

SUPPORTING INFORMATION for:

Water oxidation catalysis upon evolution of Co(III) cubanes in aqueous media

Andrea Genoni,^a Giuseppina La Ganga,^c Andrea Volpe,^a Fausto Puntoriero,^c Marilena Di Valentin,^a
Marcella Bonchio,^a Mirco Natali^{b*} and Andrea Sartorel^{a*}

^{a.} *Department of Chemical Sciences, University of Padova and ITM-CNR, via Marzolo 1, 35131 Padova (Italy).*

^{b.} *Dipartimento di Scienze Chimiche e Farmaceutiche, Università di Ferrara and Centro Interuniversitario per la Conversione Chimica dell'Energia Solare, sezione di Ferrara, Via Fossato di Mortara 17-19, 44121 Ferrara, Italy.*

^{c.} *Dipartimento di Scienze Chimiche, Università di Messina and Centro Interuniversitario per la Conversione Chimica dell'Energia Solare, sezione di Messina, Via Sperone 31, 98166 Messina, Italy.*

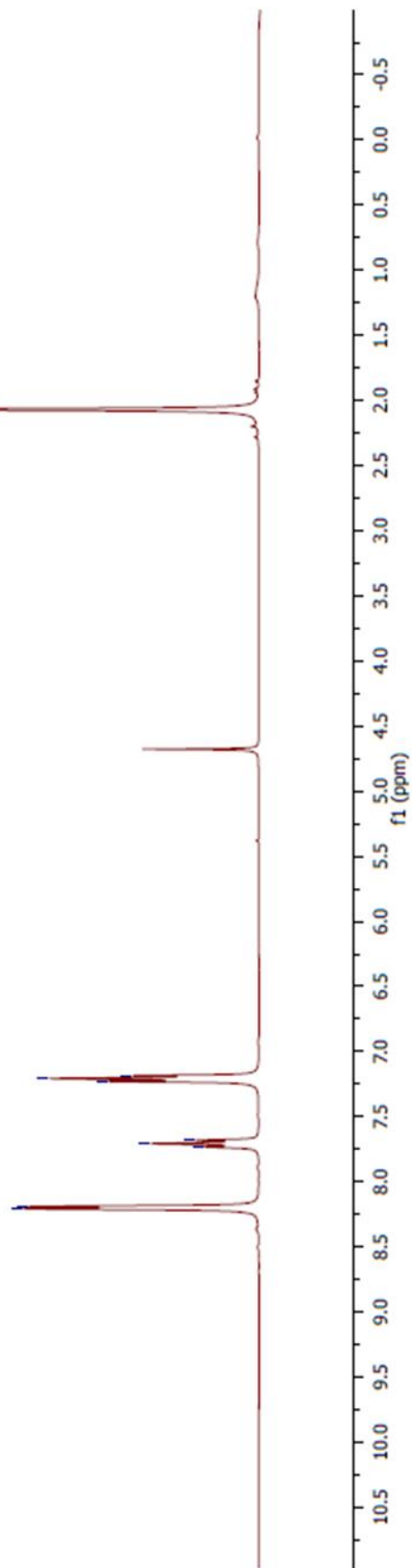
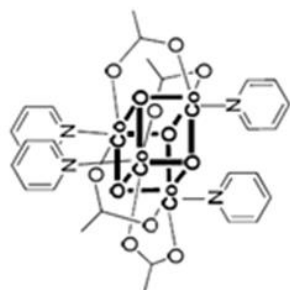
Content

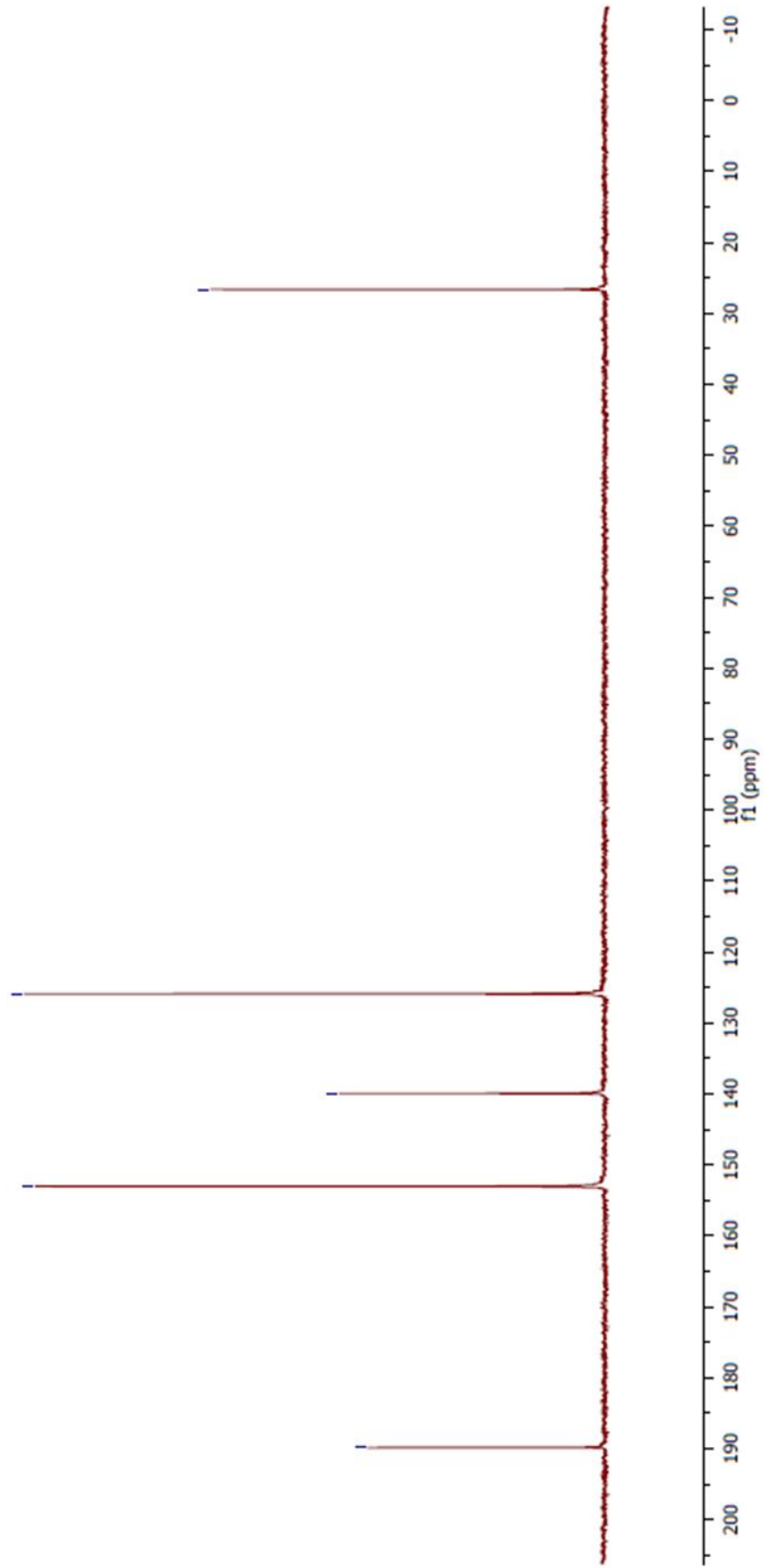
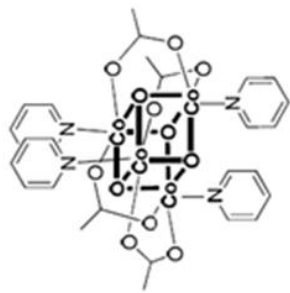
¹H and ¹³C-NMR, ESI-MS of purified **1-X**.

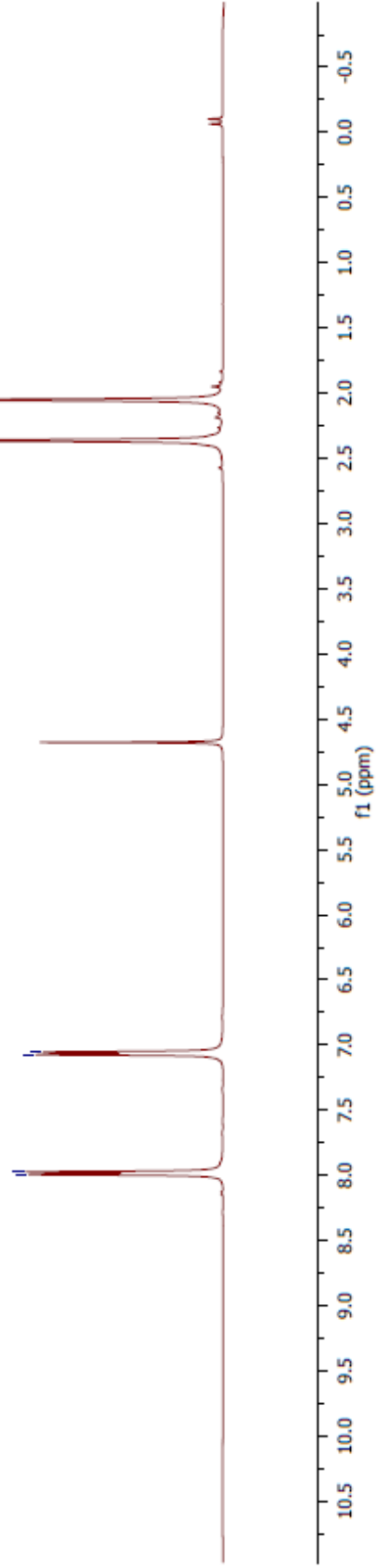
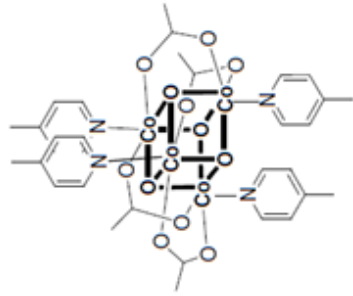
UV-Vis evolution of **1-H** in 10 mM borate buffer, pH 8.

