Electronic Supplementary Material (ESI) for Green Chemistry. This journal is © The Royal Society of Chemistry 2015

Supporting Information

Photoactive bimetallic framework for direct aminoformylation of nitroarenes

R. B. Nasir Baig^{a†}, Sanny Verma^{a†}, Mallikarjuna N. Nadagouda^b and Rajender S. Varma^{a*}

^aSustainable Technology Division, National Risk Management Research Laboratory, U. S. Environmental Protection Agency, MS 443, Cincinnati, Ohio 45268, USA. Fax: 513-569-7677; Tel: 513-487-2701.E-mail:varma.rajender@epa.gov

^bWQMB,WSWRD, National Risk Management Research Laboratory, U. S. Environmental Protection Agency, MS 443, Cincinnati, Ohio 45268, USA

- 1. Synthesis of g-C₃N₄ and AgPd@g-C₃N₄ catalyst
 - a) Synthesis of g-C₃N₄
 - b) Synthesis of AgPd@g-C₃N₄ catalyst
- 3. TEM image of g-C₃N₄ support (S1)
- 4. Recycling of AgPd@g-C₃N₄ catalyst (S2)
- 5. XPS spectra of recycled AgPd@g-C₃N₄ catalyst (S3)
- 6. XRD spectra of recycled AgPd@g-C₃N₄ catalyst (S4)
- 7. TEM spectra of recycled AgPd@g-C₃N₄ catalyst (S5)
- 8. GC-MASS data of the product
- 9. ¹H and ¹³C NMR of the product

1. Synthesis of g-C₃N₄ and AgPd@g-C₃N₄ catalyst

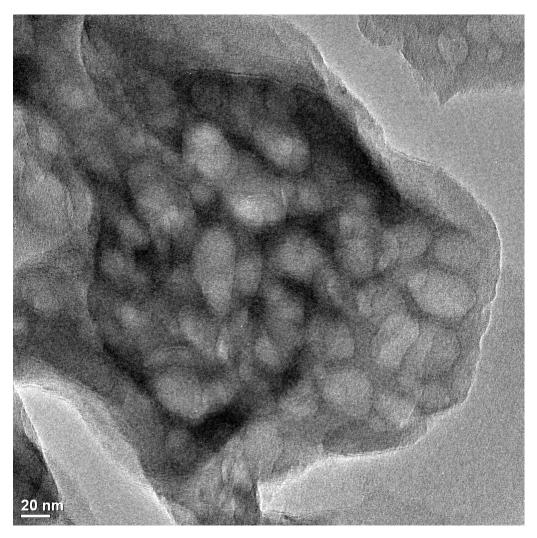
a) Synthesis of $g-C_3N_4$:

The pure urea obtained from Aldrich was calcinated at 500 °C in a closed furnace. A pale yellow solid of pure graphitic carbon nitride (g-C₃N₄) was obtained and used directly as a heterogeneous support in the synthesis of AgPd@g-C₃N₄ catalyst.

b) Synthesis of AgPd@g-C₃N₄ catalyst:

Graphitic carbon nitride, g-C₃N₄ (1000 mg) was dispersed in 500 mL of water using sonication. The aqueous solution of AgNO₃ was added in the aqueous suspension of g-C₃N₄. After 10 min, the aqueous solution of Pd(NO₃)₂ was added and the reaction mixture was stirred for 12 h. The reaction temperature was raised to 50 °C and excess of sodium borohydride (NaBH₄) was added in portions to reduce the Ag and Pd salts. The reaction mixture turned black after addition of NaBH₄. The stirring of reaction mixture was continued for another 12 hours. The catalyst was filtered off, washed with methanol and dried under vacuum at 50 °C. The isolated AgPd@g-C₃N₄ catalyst was than characterized using TEM, SEM, XRD, XPS and ICP-AES analysis.

