

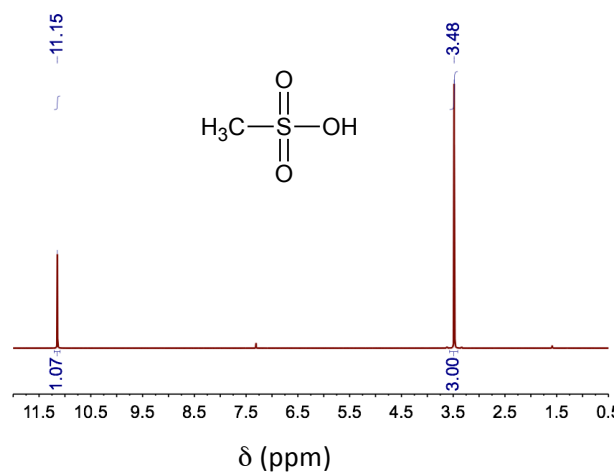
Supplementary Information of the Manuscript entitled

## Eutectic Mixtures as Bifunctional Catalysts in the Low-Temperature-Synthesis of Polycaprolactone

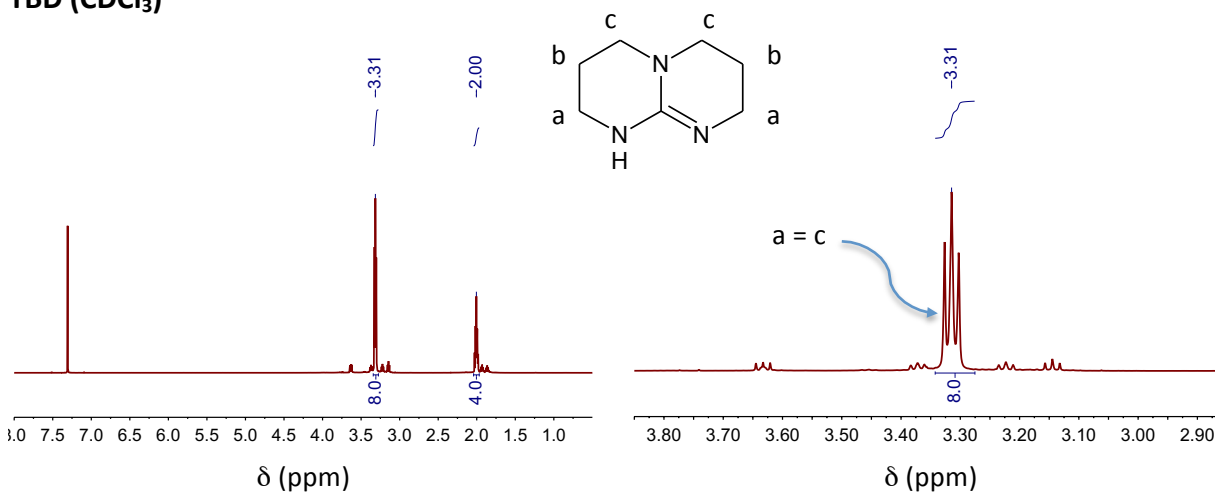
by Sara García-Argüelles et al.

**Figure S1:**  $^1\text{H}$  NMR spectra of the components that form the DESs (e.g. TBD and  $\text{MeSO}_3\text{H}$ ) and of the TBD: $\text{MeSO}_3\text{H}$  mixtures with 0.1:1.5 and 0.5:1.5 molar ratios.