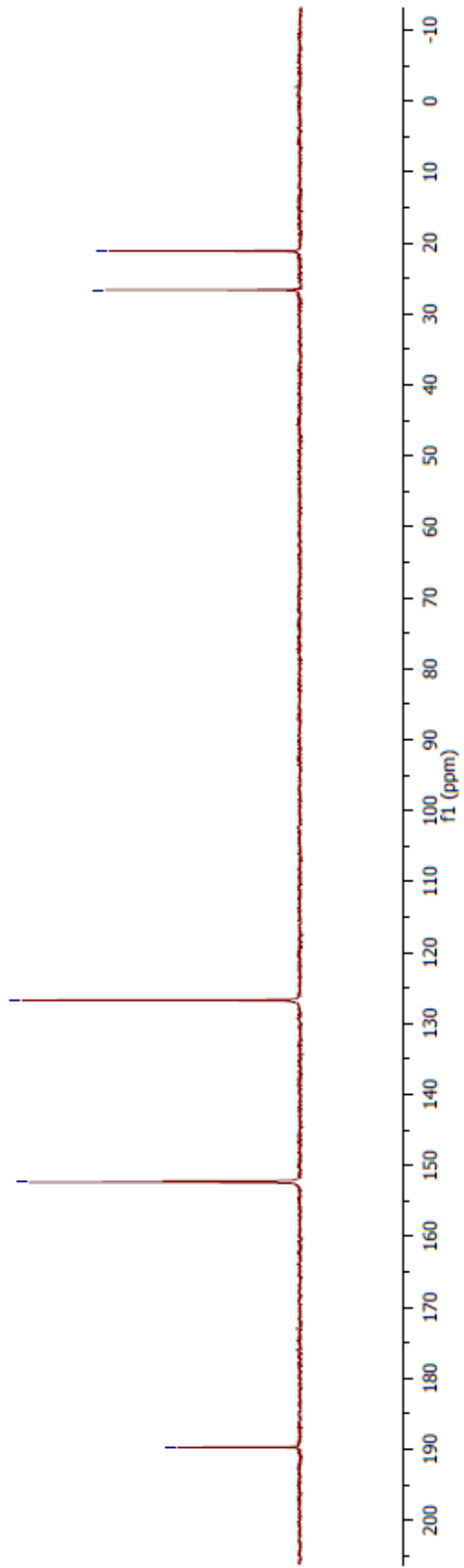
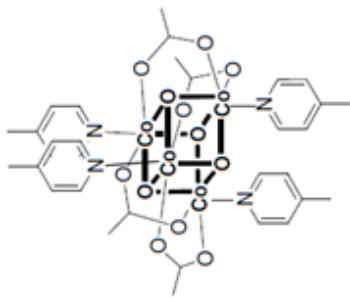
Hole scavenging analysis of **1-X** in 50:50 acetonitrile : 10 mM borate buffer, pH 8.

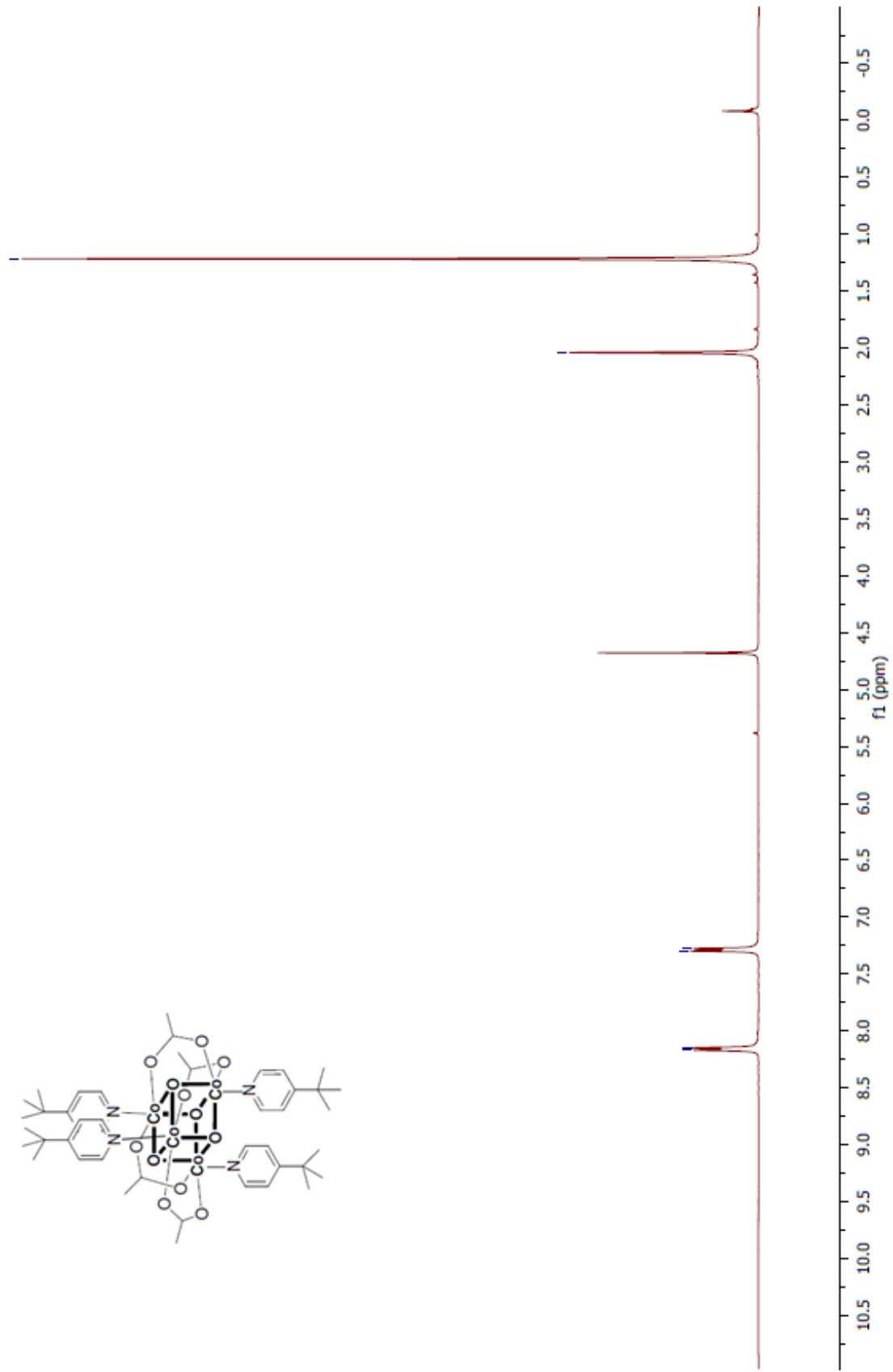
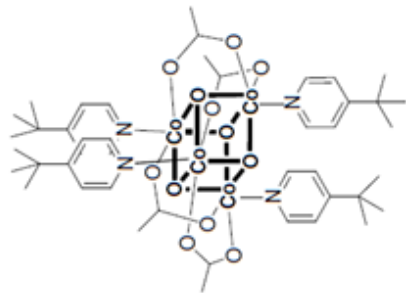
ESI-MS analysis of a photocatalytic reaction mixture.

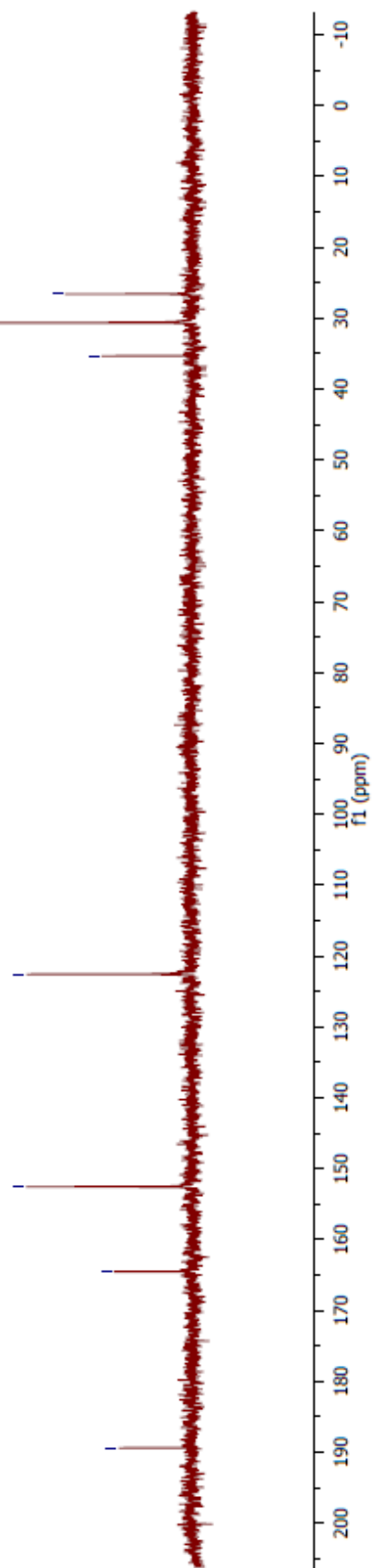
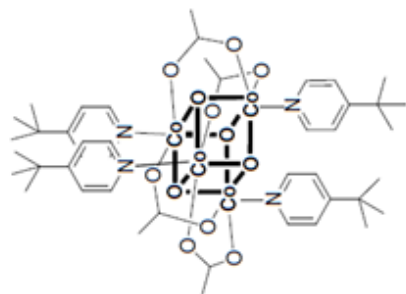


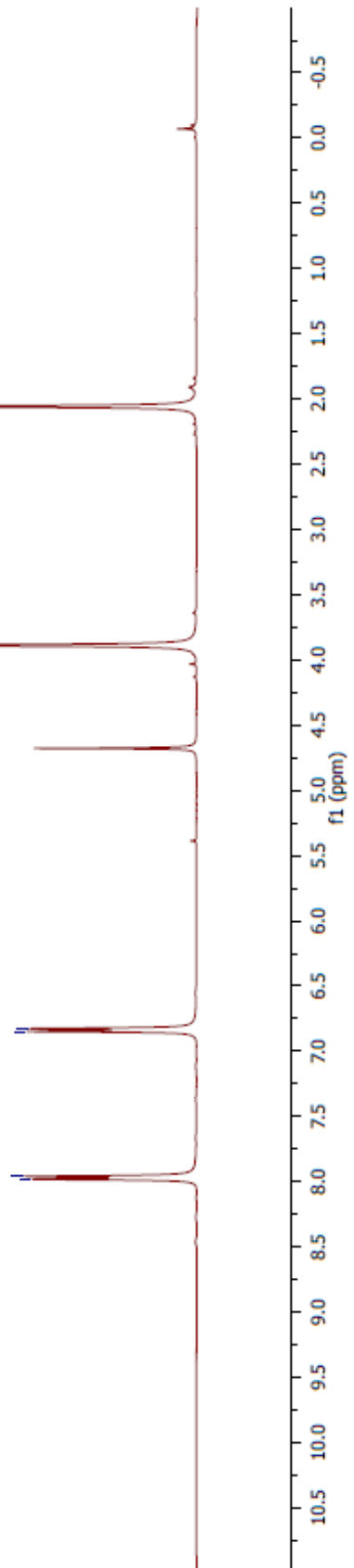
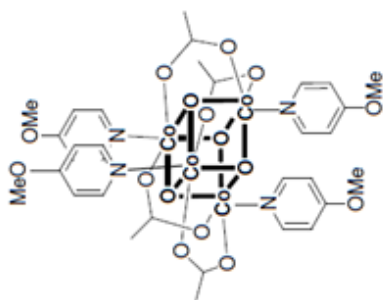


