

SUPPLEMENTARY SECTION

Single step incorporation of carboxylic acid groups in the lower rim of calix[4]arenes: a recyclable catalyst towards assembly of diverse five ring fused acridines

Piyali Sarkar[†] and Chhanda Mukhopadhyay^{*†}

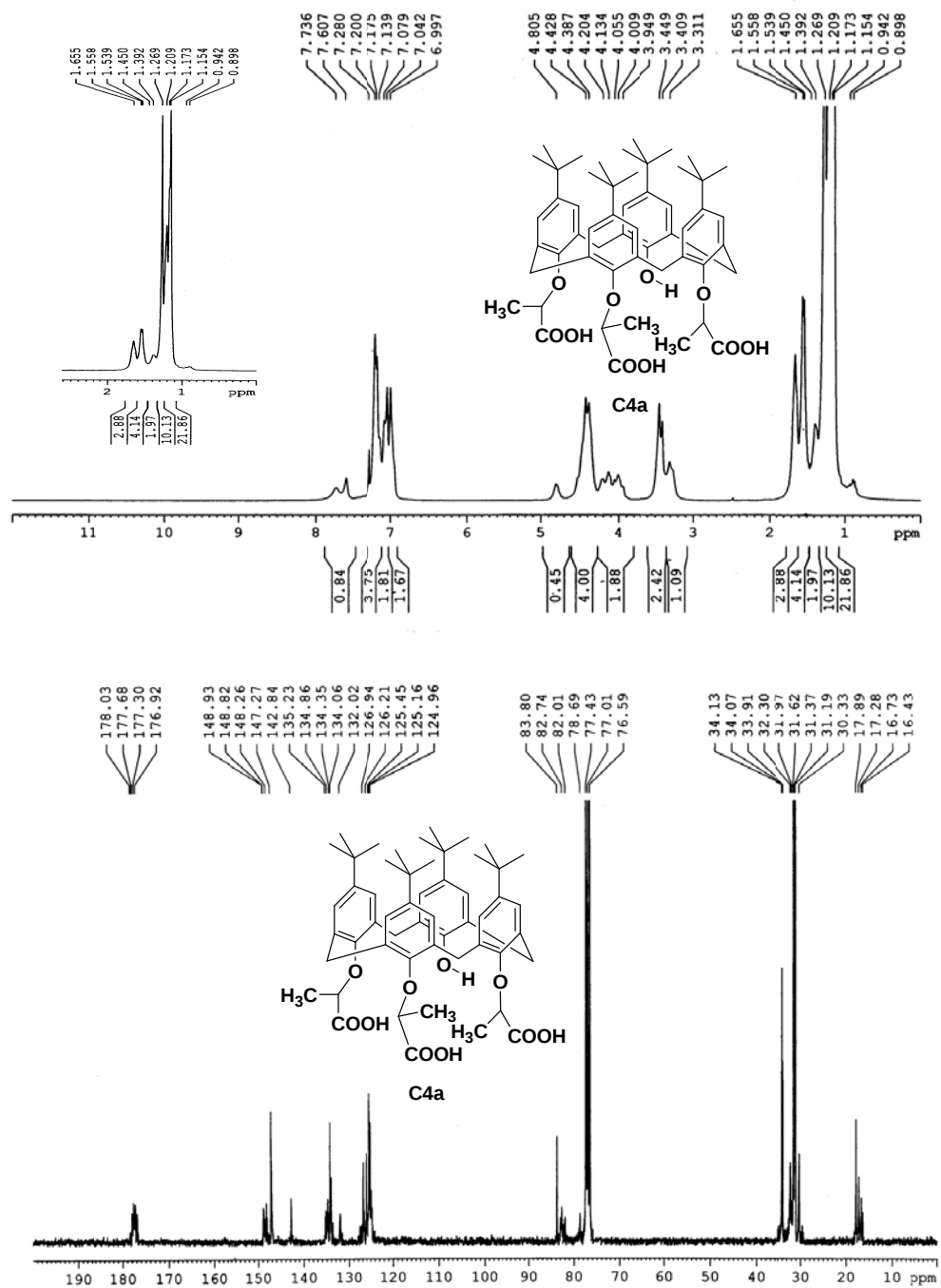
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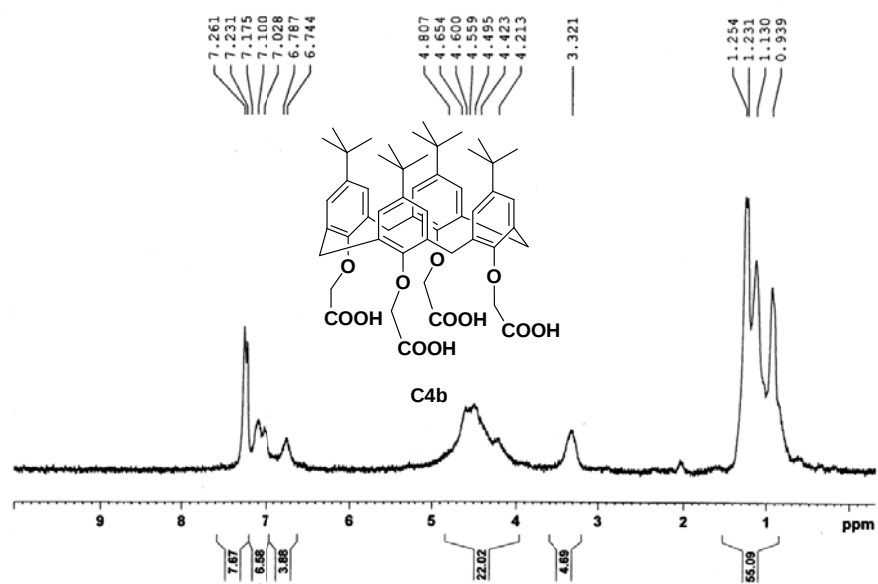
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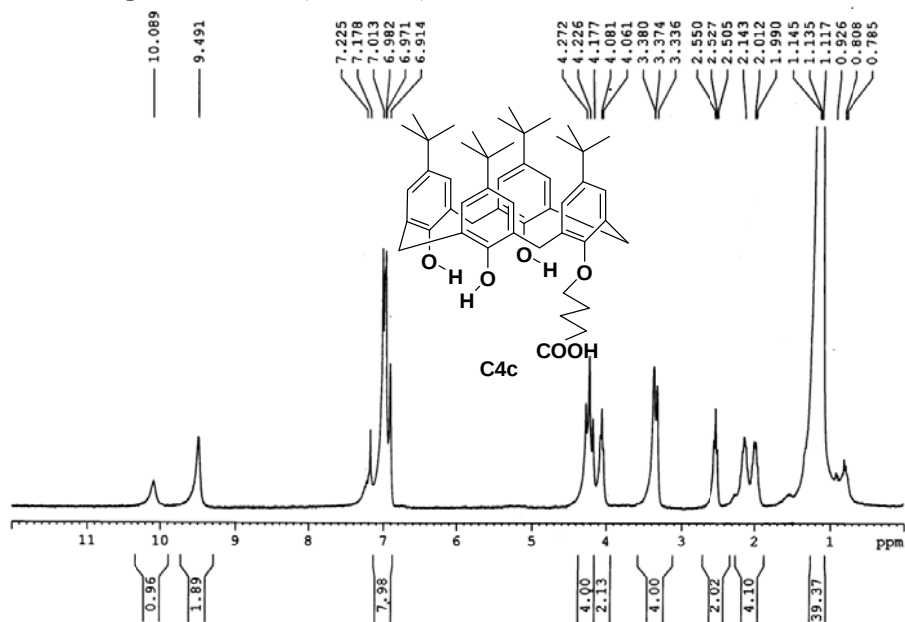
1. ^1H and ^{13}C spectra of **C4a** (scheme 1):

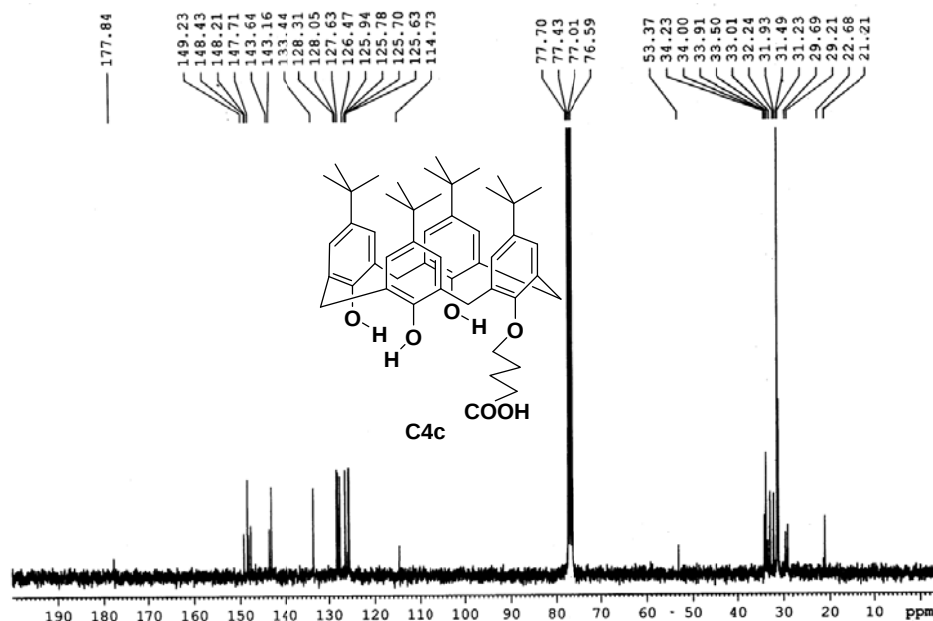


2. ^1H spectra of **C4b** (scheme 2):