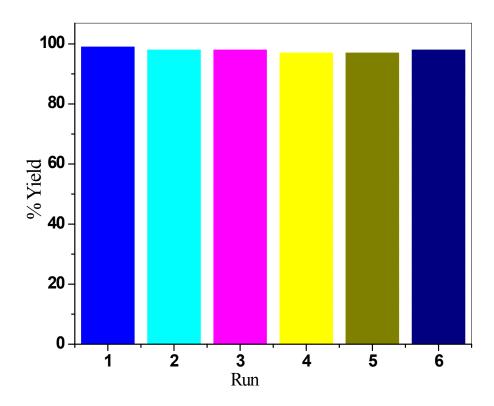
3. TEM image of g- C_3N_4 support



S1 TEM image of g-C₃N₄ support

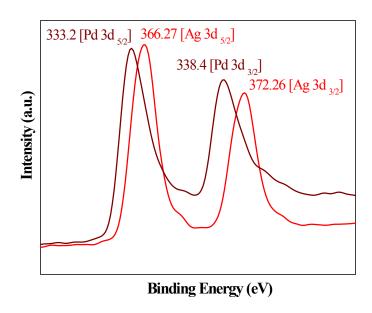
3. Recycling of catalyst:

After the completion of each reaction, the AgPd@g-C₃N₄ catalyst was recovered using centrifuge, washed with water followed by acetone: methanol (1:1) mixture and dried under vacuum and used for the fresh set of reactants. It was observed that the catalyst remains active even after sixth cycle and could be reused several times without losing activity.



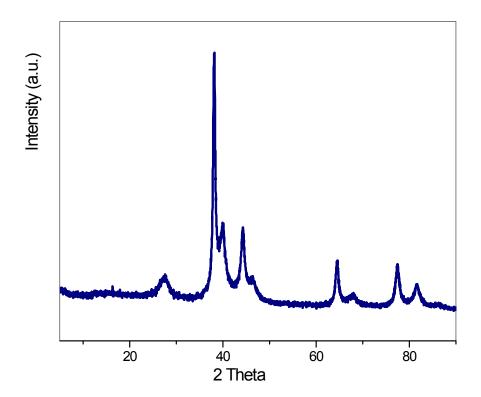
S2 Recycling of AgPd@g-C₃N₄ catalyst

5. XPS spectra of recycled AgPd@g- C_3N_4 catalyst



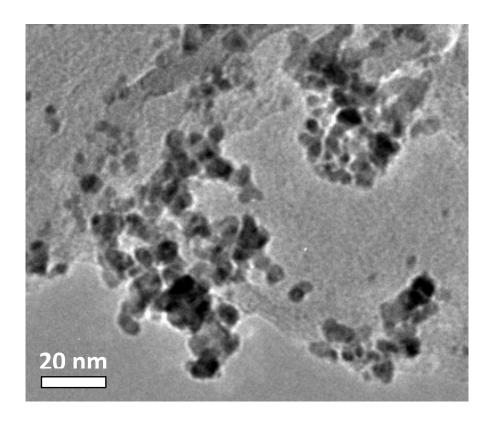
 $\textbf{S3} \text{ XPS spectra of recycled AgPd} \\ @g\text{-}C_3N_4 \text{ catalyst}$

6. XRD spectra of recycled AgPd@g-C₃N₄ catalyst



 ${f S4}$ XRD spectra of recycled AgPd@g-C₃N₄ catalyst

7. TEM spectra of recycled AgPd@g-C₃N₄ catalyst



S5 TEM spectra of recycled AgPd@g-C₃N₄ catalyst

GC-MASS Spectra of the product

Table 2, Entry 1

C:\Sanny\Data\ns 90tyr.D

Operator : kt

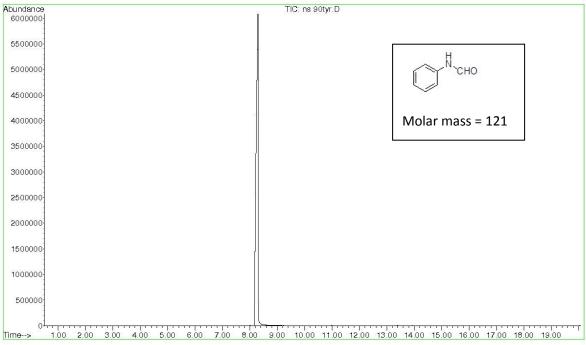
Acquired : 10 Nov 2015 12:25 pm using AcqMethod SAMPLES- 20 MIN_20151015_A.M

Instrument : Instrument #1

Sample Name: ns 90tyr

Misc Info :

