

SUPPORTING INFORMATION

Versatile pyridine-2,6-bis-tetrazolate scaffolds for the formation of highly luminescent lanthanide complexes

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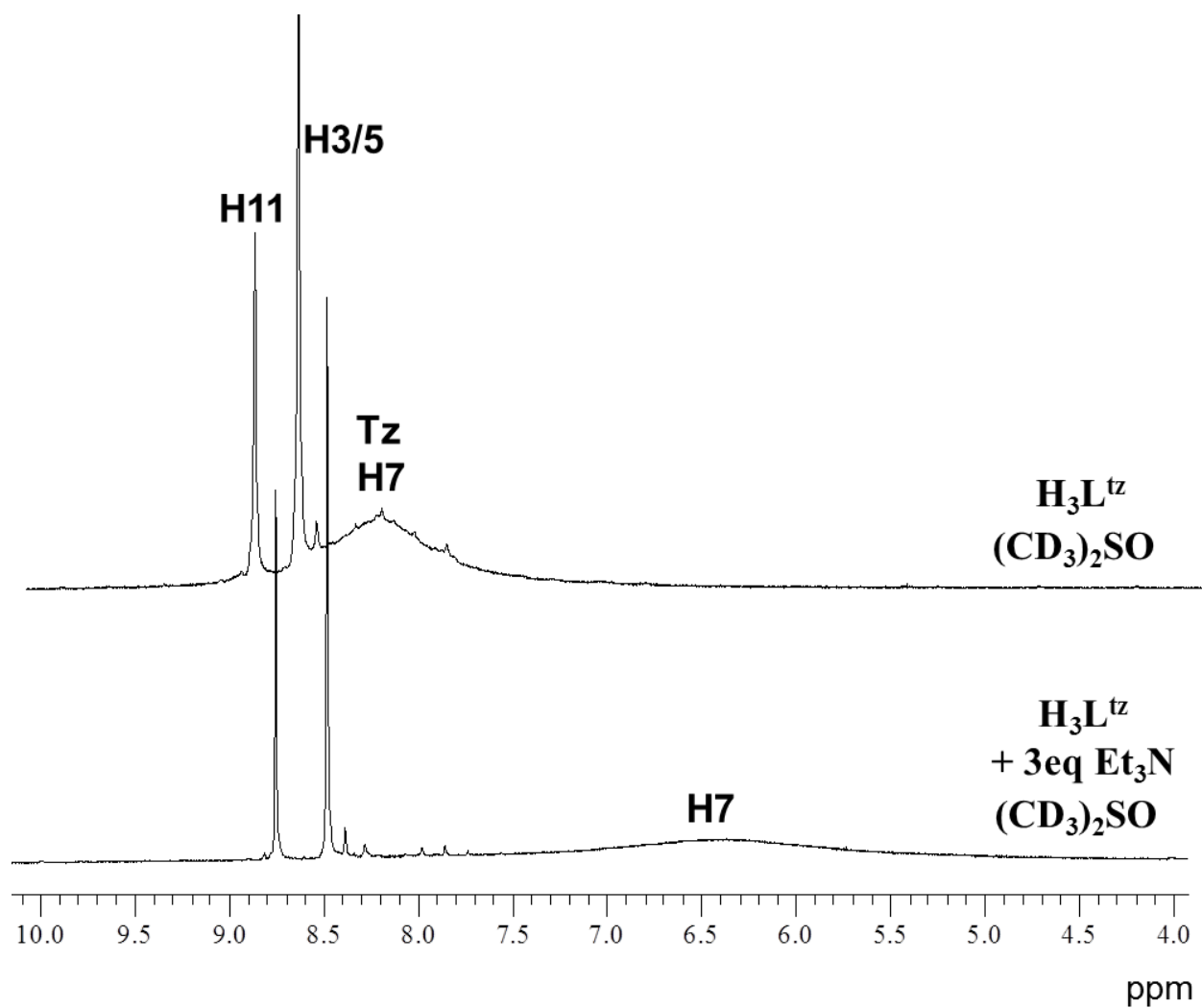


Figure S 1. ^1H NMR spectra (200 MHz) of $\text{H}_3\text{L}^{\text{tz}}$ in DMSO_d_6 .

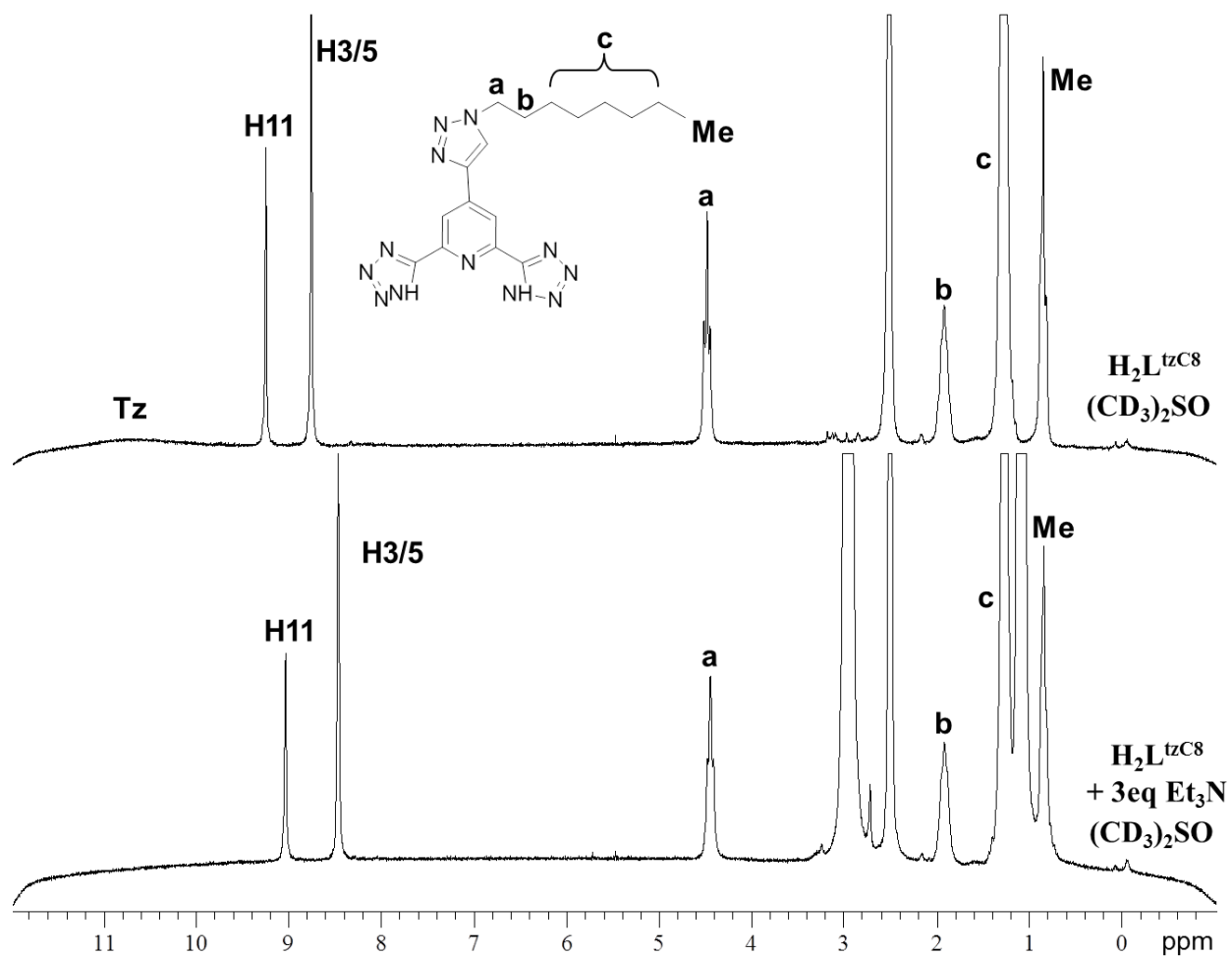


Figure S 2. ^1H NMR spectra (200 MHz) $\text{H}_2\text{L}^{\text{tzC8}}$ in DMSO-d_6 .

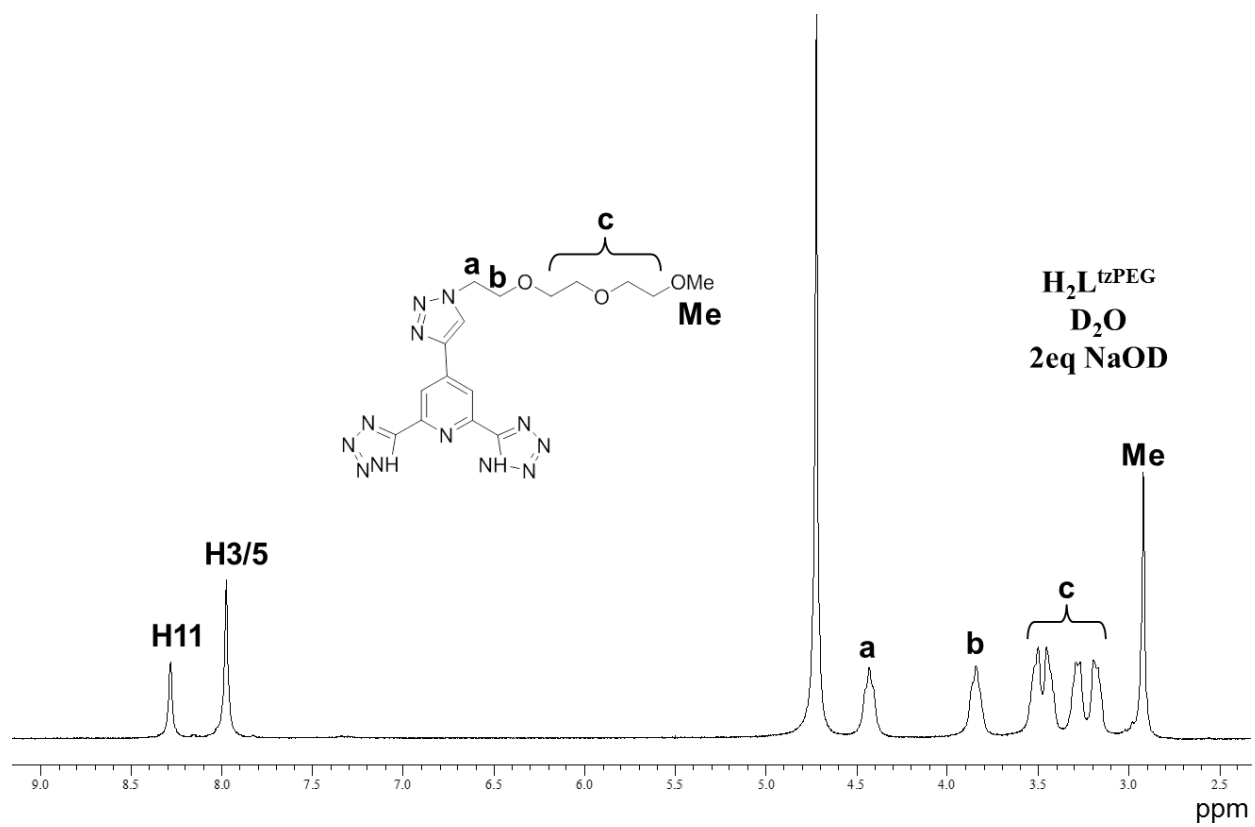


Figure S 3. ^1H NMR spectrum (200 MHz) of $\text{H}_2\text{L}^{\text{tzPEG}}$ in D_2O .

