

Electronic Supplementary Information

Mononuclear and polynuclear complexes ligated by an iminodiacetic acid derivative: synthesis, structure, solution studies and magnetic properties

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Table S1. Calculated spin configurations and their relative energies as a function of different J_i constants for model **Gd₄**. The spin configuration used as a reference is that with the maximum multiplicity that is generated from the parallel alignment of all local spin moments of the Gd(III) ions. Only the centers with an antiparallel (negative) alignment of their spin moment are noted.

Spin conf	S	J_a	J_b	J_c
{1,4}	0	56	0	56
{1,2}	0	0	28	56
{1,3}	0	56	28	0
{1}	7	28	0	28
{2}	7	28	28	28

Table S2. Hydrolysis constants of the Ln ions ($\text{Ln}^{3+} + \text{H}_2\text{O} \rightarrow [\text{Ln}(\text{OH})]^{2+} + \text{H}^+$, $*K_1$) and protonation constants of bzilda determined at 25.0 °C in 0.50 mol L⁻¹ Me₄NCl.

Protonation of bzilda		Hydrolysis of Ln		
Equilibrium	log K	Ln	log $*K_1$	σ
$\text{L}^{2-} + \text{H}^+ \rightarrow \text{HL}^-$	8.80(1)	Ce	-8.0(1)	0.3
$\text{L}^{2-} + 2 \text{H}^+ \rightarrow \text{H}_2\text{L}$	11.27(3)	Sm	-7.3(1)	0.7
$\text{L}^{2-} + 3 \text{H}^+ \rightarrow \text{H}_3\text{L}^+$	13.46(9)	Gd	-7.51(6)	0.1
σ	0.8	Yb	-7.4(4)	0.4

Table S3 Bond distances and angles in the metal coordination

Bond distances (Angstrom) with su's and angles (°) in the coordination sphere for [Nd₂(bzilda)₃].3H₂O

Nd1 - N1 2.8564(4)	Nd1 - O5 2.5060(4)	Nd2 - N2 2.6841(1)
Nd1 - O1 2.5433(3)	Nd1 - O11 2.6120(3)	Nd2 - N3 2.7603(1)
Nd1 - O3 2.4343(4)	Nd1 - O14 2.4982(3)	Nd2 - O1 2.5142(1)

Nd2 - O5	2.4997(1)	O3 - Nd1 - O14	136.62	O1 - Nd2 - O7	135.01
Nd2 - O7	2.4390(1)	O5 - Nd1 - O11	62.71	O1 - Nd2 - O9	142.91
Nd2 - O9	2.4238(1)	O5 - Nd1 - O14	78.97	O1 - Nd2 - O11	65.48
Nd2 - O11	2.5388(1)	O11 - Nd1 - O14	124.18	O1 - Nd2 - O13	80.00
Nd2 - O13	2.4798(1)	N2 - Nd2 - N3	123.12	O1 - Nd2 - O15	69.00
Nd2 - O15	2.5096(1)	N2 - Nd2 - O1	134.58	O5 - Nd2 - O7	104.72
Nd1 ...O3	2.4902(4)-x+1,-y+1,-z	N2 - Nd2 - O5	63.73	O5 - Nd2 - O9	132.32
Nd1 ..O8	2.4416(3) -x+3/2,+y+1/2,-z+1/2	N2 - Nd2 - O7	63.09	O5 - Nd2 - O11	63.85
Nd1 ...O12	2.4626(3) -x+1,-y+1,-z	N2 - Nd2 - O9	74.63	O5 - Nd2 - O13	141.59
N1 - Nd1 - O1	60.22	N2 - Nd2 - O11	91.70	O5 - Nd2 - O15	72.01
N1 - Nd1 - O3	63.80	N2 - Nd2 - O13	140.37	O7 - Nd2 - O9	73.70
N1 - Nd1 - O5	125.77	N2 - Nd2 - O15	98.14	O7 - Nd2 - O11	154.46
N1 - Nd1 - O11	109.34	N3 - Nd2 - O1	81.35	O7 - Nd2 - O13	78.59
N1 - Nd1 - O14	126.12	N3 - Nd2 - O5	125.00	O7 - Nd2 - O15	67.32
O1 - Nd1 - O3	84.63	N3 - Nd2 - O7	127.68	O9 - Nd2 - O11	96.77
O1 - Nd1 - O5	70.34	N3 - Nd2 - O9	61.68	O9 - Nd2 - O13	85.79
O1 - Nd1 - O11	64.01	N3 - Nd2 - O11	61.53	O9 - Nd2 - O15	138.90
O1 - Nd1 - O14	138.54	N3 - Nd2 - O13	72.55	O11 - Nd2 - O13	125.11
O3 - Nd1 - O5	133.64	N3 - Nd2 - O15	138.71	O11 - Nd2 - O15	124.13
O3 - Nd1 - O11	71.38	O1 - Nd2 - O5	70.92	O13 - Nd2 - O15	74.47

Bond distances (Angstrom) with su's and angles (°) in the coordination sphere for [Sm₂(bzlida)₃].3H₂O

Sm1 - N1	2.8247(4)	O1 - Sm1 - O3	85.40	N3 - Sm2 - O15	138.56
Sm1 - O1	2.5167(3)	O1 - Sm1 - O5	70.30	O1 - Sm2 - O5	70.70
Sm1 - O3	2.4130(4)	O1 - Sm1 - O11	63.97	O1 - Sm2 - O7	134.31
Sm1 - O5	2.4801(4)	O1 - Sm1 - O14	138.69	O1 - Sm2 - O9	143.19
Sm1 - O11	2.5933(3)	O3 - Sm1 - O5	133.92	O1 - Sm2 - O11	65.60
Sm1 - O14	2.4718(3)	O3 - Sm1 - O11	71.85	O1 - Sm2 - O13	80.28
Sm2 - N2	2.6585(1)	O3 - Sm1 - O14	135.74 1	O1 - Sm2 - O15	68.60
Sm2 - N3	2.7434(1)	O5 - Sm1 - O11	62.42	O5 - Sm2 - O7	105.33
Sm2 - O1	2.4903(1)	O5 - Sm1 - O14	78.94	O5 - Sm2 - O9	132.82
Sm2 - O5	2.4819(1)	O11 - Sm1 - O14	123.751	O5 - Sm2 - O11	63.63
Sm2 - O7	2.4242(1)	N2 - Sm2 - N3	123.57	O5 - Sm2 - O13	141.65
Sm2 - O9	2.3975(1)	N2 - Sm2 - O1	134.89	O5 - Sm2 - O15	71.36
Sm2 - O11	2.5077(1)	N2 - Sm2 - O5	64.22	O7 - Sm2 - O9	73.49
Sm2 - O13	2.4431(1)	N2 - Sm2 - O7	63.55	O7 - Sm2 - O11	155.22
Sm2 - O15	2.4905(1)	N2 - Sm2 - O9	74.55	O7 - Sm2 - O13	77.51
Sm1 ...O3	2.4709 (4)-x+1,-y+1,-z	N2 - Sm2 - O11	91.96	O7 - Sm2 - O15	67.32
Sm1 .O8	2.4161 (3) -x+3/2,+y+1/2,-z+1/2	N2 - Sm2 - O13	139.67	O9 - Sm2 - O11	97.36
Sm1 ...O12	2.4313 (3)-x+1,-y+1,-z	N2 - Sm2 - O15	97.86	O9 - Sm2 - O13	85.18
N1 - Sm1 - O1	60.83	N3 - Sm2 - O1	81.19	O9 - Sm2 - O15	138.93
N1 - Sm1 - O3	64.53	N3 - Sm2 - O5	124.98	O11 - Sm2 - O13	125.48
N1 - Sm1 - O5	126.14	N3 - Sm2 - O7	127.47	O11 - Sm2 - O15	123.48
N1 - Sm1 - O11	110.14	N3 - Sm2 - O9	62.16	O13 - Sm2 - O15	75.05
N1 - Sm1 - O14	125.77	N3 - Sm2 - O11	61.81		
		N3 - Sm2 - O13	72.35		

Bond distances (Angstrom) with su's and angles (°) in the coordination sphere for [Eu₂(bzlida)₃].3H₂O

