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Supporting Information

Vinyl sulfone building blocks in covalently reversible reactions with thiols

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Contents: Crystallographic details, summary of hydrogen bonds in the crystals, electrostatic potentials mapped onto the Hirshfeld surfaces of all compounds **1-9**.

Table S1: Crystallographic Details									
	1	2	3	4	5	6	7	8	9
Empirical formula	$C_{11}H_{12}N_2O_2S_2$	$C_{12}H_{14}N_2O_2S_2$	$C_{17}H_{16}N_2O_2S_2$	$C_{17}H_{16}N_2O_2S_2$	$C_{16}H_{13}FN_2O_2S_2$	$C_{16}H_{13}N_3O_4S_2$	$C_{17}H_{16}N_2O_3S_2$	$C_{16}H_{13}IN_2O_2S_2$	$C_{16}H_{14}N_2O_2S_2$
Formula mass [gmol-1]	268.35	282.37	344.44	344.44	348.40	375.41	360.44	456.30	330.41
T [K]	100(2)	100(2)	100(2)	100(2)	100(2)	100(2)	100(2)	100(2)	100(2)
$\lambda (Mo_{K\alpha}) [Å]$	0.71073	0.71073	0.71073	0.71073	0.71073	0.71073	0.71073	0.71073	0.71073
Crystal system	monoclinic	triclinic	monoclinic	monoclinic	monoclinic	monoclinic	monoclinic	monoclinic	triclinic
Space group (no.)	P2 ₁ /n (14)	<i>P</i> ⁻ (2)	$P2_1/c$ (14)	C2/c (15)	$P2_{1}/c$ (14)	P2 ₁ /n (14)	$P2_{1}/c$ (14)	C 2/c (15)	P (2)
<i>a</i> [Å]	7.9295(16)	8.0971(1)	8.4275(2)	29.3586(16)	8.3288(2)	11.226(2)	8.6156(2)	28.3814(8)	11.0502(3)
<i>b</i> [Å]	11.306(2)	12.4447(2)	16.8324(4)	5.6119(2)	11.2945(2)	10.807(2)	20.5840(4)	6.2777(1)	11.5540(4)
c [Å]	14.113(3)	13.3768(3)	11.6886(2)	24.1160(15)	17.1609(4)	14.909(3)	9.6518(2)	21.5020(7)	13.7697(4)
α [°]	90.00	87.181(2)	90.0	90.0	90.0	90.00	90.0	90.0	66.109(3)
β[°]	99.94(3)	80.232(1)	97.869(2)	122.985(8)	100.643(2)	110.24(3)	101.108(2)	117.043	84.184(2)
γ[°]	90.00	88.150(1)	90.0	90.0	90.0	90.00	90.0	90.0	77.048(2)
V [Å ³]	1246.3(4)	1326.39(4)	1642.47(6)	3332.8(3)	1586.55(6)	1697.2(6)	1679.62(6)	3412.15(16)	1566.40(8)
Ζ	4	4	4	8	4	4	4	8	4
$\rho_{\text{calcd}} [\text{gcm}^{-3}]$	1.430	1.414	1.393	1.373	1.459	1.469	1.425	1.777	1.401
$\mu [{ m mm}^{-1}]$	0.4181	0.397 ²	0.335 ²	0.330 ²	0.356 ²	0.341 ²	0.335 ²	2.131 ²	0.348 ²
F_{000}	560	592	720	1440	720	776	752	1792	688
Crystal dimensions [mm]	0.37×0.36×0.20	0.36×0.27×0.26	0.33×0.25×0.09	0.70×0.45×0.15	0.32×0.20×0.12	0.34×0.14×0.08	0.37×0.15×0.13	0.40 ×0.25×0.14	0.41 ×0.27×0.06
θ range [°]	2.93-52.16	3.69-45.74	3.52-41.03	3.97-45.75	3.01-45.47	2.80-49.12	3.66-37.69	2.95-45.49	3.69-37.57
No. of collected reflections	407484	92204	77724	54264	56056	129050	33211	107459	54281
No. of independent reflections	14320	22625	10664	14231	13383	17254	8593	14388	15966
No. of reflections $> 2\sigma(I)$	12324	15703	5529	7127	8950	14297	5829	11464	9736
R _{int}	0.0498	0.0318	0.0707	0.0646	0.0576	0.0357	0.0381	0.0423	0.0567
completeness of data [%]	100.0	99.5	98.3	99.5	99.8	97.7	95.9	99.7	96.4
Number of parameters	262	437	272	272	260	283	281	260	509
Goodness-of-fit	1.707	1.004	1.004	1.005	1.001	1.127	1.006	1.001	1.007
Final R indices $(I \ge 2\sigma(I))$	R1=0.0265,	R1=0.0320,	R1=0.0394,	R1=0.0444,	R1=0.0612,	R1=0.0460,	R1=0.0342,	R1=0.0316,	R1=0.0397,
	wR2=N/A	wR2=0.0834	wR2=0.0794	wR2=0.0866	wR2=0.1231	wR2=0.1073	wR2=0.0756	wR2=0.0746	wR2=0.0798
R indices (all data)	R1=N/A,	R1=0.0510,	R1=0.0952,	R1=0.1011,	R1=0.0992,	R1=0.0572,	R1=0.0600,	R1=0.0449,	R1=0.0783,
	wR2=N/A	wR2=0.0867	wR2=0.0854	wR2=0.0915	wR2=0.1402	wR2=0.1126	wR2=0.0790	wR2=0.0815	wR2=0.0841
max./min. residual electron	0.310/-0.250	0.908/-0.606	0.442/-0.363	0.919/0.441	0.757/-0.319	1.021/-0.257	0.492/-0.377	1.550/-0.587	0.636/-0.524
density [e Å-3]									
CCDC entry	852449	897056	897057	897058	897059	897060	897061	897062	897063

