SUPPORTING INFORMATION

Aminopiperidine complexes for lactide polymerisation

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Characterisation of ligands and complexes



Figure SI1: ¹H NMR (CDCl₃ 400MHz) spectrum of aromatic region for $1H_2$ precursor. Inset: ESI-TOF spectrum



Figure SI2: ¹H EXSY NMR (CDCl₃, 500 MHz, mixing time 0.6 s) spectrum with blue cross peaks showing chemical exchange and red cross peaks showing NOE.



Figure SI3: ¹H NMR (CDCl₃ 400MHz) spectrum of 1H₂



Figure SI4: ¹³C{¹H} NMR (CDCl₃, 100MHz) spectrum of **1**H₂



Figure SI5: Solid state structure of 1H₂. Ellipsoids are shown at the 30% probability level.



Figure SI6: ¹H NMR (CDCl₃ 400MHz) spectrum of **5**H.



Figure SI7: ¹H NMR (CDCl₃ 400MHz) spectrum of aromatic region for **5**H; Inset: ESI-TOF spectrum



Figure SI8: ¹³C{¹H} NMR (CDCl₃ 100MHz) spectrum (Red) of **5**H with ¹³C Dept (Blue) overlaid.



Figure SI9: ¹H NMR (CDCl₃ 400MHz) spectrum of 7H₂.



Figure SI10: ¹³C{¹H} NMR (CDCl₃, 100MHz) spectrum of 7H₂.



Figure SI11: ¹H NMR (CDCl₃, 400MHz) spectrum of Zr(1)₂.



Figure SI12: ¹³C{¹H} NMR (CDCl₃, 100MHz) spectrum of Zr(1)₂



Figure SI13: ¹H NMR (CDCl₃, 400MHz) spectrum of Zr(*cis*-3)₂.



Figure SI14: ¹³C{¹H} NMR (CDCl₃, 100MHz) spectrum of Zr(*cis*-3)₂.



Figure SI15: ¹H NMR (CDCl₃, 400MHz) spectrum of $Zr(trans-3)_2$. Note: impurity of another unsymmetrical $Zr(3)_2$ isomer.



Figure SI16: ¹H NMR (CDCl₃, 400MHz) spectra comparing Zr(1-3)₂ and Hf(*trans*-3)₂.







Figure SI20: ¹H NMR (CDCl₃, 400MHz) spectrum of $Hf(trans-3)_{2}$. Note: Impurity of another unsymmetrical $Hf(3)_2$ isomer.





Figure SI22: ¹H NMR (CDCl₃, 400MHz) spectrum of $Hf(cis-3)_2$. Note: Impurity of $Hf(trans-3)_2$ and another unsymmetrical $Hf(3)_2$ isomer.



Figure SI23: DOSY NMR (CDCl₃, 500MHz) spectrum of $Hf(cis-3)_2$. Major series ($Hf(cis-3)_2$) diffusion coefficient = $4.90 \times 10^{-10} \text{ m}^2 \text{ s}^{-1}$, minor series ($Hf(trans-3)_2$) diffusion coefficient = $4.78 \times 10^{-10} \text{ m}^2 \text{ s}^{-1}$.



Figure SI24: ¹³C{¹H} NMR (CDCl₃, 100MHz) spectrum of Hf(*cis*-3)₂.





Figure SI26: ¹³C{¹H} NMR (CDCl₃, 100MHz) spectrum of Ti(1)(OⁱPr)₂



Figure SI27: ¹H NMR (d₈-toluene, 400MHz) spectrum of Al(1)Me



Figure SI29: ¹H NMR (d₈-toluene, 400MHz) spectrum of Al(4)Me₂



Figure SI30: ¹³C{¹H} NMR (d₈-toluene, 100MHz) spectrum Al(4)Me₂





Figure SI32: ¹³C{¹H} NMR (d₆-benzene, 100MHz) spectrum Al(7)Me



Figure SI33: DOSY NMR (d₈-toluene, 500MHz) spectrum of Al(7)Me

Selected Polymer Characterisation



Figure SI34: ¹H NMR (CDCl₃, 400MHz) spectrum of heterotactic PLA synthesised from solution polymerisation with $Zr(1)_2$ – table 2, entry 1.



Figure SI35: ¹H NMR (CDCl₃, 400MHz) spectrum of PLA synthesised from bulk polymerisation with $Zr(1)_2$ – table 2, entry 3.



Figure SI36: ¹H NMR (CDCl₃, 400MHz) spectrum of atactic PLA synthesised from bulk polymerisation with $Ti(1)(O^iPr)_2$ – table 2, entry 12.



Figure SI37: ¹H NMR (CDCl₃, 400MHz) spectrum of isotactic PLA synthesised from solution polymerisation with Al(7)Me table 3, entry 7.



