

Supplementary Information

Raman Spectroscopy of the N–N Bond in Rare Earth Dinitrogen Complexes

Megan E. Fieser, David H. Woen, Jordan F. Corbey, Thomas J. Mueller, Joseph W. Ziller, and

William J. Evans*

Department of Chemistry, University of California, Irvine, California 92697-2025, United States

Email: wevans@uci.edu

*To whom correspondence should be addressed.

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Raman Spectroscopy Glassware



Table S1 X-ray Data Collection Parameters of four crystals of **1-Lu**.

	1-Lu(A)	1-Lu(B)	1-Lu(C)	1-Lu(D)
formula	C ₄₀ H ₆₀ Lu ₂ N ₂	C ₄₀ H ₆₀ Lu ₂ N ₂	C ₄₀ H ₆₀ Lu ₂ N ₂	C ₄₀ H ₆₀ Lu ₂ N ₂
fw	918.84	918.84	918.84	918.84
temp (K)	93(2)	143(2)	88(2)	143(2)
Wavelength	0.71073 Å	0.71073 Å	0.71073 Å	0.71073 Å
cryst syst	Triclinic	Tetragonal	Tetragonal	Tetragonal
space group	P $\bar{1}$	P $\bar{4}$ 2 ₁ c	P $\bar{4}$ 2 ₁ c	P $\bar{4}$ 2 ₁ c
a (Å)	10.6530(6)	14.4028(16)	14.3199(19)	14.317(3)
b (Å)	10.9963(6)	14.4028(16)	14.3199(19)	14.317(3)
c (Å)	16.6384(9)	19.526(2)	19.597(3)	19.639(5)
Volume (Å ³)	1862.31(18)	4050.4(10)	4018.4(12)	4026(2)
α (deg)	78.1853(6)	90	90	90
β (deg)	78.1059(6)	90	90	90
γ (deg)	83.7548(6)	90	90	90
Z	2	4	4	4
ρ _{calc} (Mg/m ³)	1.639	1.507	1.519	1.516
μ (mm ⁻¹)	5.298	4.872	4.911	4.902
R1 (I > 2σ(I)) ^a	0.0232	0.0325	0.0254	0.0274
wR2 (all data) ^a	0.0572	0.0911	0.0660	0.0758

^aDefinitions: wR2 = [Σ[w(F_o² - F_c²)²] / Σ[w(F_o²)²]]^{1/2}; R1 = Σ||F_o|| - |F_c|| / Σ|F_o|.

Crystallographic Details

X-ray Data Collection, Structure Solution and Refinement for 1-Lu(A). A red crystal of approximate dimensions 0.115 x 0.122 x 0.345 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX²¹ program package and the CELL_NOW² were used to determine the unit-cell parameters. Data was collected using a 25 sec/frame scan time for a sphere of diffraction data. The raw frame data was processed using SAINT³ and TWINABS⁴ to yield the reflection data file (HKLF5 format)⁴. Subsequent calculations were carried out using the SHELXTL⁵ program. There were no systematic absences nor any diffraction symmetry other than the Friedel condition. The centrosymmetric triclinic space group $P\bar{1}$ was assigned and later determined to be correct. The structure was solved by direct methods and refined on F² by full-matrix least-squares techniques. The analytical scattering factors⁶ for neutral atoms were used throughout the analysis. Hydrogen atoms were included using a riding model. The lutetium atoms were disordered and included using multiple components, anisotropic thermal parameters and partial site-occupancy-factors (0.93/0.07). At convergence, wR2 = 0.0572 and Goof = 1.039 for 436 variables refined against 8933 data (0.74Å), R1 = 0.0232 for those 7977 data with I > 2.0σ(I). The structure was refined as a two-component twin (HKLF5 format)⁵.

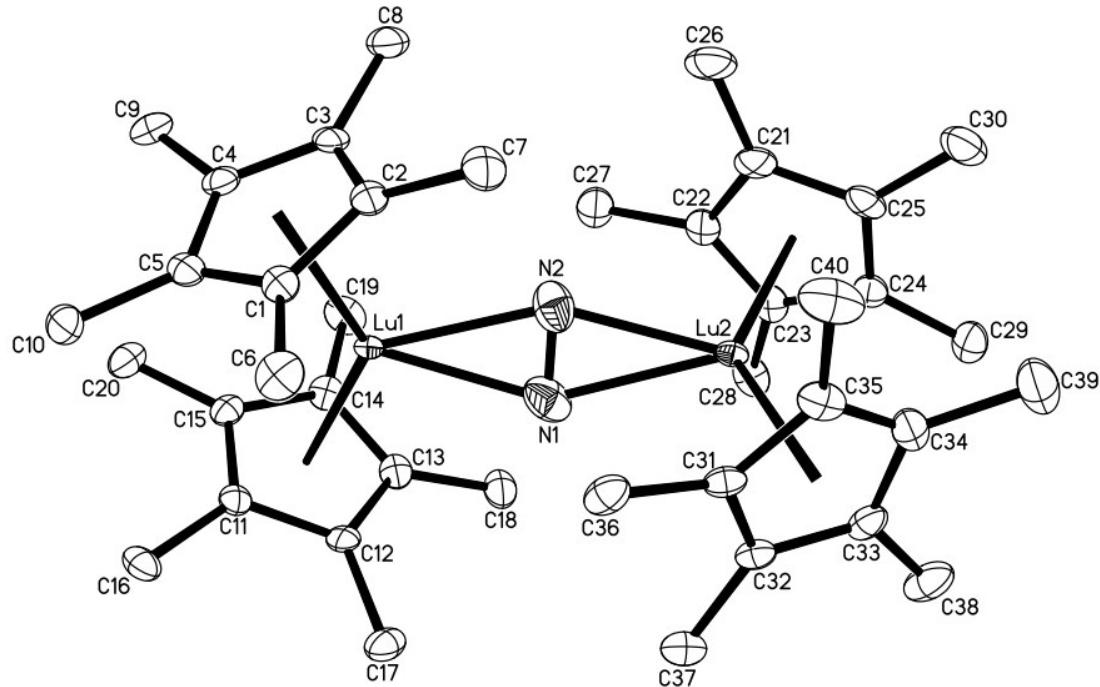


Fig. S1 Thermal ellipsoid plot of $[(C_5Me_5)_2Lu]_2(\mu-\eta^2:N_2)$, **1-Lu(A)**, drawn at the 30% probability level. Hydrogen atoms are omitted for clarity.

Table S2 Bond lengths [\AA] and angles [$^\circ$] for **1-Lu(A)**.

Lu(1)-Cnt1	2.335
Lu(1)-Cnt2	2.334
Lu(2)-Cnt1	2.332
Lu(2)-Cnt1	2.320
Lu(1)-N(1)	2.227(4)
Lu(1)-N(2)	2.265(4)
Lu(1)-C(14)	2.593(3)
Lu(1)-C(1)	2.595(3)
Lu(1)-C(13)	2.604(3)
Lu(1)-C(3)	2.614(3)
Lu(1)-C(2)	2.614(3)
Lu(1)-C(12)	2.618(3)
Lu(1)-C(5)	2.655(3)
Lu(1)-C(15)	2.656(3)

Lu(1)-C(11)	2.658(3)
Lu(1)-C(4)	2.661(3)
Lu(2)-N(2)	2.205(4)
Lu(2)-N(1)	2.267(4)
Lu(2)-C(21)	2.582(3)
Lu(2)-C(32)	2.590(3)
Lu(2)-C(33)	2.605(3)
Lu(2)-C(22)	2.611(3)
Lu(2)-C(31)	2.618(3)
Lu(2)-C(35)	2.625(3)
Lu(2)-C(34)	2.634(4)
Lu(2)-C(25)	2.635(3)
Lu(2)-C(23)	2.635(3)
Lu(2)-C(24)	2.662(3)
N(1)-N(2)	1.111(5)
C(1)-C(2)	1.414(5)
C(1)-C(5)	1.423(5)
C(1)-C(6)	1.502(5)
C(2)-C(3)	1.410(5)
C(2)-C(7)	1.511(5)
C(3)-C(4)	1.421(6)
C(3)-C(8)	1.506(5)
C(4)-C(5)	1.426(5)
C(4)-C(9)	1.500(5)
C(5)-C(10)	1.499(5)
C(11)-C(15)	1.406(5)
C(11)-C(12)	1.417(5)
C(11)-C(16)	1.503(5)
C(12)-C(13)	1.415(5)
C(12)-C(17)	1.498(5)
C(13)-C(14)	1.416(5)
C(13)-C(18)	1.503(5)
C(14)-C(15)	1.420(5)
C(14)-C(19)	1.503(5)
C(15)-C(20)	1.503(5)
C(21)-C(22)	1.416(6)

C(21)-C(25)	1.435(6)
C(21)-C(26)	1.488(6)
C(22)-C(23)	1.396(5)
C(22)-C(27)	1.512(6)
C(23)-C(24)	1.419(5)
C(23)-C(28)	1.499(5)
C(24)-C(25)	1.425(5)
C(24)-C(29)	1.504(6)
C(25)-C(30)	1.499(6)
C(31)-C(35)	1.407(5)
C(31)-C(32)	1.423(4)
C(31)-C(36)	1.500(5)
C(32)-C(33)	1.416(5)
C(32)-C(37)	1.492(5)
C(33)-C(34)	1.428(6)
C(33)-C(38)	1.497(5)
C(34)-C(35)	1.412(6)
C(34)-C(39)	1.510(6)
C(35)-C(40)	1.509(5)

Cnt1-Lu(1)-N(1)	123.8
Cnt1-Lu(1)-N(2)	101.9
Cnt2-Lu(1)-N(1)	102.3
Cnt2-Lu(1)-N(2)	124.1
Cnt1-Lu(1)-Cnt2	133.3
Cnt3-Lu(2)-N(1)	119.4
Cnt3-Lu(2)-N(2)	102.8
Cnt4-Lu(2)-N(1)	103.2
Cnt4-Lu(2)-N(2)	119.1
Cnt3-Lu(2)-Cnt4	136.7
N(1)-Lu(1)-N(2)	28.62(13)
N(1)-Lu(1)-C(14)	102.83(12)
N(2)-Lu(1)-C(14)	112.87(12)
N(1)-Lu(1)-C(1)	111.09(12)
N(2)-Lu(1)-C(1)	100.51(12)
C(14)-Lu(1)-C(1)	145.36(10)

N(1)-Lu(1)-C(13)	76.75(13)
N(2)-Lu(1)-C(13)	96.55(13)
C(14)-Lu(1)-C(13)	31.62(11)
C(1)-Lu(1)-C(13)	154.75(11)
N(1)-Lu(1)-C(3)	112.87(13)
N(2)-Lu(1)-C(3)	85.25(14)
C(14)-Lu(1)-C(3)	119.93(12)
C(1)-Lu(1)-C(3)	52.07(11)
C(13)-Lu(1)-C(3)	148.80(11)
N(1)-Lu(1)-C(2)	96.41(13)
N(2)-Lu(1)-C(2)	75.79(13)
C(14)-Lu(1)-C(2)	151.10(11)
C(1)-Lu(1)-C(2)	31.51(11)
C(13)-Lu(1)-C(2)	172.28(10)
C(3)-Lu(1)-C(2)	31.30(11)
N(1)-Lu(1)-C(12)	84.17(13)
N(2)-Lu(1)-C(12)	111.59(14)
C(14)-Lu(1)-C(12)	52.19(11)
C(1)-Lu(1)-C(12)	123.58(11)
C(13)-Lu(1)-C(12)	31.45(11)
C(3)-Lu(1)-C(12)	162.95(11)
C(2)-Lu(1)-C(12)	152.78(11)
N(1)-Lu(1)-C(5)	142.50(12)
N(2)-Lu(1)-C(5)	127.42(13)
C(14)-Lu(1)-C(5)	114.40(11)
C(1)-Lu(1)-C(5)	31.41(11)
C(13)-Lu(1)-C(5)	136.03(11)
C(3)-Lu(1)-C(5)	51.60(11)
C(2)-Lu(1)-C(5)	51.65(11)
C(12)-Lu(1)-C(5)	114.88(11)
N(1)-Lu(1)-C(15)	128.20(12)
N(2)-Lu(1)-C(15)	144.12(12)
C(14)-Lu(1)-C(15)	31.37(11)
C(1)-Lu(1)-C(15)	115.19(11)
C(13)-Lu(1)-C(15)	51.51(11)
C(3)-Lu(1)-C(15)	113.29(11)

C(2)-Lu(1)-C(15)	135.37(11)
C(12)-Lu(1)-C(15)	51.28(10)
C(5)-Lu(1)-C(15)	86.19(10)
N(1)-Lu(1)-C(11)	115.01(13)
N(2)-Lu(1)-C(11)	142.73(13)
C(14)-Lu(1)-C(11)	51.77(10)
C(1)-Lu(1)-C(11)	105.18(11)
C(13)-Lu(1)-C(11)	51.60(11)
C(3)-Lu(1)-C(11)	131.94(11)
C(2)-Lu(1)-C(11)	135.86(11)
C(12)-Lu(1)-C(11)	31.15(11)
C(5)-Lu(1)-C(11)	86.43(10)
C(15)-Lu(1)-C(11)	30.67(10)
N(1)-Lu(1)-C(4)	144.02(13)
N(2)-Lu(1)-C(4)	116.38(14)
C(14)-Lu(1)-C(4)	102.41(11)
C(1)-Lu(1)-C(4)	51.93(11)
C(13)-Lu(1)-C(4)	133.50(11)
C(3)-Lu(1)-C(4)	31.23(12)
C(2)-Lu(1)-C(4)	51.65(11)
C(12)-Lu(1)-C(4)	131.76(11)
C(5)-Lu(1)-C(4)	31.11(10)
C(15)-Lu(1)-C(4)	85.15(11)
C(11)-Lu(1)-C(4)	100.73(10)
N(2)-Lu(2)-N(1)	28.73(14)
N(2)-Lu(2)-C(21)	85.02(15)
N(1)-Lu(2)-C(21)	110.05(14)
N(2)-Lu(2)-C(32)	106.73(14)
N(1)-Lu(2)-C(32)	82.78(13)
C(21)-Lu(2)-C(32)	166.94(13)
N(2)-Lu(2)-C(33)	138.37(13)
N(1)-Lu(2)-C(33)	113.28(14)
C(21)-Lu(2)-C(33)	136.26(13)
C(32)-Lu(2)-C(33)	31.64(12)
N(2)-Lu(2)-C(22)	77.05(13)
N(1)-Lu(2)-C(22)	92.17(13)

C(21)-Lu(2)-C(22)	31.64(13)
C(32)-Lu(2)-C(22)	155.34(12)
C(33)-Lu(2)-C(22)	139.13(12)
N(2)-Lu(2)-C(31)	91.72(13)
N(1)-Lu(2)-C(31)	79.01(12)
C(21)-Lu(2)-C(31)	145.70(12)
C(32)-Lu(2)-C(31)	31.71(10)
C(33)-Lu(2)-C(31)	52.41(11)
C(22)-Lu(2)-C(31)	168.34(11)
N(2)-Lu(2)-C(35)	109.10(13)
N(1)-Lu(2)-C(35)	106.45(13)
C(21)-Lu(2)-C(35)	119.40(13)
C(32)-Lu(2)-C(35)	51.95(11)
C(33)-Lu(2)-C(35)	52.24(12)
C(22)-Lu(2)-C(35)	150.99(12)
C(31)-Lu(2)-C(35)	31.12(11)
N(2)-Lu(2)-C(34)	139.99(13)
N(1)-Lu(2)-C(34)	129.99(12)
C(21)-Lu(2)-C(34)	115.33(13)
C(32)-Lu(2)-C(34)	51.86(11)
C(33)-Lu(2)-C(34)	31.62(12)
C(22)-Lu(2)-C(34)	137.83(11)
C(31)-Lu(2)-C(34)	51.61(11)
C(35)-Lu(2)-C(34)	31.15(12)
N(2)-Lu(2)-C(25)	116.72(14)
N(1)-Lu(2)-C(25)	141.52(13)
C(21)-Lu(2)-C(25)	31.90(13)
C(32)-Lu(2)-C(25)	135.57(12)
C(33)-Lu(2)-C(25)	104.35(12)
C(22)-Lu(2)-C(25)	52.12(12)
C(31)-Lu(2)-C(25)	133.65(12)
C(35)-Lu(2)-C(25)	102.60(12)
C(34)-Lu(2)-C(25)	87.00(12)
N(2)-Lu(2)-C(23)	102.01(13)
N(1)-Lu(2)-C(23)	106.12(12)
C(21)-Lu(2)-C(23)	51.94(13)

C(32)-Lu(2)-C(23)	128.25(11)
C(33)-Lu(2)-C(23)	108.69(11)
C(22)-Lu(2)-C(23)	30.87(12)
C(31)-Lu(2)-C(23)	159.63(12)
C(35)-Lu(2)-C(23)	146.99(11)
C(34)-Lu(2)-C(23)	117.75(12)
C(25)-Lu(2)-C(23)	51.86(12)
N(2)-Lu(2)-C(24)	128.35(13)
N(1)-Lu(2)-C(24)	137.19(12)
C(21)-Lu(2)-C(24)	51.91(12)
C(32)-Lu(2)-C(24)	120.12(11)
C(33)-Lu(2)-C(24)	90.61(11)
C(22)-Lu(2)-C(24)	51.29(12)
C(31)-Lu(2)-C(24)	139.82(11)
C(35)-Lu(2)-C(24)	116.14(12)
C(34)-Lu(2)-C(24)	88.99(11)
C(25)-Lu(2)-C(24)	31.21(12)
C(23)-Lu(2)-C(24)	31.08(11)
N(2)-N(1)-Lu(1)	77.6(4)
N(2)-N(1)-Lu(2)	72.5(3)
Lu(1)-N(1)-Lu(2)	150.0(2)
N(1)-N(2)-Lu(2)	78.7(4)
N(1)-N(2)-Lu(1)	73.8(3)
Lu(2)-N(2)-Lu(1)	152.4(2)
C(2)-C(1)-C(5)	108.0(3)
C(2)-C(1)-C(6)	126.6(3)
C(5)-C(1)-C(6)	125.2(3)
C(2)-C(1)-Lu(1)	74.97(18)
C(5)-C(1)-Lu(1)	76.61(18)
C(6)-C(1)-Lu(1)	118.5(2)
C(3)-C(2)-C(1)	108.1(3)
C(3)-C(2)-C(7)	125.4(3)
C(1)-C(2)-C(7)	126.3(3)
C(3)-C(2)-Lu(1)	74.34(17)
C(1)-C(2)-Lu(1)	73.52(18)
C(7)-C(2)-Lu(1)	121.8(2)

