF₂₄] (250.1 MHz, CDCl₃)

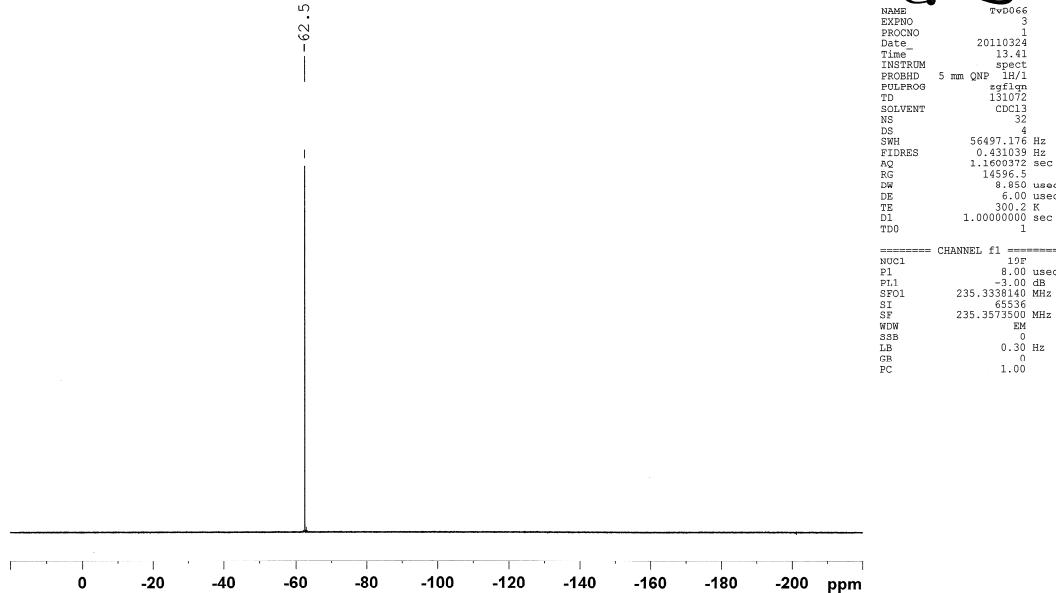


¹³C{¹H} NMR of [C^{Me2}Cu(NCMe)][BArF₂₄] (62.9 MHz, CDCl₃)



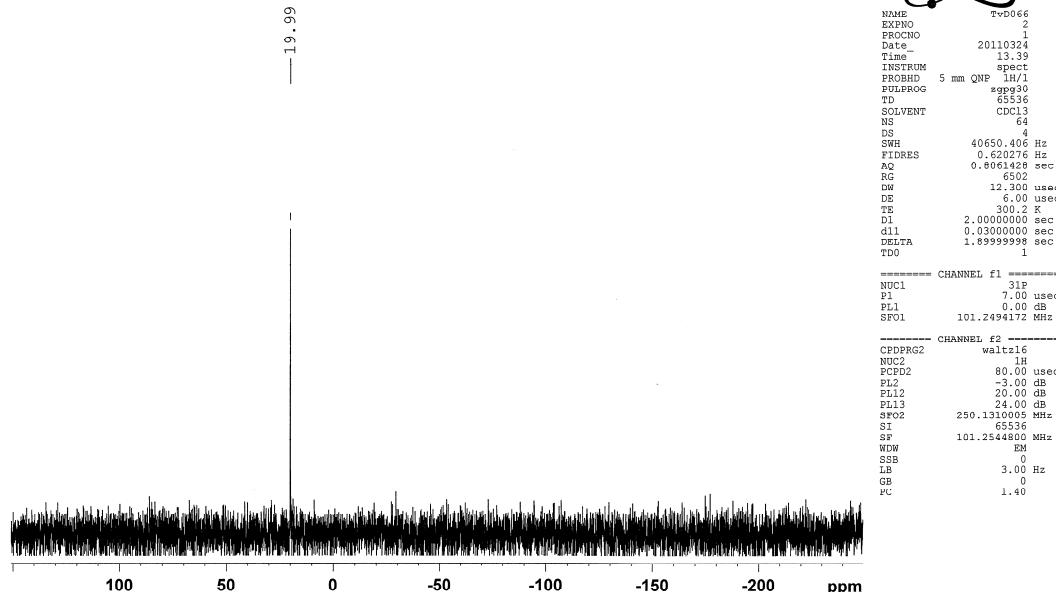
¹⁹F NMR of [C^{Me2}Cu(NCMe)][BArF₂₄] (235.3 MHz, CDCl₃)

19F (CDCl₃ GB)
 TvD066 P(Me2pz)3Cu(NCMe) BF24: Sample of 500expl-3
 F19_NO_processing CDC13 {D:\NMRDATA} NielsT 41



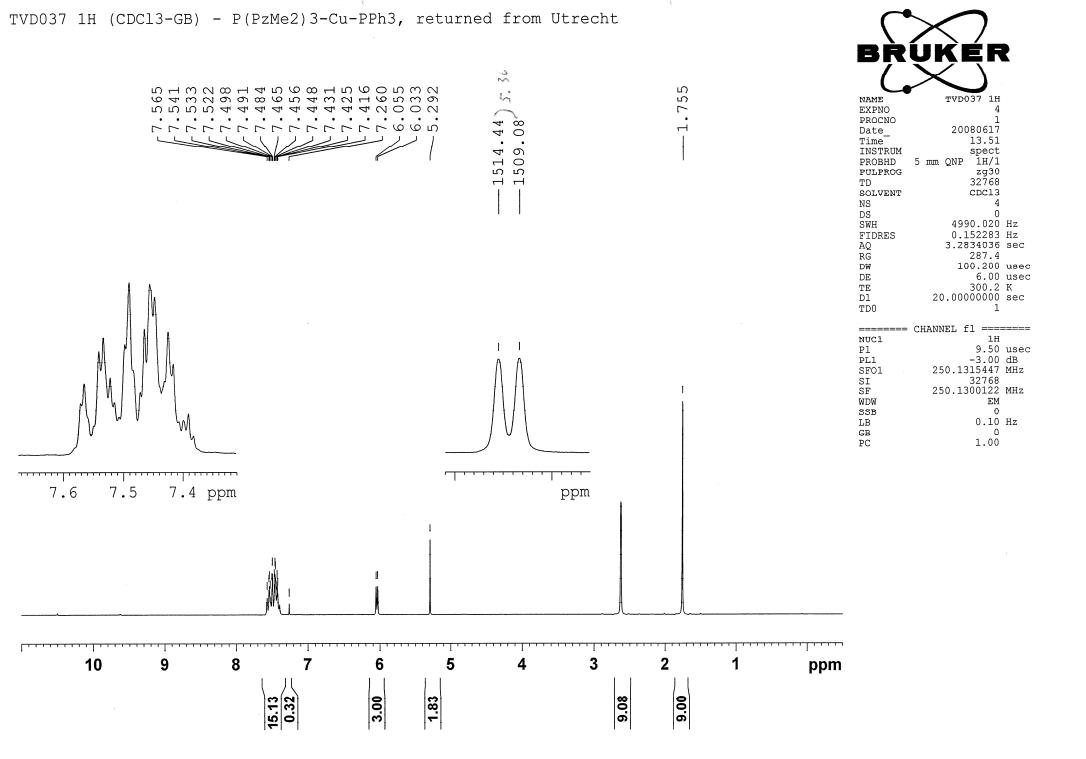
³¹P NMR of [C^{Me2}Cu(NCMe)][BArF₂₄] (101.3 MHz, CDCl₃)

31P{1H} (CDCl₃ GB)
 TvD066 P(Me2pz)3Cu(NCMe) BF24: Sample of 500expl-3
 P31CPD_NO_d-solvent_NO_processing CDC13 {D:\NMRDATA} NielsT 41

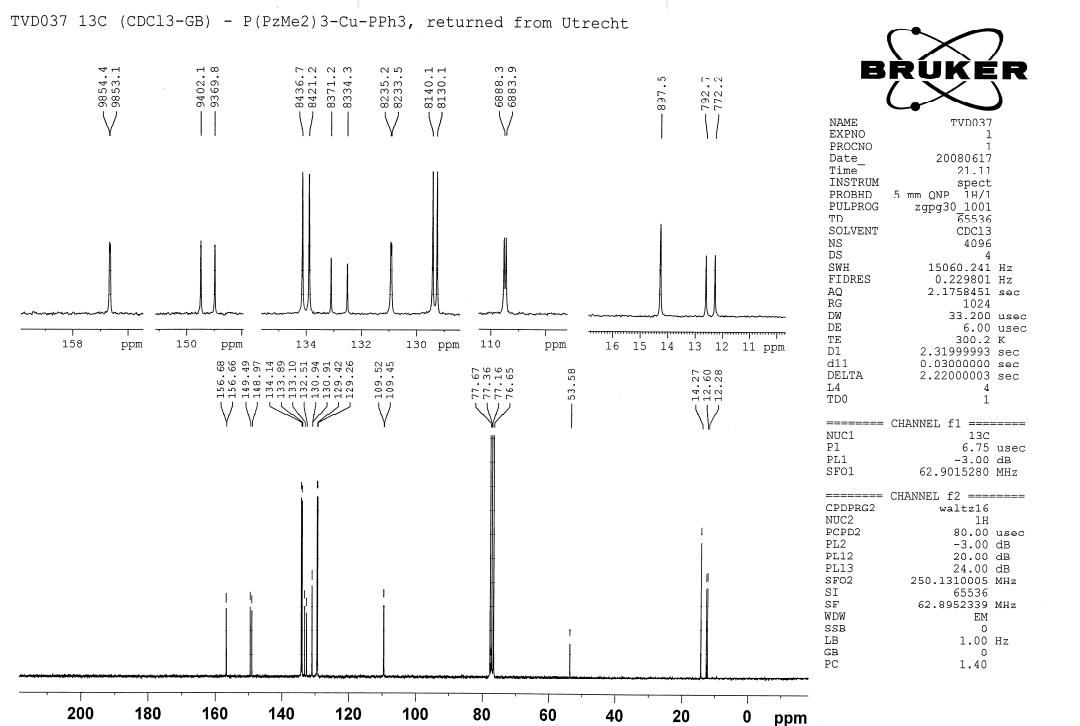


NMR spectra of tris(3,5-dimethylpyrazolyl)phosphine triphenylphosphine copper(I) hexafluorophosphate ($[C^{Me_2}Cu(PPh_3)][PF_6]$)

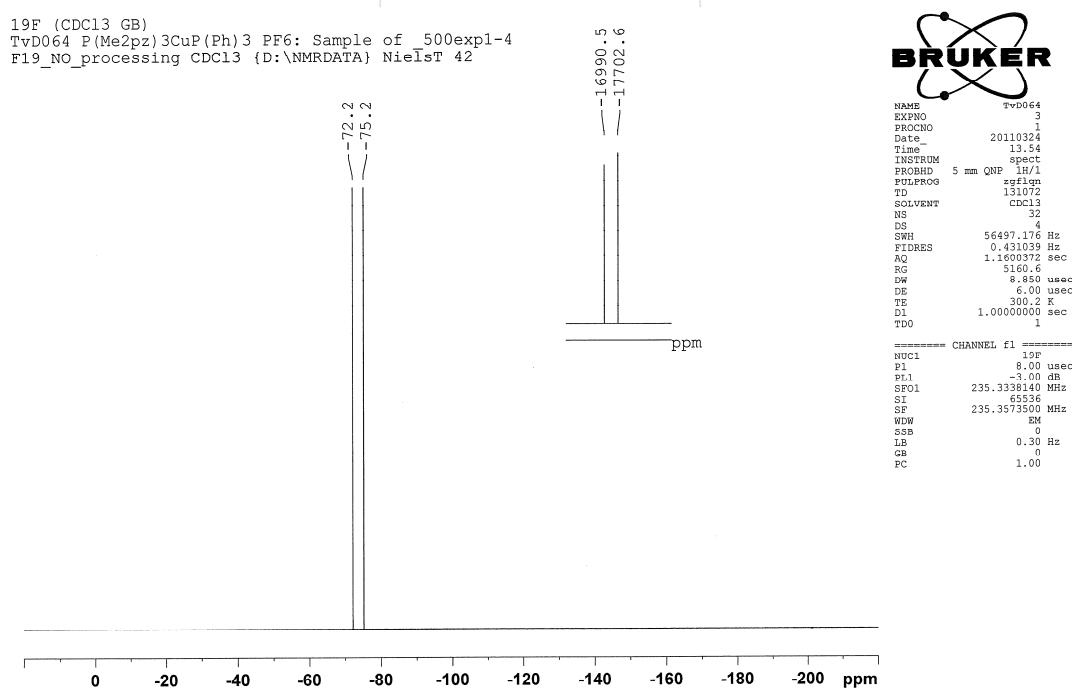
1H NMR of $[C^{Me_2}Cu(PPh_3)][PF_6]$ (250.1 MHz, $CDCl_3$)



