

Electronic Supplementary Information (ESI)

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Influence of noncovalent interactions on the structures of metal–organic hybrids based on $[\text{VO}_2(2,6\text{-pydc})]^-$ tecton with cations of imidazole, pyridine and its derivatives

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Figs. S1–S7 Asymmetric units

Table S1 Selected bond distances for compounds 1–5.

Table S2 Selected bond angles for compounds 1–5.

Tables S3–S7 Hydrogen bonding for 1–5 [$\text{\AA}$ and $^\circ$]

Table S8 Geometrical parameters ($\text{\AA}$, $^\circ$) for $\pi\cdots\pi$ stacking interactions in $1\cdot 2\text{H}_2\text{O}$, 1 , $4\cdot\text{H}_2\text{O}$ and 5

Fig. S8–S9 Comparison of PXRD patterns

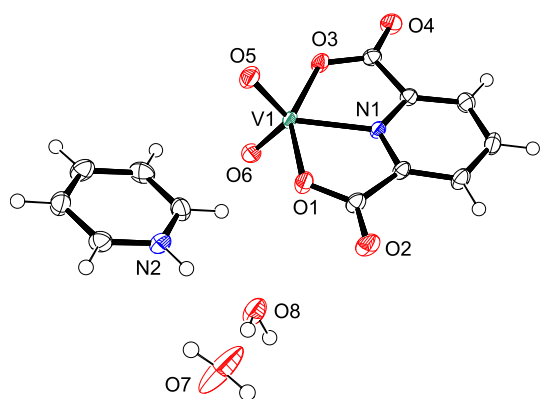


Fig. S1 Asymmetric unit of **1**·2H₂O.

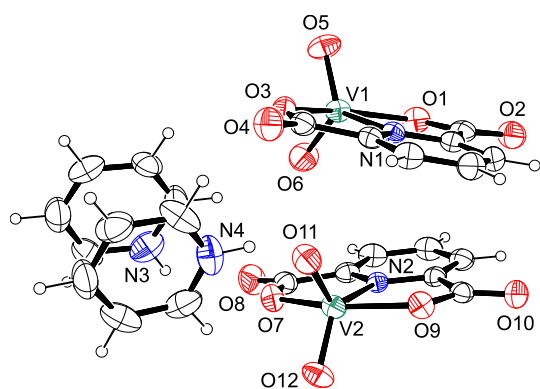


Fig. S2 Asymmetric unit of anhydrous **1**.

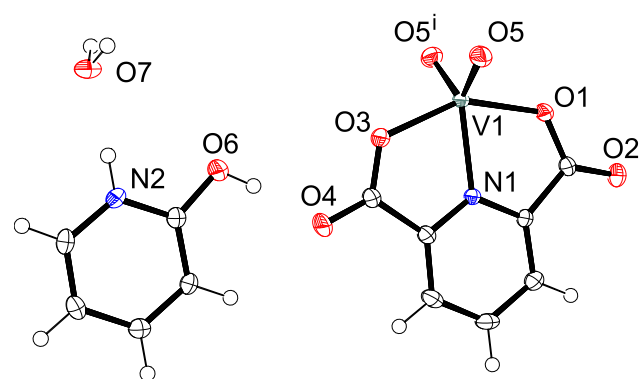


Fig. S3 Structure of **2**·H₂O. Symmetry code: (i) $x, -y + \frac{1}{2}, z$.

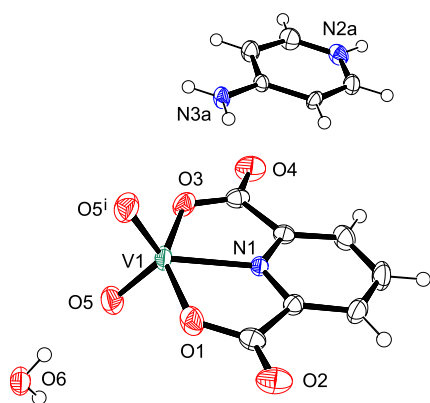


Fig. S4 Structure of $3 \cdot \text{H}_2\text{O}$. Symmetry code: (i) $x, -y + 1\frac{1}{2}, z$.