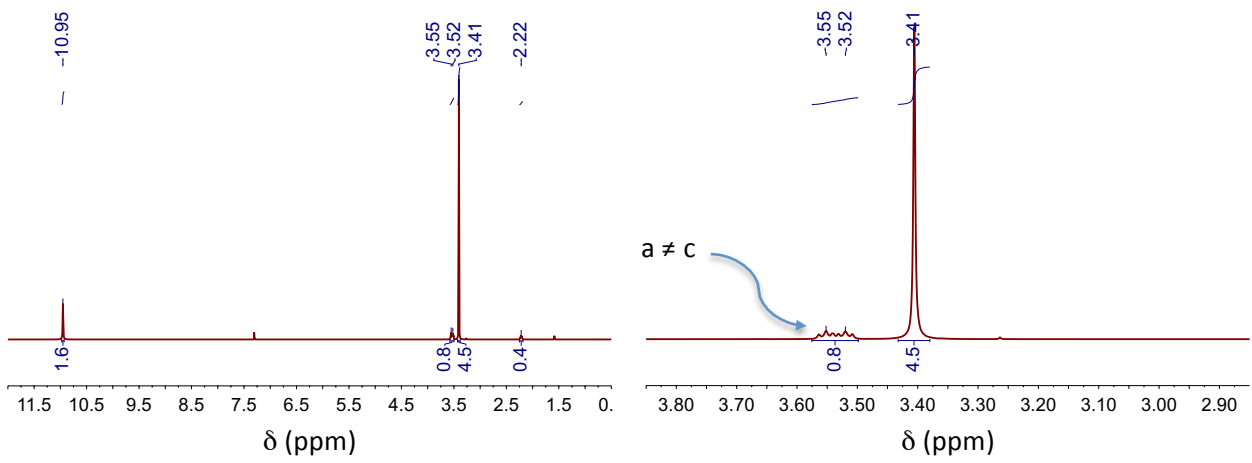
### $\text{MeSO}_3\text{H}$ ( $\text{CDCl}_3$ )



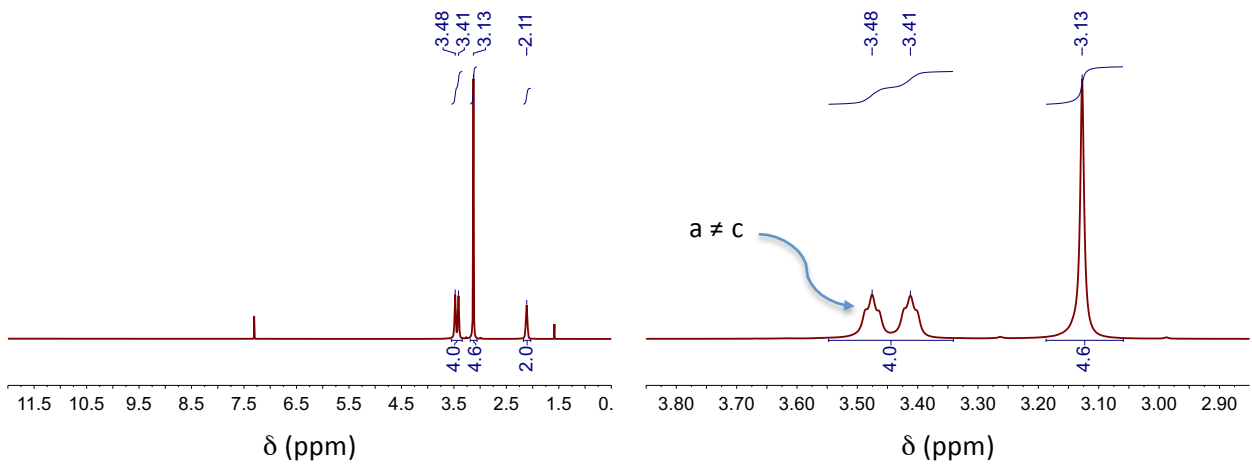
### TBD ( $\text{CDCl}_3$ )



