

Supporting Information

Halogen-bridged metal–organic frameworks constructed from bipyridinium-based ligand: structures, photochromism and non-destructive readout luminescence switching

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Section 1. Kinetic rate calculations

The photochemical reactions of compounds **1** and **2** all exhibit first order kinetics which can be analyzed with eq^[1]:

$$\ln\left(\frac{R_0 - R_\infty}{R_t - R_\infty}\right) = kt$$

where k is the first-order rate constant, R_0 , R_t , R_∞ refer to the UV–vis diffuse reflectance intensity values (400nm for compound **1**, 410nm for compound **2**) at the beginning, versus time, and at the end of the reaction, respectively.

Reference

[1] J. Sworakowski, K. Janus and S. Nešpůrek, *Adv. Colloid Interface Sci.*, 2005, **116**, 97.

Section 2. Additional Data and Figures

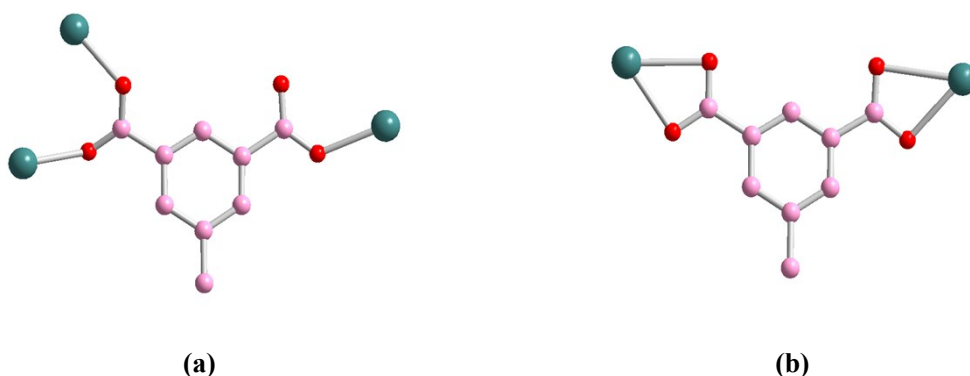


Fig.S1. The two coordination modes of the mip^{2-} ligands in compound **1**. (a) mode I: $\mu_3-\eta^1:\eta^1:\eta^1:\eta^1$; (b) mode II: $\mu_2-\eta^1:\eta^1:\eta^1:\eta^1$.

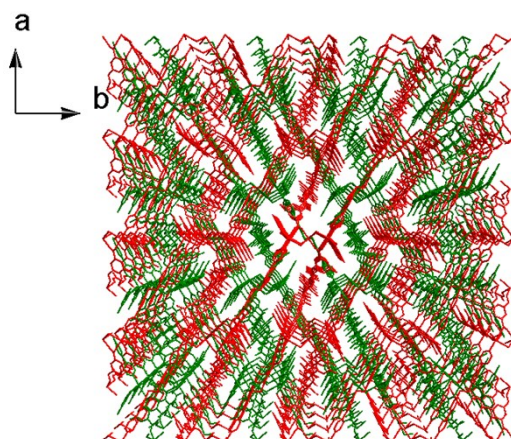


Fig.S2. The 2-fold interpenetrated structure in compound 1.

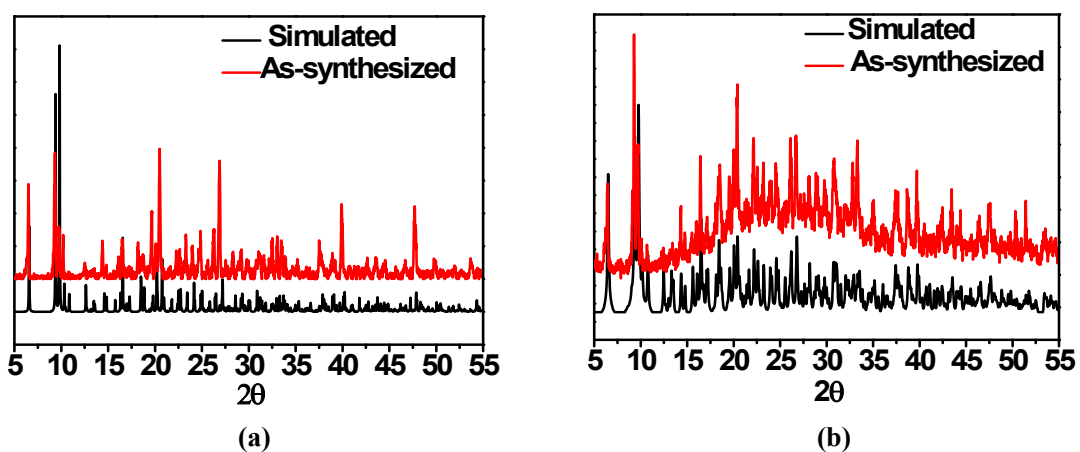


Fig.S3. PXRD patterns of compounds 1 (a) and 2 (b).