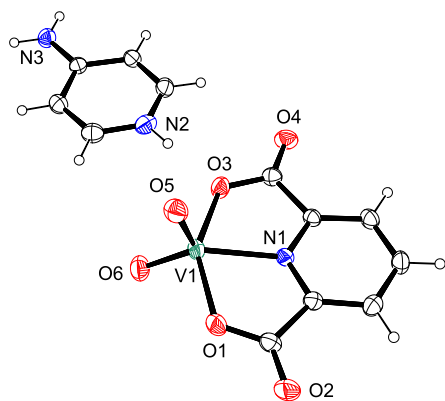


Fig. S5 Asymmetric unit of anhydrous **3**.

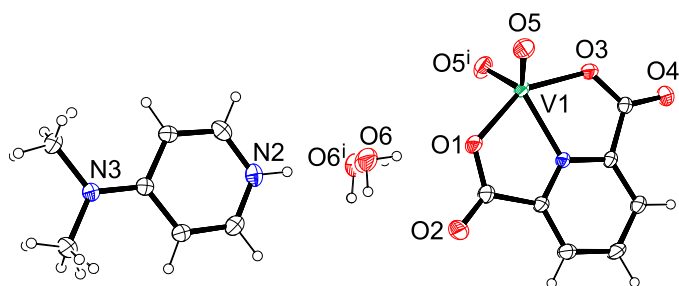


Fig. S6 Asymmetric unit of $4 \cdot \text{H}_2\text{O}$.

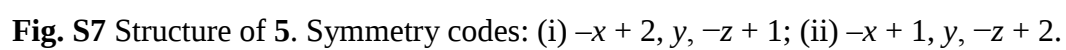


Fig. S7 Structure of **5**. Symmetry codes: (i) $-x + 2, y, -z + 1$; (ii) $-x + 1, y, -z + 2$.

Table S1. Selected bond distances for compounds **1**–**5**.

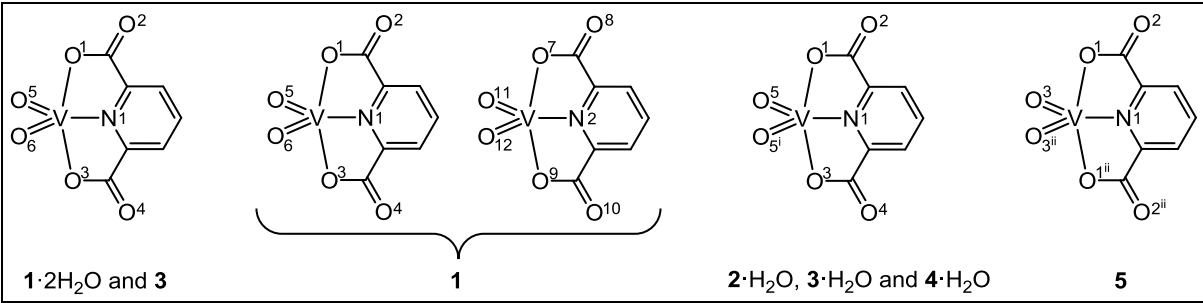
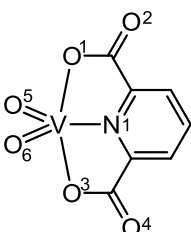
Distance	1 ·2H ₂ O [Å]	1 [Å]	2 ·H ₂ O [Å]	3 ·H ₂ O [Å]	3 [Å]	4 ·H ₂ O [Å]	5 [Å]
V1–O1	1.9926(14)	1.992(3)	1.9783(17)	1.9865(19)	1.9914(14)	2.0013(17)	1.9803(12)
V1–O3	1.9985(14)	1.999(3)	2.0109(17)	1.9746(17)	2.0051(13)	1.9869(16)	1.6212(15)
V1–O5	1.6299(14)	1.603(3)	1.6212(13)	1.6186(13)	1.6078(14)	1.6133(13)	
V1–O6	1.6127(15)	1.612(3)			1.6311(14)		
V1–N1	2.0941(15)	2.087(3)	2.0848(19)	2.0941(18)	2.0993(13)	2.0912(18)	2.0824(19)
V2–O7		1.989(3)					
V2–O9		1.981(3)					
V2–O11		1.603(3)					
V2–O12		1.619(3)					
V2–N2		2.088(3)					
							
