

Supporting Information

Ultrastrong Composites from Dopamine Modified Polymer-Infiltrated Colloidal Crystals

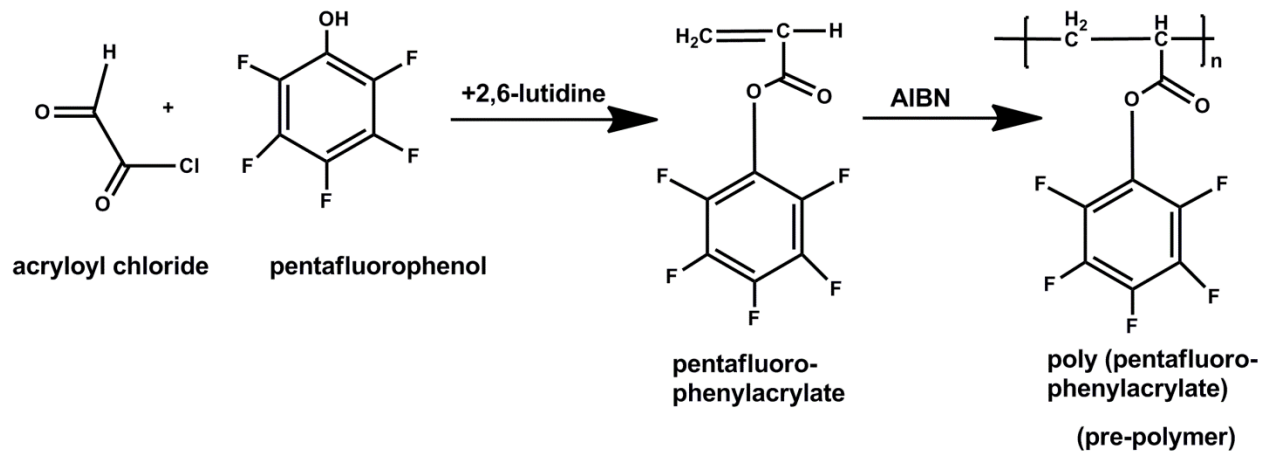
Faroha Liaqat¹, Muhammad Nawaz Tahir^{1*}, Hannah Huesmann¹, Phillip Daniel,¹
Michael Kappl², Günter K. Auernhammer², Dirk Schneider², Ingo Lieberwirth²,
Kookheon Char³, George Fytas^{2, 4}, Hans-Jürgen Butt², and Wolfgang Tremel^{1*}

¹ Institut für Anorganische Chemie und Analytische Chemie, Johannes Gutenberg-Universität, Duesbergweg
10-14, 55099 Mainz, Germany

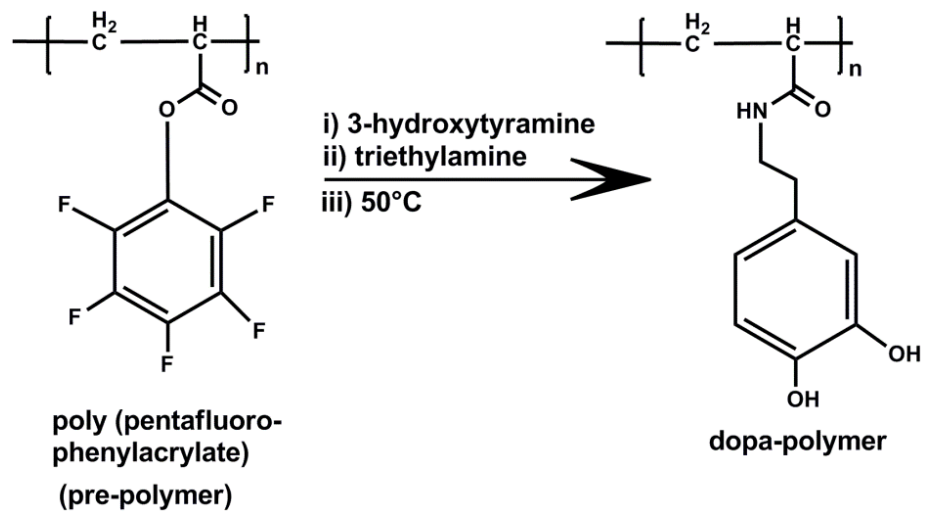
² Max Planck-Institut für Polymerforschung, Ackermannweg 10, 55128 Mainz, Germany

