## **Electronic Supplementary Information**

ESI Table 1: Typical instrument working conditions.

HPLC						
Agilent 1200 coupled to Agilent ICP-MS 7700						
Autosampler temperature	4 °C					
Injection volume	40 µL					
Flow rate	0.5 mL/min					
Column	MonoQ 5/50 GL					
Oven temperature	30 °C					
Eluent	A: 12.5 mM Tris pH 7.2					
	B: A + 125 mM NH <sub>4</sub> Ac pH 7.2					
Gradient	7 % B 0 – 5 min					
	7 - 50 % B 5 - 10 min					
	50 % B 10 – 15 min					
	50 - 86 % B 15 - 17 min					
	86 % B 17 – 19 min					
	86 – 100 % B 19 – 20 min					
	100 % B 20 – 25 min					
Agilent 1200 coupled to Bruker micrOTOF-Q						
Flow rate	0.2 mL/min					
Column	Jupiter300-C4					
Oven temperature	40 °C					
Eluent	A: 0.1% HCOOH in H <sub>2</sub> O					
	B: 0.1% HCOOH in CH <sub>3</sub> CN					
Gradient	5 % B 0 – 5 min					
	5 - 100% B 5 - 55 min					
	100% B 55 – 65 min					
	100 - 5% B 65 - 70 min					
	5% B 70 – 80 min					
Accela coupled to Element2						
Injection volume	50 µL					
Flow rate	1 mL/min					
Column	HiTrap (SEC) / MonoQ 5/50 GL					
Oven temperature	30 °C					
Eluent (SEC)	20 mM Tris + 50 mM KCl pH 7.4					
Eluent (MonoQ)	A: 25 mM Tris pH 7.4					
	B: A + 250 mM NH <sub>4</sub> Ac pH 7.4					
Gradient (MonoQ)	0 - 100 % B 0 - 30 min					
	100 % B 30 – 35 min					
ICP-MS						
Agilent 7700						
RF Power	1500 W					
Sample depth	8.0 mm					
Plasma gas flow	15 L/min					

Spraying chamber temperature	2 °C
Isotopes monitored	63, 65, 66, 67
Collision gas	4.5 mL He / no gas



*ESI* Figure 1. ESI-MS data for the verification of the identity of native SOD1. A) Total ion chromatogram of SOD1. B) Charge distribution of SOD1. C) Mass calculation by charge deconvolution.

## Equation for the determination of the Cu mass fraction in native SOD1 by single post-column IDMS

Calculation of K

$$R^{true} = K * R^{meas}$$

Single post-column IDMS

$$w_{x} = \frac{m_{y} * w_{y}}{\rho_{x} * V_{inj}} * \frac{M_{x}}{M_{y}} * \frac{\sum R_{x}}{\sum R_{y}} * \int_{t_{1}}^{t_{2}} \frac{R_{y} - R_{bx}}{R_{bx} - R_{x}} dt$$
(3)

R<sup>IUPAC</sup>

 $\overline{R_K}$ (1)

(2)

 $K = \frac{1}{2}$ 

symbol	unit	definition						
$W_{\rm x}, W_{\rm y}$	g/kg	mass fraction of Cu in sample x and spike y, resp.						
my	g	mass flow of spike y						
$ ho_{ m x}$	g/cm <sup>3</sup>	density of sample x						
$V_{ m inj}$	mL	injection volume of sample x						
$M_{ m x},M_{ m y}$	g/mol	molar masses of natural Cu in sample x and isotopically						
		labelled Cu in spike y						
$R_{\rm x}, R_{\rm y}$	mol/mol	isotope amount ratio of spike and reference isotope						
		( <sup>65</sup> Cu/ <sup>63</sup> Cu) in sample x and spike y, resp.						
R <sub>bx</sub>	mol/mol	isotope amount ratio of spike and reference isotope						
		$(^{65}Cu/^{63}Cu)$ in blend bx (sample x + spike y)						
Κ	1	Correction factor for mass bias						
R <sup>IUPAC</sup>	mol/mol	natural isotope ratio according to IUPAC						
$R_{\rm k}$	mol/mol	measured isotope ratio in reference k						
R <sup>true</sup>	mol/mol	isotope ratio after correction for mass discrimination						
R <sup>meas</sup>	mol/mol	measured isotope ratio in sample x						