O1–C1	1.296(2)	1.282(5)	1.305(3)	1.302(3)	1.291(2)	1.294(3)	1.298(2)
O2–C1	1.224(2)	1.226(5)	1.217(3)	1.213(3)	1.222(2)	1.213(3)	1.213(2)
O3–C7	1.292(2)	1.303(5)	1.281(3)	1.299(3)	1.309(2)	1.304(3)	
O4–C7	1.224(2)	1.222(5)	1.234(3)	1.215(3)	1.212(2)	1.211(3)	
O7–C8		1.310(5)					
O8–C8		1.210(5)					
O9–C14		1.294(5)					
O10–C14		1.217(5)					

Table S2. Selected bond angles for compounds **1–5**.

	1 ·2H ₂ O	1	3
Angle	[°]	[°]	[°]
O1–V1–O3	148.95(6)	148.71(12)	147.97(5)
O1–V1–O5	98.85(7)	100.15(15)	98.95(7)
O1–V1–O6	99.27(7)	98.98(15)	100.64(7)
O3–V1–O5	98.55(7)	98.13(16)	99.66(7)
O3–V1–O6	98.93(7)	97.92(15)	98.11(7)
O5–V1–O6	109.37(8)	110.85(18)	107.92(7)
N1–V1–O1	74.59(6)	74.38(12)	74.86(5)
N1–V1–O3	74.43(6)	74.36(12)	73.86(5)
N1–V1–O5	128.17(7)	123.73(15)	113.63(7)
N1–V1–O6	122.46(7)	125.40(14)	138.42(7)
			
O7–V2–O9		148.89(12)	
O7–V2–O11		97.25(14)	
O7–V2–O12		98.90(15)	
O9–V2–O11		99.29(15)	
O9–V2–O12		99.78(14)	
O11–V2–O12		110.30(17)	
N2–V2–O7		74.45(12)	
N2–V2–O9		74.49(11)	
N2–V2–O11		125.37(14)	
N2–V2–O12		124.31(14)	

Angle	2 ·H ₂ O	3 ·H ₂ O	4 ·H ₂ O
	[°]	[°]	[°]
O1–V1–O3	149.16(7)	148.86(7)	149.22(7)
O1–V1–O5	99.01(5)	98.55(6)	98.39(5)
O3–V1–O5	98.66(5)	99.59(6)	99.20(5)
O5–V1–O5 ⁱ	109.39(9)	108.07(10)	109.66(10)
N1–V1–O1	74.71(7)	73.94(7)	74.53(7)
N1–V1–O3	74.45(7)	74.93(7)	74.69(7)
N1–V1–O5	125.30(5)	125.95(5)	125.17(5)
Symmetry code:	(i) $x, -y + \frac{1}{2}, z$	(i) $x, -y + \frac{1}{2}, z$	(i) $x, y, -z + \frac{1}{2}$

Angle	5
	[°]
O1–V1–O1 ⁱⁱ	150.04(7)
O1–V1–O3	98.52(7)
O1–V1–O3 ⁱⁱ	98.43(6)
O3–V1–O3 ⁱⁱ	110.49(13)
N1–V1–O1	75.02(4)
N1–V1–O3	124.76(6)
Symmetry code:	(ii) $-x + 1, y, -z + 2$