³ School of Biological and Chemical Engineering, Center for Functional Polymer Thin Films, Seoul National
University, Seoul 151-744, South Korea

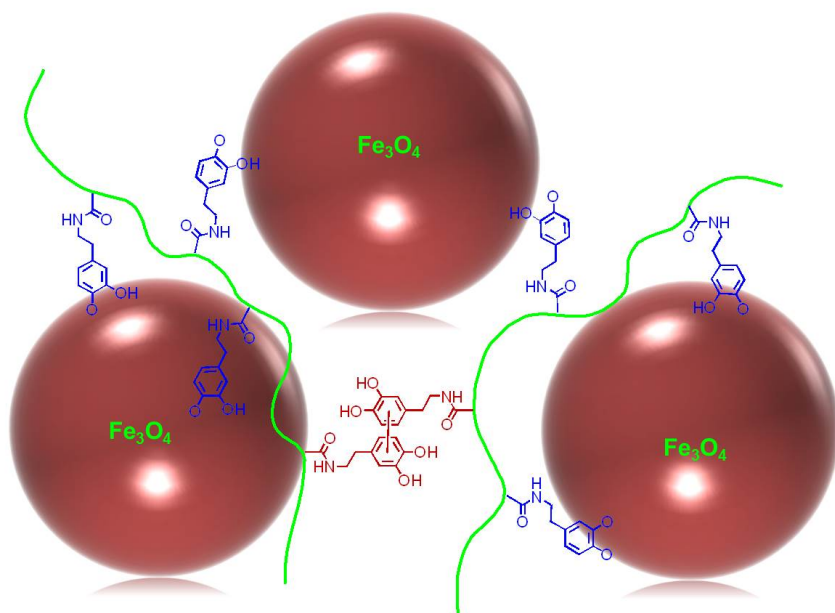
⁴ Department of Materials Science, University of Crete and IESL/FORTH, 71110 Heraklion, Greece



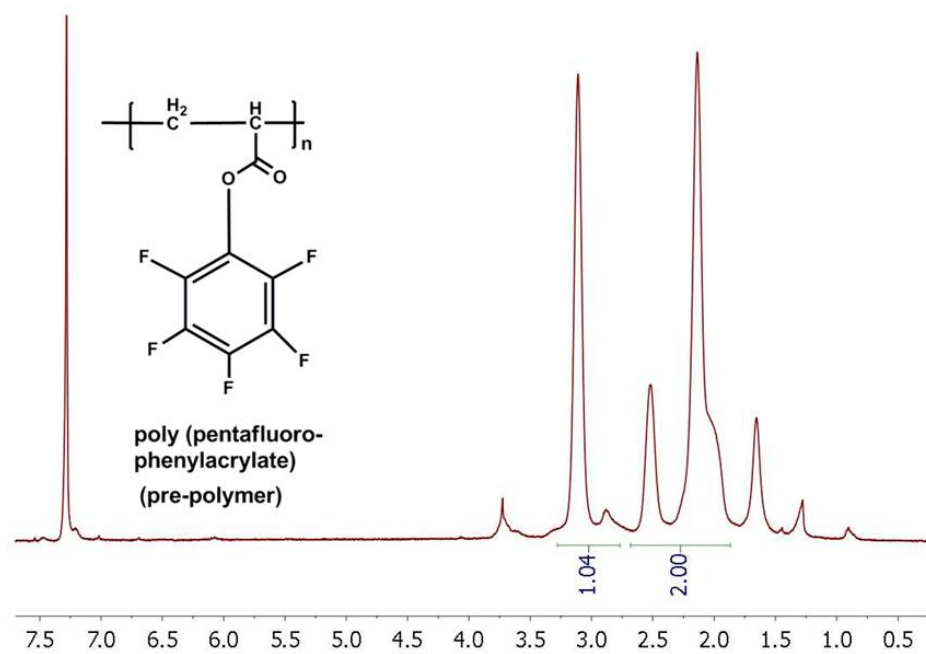
Scheme S1. Synthesis of the poly (active ester) poly (pentafluoro-phenylacrylate) (PFA).