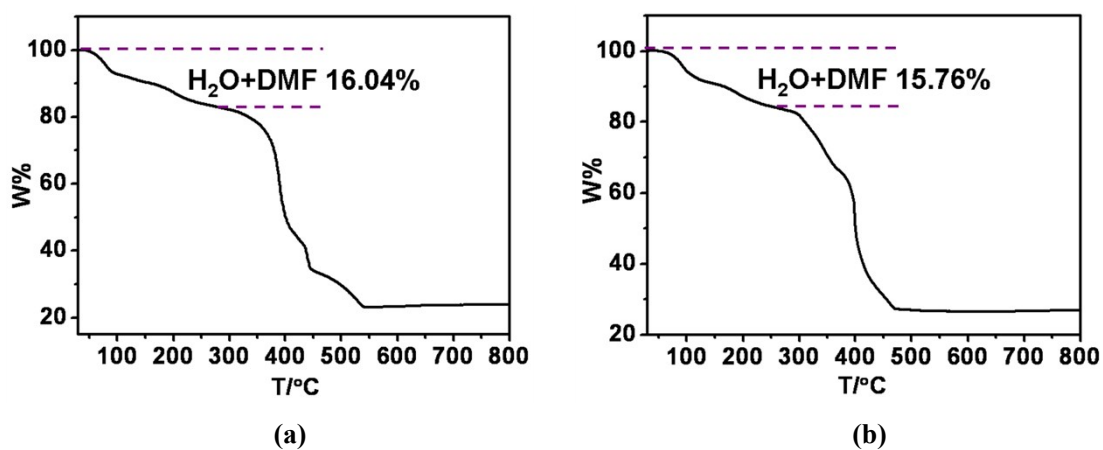


Fig.S4. TGA curves of compounds 1 (a) and 2 (b).

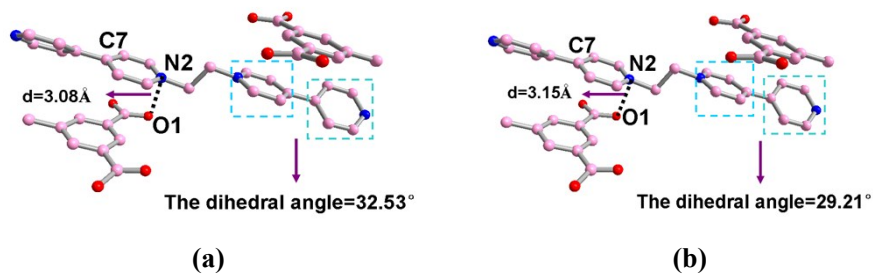


Fig.S5. The orientation diagrams of mip^{2-} ligands and $Bpyen^{2+}$ ligands in compounds **1** (a) and **2** (b).

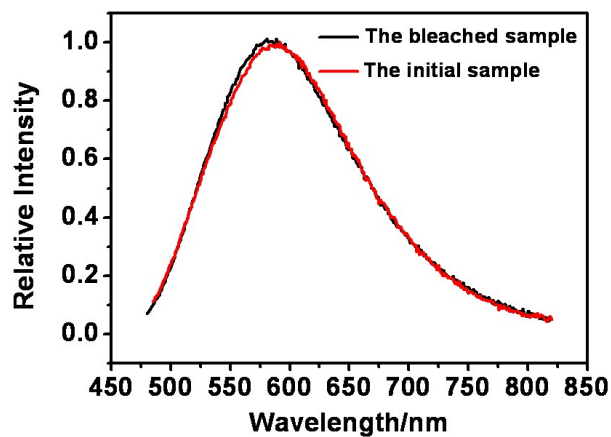
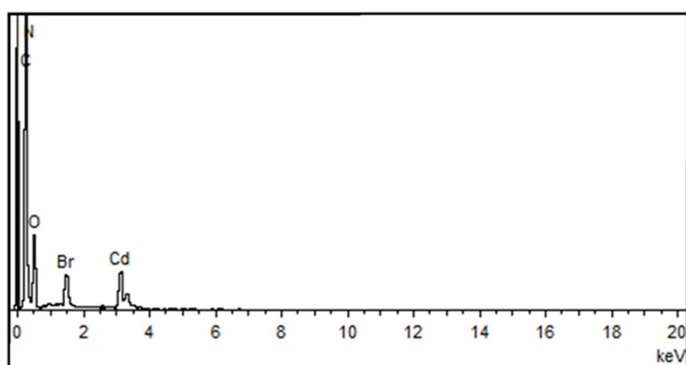
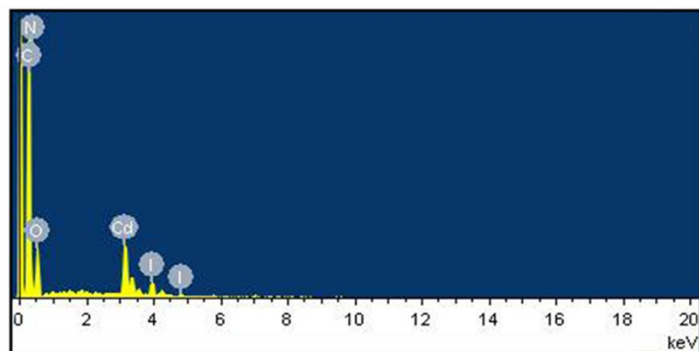


Fig.S6. The luminescence emission spectra of initial sample (red) and bleached sample (dark).



(a)



(b)

Fig.S7. Energy-dispersive X-ray spectroscopy analysis of compounds **1** (a), **2** (b) confirms the existence of elements Br or I in the structure, respectively.