C(2)-C(3)-C(4)	108.6(3)
C(2)-C(3)-C(8)	127.1(4)
C(4)-C(3)-C(8)	123.8(4)
C(2)-C(3)-Lu(1)	74.36(18)
C(4)-C(3)-Lu(1)	76.24(19)
C(8)-C(3)-Lu(1)	122.2(2)
C(3)-C(4)-C(5)	107.4(3)
C(3)-C(4)-C(9)	124.2(3)
C(5)-C(4)-C(9)	127.6(4)
C(3)-C(4)-Lu(1)	72.53(19)
C(5)-C(4)-Lu(1)	74.19(19)
C(9)-C(4)-Lu(1)	126.8(2)
C(1)-C(5)-C(4)	107.9(3)
C(1)-C(5)-C(10)	123.8(3)
C(4)-C(5)-C(10)	127.1(4)
C(1)-C(5)-Lu(1)	71.98(19)
C(4)-C(5)-Lu(1)	74.70(19)
C(10)-C(5)-Lu(1)	128.7(2)
C(15)-C(11)-C(12)	107.9(3)
C(15)-C(11)-C(16)	128.0(3)
C(12)-C(11)-C(16)	123.3(3)
C(15)-C(11)-Lu(1)	74.60(19)
C(12)-C(11)-Lu(1)	72.88(18)
C(16)-C(11)-Lu(1)	126.3(2)
C(13)-C(12)-C(11)	108.0(3)
C(13)-C(12)-C(17)	127.5(3)
C(11)-C(12)-C(17)	124.2(3)
C(13)-C(12)-Lu(1)	73.71(18)
C(11)-C(12)-Lu(1)	75.98(18)
C(17)-C(12)-Lu(1)	121.8(2)
C(12)-C(13)-C(14)	108.1(3)
C(12)-C(13)-C(18)	126.1(3)
C(14)-C(13)-C(18)	125.6(3)
C(12)-C(13)-Lu(1)	74.85(18)
C(14)-C(13)-Lu(1)	73.76(18)
C(18)-C(13)-Lu(1)	121.2(2)

C(13)-C(14)-C(15)	107.4(3)
C(13)-C(14)-C(19)	127.3(3)
C(15)-C(14)-C(19)	125.0(3)
C(13)-C(14)-Lu(1)	74.62(18)
C(15)-C(14)-Lu(1)	76.79(18)
C(19)-C(14)-Lu(1)	119.8(2)
C(11)-C(15)-C(14)	108.5(3)
C(11)-C(15)-C(20)	126.3(3)
C(14)-C(15)-C(20)	123.8(3)
C(11)-C(15)-Lu(1)	74.73(18)
C(14)-C(15)-Lu(1)	71.84(18)
C(20)-C(15)-Lu(1)	130.1(2)
C(22)-C(21)-C(25)	107.9(4)
C(22)-C(21)-C(26)	127.1(4)
C(25)-C(21)-C(26)	124.9(4)
C(22)-C(21)-Lu(2)	75.30(19)
C(25)-C(21)-Lu(2)	76.07(19)
C(26)-C(21)-Lu(2)	116.9(3)
C(23)-C(22)-C(21)	108.7(3)
C(23)-C(22)-C(27)	126.2(4)
C(21)-C(22)-C(27)	125.1(4)
C(23)-C(22)-Lu(2)	75.52(18)
C(21)-C(22)-Lu(2)	73.06(19)
C(27)-C(22)-Lu(2)	120.2(2)
C(22)-C(23)-C(24)	108.3(3)
C(22)-C(23)-C(28)	127.2(4)
C(24)-C(23)-C(28)	124.1(4)
C(22)-C(23)-Lu(2)	73.61(19)
C(24)-C(23)-Lu(2)	75.52(19)
C(28)-C(23)-Lu(2)	122.4(2)
C(23)-C(24)-C(25)	108.2(3)
C(23)-C(24)-C(29)	124.4(4)
C(25)-C(24)-C(29)	126.0(4)
C(23)-C(24)-Lu(2)	73.40(19)
C(25)-C(24)-Lu(2)	73.3(2)
C(29)-C(24)-Lu(2)	129.7(2)

C(24)-C(25)-C(21)	106.8(3)
C(24)-C(25)-C(30)	126.4(4)
C(21)-C(25)-C(30)	126.1(4)
C(24)-C(25)-Lu(2)	75.5(2)
C(21)-C(25)-Lu(2)	72.03(19)
C(30)-C(25)-Lu(2)	125.3(3)
C(35)-C(31)-C(32)	107.6(3)
C(35)-C(31)-C(36)	127.5(3)
C(32)-C(31)-C(36)	124.7(3)
C(35)-C(31)-Lu(2)	74.70(18)
C(32)-C(31)-Lu(2)	73.04(17)
C(36)-C(31)-Lu(2)	121.3(2)
C(33)-C(32)-C(31)	108.6(3)
C(33)-C(32)-C(37)	126.8(3)
C(31)-C(32)-C(37)	124.5(3)
C(33)-C(32)-Lu(2)	74.76(19)
C(31)-C(32)-Lu(2)	75.25(18)
C(37)-C(32)-Lu(2)	119.2(2)
C(32)-C(33)-C(34)	106.9(3)
C(32)-C(33)-C(38)	125.1(4)
C(34)-C(33)-C(38)	127.7(4)
C(32)-C(33)-Lu(2)	73.60(19)
C(34)-C(33)-Lu(2)	75.3(2)
C(38)-C(33)-Lu(2)	122.0(2)
C(35)-C(34)-C(33)	108.4(3)
C(35)-C(34)-C(39)	125.3(4)
C(33)-C(34)-C(39)	124.9(4)
C(35)-C(34)-Lu(2)	74.1(2)
C(33)-C(34)-Lu(2)	73.1(2)
C(39)-C(34)-Lu(2)	129.8(3)
C(31)-C(35)-C(34)	108.4(3)
C(31)-C(35)-C(40)	125.6(4)
C(34)-C(35)-C(40)	125.8(4)
C(31)-C(35)-Lu(2)	74.18(19)
C(34)-C(35)-Lu(2)	74.78(19)
C(40)-C(35)-Lu(2)	120.6(3)

X-ray Data Collection, Structure Solution and Refinement for 1-Lu(B). A red crystal of approximate dimensions 0.107 x 0.111 x 0.390 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2⁷ program package was used to determine the unit-cell parameters and for data collection (25 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT⁸ and SADABS⁹ to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL⁵ program. The diffraction symmetry was $4/mmm$ and the systematic absences were consistent with the tetragonal space group $P\bar{4}2_1c$ that was later determined to be correct. The structure was solved by direct methods and refined on F^2 by full-matrix least-squares techniques. The analytical scattering factors⁶ for neutral atoms were used throughout the analysis. Hydrogen atoms were included using a riding model. The molecule was located about a two-fold rotation axis. The lutetium atom was disordered approximately 84% (Lu1), 16% (Lu2). Metric data involving the lutetium atom was referenced to the major component Lu(1). Carbon atoms C(16)-C(20) were disordered and included using multiple components, partial site-occupancy-factors and isotropic thermal parameters. At convergence, $wR2 = 0.0911$ and $Goof = 1.145$ for 219 variables refined against 4478 data (0.80Å), $R1 = 0.0325$ for those 3999 data with $I > 2.0\sigma(I)$. The absolute structure was assigned by refinement of the Flack parameter.¹⁰

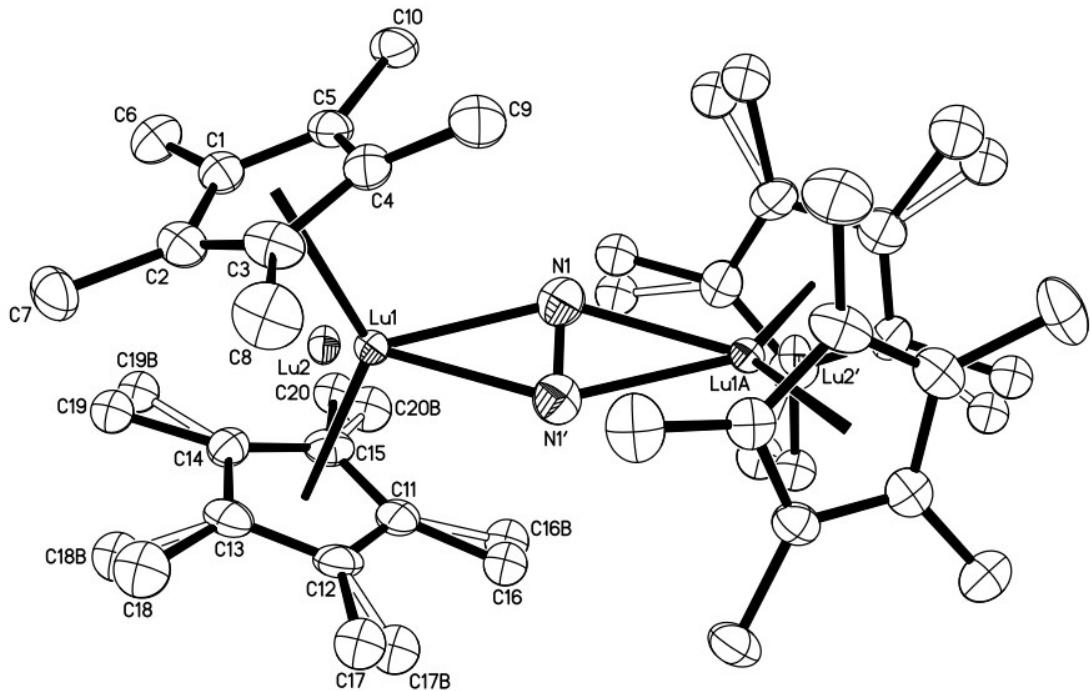


Fig. S2 Thermal ellipsoid plot of $[(\text{C}_5\text{Me}_5)_2\text{Lu}]_2(\mu\text{-}\eta^2\text{:}\eta^2\text{-N}_2)$, **1-Lu(B)**, drawn at the 30% probability level. Hydrogen atoms are omitted for clarity.

Table S3 Bond lengths [\AA] and angles [$^\circ$] for **1-Lu(B)**.

Lu(1)-Cnt1	2.352
Lu(1)-Cnt2	2.349
Lu(1)-N(1)#1	2.271(9)
Lu(1)-N(1)	2.286(9)
Lu(1)-C(11)	2.587(9)
Lu(1)-C(12)	2.593(9)
Lu(1)-C(5)	2.607(9)
Lu(1)-C(4)	2.611(9)
Lu(1)-C(3)	2.612(9)
Lu(1)-C(15)	2.636(9)
Lu(1)-C(13)	2.675(10)
Lu(1)-C(14)	2.684(9)
Lu(1)-C(1)	2.684(10)

Lu(1)-C(2)	2.693(10)
N(1)-N(1)#1	1.190(16)
N(1)-Lu(1)#1	2.271(9)
C(1)-C(5)	1.390(13)
C(1)-C(2)	1.423(13)
C(1)-C(6)	1.501(14)
C(2)-C(3)	1.402(15)
C(2)-C(7)	1.518(14)
C(3)-C(4)	1.414(14)
C(3)-C(8)	1.518(14)
C(4)-C(5)	1.439(14)
C(4)-C(9)	1.466(15)
C(5)-C(10)	1.511(13)
C(11)-C(12)	1.405(16)
C(11)-C(15)	1.420(14)
C(11)-C(16)	1.45(2)
C(11)-C(16B)	1.54(3)
C(12)-C(13)	1.396(16)
C(12)-C(17B)	1.48(3)
C(12)-C(17)	1.61(3)
C(13)-C(14)	1.410(15)
C(13)-C(18B)	1.51(3)
C(13)-C(18)	1.61(3)
C(14)-C(15)	1.383(15)
C(14)-C(19)	1.55(2)
C(14)-C(19B)	1.56(2)
C(15)-C(20B)	1.51(3)
C(15)-C(20)	1.54(3)

Cnt1-Lu(1)-N(1)	104.9
Cnt1-Lu(1)-N(1)#1	123.4
Cnt2-Lu(1)-N(1)	122.5
Cnt2-Lu(1)-N(1)#1	102.7
Cnt1-Lu(1)-Cnt2	132.1
N(1)#1-Lu(1)-N(1)	30.3(4)
N(1)#1-Lu(1)-C(11)	77.0(3)

N(1)-Lu(1)-C(11)	95.2(3)
N(1)#1-Lu(1)-C(12)	84.9(3)
N(1)-Lu(1)-C(12)	112.6(3)
C(11)-Lu(1)-C(12)	31.5(4)
N(1)#1-Lu(1)-C(5)	113.6(3)
N(1)-Lu(1)-C(5)	86.8(3)
C(11)-Lu(1)-C(5)	154.0(3)
C(12)-Lu(1)-C(5)	160.6(4)
N(1)#1-Lu(1)-C(4)	95.7(3)
N(1)-Lu(1)-C(4)	79.3(3)
C(11)-Lu(1)-C(4)	172.4(3)
C(12)-Lu(1)-C(4)	146.7(3)
C(5)-Lu(1)-C(4)	32.0(3)
N(1)#1-Lu(1)-C(3)	110.3(3)
N(1)-Lu(1)-C(3)	105.4(3)
C(11)-Lu(1)-C(3)	149.2(3)
C(12)-Lu(1)-C(3)	117.9(4)
C(5)-Lu(1)-C(3)	52.0(3)
C(4)-Lu(1)-C(3)	31.4(3)
N(1)#1-Lu(1)-C(15)	103.0(3)
N(1)-Lu(1)-C(15)	109.9(3)
C(11)-Lu(1)-C(15)	31.5(3)
C(12)-Lu(1)-C(15)	51.8(3)
C(5)-Lu(1)-C(15)	124.2(3)
C(4)-Lu(1)-C(15)	155.7(3)
C(3)-Lu(1)-C(15)	144.2(4)
N(1)#1-Lu(1)-C(13)	115.4(3)
N(1)-Lu(1)-C(13)	143.0(3)
C(11)-Lu(1)-C(13)	51.0(3)
C(12)-Lu(1)-C(13)	30.7(3)
C(5)-Lu(1)-C(13)	130.1(4)
C(4)-Lu(1)-C(13)	132.4(4)
C(3)-Lu(1)-C(13)	101.5(4)
C(15)-Lu(1)-C(13)	50.7(4)
N(1)#1-Lu(1)-C(14)	127.7(3)
N(1)-Lu(1)-C(14)	140.1(3)

C(11)-Lu(1)-C(14)	50.7(3)
C(12)-Lu(1)-C(14)	50.8(3)
C(5)-Lu(1)-C(14)	114.7(3)
C(4)-Lu(1)-C(14)	136.4(3)
C(3)-Lu(1)-C(14)	114.4(4)
C(15)-Lu(1)-C(14)	30.1(3)
C(13)-Lu(1)-C(14)	30.5(3)
N(1)#1-Lu(1)-C(1)	143.8(3)
N(1)-Lu(1)-C(1)	116.9(3)
C(11)-Lu(1)-C(1)	136.3(3)
C(12)-Lu(1)-C(1)	130.4(3)
C(5)-Lu(1)-C(1)	30.4(3)
C(4)-Lu(1)-C(1)	51.3(3)
C(3)-Lu(1)-C(1)	51.0(3)
C(15)-Lu(1)-C(1)	105.9(3)
C(13)-Lu(1)-C(1)	99.8(3)
C(14)-Lu(1)-C(1)	87.2(3)
N(1)#1-Lu(1)-C(2)	140.9(3)
N(1)-Lu(1)-C(2)	130.5(3)
C(11)-Lu(1)-C(2)	134.3(3)
C(12)-Lu(1)-C(2)	111.1(3)
C(5)-Lu(1)-C(2)	51.0(3)
C(4)-Lu(1)-C(2)	51.2(3)
C(3)-Lu(1)-C(2)	30.6(3)
C(15)-Lu(1)-C(2)	115.0(3)
C(13)-Lu(1)-C(2)	84.2(3)
C(14)-Lu(1)-C(2)	86.8(3)
C(1)-Lu(1)-C(2)	30.7(3)
N(1)#1-N(1)-Lu(1)#1	75.6(8)
N(1)#1-N(1)-Lu(1)	74.2(7)
Lu(1)#1-N(1)-Lu(1)	149.6(4)
C(5)-C(1)-C(2)	108.4(8)
C(5)-C(1)-C(6)	123.4(9)
C(2)-C(1)-C(6)	127.9(9)
C(5)-C(1)-Lu(1)	71.7(5)
C(2)-C(1)-Lu(1)	75.0(5)

C(6)-C(1)-Lu(1)	124.5(6)
C(3)-C(2)-C(1)	107.6(9)
C(3)-C(2)-C(7)	124.0(9)
C(1)-C(2)-C(7)	127.0(10)
C(3)-C(2)-Lu(1)	71.5(6)
C(1)-C(2)-Lu(1)	74.3(5)
C(7)-C(2)-Lu(1)	130.0(7)
C(2)-C(3)-C(4)	108.9(9)
C(2)-C(3)-C(8)	124.4(10)
C(4)-C(3)-C(8)	126.5(11)
C(2)-C(3)-Lu(1)	77.9(5)
C(4)-C(3)-Lu(1)	74.3(5)
C(8)-C(3)-Lu(1)	118.3(8)
C(3)-C(4)-C(5)	106.6(8)
C(3)-C(4)-C(9)	128.3(10)
C(5)-C(4)-C(9)	124.7(10)
C(3)-C(4)-Lu(1)	74.3(5)
C(5)-C(4)-Lu(1)	73.8(5)
C(9)-C(4)-Lu(1)	122.7(7)
C(1)-C(5)-C(4)	108.3(8)
C(1)-C(5)-C(10)	126.1(10)
C(4)-C(5)-C(10)	125.3(9)
C(1)-C(5)-Lu(1)	77.9(5)
C(4)-C(5)-Lu(1)	74.2(5)
C(10)-C(5)-Lu(1)	119.4(6)
C(12)-C(11)-C(15)	108.0(9)
C(12)-C(11)-C(16)	112.2(14)
C(15)-C(11)-C(16)	139.6(15)
C(12)-C(11)-C(16B)	134.2(13)
C(15)-C(11)-C(16B)	117.4(14)
C(12)-C(11)-Lu(1)	74.5(5)
C(15)-C(11)-Lu(1)	76.1(5)
C(16)-C(11)-Lu(1)	117.9(11)
C(16B)-C(11)-Lu(1)	120.9(10)
C(13)-C(12)-C(11)	108.0(9)
C(13)-C(12)-C(17B)	133.6(15)

C(11)-C(12)-C(17B)	118.2(14)
C(13)-C(12)-C(17)	113.1(14)
C(11)-C(12)-C(17)	138.7(13)
C(13)-C(12)-Lu(1)	77.9(6)
C(11)-C(12)-Lu(1)	74.0(5)
C(17B)-C(12)-Lu(1)	118.1(11)
C(17)-C(12)-Lu(1)	111.1(11)
C(12)-C(13)-C(14)	107.7(10)
C(12)-C(13)-C(18B)	129.3(14)
C(14)-C(13)-C(18B)	121.8(14)
C(12)-C(13)-C(18)	111.4(14)
C(14)-C(13)-C(18)	141.0(15)
C(12)-C(13)-Lu(1)	71.4(5)
C(14)-C(13)-Lu(1)	75.1(5)
C(18B)-C(13)-Lu(1)	129.0(11)
C(18)-C(13)-Lu(1)	118.0(12)
C(15)-C(14)-C(13)	109.0(9)
C(15)-C(14)-C(19)	136.4(13)
C(13)-C(14)-C(19)	113.1(12)
C(15)-C(14)-C(19B)	115.9(12)
C(13)-C(14)-C(19B)	134.2(12)
C(15)-C(14)-Lu(1)	73.0(5)
C(13)-C(14)-Lu(1)	74.4(5)
C(19)-C(14)-Lu(1)	128.6(10)
C(19B)-C(14)-Lu(1)	126.7(10)
C(14)-C(15)-C(11)	107.3(9)
C(14)-C(15)-C(20B)	134.4(15)
C(11)-C(15)-C(20B)	117.4(15)
C(14)-C(15)-C(20)	115.8(13)
C(11)-C(15)-C(20)	136.1(14)
C(14)-C(15)-Lu(1)	76.9(6)
C(11)-C(15)-Lu(1)	72.3(5)
C(20B)-C(15)-Lu(1)	123.7(11)
C(20)-C(15)-Lu(1)	124.1(10)