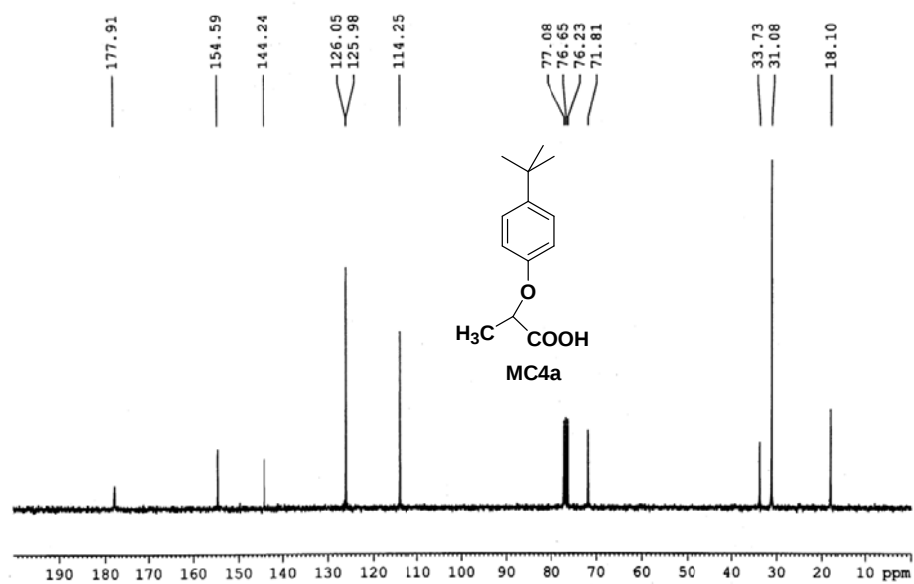
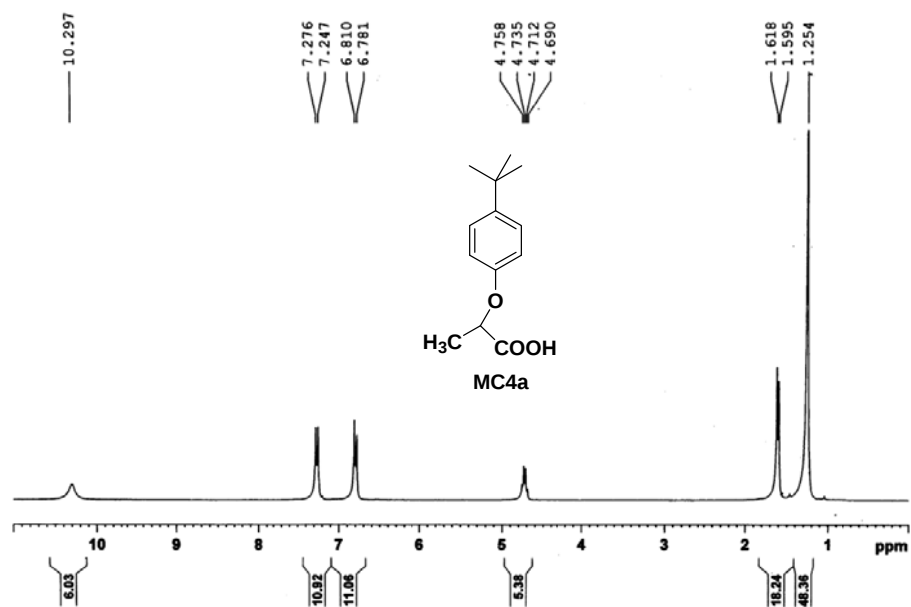


3. ^1H and ^{13}C spectra of **C4c** (scheme 2):



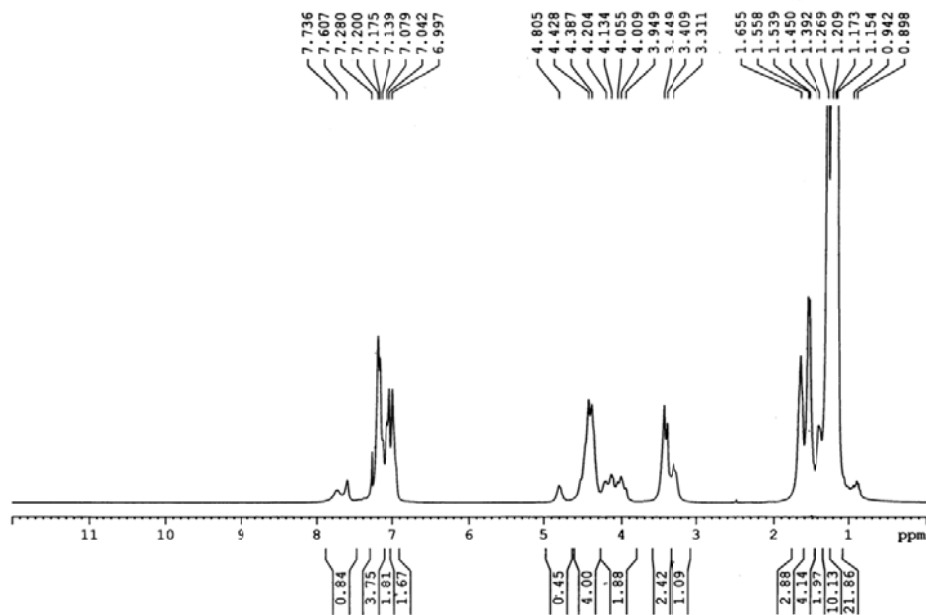


4. **Synthesis of the monomer unit of the parent calixarene (C4a), 1-(p-tert-butylphenoxy)-1-methylacetic acid (MC4a):** 4-tert-Butylphenol (150mg, 1 mmol) was mixed with NaH (48 mg, 2 mmol) in a 25 ml round bottom flask, 10 ml of dry DMF was added to it. The mixture was allowed to stir for 30 mins at room temperature (25-30 °C). 2-Bromopropionic acid (0.9 ml, 1 mmol) was added drop wise maintaining the temperature at 20-25 °C. After the complete addition the reaction mixture was stirred for 4 h. As the reaction progressed, the suspended particles changed into a form of viscous mass. The disappearance of starting 4-tert-Butylphenol was checked by TLC. After complete conversion the reaction mixture was placed in an ice bath and was quenched via acidification using 5% HCl solution. Then 20 ml of chilled water was added to it and we got the crude product by filtration. 210 mg (95%) pure product, color less solid was gained via column chromatography at 10% ethyl acetate in petroleum ether as eluent. mp. 64 °C; IR (KBr, ν cm^{-1}): 3394, 2963, 2865, 1738, 1654, 1362, 1199, 1068. ^1H NMR (300 MHz, CDCl_3): δ 10.30 (br s., 1H, -COOH), 7.26 (d, J = 8.7 Hz, 2H, Ar-H), 6.80 (d, J = 8.7 Hz, 2H, Ar-H), 4.72 (q, J = 6.8 Hz, 1H, -CH), 1.61 (d, J = 6.9 Hz, 3H, -CH₃), 1.25 (s, 9H, -CH₃); ^{13}C NMR (75 MHz, CDCl_3): δ 177.9, 154.6, 144.2, 126.1, 126.0, 114.2, 71.8, 33.7, 31.1, 18.1; Anal. calcd. for $\text{C}_{13}\text{H}_{18}\text{O}_3$; C: 70.24; H: 8.16. Found: C: 70.29; H: 8.15.

