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Electronic Supplementary Information

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Alginic acid aerogel: a heterogeneous Brønsted acid promoter for the direct Mannich reaction

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General methods and materials

¹H, ¹³C NMR spectra were recorded on a Varian AS 400 or 600 spectrometer. Chemical shifts (δ) are reported in ppm relative to residual solvent signals for ¹H and ¹³C NMR,¹ and using CF₃C₆H₅ as external reference (-63.7 ppm) for ¹⁹F NMR. ¹³C NMR spectra were acquired with ¹H broad band decoupled mode. Chromatographic purifications were performed using 70-230 mesh silica. Mass spectra were recorded on using electronic impact (EI) ionisation techniques. The relative configuration of the major diastereoisomer in compounds **4a-c** and **4e** was determined as *anti* by comparing their NMR spectra with the reported ones, as outlined below. We assume a similar reaction pathway for the remaining compounds **4**, leading to the same *anti* relative configuration for the major diastereoisomer. This assignment is further substantiated by observing the signals related to the α -N protons, which follow a similar pattern in all compounds **4**: in particular, in the minor diatereoisomers of all compounds **4** this proton resonates at higher ppm and features a smaller J coupling constant with the α -CH, compared to the major diastereoisomers.

Analytical grade solvents and commercially available reagents were used as received, unless otherwise stated. To avoid reactivity due to the presence of residual acid, the purity of aldehydes **1** was checked by ¹H NMR before use. Benzaldehyde **1a** was purified by distillation, 4-chlorobenzaldehyde **1b** and 4-fluorobenzaldehyde **1e** by washing a CH₂Cl₂ solution with sat.NaHCO₃, drying with MgSO₄, filtration and evaporation. Sodium alginates Protanal 200S and Protanal 240D, provided by FMC Biopolymer, and commercially available Laminaria Digitata for food uses, purchased from Celnat, were suitably treated to obtain the catalytic materials. Alginic acid powder Satialgine® H8 has been supplied by JRS Pharma and used as obtained.

¹ H. E. Gottlieb, V. Kotlyar and A. Nudelman, J. Org. Chem., 1997, **62**, 7512.

Preparation of the gel materials

Alginic acid alcogel. The catalyst beads were obtained using a procedure previously reported in the literature.² A 2% w/V solution of sodium alginate was obtained, adding 2 g of the desired type of alginate (Protanal 200S, Protanal 240D) to 100 mL of distilled water and stirring it until a clear and viscous solution was obtained. The prepared solution was added dropwise into 400 mL of 1 mol/L HCl at room temperature and the resulting solution system was stirred slowly overnight to allow the maturation of the beads, whose formation is immediately evident. After filtration, the beads were carefully rinsed with distilled water and dehydrated by immersion in a series of ethanol/water baths, with increasing alcohol content (10, 30, 50, 70, 90% and absolute ethanol), for 15 min each.

Alginic acid aerogel beads. After solvent replacement, the gel was filtered and loaded into a stirred flask containing absolute ethanol and molecular sieves for 24 h, to guarantee the total dehydration of the beads. The wet gel was converted to an aerogel (denoted as AG-1 or AG-2, depending on the composition) by low temperature drying under supercritical CO₂ conditions (74 bar, 31.5 °C) in a Polaron 3100 apparatus.

Alginic acid xerogel beads. After solvent replacement, the gel was filtered and dried by evaporative drying using a rotary evaporator, at 60 °C for almost 1h, to obtain the final material (denoted as XG).

Alginic acid solvogel. The alginic acid beads, after maturation in the acidic solution, were filtered, rinsed with distilled water and dehydrated using a series of solvent/water baths (es. acetonitrile/water instead of ethanol/water ones), with increasing acetonitrile content to obtain the desired solvent mixture (80:20 acetonitrile/water, denoted as SG).

Laminaria digitata aerogel. The desired amount of Laminaria Digitata was washed using distilled water at 60 °C, to remove salts present on the surface, and immersed in a 1 M solution of HCl, stirring it slowly overnight. The amount of HCl necessary to guarantee a perfect acidification was calculated considering the presence of 45% alginic acid groups in the seaweed.³ After filtration, the seaweed was carefully rinsed with distilled water and dehydrated by immersion in a series of

² F. Quignard, R. Valentin and F. Di Renzo, New J. Chem., 2008, **32**, 1300.