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+1,z

X-ray Data Collection, Structure Solution and Refinement for 1-Lu(C). A red crystal of approximate dimensions 0.083 x 0.096 x 0.230 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2⁷ program package was used to determine the unit-cell parameters and for data collection (25 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT⁸ and SADABS⁹ to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL⁵ program. The diffraction symmetry was *4/mmm* and the systematic absences were consistent with the tetragonal space group $P\bar{4}2_1c$ that was later determined to be correct. The structure was solved by direct methods and refined on F² by full-matrix least-squares techniques. The analytical scattering factors⁶ for neutral atoms were used throughout the analysis. Hydrogen atoms were included using a riding model. The molecule was located about a two-fold rotation axis. The lutetium atom was disordered approximately 92% (Lu1), 8% (Lu2). Metric data involving the lutetium atom was referenced to the major component Lu(1). Carbon atoms C(16)-C(20) were disordered and included using multiple components, partial site-occupancy-factors and isotropic thermal parameters. At convergence, wR2 = 0.0660 and Goof = 1.082 for 218 variables refined against 5242 data (0.73Å), R1 = 0.0254 for those 4919 data with I > 2.0σ(I). The absolute structure was assigned by refinement of the Flack parameter.¹⁰

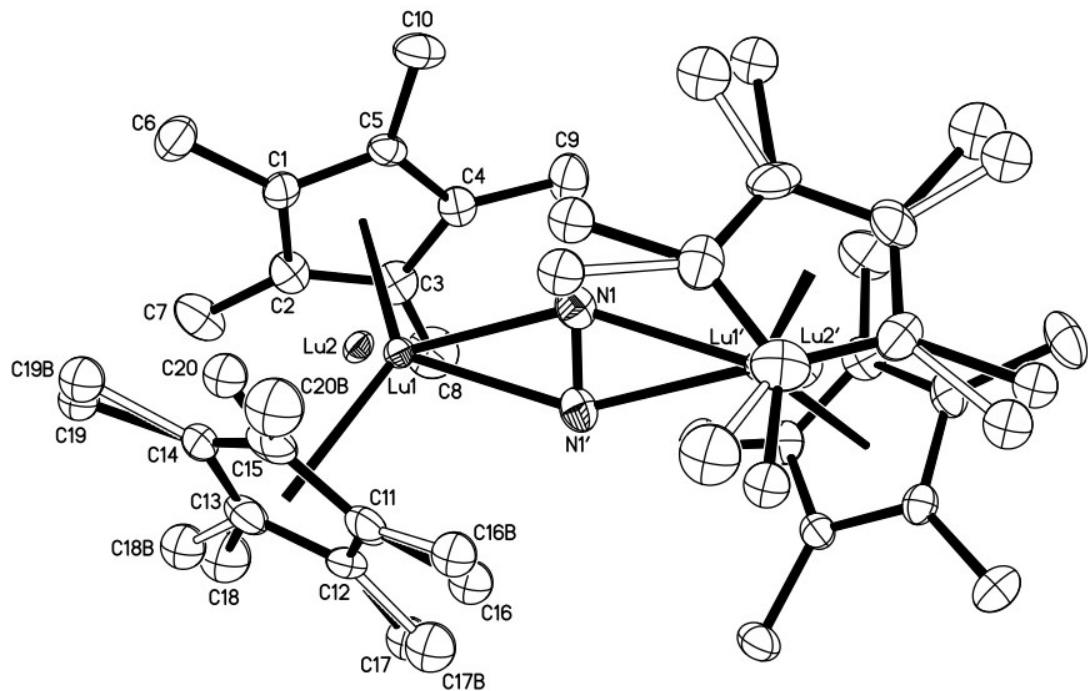


Fig. S3 Thermal ellipsoid plot of $[(C_5Me_5)_2Lu]_2(\mu\text{-}\eta^2\text{:}\eta^2\text{-}N_2)$, **1-Lu(C)**, drawn at the 30% probability level. Hydrogen atoms are omitted for clarity.

Table S4 Bond lengths [\AA] and angles [$^\circ$] for **1-Lu(C)**.

Lu(1)-Cnt1	2.336
Lu(1)-Cnt2	2.351
Lu(1)-N(1)	2.267(5)
Lu(1)-N(1)#1	2.279(5)
Lu(1)-C(12)	2.589(6)
Lu(1)-C(3)	2.604(6)
Lu(1)-C(5)	2.610(5)
Lu(1)-C(11)	2.614(6)
Lu(1)-C(4)	2.615(6)
Lu(1)-C(15)	2.636(7)
Lu(1)-C(13)	2.653(6)
Lu(1)-C(1)	2.654(6)
Lu(1)-C(2)	2.660(6)

Lu(1)-C(14)	2.679(6)
N(1)-N(1)#1	1.233(10)
N(1)-Lu(1)#1	2.279(5)
C(1)-C(5)	1.401(8)
C(1)-C(2)	1.427(9)
C(1)-C(6)	1.500(8)
C(2)-C(3)	1.408(9)
C(2)-C(7)	1.524(9)
C(3)-C(4)	1.426(10)
C(3)-C(8)	1.514(9)
C(4)-C(5)	1.421(9)
C(4)-C(9)	1.479(9)
C(5)-C(10)	1.512(8)
C(11)-C(12)	1.394(11)
C(11)-C(15)	1.441(10)
C(11)-C(16)	1.464(15)
C(11)-C(16B)	1.589(16)
C(12)-C(13)	1.390(11)
C(12)-C(17B)	1.482(18)
C(12)-C(17)	1.614(16)
C(13)-C(14)	1.385(11)
C(13)-C(18B)	1.514(17)
C(13)-C(18)	1.66(2)
C(14)-C(15)	1.380(11)
C(14)-C(19)	1.530(15)
C(14)-C(19B)	1.533(16)
C(15)-C(20B)	1.48(2)
C(15)-C(20)	1.516(16)

Cnt1-Lu(1)-N(1)	104.2
Cnt1-Lu(1)-N(1)#1	123.3
Cnt2-Lu(1)-N(1)	122.0
Cnt2-Lu(1)-N(1)#1	101.8
Cnt1-Lu(1)-Cnt2	133.1
N(1)-Lu(1)-N(1)#1	31.5(2)
N(1)-Lu(1)-C(12)	112.8(2)

N(1)#1-Lu(1)-C(12)	84.3(2)
N(1)-Lu(1)-C(3)	104.8(2)
N(1)#1-Lu(1)-C(3)	109.7(2)
C(12)-Lu(1)-C(3)	118.5(2)
N(1)-Lu(1)-C(5)	86.04(18)
N(1)#1-Lu(1)-C(5)	113.83(18)
C(12)-Lu(1)-C(5)	161.1(2)
C(3)-Lu(1)-C(5)	52.2(2)
N(1)-Lu(1)-C(11)	94.9(2)
N(1)#1-Lu(1)-C(11)	76.1(2)
C(12)-Lu(1)-C(11)	31.1(2)
C(3)-Lu(1)-C(11)	149.6(2)
C(5)-Lu(1)-C(11)	154.3(2)
N(1)-Lu(1)-C(4)	78.69(19)
N(1)#1-Lu(1)-C(4)	95.86(19)
C(12)-Lu(1)-C(4)	147.9(2)
C(3)-Lu(1)-C(4)	31.7(2)
C(5)-Lu(1)-C(4)	31.6(2)
C(11)-Lu(1)-C(4)	171.8(2)
N(1)-Lu(1)-C(15)	108.9(2)
N(1)#1-Lu(1)-C(15)	102.2(2)
C(12)-Lu(1)-C(15)	51.8(2)
C(3)-Lu(1)-C(15)	145.8(2)
C(5)-Lu(1)-C(15)	124.2(2)
C(11)-Lu(1)-C(15)	31.9(2)
C(4)-Lu(1)-C(15)	155.5(2)
N(1)-Lu(1)-C(13)	143.0(2)
N(1)#1-Lu(1)-C(13)	114.9(2)
C(12)-Lu(1)-C(13)	30.7(2)
C(3)-Lu(1)-C(13)	103.2(2)
C(5)-Lu(1)-C(13)	130.6(2)
C(11)-Lu(1)-C(13)	50.5(2)
C(4)-Lu(1)-C(13)	134.2(2)
C(15)-Lu(1)-C(13)	50.2(3)
N(1)-Lu(1)-C(1)	116.49(18)
N(1)#1-Lu(1)-C(1)	144.43(19)

C(12)-Lu(1)-C(1)	130.5(2)
C(3)-Lu(1)-C(1)	51.79(19)
C(5)-Lu(1)-C(1)	30.86(18)
C(11)-Lu(1)-C(1)	136.6(2)
C(4)-Lu(1)-C(1)	51.61(19)
C(15)-Lu(1)-C(1)	106.0(2)
C(13)-Lu(1)-C(1)	99.8(2)
N(1)-Lu(1)-C(2)	130.15(19)
N(1)#1-Lu(1)-C(2)	140.7(2)
C(12)-Lu(1)-C(2)	111.4(2)
C(3)-Lu(1)-C(2)	31.0(2)
C(5)-Lu(1)-C(2)	51.36(18)
C(11)-Lu(1)-C(2)	134.9(2)
C(4)-Lu(1)-C(2)	51.50(19)
C(15)-Lu(1)-C(2)	116.0(2)
C(13)-Lu(1)-C(2)	85.0(2)
C(1)-Lu(1)-C(2)	31.16(19)
N(1)-Lu(1)-C(14)	139.0(2)
N(1)#1-Lu(1)-C(14)	126.9(2)
C(12)-Lu(1)-C(14)	50.9(2)
C(3)-Lu(1)-C(14)	115.9(2)
C(5)-Lu(1)-C(14)	115.0(2)
C(11)-Lu(1)-C(14)	50.8(2)
C(4)-Lu(1)-C(14)	137.2(2)
C(15)-Lu(1)-C(14)	30.1(2)
C(13)-Lu(1)-C(14)	30.1(2)
C(1)-Lu(1)-C(14)	87.19(19)
C(2)-Lu(1)-C(14)	87.8(2)
N(1)#1-N(1)-Lu(1)	74.8(4)
N(1)#1-N(1)-Lu(1)#1	73.7(4)
Lu(1)-N(1)-Lu(1)#1	148.5(2)
C(5)-C(1)-C(2)	107.7(5)
C(5)-C(1)-C(6)	124.5(6)
C(2)-C(1)-C(6)	127.0(6)
C(5)-C(1)-Lu(1)	72.9(3)
C(2)-C(1)-Lu(1)	74.7(3)

C(6)-C(1)-Lu(1)	125.9(4)
C(3)-C(2)-C(1)	108.2(6)
C(3)-C(2)-C(7)	124.2(6)
C(1)-C(2)-C(7)	126.0(6)
C(3)-C(2)-Lu(1)	72.3(3)
C(1)-C(2)-Lu(1)	74.2(3)
C(7)-C(2)-Lu(1)	130.7(4)
C(2)-C(3)-C(4)	107.9(6)
C(2)-C(3)-C(8)	124.6(7)
C(4)-C(3)-C(8)	127.3(7)
C(2)-C(3)-Lu(1)	76.7(3)
C(4)-C(3)-Lu(1)	74.6(3)
C(8)-C(3)-Lu(1)	119.0(5)
C(5)-C(4)-C(3)	107.3(5)
C(5)-C(4)-C(9)	124.9(6)
C(3)-C(4)-C(9)	127.5(7)
C(5)-C(4)-Lu(1)	74.0(3)
C(3)-C(4)-Lu(1)	73.7(3)
C(9)-C(4)-Lu(1)	122.7(4)
C(1)-C(5)-C(4)	108.7(5)
C(1)-C(5)-C(10)	125.6(6)
C(4)-C(5)-C(10)	125.5(6)
C(1)-C(5)-Lu(1)	76.3(3)
C(4)-C(5)-Lu(1)	74.4(3)
C(10)-C(5)-Lu(1)	119.5(4)
C(12)-C(11)-C(15)	107.3(6)
C(12)-C(11)-C(16)	113.3(9)
C(15)-C(11)-C(16)	139.4(9)
C(12)-C(11)-C(16B)	139.5(8)
C(15)-C(11)-C(16B)	112.9(8)
C(12)-C(11)-Lu(1)	73.5(4)
C(15)-C(11)-Lu(1)	74.9(4)
C(16)-C(11)-Lu(1)	118.6(6)
C(16B)-C(11)-Lu(1)	121.9(6)
C(13)-C(12)-C(11)	107.5(6)
C(13)-C(12)-C(17B)	140.5(10)

C(11)-C(12)-C(17B)	111.5(10)
C(13)-C(12)-C(17)	115.1(9)
C(11)-C(12)-C(17)	137.4(8)
C(13)-C(12)-Lu(1)	77.2(4)
C(11)-C(12)-Lu(1)	75.4(3)
C(17B)-C(12)-Lu(1)	118.2(8)
C(17)-C(12)-Lu(1)	112.5(7)
C(14)-C(13)-C(12)	109.4(7)
C(14)-C(13)-C(18B)	122.0(10)
C(12)-C(13)-C(18B)	127.6(10)
C(14)-C(13)-C(18)	141.3(10)
C(12)-C(13)-C(18)	109.2(10)
C(14)-C(13)-Lu(1)	76.0(4)
C(12)-C(13)-Lu(1)	72.1(4)
C(18B)-C(13)-Lu(1)	127.6(8)
C(18)-C(13)-Lu(1)	115.1(8)
C(15)-C(14)-C(13)	108.4(6)
C(15)-C(14)-C(19)	136.7(9)
C(13)-C(14)-C(19)	113.7(8)
C(15)-C(14)-C(19B)	115.1(8)
C(13)-C(14)-C(19B)	135.3(9)
C(15)-C(14)-Lu(1)	73.2(4)
C(13)-C(14)-Lu(1)	73.9(4)
C(19)-C(14)-Lu(1)	127.5(6)
C(19B)-C(14)-Lu(1)	127.4(7)
C(14)-C(15)-C(11)	107.2(7)
C(14)-C(15)-C(20B)	141.6(11)
C(11)-C(15)-C(20B)	109.5(11)
C(14)-C(15)-C(20)	116.8(8)
C(11)-C(15)-C(20)	134.9(9)
C(14)-C(15)-Lu(1)	76.7(4)
C(11)-C(15)-Lu(1)	73.2(4)
C(20B)-C(15)-Lu(1)	124.7(8)
C(20)-C(15)-Lu(1)	124.7(6)

Symmetry transformations used to generate equivalent atoms:

#1 -x+2,-y+1,z

X-ray Data Collection, Structure Solution and Refinement for 1-Lu(D). A red crystal of approximate dimensions 0.109 x 0.129 x 0.229 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2⁷ program package was used to determine the unit-cell parameters and for data collection (20 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT⁸ and SADABS¹¹ to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL⁵ program. The diffraction symmetry was 4/mmm and the systematic absences were consistent with the tetragonal space group $P\bar{4}2_1c$ which was later determined to be correct. The structure was solved by direct methods and refined on F² by full-matrix least-squares techniques. The analytical scattering factors⁶ for neutral atoms were used throughout the analysis. Hydrogen atoms were included using a riding model. The molecule was located about a two-fold rotation axis. The lutetium atom was disordered approximately 96% (Lu1), 4% (Lu2). Metric data involving the lutetium atom was referenced to the major component Lu(1). Carbon atoms C(16)-C(20) were disordered and included using multiple components with partial site-occupancy-factors and isotropic thermal parameters. At convergence, wR2 = 0.0758 and Goof = 1.096 for 213 variables refined against 4430 data (0.78Å), R1 = 0.0274 for those 4204 data with I > 2.0σ(I). The absolute structure was assigned by refinement of the Flack parameter¹⁰. Solvent accessible voids were examined with the PLATON program SQUEEZE¹². A model based on the application of the SQUEEZE routine was not deemed appropriate as no solvent molecules were observed.

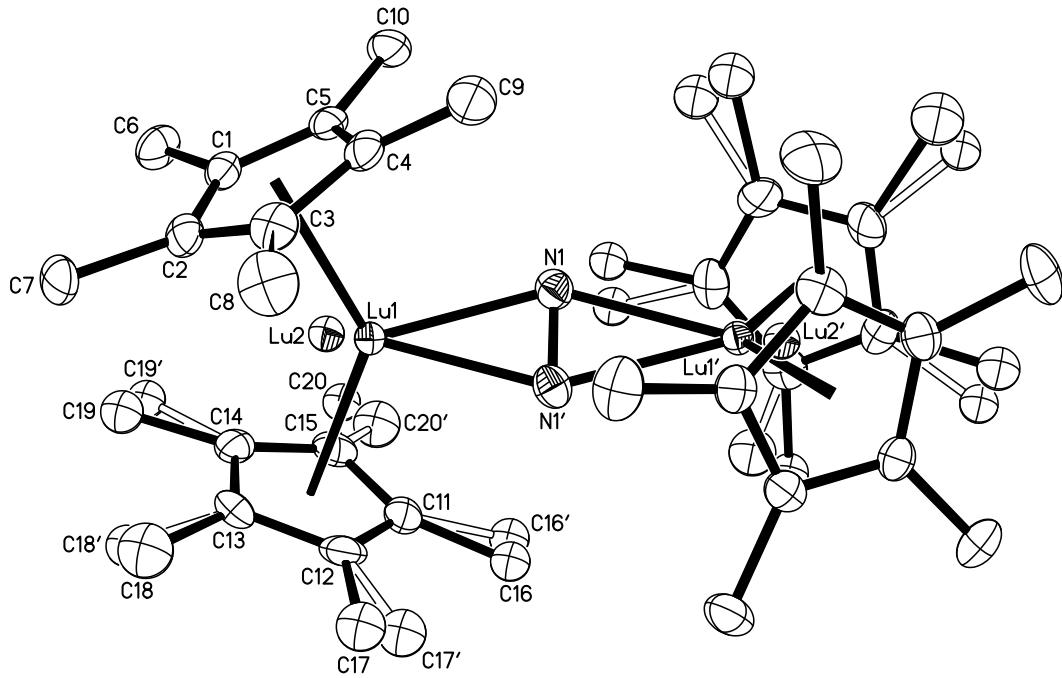


Fig. S4 Thermal ellipsoid plot of $[(C_5Me_5)_2Lu]_2(\mu-\eta^2:N_2)$, **1-Lu(D)**, drawn at the 30% probability level. Hydrogen atoms are omitted for clarity.

Table S5 Bond lengths [\AA] and angles [$^\circ$] for **1-Lu(D)**.