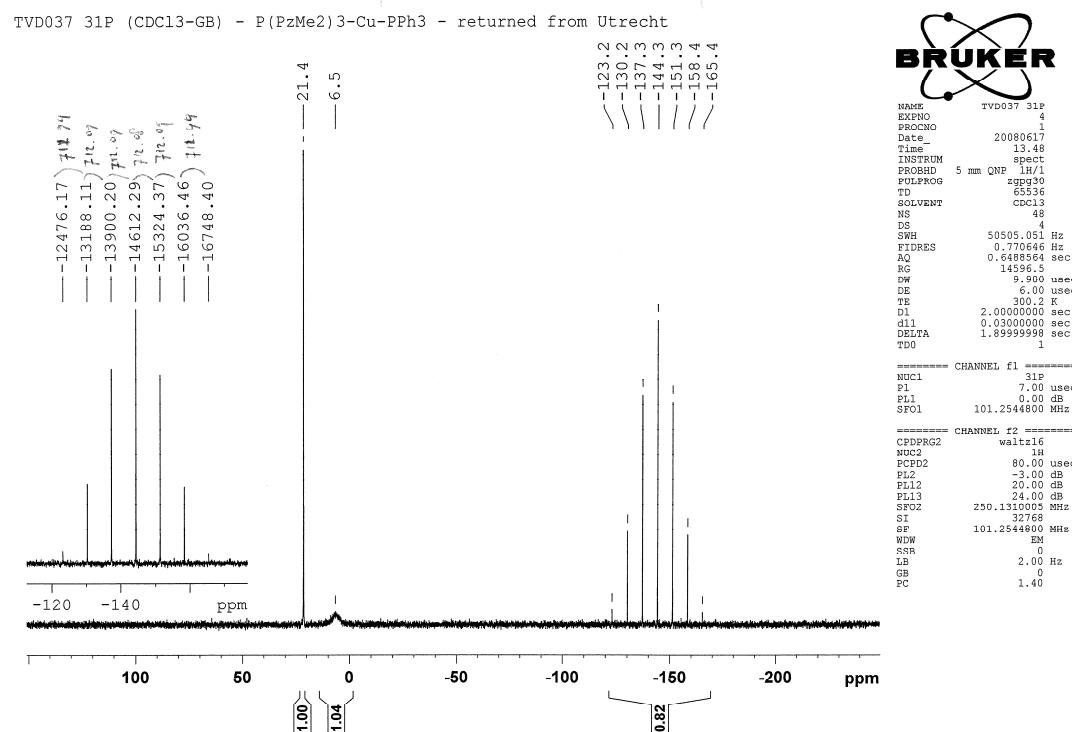
$^{13}C\{^1H\}$ NMR of $[C^{Me_2}Cu(PPh_3)][PF_6]$ (62.9 MHz, $CDCl_3$)



¹⁹F NMR of [C^{Me2}Cu(PPh₃)][PF₆] (235.3 MHz, CDCl₃)

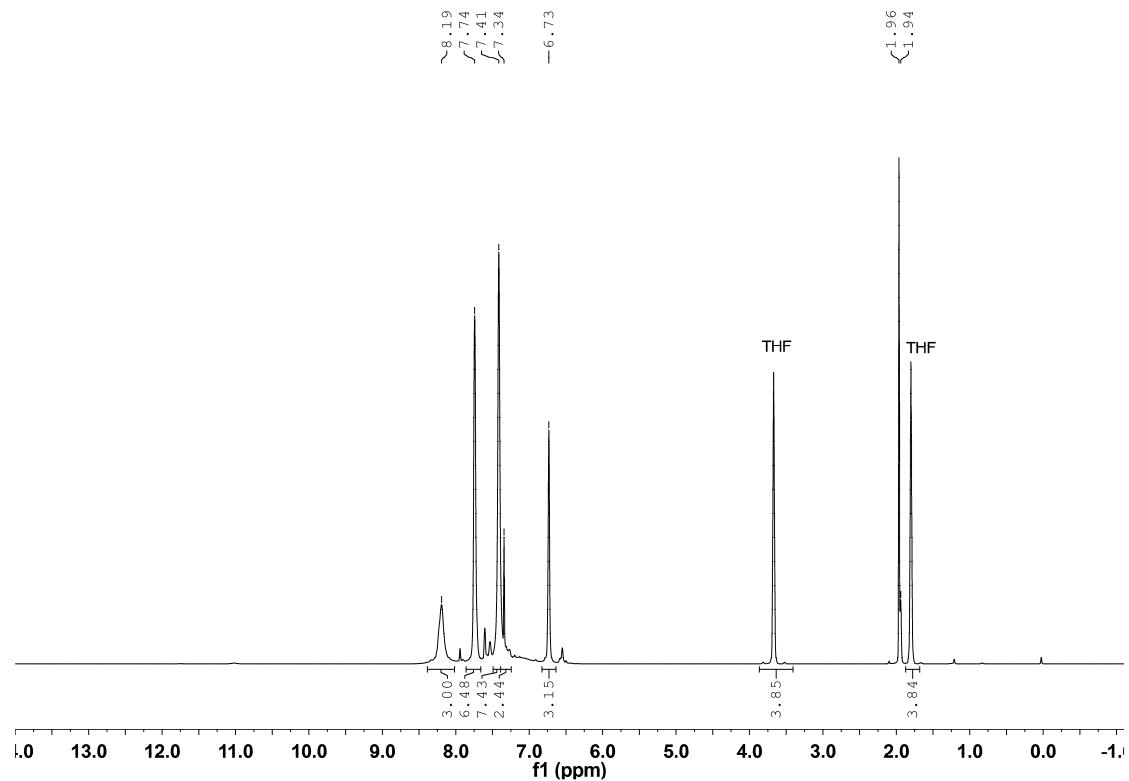


³¹P NMR of [C^{Me2}Cu(PPh₃)][PF₆] (101.3 MHz, CDCl₃)

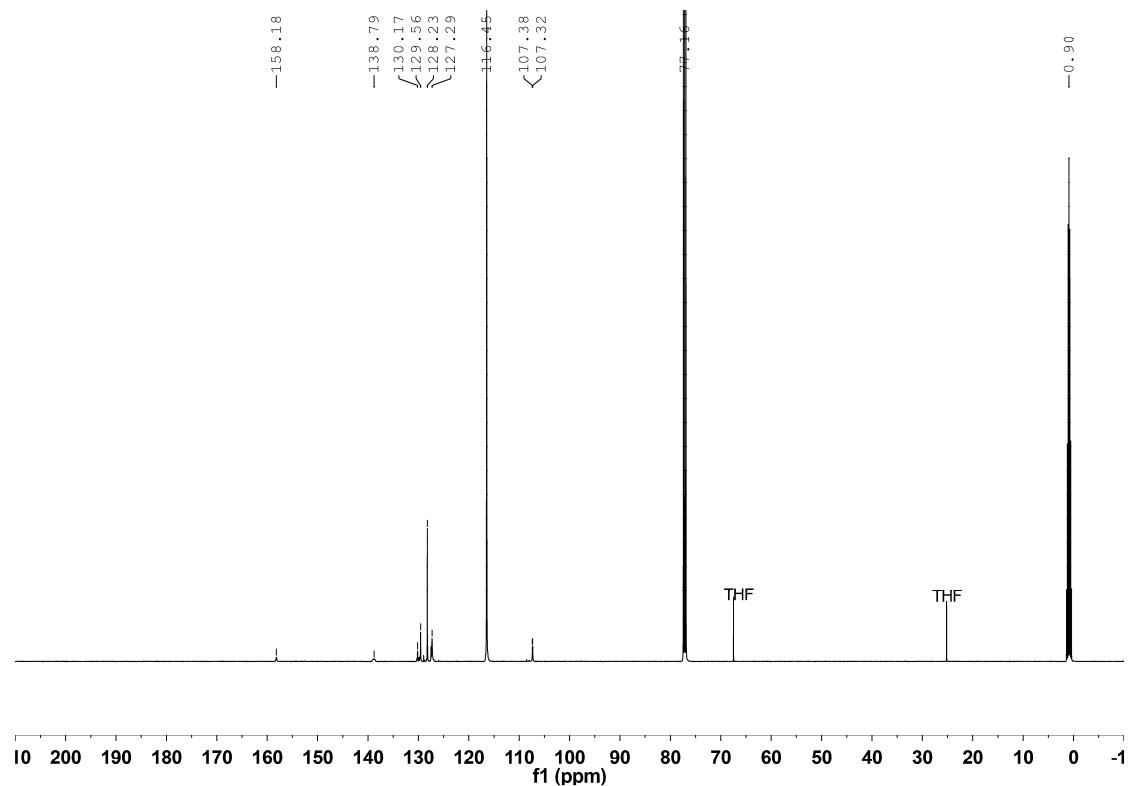


NMR spectra of acetonitrile tris(3-phenylpyrazolyl)phosphine copper(I) hexafluorophosphate ($[\text{C}^{\text{Ph}}\text{Cu}(\text{NCMe})][\text{PF}_6]$)

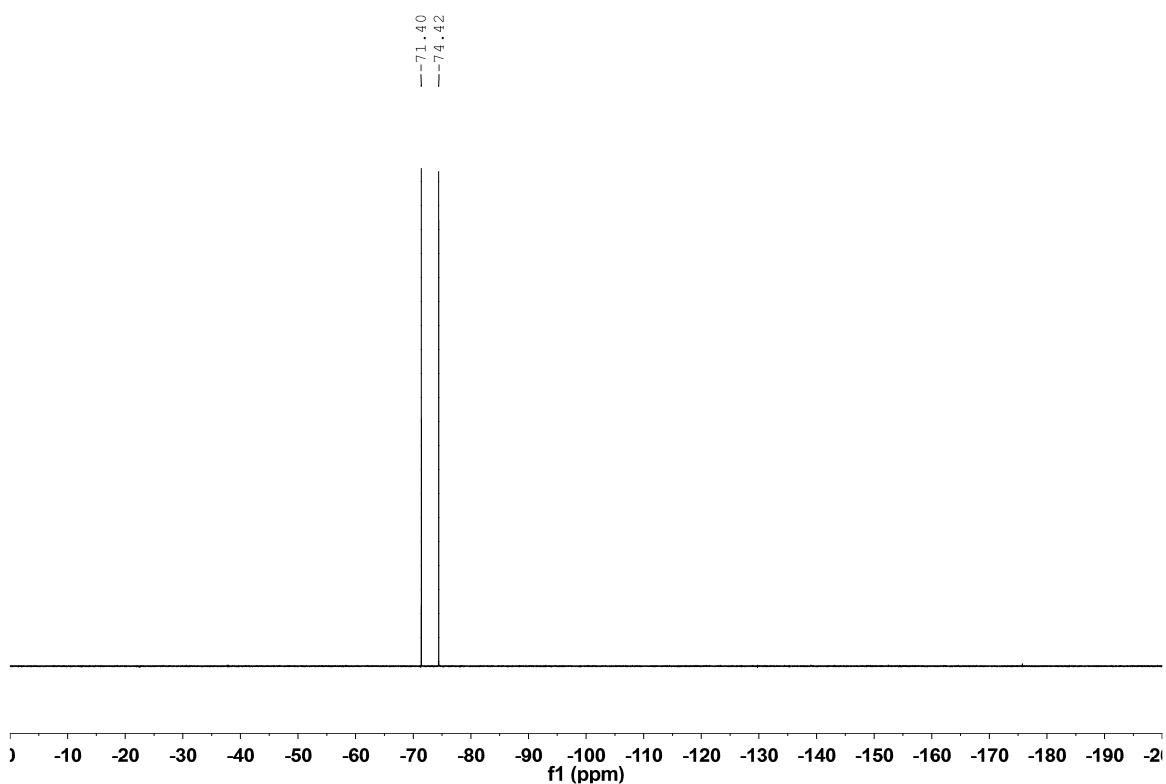
^1H NMR of $[\text{C}^{\text{Ph}}\text{Cu}(\text{NCMe})][\text{PF}_6]$ (500.2 MHz, $\text{CDCl}_3 + \text{CD}_3\text{CN}$)