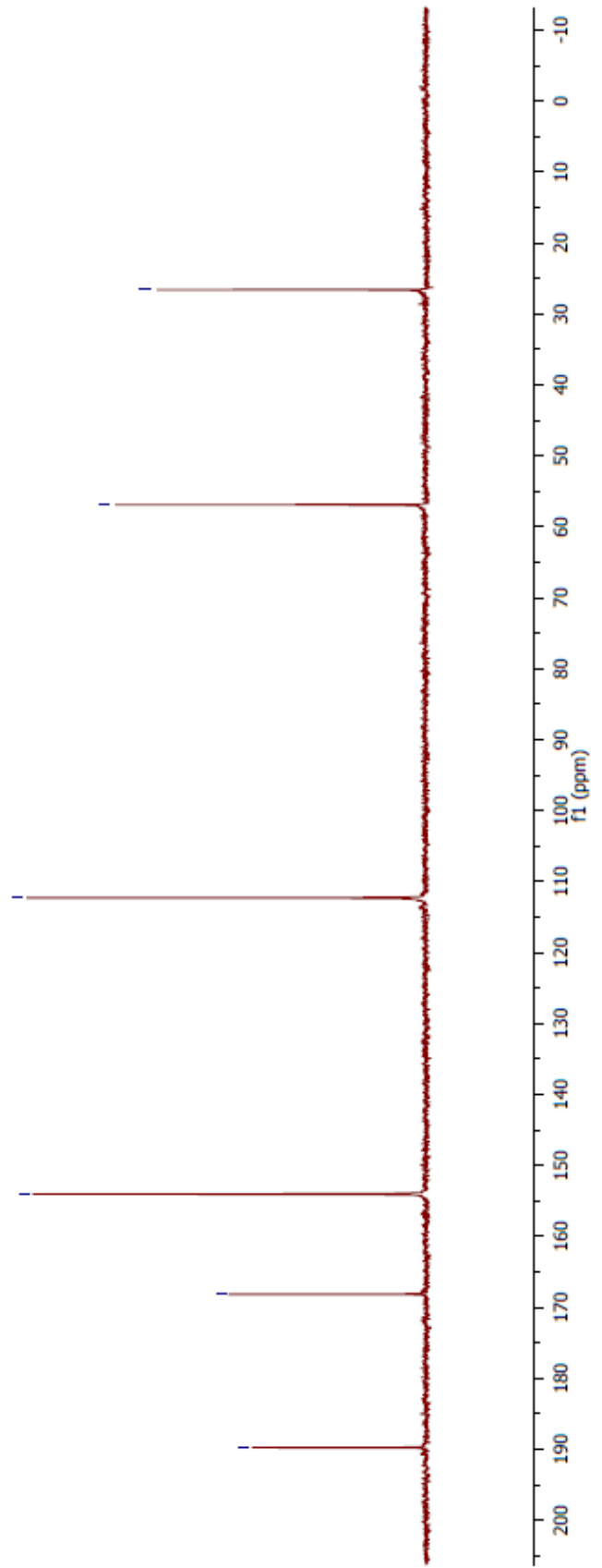
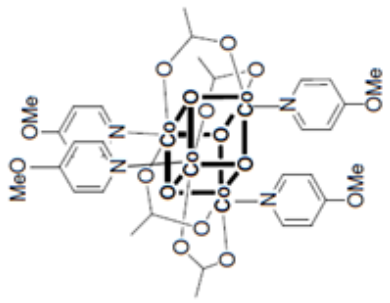


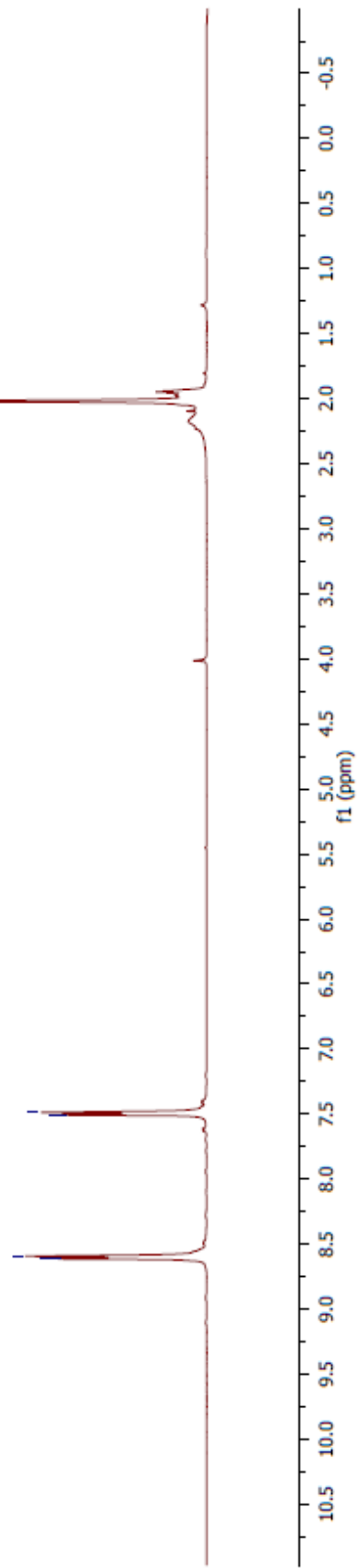
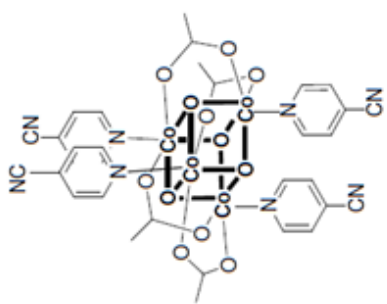


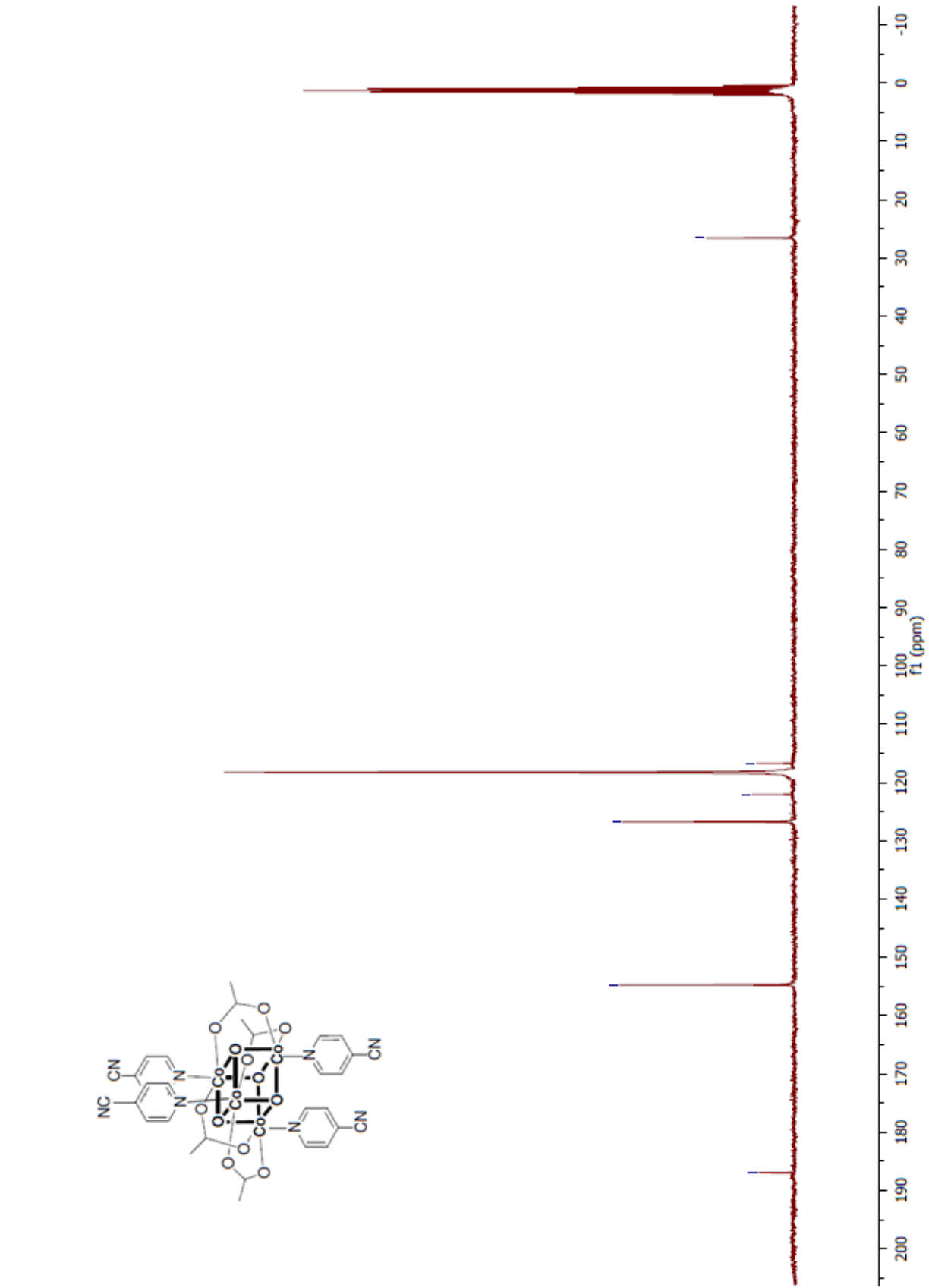
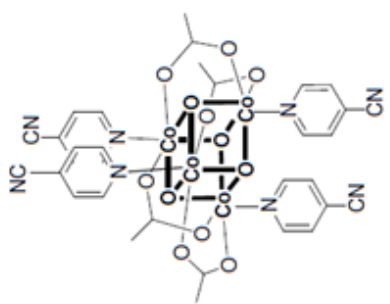


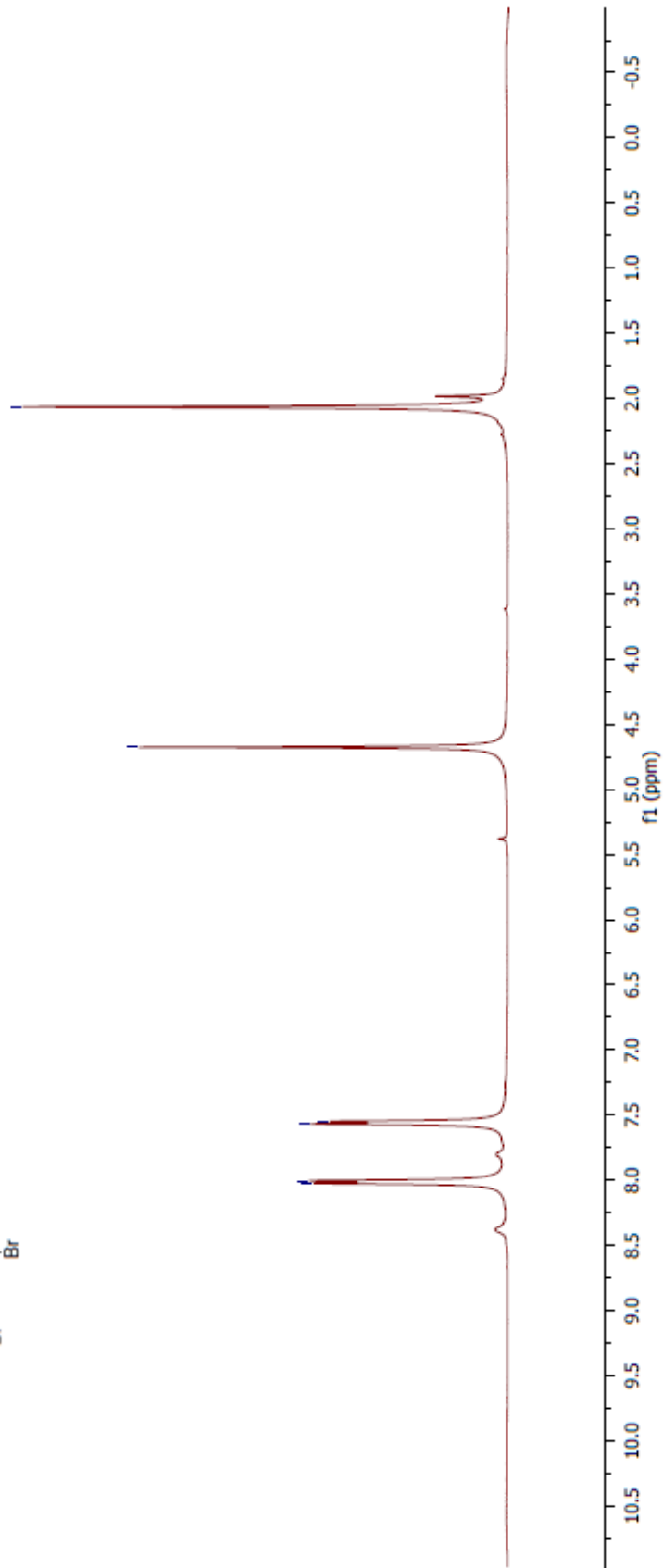
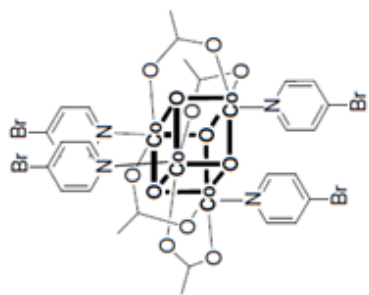