³ E. Fourest and B. Volesky, Appl. Biochem. Biotech., 1997, 67, 33.

ethanol/water baths, with increasing alcohol content (10, 30, 50, 70, 90% and absolute ethanol), for 15 min each. The seaweed was then, dried following the procedure described for the alginic acid aerogel obtaining the final material (denoted as S).

Surface area of the catalytic materials

Surface areas were measured by the BET method by nitrogen gas adsorption/desorption at 77 °K, using a Micrometrics *Tristar* apparatus on samples outgassed at 50 °C for 6 hours.

Polysaccharide gel	Surface area (m ² g ⁻¹)
AG-1	250
AG-2	280
XG	<2
S	122

Optimisation of reaction parameters: selected results

Stoichiometric ratio



Entry	3a:5	Y ¹ 4a (%)	Y ¹ 6 (%)
1	1:2	27	7
2	1:1	56	16
3	2:1	54	11
4	5:1	58	21

¹ The yield was calculated by ¹H-NMR and refers to the product: imine ratio, without use of an internal standard.

A 2:1 ketone/imine ratio was selected because it allowed obtaining the best product/byproduct ratio.

Reactions in pure water



Entry	AG-1	additive	Y ¹ 4a (%)
1	yes	-	<10
2	yes	Aliquat-336	17
3	no	Aliquat-336	<10
4	yes	SDS	54
5	no	SDS	91

¹ Yield calculated by ¹H NMR using bibenzyl as internal standard.

The reaction without phase-transfer catalyst or with aliquat-336 did not proceed, presumably for the poor solubility of the reactants. In the reactions in the presence of SDS, alginic acid did not show catalytic activity. We attribute the lower yields obtained in the reactions with SDS and AG-1, compared to the reactions in the absence of AG-1, to the capability of alginic acid in sequestering

the surfactant from the reaction mixture, and thus preventing the reaction from occurring. Reactions in pure water are thus not useful for our purposes.

Concentration



Entry	Conc (M)	Y ¹ Catalysed (%)	Y ¹ Blank reaction (%)
1	0.1	43	21
2	0.2	78	34
3	0.35	91	22
4	0.5	95	61

A 0.35M concentration was selected for the following tests considering the higher yields and the presence of a limited background reaction in absence of the catalyst.

Amount of catalyst



Entry	t (h)	Y ¹ 10 mol% AG-1 (%)	Y ¹ 20 mol% AG-1 (%)
1	5	26	53
2	16	70	82
3	24	75	91

³ ²⁴ ⁷⁵ ⁹¹ ¹ The yield was calculated by ¹H-NMR and refers to the product:imine ratio, without use of an internal standard

A higher amount of catalyst led to a slightly higher yield. Considering its cheapness, a 20 mol% catalyst loading was used throughout.

Reaction temperature



¹ Yield calculated by ¹H NMR using bibenzyl as internal standard.

Room temperature appears as the most suitable temperature to perform this AG-1 catalysed transformation, as neither lower nor higher temperatures are advantageous.

Kinetic study on the 2C Mannich reaction

In order to ascertain the optimal reaction time for the alginic acid aerogel AG-1 catalysed Mannich reaction (i.e. reaching a satisfactory yield while keeping at acceptable levels the difference with the non-catalysed process), the evolution (**4a** yield vs time) of the 2C Mannich reaction between imine **5** and cyclohexanone **3a** (2 equiv.), under the optimal reaction conditions (CH₃CN/H₂O 8:2 solvent mixture, 20 mol% catalyst loading, 0.35 M, RT) was determined by ¹H NMR using bibenzyl as internal standard. Instead of sampling from a single reaction, which would modify the catalyst loading due to its heterogeneous nature, it was preferred to run a different experiment for each point of the curve corresponding to the catalysed process.



¹H NMR yield of **4a** vs time in the AG-**1** catalysed (20 mol%) Mannich reaction between **5** and **3a** (2 equiv.), and comparison with the non-catalysed process. Conditions: imine **5** (0.15 mmol), ketone **3a** (0.30 mmol), AG-**1** (20 mol%), CH₃CN/H₂O 8:2 (480 μ L), bibenzyl (0.075 mmol).

A reaction time of ca 24 h gives a satisfactory yield in the product **4a**, with a minor non-catalysed contribution. Seen that the reaction considerably slows down after approximately 9 h, we could confirm 18-20 h as the optimal time lapse for our experiments.



Furthermore, the curve is fitting with a pseudo-first order reaction rate, as shown below:

First-order kinetics plots of AG-1 catalysed (20 mol%) Mannich reaction between 5 and 3a (2 equiv.). Ct: concentration at a given time, t.