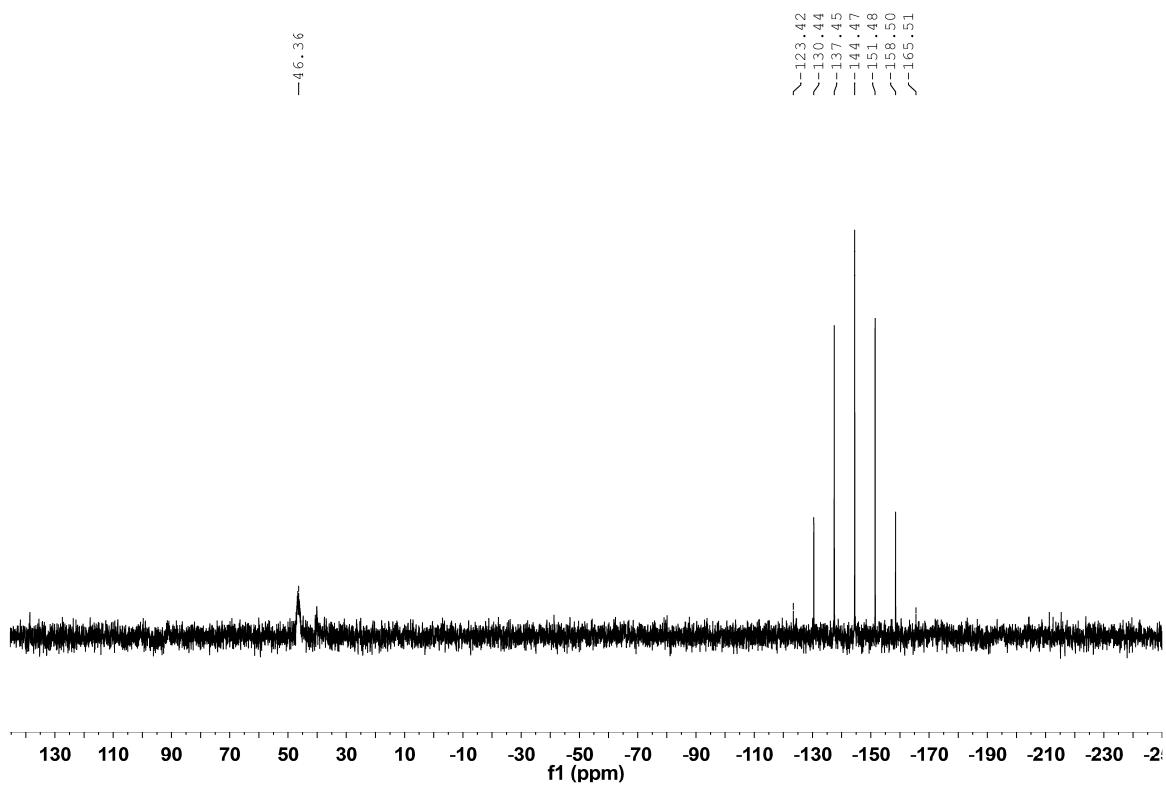
$^{13}\text{C}\{^1\text{H}\}$ NMR of $[\text{C}^{\text{Ph}}\text{Cu}(\text{NCMe})][\text{PF}_6]$ (125.8 MHz, $\text{CDCl}_3 + \text{CD}_3\text{CN}$)



¹⁹F-NMR of [C^{Ph}Cu(NCMe)][PF₆] (235.4 MHz, CDCl₃+CD₃CN)

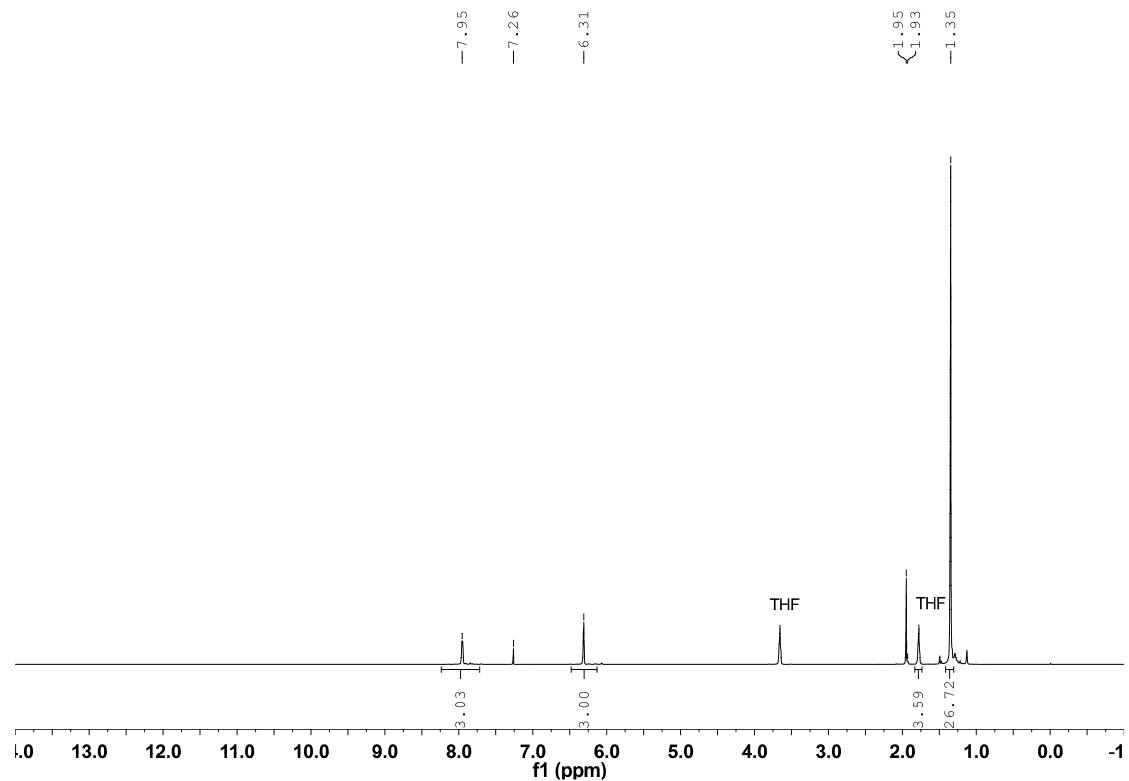


³¹P-NMR of [C^{Ph}Cu(NCMe)][PF₆] (162.0 MHz, CDCl₃+CD₃CN)

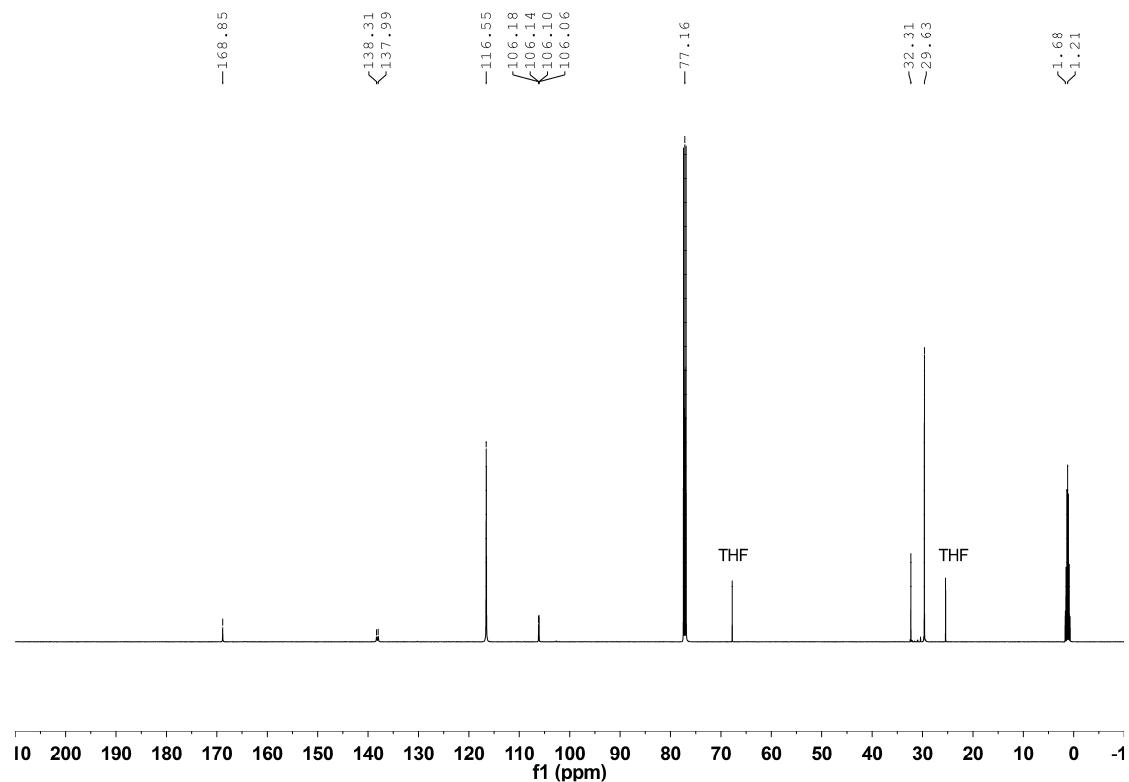


NMR spectra of acetonitrile tris(3-*tert*-butylpyrazolyl)phosphine copper(I) hexafluorophosphate ($[\text{C}^{t\text{-Bu}}\text{Cu}(\text{NCMe})]\text{[PF}_6\text{]}$)

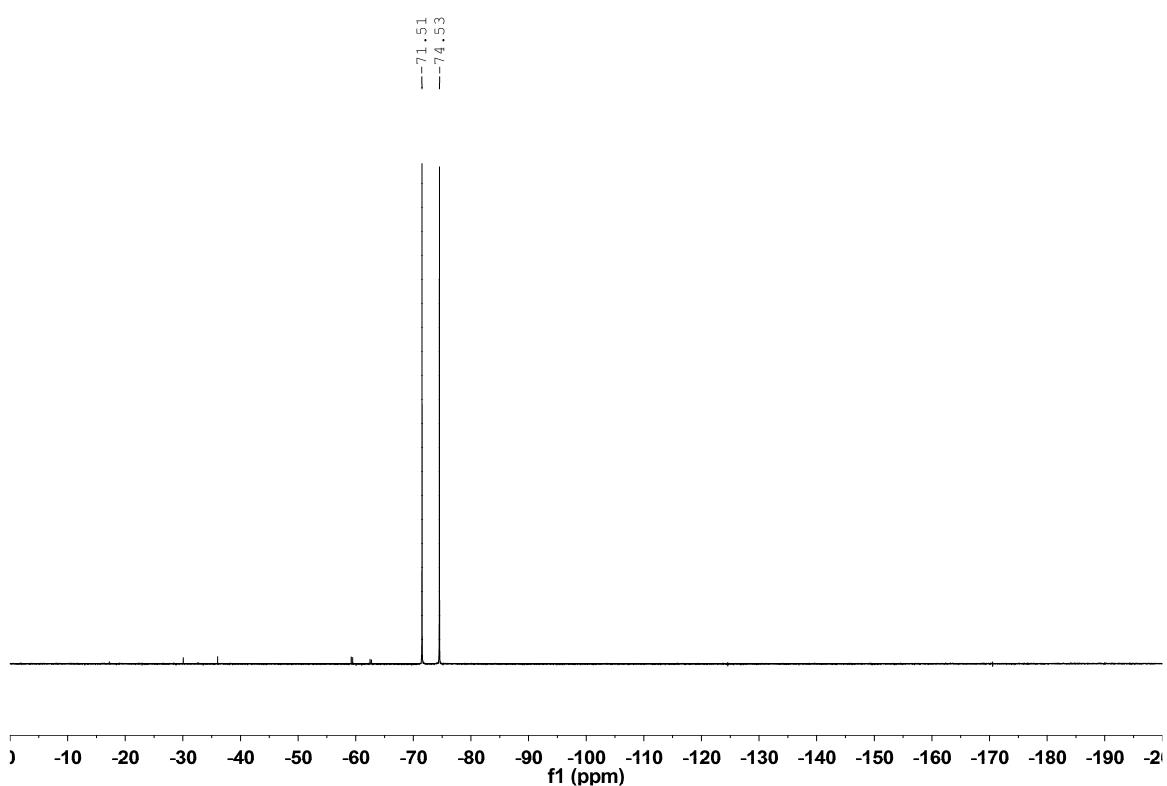
^1H -NMR of $[\text{C}^{t\text{-Bu}}\text{Cu}(\text{NCMe})]\text{[PF}_6\text{]}$ (500.2 MHz, $\text{CDCl}_3+\text{CD}_3\text{CN}$)