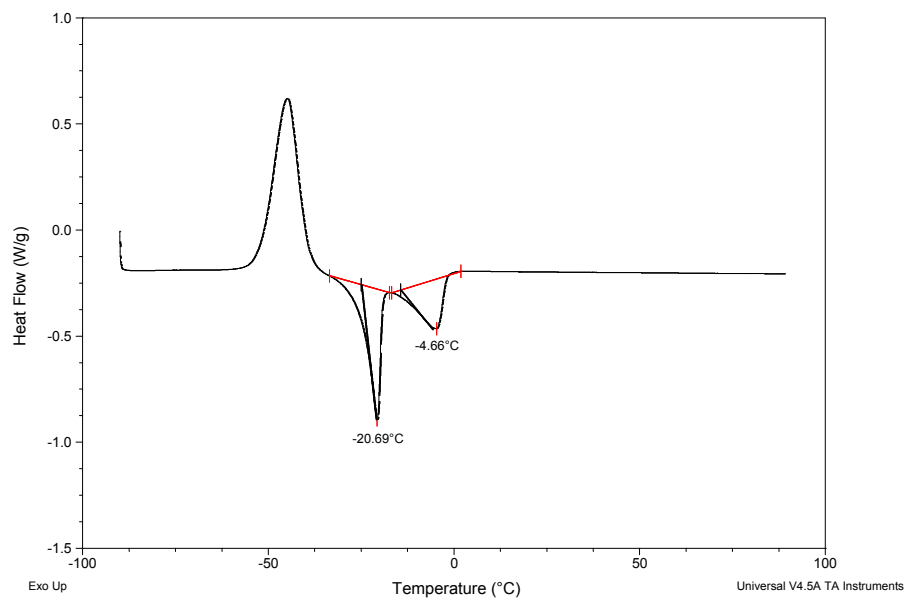
### DES 0.1:1.5



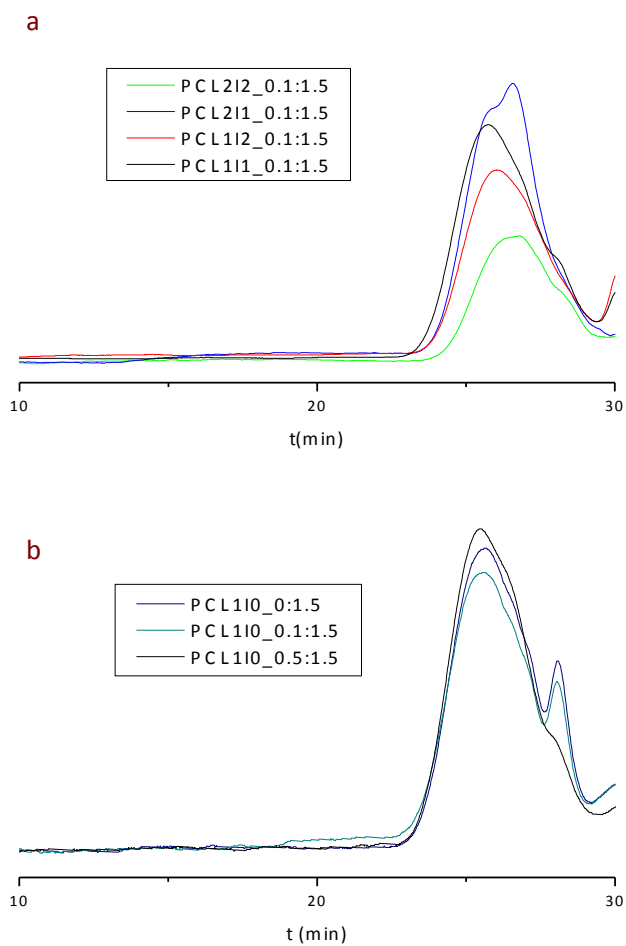
### DES 0.5:1.5



**Figure S2:** DSC scan of the eutectic TBD:MeSO<sub>3</sub>H mixture with a molar ratio of 0.05:1.5.

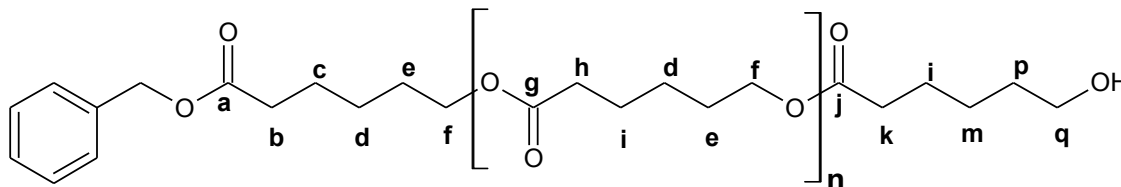


**Figure S3:** GPC separation of the PCLs obtained in this work with initiator (a) and without initiator (b).



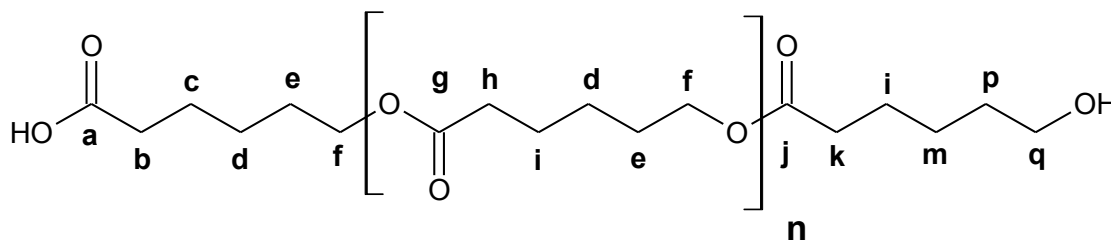
**Table S1:**  $^1\text{H}$  NMR chemical shifts ( $\delta$ , in ppm) of different PCLs obtained with initiator.

Set 1	e-caprolactone	Polycaprolactone						Initiator	
	$-\text{CO}-\text{O}-\text{CH}_2-$	$-\text{CO}-\text{O}-\text{CH}_2-$ (f)	$\text{H}-\text{O}-\text{CH}_2-$ (q)	