Optimised procedure and products characterisation

General procedure for the catalytic 3C Mannich reaction:

To a 3 mL screw cap vial were sequentially added the aldehyde **1** (0.168 mmol), the solvent mixture (CH₃CN/H₂O 8:2, 480 μ L), the aniline **2** (0.168 mmol), the ketone **3** (0.336 mmol, 2 equiv.), and 6 mg of aerogel beads (12 beads, 0.5 mg per bead, 20 mol% catalyst loading considering full availability of carboxylic acid moieties). After screw-capping the vial, the mixture was gently shaken at RT for 18-20 h. The supernatant was filtered through a Pasteur pipette packed with cotton in a flask, the beads carefully washed with EtOAc or CH₂Cl₂ (3 x), the washings filtered through the same pipette and added to the same flask. The solvents were evaporated to dryness under reduced pressure, and the residue analysed by ¹H NMR spectroscopy to determine the diastereomeric ratio. The pure Mannich adducts **4** were obtained by purification of the crude by chromatography on silica gel. As detailed below, two purifications were usually necessary.

2-(Phenyl(phenylamino)methyl)cyclohexan-1-one (4a)

2-((4-Chlorophenyl)(phenylamino)methyl)cyclohexan-1-one (4b)

Following the general procedure, the title compound was obtained as a white solid in 78% yield, after two chromatographic purifications (one with *n*-hexane/EtOAc 8:2, one with CH_2Cl_2). The diastereomeric ratio of the product, determined by ¹H NMR on the crude mixture, was found to be 63:37, favouring the *anti*-isomer. ¹H

⁴ Y.-Y. Yang, W.-G. Shou and Y.-G. Wang, *Tetrahedron*, 2006, **62**, 10079.

NMR (CDCl₃, 400 MHz): $\delta = 7.35-7.23$ (m, 4H_{maj}, 4H_{min}), 7.11-7.04 (m, 2H_{maj}, 2H_{min}), 6.70-6.62 (m, 1H_{maj}, 1H_{min}), 6.55-6.48 (m, 2H_{maj}, 2H_{min}), 4.74 (d, J = 4.6 Hz, 1H_{min}), 4.66 (br s, 1H_{maj}, 1H_{min}), 4.60 (d, J = 6.1 Hz, 1H_{maj}), 2.82-2.69 (m, 1H_{maj}, 1H_{min}), 2.46-2.38 (m, 1H_{maj}, 1H_{min}), 2.37-2.26 (m, 1H_{maj}, 1H_{min}), 2.09-1.54 (m, 6H_{maj}, 6H_{min}); ¹³C NMR (CDCl₃, 400 MHz): δ [aromatic CH signals not assigned] = 212.4 (maj), 212.1 (min), 147.1 (min), 147.0 (maj), 140.4 (maj), 140.0 (min), 132.8 (maj), 132.7 (min), 129.1, 129.1, 129.0, 128.7, 128.6, 128.5, 117.9, 117.7, 114.0, 113.6, 57.5 (min), 57.3 (maj), 56.9 (min), 56.4 (maj), 42.4 (min), 42.0 (maj), 31.5 (maj), 28.9 (min), 27.8 (maj), 27.0 (min), 24.9 (min), 24.0 (maj). Spectral data are consistent with literature values,⁵ which also allowed assignment of the relative configuration of the major diastereoisomer as *anti*.

2-((4-Nitrophenyl)(phenylamino)methyl)cyclohexan-1-one (4c)

Following the general procedure, the title compound was obtained as a white solid in 76% yield after two chromatographic purifications (one with CH₂Cl₂, one with *n*-hexane/EtOAc 8:2). The diastereomeric ratio of the product, determined by ¹H NMR on the crude mixture, was found to be 64:36, favouring the *anti*-isomer. ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.17-8.12$ (m, 2H_{maj}, 2H_{min}), 7.60-7.53 (m, 2H_{maj}, 2H_{min}), 7.11-7.05 (m, 2H_{maj}, 2H_{min}), 6.71-6.64 (m, 1H_{maj}, 1H_{min}), 6.52-6.47 (m, 2H_{maj}, 2H_{min}), 4.86 (d, J = 4.5 Hz, 1H_{min}), 4.85 (br s, 1H_{maj}), 4.71 (d, J = 5.2 Hz, 1H_{maj}), 4.60 (br s, 1H_{min}), 2.89-2.81 (m, 1H_{maj}, 1H_{min}), 2.48-2.28 (m, 2H_{maj}, 2H_{min}), 2.11-1.55 (m, 6H_{maj}, 6H_{min}). ¹³C NMR (CDCl₃, 400 MHz): δ [aromatic C signals not assigned] = 211.8 (maj), 210.6 (min), 149.9, 149.6, 147.1, 146.6, 129.3, 129.2, 128.6, 128.2, 123.7, 123.6, 118.4, 118.1, 114.0, 113.5, 57.8 (maj), 57.2 (min), 57.0 (maj), 56.2 (min), 42.5 (min), 42.4 (maj), 32.0 (maj), 29.0 (min), 27.8 (maj), 27.0 (min), 24.9 (min), 24.5 (maj). Spectral data are consistent with literature values,⁶ which also allowed assignment of the relative configuration of the major diastereoisomer as *anti*.