Lu(1)-Cnt1	2.330
Lu(1)-Cnt1	2.338
Lu(1)-N(1)#1	2.272(6)
Lu(1)-N(1)	2.279(6)
Lu(1)-C(12)	2.579(8)
Lu(1)-C(3)	2.593(7)
Lu(1)-C(5)	2.605(7)
Lu(1)-C(4)	2.612(7)
Lu(1)-C(11)	2.613(8)
Lu(1)-C(15)	2.637(9)
Lu(1)-C(13)	2.642(8)

Lu(1)-C(1)	2.649(8)
Lu(1)-C(2)	2.655(7)
Lu(1)-C(14)	2.658(8)
N(1)-N(1)#1	1.239(12)
N(1)-Lu(1)#1	2.272(6)
C(1)-C(5)	1.391(11)
C(1)-C(2)	1.430(11)
C(1)-C(6)	1.508(10)
C(2)-C(3)	1.404(11)
C(2)-C(7)	1.505(11)
C(3)-C(4)	1.432(12)
C(3)-C(8)	1.523(12)
C(4)-C(5)	1.423(11)
C(4)-C(9)	1.482(12)
C(5)-C(10)	1.515(11)
C(11)-C(12)	1.397(13)
C(11)-C(15)	1.443(13)
C(11)-C(16)	1.46(2)
C(11)-C(16A)	1.55(2)
C(12)-C(13)	1.399(13)
C(12)-C(17A)	1.50(2)
C(12)-C(17)	1.61(2)
C(13)-C(14)	1.395(14)
C(13)-C(18A)	1.492(19)
C(13)-C(18)	1.68(2)
C(14)-C(15)	1.391(13)
C(14)-C(19)	1.54(2)
C(14)-C(19A)	1.56(2)
C(15)-C(20A)	1.44(3)
C(15)-C(20)	1.53(2)

Cnt1-Lu(1)-N(1)	104.2
Cnt1#1-Lu(1)-N(1)	123.3
Cnt2-Lu(1)-N(1)	121.6
Cnt2#1-Lu(1)-N(1)	101.3
Cnt1-Lu(1)-Cnt2	133.6

N(1)#1-Lu(1)-N(1)	31.6(3)
N(1)#1-Lu(1)-C(12)	83.6(3)
N(1)-Lu(1)-C(12)	112.2(3)
N(1)#1-Lu(1)-C(3)	109.8(3)
N(1)-Lu(1)-C(3)	104.9(3)
C(12)-Lu(1)-C(3)	118.8(3)
N(1)#1-Lu(1)-C(5)	113.8(2)
N(1)-Lu(1)-C(5)	85.9(2)
C(12)-Lu(1)-C(5)	161.9(3)
C(3)-Lu(1)-C(5)	52.4(3)
N(1)#1-Lu(1)-C(4)	95.9(2)
N(1)-Lu(1)-C(4)	78.6(2)
C(12)-Lu(1)-C(4)	148.1(3)
C(3)-Lu(1)-C(4)	31.9(3)
C(5)-Lu(1)-C(4)	31.7(3)
N(1)#1-Lu(1)-C(11)	75.6(2)
N(1)-Lu(1)-C(11)	94.4(3)
C(12)-Lu(1)-C(11)	31.2(3)
C(3)-Lu(1)-C(11)	149.9(3)
C(5)-Lu(1)-C(11)	154.0(3)
C(4)-Lu(1)-C(11)	171.3(3)
N(1)#1-Lu(1)-C(15)	101.9(3)
N(1)-Lu(1)-C(15)	108.6(3)
C(12)-Lu(1)-C(15)	52.0(3)
C(3)-Lu(1)-C(15)	145.9(3)
C(5)-Lu(1)-C(15)	124.2(3)
C(4)-Lu(1)-C(15)	155.6(3)
C(11)-Lu(1)-C(15)	31.9(3)
N(1)#1-Lu(1)-C(13)	114.5(3)
N(1)-Lu(1)-C(13)	142.8(3)
C(12)-Lu(1)-C(13)	31.1(3)
C(3)-Lu(1)-C(13)	102.9(3)
C(5)-Lu(1)-C(13)	131.0(3)
C(4)-Lu(1)-C(13)	134.1(3)
C(11)-Lu(1)-C(13)	51.1(3)
C(15)-Lu(1)-C(13)	50.9(3)

N(1)#1-Lu(1)-C(1)	144.2(2)
N(1)-Lu(1)-C(1)	116.2(2)
C(12)-Lu(1)-C(1)	131.4(3)
C(3)-Lu(1)-C(1)	51.5(3)
C(5)-Lu(1)-C(1)	30.7(2)
C(4)-Lu(1)-C(1)	51.4(2)
C(11)-Lu(1)-C(1)	137.3(3)
C(15)-Lu(1)-C(1)	106.4(3)
C(13)-Lu(1)-C(1)	100.4(3)
N(1)#1-Lu(1)-C(2)	140.8(2)
N(1)-Lu(1)-C(2)	130.3(2)
C(12)-Lu(1)-C(2)	111.8(3)
C(3)-Lu(1)-C(2)	31.0(2)
C(5)-Lu(1)-C(2)	51.8(2)
C(4)-Lu(1)-C(2)	51.8(2)
C(11)-Lu(1)-C(2)	135.2(2)
C(15)-Lu(1)-C(2)	116.1(3)
C(13)-Lu(1)-C(2)	84.8(3)
C(1)-Lu(1)-C(2)	31.3(2)
N(1)#1-Lu(1)-C(14)	126.8(3)
N(1)-Lu(1)-C(14)	139.1(3)
C(12)-Lu(1)-C(14)	51.3(3)
C(3)-Lu(1)-C(14)	115.7(3)
C(5)-Lu(1)-C(14)	115.2(3)
C(4)-Lu(1)-C(14)	137.3(3)
C(11)-Lu(1)-C(14)	51.2(3)
C(15)-Lu(1)-C(14)	30.5(3)
C(13)-Lu(1)-C(14)	30.5(3)
C(1)-Lu(1)-C(14)	87.6(2)
C(2)-Lu(1)-C(14)	87.6(3)
N(1)#1-N(1)-Lu(1)#1	74.5(5)
N(1)#1-N(1)-Lu(1)	73.9(5)
Lu(1)#1-N(1)-Lu(1)	148.4(3)
C(5)-C(1)-C(2)	109.0(7)
C(5)-C(1)-C(6)	124.0(8)
C(2)-C(1)-C(6)	126.6(7)

C(5)-C(1)-Lu(1)	72.9(4)
C(2)-C(1)-Lu(1)	74.6(4)
C(6)-C(1)-Lu(1)	124.6(5)
C(3)-C(2)-C(1)	107.1(7)
C(3)-C(2)-C(7)	124.6(8)
C(1)-C(2)-C(7)	126.4(7)
C(3)-C(2)-Lu(1)	72.1(4)
C(1)-C(2)-Lu(1)	74.2(4)
C(7)-C(2)-Lu(1)	131.4(6)
C(2)-C(3)-C(4)	108.5(7)
C(2)-C(3)-C(8)	125.1(9)
C(4)-C(3)-C(8)	126.1(8)
C(2)-C(3)-Lu(1)	76.9(4)
C(4)-C(3)-Lu(1)	74.8(4)
C(8)-C(3)-Lu(1)	119.8(6)
C(5)-C(4)-C(3)	107.0(7)
C(5)-C(4)-C(9)	125.7(8)
C(3)-C(4)-C(9)	126.9(8)
C(5)-C(4)-Lu(1)	73.9(4)
C(3)-C(4)-Lu(1)	73.3(4)
C(9)-C(4)-Lu(1)	122.9(5)
C(1)-C(5)-C(4)	108.2(7)
C(1)-C(5)-C(10)	126.1(8)
C(4)-C(5)-C(10)	125.4(8)
C(1)-C(5)-Lu(1)	76.4(4)
C(4)-C(5)-Lu(1)	74.4(4)
C(10)-C(5)-Lu(1)	120.1(5)
C(12)-C(11)-C(15)	107.3(8)
C(12)-C(11)-C(16)	113.7(11)
C(15)-C(11)-C(16)	138.8(12)
C(12)-C(11)-C(16A)	138.4(11)
C(15)-C(11)-C(16A)	113.8(11)
C(12)-C(11)-Lu(1)	73.0(5)
C(15)-C(11)-Lu(1)	75.0(5)
C(16)-C(11)-Lu(1)	119.7(8)
C(11)-C(12)-C(13)	108.3(8)

C(11)-C(12)-C(17A)	116.1(12)
C(13)-C(12)-C(17A)	135.3(12)
C(11)-C(12)-C(17)	139.0(11)
C(13)-C(12)-C(17)	112.7(11)
C(11)-C(12)-Lu(1)	75.7(4)
C(13)-C(12)-Lu(1)	77.0(5)
C(17)-C(12)-Lu(1)	112.2(9)
C(14)-C(13)-C(12)	108.4(8)
C(14)-C(13)-C(18A)	123.0(11)
C(12)-C(13)-C(18A)	127.5(11)
C(14)-C(13)-C(18)	140.7(12)
C(12)-C(13)-C(18)	110.8(12)
C(14)-C(13)-Lu(1)	75.4(4)
C(12)-C(13)-Lu(1)	72.0(4)
C(18)-C(13)-Lu(1)	114.5(10)
C(15)-C(14)-C(13)	108.9(8)
C(15)-C(14)-C(19)	136.4(12)
C(13)-C(14)-C(19)	113.6(11)
C(15)-C(14)-C(19A)	115.5(10)
C(13)-C(14)-C(19A)	134.2(11)
C(15)-C(14)-Lu(1)	74.0(5)
C(13)-C(14)-Lu(1)	74.1(5)
C(19)-C(14)-Lu(1)	126.8(9)
C(14)-C(15)-C(11)	107.0(8)
C(14)-C(15)-C(20A)	137.3(15)
C(11)-C(15)-C(20A)	114.4(14)
C(14)-C(15)-C(20)	116.4(10)
C(11)-C(15)-C(20)	135.8(11)
C(14)-C(15)-Lu(1)	75.6(5)
C(11)-C(15)-Lu(1)	73.1(5)
C(20)-C(15)-Lu(1)	124.3(8)

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+1,z

Table S6 X-ray Data Collection Parameters of **1-Gd** and **1-Tb**.

	1-Gd	1-Tb
formula	C ₄₀ H ₆₀ Gd ₂ N ₂	C ₄₀ H ₆₀ Tb ₂ N ₂
fw	883.40	886.74
temp (K)	88(2)	88(2)
Wavelength	0.71073 Å	0.71073 Å
cryst syst	Tetragonal	Tetragonal
space group	<i>P</i> 4̄2 ₁ <i>c</i>	<i>P</i> 4̄2 ₁ <i>c</i>
a (Å)	14.4239(19)	14.4075(15)
b (Å)	14.4239(19)	14.4075(15)
c (Å)	19.665(3)	19.627(2)
Volume (Å ³)	4091.3(12)	4074.0(10)
α (deg)	90	90
β (deg)	90	90
γ (deg)	90	90
Z	4	4
ρ _{calc} (Mg/m ³)	1.434	1.446
μ (mm ⁻¹)	3.240	3.469
R1 (I > 2σ(I)) ^a	0.0169	0.0159
wR2 (all data) ^a	0.0424	0.0377

^aDefinitions: wR2 = [Σ[w(F_o² - F_c²)²] / Σ[w(F_o²)²]]^{1/2}; R1 = Σ||F_o|| - |F_c|| / Σ|F_o|.

X-ray Data Collection, Structure Solution and Refinement for 1-Gd. A red crystal of approximate dimensions 0.248 x 0.296 x 0.618 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2¹³ program package was used to

determine the unit-cell parameters and for data collection (5 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT⁸ and SADABS⁹ to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL⁵ program. The diffraction symmetry was $4/mmm$ and the systematic absences were consistent with the tetragonal space group $P\bar{4}2_1c$ that was later determined to be correct. The structure was solved by direct methods and refined on F^2 by full-matrix least-squares techniques. The analytical scattering factors⁶ for neutral atoms were used throughout the analysis. Hydrogen atoms were included using a riding model. The molecule was located about a two-fold rotation axis. At convergence, $wR2 = 0.0424$ and $Goof = 1.052$ for 209 variables refined against 5262 data (0.73\AA), $R1 = 0.0169$ for those 5068 data with $I > 2.0\sigma(I)$. The absolute structure was assigned by refinement of the Flack parameter.¹⁰ There were high residuals present in the final difference-Fourier map. It was not possible to determine the nature of the residuals although it was probable that benzene or methylcyclohexane solvent was present. The SQUEEZE routine in the PLATON^{12,14} program package was used to account for the electrons in the solvent accessible voids.

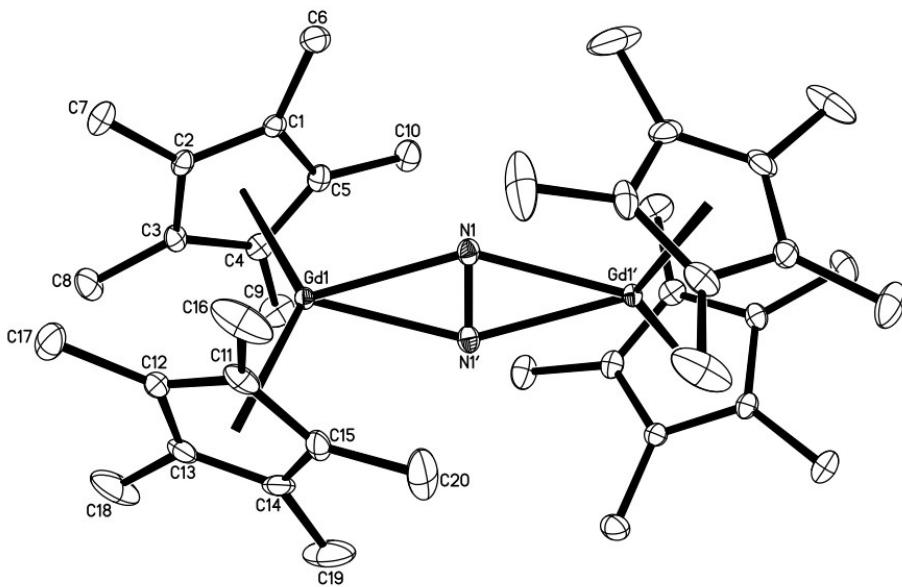


Fig. S5 Thermal ellipsoid plot of $[(C_5Me_5)_2Gd]_2(\mu-\eta^2:N_2)$, **1-Gd**, drawn at the 30% probability level. Hydrogen atoms are omitted for clarity.

Table S7 Bond lengths [\AA] and angles [$^\circ$] for **1-Gd**.

Gd(1)-Cnt1	2.409
Gd(1)-Cnt2	2.429
Gd(1)-N(1)#1	2.363(2)
Gd(1)-N(1)	2.370(2)
Gd(1)-C(14)	2.670(3)
Gd(1)-C(4)	2.679(3)
Gd(1)-C(1)	2.685(3)
Gd(1)-C(5)	2.695(3)
Gd(1)-C(15)	2.698(3)
Gd(1)-C(2)	2.707(3)
Gd(1)-C(3)	2.709(3)
Gd(1)-C(13)	2.717(3)
Gd(1)-C(11)	2.724(3)
Gd(1)-C(12)	2.735(3)
N(1)-N(1)#1	1.236(5)
N(1)-Gd(1)#1	2.363(2)

C(1)-C(2)	1.416(5)
C(1)-C(5)	1.420(5)
C(1)-C(6)	1.502(5)
C(2)-C(3)	1.415(5)
C(2)-C(7)	1.500(4)
C(3)-C(4)	1.423(5)
C(3)-C(8)	1.499(5)
C(4)-C(5)	1.424(5)
C(4)-C(9)	1.512(5)
C(5)-C(10)	1.510(5)
C(11)-C(12)	1.410(5)
C(11)-C(15)	1.420(5)
C(11)-C(16)	1.488(6)
C(12)-C(13)	1.410(5)
C(12)-C(17)	1.504(5)
C(13)-C(14)	1.400(5)
C(13)-C(18)	1.522(5)
C(14)-C(15)	1.412(6)
C(14)-C(19)	1.514(6)
C(15)-C(20)	1.504(5)

Cnt1-Gd(1)-N(1)	103.1
Cnt1-Gd(1)-N(1)#1	120.6
Cnt2-Gd(1)-N(1)	121.3
Cnt2-Gd(1)-N(1)#1	101.6
Cnt1-Gd(1)-Cnt2	135.4
N(1)#1-Gd(1)-N(1)	30.28(11)
N(1)#1-Gd(1)-C(14)	84.31(10)
N(1)-Gd(1)-C(14)	111.58(11)
N(1)#1-Gd(1)-C(4)	107.31(10)
N(1)-Gd(1)-C(4)	102.99(10)
C(14)-Gd(1)-C(4)	119.53(12)
N(1)#1-Gd(1)-C(1)	111.93(9)
N(1)-Gd(1)-C(1)	85.82(9)
C(14)-Gd(1)-C(1)	162.44(11)
C(4)-Gd(1)-C(1)	50.72(10)

N(1)#1-Gd(1)-C(5)	94.03(10)
N(1)-Gd(1)-C(5)	78.07(10)
C(14)-Gd(1)-C(5)	147.55(12)
C(4)-Gd(1)-C(5)	30.73(10)
C(1)-Gd(1)-C(5)	30.61(10)
N(1)#1-Gd(1)-C(15)	76.98(10)
N(1)-Gd(1)-C(15)	94.84(10)
C(14)-Gd(1)-C(15)	30.51(12)
C(4)-Gd(1)-C(15)	150.04(11)
C(1)-Gd(1)-C(15)	156.24(12)
C(5)-Gd(1)-C(15)	170.68(11)
N(1)#1-Gd(1)-C(2)	141.88(9)
N(1)-Gd(1)-C(2)	116.03(9)
C(14)-Gd(1)-C(2)	132.37(11)
C(4)-Gd(1)-C(2)	50.38(10)
C(1)-Gd(1)-C(2)	30.45(10)
C(5)-Gd(1)-C(2)	50.26(10)
C(15)-Gd(1)-C(2)	139.05(10)
N(1)#1-Gd(1)-C(3)	137.92(10)
N(1)-Gd(1)-C(3)	128.50(9)
C(14)-Gd(1)-C(3)	113.06(12)
C(4)-Gd(1)-C(3)	30.62(10)
C(1)-Gd(1)-C(3)	50.41(9)
C(5)-Gd(1)-C(3)	50.44(10)
C(15)-Gd(1)-C(3)	136.39(10)
C(2)-Gd(1)-C(3)	30.29(10)
N(1)#1-Gd(1)-C(13)	114.17(10)
N(1)-Gd(1)-C(13)	141.41(10)
C(14)-Gd(1)-C(13)	30.10(11)
C(4)-Gd(1)-C(13)	104.52(11)
C(1)-Gd(1)-C(13)	132.72(11)
C(5)-Gd(1)-C(13)	134.72(11)
C(15)-Gd(1)-C(13)	49.81(11)
C(2)-Gd(1)-C(13)	102.34(10)
C(3)-Gd(1)-C(13)	87.17(11)
N(1)#1-Gd(1)-C(11)	102.16(10)