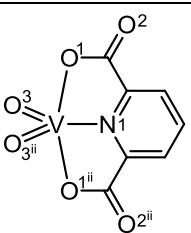
	
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Table S3 Hydrogen bonds for **1**·2H₂O and anhydrous **1** [Å and °]

D–H···A	<i>d</i> (D–H)	<i>d</i> (H···A)	<i>d</i> (D···A)	<(DHA)	Symmetry transformation for acceptors
1 ·2H ₂ O					
N2–H2···O8	0.88	1.80	2.661(2)	167	<i>x</i> , <i>y</i> , <i>z</i>
O7–H7A···O4	0.83(2)	1.99(2)	2.807(2)	169(5)	<i>x</i> , <i>y</i> + 1, <i>z</i> + 1
O7–H7B···O5	0.83(2)	2.03(3)	2.796(2)	153(5)	– <i>x</i> , – <i>y</i> + 1, – <i>z</i> + 1
O8–H8A···O2	0.84(2)	2.09(2)	2.877(2)	158(3)	– <i>x</i> + 1, – <i>y</i> + 2, – <i>z</i> + 1
O8–H8B···O7	0.84(2)	1.83(2)	2.670(3)	173(3)	<i>x</i> , <i>y</i> , <i>z</i>
C3–H3···O6	0.95	2.41	3.208(2)	142	<i>x</i> , <i>y</i> + 1, <i>z</i>
C5–H5···O5	0.95	2.57	3.202(2)	124	– <i>x</i> , – <i>y</i> + 1, – <i>z</i>
C5–H5···O6	0.95	2.58	3.271(2)	130	– <i>x</i> + 1, – <i>y</i> + 1, – <i>z</i>
C8–H8···O1	0.95	2.25	3.150(3)	157	<i>x</i> , <i>y</i> , <i>z</i>
C10–H10···O2	0.95	2.56	3.231(3)	128	<i>x</i> + 1, <i>y</i> – 1, <i>z</i>
1					
N3–H3N···O7	0.86	2.54	3.129 (6)	126	<i>x</i> , <i>y</i> , <i>z</i>
N4–H4N···O11	0.86	2.19	2.876 (6)	137	<i>x</i> , <i>y</i> , <i>z</i>
C10–H10···O12	0.93	2.49	3.402(5)	166	<i>x</i> – 1, <i>y</i> , <i>z</i>
C15–H15···O3	0.93	2.58	3.218(6)	126	<i>x</i> , <i>y</i> , <i>z</i>
C21–H21···O5	0.93	2.58	3.476(6)	163	– <i>x</i> + 1, – <i>y</i> + 1, – <i>z</i> + 1
C22–H22···O10	0.93	2.53	3.403(7)	156	<i>x</i> + 1, <i>y</i> – 1, <i>z</i>
C23–H23···O7	0.93	2.56	3.412(6)	153	– <i>x</i> + 1, – <i>y</i> + 1, – <i>z</i>

Table S4 Hydrogen bonds for **2**·H₂O [Å and °]

D–H···A	<i>d</i> (D–H)	<i>d</i> (H···A)	<i>d</i> (D···A)	<(DHA)	Symmetry transformation for acceptors
N2–H2···O7	0.88	1.80	2.677(3)	177	<i>x</i> , <i>y</i> , <i>z</i>
O6–H6O···O4	0.84	1.74	2.577(2)	175	<i>x</i> , <i>y</i> , <i>z</i>
O7–H7O···O5	0.83(1)	2.00(1)	2.8090(18)	166(2)	– <i>x</i> + 1, – <i>y</i> + 1, – <i>z</i> + 1
C3–H3···O5	0.95	2.56	3.280(2)	132	– <i>x</i> + 1, – <i>y</i> + 1, – <i>z</i>
C3–H3···O5	0.95	2.56	3.280(2)	132	– <i>x</i> + 1, <i>y</i> – ½, – <i>z</i>
C5–H5···O1	0.95	2.40	3.144(3)	135	<i>x</i> + 1, <i>y</i> , <i>z</i>
C8–H8···O2	0.95	2.55	3.475(3)	164	<i>x</i> + 1, <i>y</i> , <i>z</i> + 1
C10–H10···O3	0.95	2.55	3.442(3)	157	<i>x</i> + 1, <i>y</i> , <i>z</i>
C11–H11···O4	0.95	2.50	3.143(3)	125	<i>x</i> , <i>y</i> , <i>z</i>