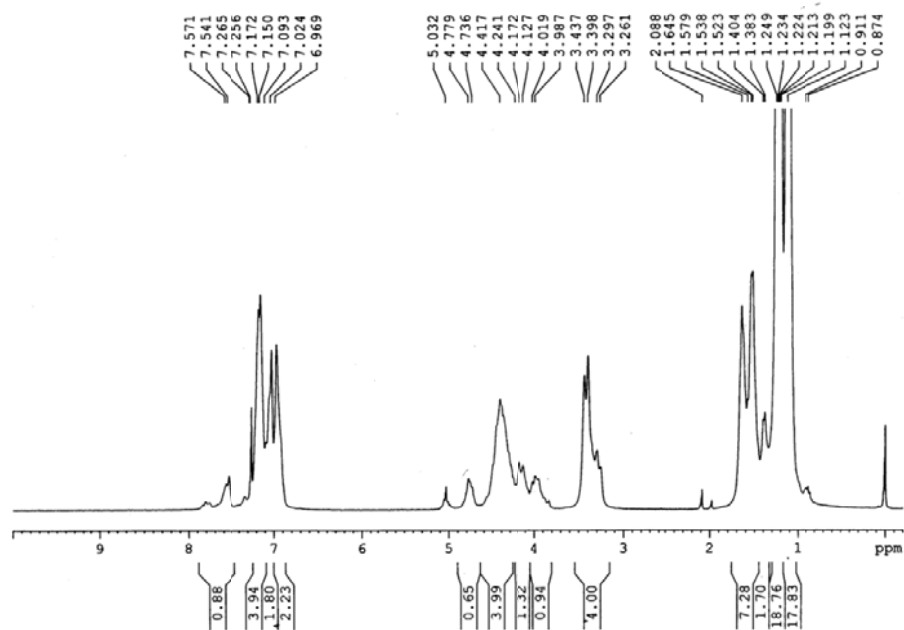


5. The ^1H NMR of the catalyst after each 2 cycles :

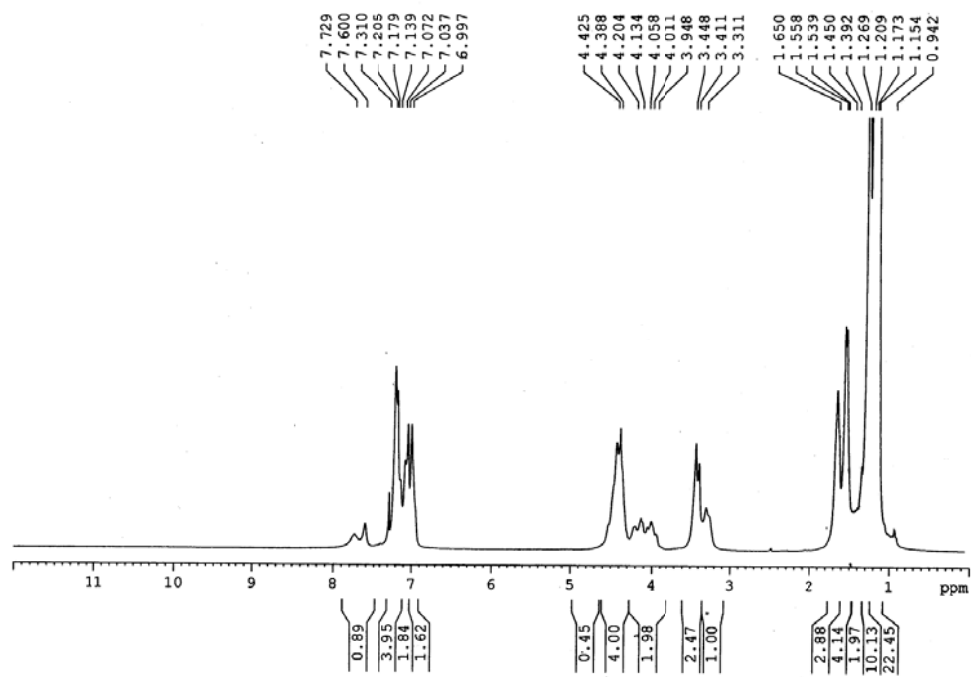
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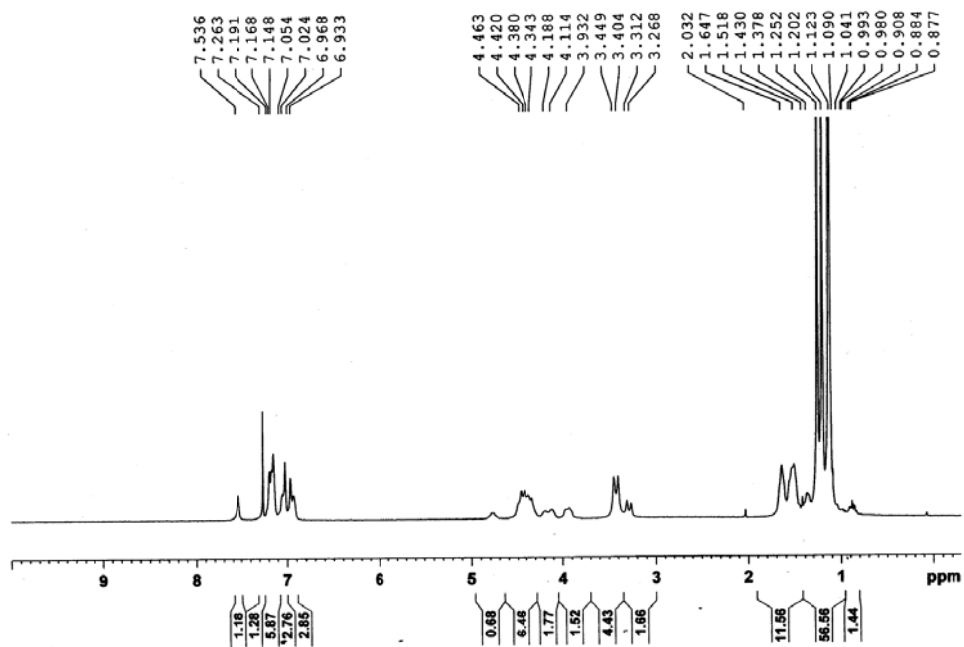
After 2nd run:



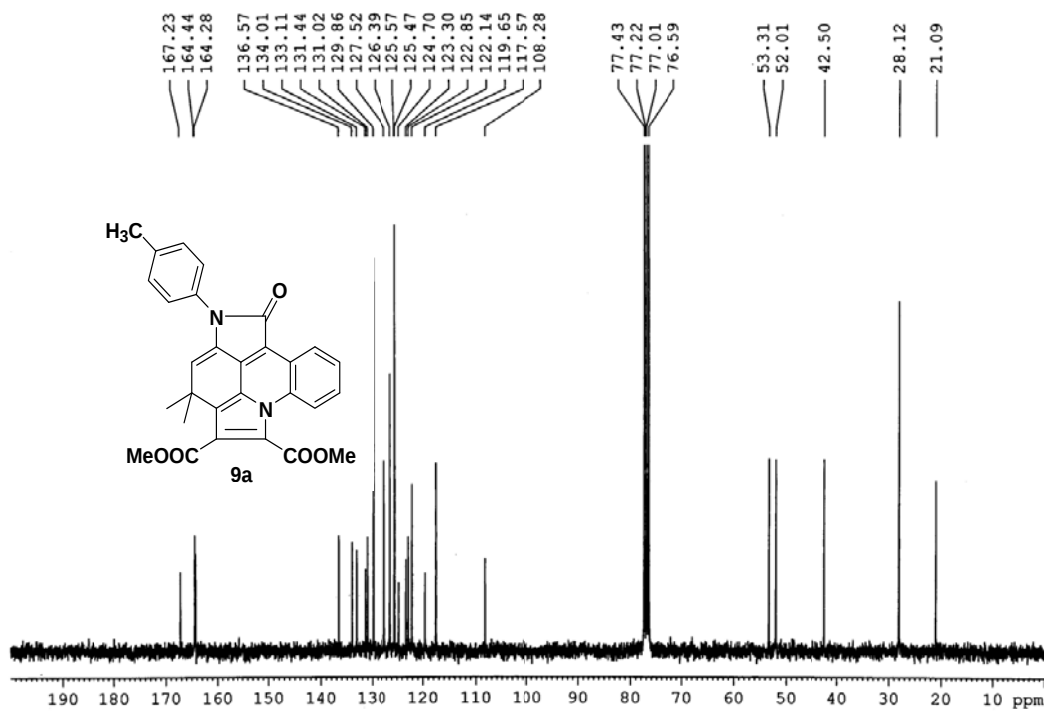
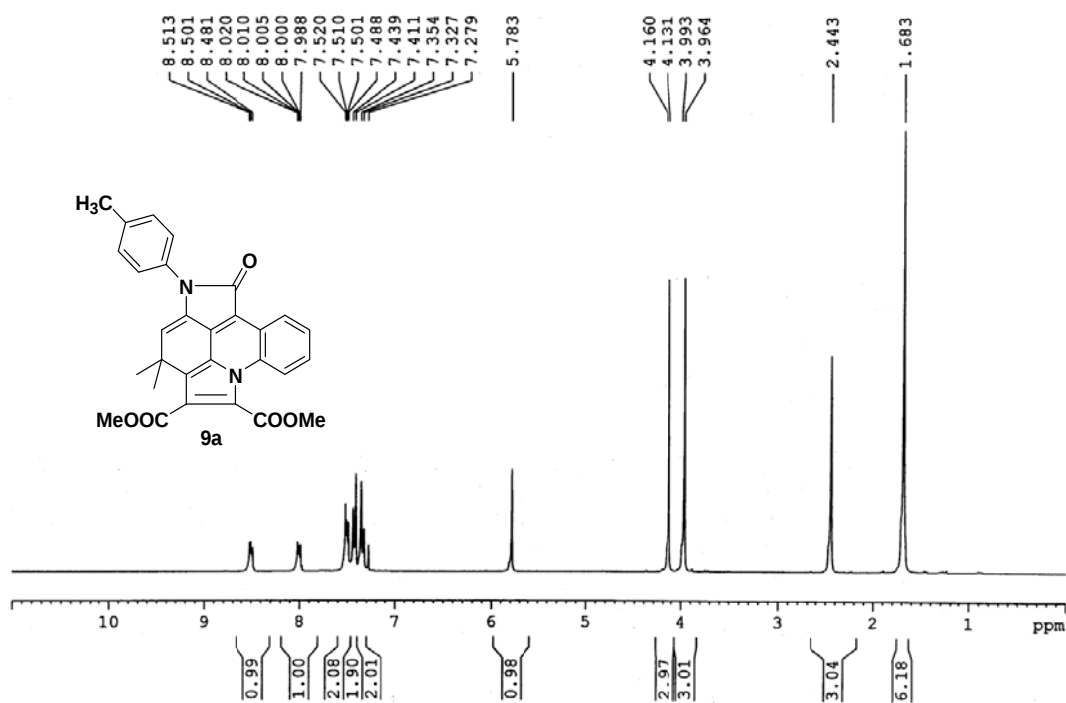
After 4th run:

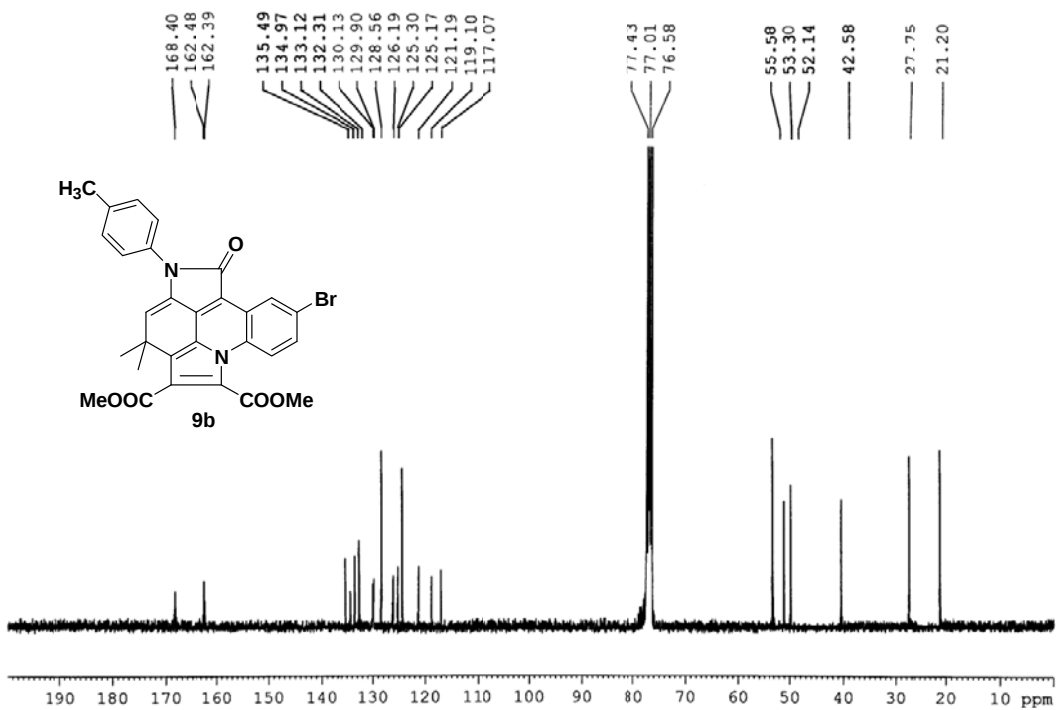
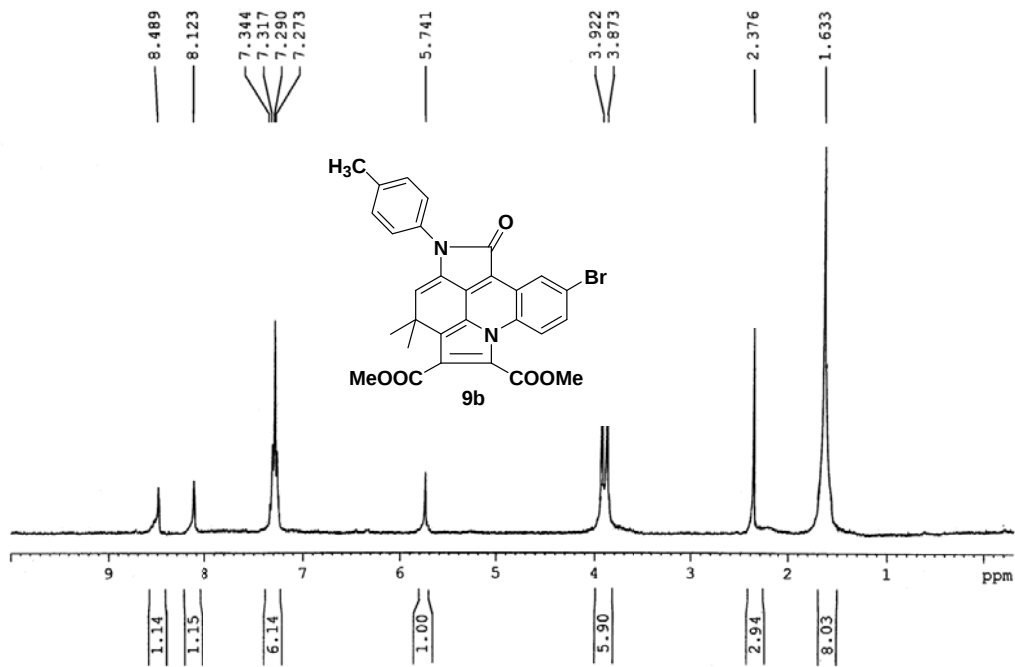


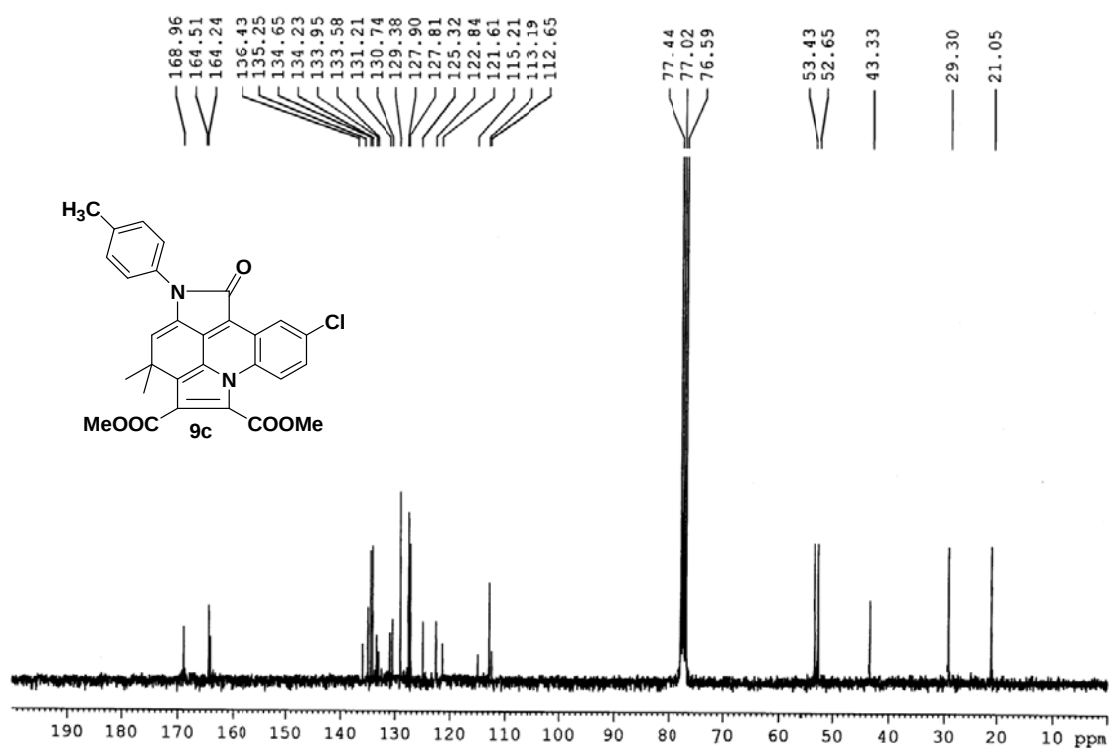
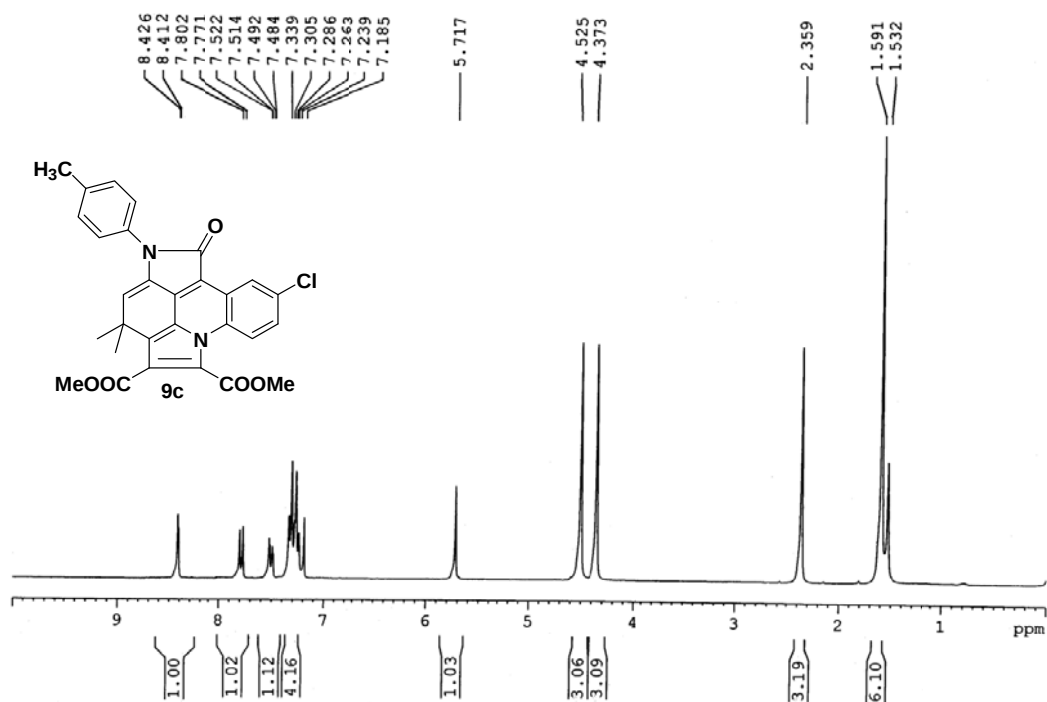
After 6th run:

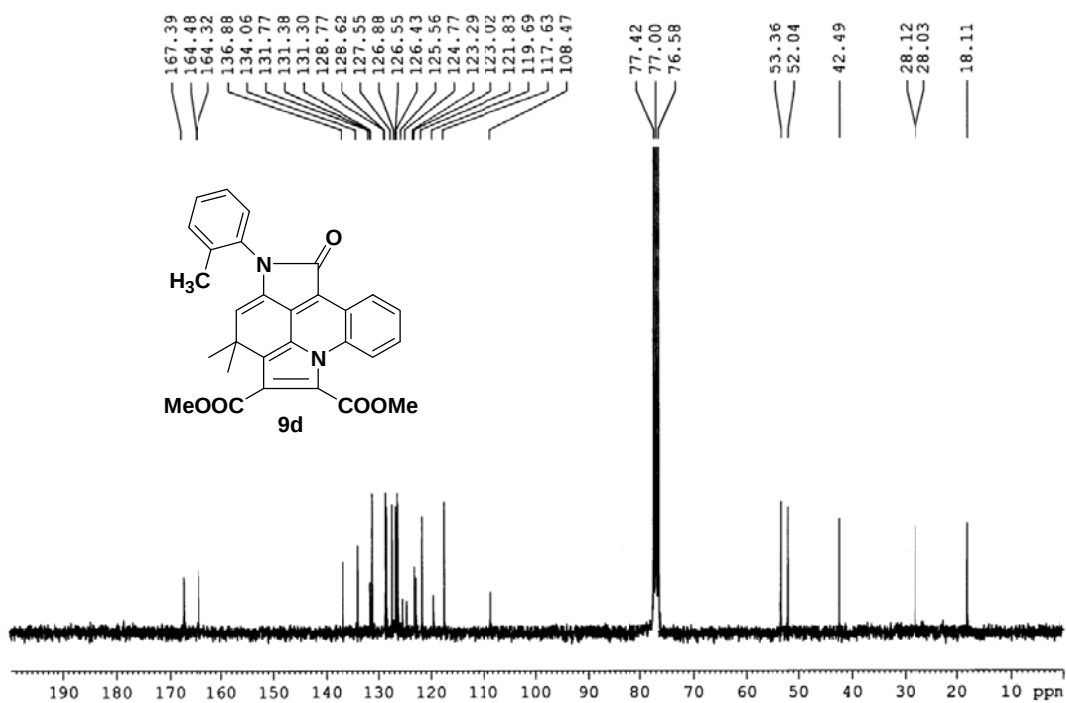
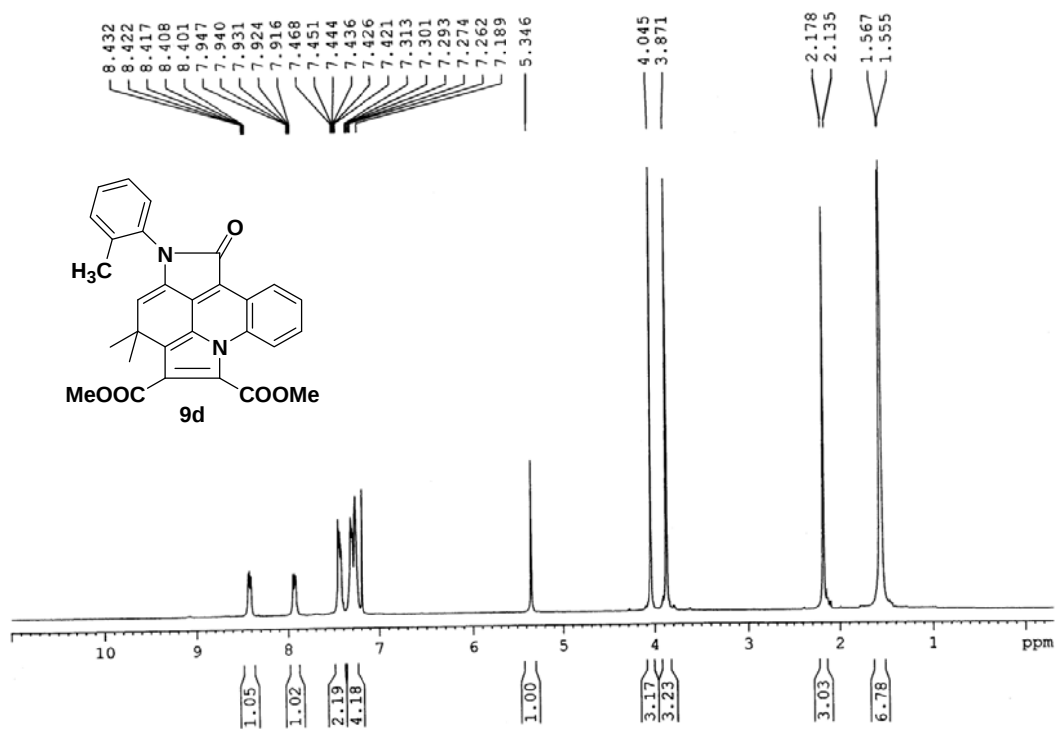


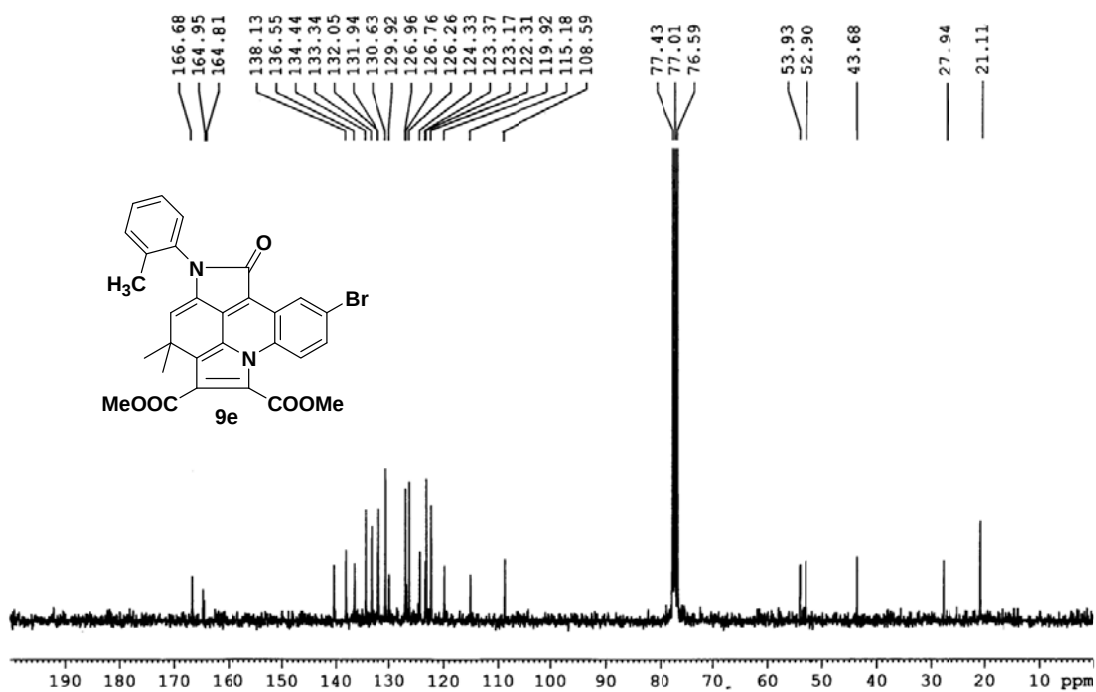
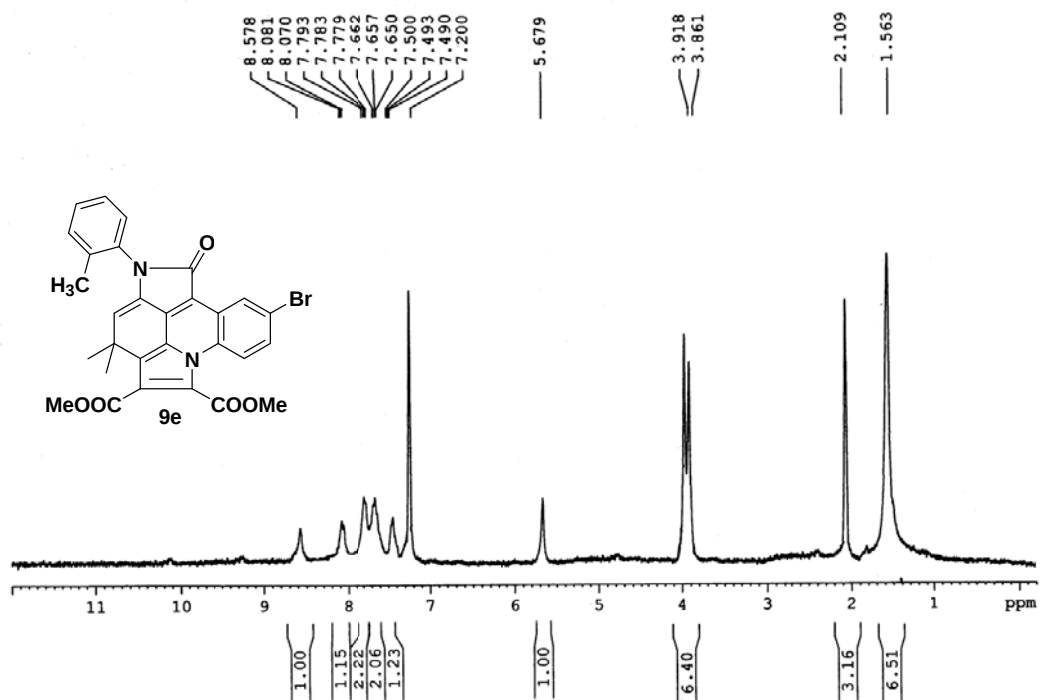
6. ^1H and ^{13}C spectra of **9a-9z** and **9a'-9e'** (scheme 3):