Eu1 - N1	2.8091(4)	Eu2 - Eu1 - N1	99.64	O3 - Eu1 - O14	135.32
Eu1 - O1	2.5007(3)	Eu2 - Eu1 - O1	38.70	O5 - Eu1 - O11	62.30
Eu1 - O3	2.3983(4)	Eu2 - Eu1 - O3	100.30	O5 - Eu1 - O14	78.77
Eu1 - O5	2.4659(4)	Eu2 - Eu1 - O5	38.42	O11 - Eu1 - O14	123.34
Eu1 - O11	2.5823(3)	Eu2 - Eu1 - O11	39.51	Eu1 - Eu2 - N2	98.33
Eu1 - O14	2.4578(3)	Eu2 - Eu1 - O14	117.19	Eu1 - Eu2 - N3	92.05
Eu2 - N2	2.6497(2)	N1 - Eu1 - O1	61.02	Eu1 - Eu2 - O1	39.13
Eu2 - N3	2.7377(1)	N1 - Eu1 - O3	65.05	Eu1 - Eu2 - O5	38.29
Eu2 - O1	2.4773(1)	N1 - Eu1 - O5	126.22	Eu1 - Eu2 - O7	140.60
Eu2 - O5	2.4730(2)	N1 - Eu1 - O11	110.54	Eu1 - Eu2 - O9	138.38
Eu2 - O7	2.4110(1)	N1 - Eu1 - O14	125.77	Eu1 - Eu2 - O11	41.14
Eu2 - O9	2.3870(2)	O1 - Eu1 - O3	85.90	Eu1 - Eu2 - O13	119.60
Eu2 - O11	2.4973(1)	O1 - Eu1 - O5	70.28	Eu1 - Eu2 - O15	82.24
Eu2 - O13	2.4146(1)	O1 - Eu1 - O11	64.07	N2 - Eu2 - N3	123.62
Eu2 - O15	2.4732(1)	O1 - Eu1 - O14	138.66	N2 - Eu2 - O1	134.98
Eu1 ...O3	2.4553(4)-x+1,-y+1,-z	O3 - Eu1 - O5	134.03	N2 - Eu2 - O5	64.47
Eu1 .O8	2.4016 (3) -x+3/2,+y+1/2,-z+1/2	O3 - Eu1 - O11	72.04	N2 - Eu2 - O7	63.85
Eu1 ...O12	2.4189 (3)-x+1,-y+1,-z			N2 - Eu2 - O9	74.48

N2 - Eu2 - O11	91.94	O1 - Eu2 - O7	133.99	O7 - Eu2 - O11	155.52
N2 - Eu2 - O13	139.35	O1 - Eu2 - O9	143.24	O7 - Eu2 - O13	76.93
N2 - Eu2 - O15	97.74	O1 - Eu2 - O11	65.66	O7 - Eu2 - O15	67.09
N3 - Eu2 - O1	81.24	O1 - Eu2 - O13	80.50	O9 - Eu2 - O11	97.46
N3 - Eu2 - O5	125.00	O1 - Eu2 - O15	68.61	O9 - Eu2 - O13	84.99
N3 - Eu2 - O7	127.29	O5 - Eu2 - O7	105.65	O9 - Eu2 - O15	138.91
N3 - Eu2 - O9	62.16	O5 - Eu2 - O9	132.95	O11 - Eu2 - O13	125.79
N3 - Eu2 - O11	62.05	O5 - Eu2 - O11	63.43	O11 - Eu2 - O15	123.38
N3 - Eu2 - O13	72.31	O5 - Eu2 - O13	141.70	O13 - Eu2 - O15	75.21
N3 - Eu2 - O15	138.62	O5 - Eu2 - O15	71.23		
O1 - Eu2 - O5	70.55	O7 - Eu2 - O9	73.58		

Bond distances (Angstrom) with su's and angles (°) in the coordination sphere for [Gd₂(bzldia)₃].3H₂O

Gd1 - N2	2.7968(2)	O2 - Gd1 - O3	72.25	N3 - Gd2 - O10	64.78
Gd1 - O2	2.5688(3)	O2 - Gd1 - O7	63.93	N3 - Gd2 - O13	64.09
Gd1 - O3	2.3993(2)	O2 - Gd1 - O10	62.28	O1 - Gd2 - O2	97.55
Gd1 - O7	2.4924(2)	O2 - Gd1 - O15	123.46	O1 - Gd2 - O4	85.35
Gd1 - O10	2.4555(2)	O3 - Gd1 - O7	86.25	O1 - Gd2 - O7	143.19
Gd1 - O15	2.4540(2)	O3 - Gd1 - O10	134.25	O1 - Gd2 - O9	138.97
Gd2 - N1	2.7350(2)	O3 - Gd1 - O15	134.90	O1 - Gd2 - O10	132.92
Gd2 - N3	2.6383(2)	O7 - Gd1 - O10	70.12	O1 - Gd2 - O13	73.80
Gd2 - O1	2.3783(2)	O7 - Gd1 - O15	138.72	O2 - Gd2 - O4	125.85
Gd2 - O2	2.4922(2)	O10 - Gd1 - O15	78.94	O2 - Gd2 - O7	65.52
Gd2 - O4	2.4006(2)	N1 - Gd2 - N3	123.58	O2 - Gd2 - O9	123.20
Gd2 - O7	2.4606(2)	N1 - Gd2 - O1	62.36	O2 - Gd2 - O10	63.32
Gd2 - O9	2.4700(2)	N1 - Gd2 - O2	62.00	O2 - Gd2 - O13	155.97
Gd2 - O10	2.4607(2)	N1 - Gd2 - O4	72.58	O4 - Gd2 - O7	80.28
Gd2 - O13	2.4029(2)	N1 - Gd2 - O7	80.99	O4 - Gd2 - O9	75.09
Gd1 ...O3	2.4446 (4) -x+1,-y+1,-z	N1 - Gd2 - O9	138.70	O4 - Gd2 - O10	141.38
Gd1 ...O5	2.4061 (3) -x+1,-y+1,-z	N1 - Gd2 - O10	124.82	O4 - Gd2 - O13	76.58
Gd1 ...O8	2.3933 (4) -x+3/2,+y-1/2,-z+1/2	N1 - Gd2 - O13	127.45	O7 - Gd2 - O9	68.74
N2 - Gd1 - O2	110.76	N3 - Gd2 - O1	74.17	O7 - Gd2 - O10	70.56
N2 - Gd1 - O3	65.14	N3 - Gd2 - O2	92.10	O7 - Gd2 - O13	133.69
N2 - Gd1 - O7	61.43	N3 - Gd2 - O4	139.22	O9 - Gd2 - O10	71.09
N2 - Gd1 - O10	126.40	N3 - Gd2 - O7	135.30	O9 - Gd2 - O13	66.79
N2 - Gd1 - O15	125.46	N3 - Gd2 - O9	97.70	O10 - Gd2 - O13	105.81

Bond distances (Angstrom) with su's and angles (°) in the coordination sphere for [La(Hbzldia)(bzldia)].H₂O

La - N1	2.8800(4)	La ...O8	2.4148 (4) x-1/2,-y+1/2,+z	O2 - La - O3	100.97
La - O2	2.4654(3)	La ...O10	2.6158 (5) x+1/2,-y+1/2,+z	O2 - La - O5	149.67
La - O3	2.6189(3)			O2 - La - O9	72.17
La - O5	2.5100(3)	N1 - La - O2	61.65	O3 - La - O5	73.25
La - O9	2.6342(3)	N1 - La - O3	59.15	O3 - La - O9	132.07
La ...O3	2.6951 (3) x+1/2,-y+1/2,+z	N1 - La - O5	129.82	O5 - La - O9	134.02
La ...O6	2.4632 (4) x+1/2,-y+1/2,+z	N1 - La - O9	77.52		

Bond distances (Angstrom) with su's and angles (°) with su's in the coordination sphere for [Cu(Hbzldia)₂].4H₂O

Cu1 - O2	2.426(1)	O2 - Cu1 - O3	95.34	0.02
Cu1 - O3	1.9529(9)	O2 - Cu1 - N1	75.60	0.02
Cu1 - N1	2.0236(9)	O3 - Cu1 - N1	86.16	0.02

Bond distances (Angstrom) with su's and angles (°) with su's in the coordination sphere for [Cu(bzldia)₂]{Er(AcO)(H₂O)₅}₂] [Cu(bzldia)₂].6H₂O

Er1 - O1	2.3339(9)	Cu1 - O7	1.9726(6)	O1 - Er1 - O9	114.64	0.03
Er1 - O9	2.4475(14)	Cu2 - N1	2.0444(8)	O1 - Er1 - O10	75.62	0.02
Er1 - O10	2.3819(10)	Cu2 - O2	2.3549(8)	O1 - Er1 - O11	74.72	0.03
Er1 - O11	2.3327(11)	Cu2 - O3	1.9951(7)	O1 - Er1 - O12	141.38	0.03
Er1 - O12	2.3274(9)	Cu1 ...N2	2.0645 (7) -x+1,-y+1,-z	O1 - Er1 - O13	141.08	0.04
Er1 - O13	2.3822(12)	Cu1 ...O6	2.4730 (8) -x+1,-y+1,-z	O1 - Er1 - O14	74.49	0.03
Er1 - O14	2.2878(10)	Cu1 ...O7	1.9726 (6) -x+1,-y+1,-z	O1 - Er1 - O15	82.26	0.02
Er1 - O15	2.3508(10)	Cu2 ...N1	2.0444 (8) -x+2,-y+1,-z	O9 - Er1 - O10	54.17	0.02
Er1 - C100	2.7924(11)	Cu2 ...O2	2.3549 (8) -x+2,-y+1,-z	O9 - Er1 - O11	128.78	0.03
Cu1 - N2	2.0645(7)	Cu2 ...O3	1.9951 (7) -x+2,-y+1,-z	O9 - Er1 - O12	82.97	0.02
Cu1 - O6	2.4730(8)			O9 - Er1 - O13	75.12	0.02