Table S5 Hydrogen bonds for **3**·H₂O and anhydrous **3** [Å and °]

D–H···A	<i>d</i> (D–H)	<i>d</i> (H···A)	<i>d</i> (D···A)	<(DHA)	Symmetry transformation for acceptors
3 ·H ₂ O					
O6–H6O···O5	0.84	1.93	2.7624(18)	168	<i>x</i> , <i>y</i> , <i>z</i>
N2A–H2A···O1	0.88	1.89	2.765(4)	173	<i>x</i> , <i>y</i> , <i>z</i> – 1
N3A–H3A···O4	0.88	1.97	2.844(3)	176	<i>x</i> , <i>y</i> , <i>z</i>
N3A–H3B···O6	0.88	1.94	2.819(4)	175	– <i>x</i> , <i>y</i> + ½, – <i>z</i> + 1
N2B–H2B···O6	0.88	1.94	2.770(8)	156	– <i>x</i> , <i>y</i> + ½, – <i>z</i> + 1
N3B–H3C···O6	0.88	2.11	2.978(7)	169	– <i>x</i> + 1, <i>y</i> + ½, – <i>z</i> + 1
N3B–H3D···O1	0.88	2.28	3.130(7)	164	<i>x</i> , <i>y</i> , <i>z</i> – 1
N3B–H3D···O2	0.88	2.43	3.153(8)	140	<i>x</i> , <i>y</i> , <i>z</i> – 1
3					
N2–H2···O3	0.86	1.92	2.783(2)	178	<i>x</i> , <i>y</i> , <i>z</i>
N3–H3A···O6	0.86	2.15	3.004(2)	170	– <i>x</i> + 1, <i>y</i> + ½, – <i>z</i> + ½
N3–H3B···O2	0.86	2.04	2.888(2)	171	<i>x</i> – 1, – <i>y</i> + ½, <i>z</i> – ½
C3–H3···O6	0.93	2.57	3.477(3)	165	– <i>x</i> + 2, <i>y</i> – ½, – <i>z</i> + 1½
C8–H8···O5	0.93	2.48	3.084(3)	122	– <i>x</i> + 1, – <i>y</i> , – <i>z</i> + 1
C12–H12···O4	0.93	2.55	3.230(3)	130	<i>x</i> , – <i>y</i> + ½, <i>z</i> – ½

Table S6 Hydrogen bonds for **4**·H₂O [Å and °]

D–H···A	<i>d</i> (D–H)	<i>d</i> (H···A)	<i>d</i> (D···A)	<(DHA)	Symmetry transformation for acceptors
N2–H2···O6	0.86	1.98	2.824(3)	168	<i>x</i> , <i>y</i> , <i>z</i>
N2–H2···O6	0.86	1.98	2.824(3)	168	<i>x</i> , <i>y</i> , – <i>z</i> + 1½
O6–H6A···O5	0.82(1)	2.38(3)	3.071(3)	144(5)	<i>x</i> , – <i>y</i> + 1½, – <i>z</i> + 1
O6–H6B···O1	0.82(1)	2.33(2)	3.140(3)	169(5)	<i>x</i> , <i>y</i> , <i>z</i>
O6–H6B···O2	0.82(1)	2.53(4)	3.080(3)	126(4)	<i>x</i> , <i>y</i> , <i>z</i>
C8–H8···O2	0.93	2.38	3.222(3)	151	<i>x</i> + 1, <i>y</i> , <i>z</i>
C11–H11···O4	0.93	2.43	3.358(3)	173	– <i>x</i> + 1, <i>y</i> – ½, – <i>z</i> + 1½
C14–H14B···O3	0.96	2.56	3.463(3)	157	– <i>x</i> + 2, <i>y</i> – ½, – <i>z</i> + 1½