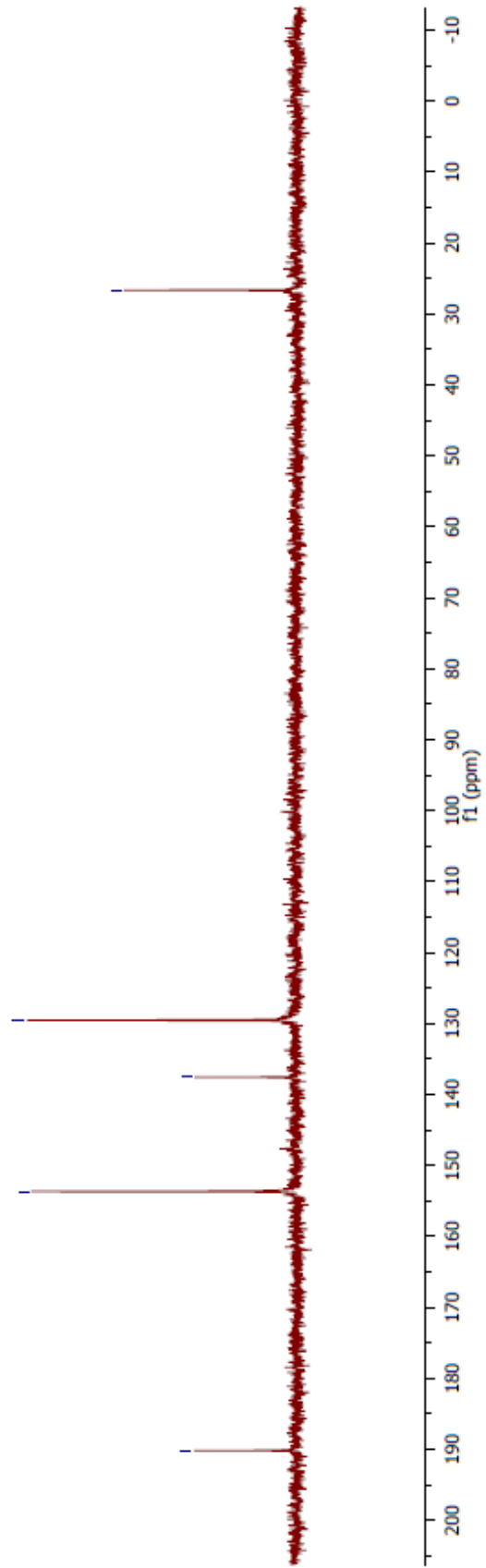
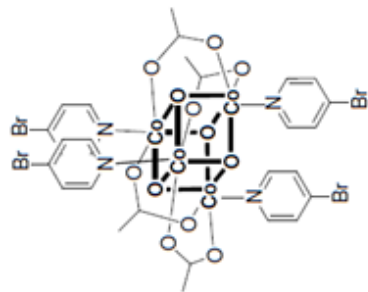


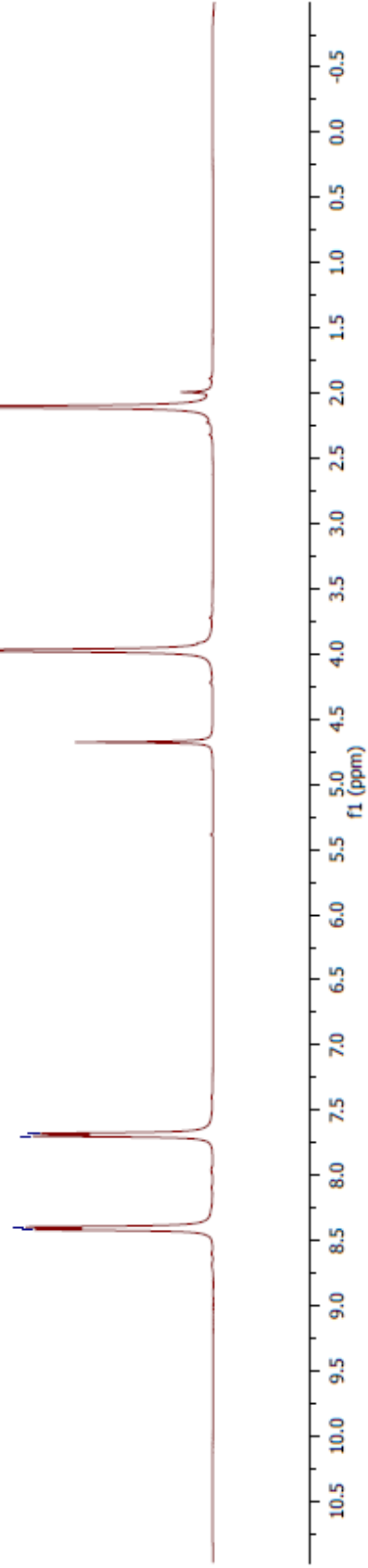
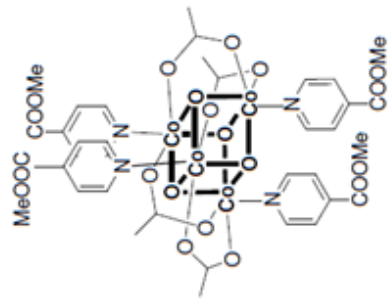


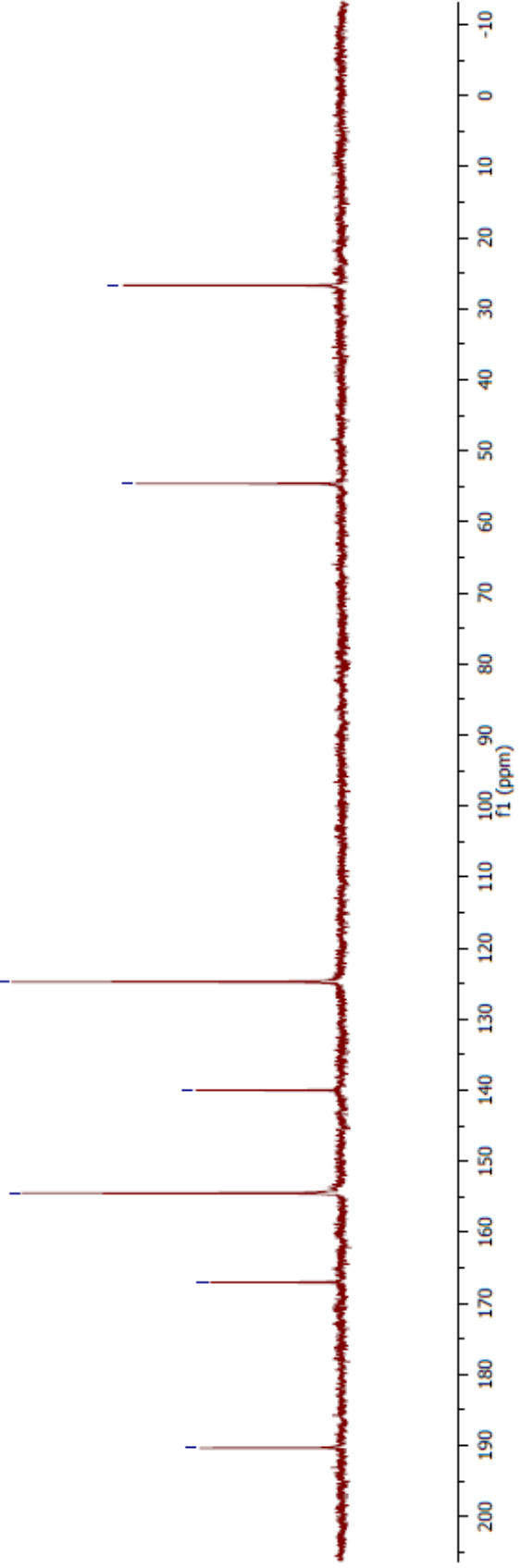
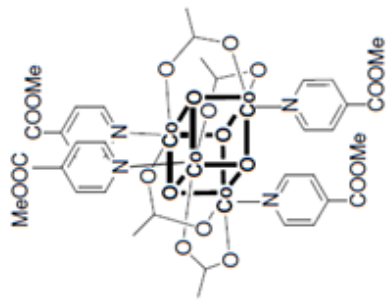