$\text{Ph}-\text{CH}_2-\text{O}-\text{CO}-\text{CH}_2-$ (b)	$-\text{O}-\text{CO}-\text{CH}_2-$ (h)	$-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{O}-$ (d,e)	$-\text{O}-\text{CO}-\text{CH}_2-\text{CH}_2-$ (l)	$\text{Ph}-\text{CH}_2-\text{O}-\text{CO}-$	$\text{Ph}-\text{CH}_2-\text{O}-\text{CO}-$
PCL1I1_0.1:1.5	4.25	4.09	3.68	2.39	2.34	1.68	1.42	7.38	5.15
PCL1I2_0.1:1.5	4.26	4.10	3.69	2.40	2.34	1.69	1.42	7.39	5.15
PCL2I1_0.1:1.5	4.24	4.09	3.68	2.39	2.34	1.69	1.42	7.38	5.15
PCL2I2_0.1:1.5	4.26	4.10	3.69	2.40	2.34	1.69	1.42	7.39	5.15



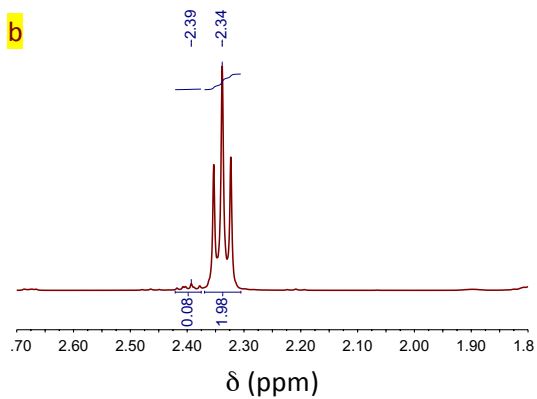
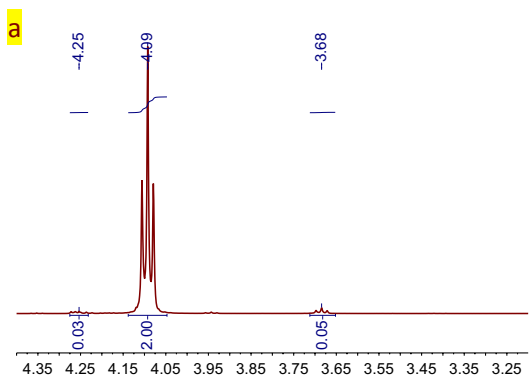
**Table S2:**  $^1\text{H}$  NMR chemical shifts ( $\delta$ , in ppm) of different PCLs obtained without initiator.