2-(Naphthalen-2-yl(phenylamino)methyl)cyclohexanone (4d)



Following the general procedure, the title compound was obtained as a pale yellow solid in 60% yield, after two chromatographic purifications (one with *n*-hexane/EtOAc 7:3, one with CH_2Cl_2). The diastereomeric ratio of the product,

determined by ¹H NMR on the crude mixture, was found to be 61:39, favouring the *anti*-isomer. ¹H NMR (CDCl₃, 400 MHz): $\delta = 7.86-7.78$ (m, 4H_{maj}, 4H_{min}), 7.55-7.41 (m, 3H_{maj}, 3H_{min}), 7.10-7.02 (m, 2H_{maj}, 2H_{min}), 6.07-6.56 (m, 3H_{maj}, 3H_{min}), 4.98 (d, J = 4.3 Hz, 1H_{min}), 4.81 (d, J = 7.3 Hz,

⁵ anti-isomer: Q.-X. Guo, H. Liu, C. Guo, S.-W. Luo, Y. Gu and L.-Z. Gong, J. Am. Chem. Soc., 2007, **129**, 3790; syn isomer: X. Zheng, Y.-B. Qian and Y. Wang, *Eur. J. Org. Chem.*, 2010, 515.

⁶ B. Eftekhari-Sis, A. Abdollahifar, M. M. Hashemi and M. Zirak, Eur. J. Org. Chem., 2006, 5152.

1H_{maj}), 4.74 (br s, 1H_{maj}, 1H_{min}), 2.93-2.81 (m, 1H_{maj}, 1H_{min}), 2.50-2.41 (m, 1H_{maj}, 1H_{min}), 2.40-2.28 (m, 1H_{maj}, 1H_{min}), 2.15-1.49 (m, 6H_{maj}, 6H_{min}). ¹³C NMR (CDCl₃, 400 MHz): δ [aromatic signals not assigned] = 212.8 (min), 211.3 (maj), 147.5, 147.2, 139.3, 139.2, 133.3, 133.3, 132.8, 132.7, 129.0, 129.0, 128.4, 128.1, 127.9, 127.9, 127.6, 127.6, 126.4, 126.3, 126.1, 126.0, 125.7, 125.6, 125.1, 117,8, 117.6, 114.2, 113.7, 58.2 (min), 57.5 (maj), 57.4 (min), 56.7 (maj), 42.4 (min), 41.8 (maj), 31.3 (maj), 28.7 (min), 27.9 (maj), 27.0 (min), 24.9 (min), 23.7 (maj). EI-MS: m/z = 329 [M⁺, 3], 231 [M⁺-98, 83], 230 [M⁺-99, 100]. The relative configuration of the major diastereoisomer was tentatively assigned as *anti* by analogy with the remaining compounds **4**.

2-((4-Chlorophenyl)((4-chlorophenyl)amino)methyl)cyclohexanone (4e)

