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Supporting Information

Reaction of a Bulky Amine Borane with Lanthanide Trialkyls. Formation of Alkyl Lanthanide Imide Complexes

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General Information

All manipulations involving air- and moisture-sensitive compounds were carried out under an atmosphere of dry argon by using modified Schlenk line and glovebox techniques. Elemental analyses were carried out on an Elementar Vario EL analyzer. The ¹H, ¹¹B and ¹³C NMR spectroscopic data were recorded on a Bruker AV400 spectrometer. Infrared spectra were recorded on a Bio-Rad FTS 6000 spectrophotometer. The solvents (THF, toluene and *n*-hexane) were freshly distilled from sodium and degassed prior to use. Chemical shifts were referenced to residual solvent peaks (¹H and ¹³C NMR) or external BF₃ Et₂O (¹¹B NMR). YCl₃, SmCl₃ and BH₃ THF were purchased from Alfa-Aesar and J&K Scientific. Y(CH₂SiMe₃)₃(THF)₂, Sm(CH₂SiMe₃)₃(THF)₂, and ArNH₂ (Ar = 2,6-(3,5-Me₂C₆H₃)₂C₆H₃) were prepared according to the procedures described in literature.^{S1}

Synthesis of ArNH₂ BH₃ (1): To a solution of ArNH₂ (3.3 mmol, 1.0 g) in THF (20 mL) was slowly added BH₃ THF (4 mmol, 1M in THF) at 0 °C. The mixture was warmed to room temperature and stirred for 3 h, after then the solution was concentrated to 5 mL to give colorless crystals of **1** at -40 °C overnight (0.9 g, 86%). Anal. Calcd for C₂₂H₂₆BN (315.26): C, 83.82; H, 8.31; N, 4.44; Found: C, 83.71; H, 8.14; N, 4.41. ¹H NMR (400 MHz, CDCl₃, 23 °C): δ 7.30–7.09 (m, 9H, Ar-*H*), 5.48 (s, 2H, N*H*₂), 2.37 (s, 12H, Ar-*Me*₂). ¹³C NMR (101 MHz, CDCl₃, 23 °C): δ 138.9, 137.5, 135.4, 130.3, 130.2, 127.1, 126.9, 126.3 (Ar-*C*), 21.5 (Ar-*Me*₂). ¹¹B NMR (128 MHz, CDCl₃, 23 °C): δ -16.1. IR (KBr, cm⁻¹): \tilde{v} (B-H) 2404.5, 2327.7, 2270.9.

Synthesis of $[Y(CH_2SiMe_3)(\mu_2-NArBH_3)]_2$ (2): A solution of 1 (0.50 g, 1.6 mmol) in toluene (40 mL) was slowly added to a stirred solution of $Y(CH_2SiMe_3)_3(THF)_2$ (0.79 g, 1.6 mmol) in toluene (20 mL) at -78 °C. The mixture was stirred overnight at -60 °C and then warmed to room temperature. The color of the mixture changed to pale yellow gradually. Then the solution was concentrated to 5 mL to give pale

yellow crystals of **2** at -40 °C for overnight (0.47 g, 60%). M.p.: 130 °C (dec.). Anal. Calcd for $C_{52}H_{70}B_2N_2Si_2Y_2$ (978.73): C, 63.81; H, 7.21; N, 2.86; Found: C, 63.56; H, 6.98; N, 2.47. ¹H NMR (400 MHz, C₆D₆, 23 °C): δ 7.54 (s, 2H, Ar-*H*), 7.46 (s, 2H, Ar-*H*), 7.42 (s, 2H, Ar-*H*), 6.96 (s, 2H, Ar-*H*), 6.83 (s, 2H, Ar-*H*), 6.77 (s, 4H, Ar-*H*), 6.56 (s, 2H, Ar-*H*), 6.41 (s, 2H, Ar-*H*), 2.32 (s, 6H, Ar-*Me*₂), 2.16 (s, 6H, Ar-*Me*₂), 2.01 (s, 6H, Ar-*Me*₂), 1.87 (s, 6H, Ar-*Me*₂), 0.29 (s, 18H, Si*Me*₃), -0.44 (d, 2H, ²*J*_{HH} = 10.5 Hz, ²*J*_{YH} = 4.0 Hz, Y-C*H*₂), -1.25 (d, 2H, ²*J*_{HH} = 10.5 Hz, ²*J*_{YH} = 3.2 Hz, Y-C*H*₂). ¹³C NMR (101 MHz, C₆D₆, 23 °C): δ 156.2, 147.3, 145.8, 143.7, 143.6, 138.6, 138.4, 138.3, 136.8, 136.0, 132.8, 132.3, 131.3, 130.5, 129.9, 127.0, 123.9, 119.2 (Ar-*C*), 39.2 (d, ¹*J*_{YC} = 40.4 Hz, Y-C*H*₂), 22.3, 22.2, 21.9, 21.6 (Ar-*Me*₂), 4.7 (Si*Me*₃). ¹¹B NMR (128 MHz, C₆D₆, 23 °C): δ -17.3. IR (KBr, cm⁻¹): \tilde{v} (B-H) 2364.8, 2270.3, 2161.6.