Vial Number: 15



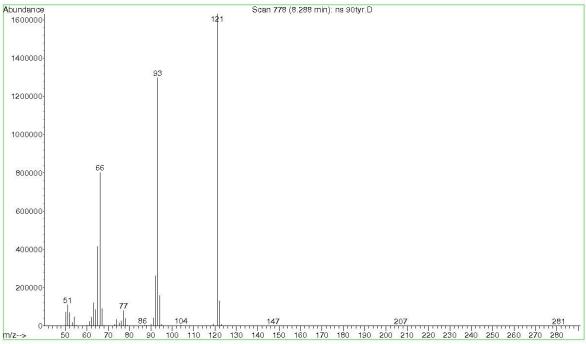


Table 2, Entry 2

:C:\Sanny\Data\NS 139 RT.D File

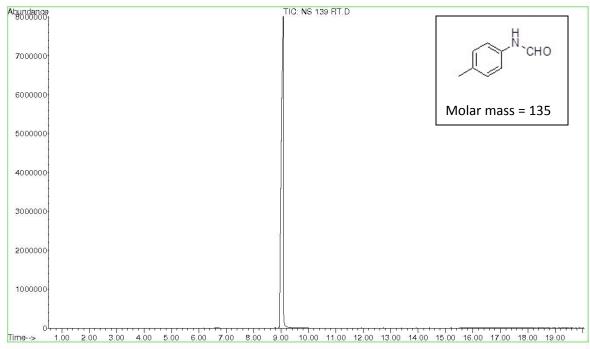
Operator

: kt : l2 Nov 2015 12:10 pm using AcqMethod SAMPLES- 20 MIN_20151015_A.M : Instrument #1 Acquired

Instrument:

Sample Name: NS 139 RT

Misc Info : Vial Number: 16



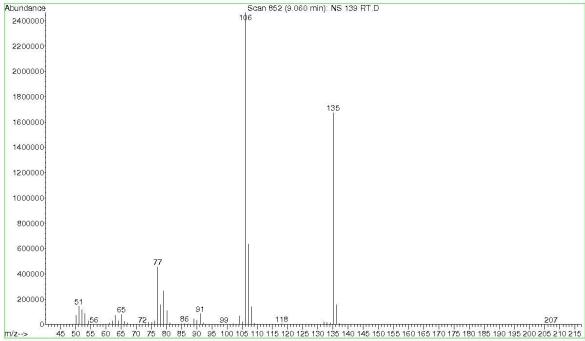
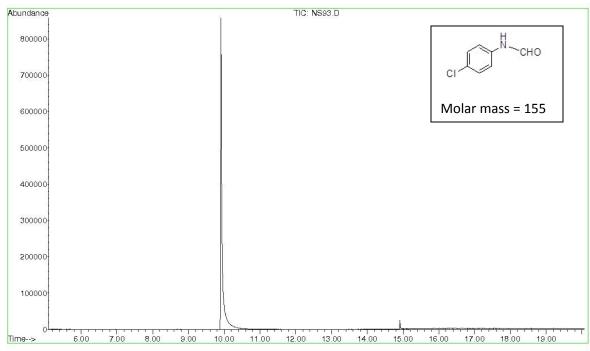


Table 2, Entry 4

Acquired: 3 Nov 2015 4:05 pm using AcqMethod SAMPLES- 20 MIN_20151015_A.M Instrument: Instrument #1 Sample Name: NS93 Misc Info: Vial Number: 7