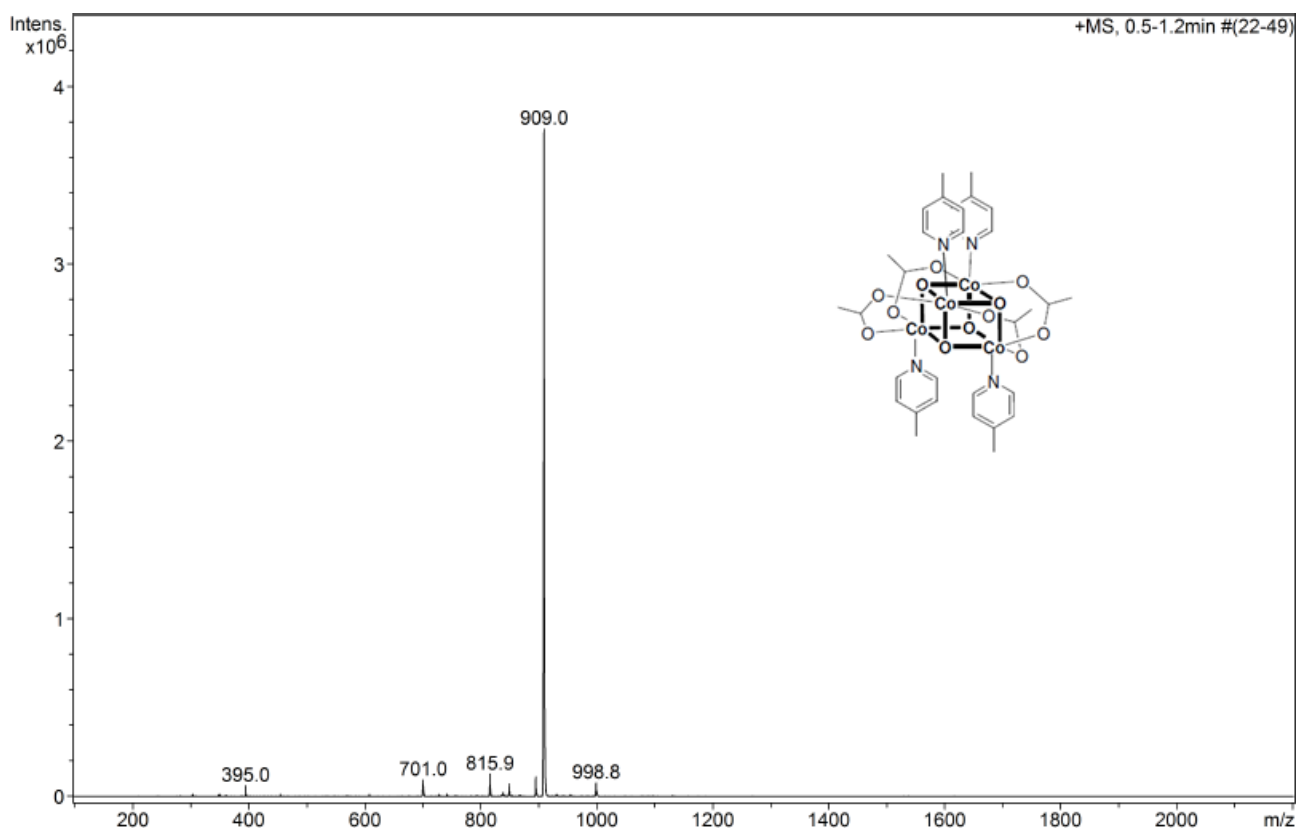
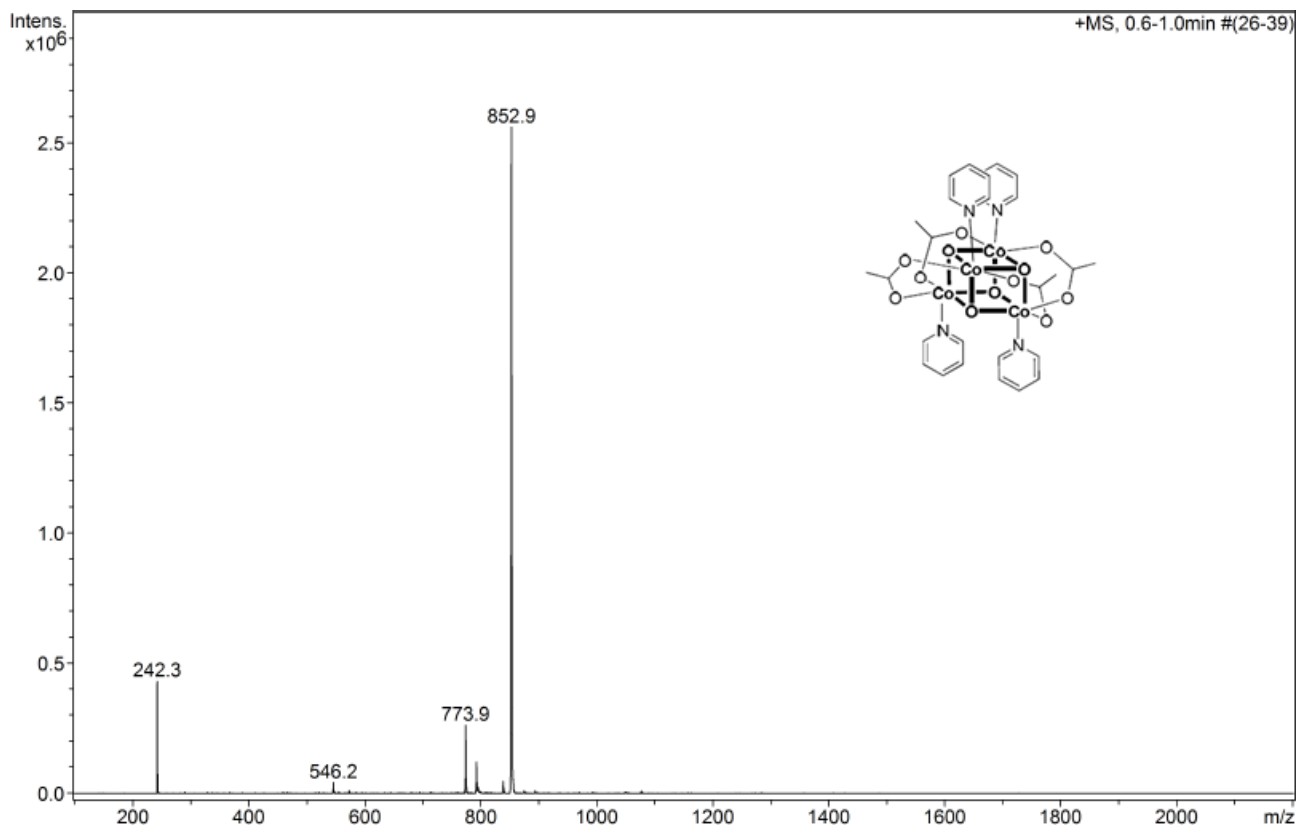


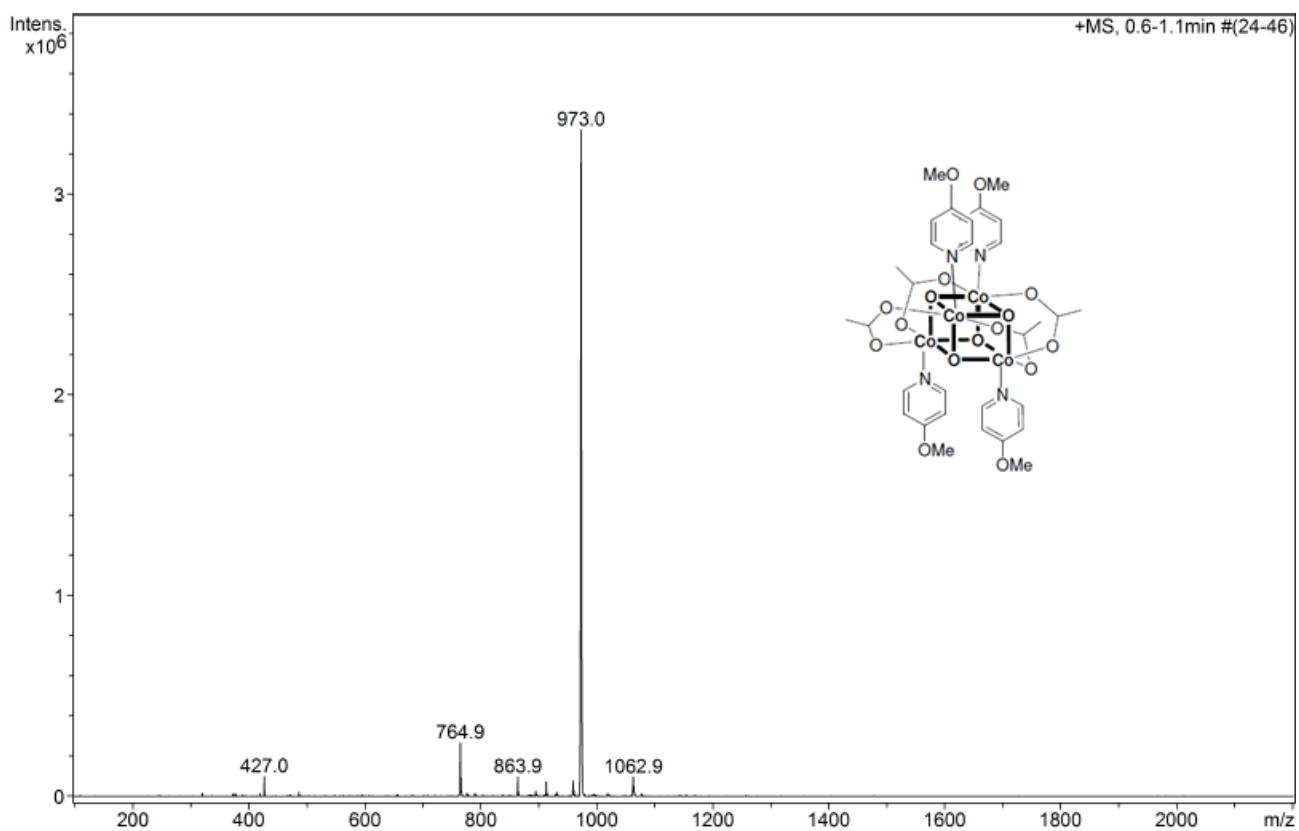
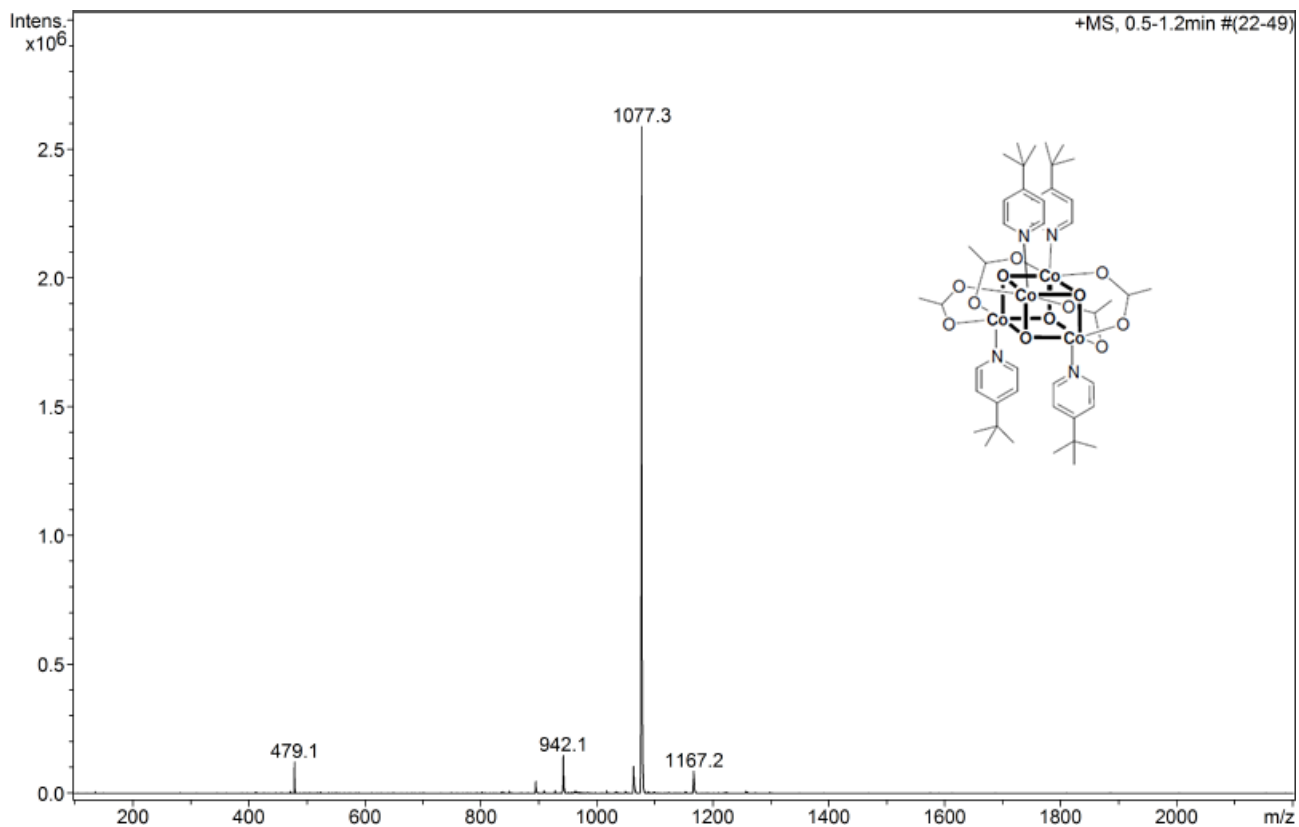


