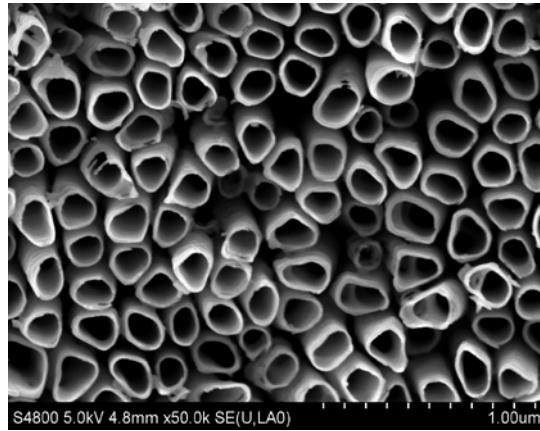


## Electronic Supporting Information

### Fabrication of coral/double-wall TiO<sub>2</sub> nanotube arrays film electrode with higher photoelectrocatalytic activity under sunlight

Yan Liu<sup>†</sup>, Kangsheng Mu<sup>†</sup>, Gang Yang, Hong Peng, Fei Shen, Lilin Wang, Shihuai Deng, Xiaohong Zhang, Yanzong Zhang<sup>\*</sup>

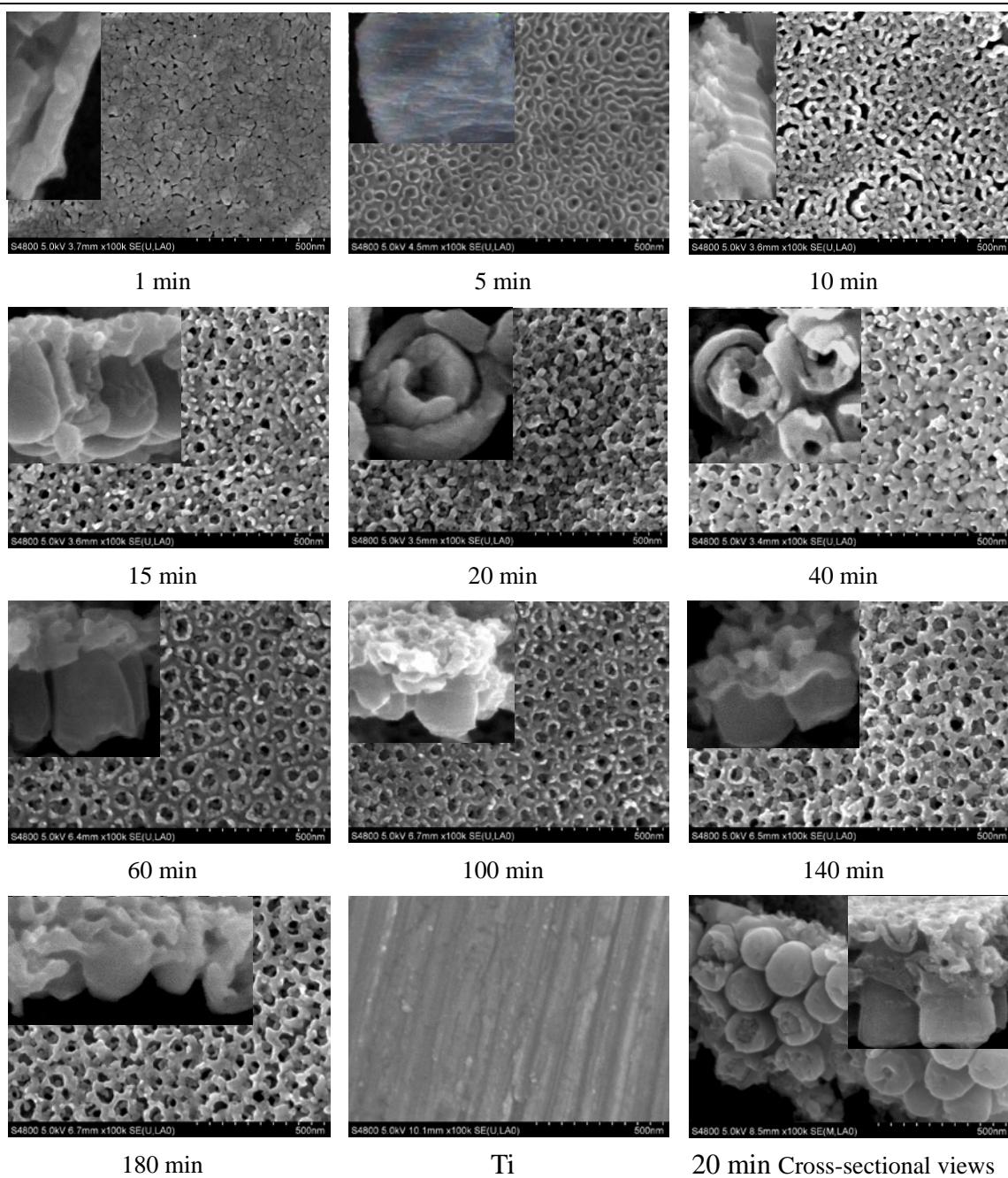
Sichuan Engineering and Technical Center for Rural Environmental Protection, College of the Environment, Sichuan Agricultural University, Chengdu 611130, Sichuan, PR China