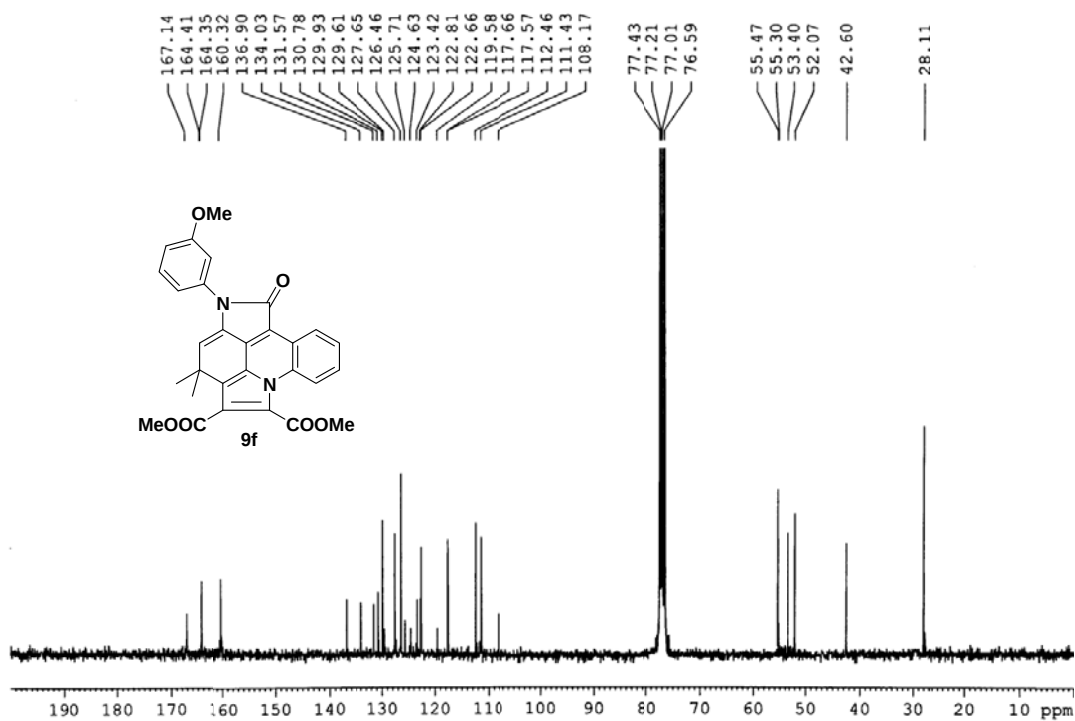
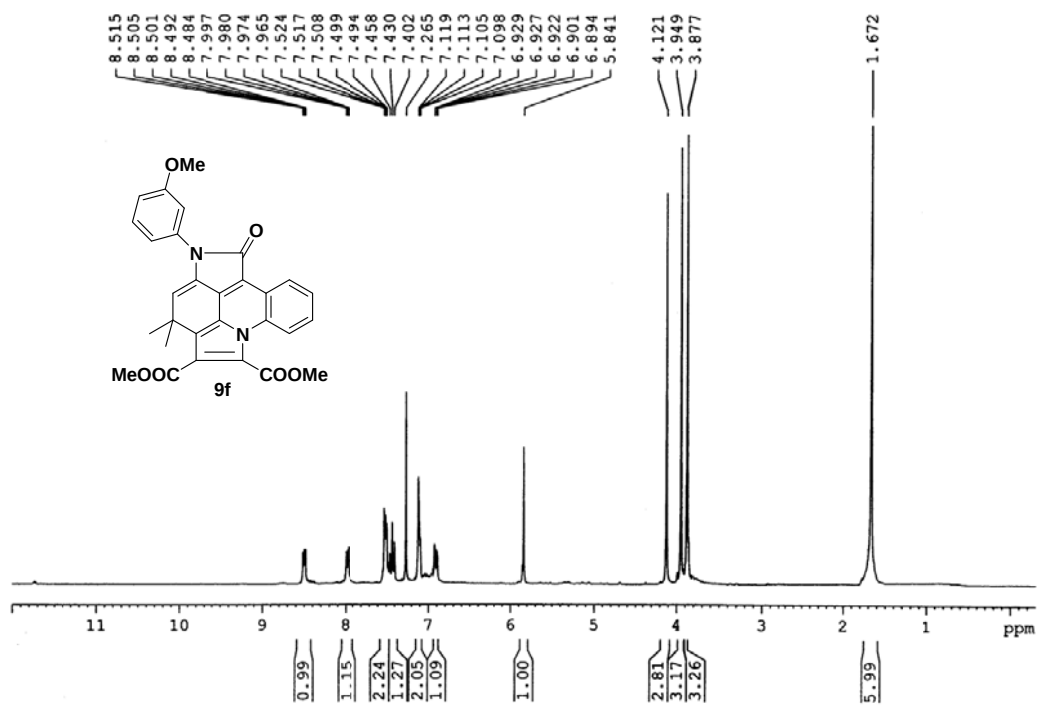


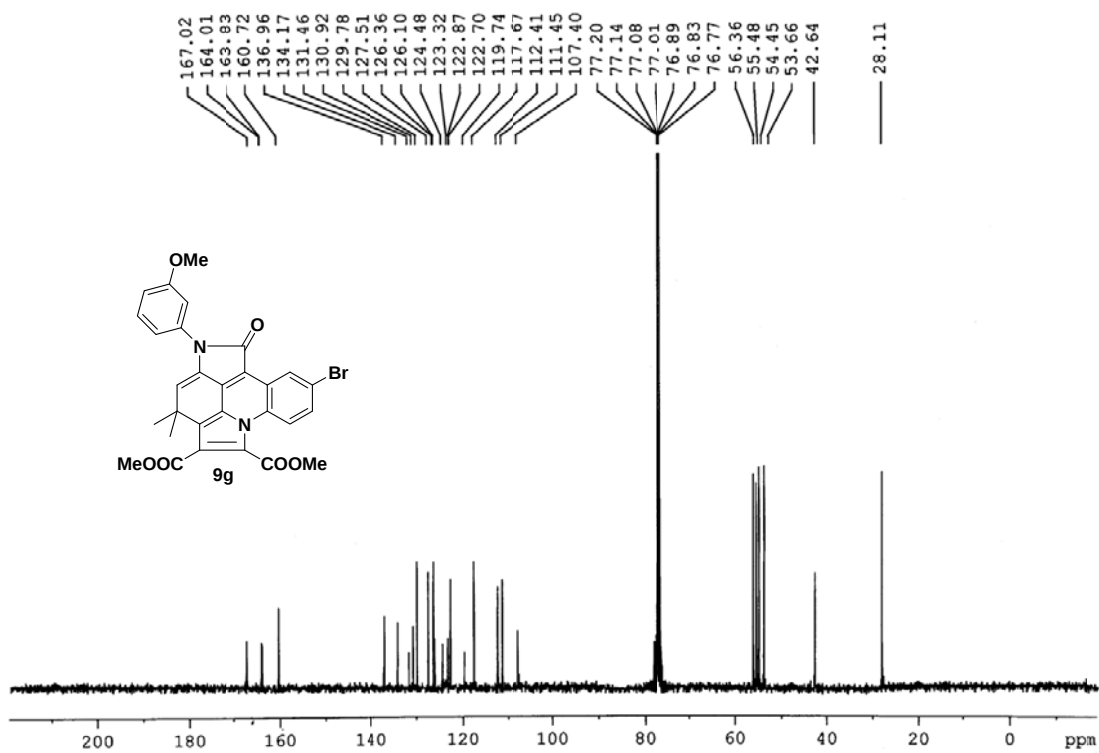
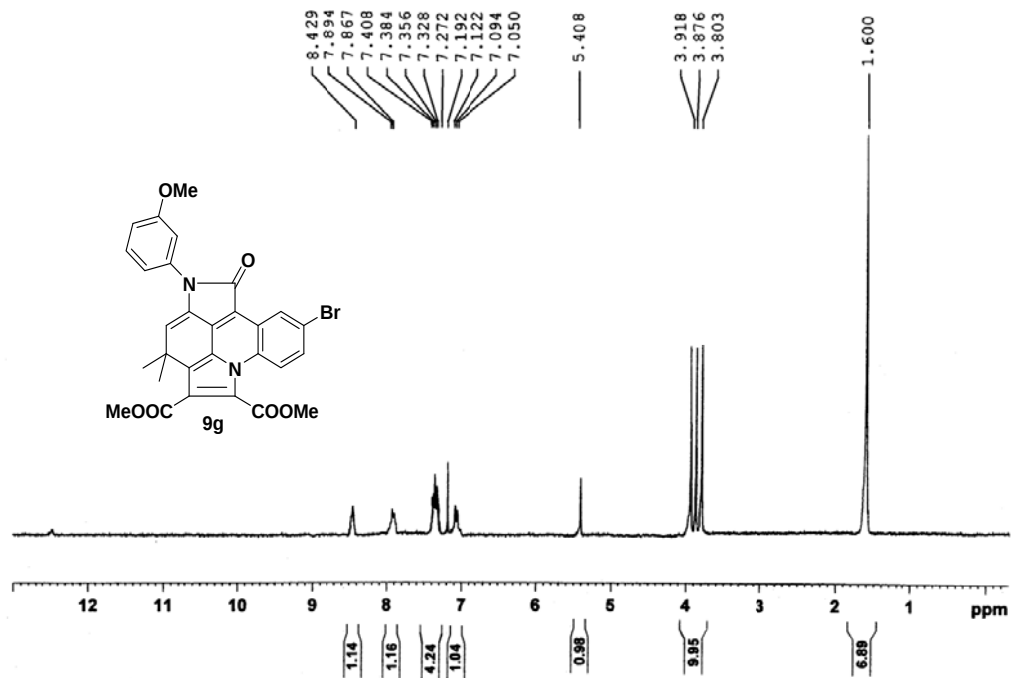