**ESI** Table 2. Meaning of the symbols used in the equations (1) - (3).

## Instrumentation for characterization of produced apo-SOD1 and \*SOD1

Both the apo- and remetallated SOD1 as well as the native protein, were characterized regarding metal content, protein integrity and activity. The characterization of the produced apo-SOD1 and the remetallated protein was performed with an Accela HPLC system (Thermo Fisher, Waltham, USA) and an sector-field ICP-MS (Element 2, Thermo Scientific, Schwerte, Germany) in medium resolution (typical operating parameters are also summerized in *ESI* table 1). For the quantification of Cu and Zn in the apo- and the remetallated form of the protein, external calibration using Cu and Zn standard solution and solutions prepared from <sup>65</sup>Cu and <sup>68</sup>Zn were used. Speciation of apo- and remetallated SOD1 was done using size exclusion chromatography (SEC) on a HiTrap column (Pharmacia, Uppsala, Sweden) and strong anion exchange (SAX) chromatography. Protein integrity was tested using ESI-FT-MS (solvent: 1 mM NH<sub>4</sub>Ac buffer at pH 7.4, flow injection). An Orbitrap Discovery system (Thermo Scientific) was used for this purpose for the determination of the spectra. The positive fourier transformation (FT) mode with a resolution of 30.000 and an extended mass range from 600 - 3000 Da was used in ESI-FT-MS. The activity was determind according to Marklund<sup>15</sup> using an UV-vis spectrophotometer (Lambda 25, Perkin Elmer, Cambridge, UK). An electrophoresis system from Atto (Tokyo, Japan) was used for native PAGE. Gels were self-casted and contained 7 % polyacrylamid and were stained with Coomassie blue.

*ESI* Table 3. Results of ICP-MS measurements expressed as protein concentration of produced apo-SOD1 and isotopically labelled SOD1 (n = 9).

	total Cu ng/mg	RSD %	<sup>65</sup> Cu ng/mg	RSD %	Total Zn ng/mg	RSD %	<sup>68</sup> Zn ng/mg	RSD %	Activity U/mg	RSD / %
native SOD	5.96	4.5			4.38	6.7			5690	4
apo-SOD A	1.91	2.5			1.61	1.1			260	20
*SOD A	6.49	4.5	4.46	4.4	5.62	5.6	3.55	4.1	4918	11

apo-SOD B	0.53	3.4			0.34	2.3			164	14
*SOD B	6.3	4.5	6.1	4.5	6.99	3.9	6.15	3.6	5752	15
apo-SOD C	0.72	4.2			0.72	7.2			311	24
*SOD C	4.3	4.7	3.76	4.4	6.23	5.1	5.18	3.8	4733	1



*ESI* Figure 2. Native PAGE to determine the structural state of A) the apo-SOD1 after production using methods A, B and C, resp., compared to native SOD1. Lane 1: native SOD1, lane 2: apo-SOD1 method A, lane 3: apo-SOD1 method B, lane 4: apo-SOD1 method C and B) the remetallated SOD1 spikes compared to native SOD1. Lane 1: native SOD1, lane 2: \*SOD1 A, lane 3: \*SOD1 B, lane 4: \*SOD1 C.



ESI Figure 3. Charge distribution of apo-SOD1 prepared according to A) method A, B) method B and C) method C.



ESI Figure 4. Mass calculation by charge deconvolution for SOD1 and \*SOD1 prepared according to method A, B and C.



*ESI* Figure 5. Stability of \*SOD1 C on MonoQ 5/50 GL column: A)  $^{65}$ Cu /  $^{63}$ Cu ratios and B)  $^{68}$ Zn /  $^{64}$ Zn ratios for pure spike material, erythrocyte lysate, erythrocyte lysate (Hb precipitated) and both of the latter spiked with \*SOD C. Y-axis is set-off for clarity purposes.



*ESI* Figure 6. Stability of \*SOD1 B on MonoQ 5/50 GL column: A)  ${}^{65}$ Cu /  ${}^{63}$ Cu ratios and B)  ${}^{68}$ Zn /  ${}^{64}$ Zn ratios for pure spike material, erythrocyte lysate, spiked erythrocyte lysate, erythrocyte lysate (Hb precipitated) and spiked erythrocyte lysate (Hb precipitated). Y-axis is set-off for clarity purposes.



*ESI* Figure 7.  ${}^{65}$ Cu/ ${}^{63}$ Cu resp  ${}^{67}$ Zn/ ${}^{66}$ Zn isotope ratio of human SOD spike in erythrocyte lysate over a time period of 48 h. The error bars represent the standard deviation (n = 3).