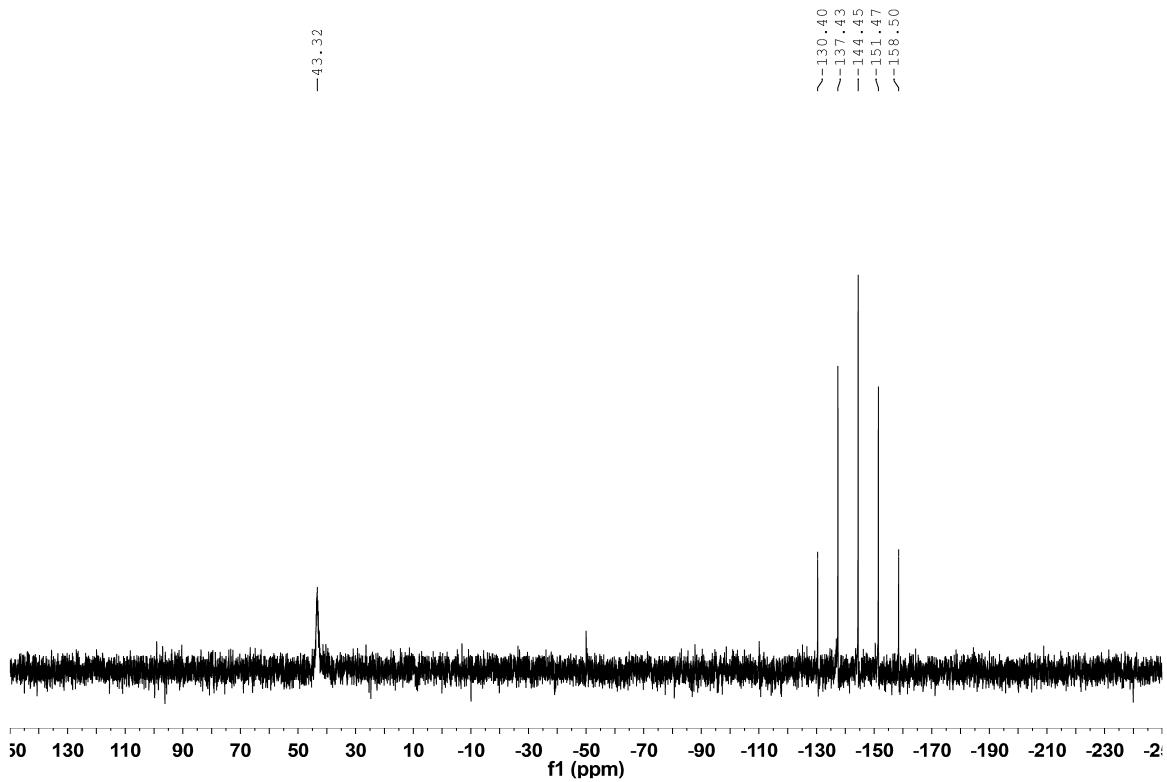
^{13}C -NMR of $[\text{C}^{t\text{-Bu}}\text{Cu}(\text{NCMe})]\text{[PF}_6\text{]}$ (125.8 MHz, $\text{CDCl}_3+\text{CD}_3\text{CN}$)



¹⁹F-NMR (235.4 MHz, CDCl₃+CD₃CN)

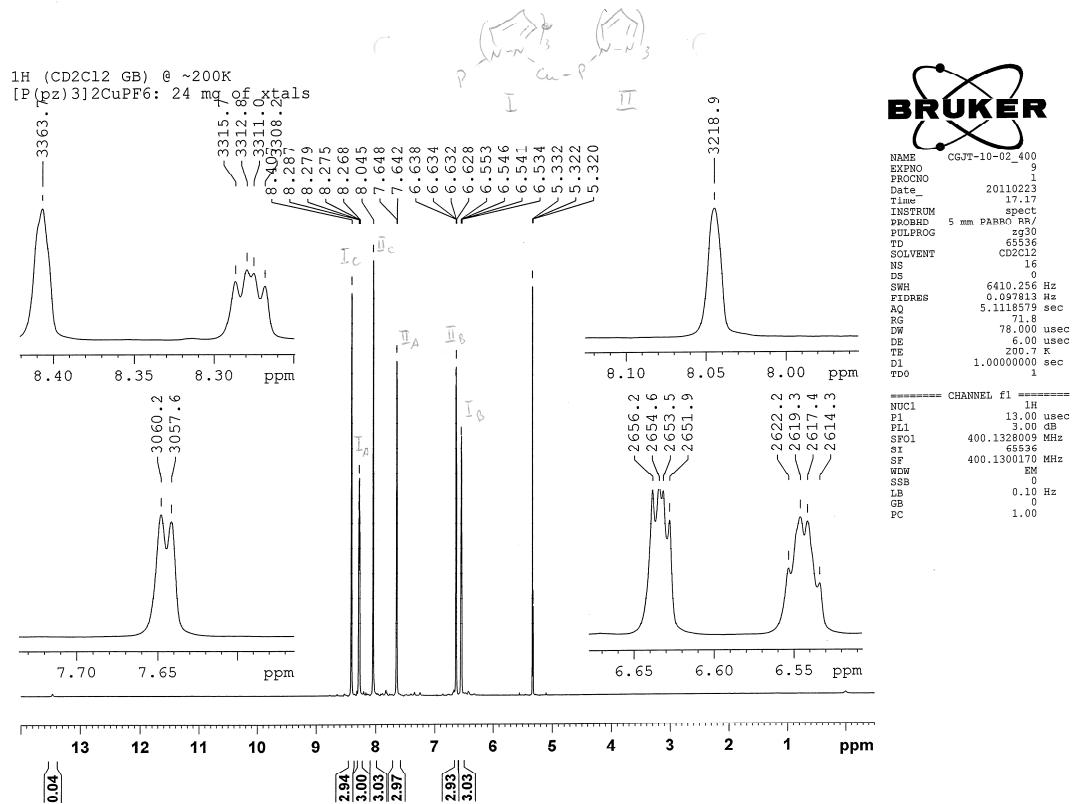


³¹P-NMR (162.0 MHz, CDCl₃+CD₃CN)

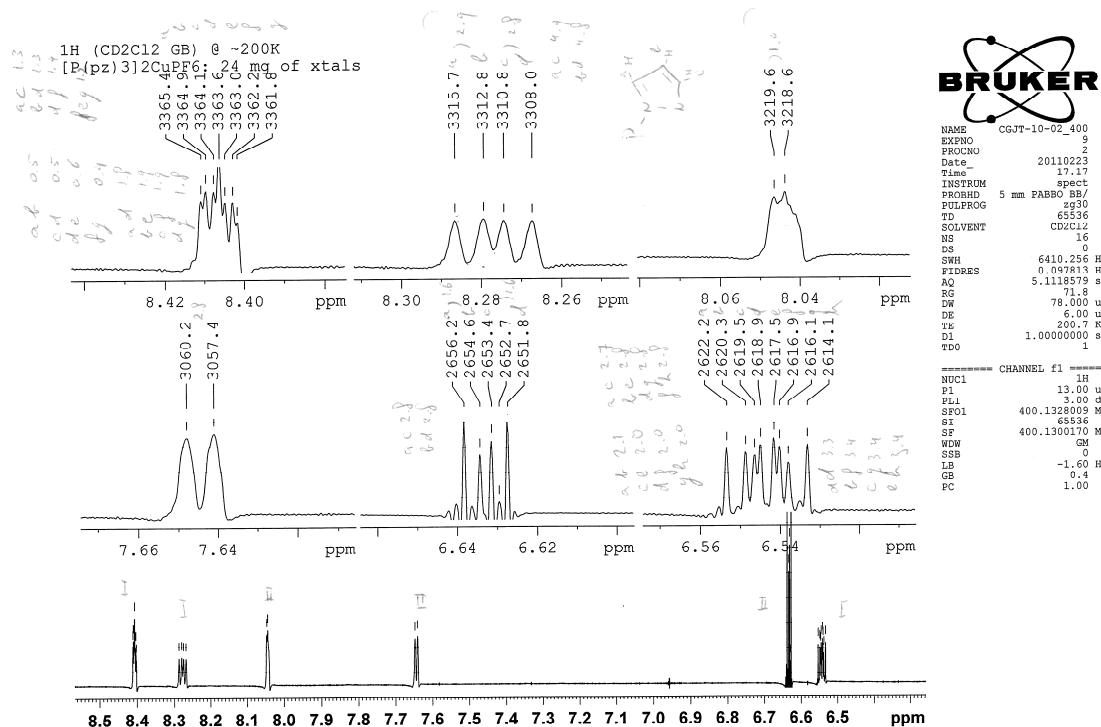


NMR spectra of bis(tris(pyrazolyl)phosphine)copper(I) hexafluorophosphate $[(\text{CH}_3)_2\text{Cu}][\text{PF}_6]$

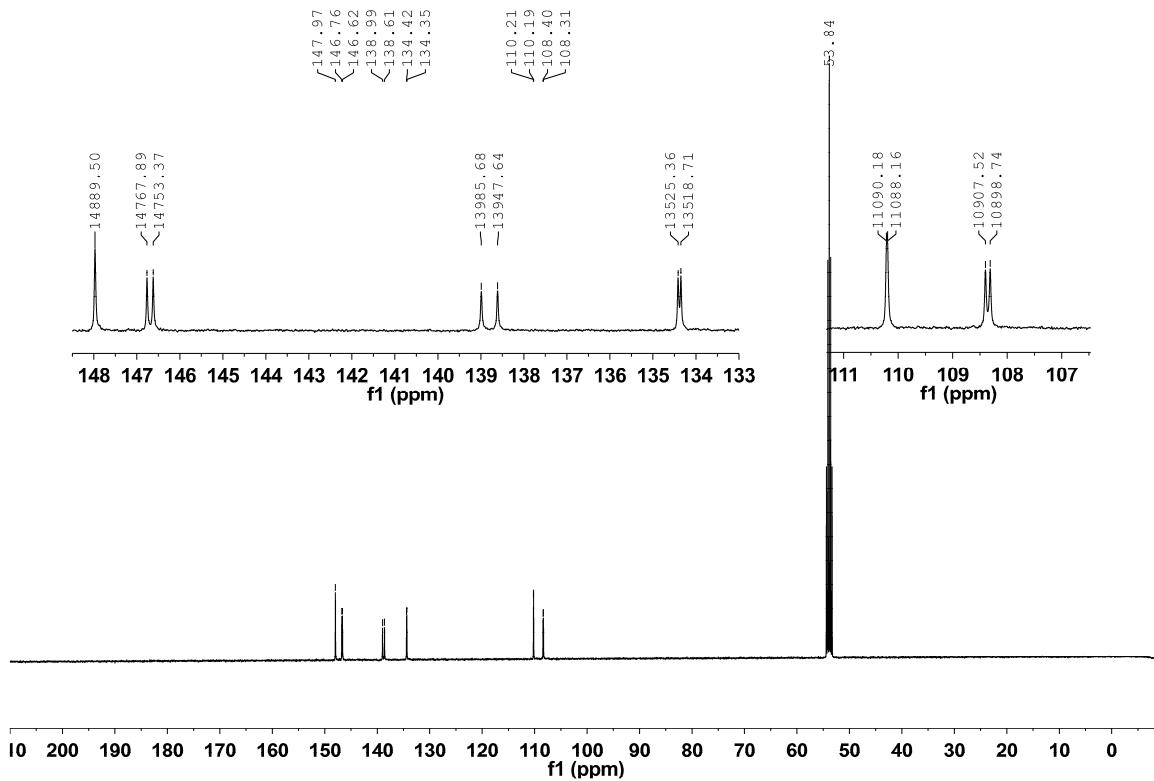
¹H NMR of $[(\text{CH}_3)_2\text{Cu}][\text{PF}_6]$ (400.1 MHz, 201 K, CD_2Cl_2)



¹H NMR of [(C^H)₂Cu][PF₆] (400.1 MHz, 201 K, CD₂Cl₂) after line narrowing

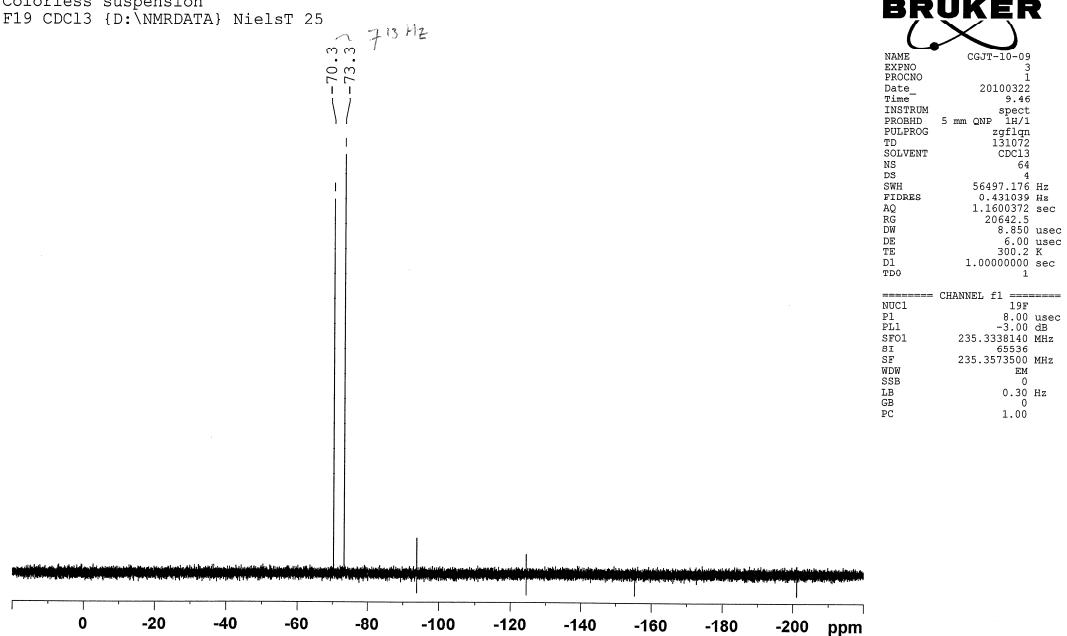


$^{13}\text{C}\{\text{H}\}$ NMR of $[(\text{CH})_2\text{Cu}][\text{PF}_6]$ (100.6 MHz, 201 K, CD_2Cl_2)