O9 - Er1 - O14	78.27	0.03
O9 - Er1 - O15	153.79	0.03
O10 - Er1 - O11	83.97	0.02
O10 - Er1 - O12	135.65	0.03
O10 - Er1 - O13	82.71	0.03
O10 - Er1 - O14	102.33	0.03
O10 - Er1 - O15	151.85	0.03
O11 - Er1 - O12	122.20	0.03

O11 - Er1 - O13	71.13	0.03
O11 - Er1 - O14	145.82	0.04
O11 - Er1 - O15	73.45	0.02
O12 - Er1 - O13	74.86	0.03
O12 - Er1 - O14	76.27	0.03
O12 - Er1 - O15	72.04	0.02
O13 - Er1 - O14	142.63	0.04
O13 - Er1 - O15	104.91	0.03

O14 - Er1 - O15	88.03	0.03
N2 - Cu1 - O6	75.84	0.02
N2 - Cu1 - O7	85.70	0.02
O6 - Cu1 - O7	94.73	0.03
N1 - Cu2 - O2	74.85	0.02
N1 - Cu2 - O3	84.98	0.02
O2 - Cu2 - O3	93.65	0.03

Table S4 Hydrogen bond geometry

(**) Values normalized following G.A.Jeffrey & L.Lewis, Carbohydr.Res.
(1978).60,179; R.Taylor, O.Kennard, Acta Cryst.(1983).B39,133.

Possible hydrogen bonds for Nd2(bzlida)3.3H2O

Donor-H	Donor...Acceptor	H...Acceptor	Donor-H.....Acceptor
O13 -H13B 0.867(.000) 0.938	O13 ...O2 (0) 2.677(.000)	H13B ...O2 (0) 1.861(.000) 1.797	O13 -H13B ...O2 (0) 156.03(0.00) 155.11 (**)
O14 -H14B 0.865(.000) 0.938	O14 ...O6 (0) 2.740(.000)	H14B ...O6 (0) 1.908(.000) 1.839	O14 -H14B ...O6 (0) 161.10(0.01) 160.36 (**)
O15 -H15B 0.869(.000) 0.938	O15 ...O2 (0) 2.948(.000)	H15B ...O2 (0) 2.638(.000) 2.624	O15 -H15B ...O2 (0) 102.29(0.00) 100.81 (**)
O15 -H15A 0.864(.000) 0.938	O15 ...O10 (1) 2.650(.000)	H15A ...O10 (1) 1.788(.000) 1.714	O15 -H15A ...O10 (1) 175.31(0.01) 175.11 (**)
O14 -H14B 0.865(.000) 0.938	O14 ...O4 (2) 2.924(.000)	H14B ...O4 (2) 2.649(.000) 2.638	O14 -H14B ...O4 (2) 99.83(0.00) 98.27 (**)
O14 -H14A 0.884(.000) 0.938	O14 ...O10 (5) 2.907(.000)	H14A ...O10 (5) 2.041(.000) 1.989	O14 -H14A ...O10 (5) 166.01(0.00) 165.63 (**)
O13 -H13A 0.893(.000) 0.938	O13 ...O6 (6) 2.700(.000)	H13A ...O6 (6) 1.828(.000) 1.785	O13 -H13A ...O6 (6) 164.51(0.00) 164.13 (**)

Equivalent positions:

- (0) x, y, z
- (1) -x+1/2+1, +y+1/2, -z+1/2
- (2) -x+1, -y+1, -z
- (5) x, +y+1, +z
- (6) -x+1/2+1, +y-1/2, -z+1/2

Possible hydrogen bonds for Sm2(bzlida)3.3H2O

Donor-H	Donor...Acceptor	H...Acceptor	Donor-H.....Acceptor
O13 -H13B 0.892(.000) 0.938	O13 ...O2 (0) 2.655(.000)	H13B ...O2 (0) 1.819(.000) 1.778	O13 -H13B ...O2 (0) 155.05(0.01) 154.43 (**)
O14 -H14B 0.897(.000) 0.938	O14 ...O6 (0) 2.728(.000)	H14B ...O6 (0) 1.883(.000) 1.846	O14 -H14B ...O6 (0) 156.16(0.01) 155.65 (**)
O15 -H15A 0.882(.000) 0.938	O15 ...O10 (1) 2.651(.000)	H15A ...O10 (1) 1.787(.000) 1.732	O15 -H15A ...O10 (1) 166.21(0.01) 165.77 (**)
O14 -H14A 0.903(.000) 0.938	O14 ...O10 (5) 2.924(.000)	H14A ...O10 (5) 2.049(.000) 2.016	O14 -H14A ...O10 (5) 162.86(0.01) 162.57 (**)
O13 -H13A 0.897(.000) 0.938	O13 ...O6 (6) 2.698(.000)	H13A ...O6 (6) 1.809(.000) 1.769	O13 -H13A ...O6 (6) 170.23(0.01) 170.00 (**)

Equivalent positions:

- (0) x, y, z
- (1) -x+1/2+1, +y+1/2, -z+1/2
- (5) x, +y+1, +z
- (6) -x+1/2+1, +y-1/2, -z+1/2

Possible hydrogen bonds for Eu2(bzlida)3.3H2O

Donor-H	Donor...Acceptor	H...Acceptor	Donor-H.....Acceptor
O13 -H13B 0.873(.000) 0.938	O13 ...O2 (0) 2.647(.000)	H13B ...O2 (0) 1.814(.000) 1.754	O13 -H13B ...O2 (0) 158.76(0.01) 157.99 (**)
O14 -H14B 0.906(.000) 0.938	O14 ...O6 (0) 2.715(.000)	H14B ...O6 (0) 1.880(.000) 1.851	O14 -H14B ...O6 (0) 152.41(0.01) 151.95 (**)
O15 -H15A 0.892(.000) 0.938	O15 ...O10 (1) 2.649(.000)	H15A ...O10 (1) 1.762(.000) 1.716	O15 -H15A ...O10 (1) 172.68(0.01) 172.49 (**)
O14 -H14A 0.898(.000) 0.938	O14 ...O10 (5) 2.944(.000)	H14A ...O10 (5) 2.073(.000) 2.035	O14 -H14A ...O10 (5) 163.05(0.01) 162.72 (**)
O13 -H13A 0.898(.000) 0.938	O13 ...O6 (6) 2.697(.000)	H13A ...O6 (6) 1.812(.000) 1.773	O13 -H13A ...O6 (6) 168.33(0.01) 168.07 (**)

Equivalent positions:

- (0) x, y, z
 (1) -x+1/2+1, +y+1/2, -z+1/2
 (5) x, +y+1, +z
 (6) -x+1/2+1, +y-1/2, -z+1/2

Possible hydrogen bonds for Gd2(bzlida)3.3H2O

Donor-H	Donor...Acceptor	H...Acceptor	Donor-H.....Acceptor
O15 -H15B 0.886(.000) 0.938	O15 ...O14 (0) 2.715(.000)	H15B ...O14 (0) 1.950(.000) 1.908	O15 -H15B ...O14 (0) 143.76(0.01) 142.83 (**)
O4 -H4A 0.884(.000) 0.938	O4 ...O14 (1) 2.701(.000)	H4A ...O14 (1) 1.819(.000) 1.765	O4 -H4A ...O14 (1) 175.17(0.01) 175.02 (**)
O15 -H15B 0.886(.000) 0.938	O15 ...O13 (2) 3.024(.000)	H15B ...O13 (2) 2.467(.000) 2.440	O15 -H15B ...O13 (2) 121.32(0.01) 120.27 (**)
O9 -H9B 0.877(.000) 0.938	O9 ...O12 (2) 2.643(.000)	H9B ...O12 (2) 1.796(.000) 1.738	O9 -H9B ...O12 (2) 161.54(0.01) 160.90 (**)
O15 -H15A 0.887(.000) 0.938	O15 ...O6 (3) 2.890(.000)	H15A ...O6 (3) 2.456(.000) 2.439	O15 -H15A ...O6 (3) 110.55(0.01) 109.43 (**)

Equivalent positions:

- (0) x, y, z
 (1) -x+1/2+1, +y+1/2, -z+1/2
 (2) -x+1/2+1, +y-1/2, -z+1/2
 (3) -x+1, -y+1, -z