Following the general procedure, the title compound was obtained as a yellow solid in 61% yield, after chromatographic purification (*n*-hexane/EtOAc 8:2). The diastereomeric ratio of the product, determined by ¹H NMR on the crude mixture, was found to be 64:36, favouring the *anti*-isomer. ¹H NMR (CDCl₃, 400 MHz): δ = 7.38-7.30 (m, 2H_{maj}, 2H_{min}), 7.12-7.04 (m, 2H_{maj}, 2H_{min}), 7.02-6.93 (m, 2H_{maj}, 2H_{min}), 6.69-6.61 (m, 1H_{maj}, 1H_{min}), 6.56-6.49 (m, 2H_{maj}, 2H_{min}), 4.75 (d, J = 4.2 Hz, 1H_{min}), 4.66 (br s, 1H_{maj}, 1H_{min}), 4.62 (d, J = 6.5 Hz, 1H_{maj}), 2.81-2.69 (m, 1H_{maj}, 1H_{min}), 2.45-2.37 (m, 1H_{maj}, 1H_{min}), 2.37-2.27 (m, 1H_{maj}, 1H_{min}), 2.09-1.53 (m, 6H_{maj}, 6H_{min}); ¹³C NMR (CDCl₃, 400 MHz): δ [aromatic CH signals not assigned] = 212.6 (maj), 211.3 (min), 161.8 (d, J = 246 Hz), 161.7 (d, J = 246 Hz), 137.4 (d, J = 3 Hz), 137.1 (d, J = 3 Hz), 129.1 (d, J = 8 Hz), 129.1, 129.05, 128.8 (d, J = 8 Hz), 117.8, 117.7, 115.4 (d, J = 21 Hz), 115.2 (d, J = 21 Hz), 114.0, 113.6, 57.5 (maj), 57.4 (maj), 56.9 (min), 56.5 (min), 42.5 (min), 42.0 (maj), 31.4 (maj), 29.0 (min), 27.8 (maj), 27.0 (min), 24.9 (min), 23.9 (maj); ¹⁹F NMR (CDCl₃, 300 MHz): δ = -115.7 (m, maj), 116 (m, min). Spectral data are consistent with literature values,⁵ which also allowed assignment of the relative configuration of the major diastereoisomer as *anti*.

2-((4-Chlorophenyl)((4-chlorophenyl)amino)methyl)cyclohexanone (4f)



Following the general procedure but using 5 equiv. of cyclohexanone, the title compound was obtained as a pale yellow solid in 57% yield, after two chromatographic purifications (one with *n*-hexane/EtOAc 7:3, one with CH_2Cl_2). The diastereomeric ratio of the product, determined by ¹H NMR on

the crude mixture, was found to be 53:47, favouring the *anti*-isomer. ¹H NMR (CDCl₃, 400 MHz): δ [NH signal not assigned] = 7.28-7.24 (m, 4H_{maj}, 4H_{min}), 7.03-6.98 (m, 2H_{maj}, 2H_{min}), 6.46-6.37 (m, 2H_{maj}, 2H_{min}), 4.77 (br s, 1H), 4.67 (d, J = 4.1 Hz, 1H_{min}), 4.61 (br s, 1H), 4.51 (d, J = 6.3 Hz,

1H_{maj}), 2.80-2.67 (m, 1H_{maj}, 1H_{min}), 2.46-2.23 (m, 2H_{maj}, 2H_{min}), 2.09-1.48 (m, 6H_{maj}, 6H_{min}). ¹³C NMR (CDCl₃, 400 MHz): δ [signals of both diastereoisomers] = 212.3, 211.1, 145.7, 145.6, 139.9, 139.5, 132.9, 132.8, 129.0, 128.9, 128.8, 128.7, 128.6, 128.6, 122.6, 122.4, 115.2, 114.7, 57.8, 57.2, 57.1, 56.2, 42.4, 42.2, 31.7, 28.7, 27.8, 26.9, 24.8, 24.1. Spectral data are consistent with literature values,⁴ which also allowed assignment of the relative configuration of the major diastereoisomer as *anti*.

2-(((4-Bromophenyl)amino)(4-chlorophenyl)methyl)cyclohexanone (4g)



Following the general procedure but using 5 equiv. of cyclohexanone, the title compound was obtained as a pale yellow solid in 95% yield, after two chromatographic purifications (one with *n*-hexane/Et₂O 6:4, one with *n*-hexane/CH₂Cl₂ 1:1). The diastereomeric ratio of the product, determined by ¹H

NMR on the crude mixture, was found to be 50:50. ¹H NMR (CDCl₃, 400 MHz): $\delta = 7.27-7.15$ (m, 4H, 4H), 7.11-7.01 (m, 2H, 2H), 6.36-6.26 (m, 2H, 2H), 4.74 (br s, 1H, 1H), 4.60 (d, J = 4.1 Hz, 1H), 4.45 (d, J = 5.6 Hz, 1H), 2.75-2.61 (m, 1H, 1H), 2.37-2.17 (m, 2H, 2H), 2.14-1.42 (m, 6H, 6H). ¹³C NMR (CDCl₃, 400 MHz): δ [signals of both diastereoisomers] = 212.2, 211.0, 146.2, 146.1, 139.8, 139.5, 131.8, 131.7, 128.9, 128.6, 128.6, 128.6, 115.6, 115.2, 109.5, 109.3, 57.6, 57.1, 56.9, 56.2, 42.4, 42.2, 31.7, 28.7, 27.9, 26.9, 24.8, 24.1. Spectral data are consistent with literature values.⁴