N(1)-Gd(1)-C(11)	109.27(10)
C(14)-Gd(1)-C(11)	50.15(11)
C(4)-Gd(1)-C(11)	147.59(11)
C(1)-Gd(1)-C(11)	127.85(11)
C(5)-Gd(1)-C(11)	158.24(11)
C(15)-Gd(1)-C(11)	30.36(11)
C(2)-Gd(1)-C(11)	109.86(11)
C(3)-Gd(1)-C(11)	118.57(10)
C(13)-Gd(1)-C(11)	49.67(11)
N(1)#1-Gd(1)-C(12)	126.64(10)
N(1)-Gd(1)-C(12)	139.19(10)
C(14)-Gd(1)-C(12)	49.76(11)
C(4)-Gd(1)-C(12)	117.79(11)
C(1)-Gd(1)-C(12)	118.15(10)
C(5)-Gd(1)-C(12)	139.11(10)
C(15)-Gd(1)-C(12)	49.69(10)
C(2)-Gd(1)-C(12)	90.65(10)
C(3)-Gd(1)-C(12)	90.31(10)
C(13)-Gd(1)-C(12)	29.99(11)
C(11)-Gd(1)-C(12)	29.93(11)
N(1)#1-N(1)-Gd(1)#1	75.2(2)
N(1)#1-N(1)-Gd(1)	74.5(2)
Gd(1)#1-N(1)-Gd(1)	149.42(11)
C(2)-C(1)-C(5)	108.0(3)
C(2)-C(1)-C(6)	126.5(3)
C(5)-C(1)-C(6)	125.4(3)
C(2)-C(1)-Gd(1)	75.66(17)
C(5)-C(1)-Gd(1)	75.09(18)
C(6)-C(1)-Gd(1)	117.8(2)
C(3)-C(2)-C(1)	108.4(3)
C(3)-C(2)-C(7)	126.3(3)
C(1)-C(2)-C(7)	124.9(3)
C(3)-C(2)-Gd(1)	74.92(17)
C(1)-C(2)-Gd(1)	73.90(17)
C(7)-C(2)-Gd(1)	122.7(2)
C(2)-C(3)-C(4)	107.8(3)

C(2)-C(3)-C(8)	125.9(3)
C(4)-C(3)-C(8)	125.1(3)
C(2)-C(3)-Gd(1)	74.79(17)
C(4)-C(3)-Gd(1)	73.55(17)
C(8)-C(3)-Gd(1)	127.3(2)
C(3)-C(4)-C(5)	107.9(3)
C(3)-C(4)-C(9)	125.1(3)
C(5)-C(4)-C(9)	126.9(3)
C(3)-C(4)-Gd(1)	75.83(17)
C(5)-C(4)-Gd(1)	75.23(18)
C(9)-C(4)-Gd(1)	117.9(2)
C(1)-C(5)-C(4)	107.8(3)
C(1)-C(5)-C(10)	124.9(3)
C(4)-C(5)-C(10)	127.1(3)
C(1)-C(5)-Gd(1)	74.30(18)
C(4)-C(5)-Gd(1)	74.04(18)
C(10)-C(5)-Gd(1)	121.7(2)
C(12)-C(11)-C(15)	107.6(3)
C(12)-C(11)-C(16)	124.6(4)
C(15)-C(11)-C(16)	127.3(4)
C(12)-C(11)-Gd(1)	75.45(19)
C(15)-C(11)-Gd(1)	73.8(2)
C(16)-C(11)-Gd(1)	123.5(3)
C(11)-C(12)-C(13)	108.3(3)
C(11)-C(12)-C(17)	125.1(4)
C(13)-C(12)-C(17)	125.6(4)
C(11)-C(12)-Gd(1)	74.63(19)
C(13)-C(12)-Gd(1)	74.32(19)
C(17)-C(12)-Gd(1)	126.3(2)
C(14)-C(13)-C(12)	108.1(3)
C(14)-C(13)-C(18)	124.9(4)
C(12)-C(13)-C(18)	126.4(4)
C(14)-C(13)-Gd(1)	73.08(19)
C(12)-C(13)-Gd(1)	75.69(18)
C(18)-C(13)-Gd(1)	124.2(3)
C(13)-C(14)-C(15)	108.4(3)

C(13)-C(14)-C(19)	125.3(5)
C(15)-C(14)-C(19)	126.3(4)
C(13)-C(14)-Gd(1)	76.82(19)
C(15)-C(14)-Gd(1)	75.82(19)
C(19)-C(14)-Gd(1)	115.6(3)
C(14)-C(15)-C(11)	107.7(3)
C(14)-C(15)-C(20)	124.6(5)
C(11)-C(15)-C(20)	127.6(5)
C(14)-C(15)-Gd(1)	73.67(19)
C(11)-C(15)-Gd(1)	75.9(2)
C(20)-C(15)-Gd(1)	120.1(3)

Symmetry transformations used to generate equivalent atoms:

#1 -x+2,-y+1,z

X-ray Data Collection, Structure Solution and Refinement for 1-Tb. A purple crystal of approximate dimensions 0.095 x 0.109 x 0.264 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2¹³ program package was used to determine the unit-cell parameters and for data collection (20 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT⁸ and SADABS⁹ to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL⁵ program. The diffraction symmetry was 4/mmm and the systematic absences were consistent with the tetragonal space group $P\bar{4}2_1c$ that was later determined to be correct. The structure was solved by direct methods and refined on F² by full-matrix least-squares techniques. The analytical scattering factors⁶ for neutral atoms were used throughout the analysis. Hydrogen atoms were included using a riding model. The molecule was located about a two-fold rotation axis. At convergence, wR2 = 0.0377 and Goof = 1.028 for 209 variables refined against 5254 data (0.73Å), R1 = 0.0159 for those 5092 data with I > 2.0σ(I). The absolute structure was

assigned by refinement of the Flack parameter.¹⁰ There were high residuals present in the final difference-Fourier map. It was not possible to determine the nature of the residuals although it was probable that benzene or methylcyclohexane solvent was present. The SQUEEZE routine in the PLATON^{12,14} program package was used to account for the electrons in the solvent accessible voids.

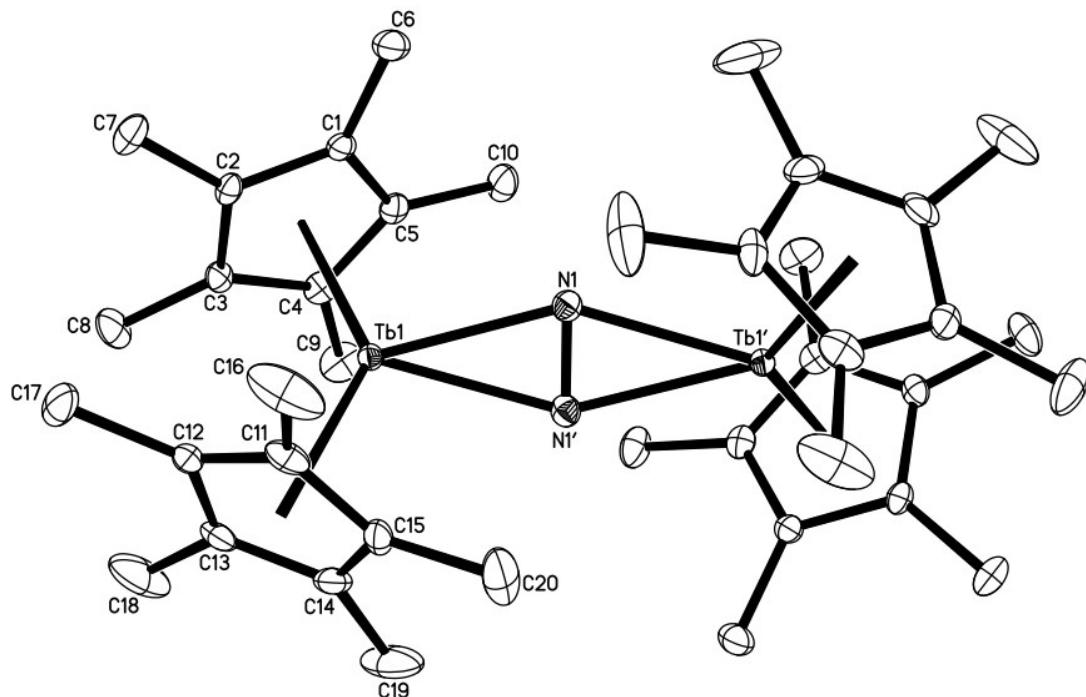


Fig. S6 Thermal ellipsoid plot of $[(C_5Me_5)_2Tb]_2(\mu-\eta^2:N_2)$, **1-Tb**, drawn at the 30% probability level. Hydrogen atoms are omitted for clarity.

Table S8 Bond lengths [\AA] and angles [$^\circ$] for **1-Tb**.

Tb(1)-Cnt1	2.392
Tb(1)-Cnt2	2.412
Tb(1)-N(1)#1	2.347(2)
Tb(1)-N(1)	2.350(2)
Tb(1)-C(14)	2.656(3)
Tb(1)-C(4)	2.663(3)
Tb(1)-C(1)	2.669(3)
Tb(1)-C(5)	2.673(3)

Tb(1)-C(15)	2.679(3)
Tb(1)-C(2)	2.693(3)
Tb(1)-C(3)	2.695(3)
Tb(1)-C(13)	2.705(3)
Tb(1)-C(11)	2.708(3)
Tb(1)-C(12)	2.718(3)
N(1)-N(1)#1	1.238(4)
N(1)-Tb(1)#1	2.347(2)
C(1)-C(5)	1.414(4)
C(1)-C(2)	1.420(4)
C(1)-C(6)	1.502(4)
C(2)-C(3)	1.414(4)
C(2)-C(7)	1.503(4)
C(3)-C(4)	1.418(4)
C(3)-C(8)	1.508(4)
C(4)-C(5)	1.421(5)
C(4)-C(9)	1.512(4)
C(5)-C(10)	1.512(4)
C(11)-C(12)	1.402(5)
C(11)-C(15)	1.415(5)
C(11)-C(16)	1.496(6)
C(12)-C(13)	1.420(5)
C(12)-C(17)	1.506(5)
C(13)-C(14)	1.402(5)
C(13)-C(18)	1.520(5)
C(14)-C(15)	1.404(5)
C(14)-C(19)	1.516(5)
C(15)-C(20)	1.504(5)

Cnt1-Tb(1)-N(1)	103.2
Cnt1-Tb(1)-N(1)#1	121.3
Cnt2-Tb(1)-N(1)	121.4
Cnt2-Tb(1)-N(1)#1	101.5
Cnt1-Tb(1)-Cnt2	135.1
N(1)#1-Tb(1)-N(1)	30.55(11)
N(1)#1-Tb(1)-C(14)	84.13(10)

N(1)-Tb(1)-C(14)	111.71(10)
N(1)#1-Tb(1)-C(4)	108.17(9)
N(1)-Tb(1)-C(4)	103.47(9)
C(14)-Tb(1)-C(4)	119.62(11)
N(1)#1-Tb(1)-C(1)	112.12(9)
N(1)-Tb(1)-C(1)	85.54(8)
C(14)-Tb(1)-C(1)	162.64(10)
C(4)-Tb(1)-C(1)	50.95(9)
N(1)#1-Tb(1)-C(5)	94.57(9)
N(1)-Tb(1)-C(5)	78.21(9)
C(14)-Tb(1)-C(5)	147.78(11)
C(4)-Tb(1)-C(5)	30.88(10)
C(1)-Tb(1)-C(5)	30.69(9)
N(1)#1-Tb(1)-C(15)	76.74(9)
N(1)-Tb(1)-C(15)	94.90(9)
C(14)-Tb(1)-C(15)	30.52(11)
C(4)-Tb(1)-C(15)	150.13(11)
C(1)-Tb(1)-C(15)	155.75(11)
C(5)-Tb(1)-C(15)	171.03(10)
N(1)#1-Tb(1)-C(2)	142.43(9)
N(1)-Tb(1)-C(2)	115.94(9)
C(14)-Tb(1)-C(2)	132.29(10)
C(4)-Tb(1)-C(2)	50.62(9)
C(1)-Tb(1)-C(2)	30.70(9)
C(5)-Tb(1)-C(2)	50.54(9)
C(15)-Tb(1)-C(2)	138.43(9)
N(1)#1-Tb(1)-C(3)	138.86(9)
N(1)-Tb(1)-C(3)	128.84(9)
C(14)-Tb(1)-C(3)	113.06(11)
C(4)-Tb(1)-C(3)	30.70(9)
C(1)-Tb(1)-C(3)	50.68(9)
C(5)-Tb(1)-C(3)	50.65(9)
C(15)-Tb(1)-C(3)	136.09(9)
C(2)-Tb(1)-C(3)	30.44(9)
N(1)#1-Tb(1)-C(13)	114.19(9)
N(1)-Tb(1)-C(13)	141.75(9)

C(14)-Tb(1)-C(13)	30.30(11)
C(4)-Tb(1)-C(13)	104.11(10)
C(1)-Tb(1)-C(13)	132.64(10)
C(5)-Tb(1)-C(13)	134.48(10)
C(15)-Tb(1)-C(13)	50.03(10)
C(2)-Tb(1)-C(13)	102.02(10)
C(3)-Tb(1)-C(13)	86.75(10)
N(1)#1-Tb(1)-C(11)	101.94(10)
N(1)-Tb(1)-C(11)	109.19(10)
C(14)-Tb(1)-C(11)	50.30(10)
C(4)-Tb(1)-C(11)	147.12(10)
C(1)-Tb(1)-C(11)	127.32(10)
C(5)-Tb(1)-C(11)	157.76(10)
C(15)-Tb(1)-C(11)	30.46(10)
C(2)-Tb(1)-C(11)	109.07(10)
C(3)-Tb(1)-C(11)	117.96(10)
C(13)-Tb(1)-C(11)	49.93(11)
N(1)#1-Tb(1)-C(12)	126.61(9)
N(1)-Tb(1)-C(12)	139.13(9)
C(14)-Tb(1)-C(12)	50.12(10)
C(4)-Tb(1)-C(12)	117.32(10)
C(1)-Tb(1)-C(12)	117.67(9)
C(5)-Tb(1)-C(12)	138.65(10)
C(15)-Tb(1)-C(12)	49.89(9)
C(2)-Tb(1)-C(12)	89.90(9)
C(3)-Tb(1)-C(12)	89.69(10)
C(13)-Tb(1)-C(12)	30.35(10)
C(11)-Tb(1)-C(12)	29.95(10)
N(1)#1-N(1)-Tb(1)#1	74.9(2)
N(1)#1-N(1)-Tb(1)	74.6(2)
Tb(1)#1-N(1)-Tb(1)	149.24(11)
C(5)-C(1)-C(2)	107.9(3)
C(5)-C(1)-C(6)	126.0(3)
C(2)-C(1)-C(6)	126.1(3)
C(5)-C(1)-Tb(1)	74.81(16)
C(2)-C(1)-Tb(1)	75.59(16)

C(6)-C(1)-Tb(1)	118.2(2)
C(3)-C(2)-C(1)	108.2(3)
C(3)-C(2)-C(7)	126.3(3)
C(1)-C(2)-C(7)	125.0(3)
C(3)-C(2)-Tb(1)	74.84(16)
C(1)-C(2)-Tb(1)	73.71(16)
C(7)-C(2)-Tb(1)	123.8(2)
C(2)-C(3)-C(4)	107.9(3)
C(2)-C(3)-C(8)	125.7(3)
C(4)-C(3)-C(8)	125.2(3)
C(2)-C(3)-Tb(1)	74.72(15)
C(4)-C(3)-Tb(1)	73.41(16)
C(8)-C(3)-Tb(1)	127.6(2)
C(3)-C(4)-C(5)	108.0(3)
C(3)-C(4)-C(9)	125.2(3)
C(5)-C(4)-C(9)	126.7(3)
C(3)-C(4)-Tb(1)	75.90(16)
C(5)-C(4)-Tb(1)	74.95(16)
C(9)-C(4)-Tb(1)	118.1(2)
C(1)-C(5)-C(4)	108.0(3)
C(1)-C(5)-C(10)	124.7(3)
C(4)-C(5)-C(10)	127.0(3)
C(1)-C(5)-Tb(1)	74.50(16)
C(4)-C(5)-Tb(1)	74.17(17)
C(10)-C(5)-Tb(1)	121.7(2)
C(12)-C(11)-C(15)	107.8(3)
C(12)-C(11)-C(16)	124.3(4)
C(15)-C(11)-C(16)	127.3(4)
C(12)-C(11)-Tb(1)	75.45(18)
C(15)-C(11)-Tb(1)	73.64(18)
C(16)-C(11)-Tb(1)	123.6(2)
C(11)-C(12)-C(13)	108.1(3)
C(11)-C(12)-C(17)	125.4(4)
C(13)-C(12)-C(17)	125.3(3)
C(11)-C(12)-Tb(1)	74.60(17)
C(13)-C(12)-Tb(1)	74.29(17)

C(17)-C(12)-Tb(1)	126.9(2)
C(14)-C(13)-C(12)	107.6(3)
C(14)-C(13)-C(18)	125.0(4)
C(12)-C(13)-C(18)	126.7(4)
C(14)-C(13)-Tb(1)	72.92(18)
C(12)-C(13)-Tb(1)	75.36(17)
C(18)-C(13)-Tb(1)	124.3(3)
C(13)-C(14)-C(15)	108.4(3)
C(13)-C(14)-C(19)	124.5(4)
C(15)-C(14)-C(19)	127.0(4)
C(13)-C(14)-Tb(1)	76.77(18)
C(15)-C(14)-Tb(1)	75.64(17)
C(19)-C(14)-Tb(1)	115.8(3)
C(14)-C(15)-C(11)	108.0(3)
C(14)-C(15)-C(20)	124.8(4)
C(11)-C(15)-C(20)	127.0(4)
C(14)-C(15)-Tb(1)	73.85(18)
C(11)-C(15)-Tb(1)	75.91(18)
C(20)-C(15)-Tb(1)	120.4(2)

Symmetry transformations used to generate equivalent atoms:

#1 -x+2,-y+1,z

Table S9 X-ray Data Collection Parameters of **2-Gd** and **2-Dy**.

	2-Gd	2-Dy
formula	C ₄₄ H ₆₈ Gd ₂ N ₂ O ₂ •2(C ₇ H ₈)	C ₄₄ H ₆₈ Dy ₂ N ₂ O ₂
fw	1155.77	982.00
temp (K)	88(2)	133(2)
Wavelength	0.71073 Å	0.71073 Å
cryst syst	Monoclinic	Monoclinic
space group	C2/c	C2/c
a (Å)	15.2453(7)	15.7376(15)
b (Å)	13.9938(6)	14.1959(14)
c (Å)	25.6281(11)	18.9717(18)
Volume (Å ³)	5286.7(4)	4157.5(7)
α (deg)	90	90
β (deg)	104.7734(4)	101.2185(11)
γ (deg)	90	90
Z	4	4
ρ _{calc} (Mg/m ³)	1.452	1.569
μ (mm ⁻¹)	2.529	3.603
R1 (I > 2σ(I)) ^a	0.0190	0.0316
wR2 (all data) ^a	0.0449	0.0790

^aDefinitions: wR2 = [Σ[w(F_o² - F_c²)²] / Σ[w(F_o²)]]^{1/2}; R1 = Σ||F_o|| - |F_c| / Σ|F_o|.