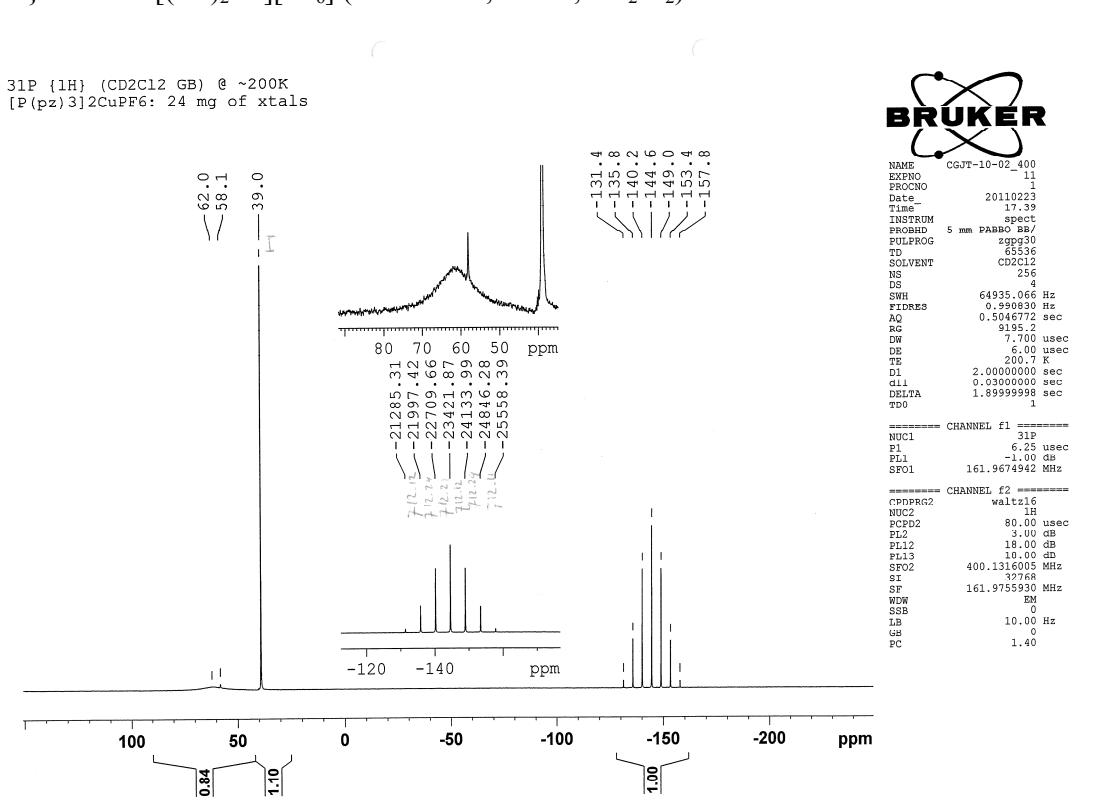


^{19}F NMR of $[(\text{CH})_2\text{Cu}][\text{PF}_6]$ (235.3 MHz, CDCl_3)

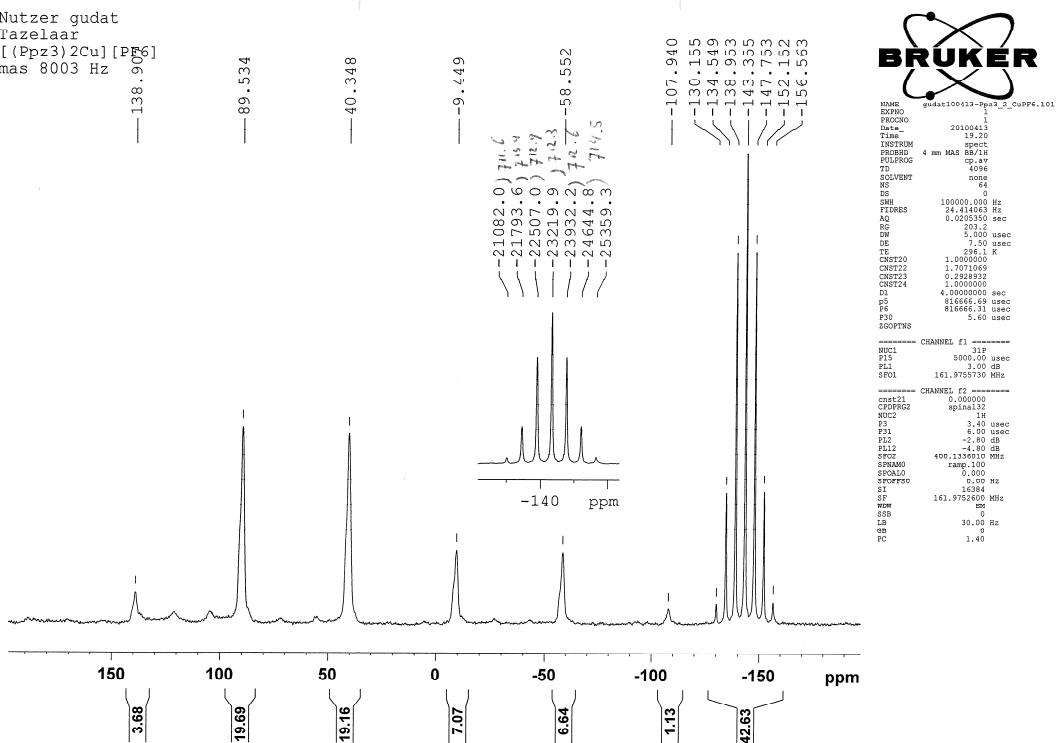
19F (CDCl_3 GB)
 $\text{P}(\text{pz})_3\text{CuPF}_6 + \text{P}(\text{pz})_3$: 11 mg of residue after removing volatiles
 Colorless suspension
 F19 CDCl_3 {D:\NMRDATA} NielsT 25



$^{31}\text{P}\{\text{H}\}$ NMR of $[(\text{C}^{\text{H}})_2\text{Cu}][\text{PF}_6]$ (162.0 MHz, 201 K, CD_2Cl_2)

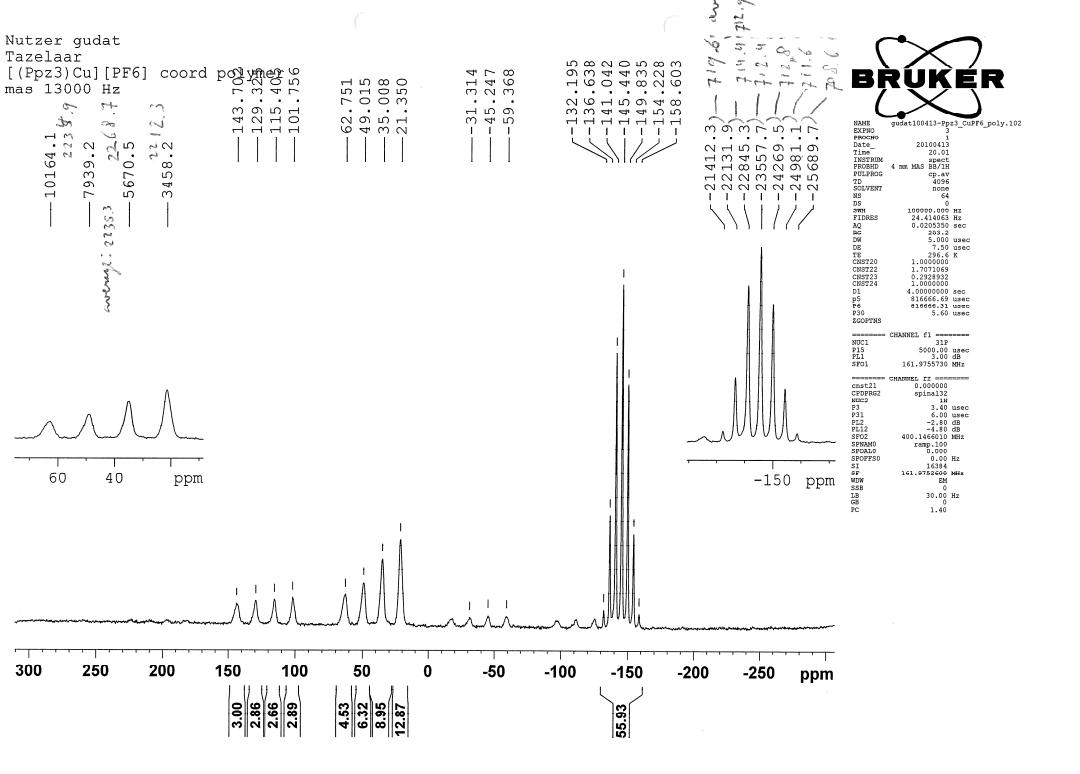


³¹P{¹H} CP/MAS NMR of [(CH)₂Cu][PF₆] (162.0 MHz)

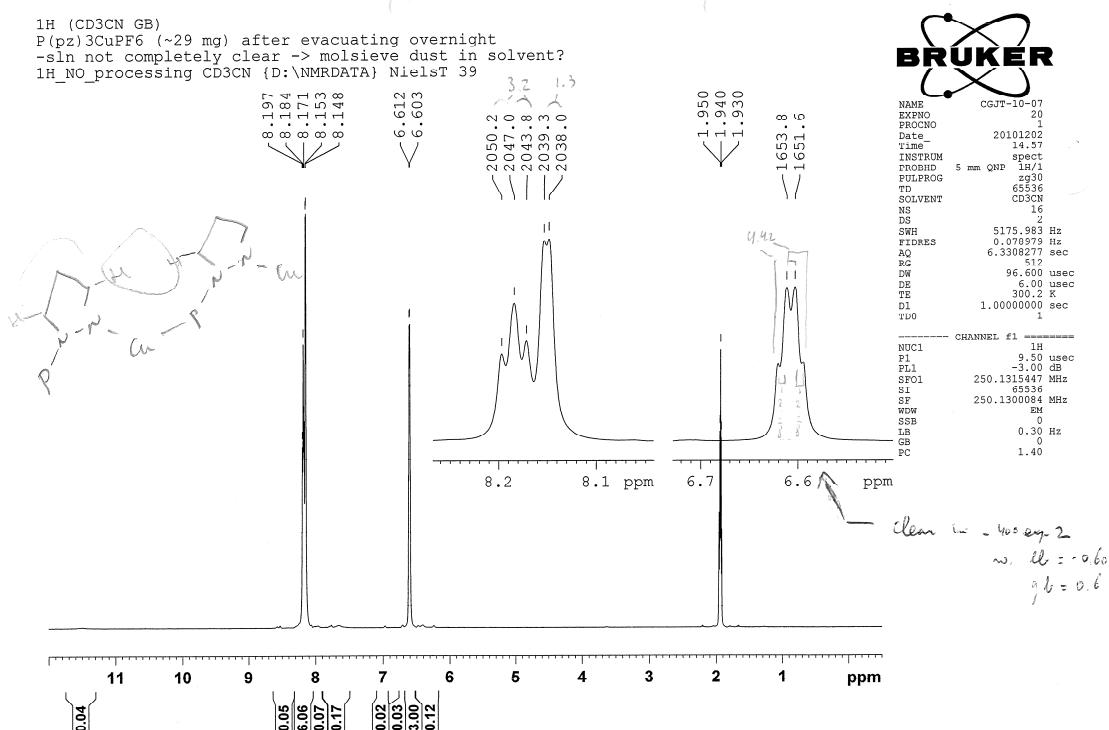


NMR spectra of tris(pyrazolyl)phosphine copper(I) hexafluorophosphate ($([C^H Cu][PF_6])_n$)

$^{31}P\{^1H\}$ MAS NMR of $([C^H Cu][PF_6])_n$ (162.0 MHz)

