Supporting Information

Bovine serum albumin-cobalt(II) Schiff base complex hybrid: an efficient artificial metalloenzyme for enantioselective sulfoxidation using hydrogen peroxide

Symmetry code: (A) -x+1, -y+1, -z.						
Co1—Cl1	2.4106 (8)	Co1—O2	2.1268 (15)	Co1—O1	2.1401 (15)	
Co2—Cl1A	2.5858 (9)	Co2—O2	1.8776 (15)	Co2—O1	1.8843 (14)	
Co2—N1	1.8771 (18)	Co2—N2	1.8818 (18)			
O2A—Co1—Cl1A	95.63 (4)	O2A—Co1—Cl1	84.37 (4)	O2A-Co1-O1A	71.03 (6)	
O2A—Co1—O1	108.97 (6)	O1A—Co1—Cl1	84.00 (4)	O1—Co1—Cl1	96.00 (4)	
O2—Co2—Cl1A	84.78 (5)	O2—Co2—O1	82.43 (6)	O2—Co2—N2	94.55 (7)	
O1—Co2—Cl1A	84.56 (5)	N1—Co2—Cl1A	94.15 (6)	N1-Co2-O2	176.29 (7)	
N1—Co2—O1	93.94 (7)	N1—Co2—N2	89.13 (8)	N2—Co2—Cl1A	99.28 (6)	
N2—Co2—O1	174.92 (7)	N2-Co2-O1				

Table S1 Selected bond lengths (Å) and angles (°) for complexes CoL



Fig. S1 The local coordination environments of complex CoL.



Fig. S2 The ESI mass spectrum of CoL in the scale of m/z = 100-800.



Fig. S3 UV-visible spectra of 10 μ M BSA (yellow solid line), 10 μ M complex CoL (green solid line), 10 μ M hybrid BSA-CoL (blue solid line) and difference spectrum of BSA-CoL minus BSA (red dashed line) in PBS (pH 7.45, 0.05 M).



Fig. S4 UV-visible titration of CoL by BSA. Evolution of the UV-visible spectrum of 4.5 μ M solution of CoL in 0.05 M PBS, pH 7.45, upon addition of increasing amounts of BSA (0 to 1.5 molar equivalents). Insert: difference between the absorbance value at 362 nm of the BSA-CoL solution and that of the cofactor alone, $\Delta A_{362nm} = A_0 - A_s$ plotted against the BSA/CoL ratio.

HPLC chromatogram of sulfoxides

Methyl phenyl sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	13.987	11888.19	24.72
2	24.027	36208.63	75.28 (R)

2-Chlorophenyl methyl sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	11.329	554.17	5.758
2	22.014	9069.67	94.242 (R)



Peak	Ret.Time (min)	Area	Area%
1	15.061	4740.51	14.370
2	21.488	28248.18	85.630 (R)

3-Bromophenyl methyl sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	10.708	1390.09	7.861
2	16.067	16293.04	92.139 (R)



Peak	Ret.Time (min)	Area	Area%
1	14.173	5893.86	16.508
2	23.699	29809.32	83.492 (R)

4-Bromophenyl methyl sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	12.155	5926.47	16.896
2	17.748	29150.65	83.104 (R)

4-Methylphenyl methyl sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	9.681	11334.11	19.178
2	16.499	47765.70	80.822 (R)

4-Methoxyphenyl methyl sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	12.192	57197.60	15.709
2	24.619	306914.00	84.291 (R)

Vinyl Phenyl Sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	18.753	2196.69	67.571 (S)
2	30.109	1054.26	32.429

Methyl Benzyl Sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	13.655	58111.96	45.118
2	18.204	70687.06	54.882 (R)

Methyl tert-Butyl Sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	32.759	15930.64	49.4
2	40.359	16335.03	50.6

Methyl *n*-Octyl Sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	26.433	2240.20	40.387
2	29.464	3307.97	59.613 (R)

Methyl *n*-Dodecyl methyl Sulfoxide



Peak	Ret.Time (min)	Area	Area%
1	22.208	3700.05	50.78
2	24.399	3445.59	49.22

¹H NMR and ¹³C NMR of sulfoxides:

Methyl phenyl sulfoxide







3-Bromophenyl methyl sulfoxide









4-Bromophenyl methyl sulfoxide





tj-9 C1.52 77.50 77.29 77.29 -2.68 -2.38 -6500 -6000 -5500 -5000 ° __S__{CH3} 4500 4000 H₃C -3500 -3000 -2500 -2000 1500 -1000 -500 -0 1 00 1 88 **4** -500 5.0 4.5 f1 (ppm) . 0 7.0 6.0 5.5 4.0 3.5 3.0 2.5 2.0 1.0 8.5 8.0 7.5 6.5 1.5





4-Methoxyphenyl methyl sulfoxide





Vinyl Phenyl Sulfoxide





Methyl Benzyl Sulfoxide



95

145

135

125

115

105

90 85 80 75 70 65 60 55 50 45 40 35 30 f1 (ppm)

-4000 -2000 -0 --2000

Methyl tert-Butyl Sulfoxide



Methyl n-Octyl Sulfoxide





Methyl n-Dodecyl methyl Sulfoxide