X-ray Data Collection, Structure Solution and Refinement for 2-Gd. A green crystal of approximate dimensions 0.237 x 0.280 x 0.374 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2¹³ program package was

used to determine the unit-cell parameters and for data collection (10 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT⁸ and SADABS⁹ to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL⁵ program. The diffraction symmetry was $2/m$ and the systematic absences were consistent with the monoclinic space groups Cc and $C2/c$. It was later determined that space group $C2/c$ was correct. The structure was solved by direct methods and refined on F^2 by full-matrix least-squares techniques. The analytical scattering factors⁶ for neutral atoms were used throughout the analysis. Hydrogen atom H(28) was located from a difference-Fourier map and refined (x,y,z and U_{iso}). The remaining hydrogen atoms were included using a riding model. The molecule was located about a two-fold rotation axis. A methyl carbon atom on one tetramethylcyclopentadienyl ligand was disordered over two positions (C(16) / C(19)) and included using multiple components with partial site-occupancy factors. The hydrogen atom associated with the disordered tetramethylcyclopentadienyl ligand was not included in the refinement. There was a molecule of toluene solvent present (two per dimeric formula-unit). The solvent was disordered. Carbon atoms C(24)-C(27) were included as above. At convergence, $wR2 = 0.0449$ and $Goof = 1.071$ for 350 variables refined against 6785 data (0.73\AA), $R1 = 0.0190$ for those 6008 data with $I > 2.0\sigma(I)$.

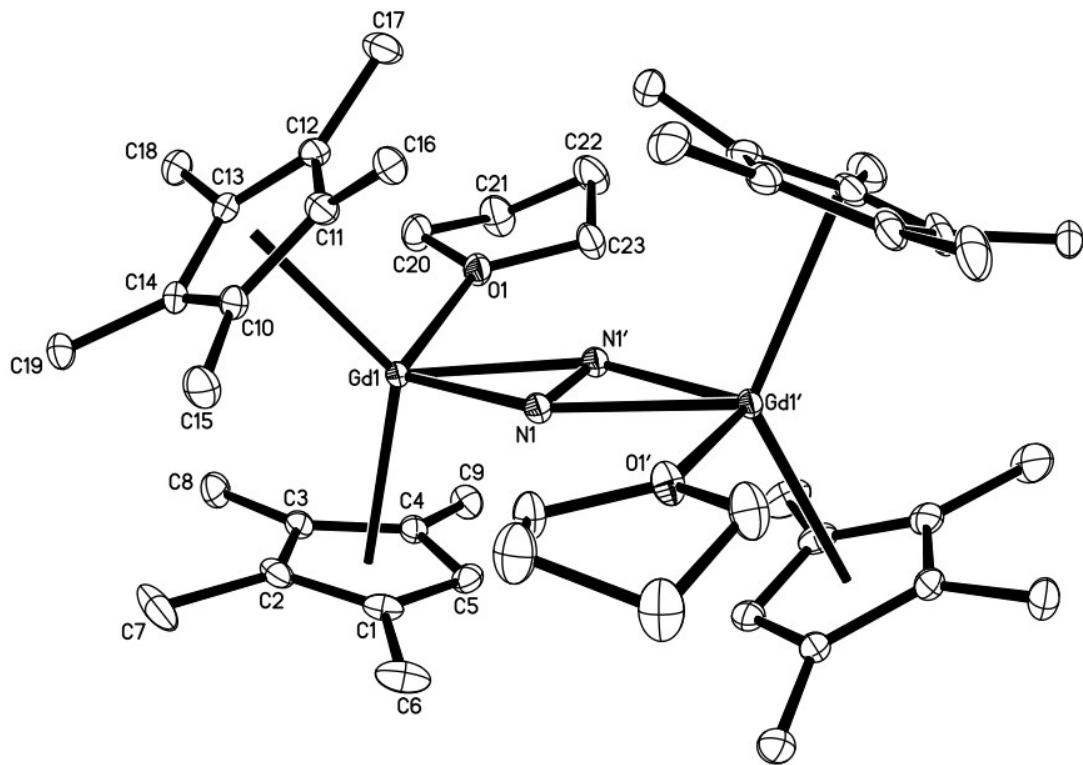


Fig. S7 Thermal ellipsoid plot of $[(C_5Me_4H)_2Gd(\text{THF})](\mu-\eta^2:\eta^2-\text{N}_2)$, **2-Gd**, drawn at the 30% probability level. Hydrogen atoms are omitted for clarity.

Table S10 Bond lengths [\AA] and angles [$^\circ$] for **2-Gd**.

Gd(1)-Cnt1	2.448
Gd(1)-Cnt2	2.459
Gd(1)-N(1)	2.3703(16)
Gd(1)-N(1)#1	2.4057(15)
Gd(1)-O(1)	2.4870(13)
Gd(1)-C(10)	2.6840(19)
Gd(1)-C(5)	2.6917(18)
Gd(1)-C(11)	2.7048(18)
Gd(1)-C(1)	2.7070(18)
Gd(1)-C(4)	2.7338(18)
Gd(1)-C(14)	2.7428(19)
Gd(1)-C(2)	2.7516(18)

Gd(1)-C(3)	2.7596(18)
Gd(1)-C(12)	2.7695(18)
Gd(1)-C(13)	2.7910(18)
O(1)-C(20)	1.453(2)
O(1)-C(23)	1.456(2)
N(1)-N(1)#1	1.247(3)
N(1)-Gd(1)#1	2.4057(15)
C(1)-C(2)	1.408(3)
C(1)-C(5)	1.423(3)
C(1)-C(6)	1.503(3)
C(2)-C(3)	1.422(3)
C(2)-C(7)	1.512(3)
C(3)-C(4)	1.419(3)
C(3)-C(8)	1.507(3)
C(4)-C(5)	1.415(3)
C(4)-C(9)	1.501(3)
C(10)-C(14)	1.407(3)
C(10)-C(11)	1.421(3)
C(10)-C(15)	1.508(3)
C(11)-C(16)	1.391(4)
C(11)-C(12)	1.415(3)
C(12)-C(13)	1.416(3)
C(12)-C(17)	1.511(3)
C(13)-C(14)	1.424(3)
C(13)-C(18)	1.507(3)
C(14)-C(19)	1.445(4)
C(20)-C(21)	1.516(3)
C(21)-C(22)	1.524(3)
C(22)-C(23)	1.518(3)
C(24)-C(25)	1.401(17)
C(24)-C(29)	1.463(5)
C(25)-C(26)	1.382(13)
C(26)-C(27)	1.383(6)
C(27)-C(28)	1.470(6)
C(24B)-C(25B)	1.399(10)
C(24B)-C(29)	1.433(6)

C(25B)-C(26B)	1.373(19)
C(26B)-C(27B)	1.369(17)
C(27B)-C(28)	1.396(7)
C(28)-C(29)	1.360(4)
C(29)-C(30)	1.491(4)

Cnt1-Gd(1)-O(1)	103.9
Cnt1-Gd(1)-N(1)	103.3
Cnt1-Gd(1)-N(1)#1	108.7
Cnt2-Gd(1)-O(1)	102.3
Cnt2-Gd(1)-N(1)	104.6
Cnt2-Gd(1)-N(1)#1	116.2
Cnt1-Gd(1)-Cnt2	130.1
N(1)-Gd(1)-N(1)#1	30.25(7)
N(1)-Gd(1)-O(1)	112.65(5)
N(1)#1-Gd(1)-O(1)	82.61(5)
N(1)-Gd(1)-C(10)	89.00(6)
N(1)#1-Gd(1)-C(10)	111.56(6)
O(1)-Gd(1)-C(10)	128.90(5)
N(1)-Gd(1)-C(5)	82.92(5)
N(1)#1-Gd(1)-C(5)	82.57(5)
O(1)-Gd(1)-C(5)	96.84(5)
C(10)-Gd(1)-C(5)	132.62(6)
N(1)-Gd(1)-C(11)	79.24(6)
N(1)#1-Gd(1)-C(11)	90.57(5)
O(1)-Gd(1)-C(11)	106.14(5)
C(10)-Gd(1)-C(11)	30.59(6)
C(5)-Gd(1)-C(11)	154.99(6)
N(1)-Gd(1)-C(1)	80.58(5)
N(1)#1-Gd(1)-C(1)	95.35(6)
O(1)-Gd(1)-C(1)	126.07(5)
C(10)-Gd(1)-C(1)	102.07(6)
C(5)-Gd(1)-C(1)	30.56(6)
C(11)-Gd(1)-C(1)	127.79(6)
N(1)-Gd(1)-C(4)	111.81(5)
N(1)#1-Gd(1)-C(4)	103.08(5)

O(1)-Gd(1)-C(4)	77.80(5)
C(10)-Gd(1)-C(4)	137.77(6)
C(5)-Gd(1)-C(4)	30.22(6)
C(11)-Gd(1)-C(4)	166.23(6)
C(1)-Gd(1)-C(4)	50.03(6)
N(1)-Gd(1)-C(14)	119.01(6)
N(1)#1-Gd(1)-C(14)	139.42(6)
O(1)-Gd(1)-C(14)	111.65(5)
C(10)-Gd(1)-C(14)	30.03(7)
C(5)-Gd(1)-C(14)	129.83(6)
C(11)-Gd(1)-C(14)	49.41(6)
C(1)-Gd(1)-C(14)	104.30(6)
C(4)-Gd(1)-C(14)	116.83(6)
N(1)-Gd(1)-C(2)	107.68(6)
N(1)#1-Gd(1)-C(2)	125.22(5)
O(1)-Gd(1)-C(2)	122.45(5)
C(10)-Gd(1)-C(2)	89.98(6)
C(5)-Gd(1)-C(2)	49.61(6)
C(11)-Gd(1)-C(2)	120.55(6)
C(1)-Gd(1)-C(2)	29.88(6)
C(4)-Gd(1)-C(2)	49.44(5)
C(14)-Gd(1)-C(2)	80.23(6)
N(1)-Gd(1)-C(3)	129.16(5)
N(1)#1-Gd(1)-C(3)	131.25(5)
O(1)-Gd(1)-C(3)	92.78(5)
C(10)-Gd(1)-C(3)	108.76(6)
C(5)-Gd(1)-C(3)	49.67(6)
C(11)-Gd(1)-C(3)	136.44(6)
C(1)-Gd(1)-C(3)	49.69(6)
C(4)-Gd(1)-C(3)	29.95(5)
C(14)-Gd(1)-C(3)	87.27(6)
C(2)-Gd(1)-C(3)	29.90(6)
N(1)-Gd(1)-C(12)	102.20(6)
N(1)#1-Gd(1)-C(12)	100.98(5)
O(1)-Gd(1)-C(12)	79.71(5)
C(10)-Gd(1)-C(12)	49.92(6)

C(5)-Gd(1)-C(12)	174.61(6)
C(11)-Gd(1)-C(12)	29.92(6)
C(1)-Gd(1)-C(12)	151.30(6)
C(4)-Gd(1)-C(12)	144.39(6)
C(14)-Gd(1)-C(12)	49.05(6)
C(2)-Gd(1)-C(12)	129.02(6)
C(3)-Gd(1)-C(12)	126.00(6)
N(1)-Gd(1)-C(13)	128.29(6)
N(1)#1-Gd(1)-C(13)	130.29(5)
O(1)-Gd(1)-C(13)	82.98(5)
C(10)-Gd(1)-C(13)	49.70(6)
C(5)-Gd(1)-C(13)	146.40(6)
C(11)-Gd(1)-C(13)	49.19(6)
C(1)-Gd(1)-C(13)	130.80(6)
C(4)-Gd(1)-C(13)	119.73(6)
C(14)-Gd(1)-C(13)	29.80(6)
C(2)-Gd(1)-C(13)	102.48(6)
C(3)-Gd(1)-C(13)	96.73(6)
C(12)-Gd(1)-C(13)	29.49(6)
C(20)-O(1)-C(23)	108.99(15)
C(20)-O(1)-Gd(1)	125.61(11)
C(23)-O(1)-Gd(1)	125.39(11)
N(1)#1-N(1)-Gd(1)	76.45(14)
N(1)#1-N(1)-Gd(1)#1	73.31(13)
Gd(1)-N(1)-Gd(1)#1	149.66(7)
C(2)-C(1)-C(5)	107.60(17)
C(2)-C(1)-C(6)	126.4(2)
C(5)-C(1)-C(6)	125.7(2)
C(2)-C(1)-Gd(1)	76.81(10)
C(5)-C(1)-Gd(1)	74.13(10)
C(6)-C(1)-Gd(1)	119.63(13)
C(1)-C(2)-C(3)	108.54(17)
C(1)-C(2)-C(7)	125.2(2)
C(3)-C(2)-C(7)	125.1(2)
C(1)-C(2)-Gd(1)	73.30(10)
C(3)-C(2)-Gd(1)	75.36(10)

C(7)-C(2)-Gd(1)	127.13(13)
C(4)-C(3)-C(2)	107.68(17)
C(4)-C(3)-C(8)	125.97(18)
C(2)-C(3)-C(8)	125.97(17)
C(4)-C(3)-Gd(1)	74.02(10)
C(2)-C(3)-Gd(1)	74.74(10)
C(8)-C(3)-Gd(1)	122.65(12)
C(5)-C(4)-C(3)	107.82(17)
C(5)-C(4)-C(9)	125.08(17)
C(3)-C(4)-C(9)	126.77(18)
C(5)-C(4)-Gd(1)	73.24(10)
C(3)-C(4)-Gd(1)	76.03(10)
C(9)-C(4)-Gd(1)	121.81(12)
C(4)-C(5)-C(1)	108.36(17)
C(4)-C(5)-Gd(1)	76.54(10)
C(1)-C(5)-Gd(1)	75.32(10)
C(14)-C(10)-C(11)	107.22(18)
C(14)-C(10)-C(15)	126.84(18)
C(11)-C(10)-C(15)	125.7(2)
C(14)-C(10)-Gd(1)	77.30(11)
C(11)-C(10)-Gd(1)	75.51(10)
C(15)-C(10)-Gd(1)	117.56(13)
C(16)-C(11)-C(12)	125.8(2)
C(16)-C(11)-C(10)	124.6(2)
C(12)-C(11)-C(10)	108.56(18)
C(16)-C(11)-Gd(1)	123.92(19)
C(12)-C(11)-Gd(1)	77.57(11)
C(10)-C(11)-Gd(1)	73.90(11)
C(11)-C(12)-C(13)	107.93(17)
C(11)-C(12)-C(17)	125.77(19)
C(13)-C(12)-C(17)	126.10(19)
C(11)-C(12)-Gd(1)	72.50(10)
C(13)-C(12)-Gd(1)	76.10(10)
C(17)-C(12)-Gd(1)	121.34(13)
C(12)-C(13)-C(14)	107.41(19)
C(12)-C(13)-C(18)	127.12(18)

C(14)-C(13)-C(18)	124.30(18)
C(12)-C(13)-Gd(1)	74.41(11)
C(14)-C(13)-Gd(1)	73.22(11)
C(18)-C(13)-Gd(1)	127.63(14)
C(10)-C(14)-C(13)	108.88(18)
C(10)-C(14)-C(19)	120.0(2)
C(13)-C(14)-C(19)	127.9(2)
C(10)-C(14)-Gd(1)	72.67(10)
C(13)-C(14)-Gd(1)	76.98(11)
C(19)-C(14)-Gd(1)	132.7(2)
O(1)-C(20)-C(21)	105.43(17)
C(20)-C(21)-C(22)	101.42(19)
C(23)-C(22)-C(21)	102.75(17)
O(1)-C(23)-C(22)	105.95(16)
C(25)-C(24)-C(29)	126.9(6)
C(26)-C(25)-C(24)	119.8(10)
C(25)-C(26)-C(27)	120.1(8)
C(26)-C(27)-C(28)	114.3(4)
C(25B)-C(24B)-C(29)	108.1(6)
C(26B)-C(25B)-C(24B)	121.1(9)
C(27B)-C(26B)-C(25B)	120.6(14)
C(26B)-C(27B)-C(28)	127.4(10)
C(29)-C(28)-C(27B)	103.1(5)
C(29)-C(28)-C(27)	131.7(3)
C(28)-C(29)-C(24B)	137.7(4)
C(28)-C(29)-C(24)	105.9(3)
C(28)-C(29)-C(30)	120.8(2)
C(24B)-C(29)-C(30)	101.0(4)
C(24)-C(29)-C(30)	133.1(3)

Symmetry transformations used to generate equivalent atoms:
#1 -x+1,y,-z+1/2

X-ray Data Collection, Structure Solution and Refinement for 2-Dy.

A green crystal of approximate dimensions 0.137 x 0.176 x 0.307 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2¹³ program package and the CELL_NOW¹⁵ program were used to determine the unit-cell parameters. Data was collected using a 15 sec/frame scan time for a sphere of diffraction data. The raw frame data was processed using SAINT⁴⁰ and TWINABS¹⁶ to yield the reflection data file (HKL F 5 format).¹⁶ Subsequent calculations were carried out using the SHELXTL⁵ program. The diffraction symmetry was $2/m$ and the systematic absences were consistent with the monoclinic space groups Cc and $C2/c$. It was later determined that space group $C2/c$ was correct. The structure was solved by direct methods and refined on F^2 by full-matrix least-squares techniques. The analytical scattering factors⁶ for neutral atoms were used throughout the analysis. Hydrogen atoms were included using a riding model. The molecule was located about a two-fold rotation axis. At convergence, $wR2 = 0.0790$ and $Goof = 1.078$ for 235 variables refined against 4928 data (0.75 Å), $R1 = 0.0316$ for those 4482 with $I > 2.0\sigma(I)$. The structure was refined as a two-component twin, $BASF^5 = 0.1165$.

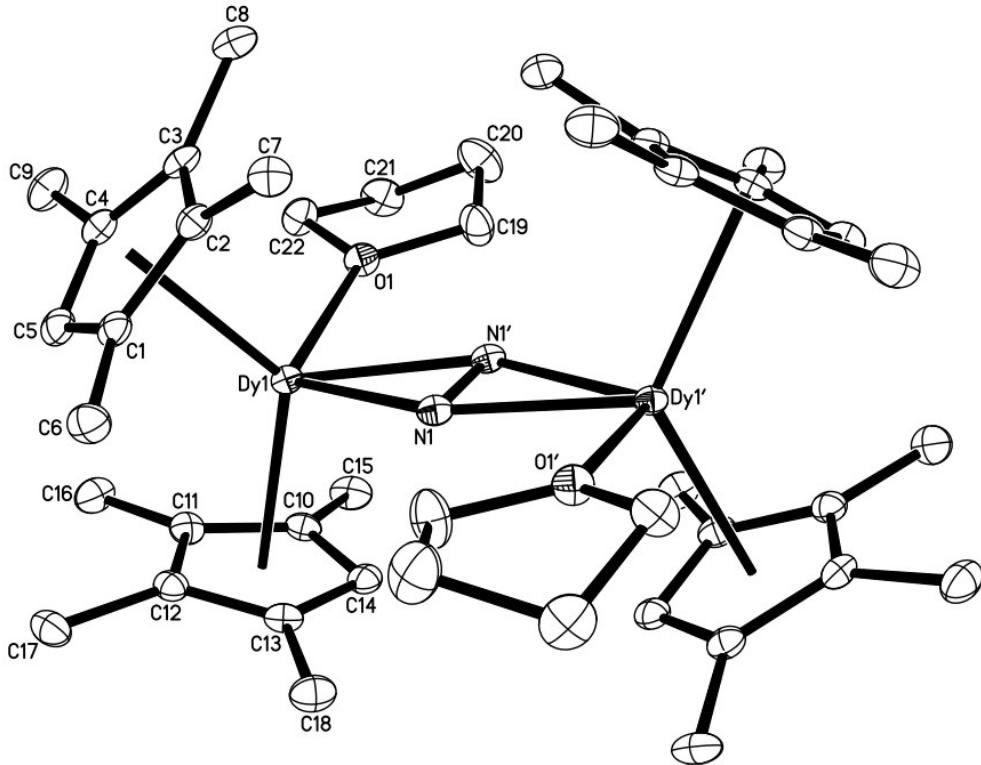


Fig. S8 Thermal ellipsoid plot of $[(\text{C}_5\text{Me}_4\text{H})_2\text{Dy}(\text{THF})](\mu\text{-}\eta^2\text{:}\eta^2\text{-N}_2)$, **2-Dy**, drawn at the 30% probability level. Hydrogen atoms are omitted for clarity.