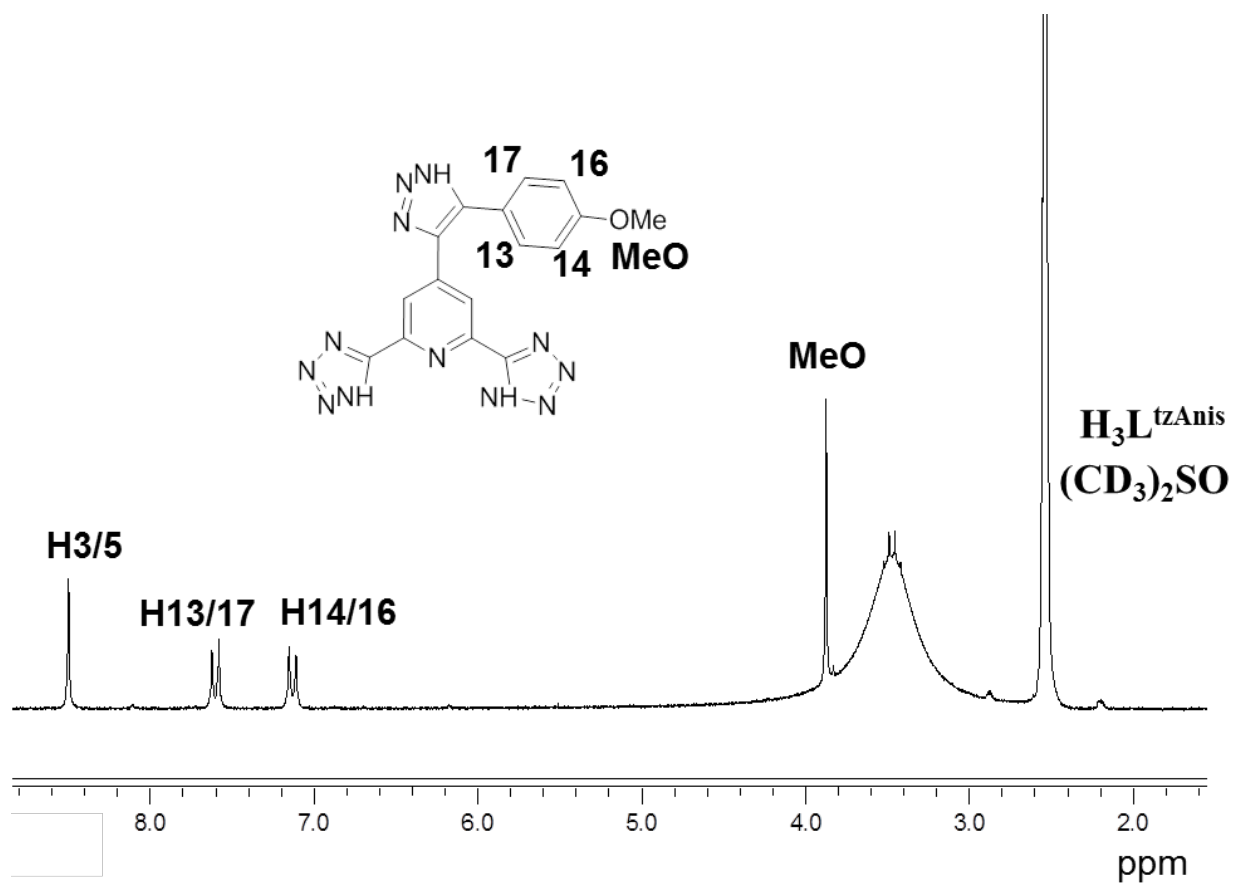


Figure S 4. ^1H NMR spectrum (200 MHz) of $\text{H}_3\text{L}^{\text{tzAnis}}$ in DMSO-d_6

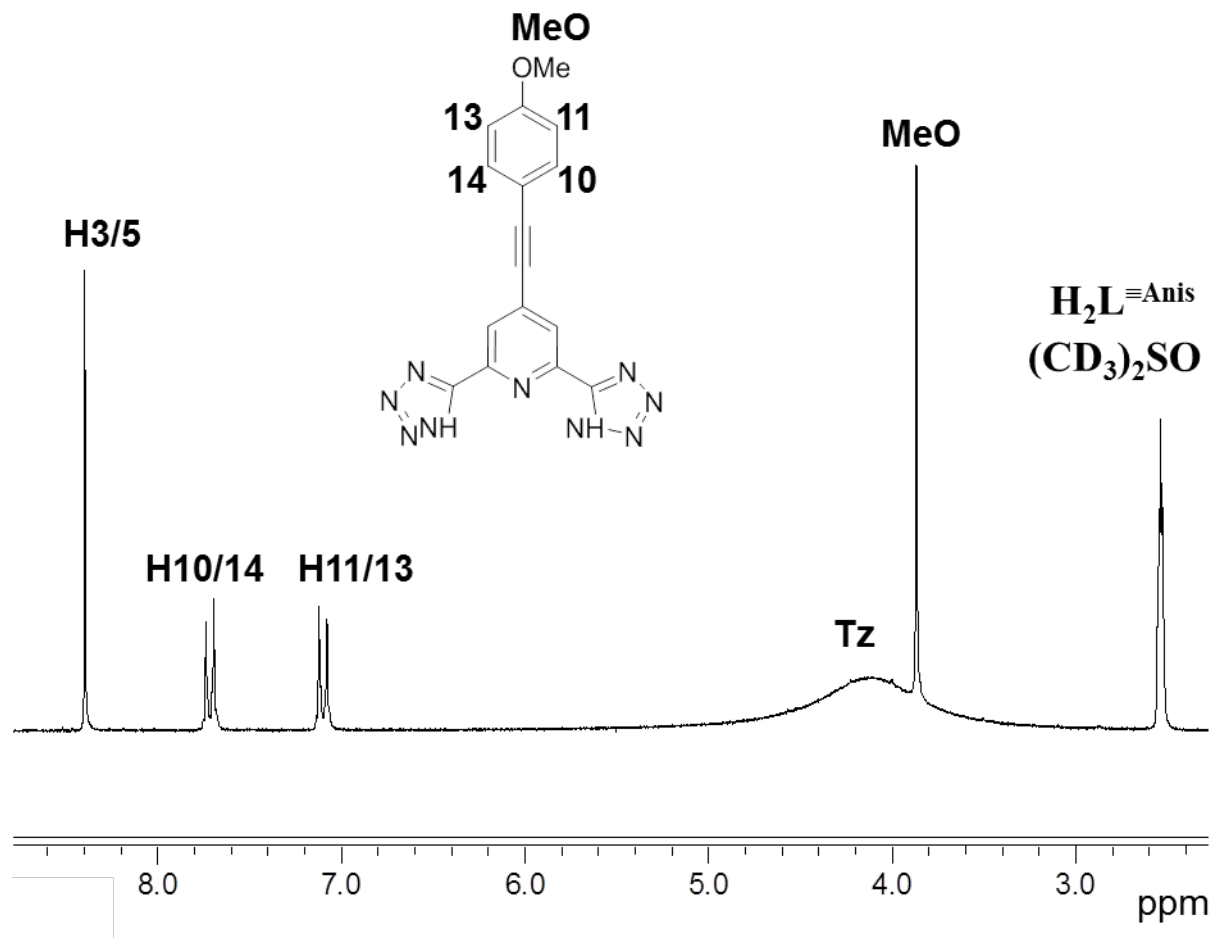


Figure S 5. H_2L_5 ^1H NMR spectrum (200 MHz) of $\text{H}_2\text{L}^{\equiv\text{Anis}}$ in DMSO-d_6 .

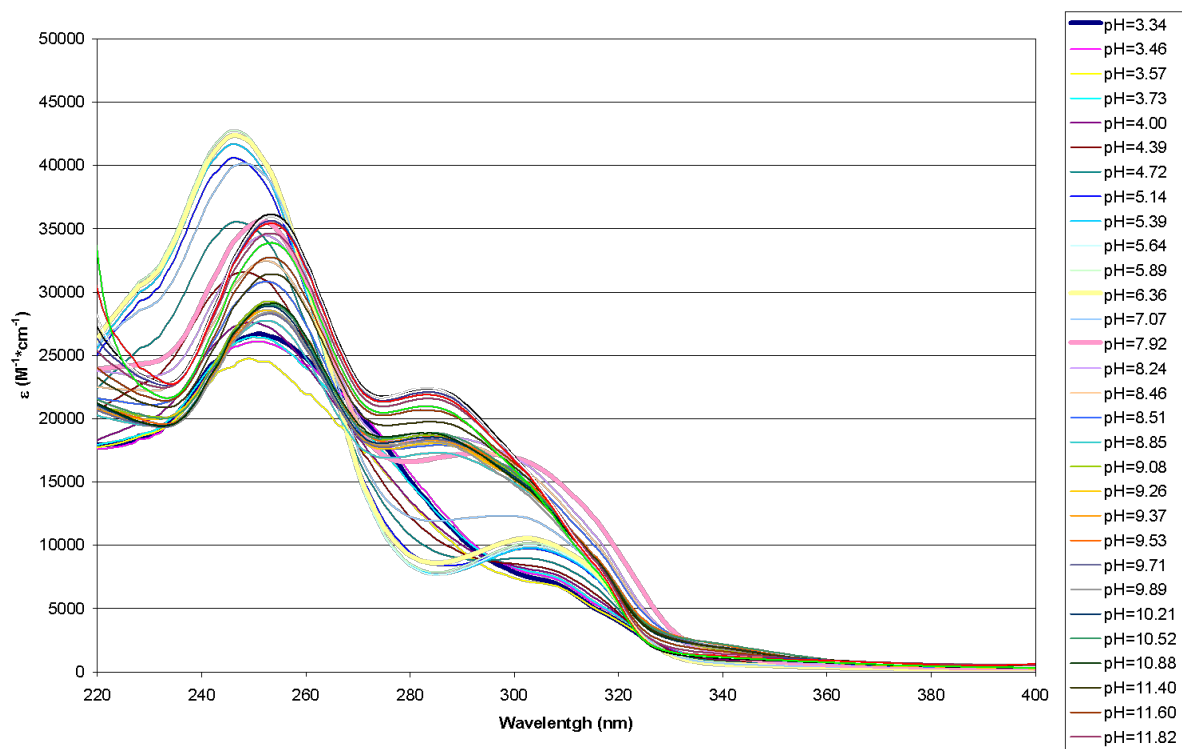


Figure S 6. UV-Vis absorption spectra of L^{tz} ($2.5 \times 10^{-5} \text{ M}$) as a function of pH, $I = 1 \text{ M}$ (KCl) in water.

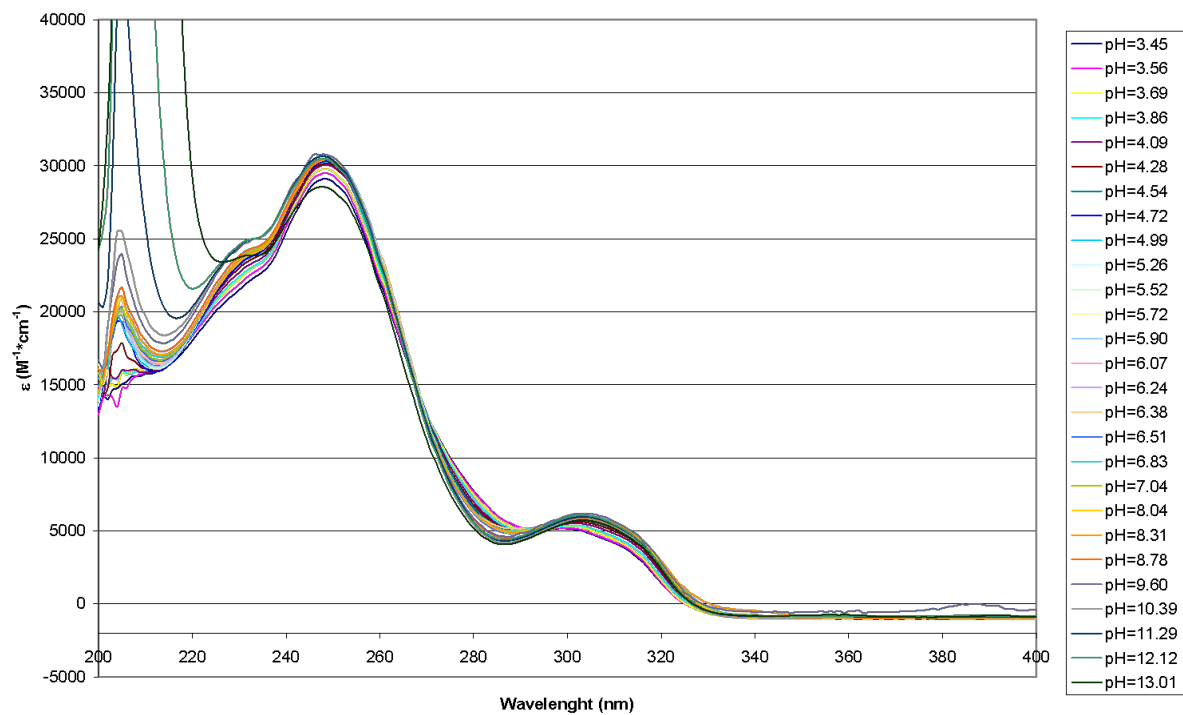


Figure S 7. UV-Vis absorption spectra of L^{tzPEG} ($2.5 \times 10^{-5} M$) as a function of pH, $I = 1M$ (KCl) in water.