Set 2	e-caprolactone	Polycaprolactone					
	$-\text{CO}-\text{O}-\text{CH}_2-$	$-\text{CO}-\text{O}-\text{CH}_2-$ (f)	$\text{H}-\text{O}-\text{CH}_2-$ (q)	$\text{H}-\text{O}-\text{CO}-\text{CH}_2-$ (b)	$-\text{O}-\text{CO}-\text{CH}_2-$ (h)	$-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{O}-$ (d,e)	$-\text{O}-\text{CO}-\text{CH}_2-\text{CH}_2-$ (l)
PCL1I0_0.1:1.5	4.26	4.10	3.69	2.40	2.36	1.69	1.42
PCL1I0_0.5:1.5	4.24	4.10	3.69	2.40	2.34	1.69	1.42
PCL1I0_0:1.5	4.26	4.10	3.69	2.40	2.34	1.69	1.42

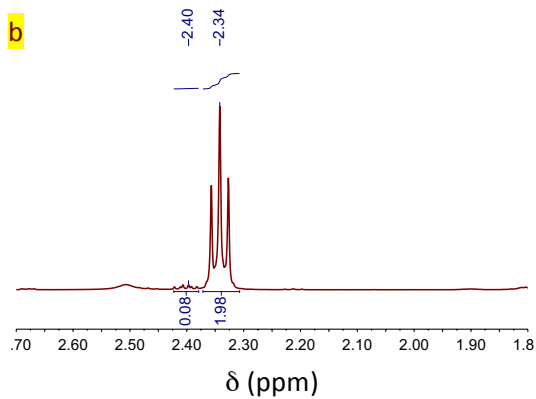
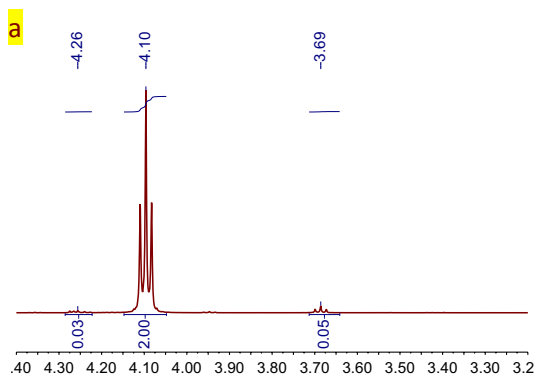


**Figure S4:** Details of the  $^1\text{H}$  NMR spectra of different PCLs obtained with initiator depicted in Figure S2 for better visualization of peaks ascribed to terminal methylene groups; (a) at ca. 3.7 ppm for those bonded to hydroxyl groups and (b) at ca. 2.4 ppm for those bonded to carboxylic groups.