2-((4-Chlorophenyl)((3,4-dichlorophenyl)amino)methyl)cyclohexanone (4h)



Following the general procedure but using 5 equiv. of cyclohexanone, the title compound was obtained as a pale yellow solid in 90% yield, after two chromatographic purifications (one with *n*-hexane/EtOAc 8:2, one with CHCl₃). The diastereomeric ratio of the product, determined by ¹H NMR on the crude mixture, was found to be 65:35, favouring the *anti*-isomer. ¹H NMR (CDCl₃)

400 MHz): $\delta = 7.30-7.24$ (m, 4H_{maj}, 4H_{min}), 7.08 (d, J = 8.9 Hz, 1H_{maj}), 7.07 (d, J = 8.9 Hz, 1H_{min}), 6.59 (d, J = 2.8 Hz, 1H_{maj}), 6.57 (d, J = 2.7 Hz, 1H_{min}), 6.33 (dd, J = 8.7, 2.7 Hz, 1H_{maj}, 1H_{min}), 4.92 (br s, 1H_{min}), 4.73 (br d, J = 6.8 Hz, 1H_{maj}), 4.64 (br dd, J = 6.1, 4.1 Hz, 1H_{maj}), 4.46 (br d, J = 4.1 Hz, 1H_{min}), 2.81-2.70 (m, 1H_{maj}, 1H_{min}), 2.46-2.25 (m, 2H_{maj}, 2H_{min}), 2.08-1.45 (m, 6H_{maj}, 6H_{min}). ¹³C NMR (CDCl₃, 400 MHz): δ [aromatic signals not assigned] = 211.3 (min), 210.1 (maj), 146.6 (2C), 139.4, 138.9, 133.1, 132.7, 132.6, 130.5, 130.4, 128.9, 128.8, 128.7, 128.5, 120.5, 120.3, 115.3, 114.8, 113.6, 113.2, 57.9 (min), 57.1 (maj + min), 55.9 (maj), 42.4 (maj), 42.3 (min), 32.0 (min), 28.7 (maj), 27.9 (min), 26.8 (maj), 24.8 (maj), 24.3 (min). EI-MS: m/z = 381,383,385 [M⁺, 5,4,2], 284,286,288 [M⁺-97, 100,90,30]. The relative configuration of the major diastereoisomer was tentatively assigned as *anti* by analogy with the remaining compounds **4**.

2-((4-methoxyphenyl)(phenylamino)methyl)cyclohexanone (4i)



Following the general procedure, the title compound was produced in 76% yield, as determined by ¹H NMR analysis of the crude using an internal standard (bibenzyl). It proved not possible to obtain the pure title compound by chromatography on silica gel, due to extensive decomposition (i.e. formation of

the α , β -unsaturated ketone byproduct deriving from aniline elimination). The diastereomeric ratio of the product, determined by ¹H NMR on the crude mixture, was found to be 61:39. The relative configuration of the major diastereoisomer was tentatively assigned as *anti* by analogy with the remaining compounds **4**.

2-((3,4-dimethoxyphenyl)(phenylamino)methyl)cyclohexanone (4j)



Following the general procedure, the title compound was produced in 60% yield, as determined by ¹H NMR analysis of the crude using an internal standard (bibenzyl). It proved not possible to obtain the pure title compound by chromatography on silica gel, due to extensive decomposition (i.e. formation of

the α , β -unsaturated ketone byproduct deriving from aniline elimination). The diastereomeric ratio of the product, determined by ¹H NMR on the crude mixture, was found to be 64:36. The relative configuration of the major diastereoisomer was tentatively assigned as *anti* by analogy with the remaining compounds **4**.

2-((4-Chlorophenyl)((4-methoxyphenyl)amino)methyl)cyclohexanone (4k)



Following the general procedure but using 5 equiv. of cyclohexanone, the title compound was produced in 72% yield, as determined by ¹H NMR analysis of the crude using an internal standard (bibenzyl). It proved not possible to obtain the pure title compound by chromatography on silica gel, due to extensive

decomposition (i.e. formation of the α , β -unsaturated ketone byproduct deriving from aniline elimination) even when using deactivated silica (small 1-2% amounts of Et₃N in the eluent). The diastereomeric ratio of the product, determined by ¹H NMR on the crude mixture, was found to be 64:36. The relative configuration of the major diastereoisomer was tentatively assigned as *anti* by analogy with the remaining compounds **4**.