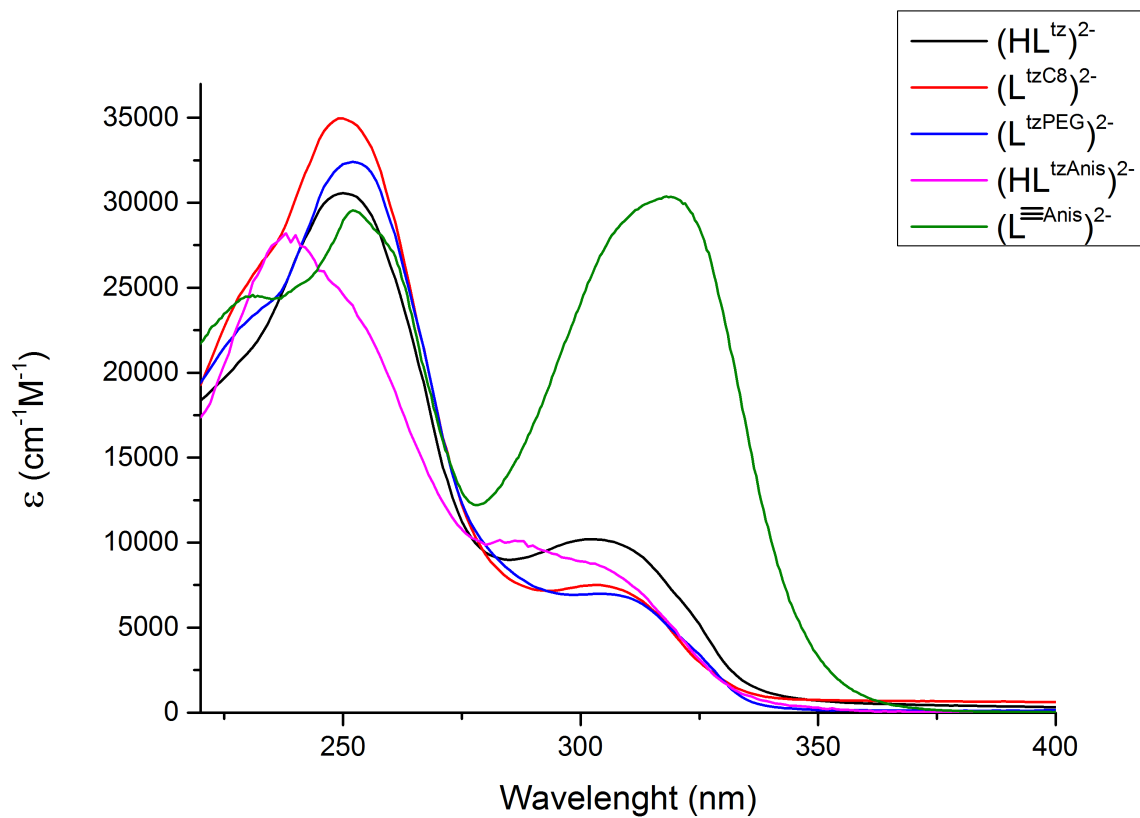


Figure S 8. Calculated extinction coefficient spectra of all the ligands ($2.5 \cdot 10^{-5}\text{M}$), $I = 1\text{M}$ (TMACl) in MeOH.

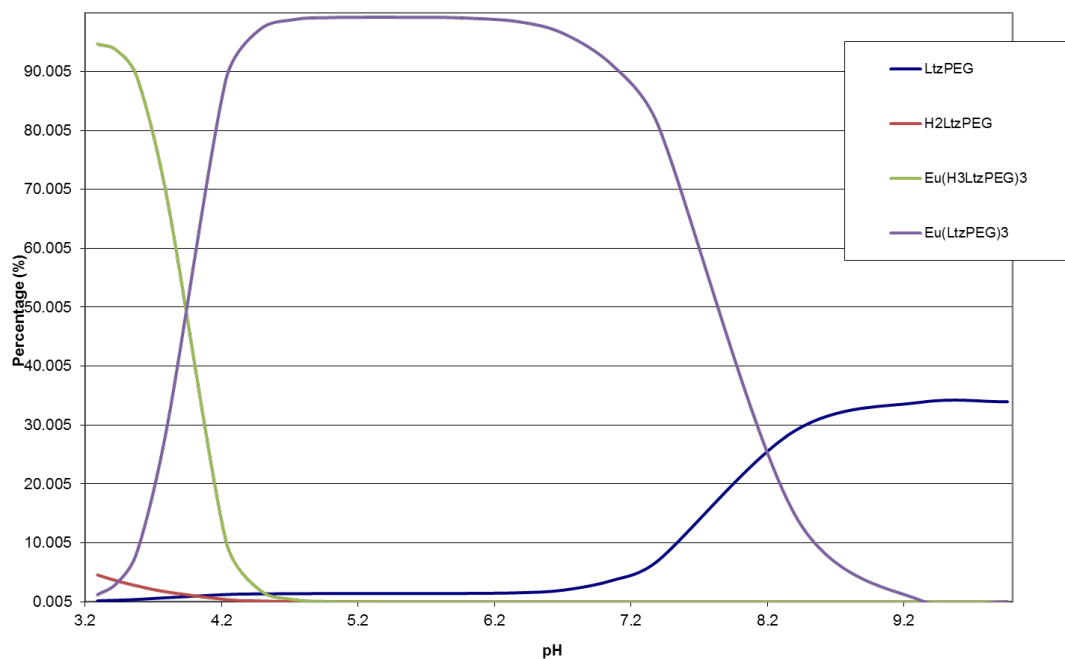


Figure S 9. Species distribution curves for $[\text{Eu}(\text{L}^{\text{tzPEG}})_3][\text{Et}_3\text{NH}]_3$ as a function of pH.

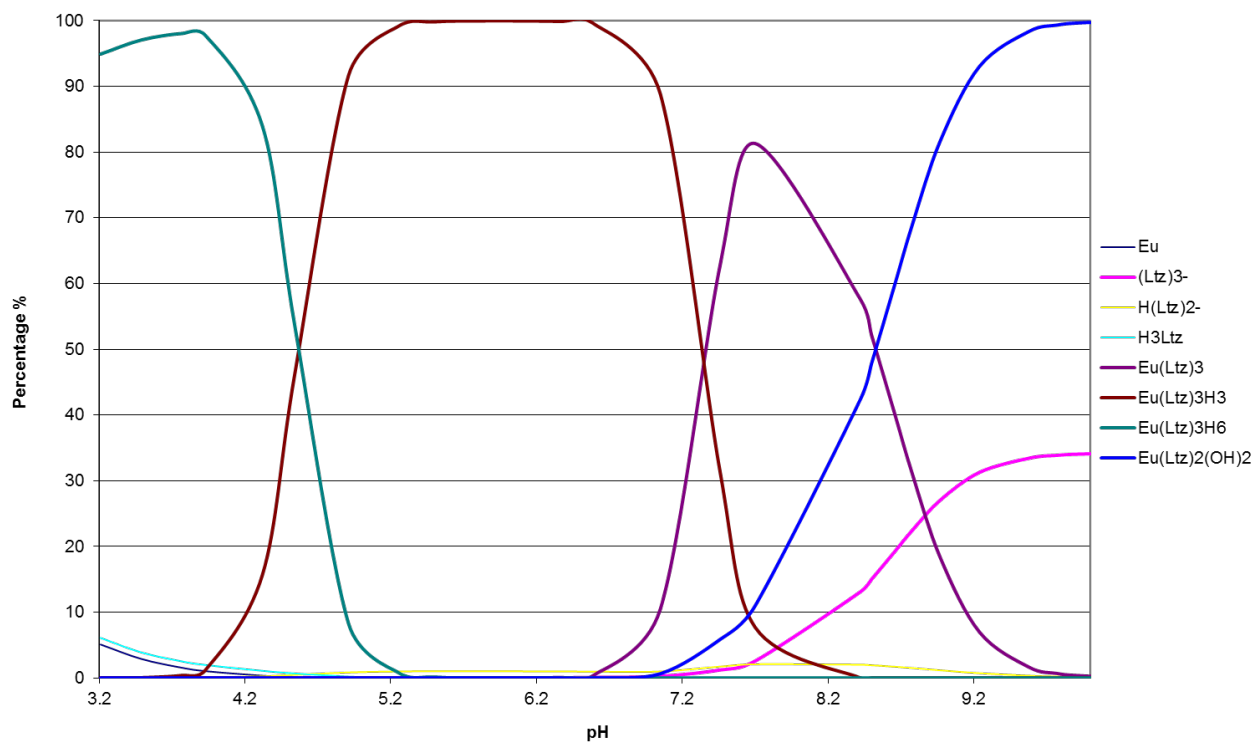


Figure S 10. Species distribution curves for $[\text{Eu}(\text{HL}^{\text{tz}})_3][\text{Et}_3\text{NH}]_3$ in function of the pH.

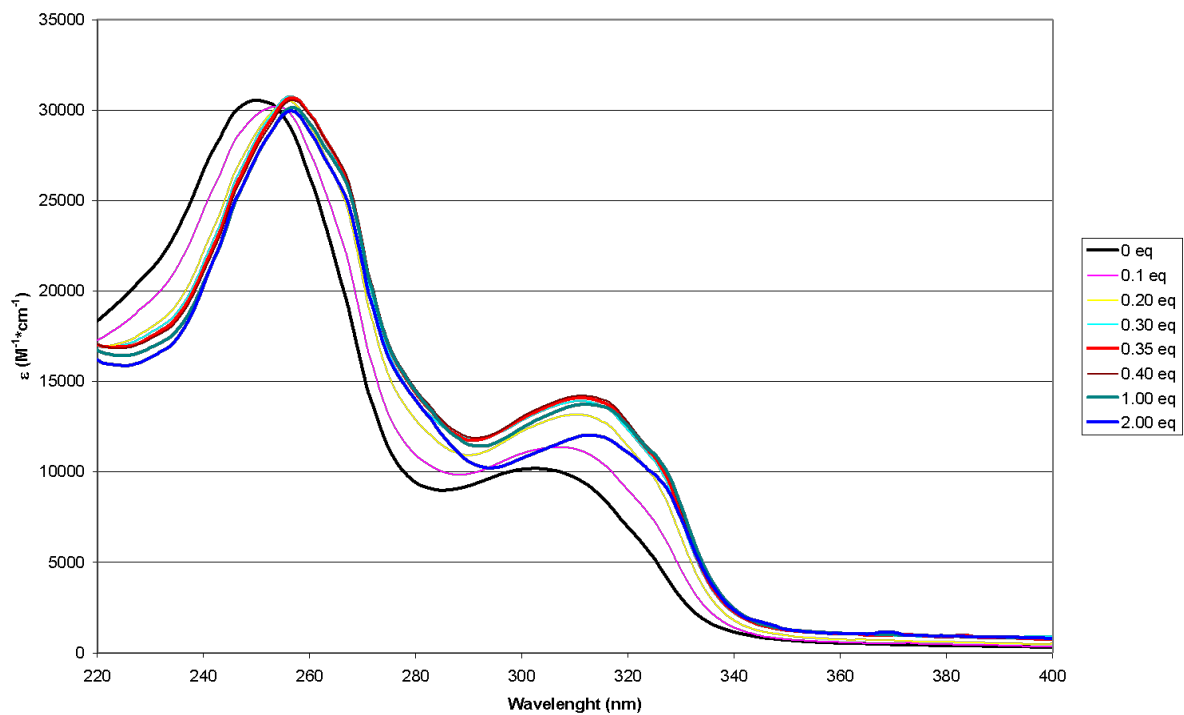


Figure S 11. Titration of L^{tz} ($2.5 \cdot 10^{-5} M$) with europium triflate, $I = 1 M$ (TMACl) in MeOH.