¹ analytical absorption correction, ² empirical 'multi-scan' absorption correction.

D-HA	d(HA) / Å	d(DA) / Å	a(D-HA) / °	symmetry
compound 1:				
N2-H2O1	2.139(8)	2.7866(8)	122.0(6)	intramolecular
N2-H2O1	2.264(8)	3.1271(8)	146.0(7)	-x,1-y,-z
C4-H4AN1	2.541(11)	3.3478(9)	132.3(7)	-1/2-x, 1/2+y, 1/2-z
C6-H6N1	2.366(8)	3.4055(9)	161.5(6)	-1/2-x,1/2+y,1/2-z
С8-Н8О2	2.555(9)	3.2879(9)	124.9(6)	1/2+x, $1/2-y$, $1/2+z$
С9-Н9О1	2.481(8)	3.3936(9)	142.5(6)	1/2-x-1/2+v.1/2-z
C11-H11CO2	2.521(15)	3.3116(9)	133.5(8)	-x.1-vz
compound 2 :	()		(-)	,- ,, =
N2A-H2AO1A	2.308(11)	2.8768(7)	128.0(9)	intramolecular
N2A-H2AN1B	2.607(11)	3.2160(8)	133.3(9)	1-x.1-vz
N2B-H2B01B	2.151(10)	2.7621(7)	128.6(9)	intramolecular
N2B-H2B O2B	2403(10)	30777(7)	136 7(9)	2-x 1-v -z
C4B-H4BC N1A	2.596(12)	34159(10)	140.2(9)	1 - x - 1 - y - 1 - z
C7A-H7A O1B	2.330(12) 2.417(12)	3 3103(9)	147.9(10)	-1+x v z
C7B-H7B 02A	2.435(10)	3.1750(8)	132 1(8)	1-x 1-y 1-z
C8B-H8B O1A	2.452(10)	3 2872(8)	147 5(8)	1 + x - 1 + y z
C12B-H12F 02B	2.432(10) 2.572(13)	3.2072(0) 3.4299(9)	134 1(9)	2 - x - 1 - y - 7
compound 3:	2.572(15)	5.4277(7)	154.1(5)	Σ A, 1 y, Z
$\frac{\text{compound } 3}{\text{N}^2 - \text{H}^2 - \Omega^1}$	2.095(13)	2.7780(11)	137.4(10)	intramolecular
C_{A} -HAC O_{2}	2.075(13) 2.355(11)	2.7780(11) 3.3086(12)	160 3(0)	$1_{-x} - x_{-x} - 1_{-z}$
C7-H7 N1	2.555(11) 2.555(13)	3.3080(12) 3.4012(16)	150.5(9)	$1^{-x}, -y, 1^{-z}$ x $1/2, -y, -1/2 + z$
C_{1}^{-11}	2.555(15) 2.503(13)	3.4012(10) 3.3024(14)	134.0(11) 142.6(11)	x, 1/2 - y, -1/2 + z x, 1/2 - y, 1/2 + z
compound 1 :	2.303(13)	5.5024(14)	142.0(11)	x, 1/2 - y, 1/2 + Z
$\frac{\text{compound 4.}}{\text{N2 H2 O1}}$	2.254(1.4)	2 7826(12)	128 2(12)	intromologular
N2-H201	2.234(14)	2.7820(12)	120.2(13) 120.7(12)	
N2-H2UI	2.203(12)	2.8801(10) 2.5200(15)	159.7(15)	1-X,1-Y,1-Z
$C_{0} = H_{0} = M_{1}$	2.010(13) 2.525(14)	3.3299(13)	104.4(11) 140.2(11)	x, 1 + y, z
Co-110NI	2.333(14)	5.5010(15)	149.3(11)	1/2-x, 1/2+y, 1/2-z
NO LIO OL	2.152(17)	2,8007(12)	1262(16)	intromologylar
$N_2 - \Pi_2 \dots OI$	2.132(17)	2.8097(12)	150.5(10) 157.5(15)	
C4-H4AN1	2.550(17)	3.4000(19)	157.5(15)	-X,1-Y,1-Z
C4-H4CO2	2.37(2)	3.2/4/(15)	167.4(19)	1-X,1-Y,1-Z
CI3-HI3NI	2.505(19)	3.3961(15)	149.2(14)	X,-1+Y,Z
C16-H16F1	2.4/8(18)	3.2/06(16)	142.3(14)	-x,1/2+y,1/2-z
compound 6:	0.100/17	0.7501(10)	100 0(10)	• . • •
N2-H201	2.190(17)	2.7581(10)	123.0(13)	intramolecular
N2-H2NI	2.50/(15)	3.1/58(10)	134.8(13)	1-x,-y,2-z
C4-H4BO2	2.440(14)	3.1678(11)	134.1(12)	1-x,-y,2-z
C8-H8O2	2.580(15)	3.4272(12)	146.8(12)	3/2-x,1/2+y,3/2-z
С7-Н7О3	2.546(15)	3.4308(12)	151.4(12)	-x,-y,1-z
С6-Н6О4	2.565(14)	3.3888(11)	143.1(11)	-x,-y,1-z
C16-H16N1	2.600(13)	3.2868(10)	128.2(10)	-1/2+x,1/2-y,-1/2+z
compound 7:				
N2-H2O1	2.108(13)	2.7957(10)	135.3(11)	intramolecular
C4-H4AN1	2.478(14)	3.4025(14)	164.4(10)	-x,1-y,2-z
C4-H4CO2	2.380(14)	3.3089(13)	156.1(10)	1-x,1-y,2-z
C9-H9N1	2.456(12)	3.4112(13)	168.8(10)	x,3/2-y,-1/2+z
compound 8:				
N2-H2O1	2.03(3)	2.7265(15)	136(2)	intramolecular
N2-H2O1	2.38(2)	3.1041(13)	141(2)	-x,y,1/2-z
C15-H15O1	2.52(3)	3.3339(16)	142(2)	-x,1+y,1/2-z