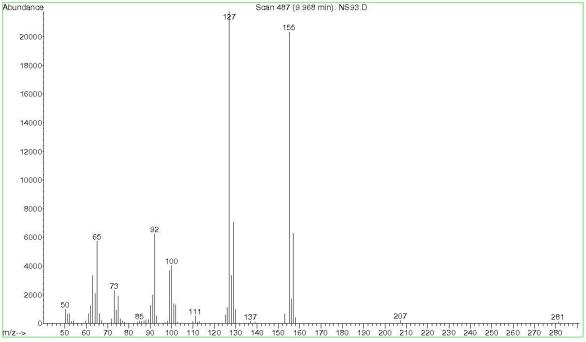


Table 2, Entry 5

:C:\Sanny\Data\NS 97 HJT.D File

File :C:\Sanny\Data\NS 97 HJT.D

Operator : kt

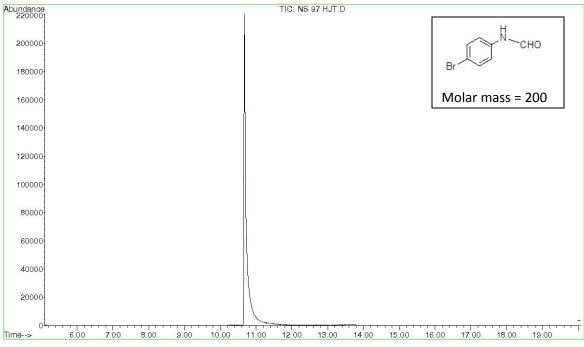
Acquired : 19 Oct 2015 1:12 pm using AcqMethod SAMPLES- 20 MIN_20151015_A.M

Instrument : Instrument #1

Sample Name: NS 97 HJT

Misc Info :

Vial Number: 8



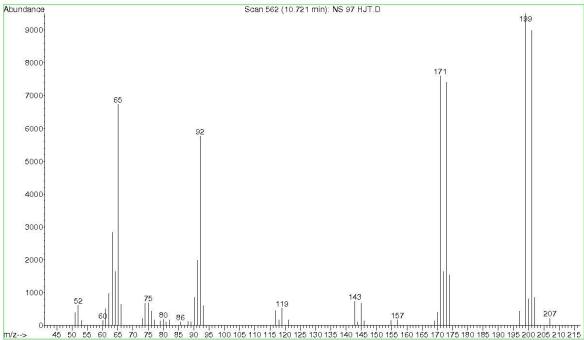


Table 2, Entry 6

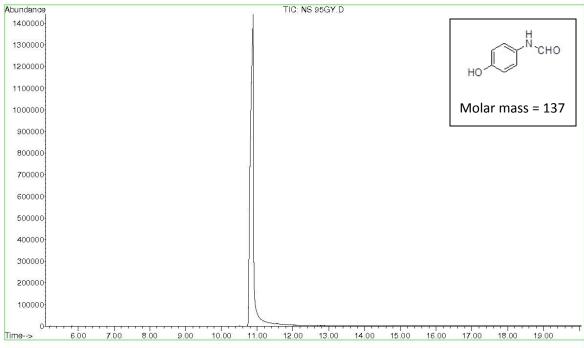
File :C:\Sanny\Data\NS 95GY.D

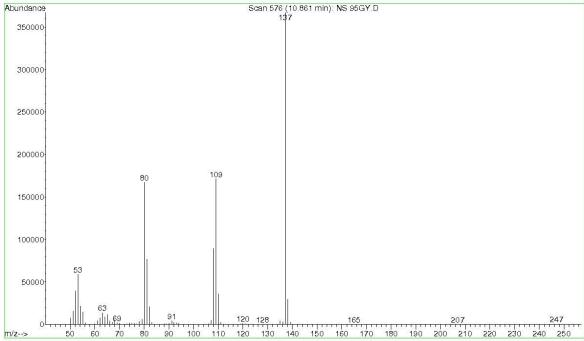
Operator

: kt : 10 Nov 2015 4:29 pm using AcqMethod SAMPLES- 20 MIN_20151015_A.M : Instrument #1 Acquired

Instrument:

Sample Name: NS 95GY Misc Info : Vial Number: 17



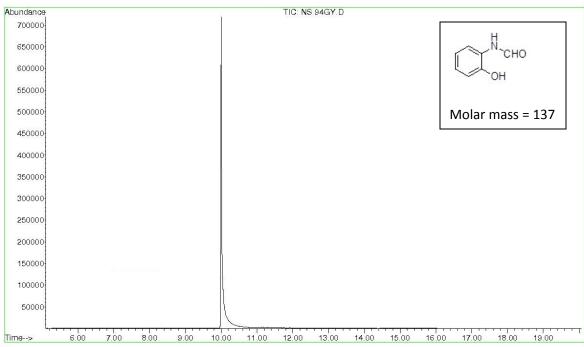


File Operator Acquired

Table 2, Entry 7

Instrument :