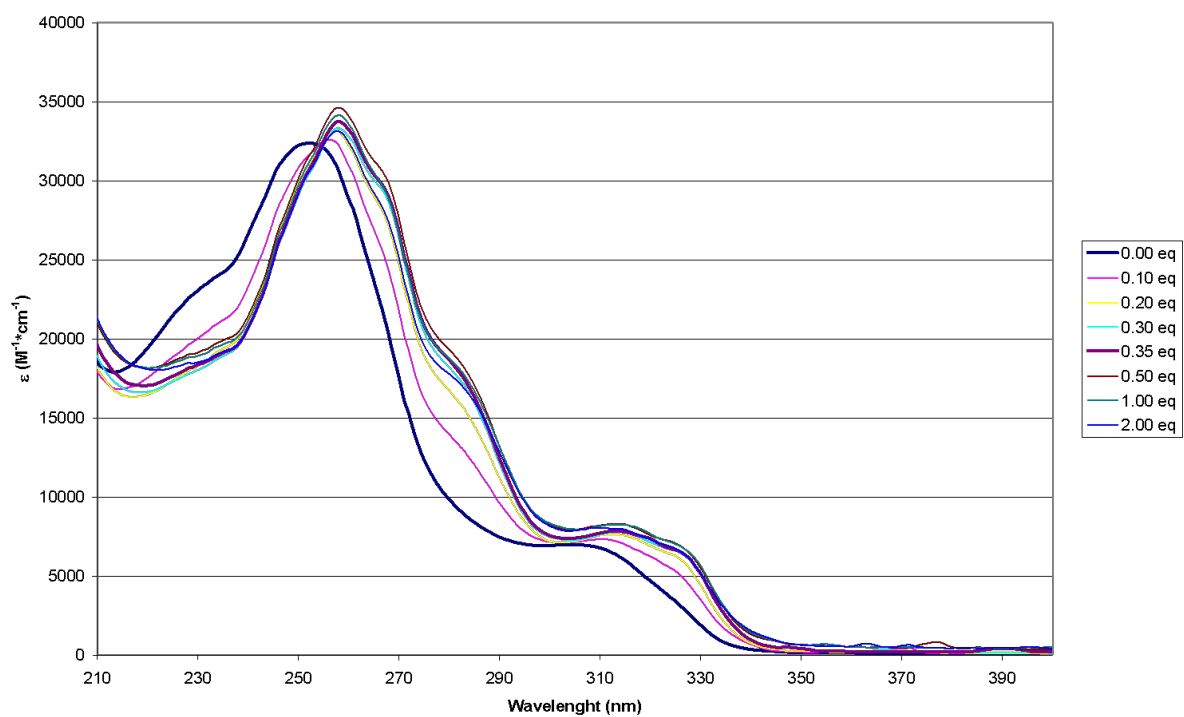


Figure S 12. Titration of L^{tzC8} ($2.5 \cdot 10^{-5} M$) with europium triflate, $I = 1 M$ (TMACl) in MeOH.

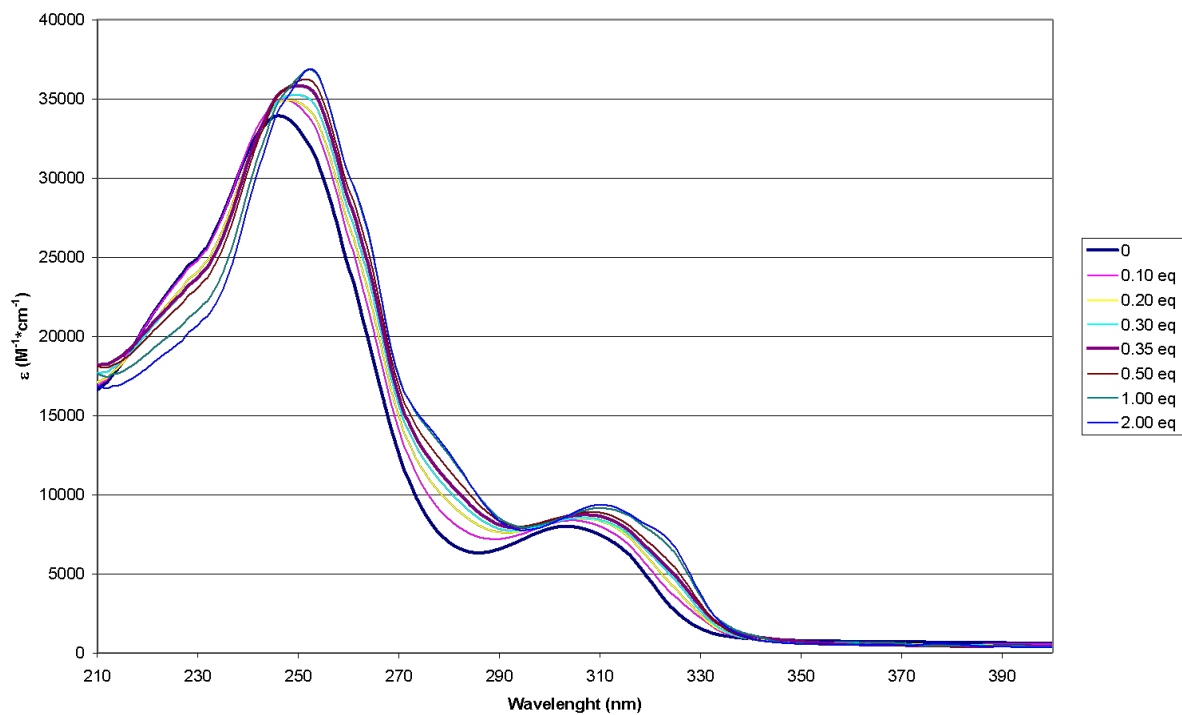


Figure S 13. Titration of L^{tz} ($2.5 \cdot 10^{-5} M$) with europium triflate in MES buffer.

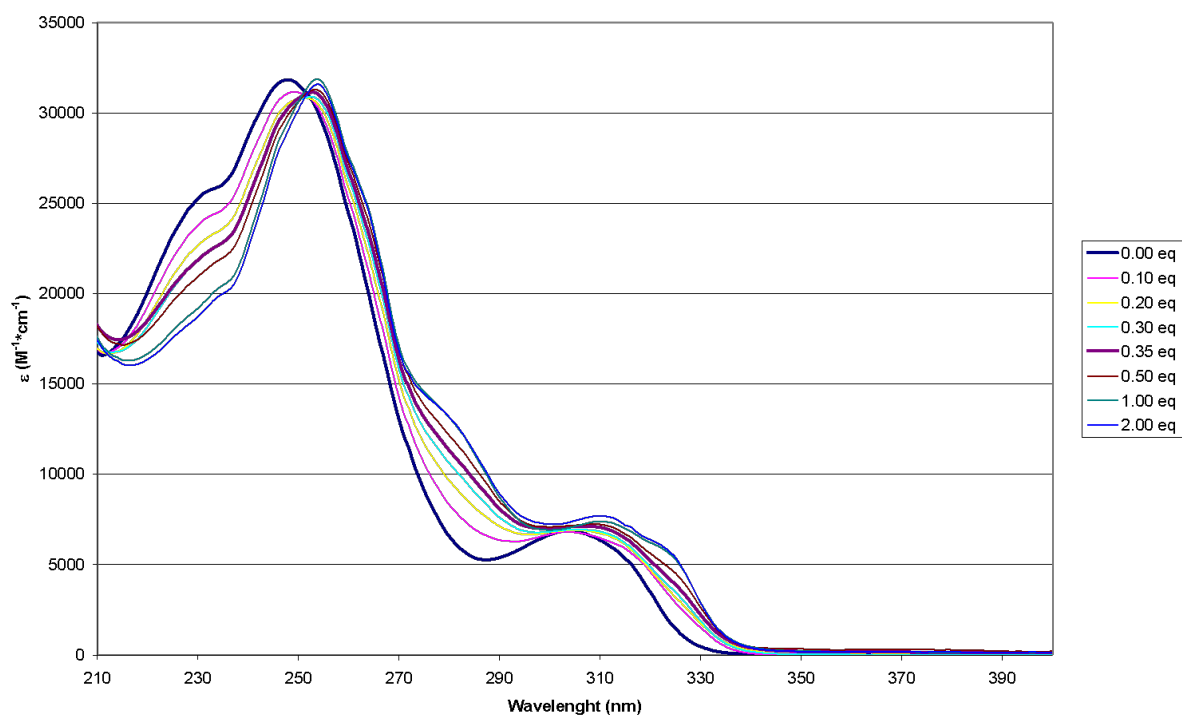


Figure S 14. Titration of L^{tzPEG} ($2.5 \cdot 10^{-5} M$) with europium triflate in water MES buffer.

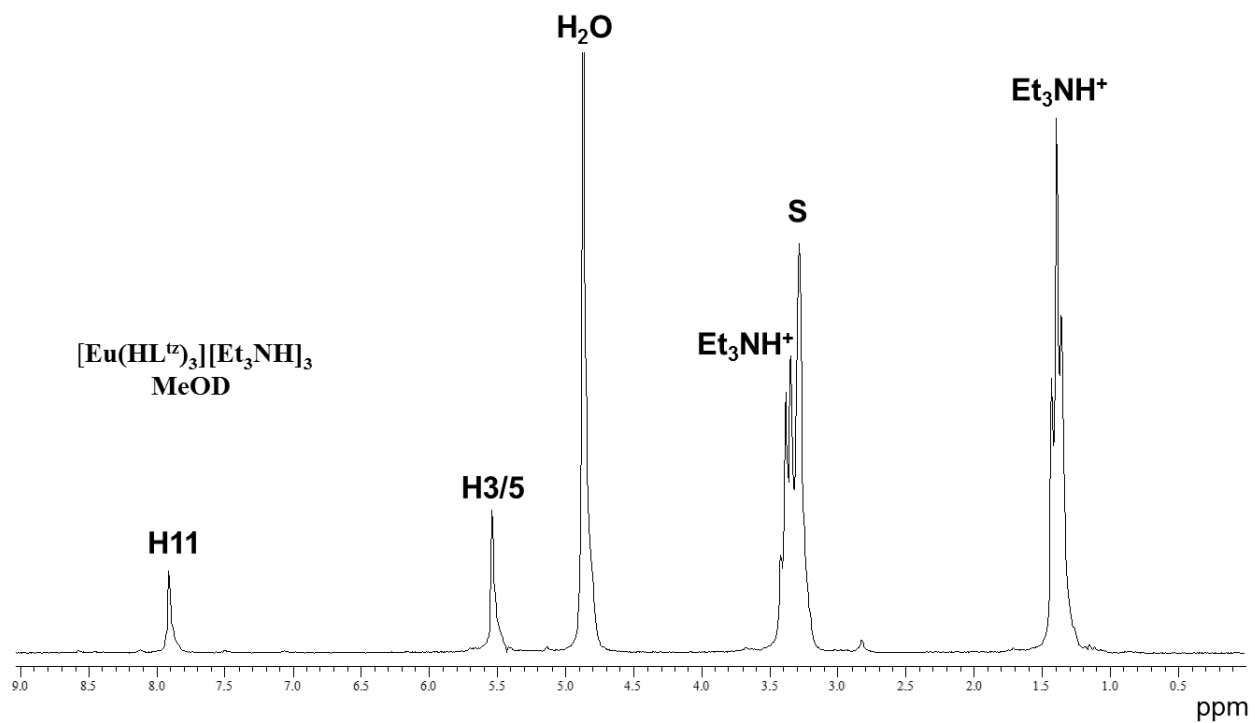


Figure S 15. ^1H NMR spectrum of $[\text{Eu}(\text{HL}^{\text{tz}})_3][\text{Et}_3\text{NH}]_3$ in CD_3OD (200 MHz).