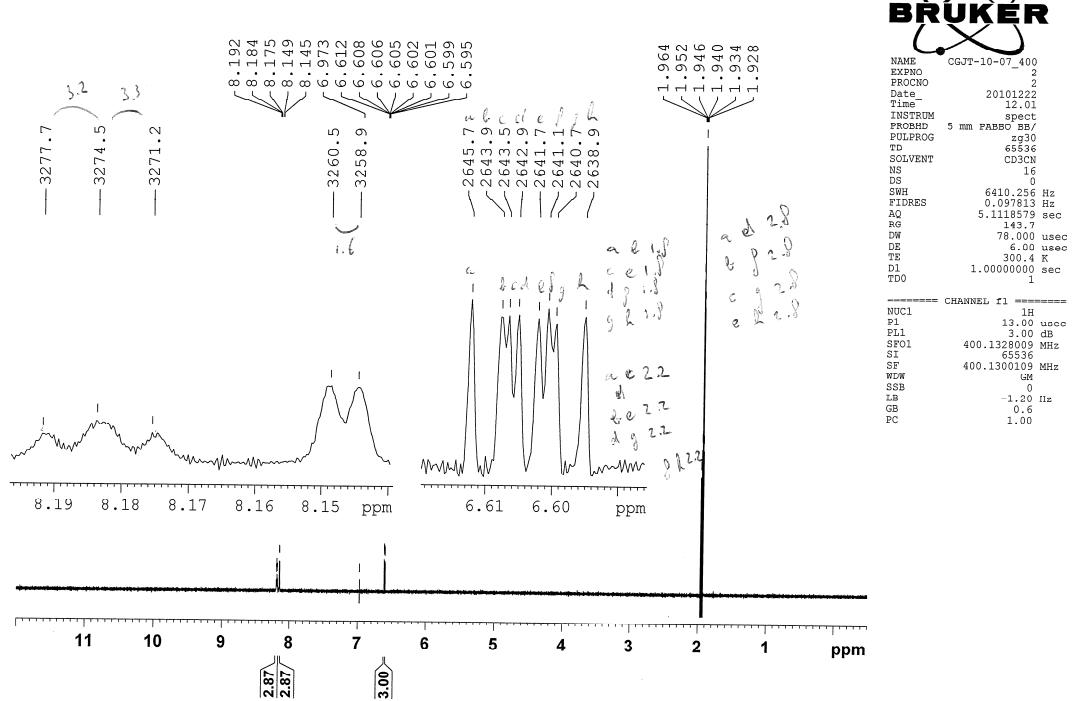


1H NMR of $([C^H Cu][PF_6])_n$ (400.1 MHz, CD₃CN)



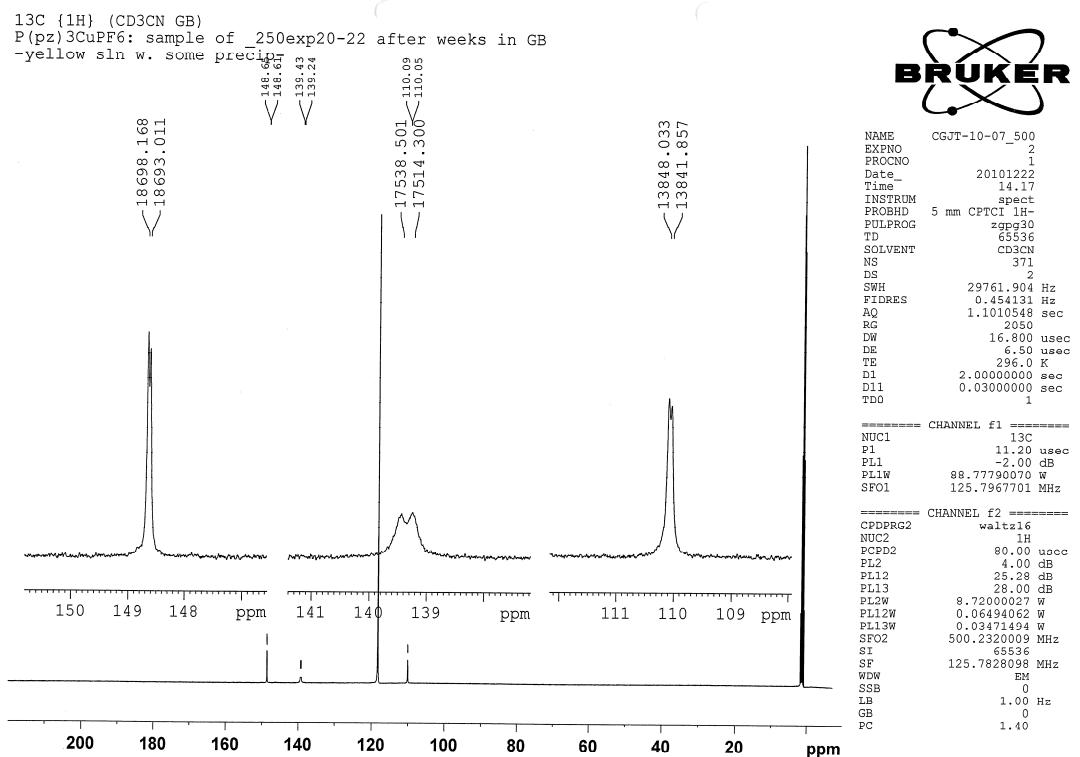
¹H NMR of ([C^HCu][PF₆])_n (400.1 MHz, CD₃CN) after line narrowing

1H (CD₃CN GB)
P(pz)3CuPF₆: Yellow sln after standing in GB for weeks, same sample as _250exp2



¹³C{¹H} NMR of ([C^HCu][PF₆])_n (125.8 MHz, CD₃CN)

13C {1H} (CD₃CN GB)
P(pz)3CuPF₆: sample of _250exp20-22 after weeks in GB
-yellow sln w. some precip.



¹⁹F NMR of ([C^HCu][PF₆])_n (235.3 MHz, CD₃CN)

19F (CD₃CN GB)
 P(pz)3CuPF₆ (~29 mg) after evacuating overnight
 -sln not completely clear -> molsieve dust in solvent?
 F19 CD₃CN {D:\NMRDATA} NielsT 39

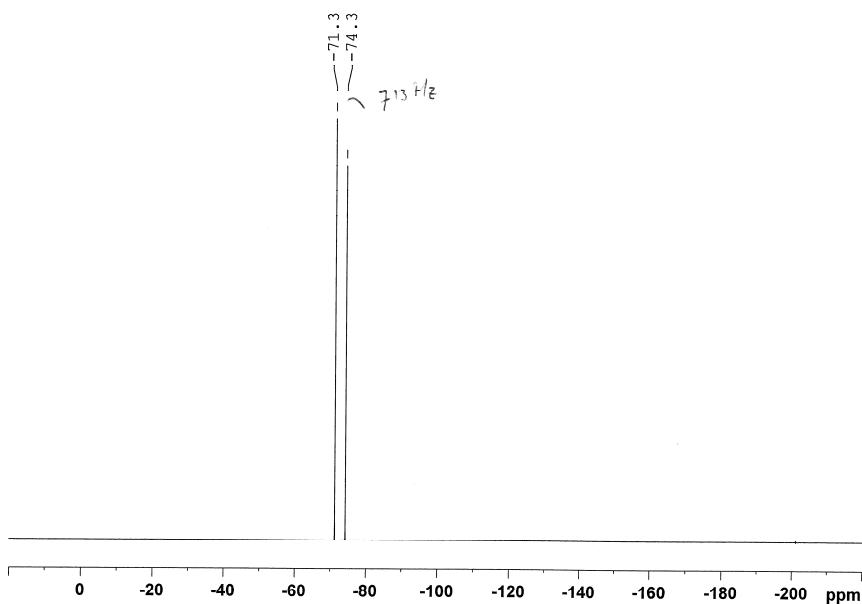


```

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EXPNO     22
PROCNO    1
Date_     20101202
Time_     15.03
INSTRUM   spect
PROBHD   5 mm QNP 1H/1
PULPROG  zgfgqn
TD        131072
SOLVENT   CD3CN
NS        32
DS        4
SWH      56497.176 Hz
FIDRES   0.439561 Hz
AQ        1.1600372 sec
RG        2298.8
DW        8.850 usec
DE        6.00 usec
TE        300.0 K
T1        1.0000000 sec
TD0      1

===== CHANNEL f1 =====
NUC1      19F
P1        8.00 usec
PL1      -3.00 dB
SF01     235.335140 MHz
SI        65536
SF       235.3573500 MHz
WDW      EM
SSB      0
LB       0.30 Hz
GB       0
PC       1.00

```



³¹P{¹H} NMR of ([C^HCu][PF₆])_n (101.3 MHz, CD₃CN)

31P CDP (CD₃CN CB)
 P(pz)3CuPF₆ (~29 mg) after evacuating overnight
 -sln not completely clear -> molsieve dust in solvent?
 31PCPD_NO_processing CD₃CN {D:\NMRDATA} NielsT 39

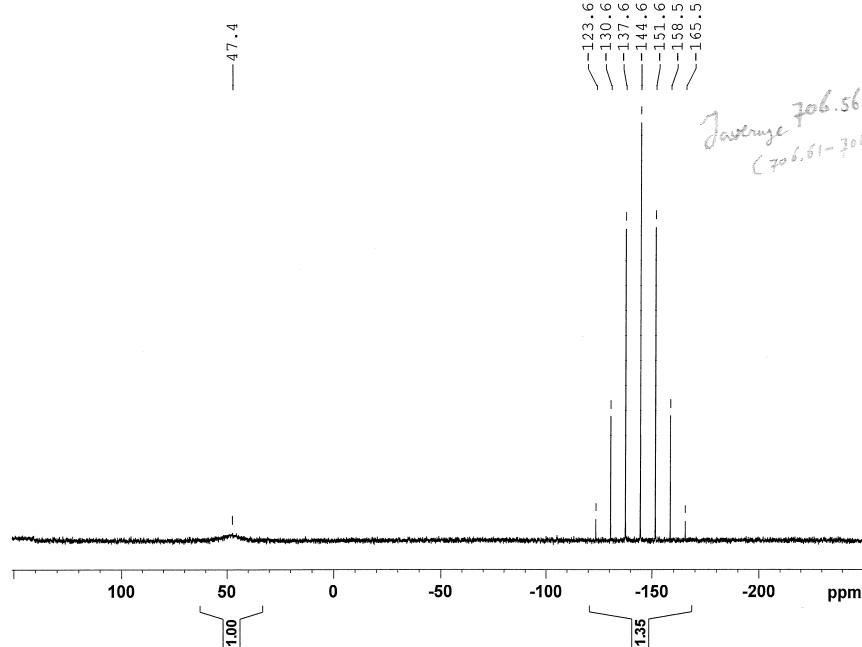


```

NAME      CGJT-10-07
EXPNO     21
PROCNO    1
Date_     20101202
Time_     15.03
INSTRUM   spect
PROBHD   5 mm QNP 1H/1
PULPROG  zgpg30
TD        65536
SOLVENT   CD3CN
NS        96
DS        4
SWH      40650.406 Hz
FIDRES   0.620276 Hz
AQ        0.8061428 sec
RG        265.5
DW        12.300 usec
DE        6.00 usec
TE        300.2 K
T1        2.0000000 sec
TD0      1
d1       0.03000000 sec
DETA     1.8999998 sec
TDO      1

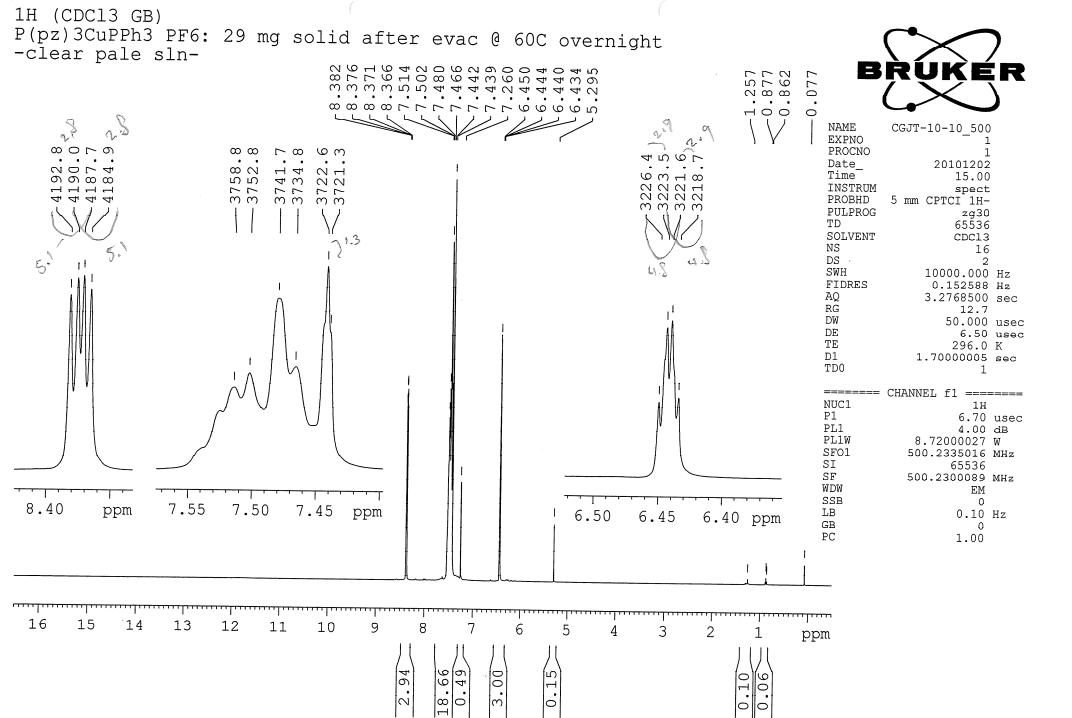
===== CHANNEL f1 =====
NUC1      31P
P1        7.00 usec
PL1      0.00 dB
SF01     101.2494172 MHz
===== CHANNEL f2 =====
CPDPG2   waltz16
NUC2      1H
PCPD2    80.00 usec
PL2      -3.00 dB
PL12     20.00 dB
PL13     24.00 dB
SF02     250.1133333 MHz
SI        65536
SF       101.2544800 MHz
WDW      EM
SSB      0
LB       3.00 Hz
GB       0
PC       1.40