Table S11 Bond lengths [\AA] and angles [$^\circ$] for **2-Dy**.

Dy(1)-Cnt1	2.421
Dy(1)-Cnt2	2.412
Dy(1)-N(1)	2.335(3)
Dy(1)-N(1)#1	2.366(3)
Dy(1)-O(1)	2.486(3)
Dy(1)-C(1)	2.657(4)
Dy(1)-C(5)	2.659(4)
Dy(1)-C(14)	2.660(4)
Dy(1)-C(10)	2.690(4)
Dy(1)-C(13)	2.692(4)
Dy(1)-C(11)	2.706(4)
Dy(1)-C(2)	2.716(4)

Dy(1)-C(4)	2.728(4)
Dy(1)-C(12)	2.730(4)
Dy(1)-C(3)	2.763(4)
O(1)-C(19)	1.448(5)
O(1)-C(22)	1.454(5)
N(1)-N(1)#1	1.243(7)
N(1)-Dy(1)#1	2.366(3)
C(1)-C(5)	1.417(6)
C(1)-C(2)	1.422(6)
C(1)-C(6)	1.496(6)
C(2)-C(3)	1.422(6)
C(2)-C(7)	1.502(5)
C(3)-C(4)	1.422(6)
C(3)-C(8)	1.503(6)
C(4)-C(5)	1.412(6)
C(4)-C(9)	1.514(6)
C(10)-C(14)	1.412(6)
C(10)-C(11)	1.418(6)
C(10)-C(15)	1.506(6)
C(11)-C(12)	1.411(6)
C(11)-C(16)	1.505(6)
C(12)-C(13)	1.410(6)
C(12)-C(17)	1.509(6)
C(13)-C(14)	1.429(6)
C(13)-C(18)	1.512(6)
C(19)-C(20)	1.502(7)
C(20)-C(21)	1.516(7)
C(21)-C(22)	1.511(6)

Cnt1-Dy(1)-O(1)	102.9
Cnt1-Dy(1)-N(1)	105.8
Cnt1-Dy(1)-N(1)#1	118.7
Cnt2-Dy(1)-O(1)	105.2
Cnt2-Dy(1)-N(1)	104.4
Cnt2-Dy(1)-N(1)#1	109.5
Cnt1-Dy(1)-Cnt2	127.0

N(1)-Dy(1)-N(1)#1	30.66(16)
N(1)-Dy(1)-O(1)	111.38(11)
N(1)#1-Dy(1)-O(1)	81.05(11)
N(1)-Dy(1)-C(1)	89.81(12)
N(1)#1-Dy(1)-C(1)	113.62(12)
O(1)-Dy(1)-C(1)	129.89(11)
N(1)-Dy(1)-C(5)	120.68(13)
N(1)#1-Dy(1)-C(5)	142.64(12)
O(1)-Dy(1)-C(5)	112.56(12)
C(1)-Dy(1)-C(5)	30.93(13)
N(1)-Dy(1)-C(14)	84.00(11)
N(1)#1-Dy(1)-C(14)	83.06(11)
O(1)-Dy(1)-C(14)	97.19(12)
C(1)-Dy(1)-C(14)	130.88(13)
C(5)-Dy(1)-C(14)	126.48(13)
N(1)-Dy(1)-C(10)	113.33(12)
N(1)#1-Dy(1)-C(10)	103.87(12)
O(1)-Dy(1)-C(10)	78.62(11)
C(1)-Dy(1)-C(10)	134.92(13)
C(5)-Dy(1)-C(10)	112.75(13)
C(14)-Dy(1)-C(10)	30.60(12)
N(1)-Dy(1)-C(13)	81.18(11)
N(1)#1-Dy(1)-C(13)	95.83(12)
O(1)-Dy(1)-C(13)	127.10(11)
C(1)-Dy(1)-C(13)	99.93(13)
C(5)-Dy(1)-C(13)	101.47(13)
C(14)-Dy(1)-C(13)	30.97(12)
C(10)-Dy(1)-C(13)	50.67(13)
N(1)-Dy(1)-C(11)	130.55(12)
N(1)#1-Dy(1)-C(11)	132.53(12)
O(1)-Dy(1)-C(11)	94.46(12)
C(1)-Dy(1)-C(11)	105.31(13)
C(5)-Dy(1)-C(11)	82.69(13)
C(14)-Dy(1)-C(11)	50.40(12)
C(10)-Dy(1)-C(11)	30.47(12)
C(13)-Dy(1)-C(11)	50.25(13)

N(1)-Dy(1)-C(2)	80.42(11)
N(1)#1-Dy(1)-C(2)	93.04(11)
O(1)-Dy(1)-C(2)	106.50(11)
C(1)-Dy(1)-C(2)	30.68(12)
C(5)-Dy(1)-C(2)	50.22(13)
C(14)-Dy(1)-C(2)	155.12(13)
C(10)-Dy(1)-C(2)	162.96(13)
C(13)-Dy(1)-C(2)	126.39(13)
C(11)-Dy(1)-C(2)	132.69(13)
N(1)-Dy(1)-C(4)	130.21(12)
N(1)#1-Dy(1)-C(4)	132.78(12)
O(1)-Dy(1)-C(4)	83.34(12)
C(1)-Dy(1)-C(4)	50.74(13)
C(5)-Dy(1)-C(4)	30.35(13)
C(14)-Dy(1)-C(4)	143.22(13)
C(10)-Dy(1)-C(4)	116.21(13)
C(13)-Dy(1)-C(4)	128.44(13)
C(11)-Dy(1)-C(4)	92.82(14)
C(2)-Dy(1)-C(4)	49.96(13)
N(1)-Dy(1)-C(12)	108.25(12)
N(1)#1-Dy(1)-C(12)	125.96(12)
O(1)-Dy(1)-C(12)	124.22(11)
C(1)-Dy(1)-C(12)	86.90(13)
C(5)-Dy(1)-C(12)	76.43(13)
C(14)-Dy(1)-C(12)	50.15(12)
C(10)-Dy(1)-C(12)	49.97(12)
C(13)-Dy(1)-C(12)	30.14(12)
C(11)-Dy(1)-C(12)	30.08(13)
C(2)-Dy(1)-C(12)	117.57(12)
C(4)-Dy(1)-C(12)	99.51(13)
N(1)-Dy(1)-C(3)	103.33(11)
N(1)#1-Dy(1)-C(3)	103.14(12)
O(1)-Dy(1)-C(3)	80.14(11)
C(1)-Dy(1)-C(3)	50.34(13)
C(5)-Dy(1)-C(3)	49.78(13)
C(14)-Dy(1)-C(3)	172.67(12)

C(10)-Dy(1)-C(3)	142.28(13)
C(13)-Dy(1)-C(3)	149.30(13)
C(11)-Dy(1)-C(3)	122.75(12)
C(2)-Dy(1)-C(3)	30.06(12)
C(4)-Dy(1)-C(3)	30.01(12)
C(12)-Dy(1)-C(3)	126.04(12)
C(19)-O(1)-C(22)	108.3(3)
C(19)-O(1)-Dy(1)	124.0(3)
C(22)-O(1)-Dy(1)	127.7(2)
N(1)#1-N(1)-Dy(1)	76.0(3)
N(1)#1-N(1)-Dy(1)#1	73.3(3)
Dy(1)-N(1)-Dy(1)#1	149.15(16)
C(5)-C(1)-C(2)	106.9(4)
C(5)-C(1)-C(6)	127.1(4)
C(2)-C(1)-C(6)	125.8(4)
C(5)-C(1)-Dy(1)	74.6(2)
C(2)-C(1)-Dy(1)	77.0(2)
C(6)-C(1)-Dy(1)	118.2(3)
C(3)-C(2)-C(1)	108.4(4)
C(3)-C(2)-C(7)	127.3(4)
C(1)-C(2)-C(7)	124.0(4)
C(3)-C(2)-Dy(1)	76.8(2)
C(1)-C(2)-Dy(1)	72.4(2)
C(7)-C(2)-Dy(1)	121.9(3)
C(2)-C(3)-C(4)	107.9(4)
C(2)-C(3)-C(8)	126.4(4)
C(4)-C(3)-C(8)	125.3(4)
C(2)-C(3)-Dy(1)	73.1(2)
C(4)-C(3)-Dy(1)	73.6(2)
C(8)-C(3)-Dy(1)	124.9(3)
C(5)-C(4)-C(3)	107.4(4)
C(5)-C(4)-C(9)	124.4(4)
C(3)-C(4)-C(9)	127.0(4)
C(5)-C(4)-Dy(1)	72.2(2)
C(3)-C(4)-Dy(1)	76.4(2)
C(9)-C(4)-Dy(1)	126.3(3)

C(4)-C(5)-C(1)	109.3(4)
C(4)-C(5)-Dy(1)	77.5(2)
C(1)-C(5)-Dy(1)	74.4(2)
C(14)-C(10)-C(11)	107.7(4)
C(14)-C(10)-C(15)	125.1(4)
C(11)-C(10)-C(15)	126.8(4)
C(14)-C(10)-Dy(1)	73.5(2)
C(11)-C(10)-Dy(1)	75.4(2)
C(15)-C(10)-Dy(1)	122.7(3)
C(12)-C(11)-C(10)	108.1(4)
C(12)-C(11)-C(16)	124.7(4)
C(10)-C(11)-C(16)	126.5(4)
C(12)-C(11)-Dy(1)	75.9(2)
C(10)-C(11)-Dy(1)	74.2(2)
C(16)-C(11)-Dy(1)	123.5(3)
C(13)-C(12)-C(11)	108.7(4)
C(13)-C(12)-C(17)	125.6(4)
C(11)-C(12)-C(17)	124.8(4)
C(13)-C(12)-Dy(1)	73.4(2)
C(11)-C(12)-Dy(1)	74.0(2)
C(17)-C(12)-Dy(1)	127.2(3)
C(12)-C(13)-C(14)	107.2(4)
C(12)-C(13)-C(18)	126.7(4)
C(14)-C(13)-C(18)	125.9(4)
C(12)-C(13)-Dy(1)	76.4(2)
C(14)-C(13)-Dy(1)	73.3(2)
C(18)-C(13)-Dy(1)	119.7(3)
C(10)-C(14)-C(13)	108.3(4)
C(10)-C(14)-Dy(1)	75.9(2)
C(13)-C(14)-Dy(1)	75.7(2)
O(1)-C(19)-C(20)	107.4(4)
C(19)-C(20)-C(21)	103.7(4)
C(22)-C(21)-C(20)	102.0(4)
O(1)-C(22)-C(21)	105.4(4)

Symmetry transformations used to generate equivalent atoms:

#1 -x,y,-z+1/2

Table S12 X-ray Data Collection Parameters of **3-Y** and **3-Dy**.

	3-Y	3-Dy
formula	C ₃₂ H ₄₄ N ₂ O ₂ Y ₂	C ₃₂ H ₄₄ N ₂ O ₂ Dy ₂
fw	666.51	813.69
temp (K)	88(2)	133(2)
Wavelength	0.71073 Å	0.71073 Å
cryst syst	Monoclinic	Monoclinic
space group	P2 ₁ /n	P2 ₁ /n
a (Å)	11.9100(9)	11.9046(7)
b (Å)	10.0376(7)	10.0279(6)
c (Å)	12.6988(9)	12.7468(8)
Volume (Å ³)	1510.39(19)	1513.95(16)
α (deg)	90	90
β (deg)	95.7802(9)	95.7793(8)
γ (deg)	90	90
Z	2	2
ρ _{calc} (Mg/m ³)	1.466	1.785
μ (mm ⁻¹)	3.853	4.927
R1 (I > 2σ(I)) ^a	0.0193	0.0164
wR2 (all data) ^a	0.0453	0.0403

^aDefinitions: wR2 = [Σ[w(F_o² - F_c²)²] / Σ[w(F_o²)]]^{1/2}; R1 = Σ||F_o| - |F_c|| / Σ|F_o|.

X-ray Data Collection, Structure Solution and Refinement for 3-Y. A blue crystal of approximate dimensions 0.148 x 0.184 x 0.273 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2¹⁷ program package was used to determine the unit-cell parameters and for data collection (15 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT⁸ and SADABS¹⁸ to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL¹⁹ program. The diffraction symmetry was $2/m$ and the systematic absences were consistent with the monoclinic space group $P2_1/n$ that was later determined to be correct. The structure was solved by direct methods and refined on F^2 by full-matrix least-squares techniques. The analytical scattering factors⁶ for neutral atoms were used throughout the analysis. Hydrogen atoms were located from a difference-Fourier map and refined (x,y,z and U_{iso}). The molecule was located about an inversion center. At convergence, $wR2 = 0.0453$ and $Goof = 1.034$ for 260 variables refined against 3825 data (0.73Å), $R1 = 0.0193$ for those 3517 data with $I > 2.0\sigma(I)$.

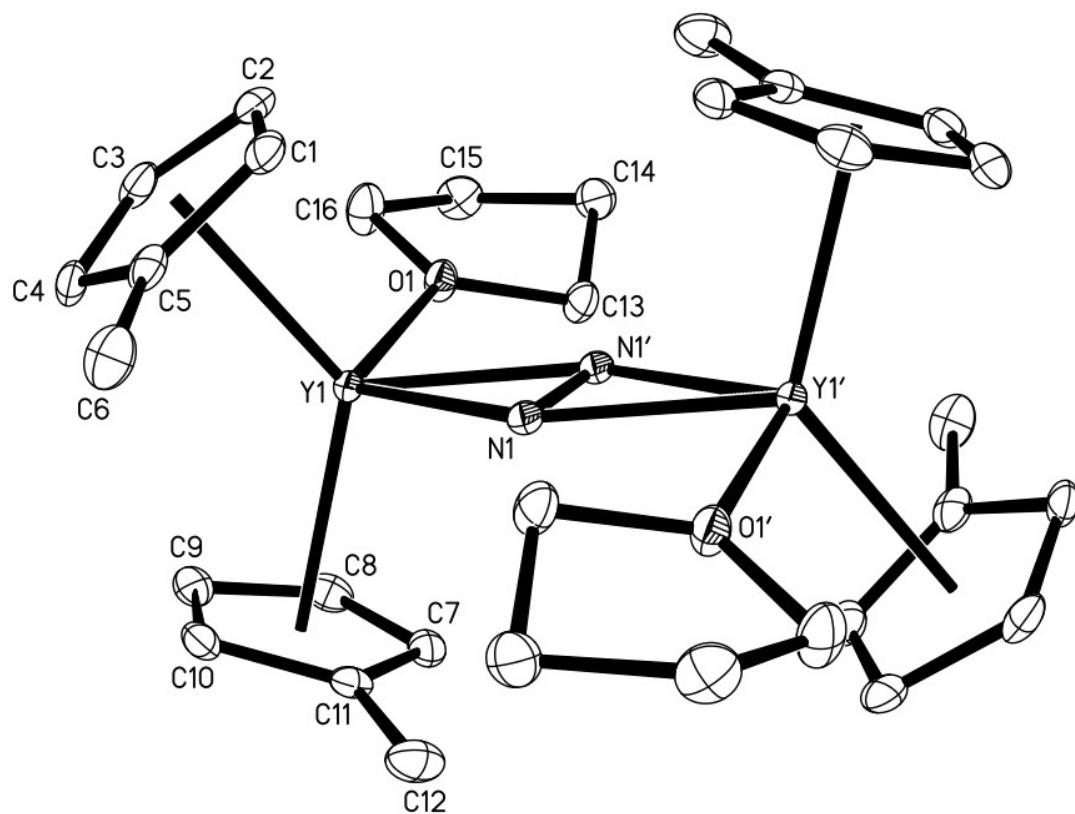


Fig. S9 Thermal ellipsoid plot of $[(\text{C}_5\text{H}_4\text{Me})_2(\text{THF})\text{Y}]_2(\mu\text{-}\eta^2\text{:}\eta^2\text{-N}_2)$, **3-Y** drawn at the 50% probability level. Hydrogen atoms are omitted for clarity.

Table S13 Bond lengths [\AA] and angles [$^\circ$] for **3-Y**.