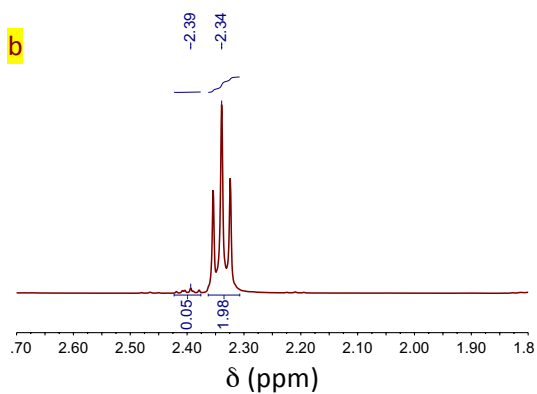
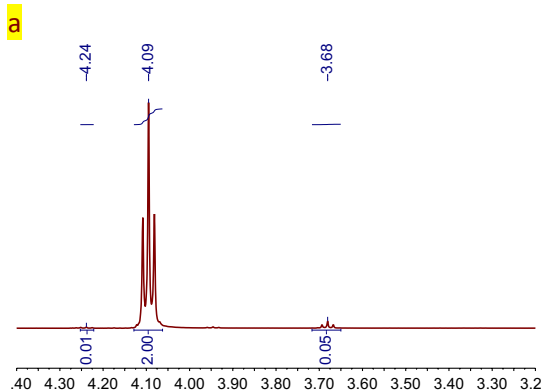
**PCL111\_0.1:1.5**



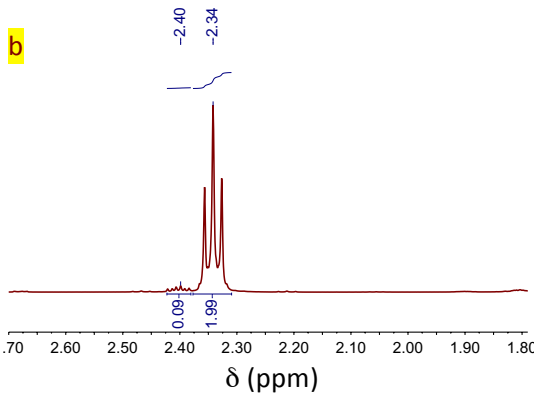
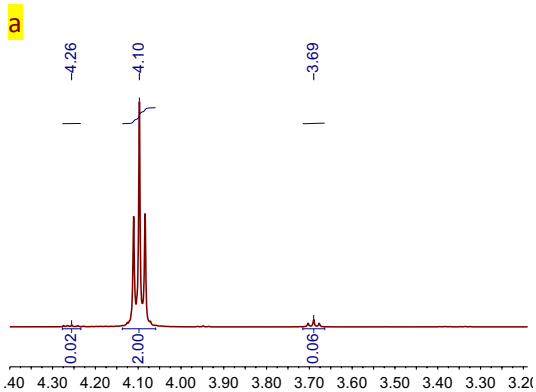
**PCL112\_0.1:1.5**



**PCL211\_0.1:1.5**

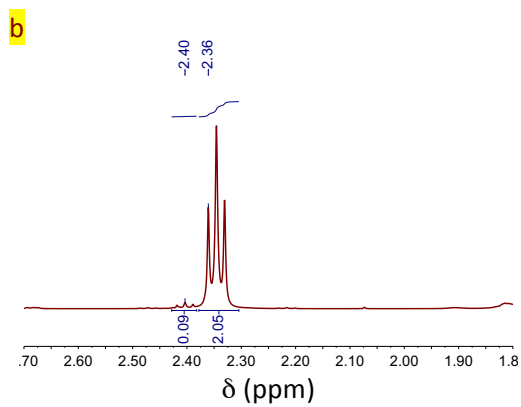
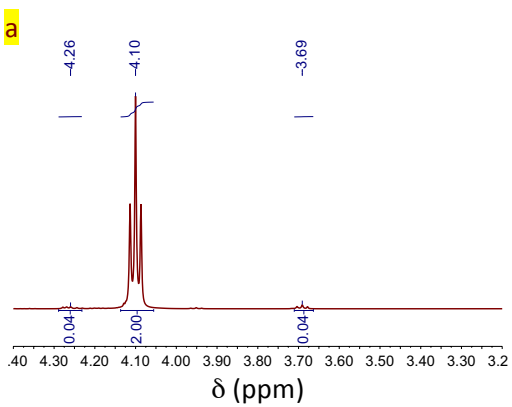