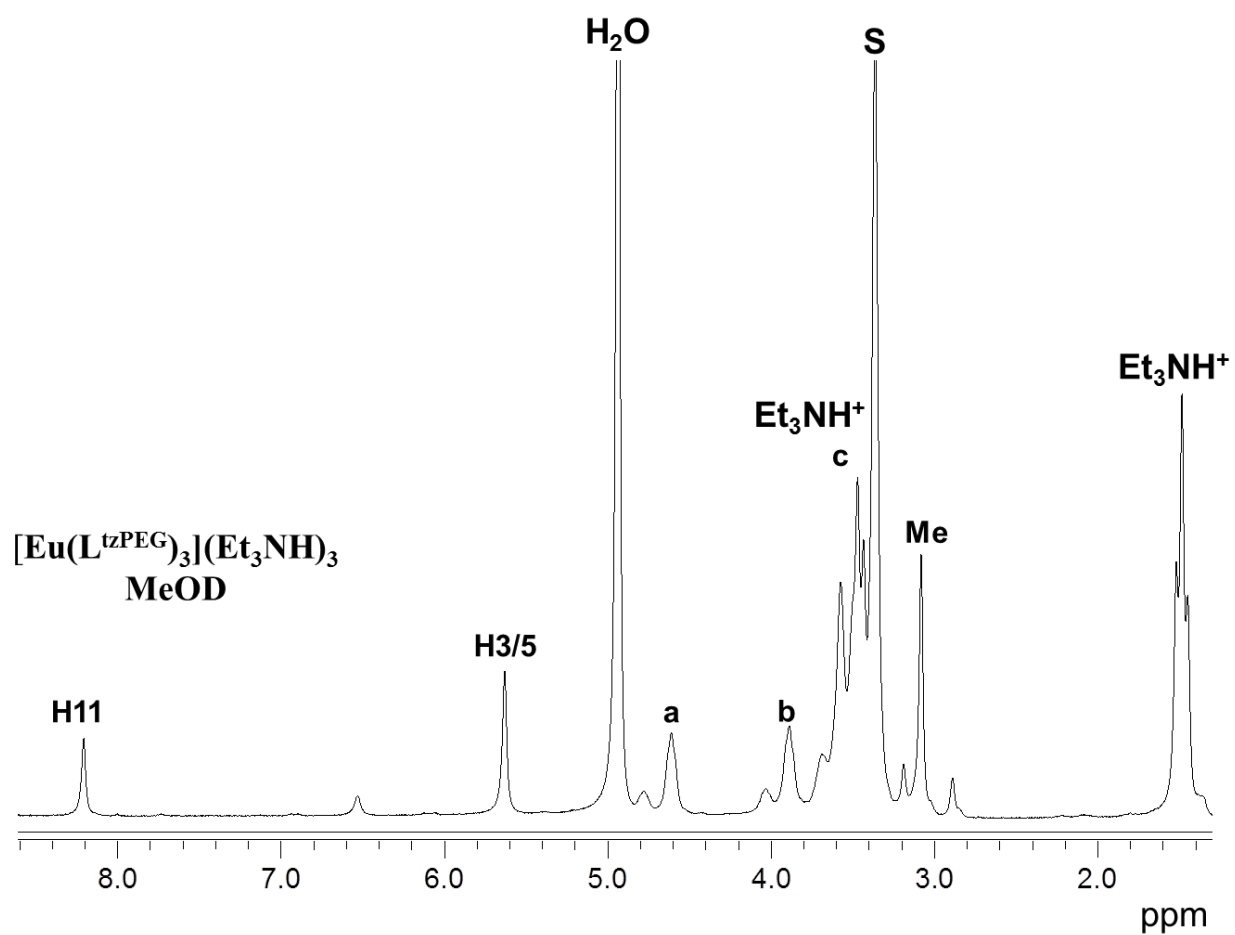


Figure S 16. ¹H NMR spectrum of [Eu(L^{tzPEG})₃][Et₃NH]₃ ¹H NMR spectrum in CD₃OD (200 MHz).

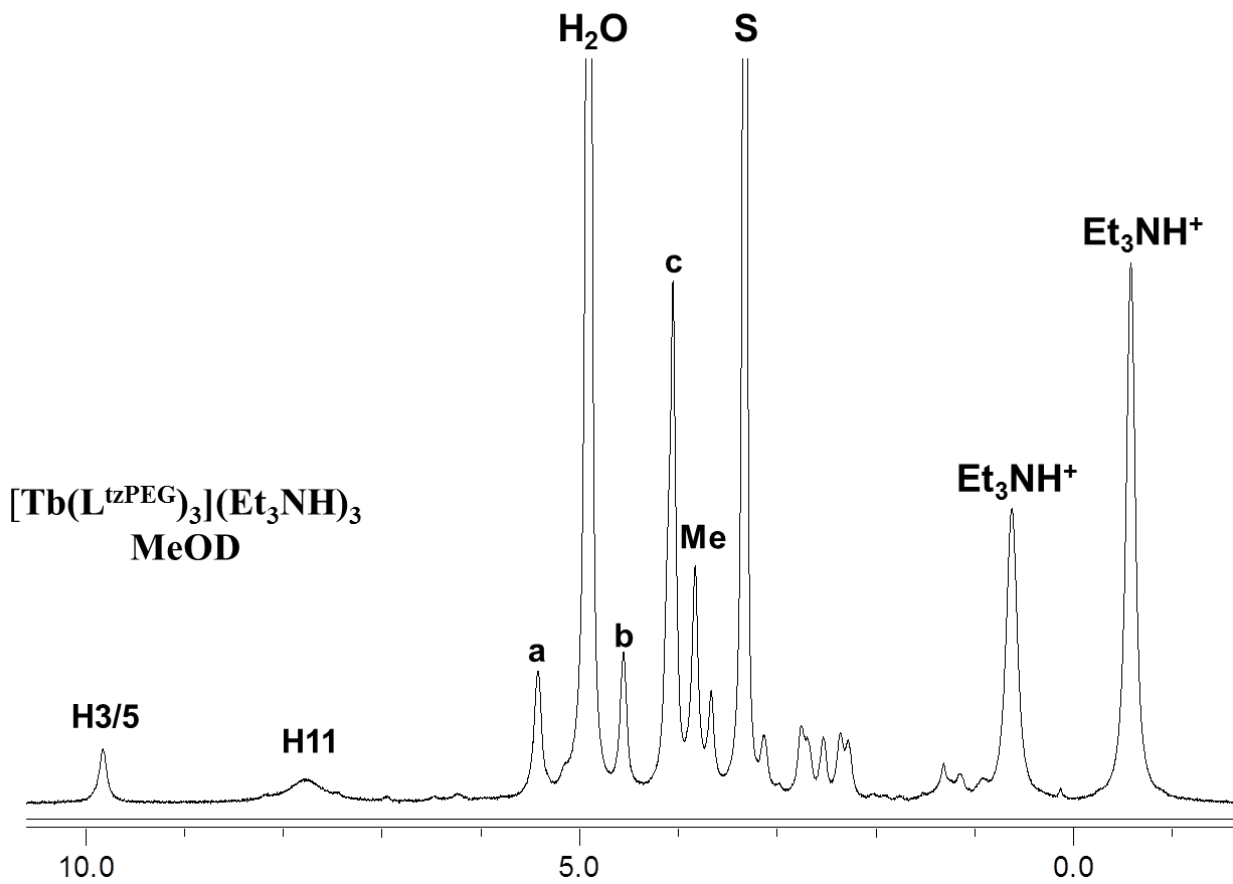


Figure S 17. ¹H NMR spectrum of [Tb(L^{tzPEG})₃](Et₃NH)₃ in CD₃OD (200 MHz).

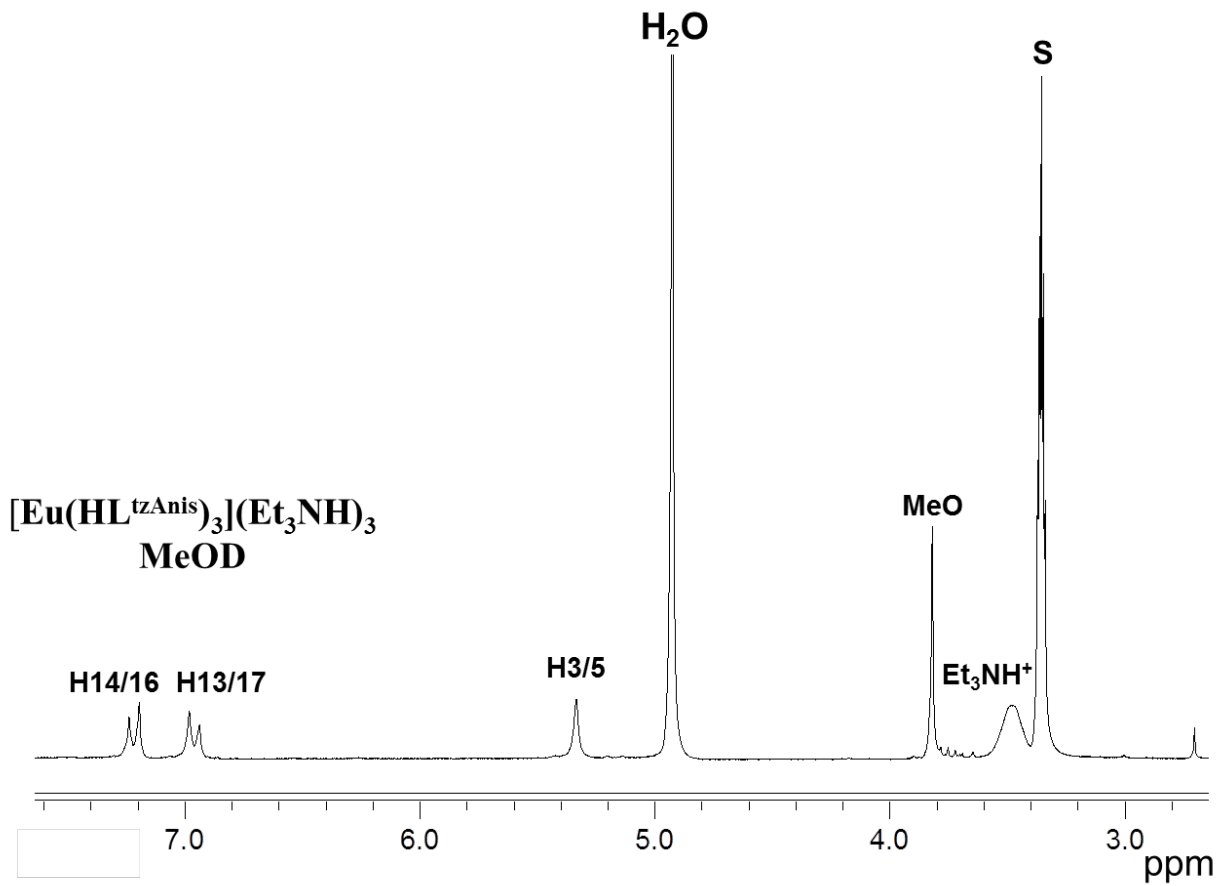


Figure S 18. ¹H NMR spectrum of [Eu(HL^{tzAnis})₃](Et₃NH)₃ in CD₃OD (200 MHz).

