Supporting Information

Oxidation of levulinic acid for the production of maleic anhydride: breathing new life into biochemicals

Anargyros Chatzidimitriou^a and Jesse Q. Bond^{a*}

^aDepartment of Biomedical and Chemical Engineering, College of Engineering & Computer Science, Syracuse University, Syracuse 13244, United States. E-mail: <u>jqbond@syr.edu</u>, phone: (315)443-2550

Esterification of product stream

Diacids and their respective anhydrides were elusive when it came to positive identification through our chromatography setup. Instead, we ran our gaseous product stream through a methanol bath with the purpose of esterifying the acids and diacids of our stream. The products were analyzed through a 7890 Agilent GC equipped with a 5975C MS detector and an Agilent HP-INNOWAX column. Esters of maleic, succinic and fumaric acid were positively identified against the "Japan AIST/NIMC Database" as shown in Figure S1. Figures S2-S7 depict the recorded mass spectra for the recovered esters.

GC-MS Chromatogram



Figure S 1: GC-MS chromatogram of esterified product stream

Dimethyl fumarate



Figure S 2: Mass spectrum of dimethyl fumarate.

Methyl levulinate



Figure S 3: Mass spectrum of methyl levulinate.

Dimethyl succinate



Figure S 4: Mass spectrum of dimethyl succinate.

Dimethyl maleate



Figure S 5: Mass spectrum of dimethyl maleate.

Methyl succinate



Figure S 6: Mass spectrum for methyl succinate

Methyl fumarate



Figure S 7: Mass spectrum for methyl fumarate