Sample Name: NS 94GY Misc Info : Vial Number: 18



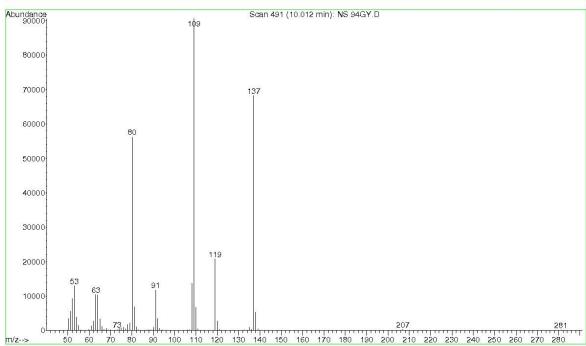


Table 2, Entry8

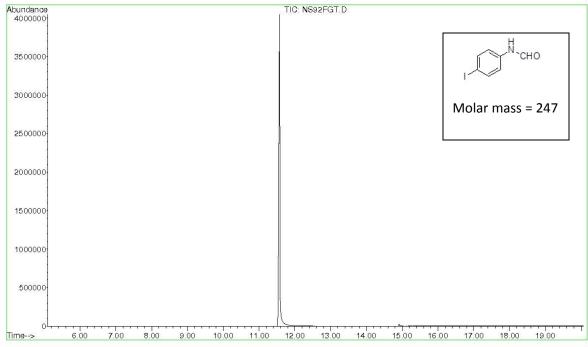
File :C:\Sanny\Data\NS92FGT.D

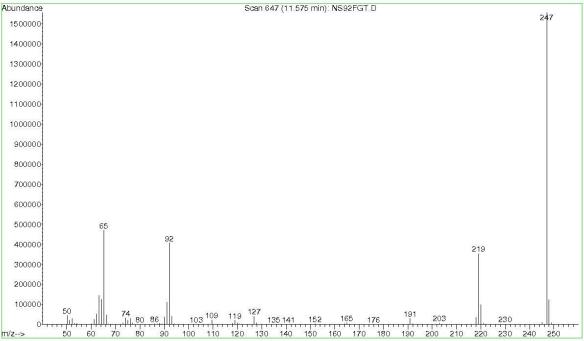
Operator : kt

Acquired : 12 Nov 2015 5:18 pm using AcqMethod SAMPLES- 20 MIN_20151015_A.M

Instrument: Instrument #1

Sample Name: NS92FGT Misc Info : Vial Number: 10



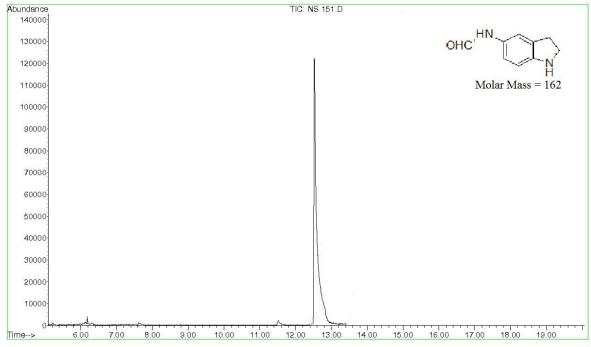


15

:C:\Sanny\Data\NS 151.D File Table 2, Entry 10

Operator: kt
Acquired: 9 Dec
Instrument: Inst
Sample Name: NS 151 : kt
: 9 Dec 2015 3:19 pm using AcqMethod SAMPLES- 20 MIN_20151015_A.M
: Instrument #1