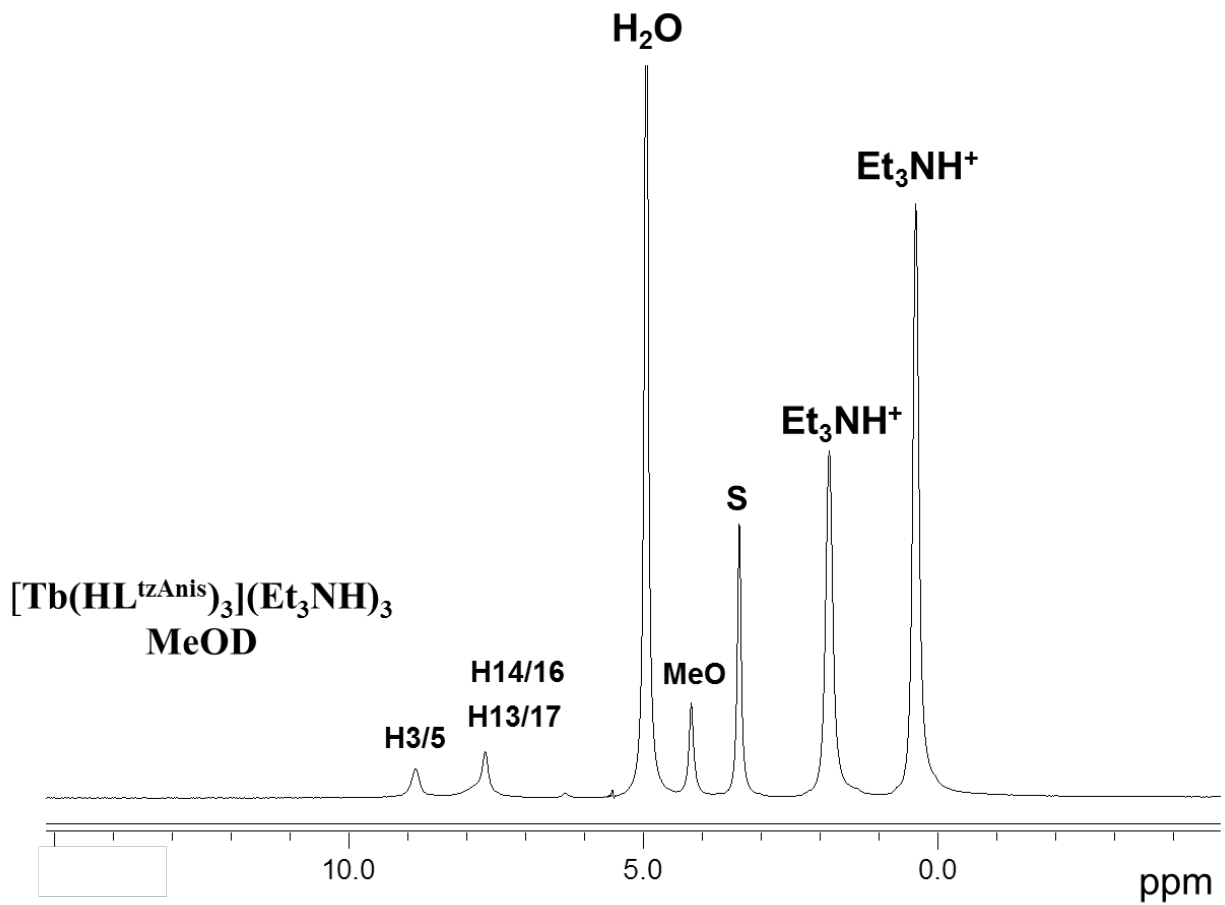


Figure S 19. ¹H NMR spectrum of [Tb(HL^{tzAnis})₃][Et₃NH]₃ in CD₃OD (200 MHz).

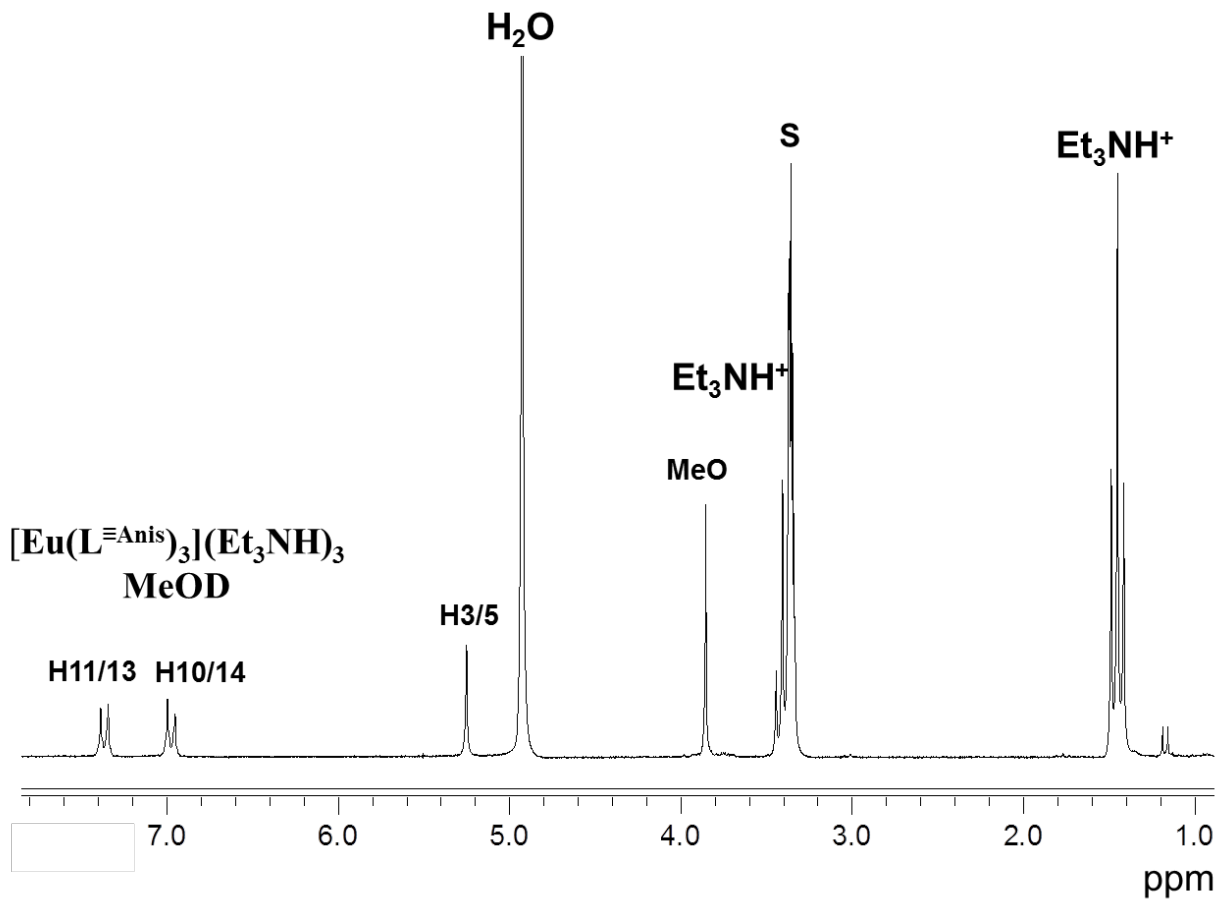


Figure S 20. ¹H NMR spectrum of [Eu(L[≡]Anis)₃](Et₃NH)₃ in CD₃OD (200 MHz).

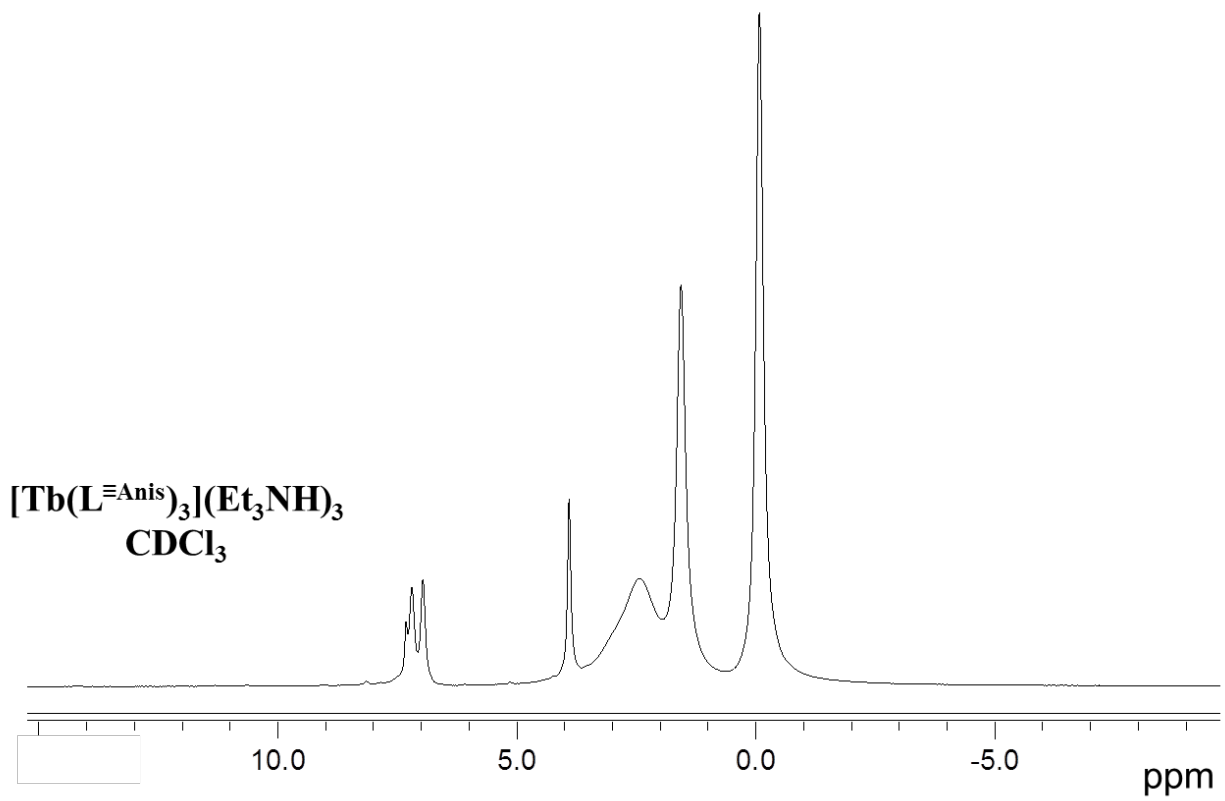


Figure S 21. ^1H NMR spectrum of $[\text{Tb}(\text{L}^{\equiv\text{Anis}})_3](\text{Et}_3\text{NH})_3$ in CD_3OD (200 MHz).

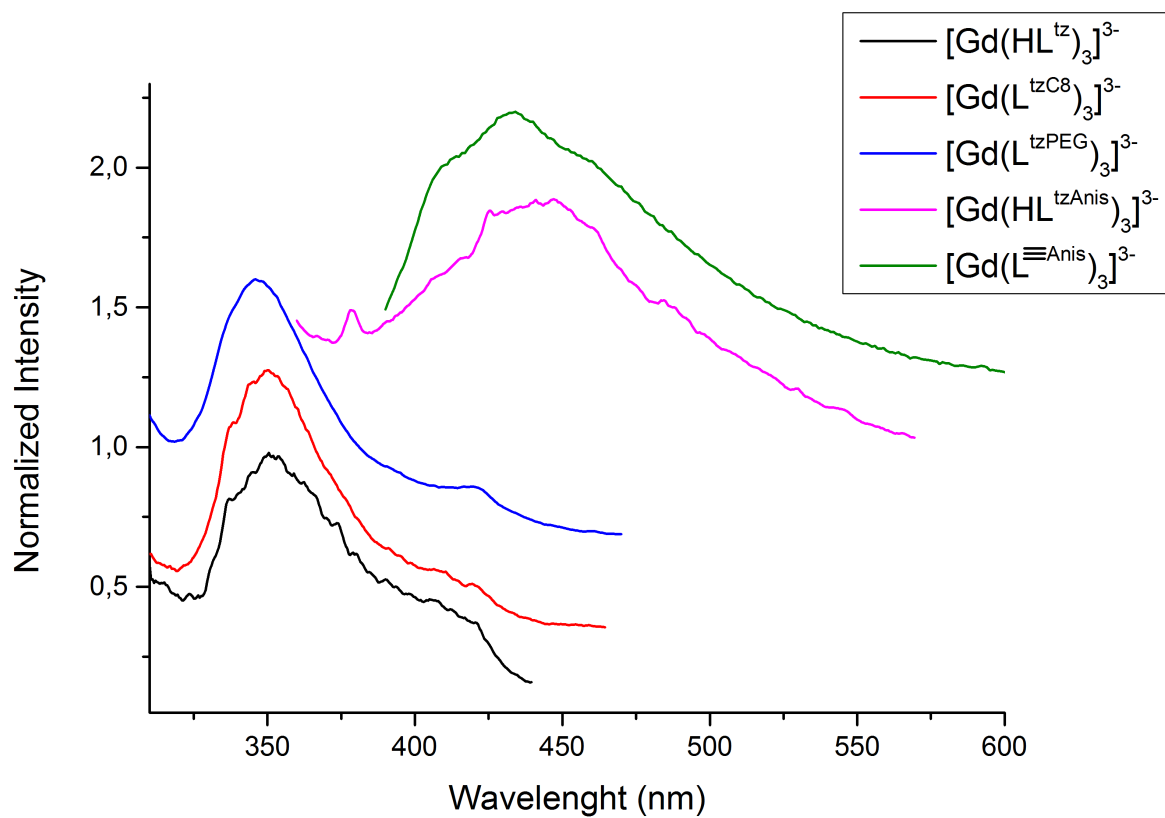


Figure S 22. Normalized emission spectra (singlet state) of the gadolinium complexes of all the ligands in MeOH.