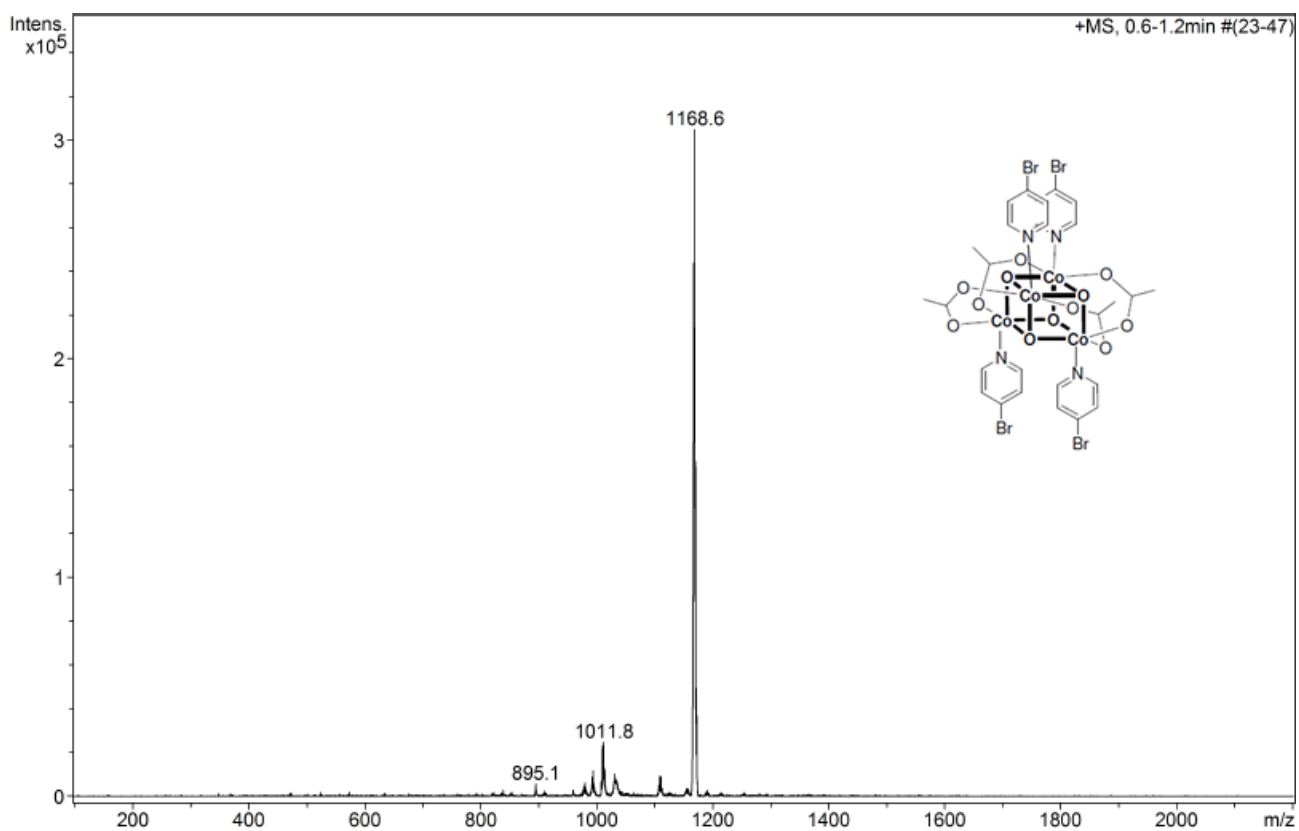
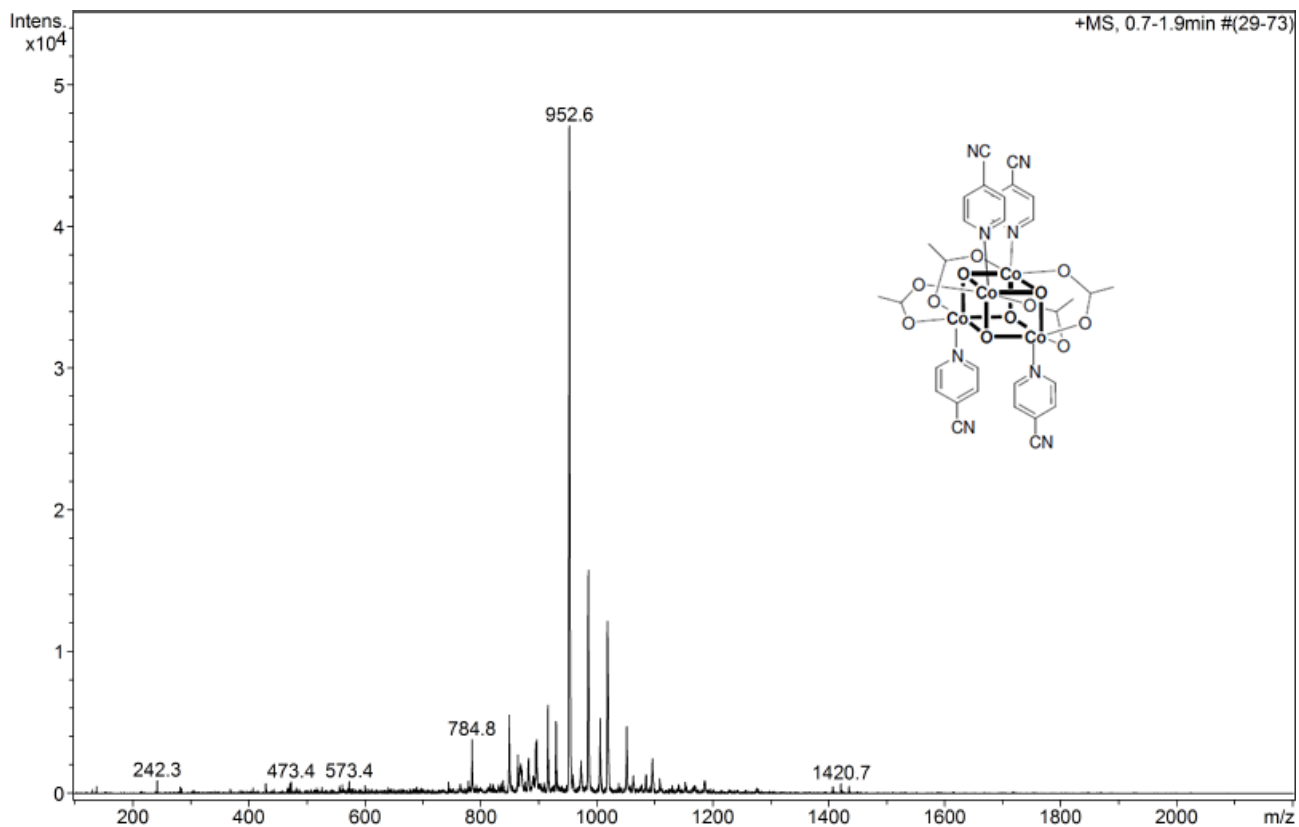


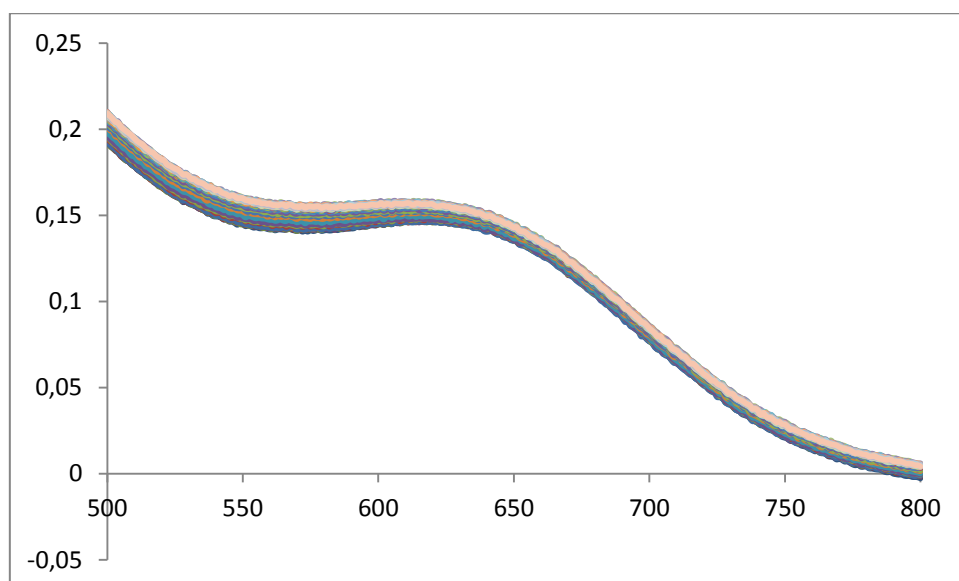
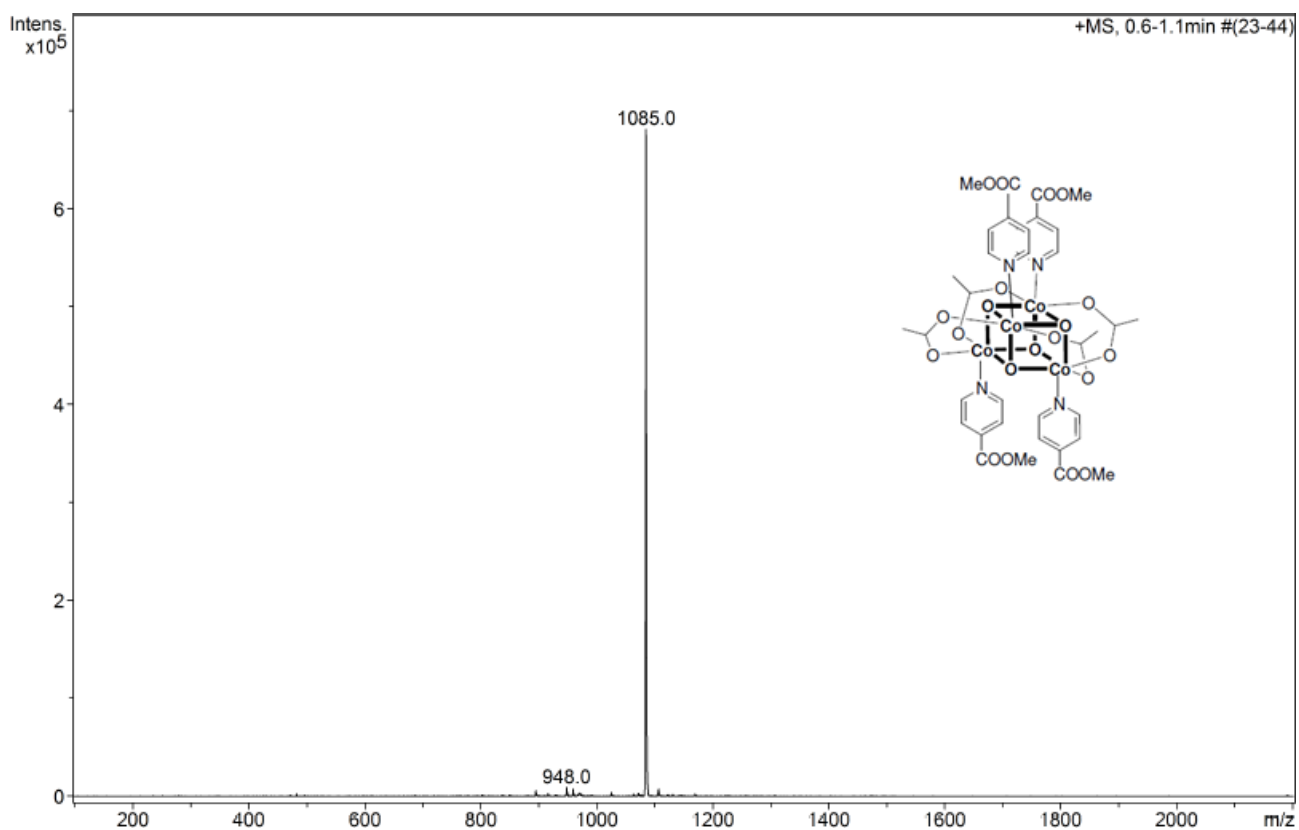