Possible hydrogen bonds for La(Hbzlida)(bzlida).H2O

Donor-H	Donor...Acceptor	H...Acceptor	Donor-H.....Acceptor
O9 -H9A 0.778(.000) 0.938	O9 ...O1 (7) 2.692(.000)	H9A ...O1 (7) 1.931(.000) 1.776	O9 -H9A ...O1 (7) 165.98(0.01) 164.72 (**)

Equivalent positions:

- (7) x+1/2, -y-1/2, +z

Possible hydrogen bonds for Cu(Hbzlida)2.4H2O

Donor-H	Donor...Acceptor	H...Acceptor	Donor-H.....Acceptor
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O1 -H1 0.820 (.000) 0.938	O1 ...O6 2.796 (.001)	(0)	H1 ...O6 2.343 (.001) 2.295	(0)	O1 -H1 ...O6 115.52 (0.03) 112.86 (**)	(0)
O5 -H5B 0.976 (.001) 0.938	O5 ...O2 2.760 (.001)	(0)	H5B ...O2 2.289 (.001) 2.302	(0)	O5 -H5B ...O2 108.57 (0.03) 109.47 (**)	(0)
O6 -H6A 0.922 (.000) 0.938	O6 ...O1 2.796 (.001)	(0)	H6A ...O1 2.276 (.001) 2.269	(0)	O6 -H6A ...O1 115.25 (0.03) 114.89 (**)	(0)
O6 -H6B 0.930 (.000) 0.938	O6 ...O1 2.756 (.001)	(5)	H6B ...O1 1.930 (.001) 1.923	(5)	O6 -H6B ...O1 146.91 (0.04) 146.77 (**)	(5)
O5 -H5B 0.976 (.001) 0.938	O5 ...O3 2.854 (.001)	(5)	H5B ...O3 2.624 (.001) 2.626	(5)	O5 -H5B ...O3 93.46 (0.03) 94.29 (**)	(6)

Equivalent positions:

(0) x,y,z
(5) x-1/2,+y,-z+1/2
(6) -x+3,-y+1,-z+1

Possible hydrogen bonds for [Cu(bzlida)₂(Er(CH₃COO)(H₂O)₅)₂] [Cu(bzlida)₂] 6H₂O

Donor-H	Donor...Acceptor		H...Acceptor		Donor-H.....Acceptor	
O14 -H14B 0.979 (.000) 0.938	O14 ...O2 2.585 (.001)	(0)	H14B ...O2 1.784 (.001) 1.814	(0)	O14 -H14B ...O2 136.70 (0.05) 137.60 (**)	(0)
O15 -H15B 1.083 (.000) 0.938	O15 ...O5 2.936 (.001)	(0)	H15B ...O5 2.174 (.001) 2.262	(0)	O15 -H15B ...O5 125.27 (0.04) 128.28 (**)	(0)
O16 -H16A 0.981 (.000) 0.938	O16 ...O18 2.889 (.001)	(0)	H16A ...O18 2.021 (.001) 2.057	(0)	O16 -H16A ...O18 146.31 (0.05) 146.98 (**)	(0)
O16 -H16B 0.984 (.000) 0.938	O16 ...O10 2.835 (.001)	(0)	H16B ...O10 1.988 (.001) 2.025	(0)	O16 -H16B ...O10 142.82 (0.04) 143.61 (**)	(0)
O17 -H17B 0.977 (.000) 0.938	O17 ...O4 2.841 (.001)	(0)	H17B ...O4 1.896 (.001) 1.933	(0)	O17 -H17B ...O4 161.85 (0.04) 162.21 (**)	(0)
O11 -H11A 0.969 (.000) 0.938	O11 ...O9 2.740 (.001)	(1)	H11A ...O9 2.025 (.001) 2.045	(1)	O11 -H11A ...O9 128.97 (0.04) 129.65 (**)	(1)
O13 -H13B 0.960 (.000) 0.938	O13 ...O7 2.908 (.001)	(2)	H13B ...O7 1.980 (.001) 2.001	(2)	O13 -H13B ...O7 161.81 (0.05) 162.01 (**)	(2)
O11 -H11B 0.967 (.000) 0.938	O11 ...O8 2.800 (.001)	(2)	H11B ...O8 1.878 (.001) 1.906	(2)	O11 -H11B ...O8 158.31 (0.05) 158.64 (**)	(2)
O18 -H18D 1.009 (.000) 0.938	O18 ...O8 3.004 (.001)	(2)	H18D ...O8 2.023 (.001) 2.091	(2)	O18 -H18D ...O8 163.59 (0.04) 164.14 (**)	(2)
O12 -H12A 0.967 (.001) 0.938	O12 ...O5 2.548 (.001)	(3)	H12A ...O5 1.705 (.001) 1.728	(3)	O12 -H12A ...O5 143.55 (0.05) 144.11 (**)	(3)
O13 -H13A 0.969 (.000) 0.938	O13 ...O6 2.775 (.001)	(3)	H13A ...O6 2.109 (.001) 2.127	(3)	O13 -H13A ...O6 124.42 (0.04) 125.12 (**)	(3)

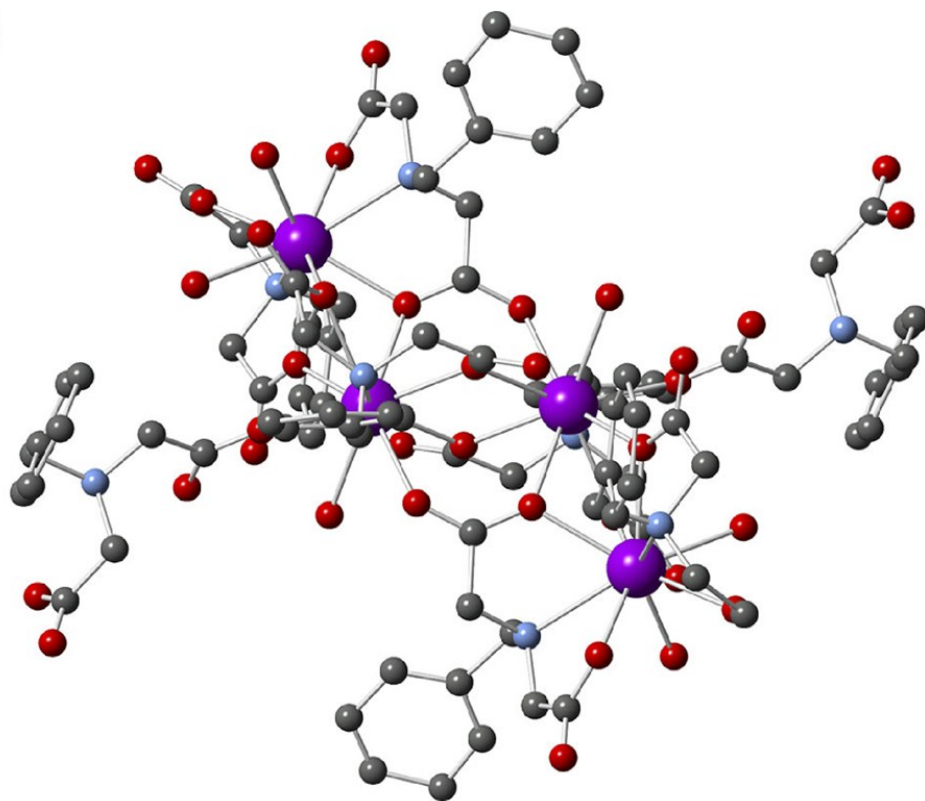
O14	-H14A	O14	...O3	(4)	H14A	...O3	(4)	O14	-H14A	...O3	(4)
0.971(.001)		2.846(.001)			2.172(.001)			125.36(0.04)			
0.938					2.191			126.07		(**)	
O18	-H18C	O18	...O17	(7)	H18C	...O17	(7)	O18	-H18C	...O17	(7)
0.975(.000)		2.957(.001)			2.073(.001)			149.91(0.04)			
0.938					2.105			150.41		(**)	
O17	-H17A	O17	...O18	(8)	H17A	...O18	(8)	O17	-H17A	...O18	(8)
1.054(.000)		2.957(.001)			2.062(.001)			141.10(0.04)			
0.938					2.153			143.03		(**)	

Equivalent positions:

- (0) x, y, z
- (1) x, +y+1, +z
- (2) -x+1, -y+1, -z
- (3) x, +y-1, +z
- (4) -x+2, -y, -z
- (7) -x+2, +y+1/2, -z+1/2
- (8) -x+2, +y-1/2, -z+1/2

Figure S1 Tetranuclear [Gd_4 (a)] and dinuclear [Gd_2 (b)] molecular models used in the theoretical study.

