

Supplementary Material

A Green Approach for the Synthesis of Bis (Substituted Sulfabenzamide) *para*-Benzoquinone Based on the Reaction of Sulfabenzamide with Electrochemically Generated *para*- Benzoquinone

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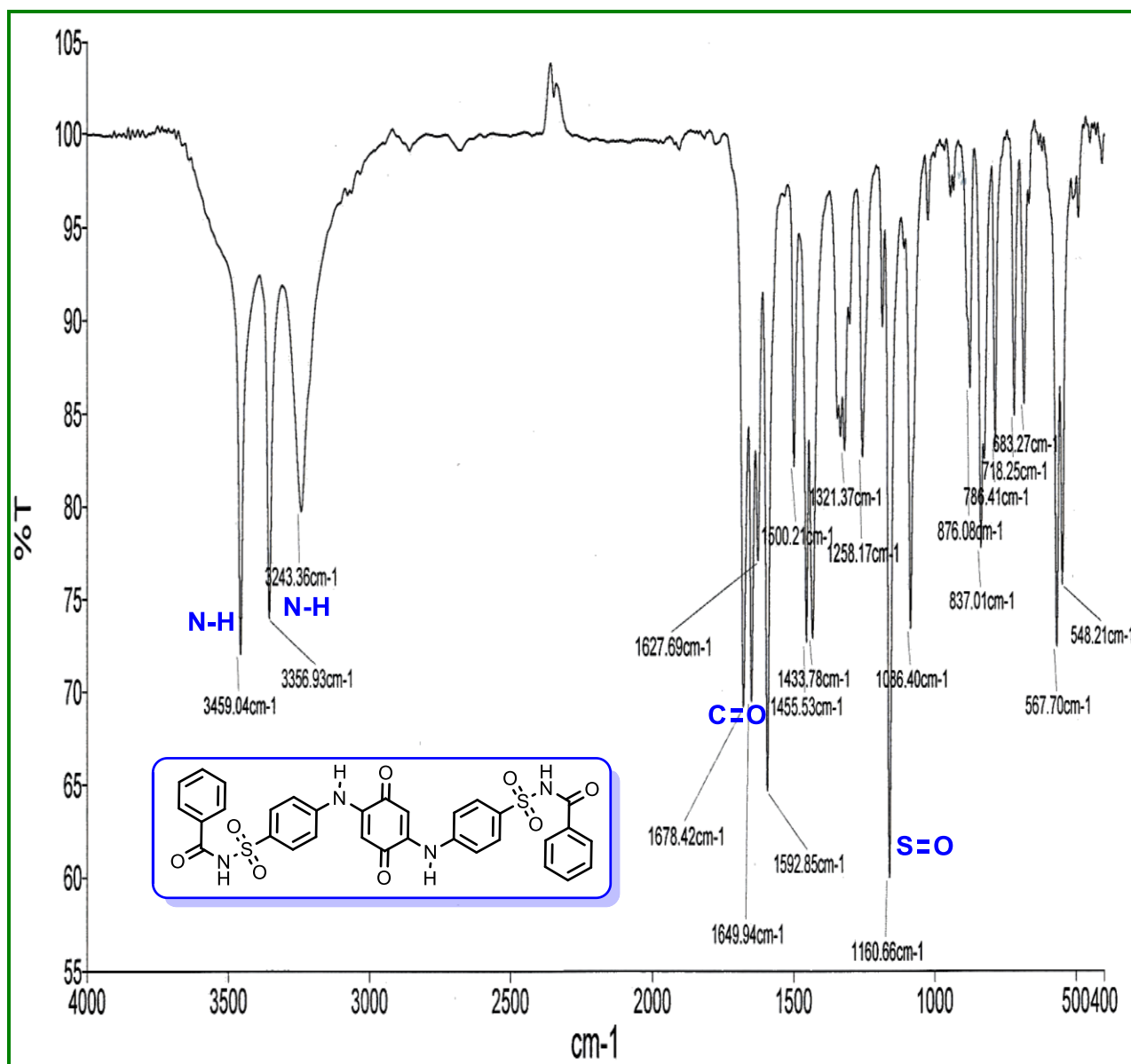
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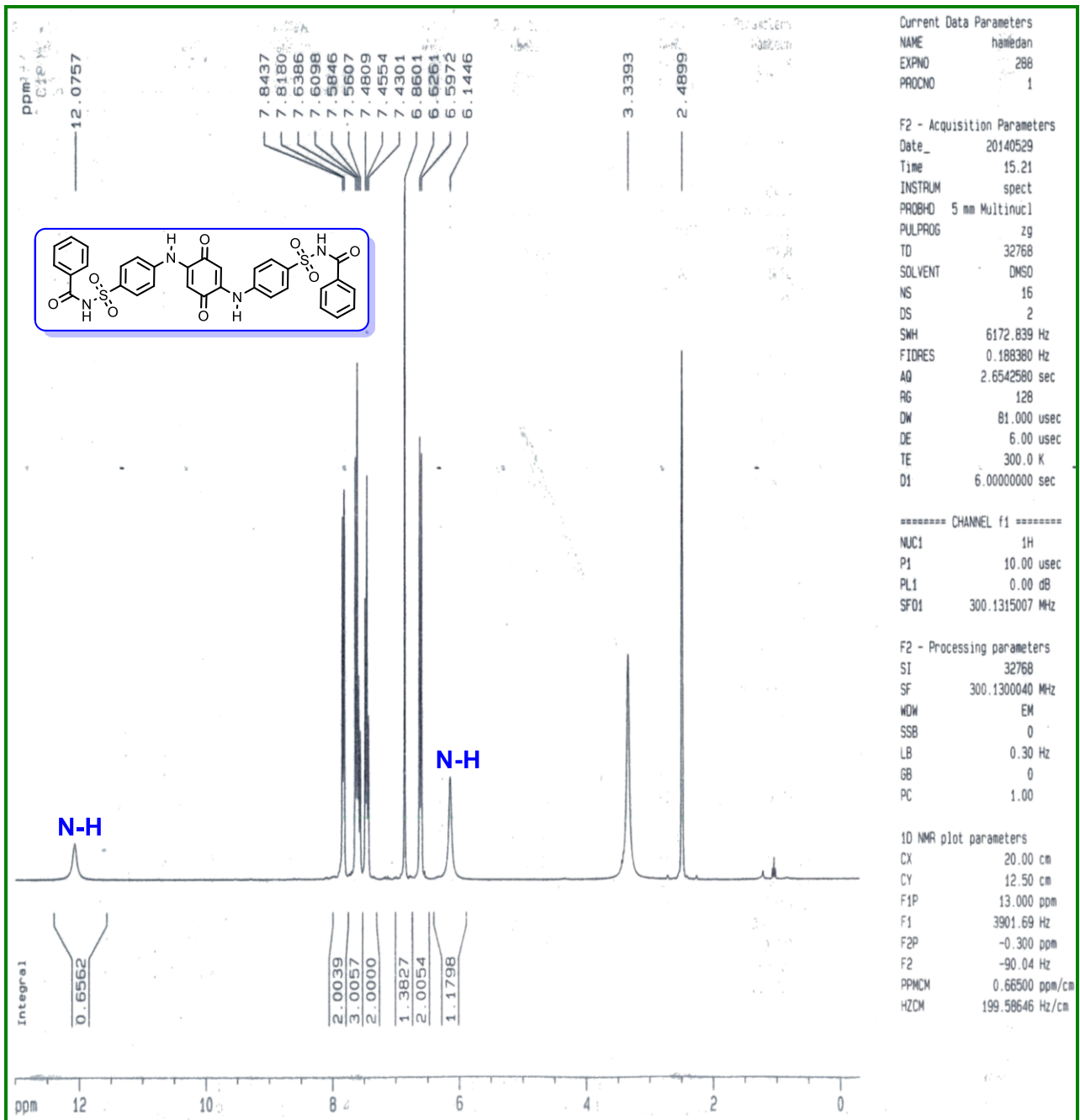
Electroorganic synthesis of *N,N'*-(4,4'-((3,6-dioxocyclohexa-1,4-diene-1,4-diyl)bis(azanediyl))bis(4,1-phenylenesulfonyl))dibenzamide (5) (C₃₂H₂₄N₄O₈S₂).