**PCL212\_0.1:1.5**

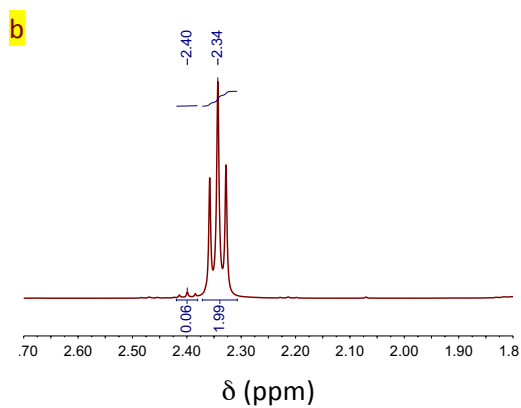
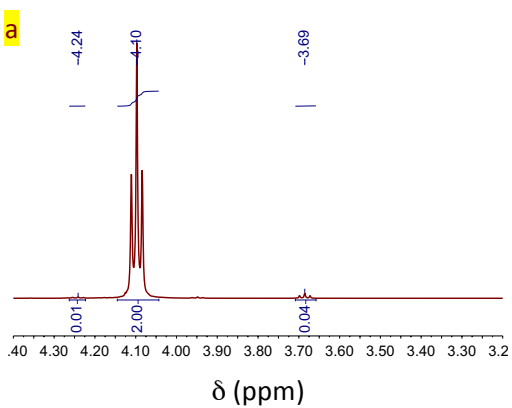


**Figure S5:** Details of the  $^1\text{H}$  NMR spectra of different PCLs obtained with initiator depicted in Figure S2 for chemical shifts ranging from (a) 3.20 to 4.40 ppm and (b) from 2.0 to 2.75 ppm.