```

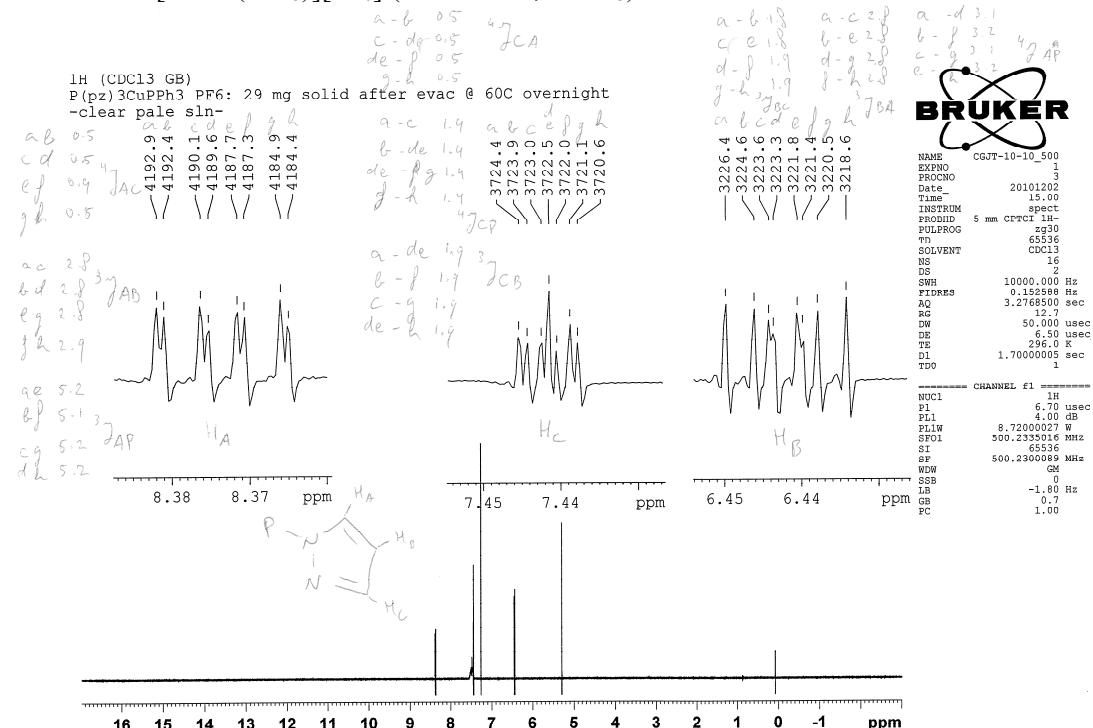


NMR spectra of triphenylphosphine tris(pyrazolyl)phosphine copper(I) hexafluorophosphate ($[C^H\text{Cu}(\text{PPh}_3)][\text{PF}_6]$)

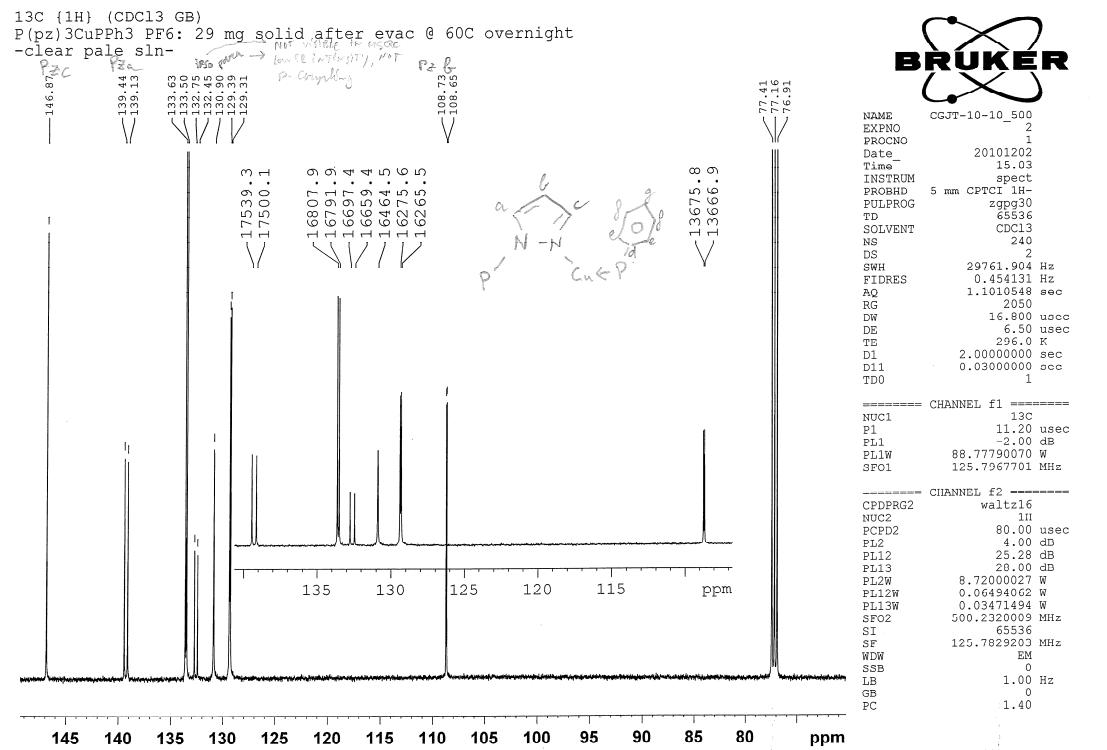
^1H NMR of $[C^H\text{Cu}(\text{PPh}_3)][\text{PF}_6]$ (500.2 MHz, CDCl_3)



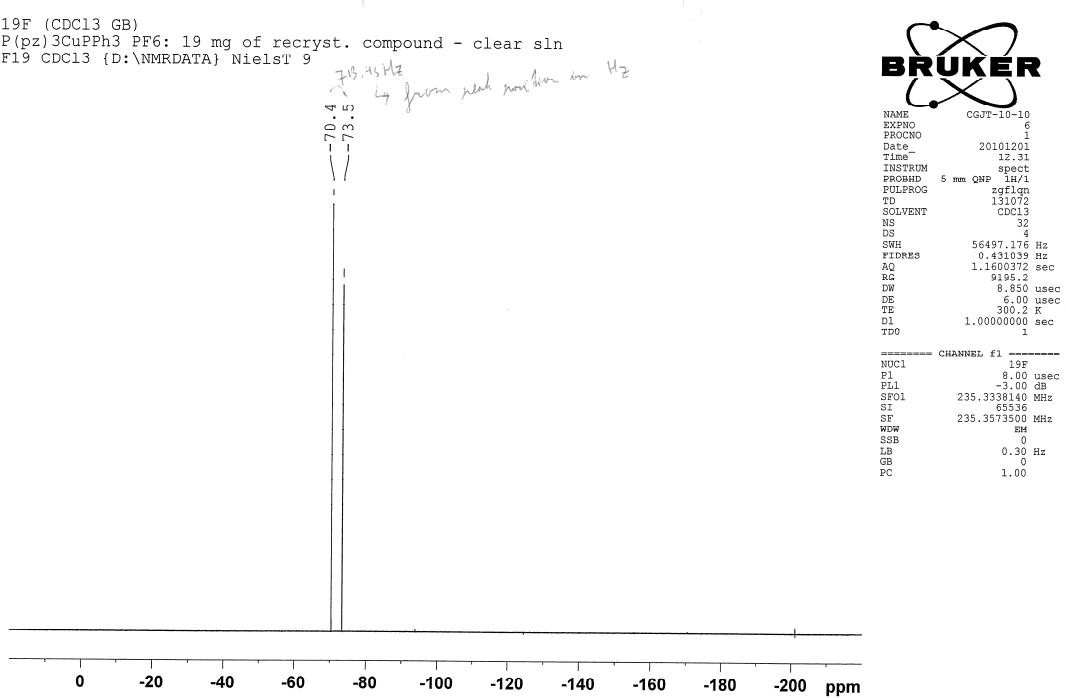
^1H NMR of $[C^H\text{Cu}(\text{PPh}_3)][\text{PF}_6]$ (500.2 MHz, CDCl_3)



$^{13}\text{C}\{^1\text{H}\}$ NMR of $[\text{C}^\text{H}\text{Cu}(\text{PPh}_3)]\text{[PF}_6]$ (125.8 MHz, CDCl_3)



¹⁹F NMR of [C^HCu(PPh₃)][PF₆] (235.3 MHz, CDCl₃)



³¹P{¹H} NMR of [C^HCu(PPh₃)][PF₆] (101.3 MHz, CDCl₃)

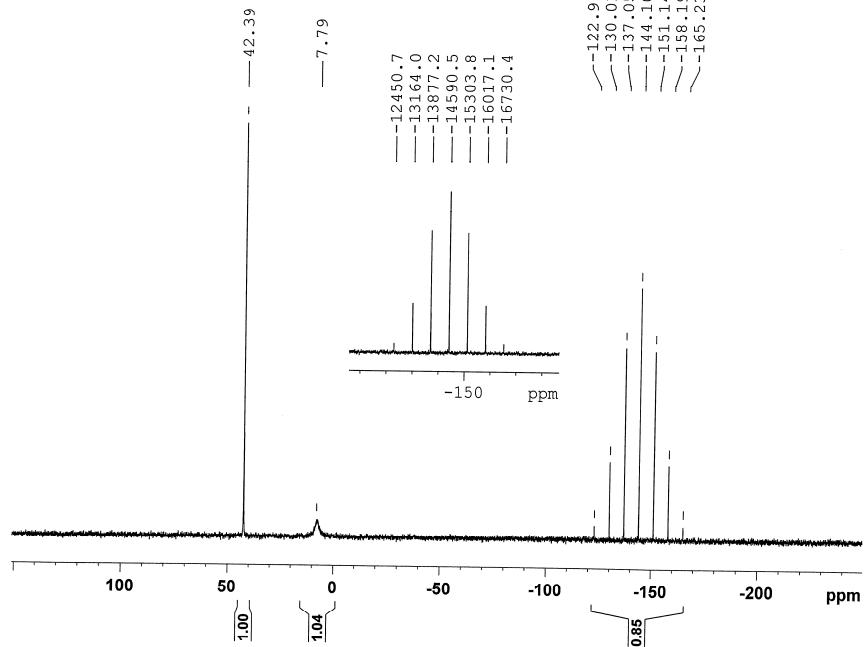
31P CPD (CDCl₃ GB)
 P(pz)3CuPPh₃ PF₆: 19 mg of recryst. compound - clear sln
 31PCPD_NO_processing CDCl₃ {D:\NMRDATA} NielsT 9



NAME: CGJT-10-10
 EXPNO: 5
 PROBNO: 1
 Date: 20101201
 Time: 12.30
 INSTRUM: spect
 PROBTYPE: 5 mm QNP IN1H
 PULPROG: zap30
 TD: 65536
 SOLVENT: CDCl₃
 NS: 96
 DS: 4
 SWH: 40650.406 Hz
 FIDRES: 0.620276 Hz
 AQ: 0.8061428 sec
 TS: 20000 sec
 DW: 12.300 usec
 DE: 6.00 usec
 TE: 300.2 K
 D1: 2.0000000 sec
 Q11: 0.03142 sec
 DELTA: 1.8999998 sec
 TDO: 1

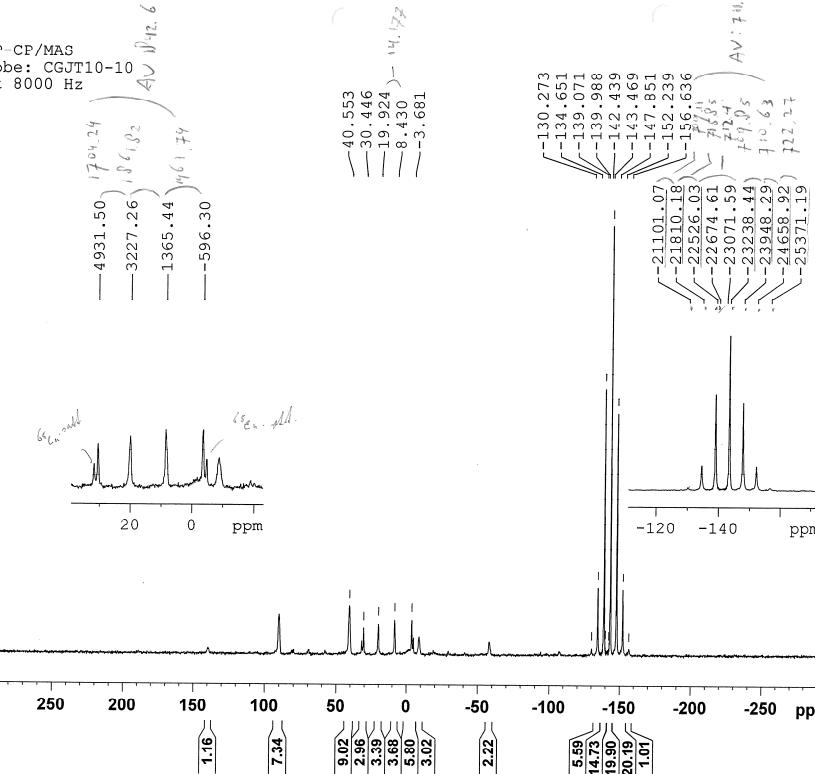
===== CHANNEL f1 =====
 NUCL: 31P
 PL: 7.00 usec
 PL1: 0.00 dB
 SF01: 101.2494172 MHz