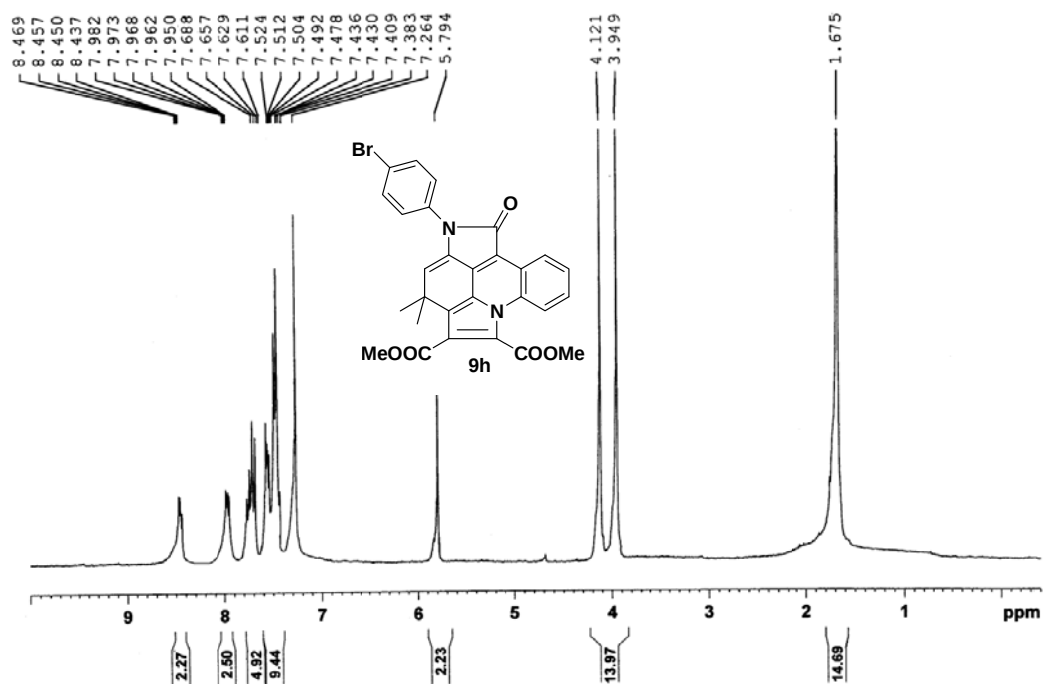




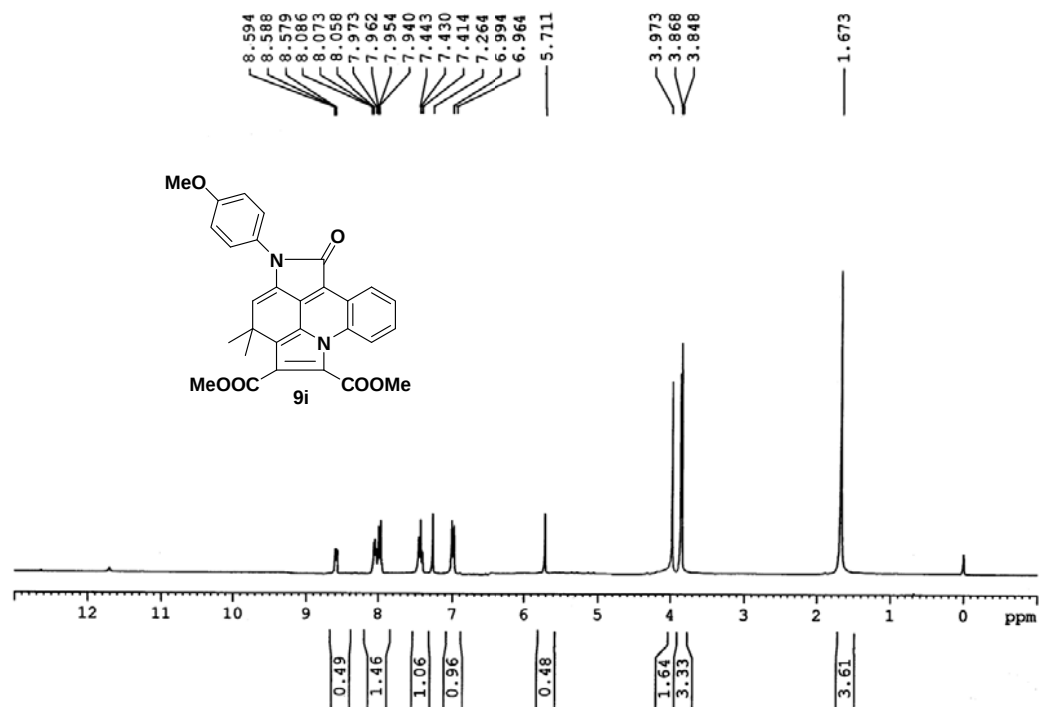


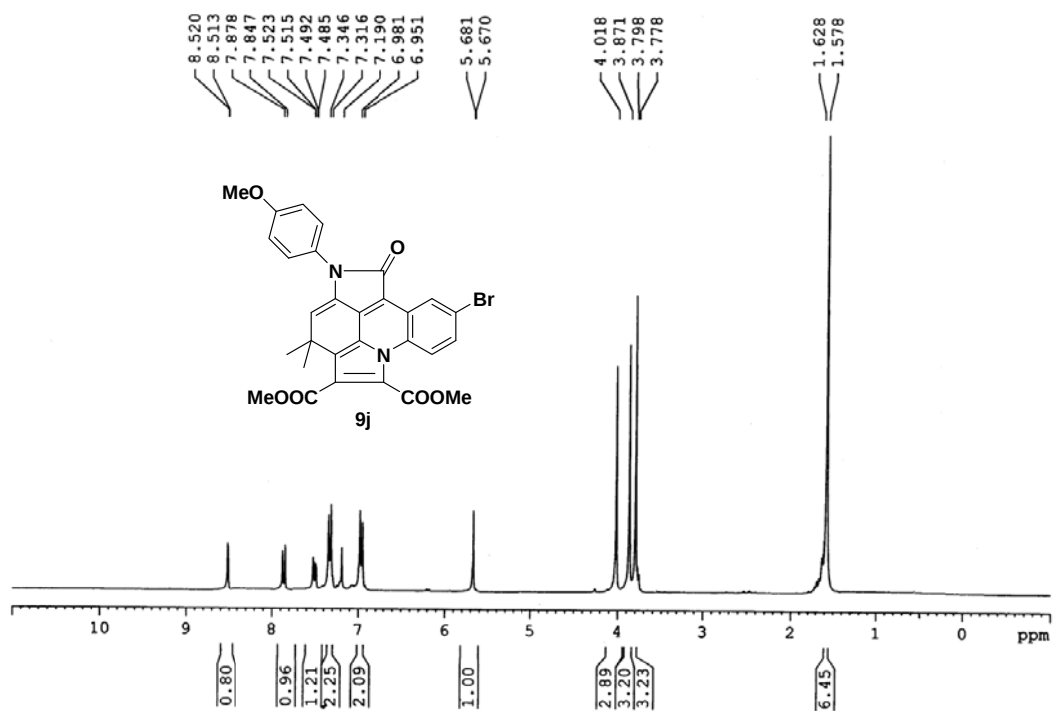
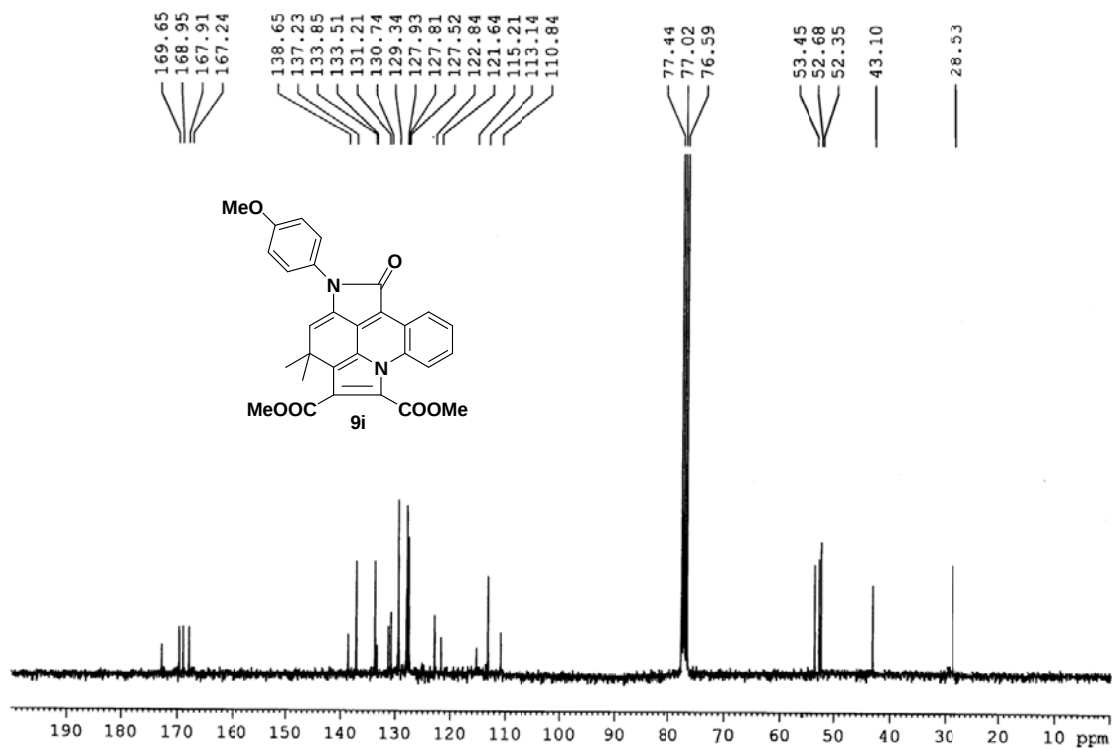