(a)



(b)

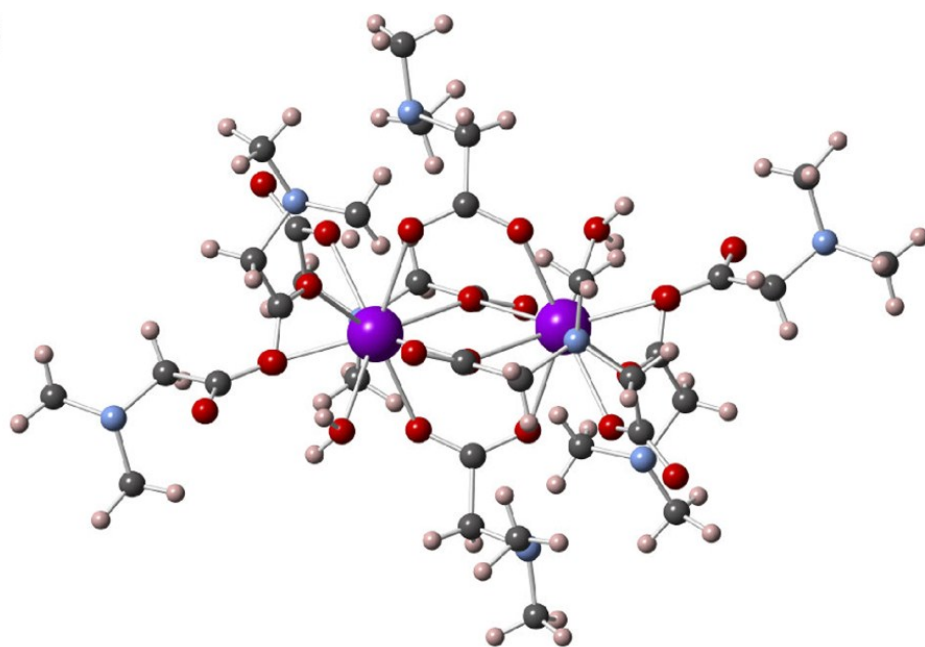


Figure S2 Scheme of the magnetic exchange interactions present in complex **6** together with a description of the pathways involved in each one of these interactions. Purple, red and grey balls represent gadolinium, oxygen and carbon atoms, respectively.

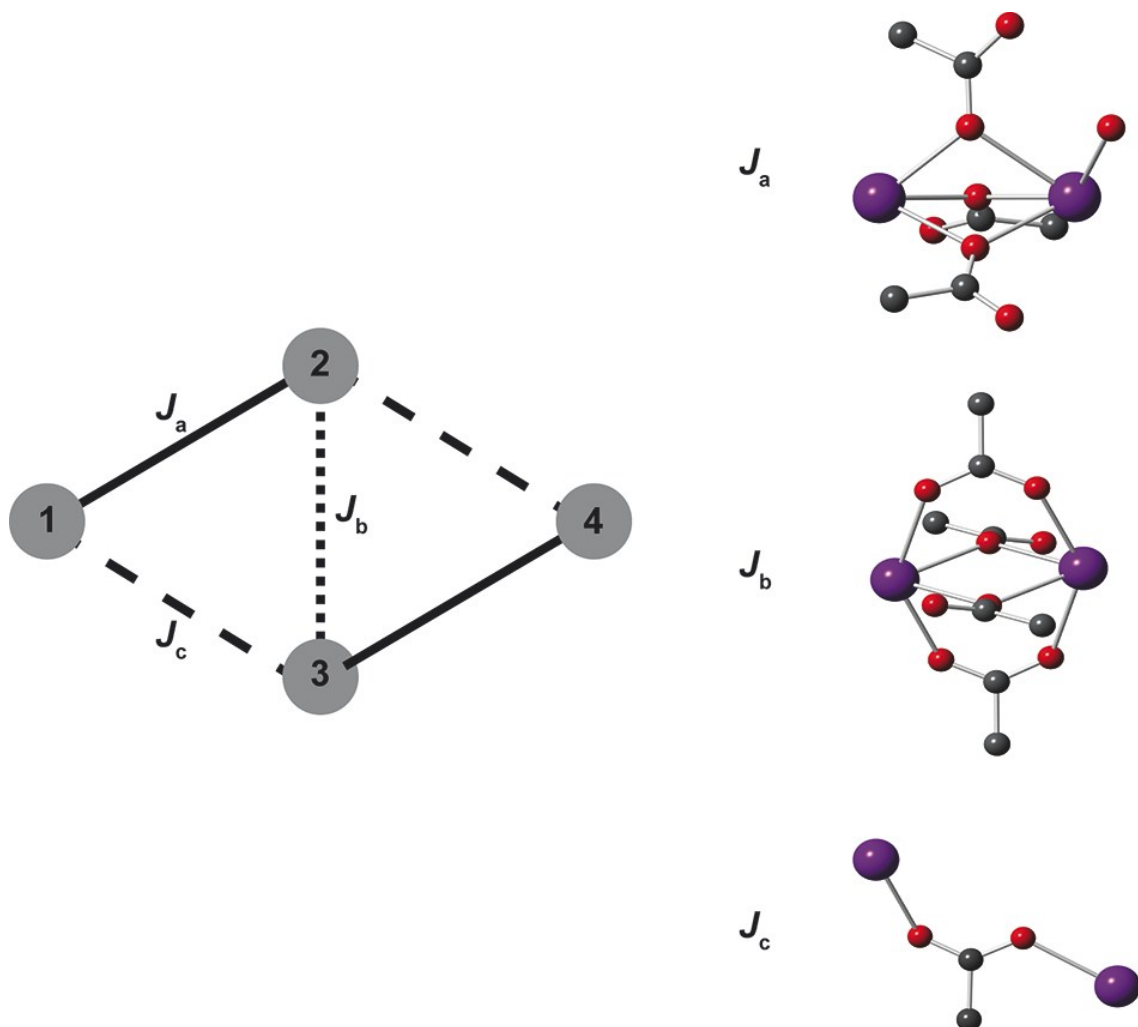


Figure S3 Species distribution diagram of the Ce-bzlida system, in 0.50 mol L⁻¹ Me₄NCl at 25.0 °C (total [Ce³⁺] = 1 mmol L⁻¹ and total [bzlida] = 2 mmol L⁻¹). The fully deprotonated ligand is represented as L²⁻.

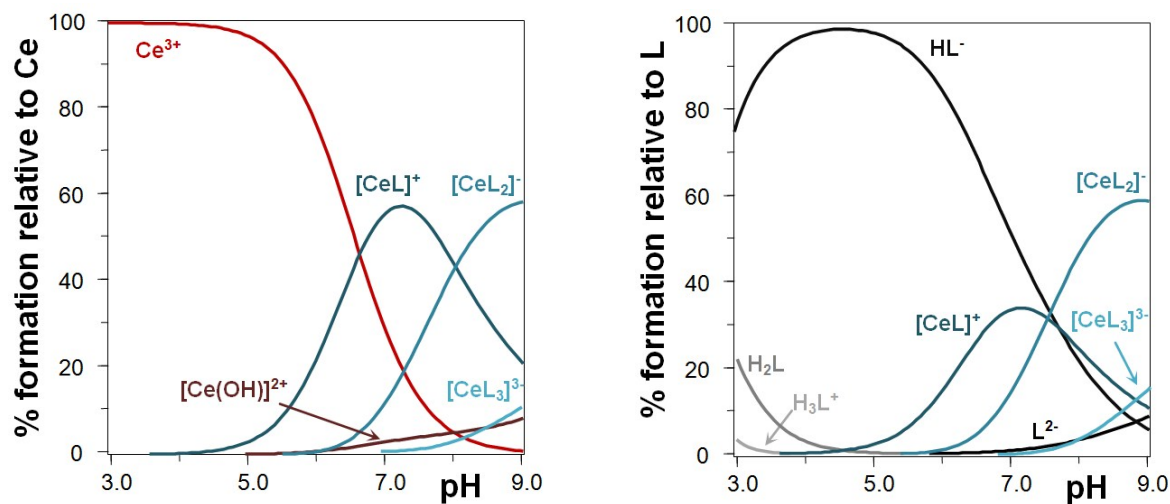


Figure S4 Species distribution diagram of the Gd-bzlida system, in 0.50 mol L⁻¹ Me₄NCl at 25.0 °C (total [Gd³⁺] = 1 mmol L⁻¹ and total [bzlida] = 2 mmol L⁻¹). The fully deprotonated ligand is represented as L²⁻.