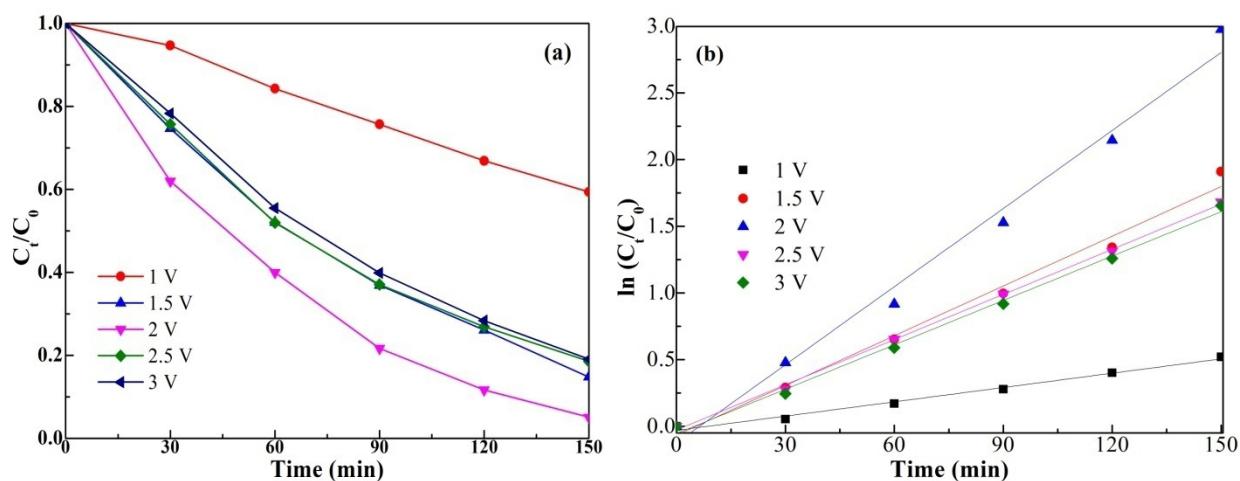
**Figure S1.** TiO<sub>2</sub> nanotube arrays formed at 60 V in 0.6 wt% NH<sub>4</sub>F-10 vol% H<sub>2</sub>O-EG electrolyte for 20 min.

<sup>†</sup> These authors contributed equally to this work (co-first authors).

\* Corresponding author: E-mail address: yzzhang@sicau.edu.cn (Y. Zhang).



**Figure S2.** FE-SEM top-views of as-prepared Coral/DWNTs  $\text{TiO}_2$  films at 60 V in 0.6 wt% [BMIM][BF<sub>4</sub>]/10 vol% H<sub>2</sub>O/EG electrolyte at different anodization time: 1 min, 5 min, 10 min, 15 min, 20 min, 40 min, 60 min, 100 min, 140 min, 180 min.



**Figure S3** (a) PEC degradation of MO at different bias voltage. (b)  $\ln(C_t/C_0)$  as a function of irradiation time in PEC degradation of MO at different bias voltage.

**Table S1** Kinetic constants and regression coefficients for PEC degradation at different bias voltage

Bias voltage (V)	Kinetic constant ( $\text{min}^{-1}$ )	$R^2$
1.0	0.002208	0.9651
1.5	0.005553	0.9954
2.0	0.009748	0.9840
2.5	0.006215	0.9857
3.0	0.005643	0.9986