Misc Info: Vial Number: 18



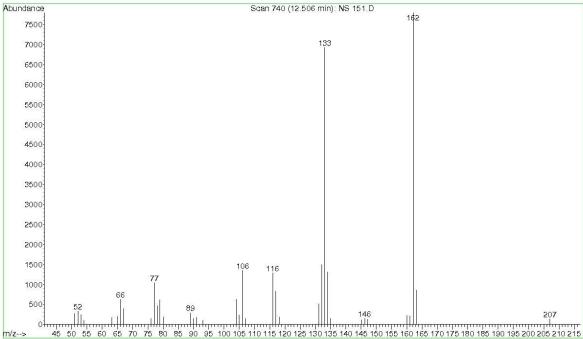


Table 2, Entry 1

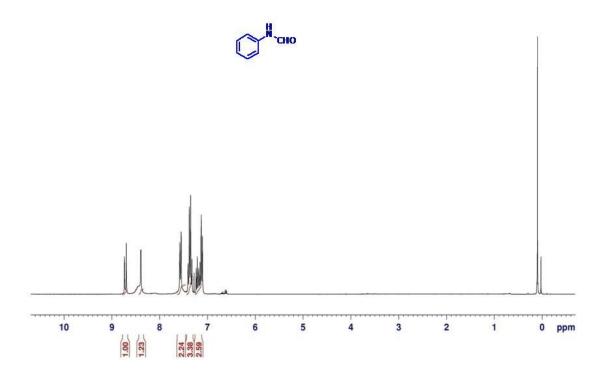


Table 2, Entry 2

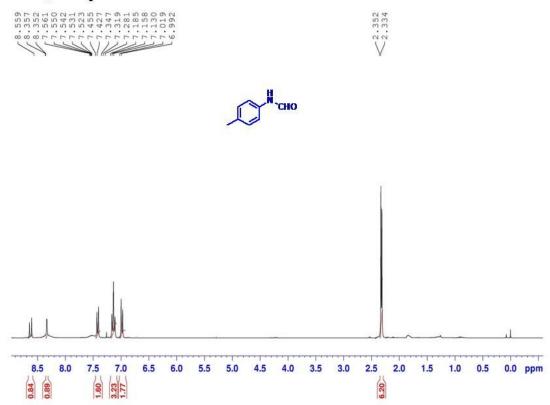


Table 2, Entry 2

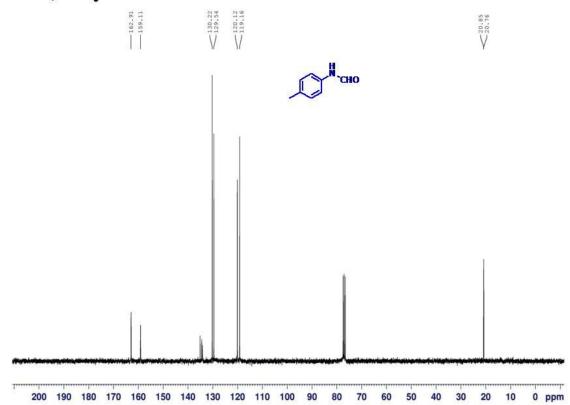
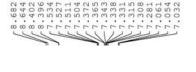


Table 2, Entry 4





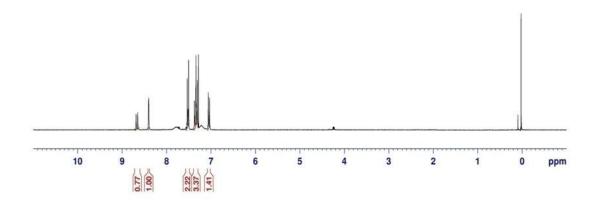


Table 2, Entry 4



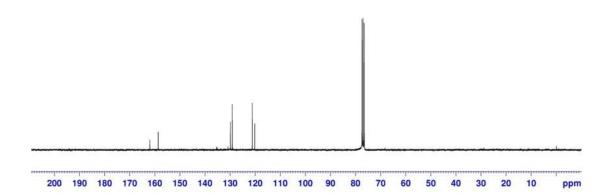
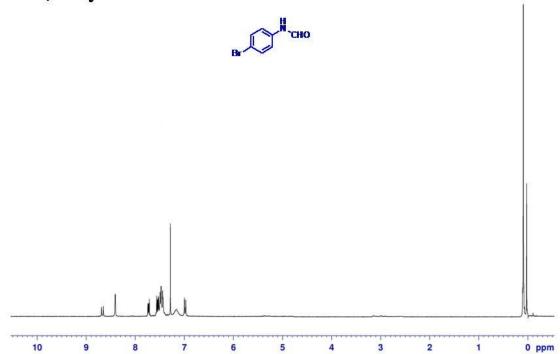
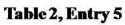
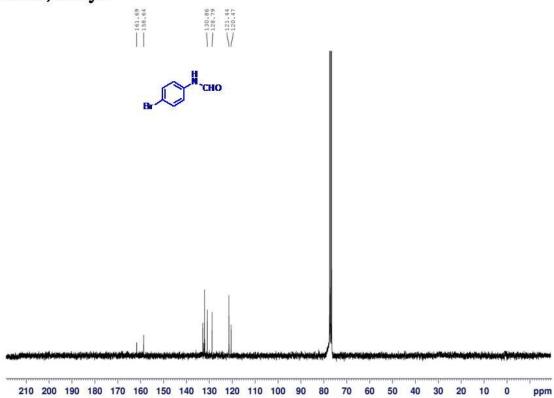
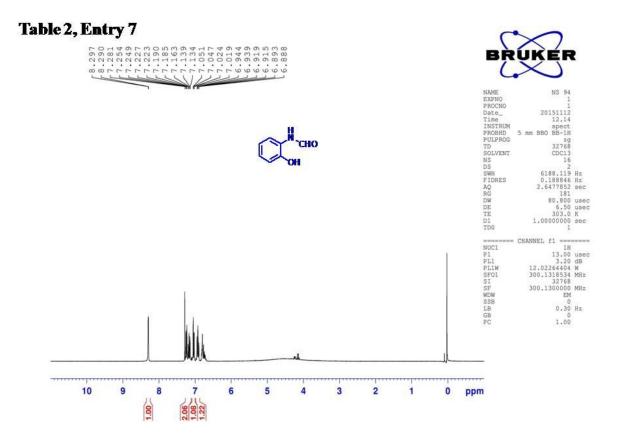