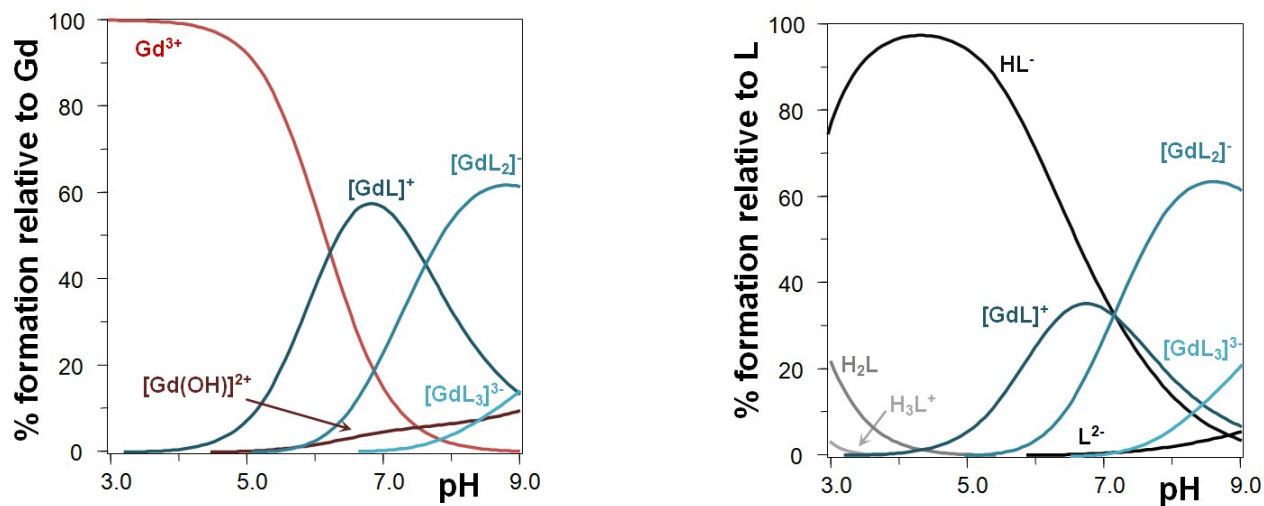


Figure S5 Species distribution diagram of the Yb-bzlida system, in 0.50 mol L⁻¹ Me₄NCl at 25.0 °C (total [Yb³⁺] = 1 mmol L⁻¹ and total [bzlida] = 2 mmol L⁻¹). The fully deprotonated ligand is represented as L²⁻.

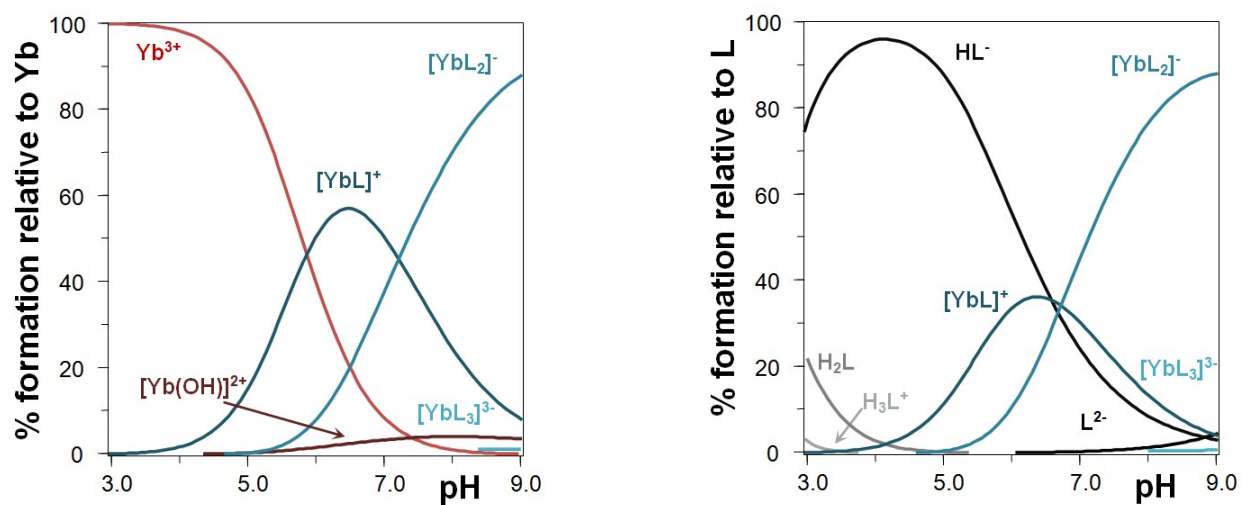


Figure S6 Thermogravimetric diagram of $[\text{La}(\text{bz}lida)(\text{Hbz}lida)] \cdot 1.5\text{H}_2\text{O}$ (1). Theoretical value for water content is 4.4%.

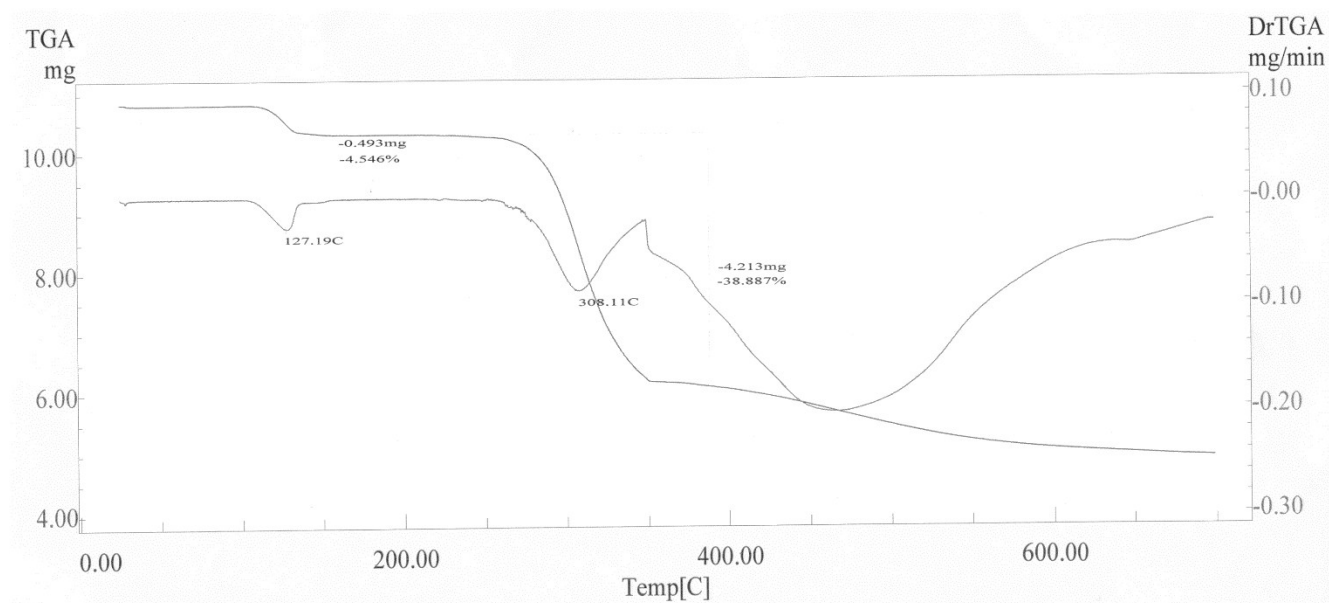


Figure S7 Thermogravimetric diagram of $[\text{Nd}(\text{bz}lida)(\text{Hbz}lida)] \cdot 1.5\text{H}_2\text{O}$ (2). Theoretical value for water content is 4.4%.