Synthesis of [Sm(CH₂SiMe₃)(\mu_2-NArBH₃)]₂ (3): A solution of 1 (0.50 g, 1.6 mmol) in toluene (50 mL) was slowly added to a stirred solution of Sm(CH₂SiMe₃)₃(THF)₂ prepared in situ (0.89 g, 1.6 mmol) in hexane (20 mL) at -78 °C. The mixture was stirred overnight at -60 °C and then warmed to room temperature. Then the solution was concentrated to 5 mL to give dark brown crystals of 3 at -40 °C for overnight (0.57 g, 65%). M.p.: 150 °C (dec.). Anal. Calcd for C₅₂H₇₀B₂N₂Si₂Sm₂ (1101.64): C, 56.69; H, 6.40; N, 2.54; Found: C, 56.33; H, 6.23; N, 2.38. ¹H NMR (400 MHz, C₆D₆, 23 °C): δ 8.80 (s, 2H, Ar-*H*), 6.02 (s, 2H, Ar-*H*), 5.94 (s, 2H, Ar-*H*), 5.88 (s, 2H, Ar-*H*), 5.45 (s, 2H, Ar-*H*), 5.28 (s, 2H, Ar-*H*), 4.87 (s, 2H, Ar-*H*), 4.57 (s, 2H, Ar-*H*), 4.05 (s, 6H, Ar-*Me*₂), 3.98 (s, 2H, Ar-*H*), 3.91 (s, 6H, Ar-*Me*₂), 1.53 (s, 9H, SiMe₃), 1.36 (s, 4H, Sm-CH₂), 1.25 (s, 6H, Ar-*Me*₂), 0.01 (s, 9H, SiMe₃), -0.63 (s, 6H, Ar-*Me*₂). ¹³C NMR (101 MHz, C₆D₆, 23 °C): δ 170.3, 142.4, 140.2, 137.8, 136.9, 136.3, 136.1, 130.8, 129.6, 129.3, 125.9, 125.7, 125.3, 120.4, 119.0, 118.0, 116.7, 114.7 (Ar-C), 30.2 (Sm-CH₂), 23.9, 23.7, 20.9, 19.3 (Ar-*Me*₂), 2.3, 0.0 (Si*Me*₃). ¹¹B NMR (128 MHz, C₆D₆, 23 °C): δ -30.7. IR (KBr, cm⁻¹): $\bar{\nu}$ (B–H) 2390.5, 2268.9, 2159.5.

X-ray Structural Determinations. Crystals for X-ray analysis of were **2** and **3** obtained from toluene at -40 °C. All intensity data of **2** and **3** were collected with a Rigaku Saturn 724 CCD diffractometer using graphite-monochromated Mo K α radiation ($\lambda = 0.71073$ Å) at 113(2) K. The structures were resolved by direct methods and refined by full-matrix least-squares on $F^{2,S2}$ Hydrogen atoms were considered in calculated positions. All non-hydrogen atoms were refined anisotropically. The ORTEP-3 program was utilized to draw the molecules.^{S3}

References

- [S1] Y. Chen, H. Song and C. Cui, Angew. Chem., Int. Ed. 2010, 49, 8958.
- [S2] G. M. Sheldrick, Acta Crystallogr. 2008, A64, 112.
- [S3] L. J. Farrugia, J. Appl. Crystallogr. 1997, 30, 837.