Table S7 Hydrogen bonds for **5** [Å and °]

D–H···A	<i>d</i> (D–H)	<i>d</i> (H···A)	<i>d</i> (D···A)	<(DHA)	Symmetry transformation for acceptors
N2–H2N···O2	0.86	2.24	2.8288 (19)	126	<i>x</i> , <i>y</i> , <i>z</i>
N2–H2N···O3	0.86	2.12	2.801 (2)	135	<i>x</i> + ½, <i>y</i> + ½, <i>z</i>
C3–H3···O3	0.93	2.46	3.359(4)	162	<i>x</i> + ½, <i>y</i> + ½, <i>z</i>
C5–H5···O1	0.93	2.58	3.323(3)	137	<i>x</i> + ½, <i>y</i> + ½, <i>z</i>
C5–H5···O1	0.93	2.58	3.323(3)	137	– <i>x</i> + 1½, <i>y</i> + ½, – <i>z</i> + 1

Table S8 Geometrical parameters (Å, °) for $\pi \cdots \pi$ stacking interactions in **1**·2H₂O, **1**, **4**·H₂O and **5**

<i>Cg</i> I⋯ <i>Cg</i> J	<i>Cg</i> I⋯ <i>Cg</i> J	α	β ($^\circ$)	<i>Cg</i> I-Perp (offset distance)	Ring slippage	Symmetry transformation for acceptors
1 ·2H ₂ O						
<i>Cg</i> 4⋯ <i>Cg</i> 4	4.1093(13)	0	51.30	3.0515(8)	3.809	− <i>x</i> , − <i>y</i> + 1, − <i>z</i> + 1
<i>Cg</i> 4 is N2/C8–C12 ring centroid.						
1						
<i>Cg</i> 7⋯ <i>Cg</i> 8	3.878(4)	6.3(3)	18.28, 16.77	−3.713(2)	/	<i>x</i> , <i>y</i> , <i>z</i>
<i>Cg</i> 7⋯ <i>Cg</i> 8	4.230(4)	6.3(3)	30.27, 34.12	3.502(2)	/	<i>x</i> − 1, <i>y</i> , <i>z</i>
<i>Cg</i> 7 and <i>Cg</i> 8 are N3/C15–C19 and N4/C20–C24 ring centroids, respectively.						
4 ·H ₂ O						
<i>Cg</i> 3⋯ <i>Cg</i> 4	3.8718(7)	0	30.97	3.3199	1.992	<i>x</i> − 1, − <i>y</i> + 1½, − <i>z</i> + 1
<i>Cg</i> 3 and <i>Cg</i> 4 are N1/C2–C6 and N2/C8–C12 ring centroids, respectively.						
5						
<i>Cg</i> 3⋯ <i>Cg</i> 4	3.6787(13)	0.34(10)	25.01, 25.31	3.3255(7)	/	<i>x</i> , <i>y</i> , <i>z</i> + 1
<i>Cg</i> 3 and <i>Cg</i> 4 are N1/C2–C4/C3 ^a /C2 ^a and N2/C5/N2 ^b /C6 ^b /C6 ring centroids, respectively. Symmetry codes: (a) − <i>x</i> + 1, <i>y</i> , − <i>z</i> + 2; (b) − <i>x</i> + 2, <i>y</i> , − <i>z</i> + 1.						
<i>Cg</i> I⋯ <i>Cg</i> J, α , β , γ and <i>Cg</i> I-Perp are the centroid-to-centroid distance between rings I and J (Å), the inter-ring dihedral angle (°), slip angles (°), and the perpendicular distance of <i>Cg</i> I from ring J (Å) , respectively. Ring slippage is in Å.						