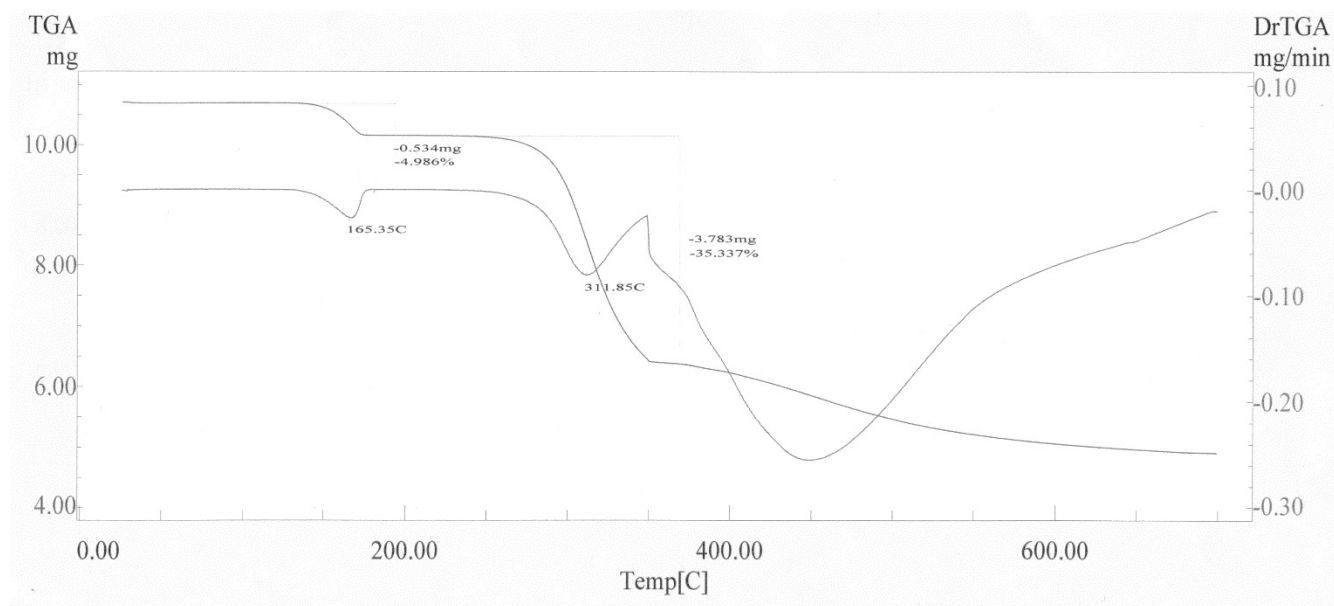


Figure S8 Thermogravimetric diagram of $[\text{Sm}_2(\text{bz}l\text{id}a)_3] \cdot 3\text{H}_2\text{O}$ (**4**). Theoretical value for water content is 5.3%.

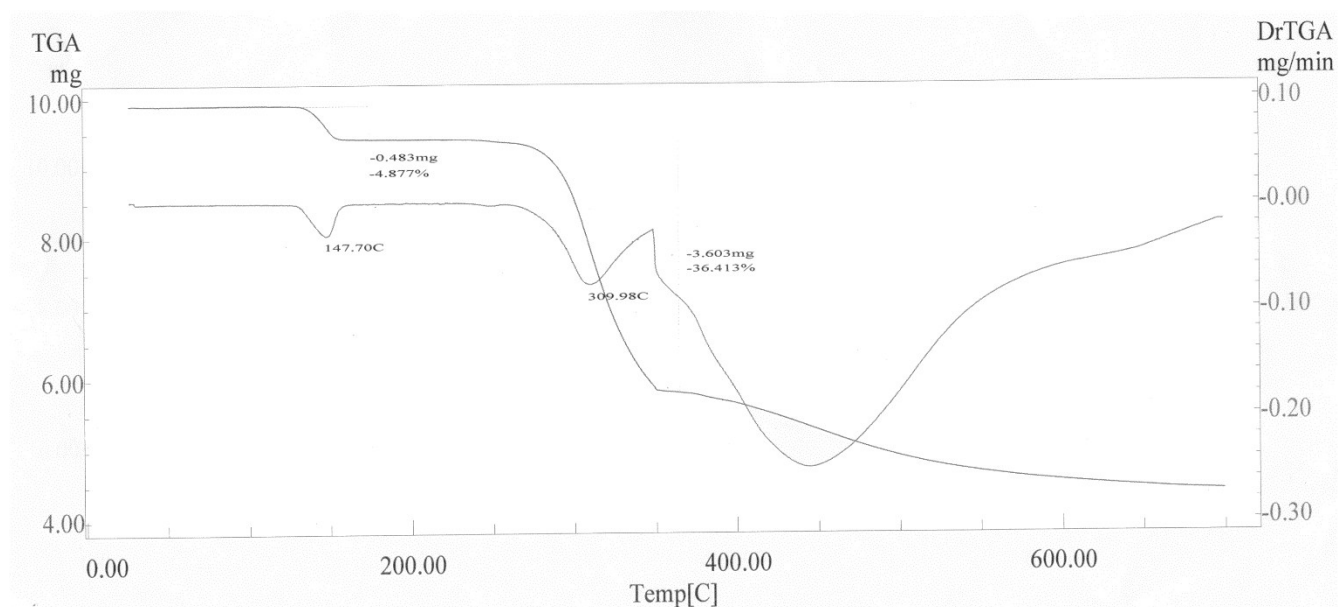


Figure S9 Thermogravimetric diagram of $[\text{Gd}_2(\text{bz}l\text{id}a)_3] \cdot 3\text{H}_2\text{O}$ (**6**). Theoretical value for water content is 5.2%.

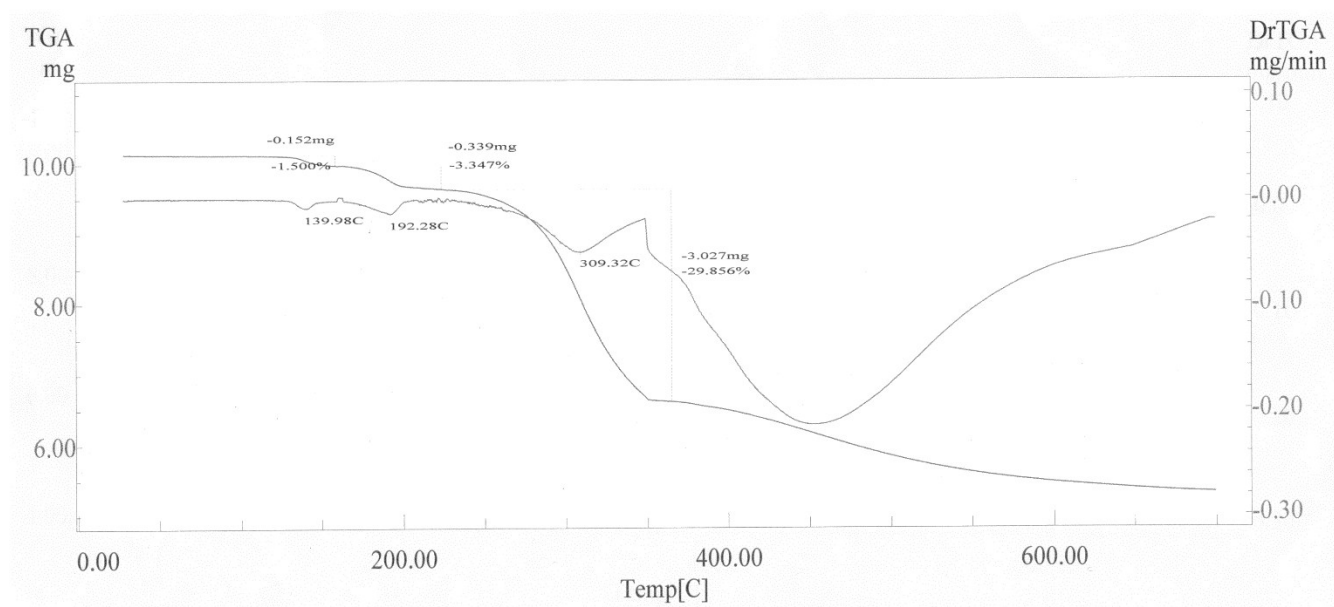


Figure S10 Powder X ray diffraction diagram of the bulk crystalline solid obtained from the hydrothermal reaction of a mixture containing H_2bzilda and Nd_2O_3 (black) compared to the calculated pattern obtained from the single crystal of $[\text{Nd}_2(\text{bzilda})_3] \cdot 3\text{H}_2\text{O}$ (**3**), isolated from the bulk (red). Inset: overlay of XRPD of the bulk solids obtained from the reaction of H_2bzilda with La_2O_3 (1) (black) and of H_2bzilda with Nd_2O_3 , (2) (blue) showing their identity.