**PCL110\_0.1:1.5**



**PCL110\_0.5:1.5**



**PCL110\_0:1.5**

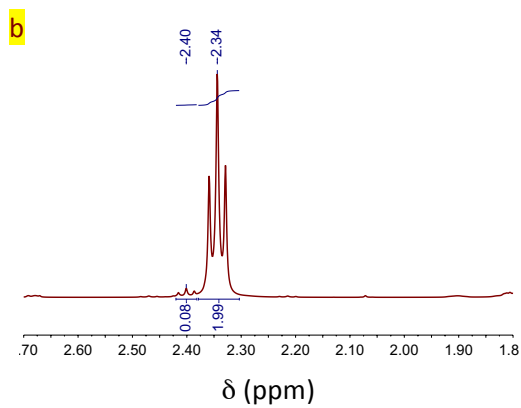
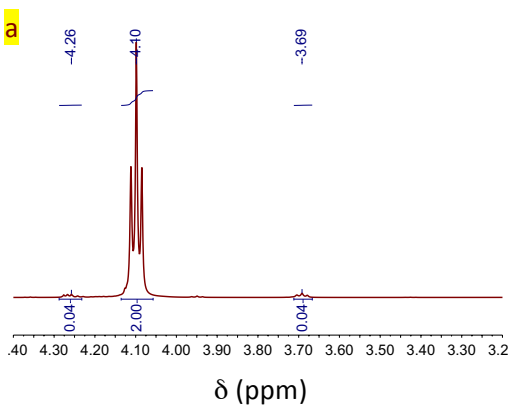
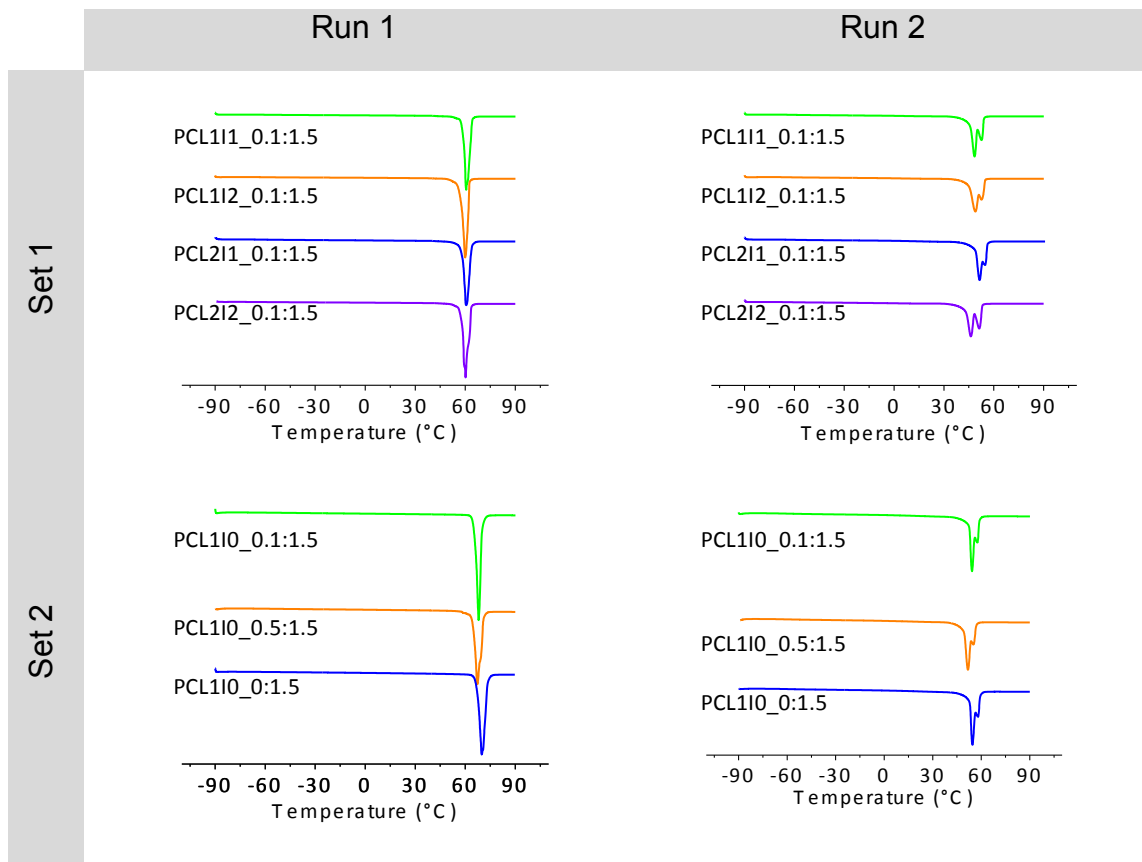


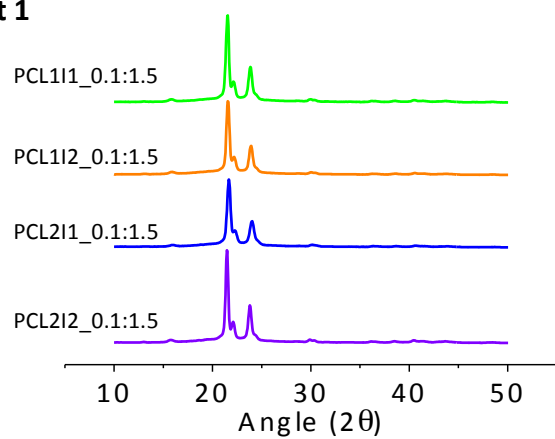
Figure S6: DSC scans – 1<sup>st</sup> and 2<sup>nd</sup> runs – of PCLs synthesized as described in Table 1.





**Figure S7:** XRD of PCLs synthesized as described in Table 1.

**a) Set 1**



**b) Set 2**