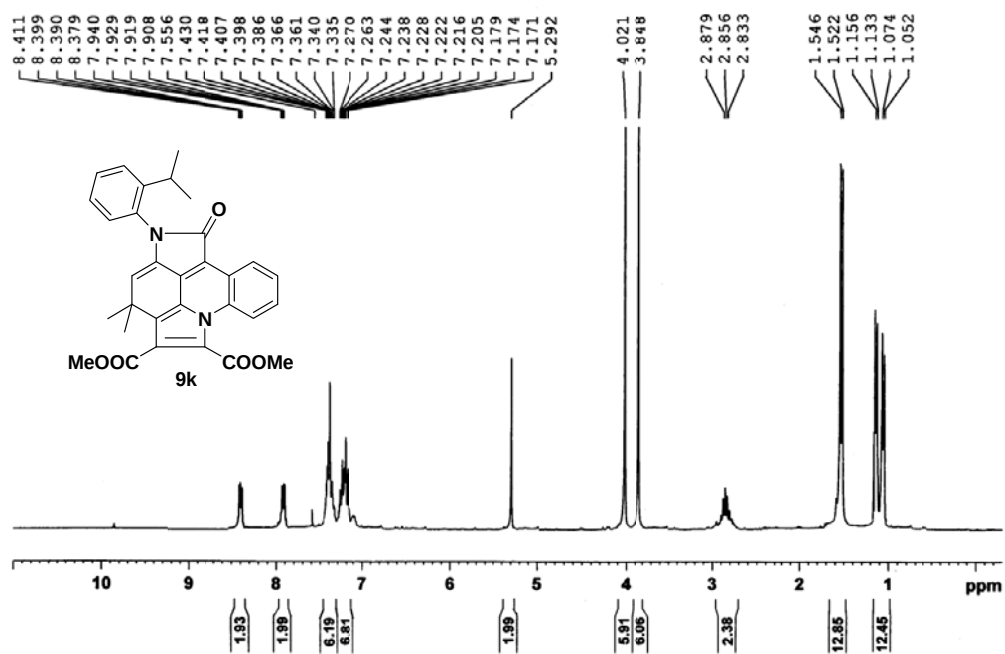
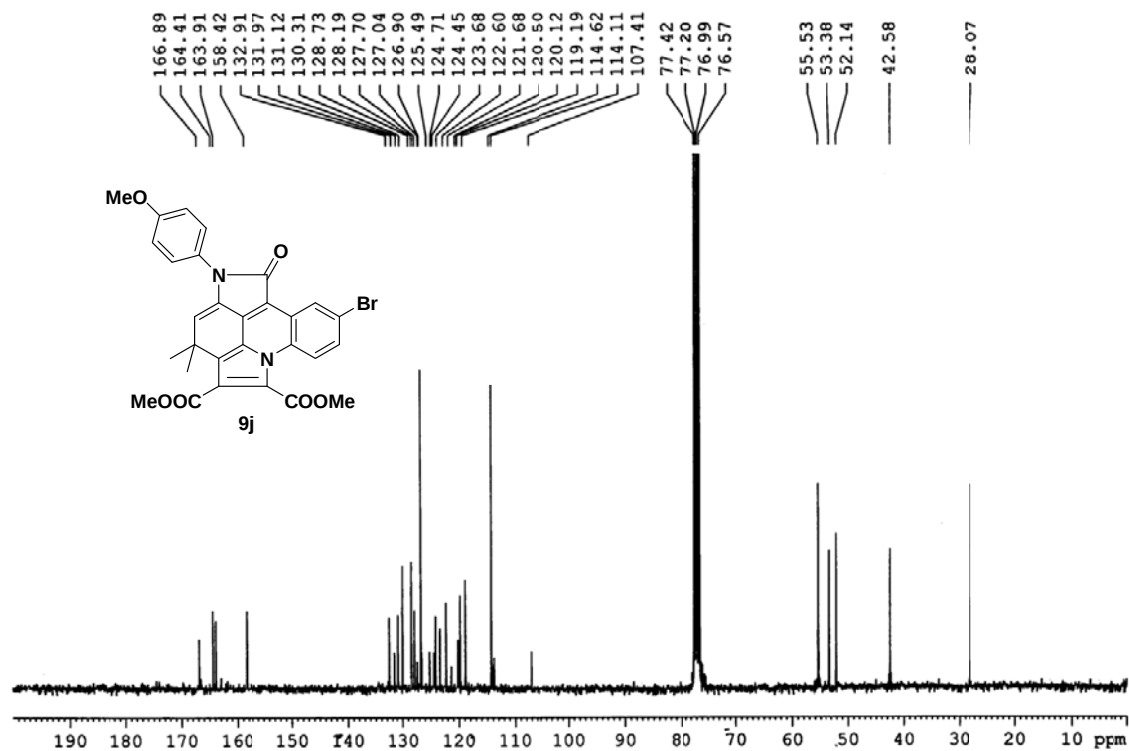


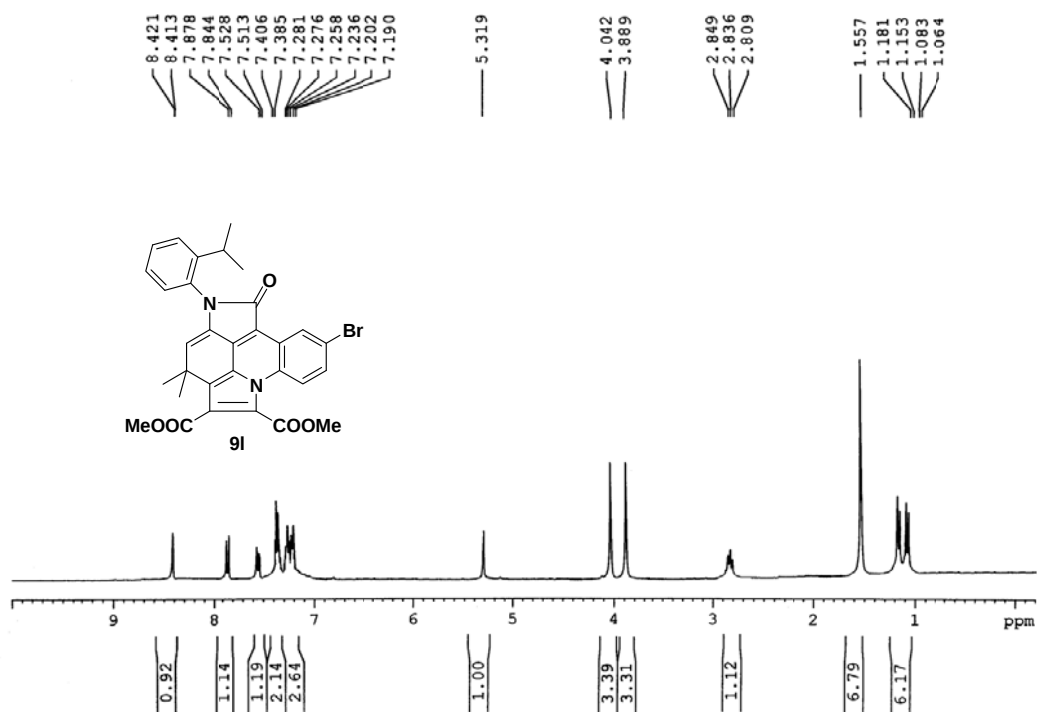
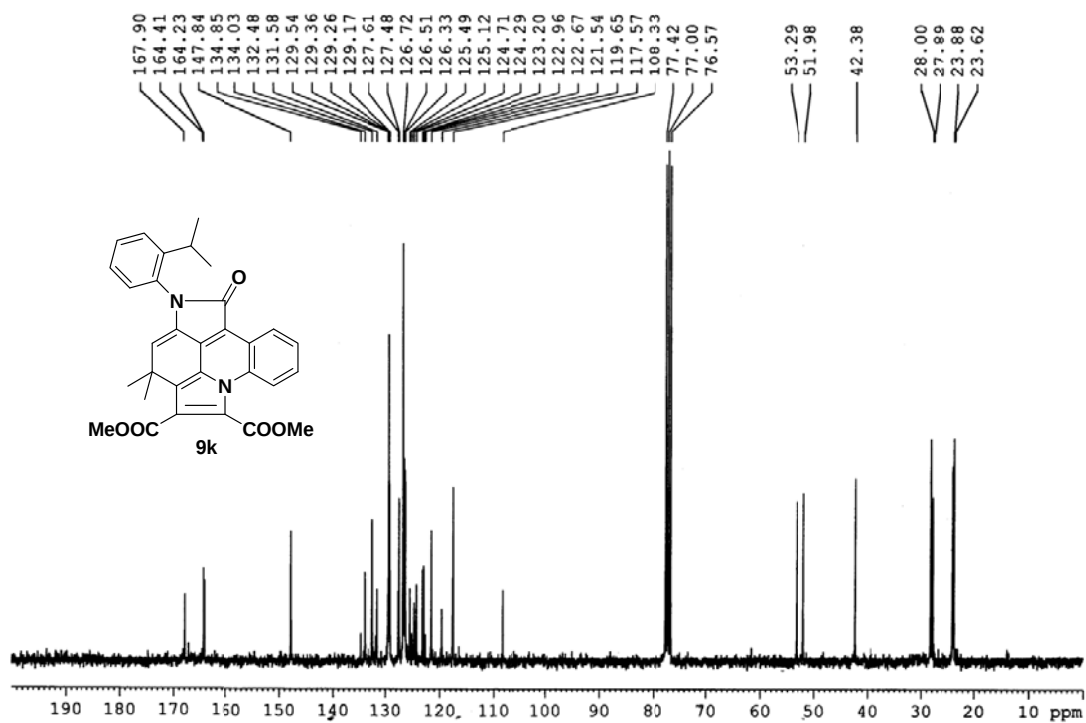


The solubility of this compound, 9h is too low to collect the ¹³C data. We got the ¹H-NMR after several scans.









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