Figure SI38: ¹H NMR (CDCl₃, 400MHz) spectrum of PLA synthesised from bulk polymerisation with $1H_2$.



Figure SI39: Semi-logarithmic plot for the solution polymerisation of $Zr(1)_2$. Conditions: Toluene, 80°C, [LA]:[Zr(1)_2]:[BnOH] = 100:1:1



Figure SI40: M_n and M_w/M_n against conversion for solution polymerisation of $Zr(2)_2$. Linear regression gave the equation of the line at $y = 151 \times + 59$.



Figure SI41: Semi-logarithmic plot for the solution polymerisation of $Zr(2)_2$. Conditions: Toluene, 80°C, [LA]:[Zr(1)_2]:[BnOH] = 100:1:1



Figure SI42: M_n and M_w/M_n against conversion for solution polymerisation of Hf(1)₂. Linear regression gave the equation of the line at $y = 239 \times -1364(R^2 = 0.92)$



Figure SI43: Semi-logarithmic plot for the solution polymerisation of $Hf(1)_2$ Conditions: Toluene, 80°C, [LA]:[Zr(1)_2]:[BnOH] = 100:1:1



Figure SI44: M_n and M_w/M_n against conversion for solution polymerisation of Al(4)Me₂. Linear regression gave the equation of the line at y = $148 \times -799(R^2 = 0.99)$.



Figure SI45: Semi-logarithmic plot for the solution polymerisation of $Al(4)Me_2$. Conditions: Toluene, 80°C, [LA]:[Zr(1)₂]:[BnOH] = 100:1:1



Figure SI46: GPC trace of PLA prepared in solution with $Zr(1)_2$ – table 2, entry 1.



Figure SI47: GPC trace of PLA prepared by bulk polymerisation with $Zr(1)_2$ – table 2, entry 3.



Figure SI48: GPC trace of PLA prepared by solution polymerisation with $Hf(1)_2$ – table 2, entry 7.



Figure SI49: GPC trace of PLA prepared by solution polymerisation with Al(9)Me – table 3, entry 9.



Figure SI50: GPC trace of PLA prepared by bulk polymerisation with $1H_2$



Figure SI51: MALDI-ToF of PLA from solution polymerisation with $Zr(1)_2$ - Table 2, Entry 1.



Figure SI52: MALDI-ToF of PLA from solution polymerisation with Al(2)Me - Table 3, Entry 2.



Figure SI53: MALDI-ToF of PLA from solution polymerisation with Al(4)Me $_2$ - Table 3, Entry 4.



Figure SI54: MALDI-ToF of PLA from solution polymerisation with Al(7)Me - Table 3, Entry 7.



Figure SI55: MALDI-ToF of PLA from bulk polymerisation with 1H₂.



Figure SI56: Schematic to illustrate the potential mechanism for the ROP with the bis-ligated complexes.

Table SI 1: Full X-ray crystallography parameters

Compound reference	$1 H_2$	Al(1)Me	Al(2)Me	Al(4)Me ₂	$Al(5)Me_2$	$Al(6)Me_2$	Al(7)Me	Al(8)Me
Chemical formula	$C_{28}H_{38}Cl_2N_2O_2\\$	$C_{32}H_{46}AlCl_2N_2O_2$	$C_{35}H_{46}AlBr_2N_2O_2$	C23H39AlN2O	$C_{34}H_{54}Al_2N_4O_2$	C ₂₆ H ₃₉ AlN ₂ O	C37H57AlN2O2	$C_{31}H_{45}AlN_2O_2$
Formula Mass	505.50	588.59	713.54	386.54	604.77	422.57	588.82	504.67
Crystal system	Triclinic	Triclinic	Monoclinic	Monoclinic	Monoclinic	Monoclinic	Monoclinic	Triclinic
a/Å	10.7434(9)	8.7940(8)	17.4539(4)	15.1677(3)	19.9805(11)	7.4788(3)	17.6460(7)	8.3422(8)
b/Å	11.2767(11)	12.2980(8)	8.5772(2)	11.97470(10)	8.6659(5)	14.4657(6)	10.2930(5)	9.3504(7)
c/Å	11.9090(11)	15.3990(8)	23.1237(6)	14.1461(2)	20.6358(14)	21.7312(11)	19.4100(8)	19.2341(18)
$a/^{\circ}$	85.543(8)	74.378(3)	90	90	90	90	90	100.197(7)
$eta/^{\circ}$	87.130(7)	87.933(5)	107.060(3)	115.389(2)	107.033(6)	97.752(4)	93.839(3)	101.412(8)
γ/°	70.262(8)	79.786(3)	90	90	90	90	90	94.028(7)
Unit cell volume/Å ³	1353.5(2)	1578.3(2)	3309.42(15)	2321.18(7)	3416.3(4)	2329.53(18)	3517.5(3)	1438.8(2)
Temperature/K	150(2)	150(2)	150(2)	150(2)	150(2)	150(2)	150(2)	150(2)
Space group	PError!	PError!	$P2_{1}/n$	$P2_{1}/c$	$P2_{1}/n$	$P2_{1}/c$	$P2_{1}/n$	PError!
No. of formula units per unit cell, Z	2	2	4	4	4	4	4	2
No. of reflections measured	12476	33901	28073	27262	14360	4091	30288	8567
No. of independent reflections	6195	7167	8618	4548	6007	4091	6145	8567
R _{int}	0.0212	0.0446	0.0352	0.0381	0.0564	-	0.0845	-
Final R_I values $(I > 2\sigma(I))$	0.0398	0.0367	0.0402	0.0391	0.1095	0.0970	0.1175	0.0942
Final $wR(F^2)$ values $(I > 2\sigma(I))$	0.0916	0.0946	0.0841	0.1008	0.2669	0.2046	0.2944	0.2368
Final R_1 values (all data)	0.0503	0.0465	0.0633	0.0447	0.1477	0.1215	0.1606	0.1234
Final $wR(F^2)$ values (all data)	0.0973	0.1011	0.0926	0.1043	0.2973	0.2156	0.3381	0.2628
Goodness of fit on F^2	1.025	1.061	1.022	1.020	1.041	1.120	1.067	1.076