3-((4-Chlorophenyl)(phenylamino)methyl)tetrahydro-4H-pyran-4-one (4l)

Following the general procedure, the title compound was obtained as a pale yellow solid in 92% yield, after two chromatographic purifications (one with *n*-hexane/Et₂O 5/5, one with *n*-hexane/acetone 8/2). The diastereomeric ratio of the product, determined by ¹H NMR on the crude mixture, was found to be 65:35, favouring the *anti*-isomer. ¹H NMR (CDCl₃, 400 MHz): $\delta = 7.37-7.35$ (m, 2H_{maj}), 7.33-7.29 (m, 2H_{maj}), 7.28 (br s, 4H_{min}), 7.13-7.04 (m, 2H_{maj}, 2H_{min}), 6.71-6.65 (m, 1H_{maj}, 1H_{min}), 6.55-6.48 (m, 2H_{maj}, 2H_{min}), 4.86 (d, J = 4.9 Hz, 1H_{min}), 4.80 (d, J = 9.0 Hz, 1H_{maj}), 4.70 (br s, 1H_{maj}, 1H_{min}), 4.19-4.12 (m, 1H_{maj}), 4.10-3.93 (m, 4H_{min}), 3.88-3.76 (m, 2H_{maj}), 3.72 (dd, J = 11.9, 4.0 Hz, 1H_{maj}), 2.84 (br q, J = 5.3 Hz, 1H_{min}), 2.78-2.50 (m, 2H_{maj}, 2H_{min}), 2.43 (dtd, J_d = 14.6, 1.4 Hz, J_t = 4.2 Hz, 1H_{maj}). ¹³C NMR (CDCl₃, 400 MHz) δ [signals of both diastereoisomers] = 207.8, 206.9, 146.3, 146.1, 139.1, 138.9, 133.4, 133.2, 129.2, 129.1, 129.0, 128.9, 128.6, 128.5, 118.2 (2C), 113.8, 113.7, 69.6, 68.6, 68.5, 67.9, 58.9, 56.9, 56.4, 55.7, 42.3, 41.4. EI-MS: m/z = 315,317 [M⁺, 5,2], 216,218 [M⁺-99, 100,30]. The relative configuration of the major diastereoisomer was tentatively assigned as *anti* by analogy with the remaining compounds **4**.

2-((4-Chlorophenyl)((4-chlorophenyl)amino)methyl)cyclopentanone (4m)



Following the general procedure, the title compound was obtained as a pale yellow solid in 50% yield, after two chromatographic purifications (one with *n*-hexane/Et₂O 6:4, one with *n*-hexane/CH₂Cl₂ 1:1). The diastereomeric ratio of the product, determined by ¹H NMR on the crude mixture, was found to be 61:39,

favouring the *anti*-isomer. ¹H NMR (CDCl₃, 400 MHz): $\delta = 7.31-7.29$ (m, 4H_{maj}), 7.28-7.24 (m, 2H_{min}), 7.21-7.16 (m, 2H_{min}), 7.03-6.97 (m, 2H_{maj}, 2H_{min}), 6.47-6.39 (m, 2H_{maj}, 2H_{min}), 5.53 (br d, J = 7.9 Hz, 1H_{min}), 5.20 (br s, 1H_{maj}), 4.64 (br dd, J = 7.0, 3.7 Hz, 1H_{min}), 4.45 (d, J = 7.9 Hz, 1H_{maj}), 2.75-2.65 (m, 1H_{min}), 2.48-2.24 (m, 2H_{maj}, 1H_{min}), 2.17-2.02 (m, 1H_{maj}, 1H_{min}), 1.98-1.58 (m, 4H_{maj}, 4H_{min}). ¹³C NMR (CDCl₃, 400 MHz) δ [signals of both diastereoisomers] = 220.3, 218.9, 145.7, 144.9, 139.8, 138.7 (2C), 133.3, 125.9, 128.9, 128.9, 128.8, 128.8, 128.4, 122.8, 122.3, 115.3, 114.7, 58.7, 57.4, 53.8, 52.9, 39.7, 39.2, 26.7, 25.9, 20.6, 20.4. EI-MS: m/z = 333,335 [M⁺, 5,3], 250,252 [M⁺-83, 100,55]. The relative configuration of the major diastereoisomer was tentatively assigned as *anti* by analogy with the remaining compounds **4**.