Scheme S2. Synthesis of the dopamine modified polymer



Scheme S3: Sketch of the connectivity of the Fe_3O_4 nanoparticles through the dopamine modified polymer.



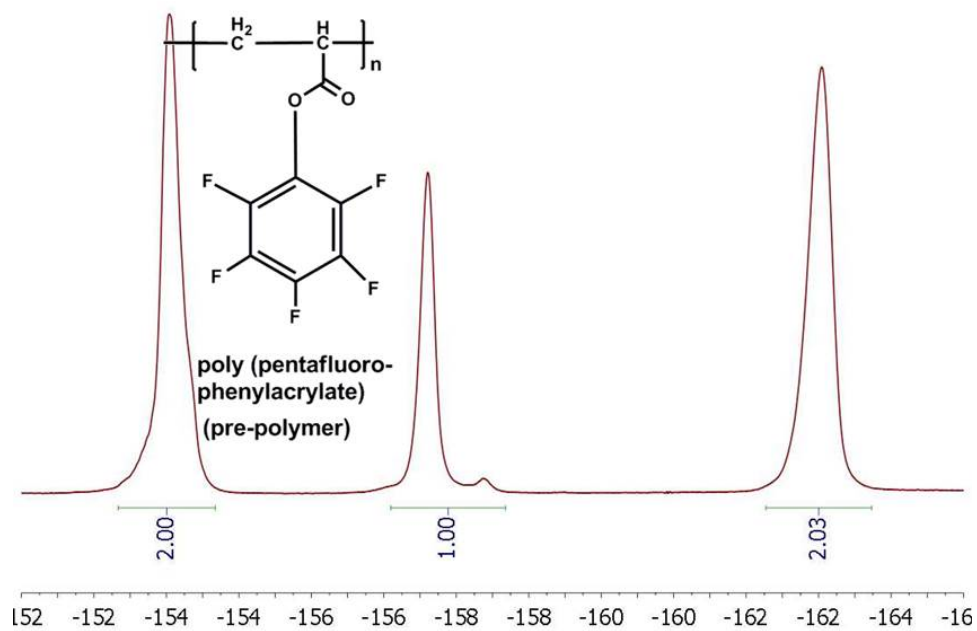


Figure S2. ^{19}F NMR spectrum of poly(pentafluorophenyl acrylate)

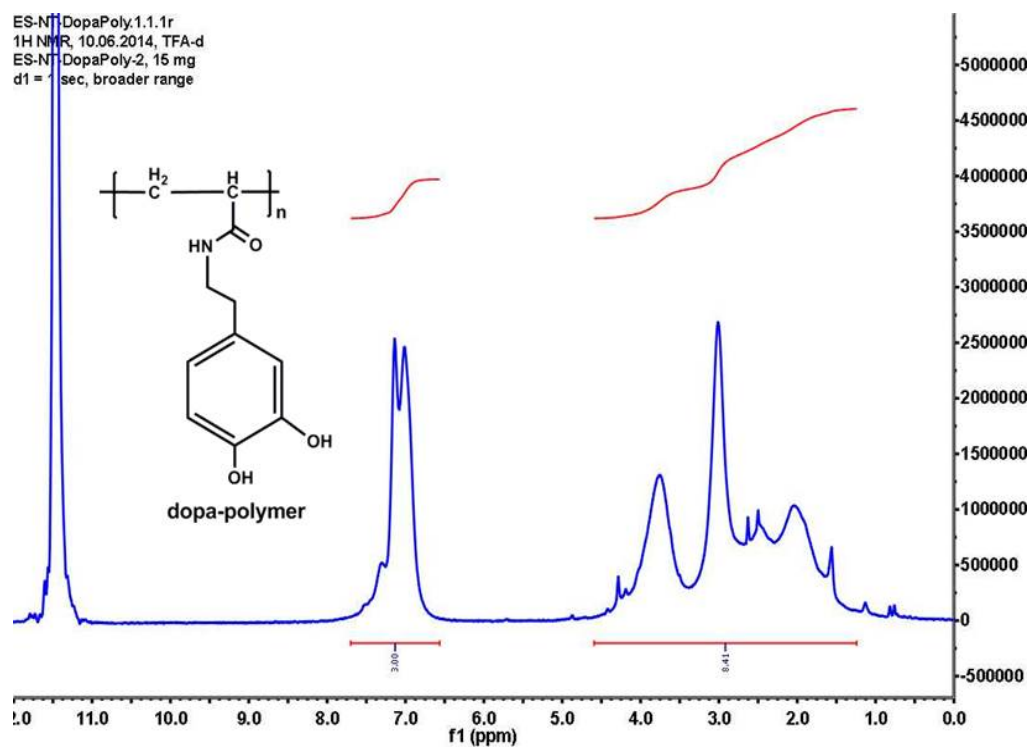


Figure S3. ¹H NMR spectrum of the dopamine modified polymer.