Compound reference	$Hf(1)_2$	$Hf(2)_2$	$\mathrm{Hf}(3)_2$	$Ti(1)(O^iPr)_2$	Zr(1) ₂	$Zr(2)_2$	$Zr(3)_2$
Chemical formula	$C_{30\cdot 25}H_{39}Cl_3Hf_{0\cdot 50}N_2O_2$	$C_{57\cdot 50}H_{75}Br_4Cl_3HfN_4O_4$	$C_{68}H_{100}HfI_4N_4O_4\\$	$C_{34}H_{49}Cl_2N_2O_4Ti$	$C_{62}H_{86}Cl_4N_4O_4Zr$	$C_{61}H_{84}Br_4N_4O_4Zr$	$C_{62}H_{86}I_4N_4O_4Zr$
Formula Mass	658.23	1490.69	1723.60	668.55	1184.36	1348.18	1550.16
Crystal system	Triclinic	Triclinic	Orthorhombic	Triclinic	Triclinic	Triclinic	Monoclinic
a/Å	10.3949(2)	10.3957(5)	54.9146(5)	10.6236(5)	13.9740(8)	13.8853(9)	19.5802(5)
b/Å	13.7142(3)	13.6050(5)	35.8227(3)	11.1469(8)	14.3110(7)	14.3894(6)	18.0805(3)
$c/{ m \AA}$	22.8673(3)	22.9024(11)	13.38270(10)	16.0773(9)	17.2770(6)	17.4919(8)	19.3122(5)
$\alpha / ^{\circ}$	91.0520(10)	89.912(3)	90	101.894(6)	74.900(3)	74.488(4)	90
$eta/^{\circ}$	96.1700(10)	84.053(4)	90	102.235(4)	87.307(3)	86.590(5)	108.576(3)
$\gamma/^{\circ}$	109.546(2)	69.887(4)	90	102.233(5)	73.892(2)	74.048(5)	90
Unit cell volume/Å3	3049.12(10)	3023.2(2)	26326.3(4)	1754.47(19)	3203.6(3)	3237.5(3)	6480.7(3)
Temperature/K	150(2)	150(2)	150(2)	150(2)	150(2)	150(2)	150(2)
Space group	PError!	PError!	Fdd2	PError!	PError!	PError!	$P2_{1}/c$
No. of formula units per unit cell, Z	4	2	16	2	2	2	4
No. of reflections measured	22737	18821	51826	11314	31292	14806	62477
No. of independent reflections	11721	12294	14439	6173	31292	14806	12703
R_{int}	0.0241	0.0295	0.0337	0.0481	-	0.0205	0.0838
Final R_I values $(I > 2\sigma(I))$	0.0329	0.0401	0.0186	0.0749	0.1065	0.0640	0.0529
Final $wR(F^2)$ values $(I > 2\sigma(I))$	0.0840	0.0945	0.0374	0.2023	0.2707	0.1458	0.1328
Final R_i values (all data)	0.0333	0.0530	0.0216	0.0798	0.1794	0.1054	0.0639
Final $wR(F^2)$ values (all data)	0.0843	0.1015	0.0377	0.2097	0.3112	0.1685	0.1423
Goodness of fit on F^2	1.061	1.065	0.990	1.036	1.003	1.028	1.018