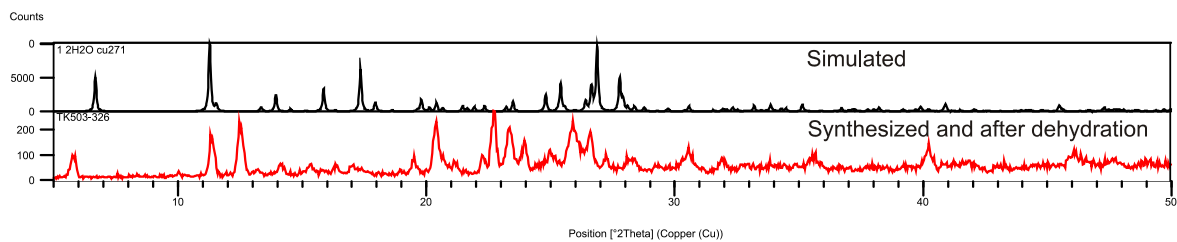


Fig. S8 Comparison of PXRD patterns of the simulated pattern from the single-crystal structure determination and the as-synthesized product after dehydration of $1 \cdot 2\text{H}_2\text{O}$ to $1 \cdot \text{H}_2\text{O}$.

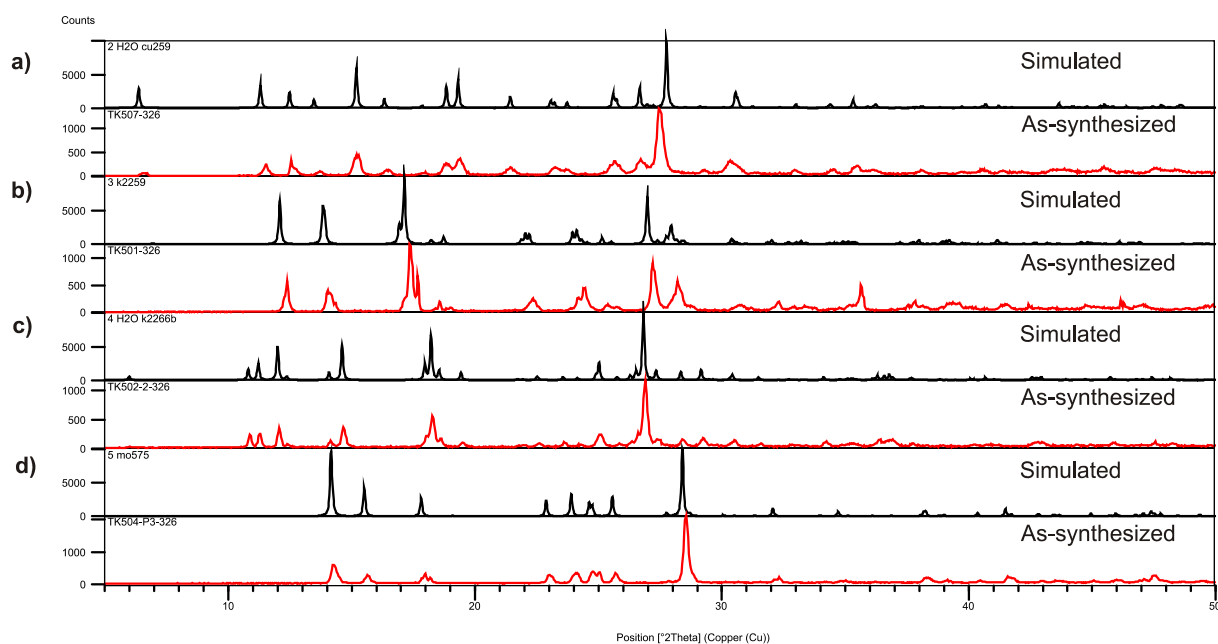


Fig. S9 Comparison of PXRD patterns of the simulated pattern from the single-crystal structure determination and the as-synthesized products in compounds a) $2 \cdot \text{H}_2\text{O}$, b) **3**, c) $4 \cdot \text{H}_2\text{O}$ and d) **5**.