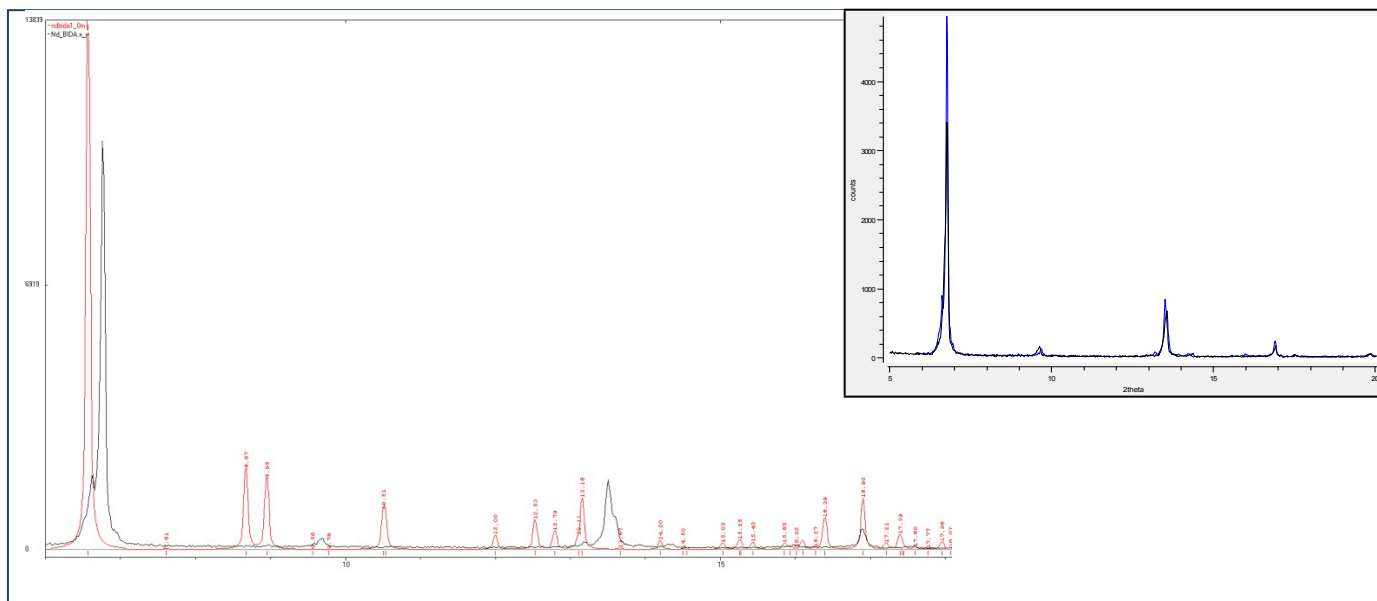


Figure S11 Hydrogen bonds between the coordinated water molecules O13, O14, O15 and the uncoordinated oxygens O2, O6, O10 in **3**.

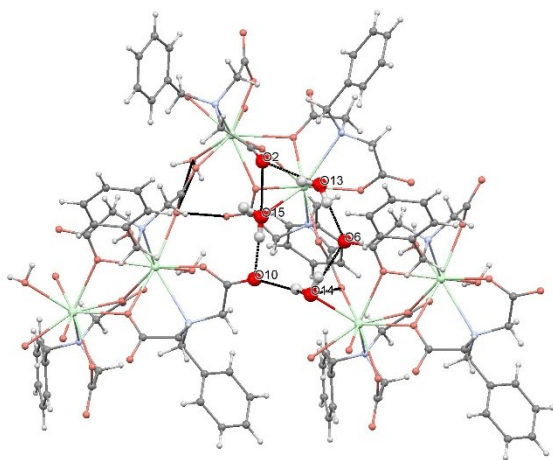


Figure S12 Top: assembly in hydrogen bonded (dotted) layers of the chains of [La(Hbzida)(bzida)]·H₂O; bottom: edge-on view of the layers, exposing phenyl groups at the surfaces.

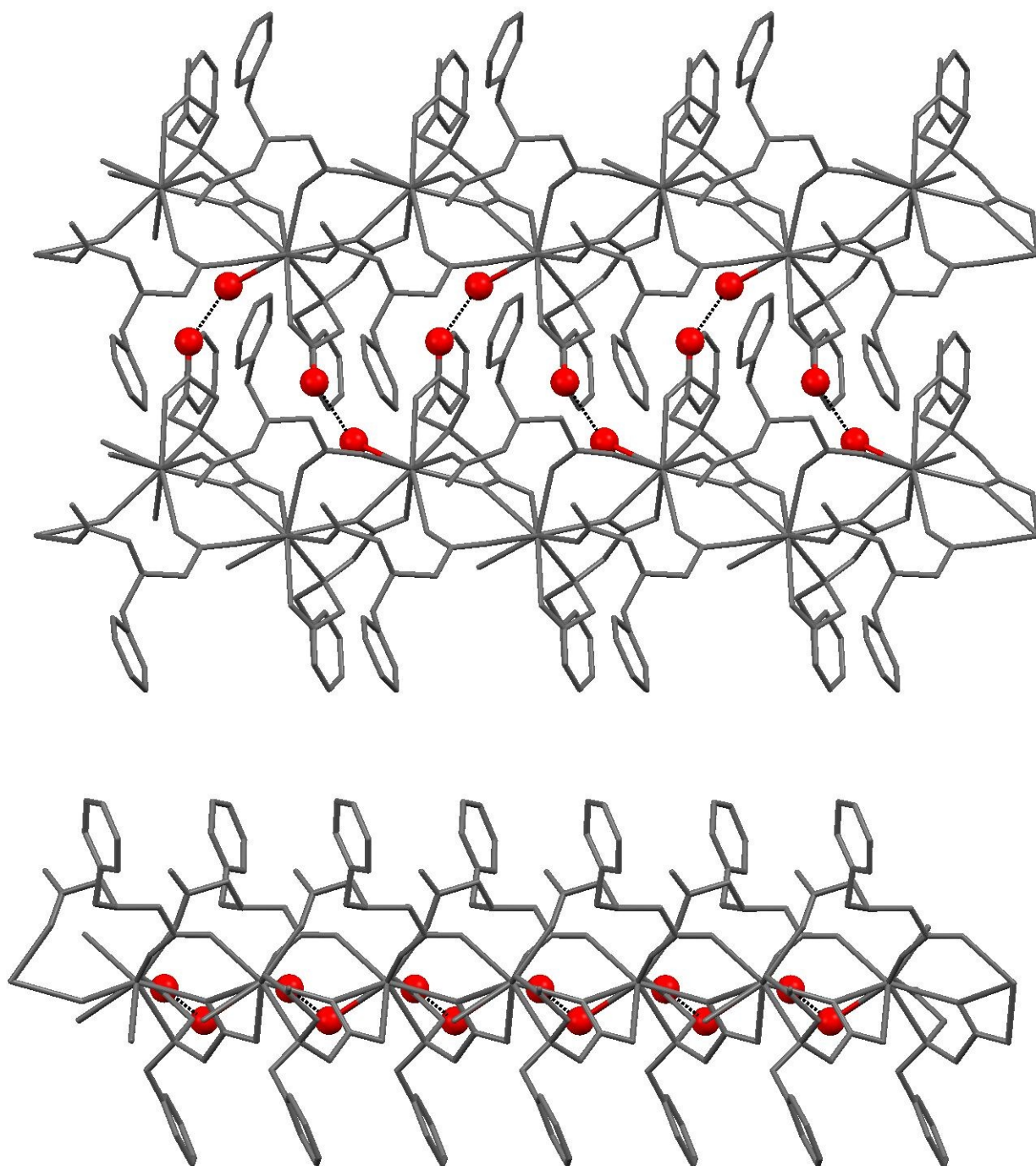


Figure S13 Hydrogen bonded layered structural arrangement in the **ab** plane of cations (coloured) and anions (blue) in **7**.

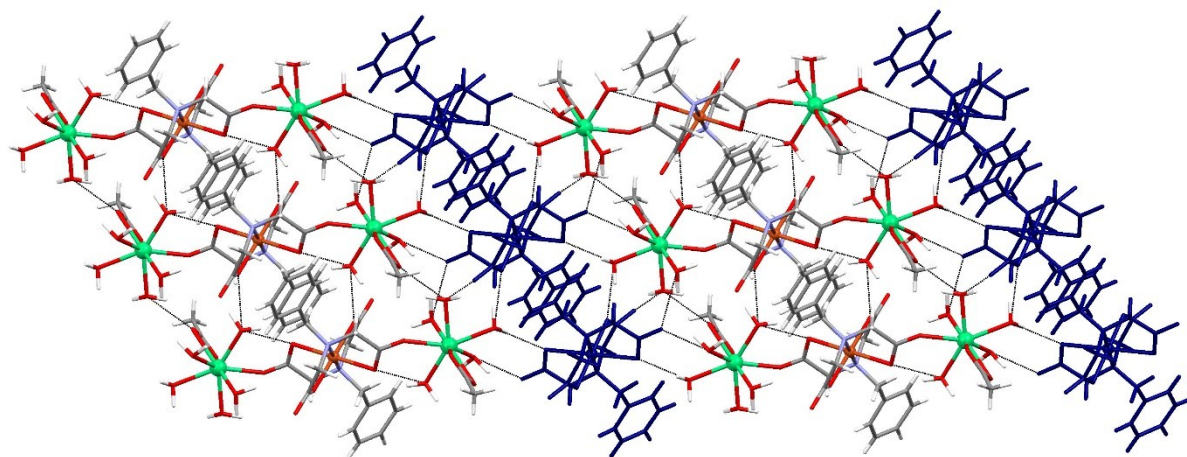


Figure S14 Location of voids hosting loosely bound water molecules in $[\text{Cu}(\text{bz}l\text{id}a)_2(\text{Er}(\text{CH}_3\text{COO})(\text{H}_2\text{O})_5)_2][\text{Cu}(\text{bz}l\text{id}a)_2] \cdot 6\text{H}_2\text{O}$. These voids contain the electron density equivalent to one water molecule each, but it was not possible to build or localize a molecular model of the disordered solvent.