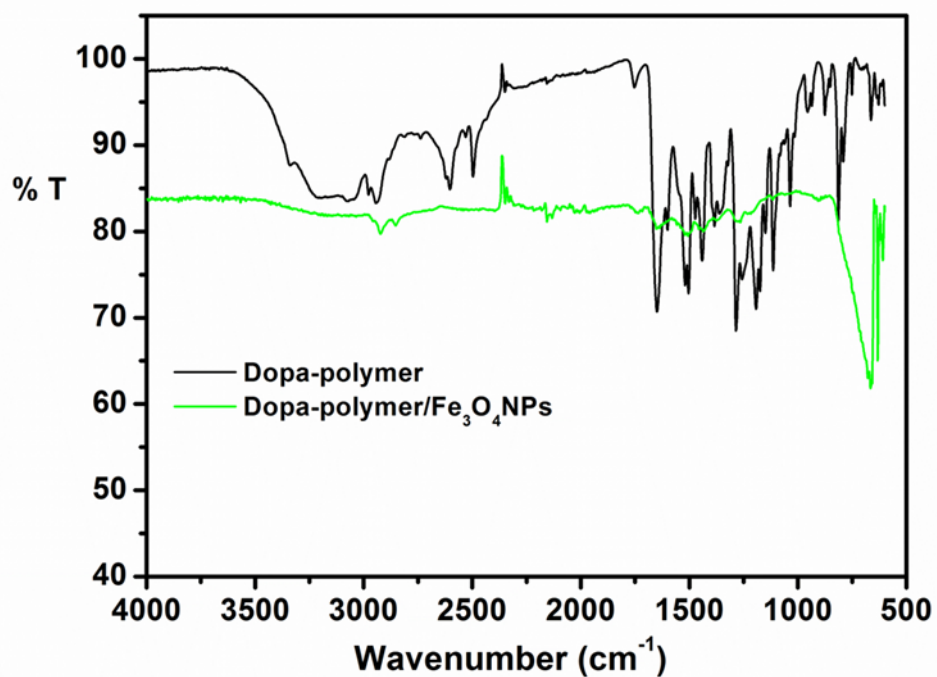


Figure S4. A comparison of the FTIR spectra of pure dopamine modified polymer (black) and the nanocomposite with Fe₃O₄ NPs (green) is shown. The binding between the polymer and iron oxide NPs is evident from the absence of the phenolic -OH stretching in the hybrid.

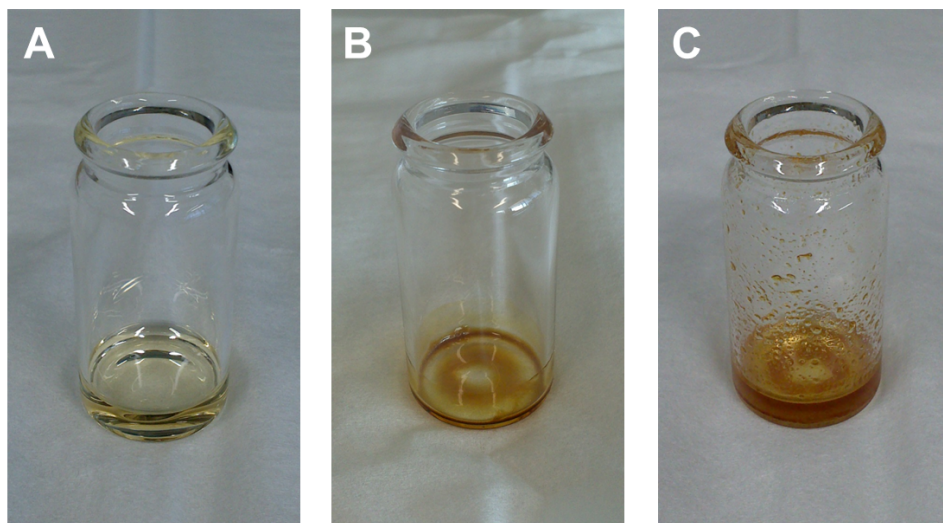


Figure S5. Digital photograph A) as prepared dopamine modified polymer solution in DMA B) after heating the solution at 120°C for 20 minutes, C) dissolution in DMA using ultrasonication for several hours.