Table 2, Entry 5









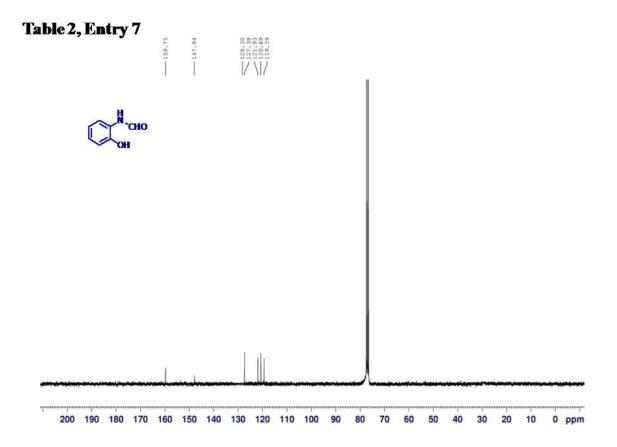
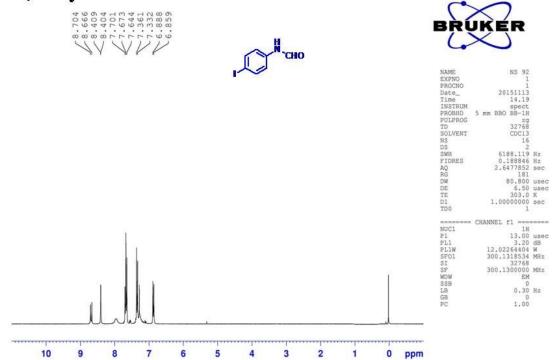
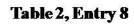
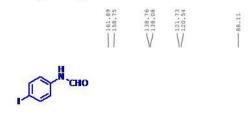
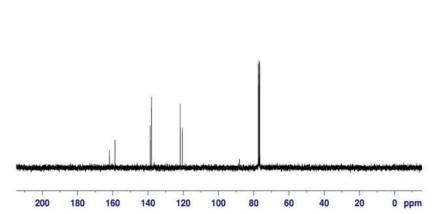


Table 2, Entry 8











_	1	
NAME	NS 92	
EXPNO	2	
PROCNO	1	
Date_	20151113	
Time	15.19	
INSTRUM	spect	
PROBHD	5 mm BBO BB-1H	
PULPROG	zgig	
TD	32768	
SOLVENT	CDC13	
NS	1024	
DS	4	
SWH	17985.611	Hz
FIDRES	0.548877	
AO	0.9110282	
RG	8192	000
DW	27,800	usec
DE		usec
TE	303.0	
D1	2,00000000	
D11	0.03000000	
TDO	1	Series !
	CHANNEL f1 ===	
NUC1	130	
P1	10.00	usec
PL1	1.80	dB
PL1W	49.78760910	W
SF01	75.4752953	MHz
	CHANNEL f2 ===	
CPDPRG2	waltz16	
NUC2	18	
PCPD2	80.00	usec
PL2	3.20	dB
PL12	18 98	dB

CHANNEL f2 -----CPDPRG2 waltzl6
NUC2 11
PCPD2 80.00 us
PL2 3.20 dB
PL12 18.98 dB
PL2W 12.0226440 W
PL12W 0.31768745 W
FU12W 0