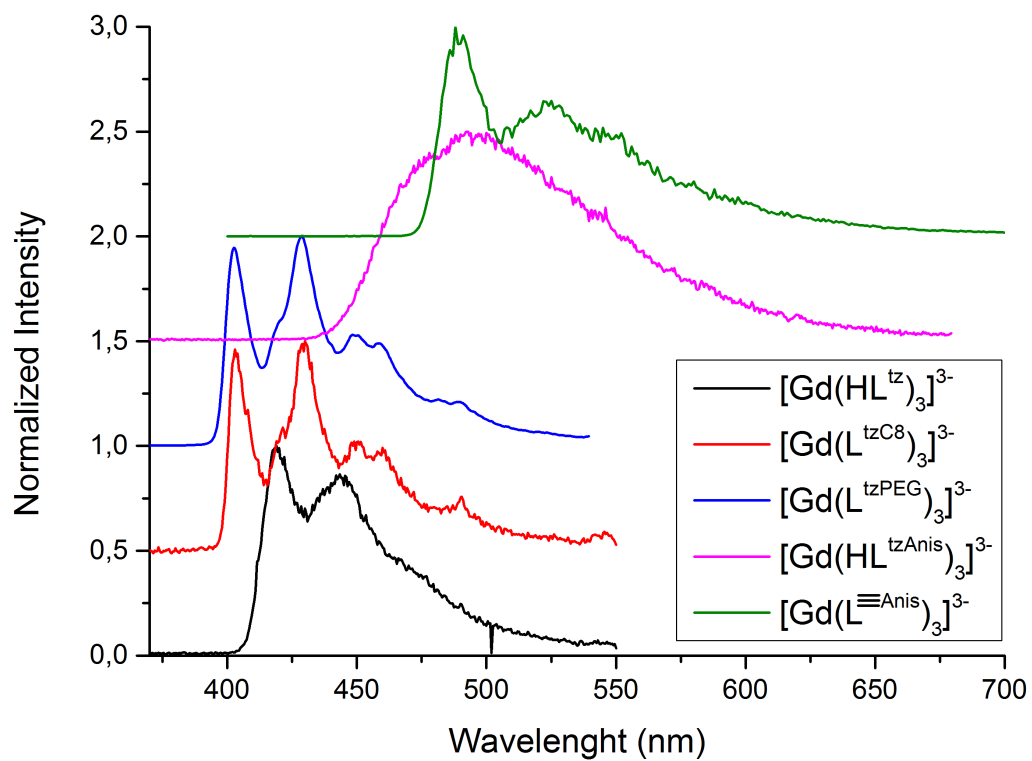


Figure S 23. Normalized emission spectra (triplet state) of the gadolinium complexes of all the ligands upon enforcement of a time delay (0.2 ms), in MeOH at 77 K

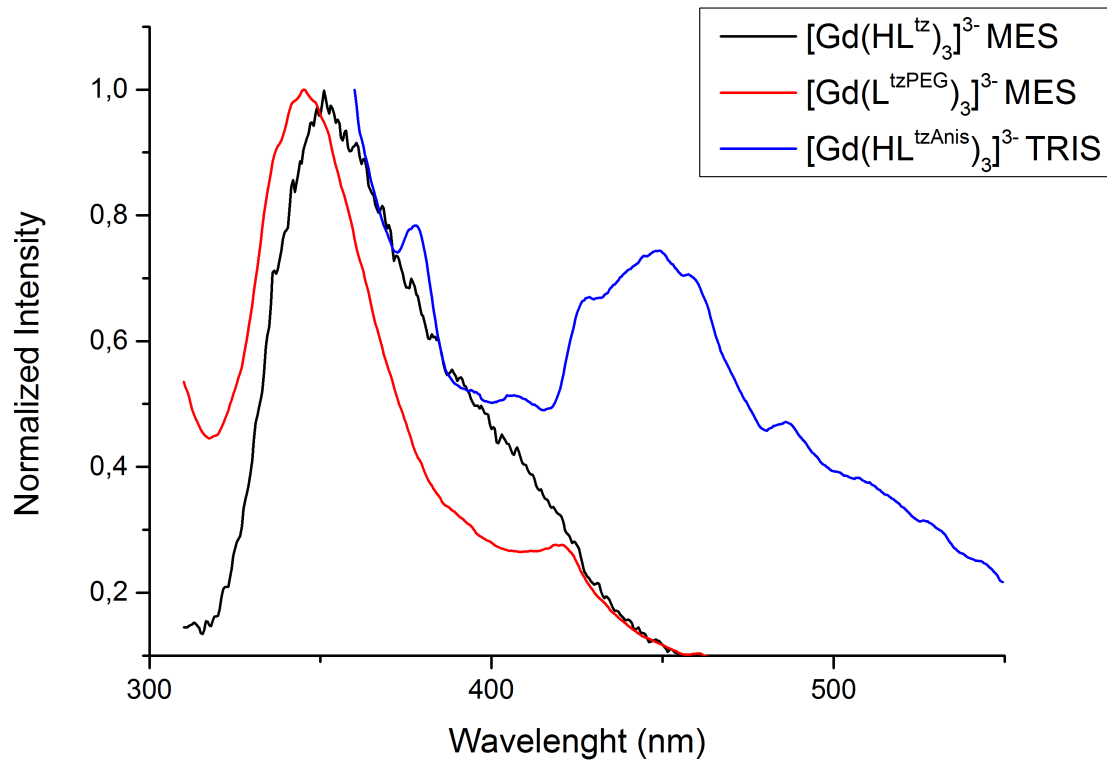


Figure S 24. Normalized emission spectra (singlet state) in water of the gadolinium complexes of L^{tz} , L^{tzPEG} and L^{tzAnis} .

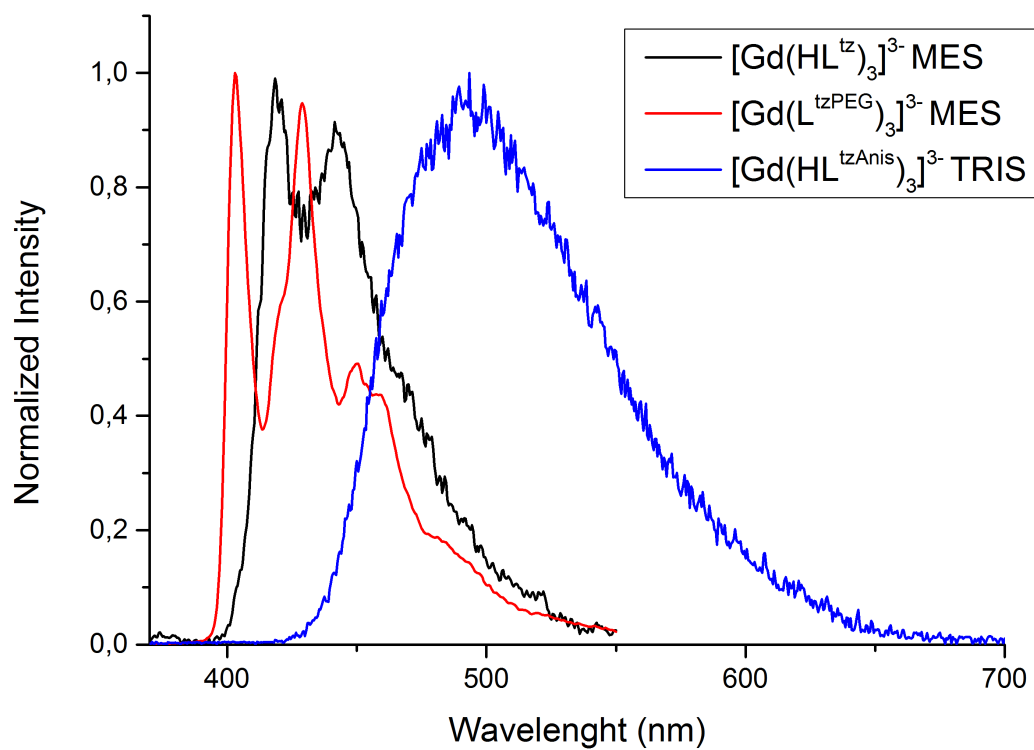


Figure S 25. Normalized emission spectra (triplet state) in water at 77K of the gadolinium complexes of L^{tz} , L^{tzPEG} and L^{tzAnis} upon enforcement of a time delay (0.2 ms).

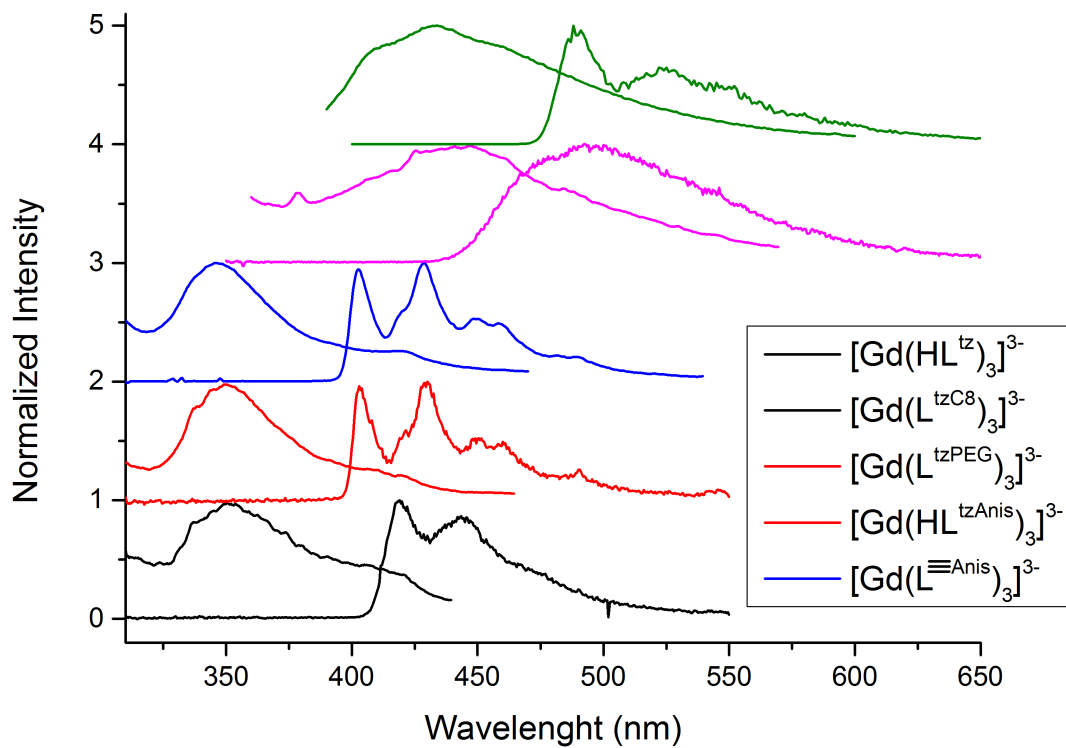


Figure S 26. Overall singlet/triplet state comparison for the gadolinium complexes of all the ligands in methanol.