Table S2: Summary of hydrogen bonds. Hydrogen atom positions were freely refined in all the determinations.

compound 9:				
N2A-H2AO1A	2.095(16)	2.7932(14)	139.1(14)	intramolecular
N2B-H2BO1B	2.040(16)	2.7655(14)	136.8(13)	intramolecular
C4A-H4AAN1B	2.579(16)	3.4614(17)	152.8(12)	1-x,2-y,-z
C4A-H4ACO2A	2.548(15)	3.1304(15)	120.7(11)	1-x,2-y,-z
C6B-H6BO1A	2.520(14)	3.1749(14)	124.3(11)	1-x,1-y,1-z



Figure S1: Electrostatic potential (-0.05 to 0.05 a.u.) mapped onto (a) the Hirshfeld surface and (b) the 0.001 a.u. electron-density isosurface of compound **1**.



(a)







(d)



S5



Figure S2: Electrostatic potential mapped onto the Hirshfeld Surfaces of (a) 2, (b) 3, (c) 4, (d) 5, (e) 6, (f) 7, (g) 8, (h) 9, for 1 see Figures S1 and 6(d). All molecules are in the same orientation as defined in Figure 5. Colour scale -0.05 (red) to 0 (white) to 0.05 a.u. (blue). Wavefunction used for the ESP calculation is at the BLYP/6-31G(d) level calculated with the programs Gaussian 03 and CrystalExplorer, except for the iodophenyl substituted compound 8 (g), where it is BLYP/3-21G because of size restrictions. For 2 (a) and 9 (h) only independent molecule A of the asymmetric unit is shown.