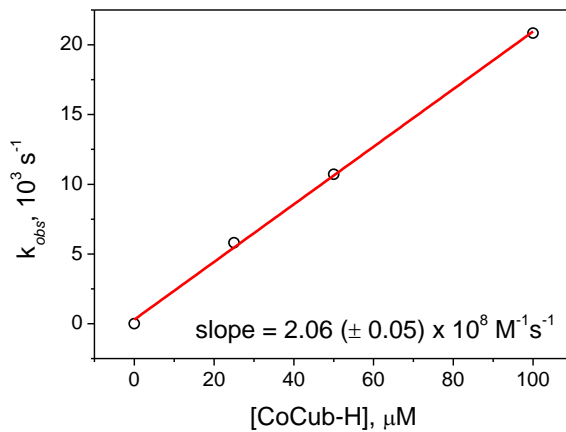
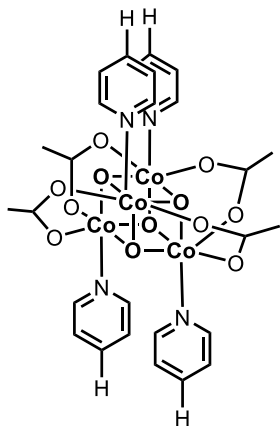


UV-Vis evolution of **1-H** (5 mM) in 10 mM borate buffer, pH 8, over the course of 12 hours. Optical path of the cuvette: 1 mm.

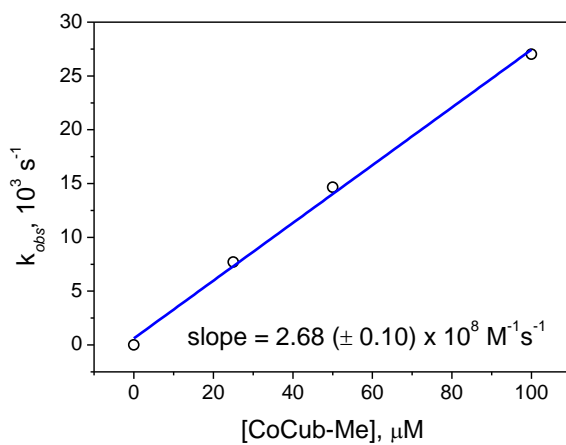
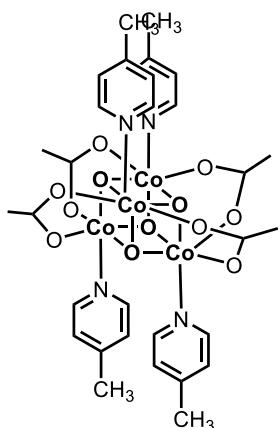
Hole scavenging in 50:50 acetonitrile:10 mM borate buffer, pH 8 (50 μ M Ru(bpy)₃Cl₂, 5 mM

Na₂S₂O₈, 0-100 μ M 1-X

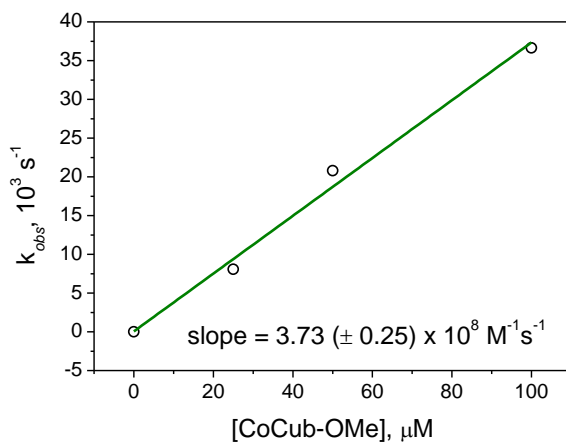
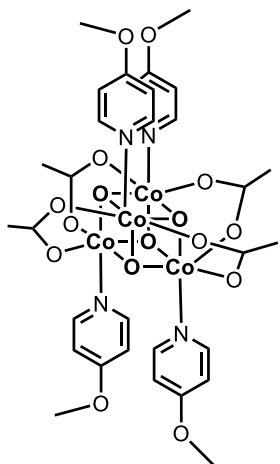
1-H



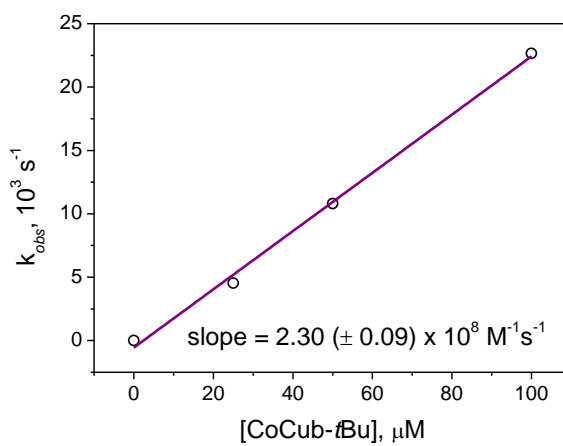
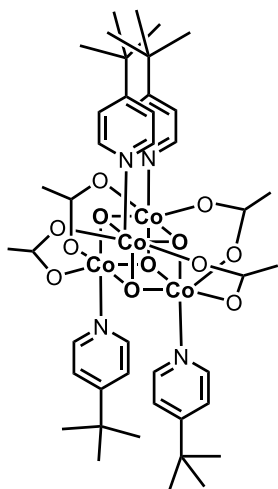
1-Me



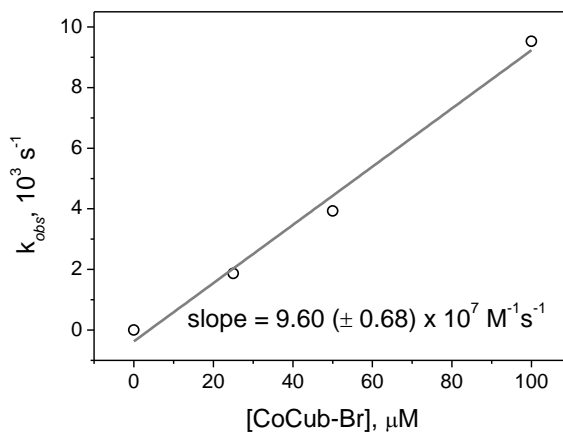
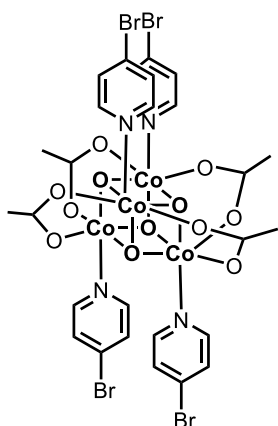
1-OMe



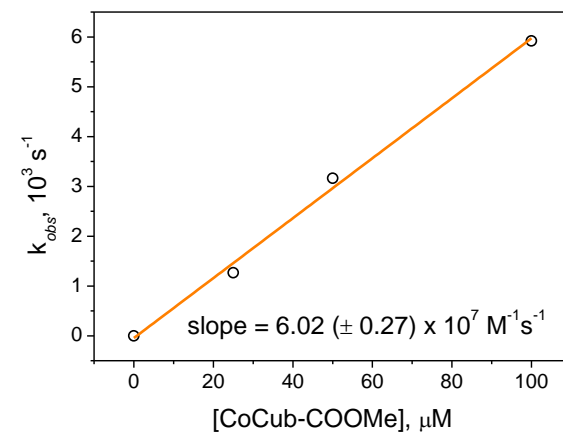
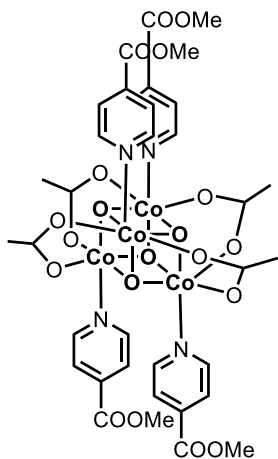
1-tBu



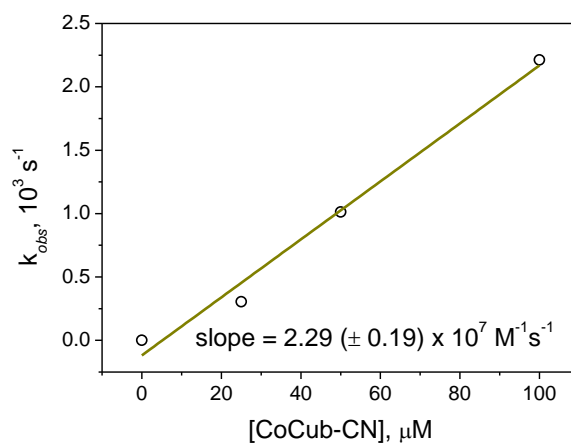
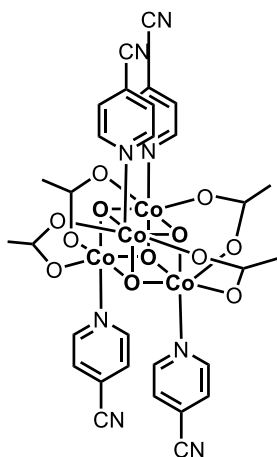
1-Br