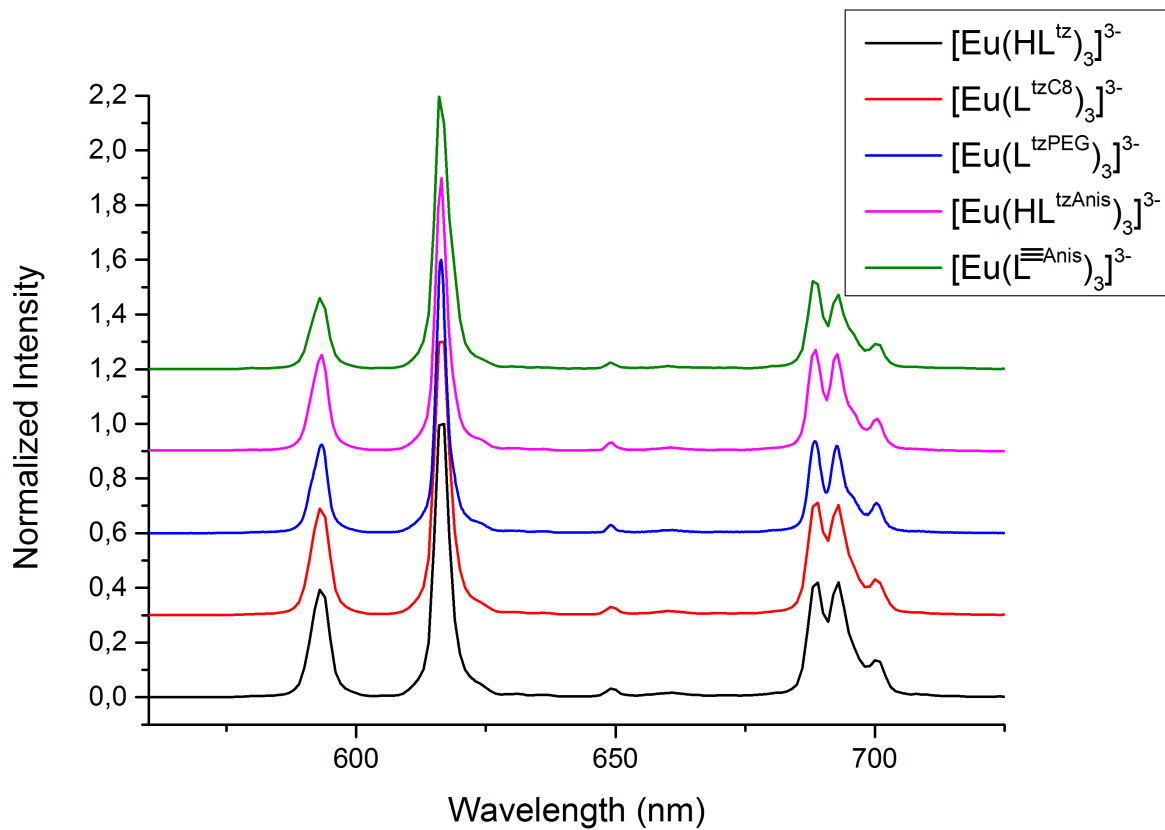


Figure S 27. Normalized emission spectra of the europium complexes of all the ligands in methanol.

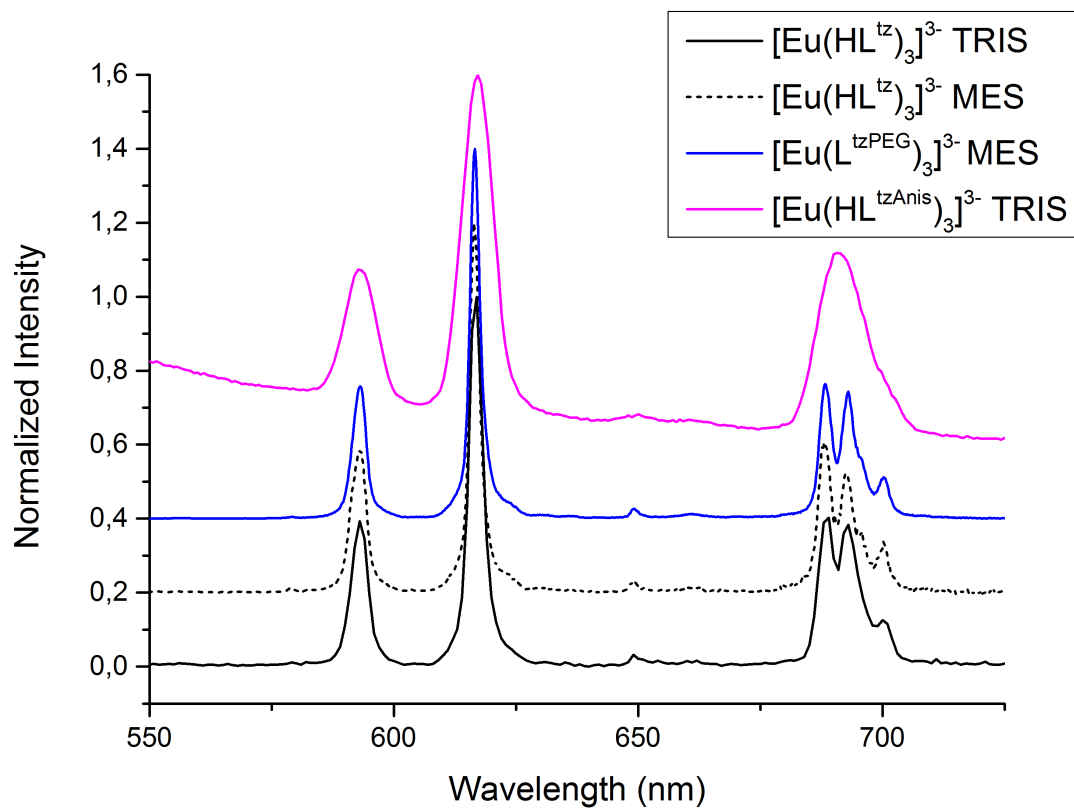


Figure S 28. Normalized emission spectra of europium complexes of L^{tz} , L^{tzPEG} and L^{tzAnis} in MES/TRIS buffer.

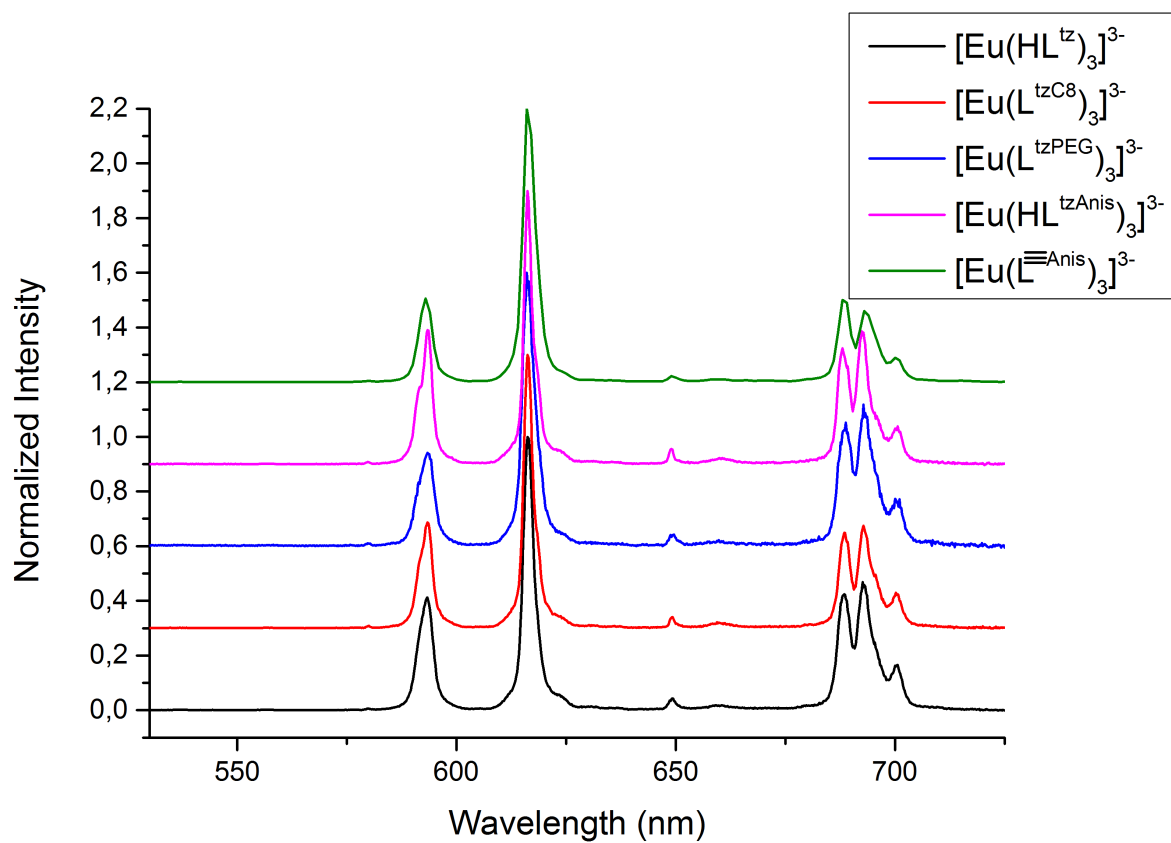


Figure S 29. Normalized emission spectra of the europium complexes of all the ligands in solid state.

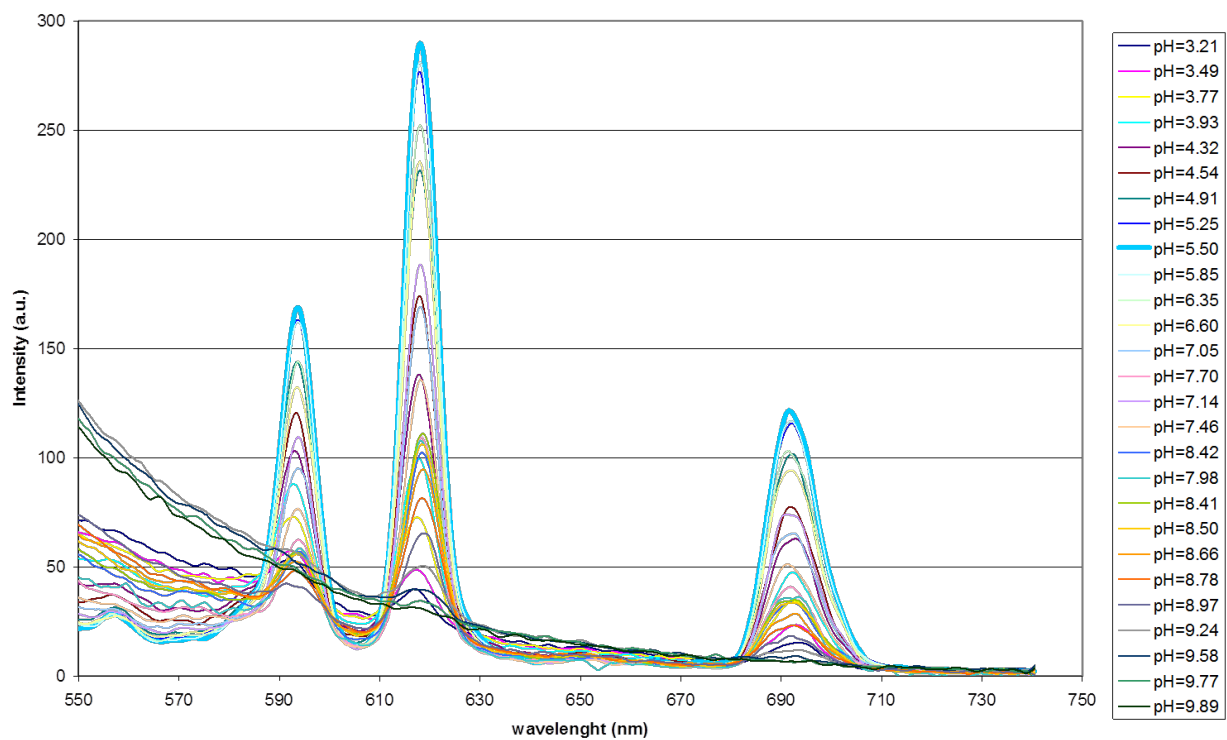


Figure S 30. Emission spectra of $[\text{Eu}(\text{HL}^{\text{tz}})]_3[\text{Et}_3\text{NH}]_3$ in water as a function of pH ($2.5 \cdot 10^{-5} \text{ M}$, $I = \text{KCl } 1 \text{ M}$).