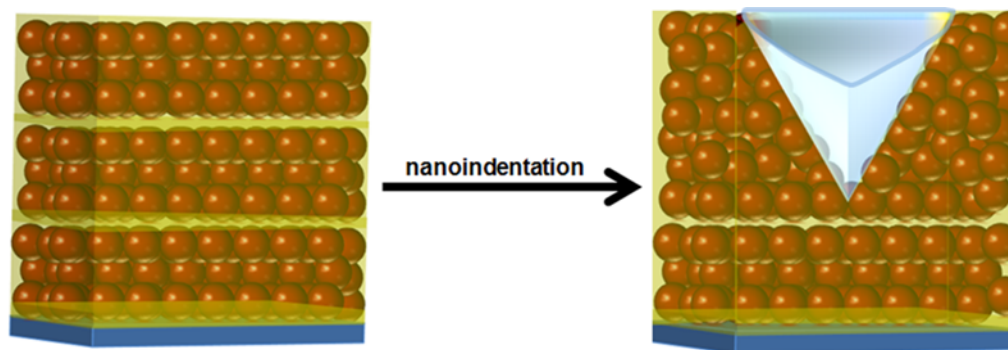


Figure S6. Nano-indentation of a multilayer Fe_3O_4 / dopamine modified polymer composite. The strong crosslinking between the Fe_3O_4 nanoparticles and the multidentate polymer ligand help the matrix resist deformation and make the composites harder and stronger. This figure shows a model illustrating the cohesive role of the catechol-polymer in the multilayer films. The multilayers contain a high crosslinking density. When the film is indented, the polymer strands have to be unfolded and a multitude of strong metal-catechol must be broken.

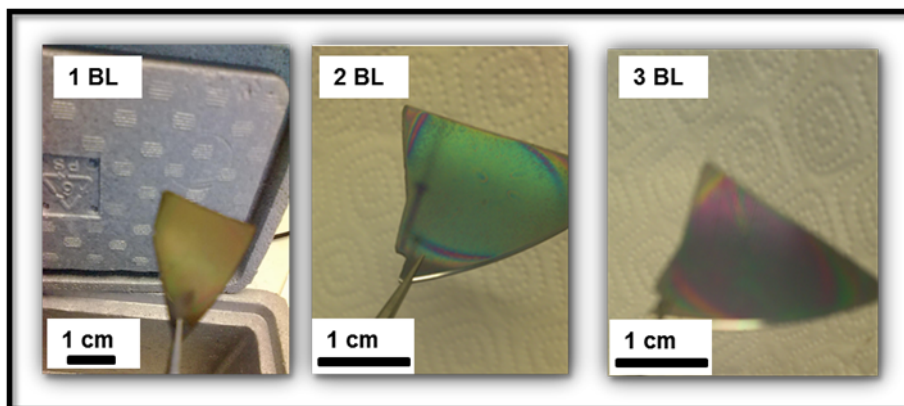


Figure S7. Structural colors shown by multilayers of dopamine modified polymer/ Fe_3O_4 due to multilayer interference with increasing bilayers (BL)