===== CHANNEL f2 =====
 CRDPGRC2 waltz16
 NUCL: 1H
 FC_PCG2: 80.00 usec
 PL2: -3.00 dB
 PL12: 20.00 dB
 PL13: 24.00 dB
 SF02: 250.1325000 MHz
 SI: 65536
 SF: 101.2544800 MHz
 WWD: EM
 SSGB: 0
 LB: 3.00 Hz
 GB: 0
 PC: 1.40



³¹P{¹H} CP/MAS NMR of [C^HCu(PPh₃)][PF₆] (162.0 MHz)

31P-CP/MAS
 Probe: CGJT10-10
 Rot 8000 Hz



NAME: gudat10707-CGJT10-10.101
 EXPNO: 20110707
 PROBNO: 1H
 Date: 2011-07-07
 Time: 18.49
 INSTRUM: spect
 PROBHD: 4 mm MAS PROB1H
 PULPROG: hpdsec.av
 TD: 4096
 SOLVENT: CDCl₃
 NS: 72
 DS: 0
 SWH: 100000.000 Hz
 FIDRES: 24.414063 Hz
 AQ: 0.02000 sec
 TS: 203.2
 DW: 5.000 usec
 DE: 2.975 usec
 D1: 2.975 sec
 SCOPTW: 4.0000000 sec

===== CHANNEL f1 =====
 NUCL: 1H
 PL: 3.90 usec
 PL1: -2.00 dB
 SF01: 161.9759350 MHz

===== CHANNEL f2 =====
 CPDPGRC2 SPIN-32
 NUCL: 1H
 DPPG: 8.00 usec
 PL2: -2.80 dB
 SF02: 400.1336010 MHz
 SF: 161.9752600 MHz
 WWD: 0
 SSGB: 0.00 Hz
 LB: 0.00 Hz
 GB: 0
 PC: 1.40

Computational Details

Calculations were performed with the Gaussian09 series of programs.¹² The B3PW91¹³⁻¹⁵ functional was used in combination with the 6-31G* basis set for all non-metal-atoms^{16,17} (C, H, N, P) and the Lanl2dz for copper.^{18,19} The stationary points were characterized as minima by full vibration frequencies calculations (no imaginary frequency).

Cartesian coordinates, three lowest frequencies and thermochemistry of PCuN3

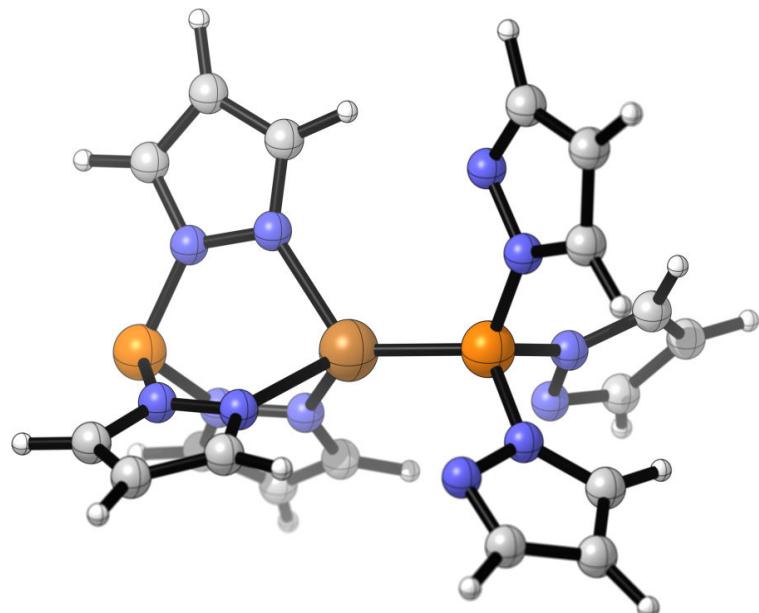


Figure S5. View of the optimized Structure of PCuN3

Element	X	Y	Z
Cu	0.00000000	0.00000000	0.47629201
P	0.00000000	0.00000000	3.84177899
P	0.00000000	0.00000000	-1.72650504
N	-0.13529100	1.53601801	3.04646993
N	-0.13052100	1.72376001	1.69552195
N	0.67523599	1.37541795	-2.52436495
N	0.46389800	2.57785511	-1.91693294
C	-0.24147400	3.02989888	1.51232505
C	-0.32127100	3.71720195	2.74373794
C	-0.24989800	2.73771095	3.70113993
C	1.01969504	3.46328402	-2.72665095
C	1.60362399	2.85307693	-3.86354303
C	1.36488199	1.51181698	-3.70460391
H	-0.25200200	3.41232204	0.49994501
H	-0.41671300	4.78109980	2.90331006
H	-0.26941100	2.78515291	4.78139400
H	0.99458098	4.51442003	-2.46866012
H	2.13827395	3.32930994	-4.67287111
H	1.64058495	0.65519100	-4.30293322

N	1.39787602	-0.65084398	3.04646993
N	-1.26258504	-0.88517398	3.04646993
N	1.55808103	-0.74884498	1.69552195
N	-1.42755997	-0.97491503	1.69552195
N	0.85352898	-1.27248096	-2.52436495
N	-1.52876496	-0.10293700	-2.52436495
N	2.00053906	-1.69067395	-1.91693294
N	-2.46443605	-0.88717997	-1.91693294
C	2.74470592	-1.30582702	1.51232505
C	-2.50323296	-1.72407198	1.51232505
C	3.37982702	-1.58037198	2.74373794
C	-3.05855489	-2.13683009	2.74373794
C	2.49587607	-1.15243697	3.70113993
C	-2.24597812	-1.58527398	3.70113993
C	2.48944402	-2.61472392	-2.72665095
C	-3.50913906	-0.84855998	-2.72665095
C	1.66902602	-2.81531811	-3.86354303
C	-3.27264905	-0.03776000	-3.86354303
C	0.62683100	-1.93793094	-3.70460391
C	-1.99171305	0.42611399	-3.70460391
H	3.08115792	-1.48792005	0.49994501
H	-2.82915592	-1.92440104	0.49994501
H	4.34890985	-2.02966595	2.90331006
H	-3.93219709	-2.75143409	2.90331006
H	2.54671907	-1.15926003	4.78139400
H	-2.27730799	-1.62589300	4.78139400
H	3.41231203	-3.11854291	-2.46866012
H	-4.40689278	-1.39587796	-2.46866012
H	1.81412995	-3.51645494	-4.67287111
H	-3.95240402	0.18714499	-4.67287111
H	-0.25288001	-1.74838400	-4.30293322
H	-1.38770497	1.09319305	-4.30293322

	1	2	3
	A	E	E
Frequencies --	17.1501	35.1384	35.1399
Red. masses --	6.2167	6.0041	6.0041
Frc consts --	0.0011	0.0044	0.0044
IR Inten --	0.0283	0.2349	0.2350

Sum of electronic and zero-point Energies=	-2231.436297
Sum of electronic and thermal Energies=	-2231.406638
Sum of electronic and thermal Enthalpies=	-2231.405694
Sum of electronic and thermal Free Energies=	-2231.499929

HF = -2231.81196597

Cartesian coordinates, three lowest frequencies and thermochemistry of N4Cu

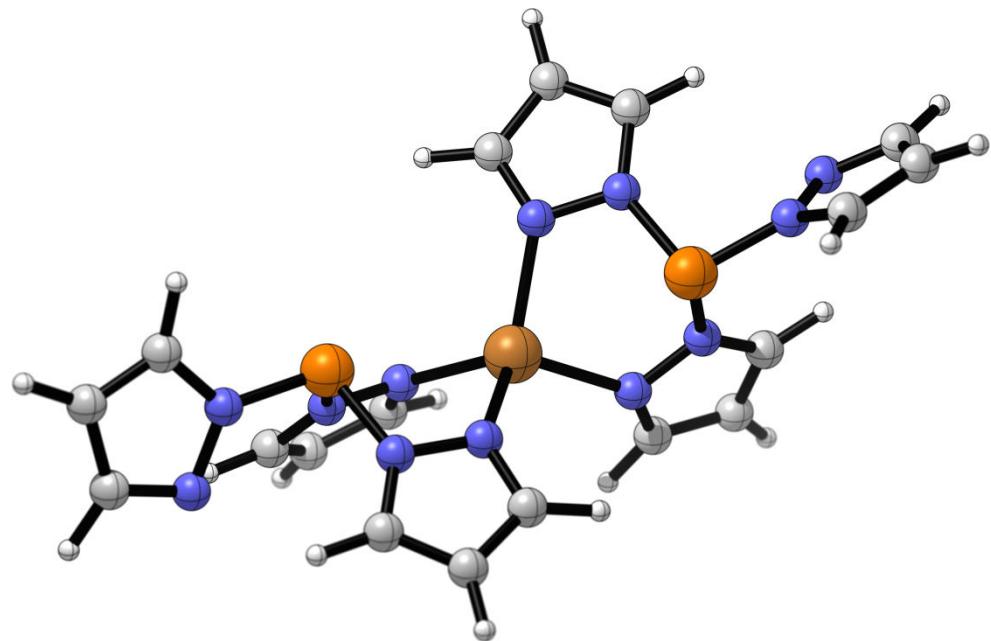


Figure S6. View of the optimized structure of N4Cu

Element	X	Y	Z
H	4.49372292	2.49232507	-0.85563999
C	3.43850803	2.27642703	-0.77130401
C	2.39855289	3.17254901	-0.88055199
H	2.37613201	4.24003077	-1.03838694
N	1.24521506	2.46304893	-0.72504002
N	1.50547695	1.14185703	-0.50878298
C	2.82936192	1.03041601	-0.53019297
H	1.37241602	-3.57283807	4.12619686
C	1.17356801	-3.12672901	3.16249990
C	0.92904198	-3.78638601	1.97819495
H	1.24407196	-0.89857101	3.48488808
H	0.85526103	-4.83576488	1.73647201
N	0.74456102	-2.83288598	1.02235901
N	0.85698700	-1.57962000	1.54849398
C	1.11013901	-1.75780106	2.84115291
H	-0.85526103	4.83576488	1.73647201
C	-0.92904198	3.78638601	1.97819495
C	-1.17356801	3.12672901	3.16249990
H	-1.37241602	3.57283807	4.12619686
C	-1.11013901	1.75780106	2.84115291
H	-1.24407196	0.89857101	3.48488808
N	-0.85698700	1.57962000	1.54849398
N	-0.74456102	2.83288598	1.02235901
H	-3.28811598	-0.06400000	-0.37125501
C	-2.82936192	-1.03041601	-0.53019297
N	-1.50547695	-1.14185703	-0.50878298
H	-4.49372292	-2.49232507	-0.85563999
N	-1.24521506	-2.46304893	-0.72504002
C	-2.39855289	-3.17254901	-0.88055199
H	-2.37613201	-4.24003077	-1.03838694
C	-3.43850803	-2.27642703	-0.77130401
Cu	0.00000000	0.00000000	0.39265999
H	3.28811598	0.06400000	-0.37125501

P	-0.47604999	2.80806899	-0.71737599
P	0.47604999	-2.80806899	-0.71737599
N	0.45613301	-4.49424219	-1.05054200
N	-0.29526299	-5.42293310	-0.38070601
C	0.00000000	-6.57081509	-0.96741199
C	0.93827498	-6.41813993	-2.01870894
C	1.21010900	-5.07743406	-2.04752493
H	-0.47110501	-7.48095989	-0.61661798
H	1.35039902	-7.18453693	-2.65872693
H	1.86247802	-4.48771906	-2.67725801
N	-0.45613301	4.49424219	-1.05054200
N	0.29526299	5.42293310	-0.38070601
C	0.00000000	6.57081509	-0.96741199
C	-0.93827498	6.41813993	-2.01870894
C	-1.21010900	5.07743406	-2.04752493
H	0.47110501	7.48095989	-0.61661798
H	-1.35039902	7.18453693	-2.65872693
H	-1.86247802	4.48771906	-2.67725801

	1	2	3
	A	B	A
Frequencies --	12.6514	14.6593	17.6812
Red. masses --	6.8231	6.7914	5.8908
Frc consts --	0.0006	0.0009	0.0011
IR Inten --	0.0758	0.2026	0.0702

Sum of electronic and zero-point Energies=	-2231.432543
Sum of electronic and thermal Energies=	-2231.402532
Sum of electronic and thermal Enthalpies=	-2231.401588
Sum of electronic and thermal Free Energies=	-2231.498536

HF = -2231.80742390

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