Copies of the ¹H, ¹³C and ¹⁹F NMR spectra of products 4

al89_crom.protone

Sample: a189_crom.protone File: home/ricci/Spettri/Asja/a189_crom.protone.fid

Pulse Sequence: s2pul Solvent: cdcl3 Temp. 25.0 C / 298.1 K Operator: ricci File: al89_crom.protone Hercury-4008B *m400*

Relax. delay 1.000 sec Pulee 45.0 degrees Acq. time 2.733 sec Width 6398.0 Hz 8 repetitions OBSERVE H1, 399.9245864 MHz DA7A PROCESSING FT size 6536 Total time 37 min, 3 sec



al89_fras_carbon

File: home/ricci/Spettri/Asja/al89_fras_carbon.fid

Pulse Sequence: s2pul Solvent: cdol3 Temp. 25.0 C / 298.1 K Operator: ricci Pile: al89_fras_carbon Hercury-4008B "m400"

Relax. delay 1.000 mec Pulee 45.0 degrees Acq. time 1.300 mec Width 24154.6 HE 512 repetitions 0858EVB 213, 100.5511145 MHE DECOUPLE H1, 399.5265566 MHE Power 41 dB continuously cn WALTE-16 modulated DATA PROCESSIM2 Line broadening 0.5 HE T size 65556 Total time 3 hr, 24 min, 34 sec



LG53 protone

File: home/ricci/Spettri/Lorenzo_Geraci/LG54_protone.fid

Pulse Sequence: s2pul Solvent: cdol3 Temp. 25.0 C / 298.1 K Operator: ricci File: L/254_protone Hercury-4008B "m400"





PLGA-PBG-COOH

Sample: DLCA-DB2-COOH File: home/ricci/Spettri/Lorenso_Geraci/lg_48_crom_protone.fid

Pulse Sequence: s2pul Solvent: cdcl3 Tamp. 25.0 C / 298.1 K Operator: ricci File: 1g_48_crcm_protone Hercury-400BB *m400*

Relax. delay 1.000 sec Pulse 45.0 degrees Acq. time 2.733 sec Width 6398.0 Hz 16 repetitions OBSERVE H1, 399.9245755 MHz DATA PROCESSING FT size 65536 Total time 37 min, 3 sec





Std Carbon experiment

Sample: RP159_lavato_Et20_carbonio File: home/ricci/Spettri/Lorenso_Geraci/lg_48_crom_carbonio.fid

Dulse Sequence: slpul Solvent: cdcl3 Temp. 25.0 C / 230.1 K Operator: ricci Pile: 1g.48_crom_carbonio Hercury-400B8 "m400"

Relax. delay 1.000 sec Pulse 45.0 degrees Acq. time 1.300 sec Width 24154.6 Hz 896 repetitions OBSERVE C13, 100.5611145 MHz DORCOUPLE H1, 399.3265566 MHz Power 41 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.5 Hz FT size 65536 Total time 3 hr, 19 min, 27 sec



L260_crom2_protone

File: home/ricci/Spettri/Lorenzo_Geraci/1g60_crom2_protone.fid

Pulse Sequence: s2pul

Solvent: cdol3 Temp. 25.0 C / 298.1 K Operator: ricci File: 1g60_crom2_protone Hercury-400BB "n400"

Relax. delay 1.000 sec Pulse 45.0 degrees Acq. time 2.733 sec Width 6398.0 Hz 20 repetitions OBSERVE H1, 399.9245825 MHs DATA PROCESSING FT size 65536 Total time 37 min, 3 sec



L260_crom2_carbonio

Sample: RF165_carbonio File: home/ricci/Spettri/Lorenzo_Geraci/1g60_crom2_carbonic.fid

Pulse Sequence: s2pul Solvent: cdcl3 Temp. 25.0 C / 298.1 K Operator: ricci File: 1g60_cron2_carbonic Hercury-40088 "m400"

Relax. delay 1.000 eac pulse 45.0 degrees Acq. time 1.300 eac Width 24154.6 Hz 352 repetitions OBSERVE C13, 100.551145 MHz DECOPDLE H1, 399.9265566 MHz Dever 41 dB continuously on WALTZ-16 modulated DATA PROCESSING Line Froadening 0.5 Hz FT size 65536 Total time 3 hr, 24 min, 34 sec



Std Proton parameters

File: home/ricci/Spettri/Asja/pf.fid









S22





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