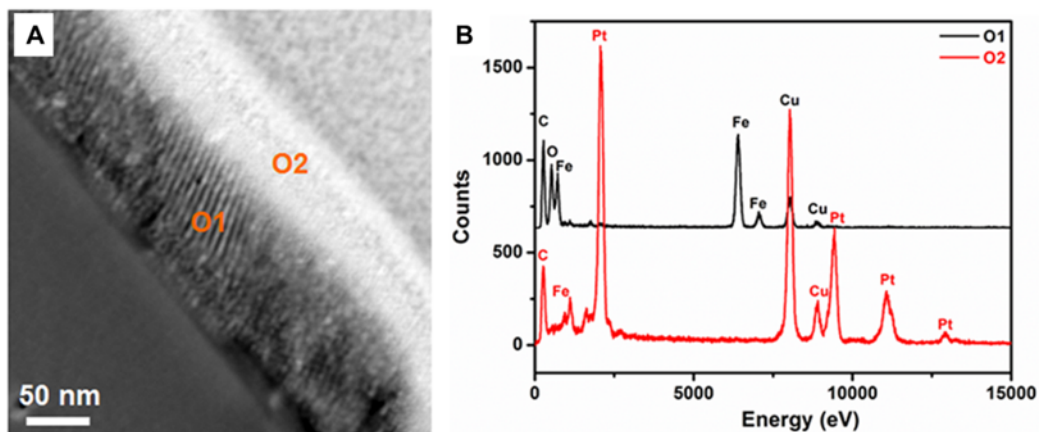


Figure S8. EDX analysis of dopamine modified polymer/ Fe_3O_4 multilayers coupled with TEM. (A) TEM image of the multilayers showing long range ordering. The two points indicated on the TEM micrographs correspond to the spectra in **(B)** representing the elemental content in the multilayers.

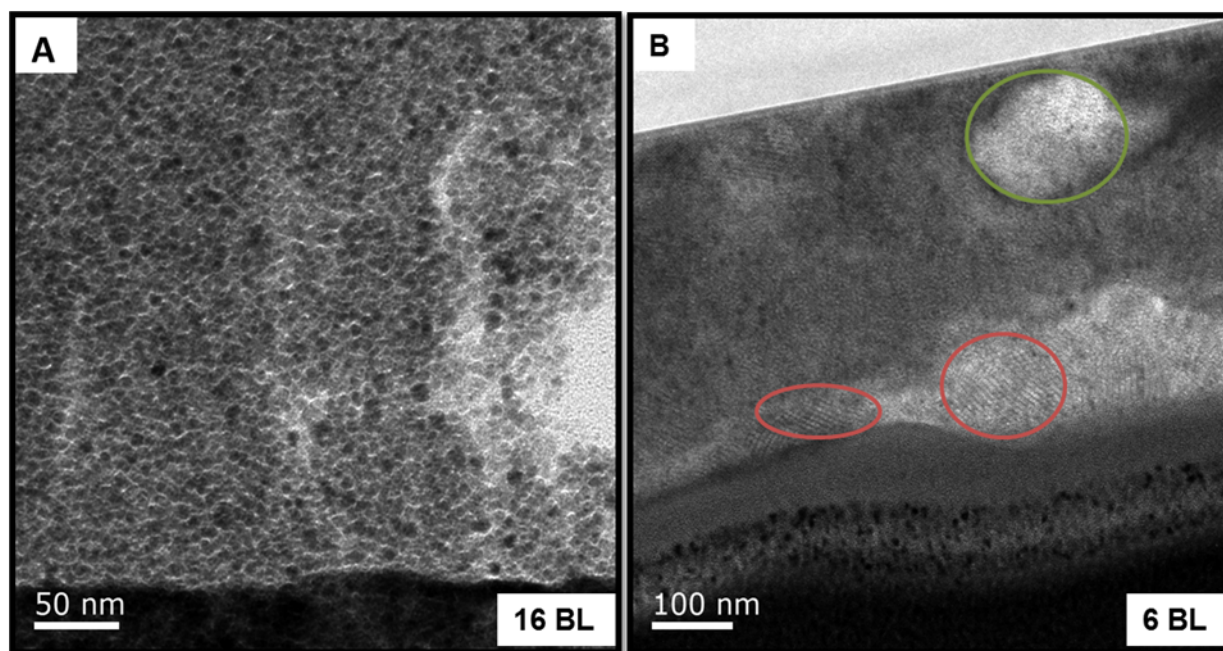


Figure S9. A comparison of TEM images of different samples of dopamine modified Fe_3O_4 nanoparticles consisting of a varying number of multilayers. **(A)** A sample prepared by 32 spin-coating cycles shows the existence of large defected regions as the number of multilayers increase. **(B)** On the other hand, a sample prepared by 12 repeated dip-coating cycles reveal the existence of ordered regions (red circles) and defects (green circle) in addition to a regular cross-linked network, highlighting the presence of inhomogeneous regions in contrast to the superior spin coating method.