1-COOMe

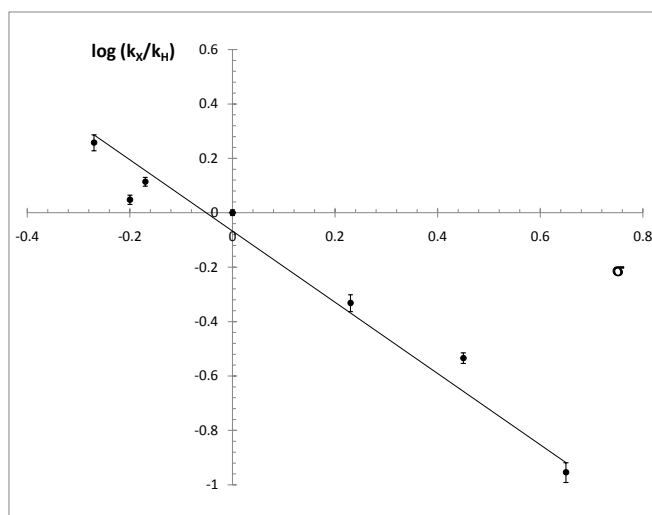


1-CN

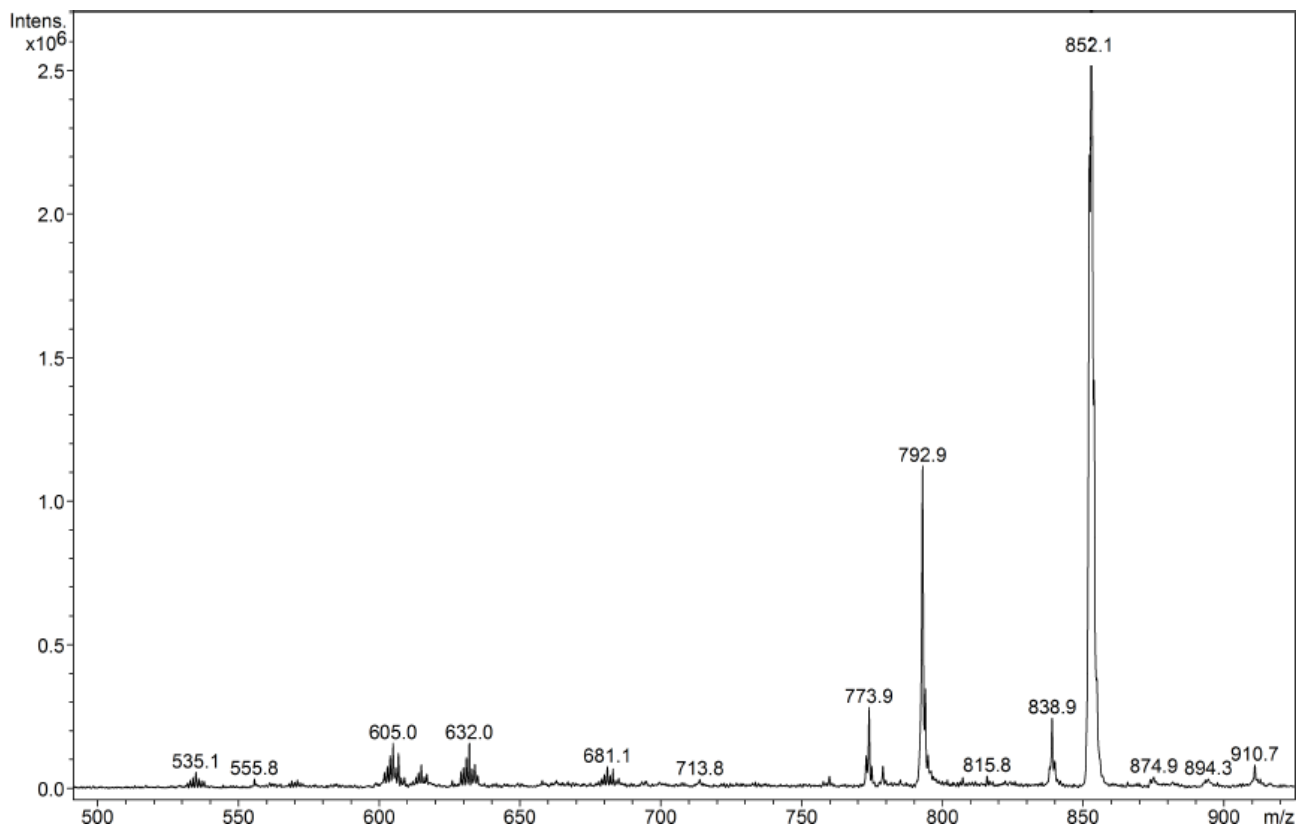


X	σ	$k_X (10^8 M^{-1}s^{-1})$	$\log(k_X / k_H)$
H	0	2.06 (± 0.05)	0
Me	-0.17	2.68 (± 0.10)	0.114
OMe	-0.27	3.73 (± 0.25)	0.258
tBu	-0.2	2.30 (± 0.09)	0.048
Br	0.23	0.96 (± 0.07)	-0.331
COOMe	0.45	0.60 (± 0.03)	-0.534
CN	0.65	0.23 (± 0.02)	-0.954

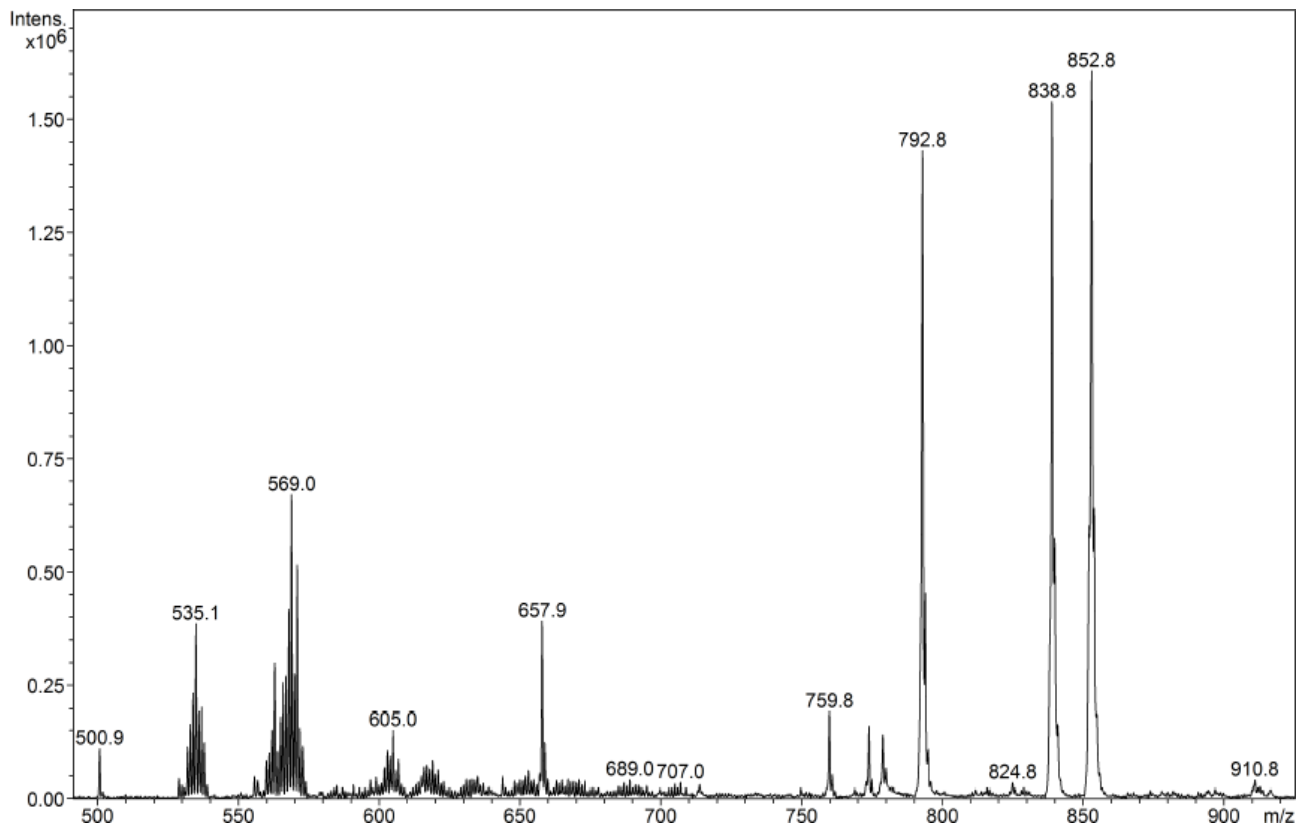
Hammett plot



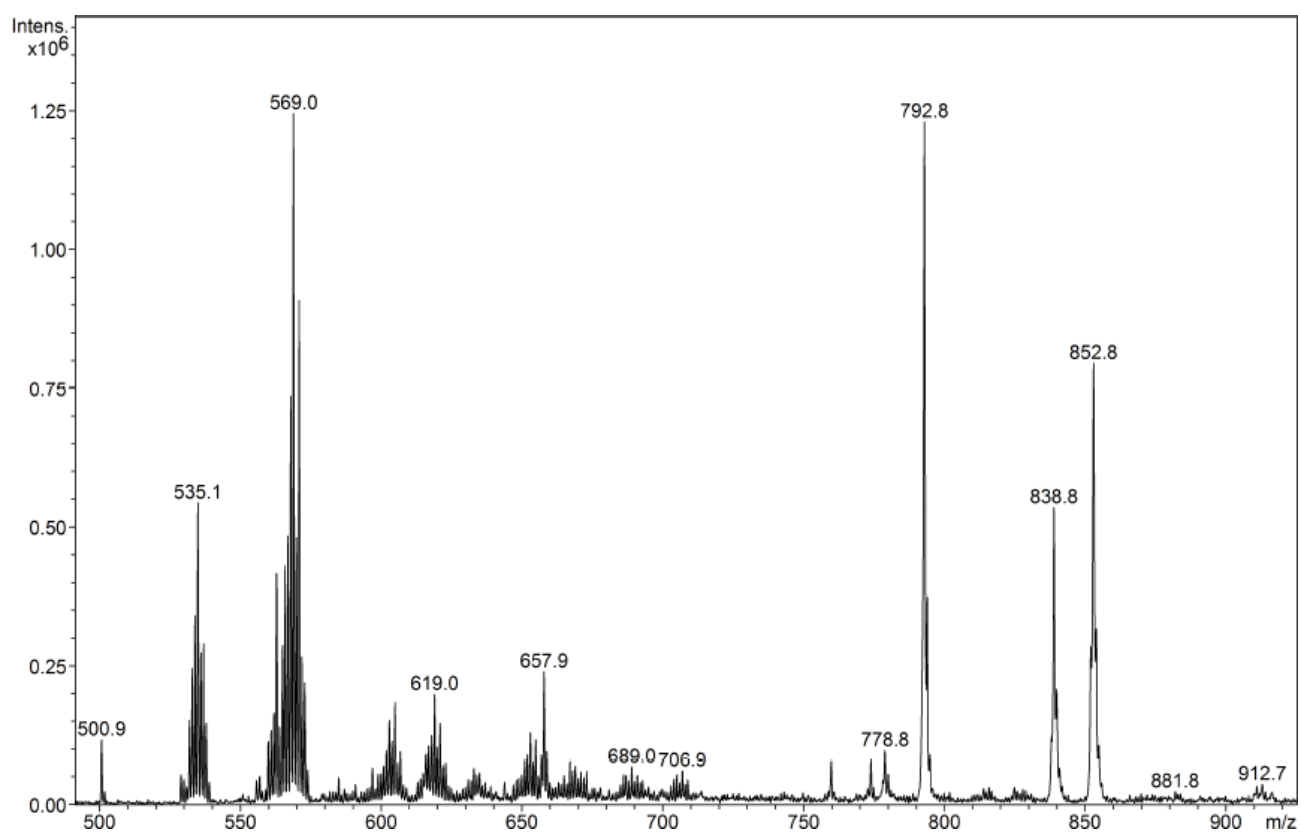
a)



b)



c)



ESI-MS spectra (5×10^{-5} M in acetonitrile, + 0.1% formic acid) of a catalytic test carried out under the conditions reported in Table 1 Entry 4, recorded on aliquots sampled at different times (respectively: a) 0 min; b) 30 min; c) 60 min.).