Y(1)-Cnt1	2.387
Y(1)-Cnt2	2.381
Y(1)-N(1)	2.3023(11)
Y(1)-N(1)#1	2.3292(11)
Y(1)-O(1)	2.4046(10)
Y(1)-C(10)	2.6427(13)
Y(1)-C(9)	2.6591(14)
Y(1)-C(3)	2.6599(13)
Y(1)-C(2)	2.6673(14)
Y(1)-C(11)	2.6732(13)

Y(1)-C(4)	2.6741(13)
Y(1)-C(1)	2.6758(14)
Y(1)-C(7)	2.6765(14)
Y(1)-C(8)	2.6842(14)
O(1)-C(16)	1.4556(17)
O(1)-C(13)	1.4617(16)
N(1)-N(1)#1	1.250(2)
N(1)-Y(1)#1	2.3292(11)
C(1)-C(2)	1.407(2)
C(1)-C(5)	1.418(2)
C(2)-C(3)	1.413(2)
C(3)-C(4)	1.415(2)
C(4)-C(5)	1.4135(19)
C(5)-C(6)	1.504(2)
C(7)-C(11)	1.4114(19)
C(7)-C(8)	1.414(2)
C(8)-C(9)	1.406(2)
C(9)-C(10)	1.417(2)
C(10)-C(11)	1.4114(19)
C(11)-C(12)	1.5031(19)
C(13)-C(14)	1.512(2)
C(14)-C(15)	1.525(2)
C(15)-C(16)	1.522(2)

Cnt1-Y(1)-N(1)	104.0
Cnt1-Y(1)-N(1')	115.8
Cnt1-Y(1)-O(1)	103.0
Cnt1-Y(1)-N(1)	102.6
Cnt1-Y(1)-N(1')	109.4
Cnt1-Y(1)-O(1)	103.9
Cnt1-Y(1)-Cnt2	129.1
N(1)-Y(1)-N(1)#1	31.30(5)
N(1)-Y(1)-O(1)	114.83(4)
N(1)#1-Y(1)-O(1)	83.65(4)
N(1)-Y(1)-C(10)	100.72(4)
N(1)#1-Y(1)-C(10)	120.95(4)

O(1)-Y(1)-C(10)	126.80(4)
N(1)-Y(1)-C(9)	127.82(4)
N(1)#1-Y(1)-C(9)	135.43(4)
O(1)-Y(1)-C(9)	97.41(4)
C(10)-Y(1)-C(9)	31.00(4)
N(1)-Y(1)-C(3)	129.50(4)
N(1)#1-Y(1)-C(3)	131.47(4)
O(1)-Y(1)-C(3)	82.99(4)
C(10)-Y(1)-C(3)	104.19(5)
C(9)-Y(1)-C(3)	92.54(5)
N(1)-Y(1)-C(2)	103.08(4)
N(1)#1-Y(1)-C(2)	100.83(4)
O(1)-Y(1)-C(2)	79.29(4)
C(10)-Y(1)-C(2)	130.70(4)
C(9)-Y(1)-C(2)	123.27(5)
C(3)-Y(1)-C(2)	30.75(5)
N(1)-Y(1)-C(11)	76.87(4)
N(1)#1-Y(1)-C(11)	90.91(4)
O(1)-Y(1)-C(11)	122.30(4)
C(10)-Y(1)-C(11)	30.79(4)
C(9)-Y(1)-C(11)	51.01(4)
C(3)-Y(1)-C(11)	134.85(4)
C(2)-Y(1)-C(11)	156.75(4)
N(1)-Y(1)-C(4)	117.60(4)
N(1)#1-Y(1)-C(4)	139.16(4)
O(1)-Y(1)-C(4)	112.76(4)
C(10)-Y(1)-C(4)	80.06(4)
C(9)-Y(1)-C(4)	81.40(4)
C(3)-Y(1)-C(4)	30.76(4)
C(2)-Y(1)-C(4)	50.66(4)
C(11)-Y(1)-C(4)	108.23(4)
N(1)-Y(1)-C(1)	78.99(4)
N(1)#1-Y(1)-C(1)	89.81(4)
O(1)-Y(1)-C(1)	106.30(4)
C(10)-Y(1)-C(1)	118.83(5)
C(9)-Y(1)-C(1)	131.25(4)

C(3)-Y(1)-C(1)	50.52(5)
C(2)-Y(1)-C(1)	30.53(5)
C(11)-Y(1)-C(1)	131.16(5)
C(4)-Y(1)-C(1)	50.31(4)
N(1)-Y(1)-C(7)	86.85(4)
N(1)#1-Y(1)-C(7)	84.84(4)
O(1)-Y(1)-C(7)	91.80(4)
C(10)-Y(1)-C(7)	50.58(4)
C(9)-Y(1)-C(7)	50.61(4)
C(3)-Y(1)-C(7)	141.93(4)
C(2)-Y(1)-C(7)	168.74(5)
C(11)-Y(1)-C(7)	30.59(4)
C(4)-Y(1)-C(7)	129.12(4)
C(1)-Y(1)-C(7)	160.46(5)
N(1)-Y(1)-C(8)	117.37(4)
N(1)#1-Y(1)-C(8)	109.70(4)
O(1)-Y(1)-C(8)	77.49(4)
C(10)-Y(1)-C(8)	50.67(4)
C(9)-Y(1)-C(8)	30.52(4)
C(3)-Y(1)-C(8)	112.45(5)
C(2)-Y(1)-C(8)	138.89(5)
C(11)-Y(1)-C(8)	50.70(4)
C(4)-Y(1)-C(8)	110.31(4)
C(1)-Y(1)-C(8)	160.48(4)
C(7)-Y(1)-C(8)	30.59(4)
C(16)-O(1)-C(13)	109.29(10)
C(16)-O(1)-Y(1)	127.91(8)
C(13)-O(1)-Y(1)	122.44(8)
N(1)#1-N(1)-Y(1)	75.54(10)
N(1)#1-N(1)-Y(1)#1	73.17(10)
Y(1)-N(1)-Y(1)#1	148.70(5)
C(2)-C(1)-C(5)	108.97(13)
C(2)-C(1)-Y(1)	74.40(8)
C(5)-C(1)-Y(1)	75.09(8)
C(1)-C(2)-C(3)	107.70(13)
C(1)-C(2)-Y(1)	75.06(8)

C(3)-C(2)-Y(1)	74.33(8)
C(2)-C(3)-C(4)	107.84(13)
C(2)-C(3)-Y(1)	74.91(8)
C(4)-C(3)-Y(1)	75.17(8)
C(5)-C(4)-C(3)	108.65(13)
C(5)-C(4)-Y(1)	75.20(8)
C(3)-C(4)-Y(1)	74.06(8)
C(4)-C(5)-C(1)	106.84(13)
C(4)-C(5)-C(6)	127.20(13)
C(1)-C(5)-C(6)	125.93(13)
C(4)-C(5)-Y(1)	74.22(8)
C(1)-C(5)-Y(1)	74.24(8)
C(6)-C(5)-Y(1)	118.74(9)
C(11)-C(7)-C(8)	108.55(12)
C(11)-C(7)-Y(1)	74.58(8)
C(8)-C(7)-Y(1)	75.00(8)
C(9)-C(8)-C(7)	107.93(13)
C(9)-C(8)-Y(1)	73.76(8)
C(7)-C(8)-Y(1)	74.40(8)
C(8)-C(9)-C(10)	107.71(12)
C(8)-C(9)-Y(1)	75.73(8)
C(10)-C(9)-Y(1)	73.86(8)
C(11)-C(10)-C(9)	108.55(12)
C(11)-C(10)-Y(1)	75.80(8)
C(9)-C(10)-Y(1)	75.14(8)
C(7)-C(11)-C(10)	107.25(12)
C(7)-C(11)-C(12)	127.14(13)
C(10)-C(11)-C(12)	125.53(13)
C(7)-C(11)-Y(1)	74.83(8)
C(10)-C(11)-Y(1)	73.41(8)
C(12)-C(11)-Y(1)	119.83(9)
O(1)-C(13)-C(14)	104.31(11)
C(13)-C(14)-C(15)	102.43(12)
C(16)-C(15)-C(14)	102.18(12)
O(1)-C(16)-C(15)	106.22(11)

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+2,-z+1

X-ray Data Collection, Structure Solution and Refinement for 3-Dy. A green crystal of approximate dimensions 0.169 x 0.199 x 0.265 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2²⁰ program package and the CELL_NOW¹⁵ were used to determine the unit-cell parameters. Data was collected using a 10 sec/frame scan time for a sphere of diffraction data. The raw frame data was processed using SAINT⁸ and TWINABS¹⁶ to yield the reflection data file (HKLF 5 format)¹⁶. Subsequent calculations were carried out using the SHELXTL⁵ program. The diffraction symmetry was $2/m$ and the systematic absences were consistent with the monoclinic space group $P2_1/n$ that was later determined to be correct. The structure was solved by direct methods and refined on F^2 by full-matrix least-squares techniques. The analytical scattering factors⁶ for neutral atoms were used throughout the analysis. Hydrogen atoms were included using a riding model. The molecule was located about an inversion center ($Z = 2$). At convergence, $wR2 = 0.0403$ and $Goof = 1.036$ for 175 variables refined against 3608 data (0.74Å), $R1 = 0.0164$ for those 3406 with $I > 2.0\sigma(I)$. The structure was refined as a two-component twin, $BASF^5 = 0.4066$.

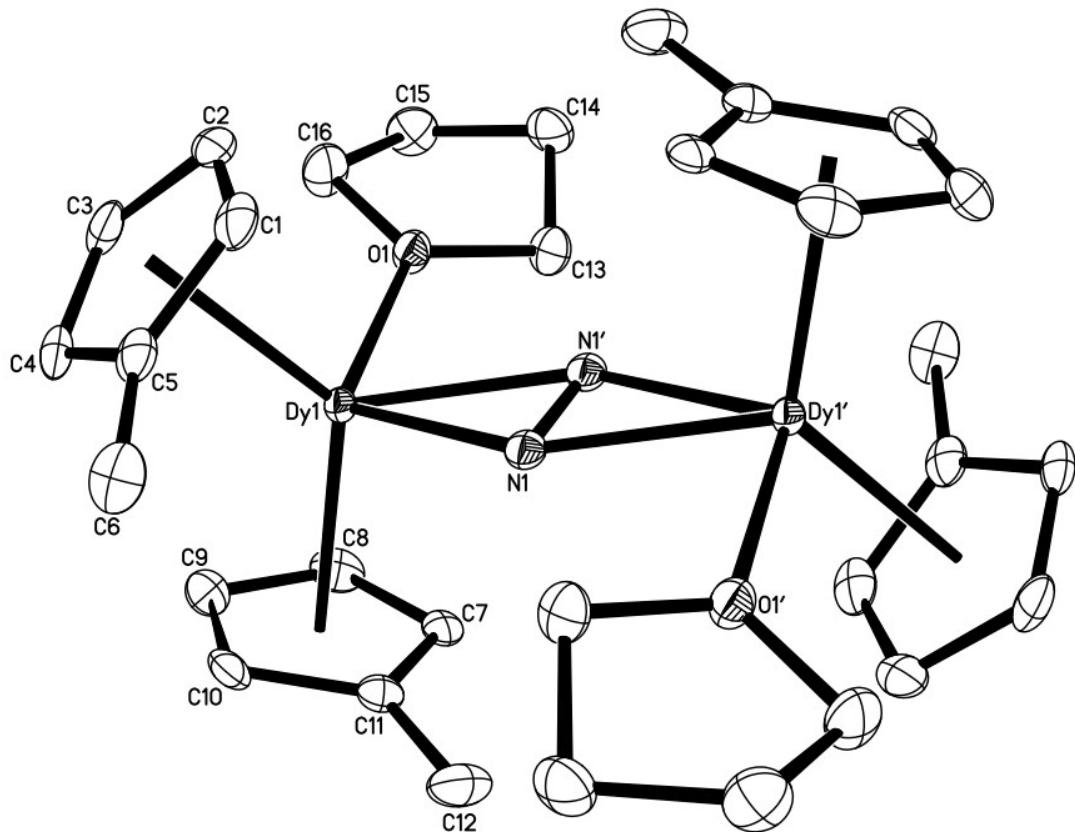


Fig. S10 Thermal ellipsoid plot of $[(C_5H_4Me)_2(\text{THF})\text{Dy}]_2(\mu-\eta^2:\eta^2\text{-N}_2)$, **3-Dy** drawn at the 50% probability level. Hydrogen atoms are omitted for clarity.

Table S14 Bond lengths [\AA] and angles [$^\circ$] for **3-Dy**.

Dy(1)-Cnt1	2.391
Dy(1)-Cnt2	2.385
Dy(1)-N(1)	2.296(2)
Dy(1)-N(1)#1	2.334(2)
Dy(1)-O(1)	2.4156(17)
Dy(1)-C(10)	2.645(2)
Dy(1)-C(9)	2.661(3)
Dy(1)-C(3)	2.662(3)
Dy(1)-C(2)	2.673(3)
Dy(1)-C(4)	2.674(2)
Dy(1)-C(7)	2.677(2)

Dy(1)-C(11)	2.677(2)
Dy(1)-C(1)	2.681(3)
Dy(1)-C(8)	2.686(3)
O(1)-C(16)	1.451(3)
O(1)-C(13)	1.460(3)
N(1)-N(1)#1	1.250(4)
N(1)-Dy(1)#1	2.334(2)
C(1)-C(2)	1.395(4)
C(1)-C(5)	1.420(4)
C(2)-C(3)	1.419(4)
C(3)-C(4)	1.412(4)
C(4)-C(5)	1.415(4)
C(5)-C(6)	1.500(4)
C(7)-C(8)	1.407(4)
C(7)-C(11)	1.411(4)
C(8)-C(9)	1.404(4)
C(9)-C(10)	1.419(4)
C(10)-C(11)	1.405(4)
C(11)-C(12)	1.497(4)
C(13)-C(14)	1.510(4)
C(14)-C(15)	1.516(4)
C(15)-C(16)	1.520(4)

Cnt1-Dy(1)-N(1)	104.0
Cnt1-Dy(1)-N(1)#1	116.0
Cnt1-Dy(1)-O(1)	103.1
Cnt2-Dy(1)-N(1)	102.6
Cnt2-Dy(1)-N(1)#1	109.3
Cnt2-Dy(1)-O(1)	103.9
Cnt1-Dy(1)-Cnt2	129.1
N(1)-Dy(1)-N(1)#1	31.32(9)
N(1)-Dy(1)-O(1)	114.57(6)
N(1)#1-Dy(1)-O(1)	83.38(6)
N(1)-Dy(1)-C(10)	100.44(8)
N(1)#1-Dy(1)-C(10)	120.54(8)
O(1)-Dy(1)-C(10)	127.02(8)

N(1)-Dy(1)-C(9)	127.68(8)
N(1)#1-Dy(1)-C(9)	135.25(8)
O(1)-Dy(1)-C(9)	97.75(8)
C(10)-Dy(1)-C(9)	31.02(8)
N(1)-Dy(1)-C(3)	129.47(8)
N(1)#1-Dy(1)-C(3)	131.60(8)
O(1)-Dy(1)-C(3)	83.21(7)
C(10)-Dy(1)-C(3)	104.47(9)
C(9)-Dy(1)-C(3)	92.61(9)
N(1)-Dy(1)-C(2)	102.86(9)
N(1)#1-Dy(1)-C(2)	100.88(8)
O(1)-Dy(1)-C(2)	79.62(8)
C(10)-Dy(1)-C(2)	130.87(8)
C(9)-Dy(1)-C(2)	123.45(9)
C(3)-Dy(1)-C(2)	30.85(9)
N(1)-Dy(1)-C(4)	117.74(8)
N(1)#1-Dy(1)-C(4)	139.43(8)
O(1)-Dy(1)-C(4)	112.89(7)
C(10)-Dy(1)-C(4)	80.24(8)
C(9)-Dy(1)-C(4)	81.27(8)
C(3)-Dy(1)-C(4)	30.69(8)
C(2)-Dy(1)-C(4)	50.63(8)
N(1)-Dy(1)-C(7)	87.17(8)
N(1)#1-Dy(1)-C(7)	84.93(8)
O(1)-Dy(1)-C(7)	91.55(7)
C(10)-Dy(1)-C(7)	50.38(8)
C(9)-Dy(1)-C(7)	50.36(8)
C(3)-Dy(1)-C(7)	141.66(9)
C(2)-Dy(1)-C(7)	168.70(9)
C(4)-Dy(1)-C(7)	128.93(8)
N(1)-Dy(1)-C(11)	76.90(8)
N(1)#1-Dy(1)-C(11)	90.71(8)
O(1)-Dy(1)-C(11)	122.06(7)
C(10)-Dy(1)-C(11)	30.61(8)
C(9)-Dy(1)-C(11)	50.86(8)
C(3)-Dy(1)-C(11)	134.98(8)

C(2)-Dy(1)-C(11)	156.76(8)
C(4)-Dy(1)-C(11)	108.39(8)
C(7)-Dy(1)-C(11)	30.57(8)
N(1)-Dy(1)-C(1)	79.15(8)
N(1)#1-Dy(1)-C(1)	90.17(8)
O(1)-Dy(1)-C(1)	106.38(8)
C(10)-Dy(1)-C(1)	118.76(9)
C(9)-Dy(1)-C(1)	130.99(8)
C(3)-Dy(1)-C(1)	50.33(9)
C(2)-Dy(1)-C(1)	30.21(9)
C(4)-Dy(1)-C(1)	50.21(8)
C(7)-Dy(1)-C(1)	160.75(9)
C(11)-Dy(1)-C(1)	131.31(9)
N(1)-Dy(1)-C(8)	117.53(8)
N(1)#1-Dy(1)-C(8)	109.73(8)
O(1)-Dy(1)-C(8)	77.60(7)
C(10)-Dy(1)-C(8)	50.63(8)
C(9)-Dy(1)-C(8)	30.44(8)
C(3)-Dy(1)-C(8)	112.31(9)
C(2)-Dy(1)-C(8)	139.00(10)
C(4)-Dy(1)-C(8)	110.01(8)
C(7)-Dy(1)-C(8)	30.42(8)
C(11)-Dy(1)-C(8)	50.61(8)
C(1)-Dy(1)-C(8)	160.09(9)
C(16)-O(1)-C(13)	109.31(18)
C(16)-O(1)-Dy(1)	127.92(14)
C(13)-O(1)-Dy(1)	122.40(14)
N(1)#1-N(1)-Dy(1)	76.03(18)
N(1)#1-N(1)-Dy(1)#1	72.65(17)
Dy(1)-N(1)-Dy(1)#1	148.68(9)
C(2)-C(1)-C(5)	109.4(3)
C(2)-C(1)-Dy(1)	74.56(15)
C(5)-C(1)-Dy(1)	74.90(15)
C(1)-C(2)-C(3)	107.7(2)
C(1)-C(2)-Dy(1)	75.23(15)
C(3)-C(2)-Dy(1)	74.15(15)

C(4)-C(3)-C(2)	107.7(3)
C(4)-C(3)-Dy(1)	75.13(14)
C(2)-C(3)-Dy(1)	75.00(15)
C(3)-C(4)-C(5)	108.7(2)
C(3)-C(4)-Dy(1)	74.18(14)
C(5)-C(4)-Dy(1)	75.21(14)
C(4)-C(5)-C(1)	106.6(2)
C(4)-C(5)-C(6)	127.2(2)
C(1)-C(5)-C(6)	126.2(3)
C(4)-C(5)-Dy(1)	74.17(14)
C(1)-C(5)-Dy(1)	74.43(15)
C(6)-C(5)-Dy(1)	118.91(17)
C(8)-C(7)-C(11)	108.9(2)
C(8)-C(7)-Dy(1)	75.15(15)
C(11)-C(7)-Dy(1)	74.72(14)
C(9)-C(8)-C(7)	107.8(2)
C(9)-C(8)-Dy(1)	73.77(15)
C(7)-C(8)-Dy(1)	74.43(15)
C(8)-C(9)-C(10)	107.7(2)
C(8)-C(9)-Dy(1)	75.78(15)
C(10)-C(9)-Dy(1)	73.86(15)
C(11)-C(10)-C(9)	108.5(2)
C(11)-C(10)-Dy(1)	75.97(14)
C(9)-C(10)-Dy(1)	75.11(14)
C(10)-C(11)-C(7)	107.1(2)
C(10)-C(11)-C(12)	126.1(2)
C(7)-C(11)-C(12)	126.8(2)
C(10)-C(11)-Dy(1)	73.42(14)
C(7)-C(11)-Dy(1)	74.71(14)
C(12)-C(11)-Dy(1)	120.03(17)
O(1)-C(13)-C(14)	104.2(2)
C(13)-C(14)-C(15)	102.7(2)
C(14)-C(15)-C(16)	102.1(2)
O(1)-C(16)-C(15)	106.2(2)

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y+1,-z+1

References

- (1) APEX2 Version 2008.3-0, Bruker AXS, Inc.; Madison, WI 2008.
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Definitions:

$$wR2 = [\sum[w(F_o^2 - F_c^2)^2] / \sum[w(F_o^2)^2]]^{1/2}$$

$$R1 = \sum|F_o| - |F_c| / \sum|F_o|$$

Goof = S = $[\sum[w(F_o^2 - F_c^2)^2] / (n-p)]^{1/2}$ where n is the number of reflections and p is the total number of parameters refined.