A mixture of phosphate buffer solution ($c = 0.2$ M, $\text{pH} = 8.0$)/ethanol mixture (50/50, v/v) containing hydroquinone (0.25 mmol) and sulfabenzamide (0.5 mmol) was subjected to electrolysis in a divided cell at 0.20 V versus Ag/AgCl. The electrolysis was terminated when the current decayed to 5% of its original value. At the end of electrolysis the precipitated solid was collected by filtration and was washed with *n*-hexane. The precipitated solid was recrystallized from methanol to give the compound **5** (isolated yield, 74%) as an orange solid, m.p: 184-185 °C. ¹H NMR (300 MHz, DMSO-*d*₆) δ /ppm: 6.14 (br, 2H, NH, D₂O exchangeable), 6.61 (d, $J = 8.7$ Hz, 4H, aromatic), 6.86 (s, 2H, quinone), 7.46 (t, $J = 7.6$ Hz, 4H, aromatic), 7.60 (m, 6H, aromatic), 7.83 (d, $J = 7.7$ Hz, 4H, aromatic), 12.0 (br, ~2H, NH, D₂O exchangeable). ¹³C NMR (75 MHz, DMSO-*d*₆) δ /ppm: 112.2, 123.7, 128.2, 128.5, 130.1, 131.9, 132.9, 136.6, 153.7, 165, 187.7 (C=O). IR (KBr) ν /cm⁻¹: 3459 (NH), 3357 (NH), 3243 (NH), 1678 (C=O), 1628, 1650, 1592, 1500, 1455, 1434, 1321, 1258, 1161, 1086, 876, 837, 786, 718, 683, 568, 548. MS (EI, 70 eV); m/z (relative intensity): 658 (M + 2H) (3), 505 (22), 503 (23), 428 (100), 426 (100), 347 (14), 276 (94).

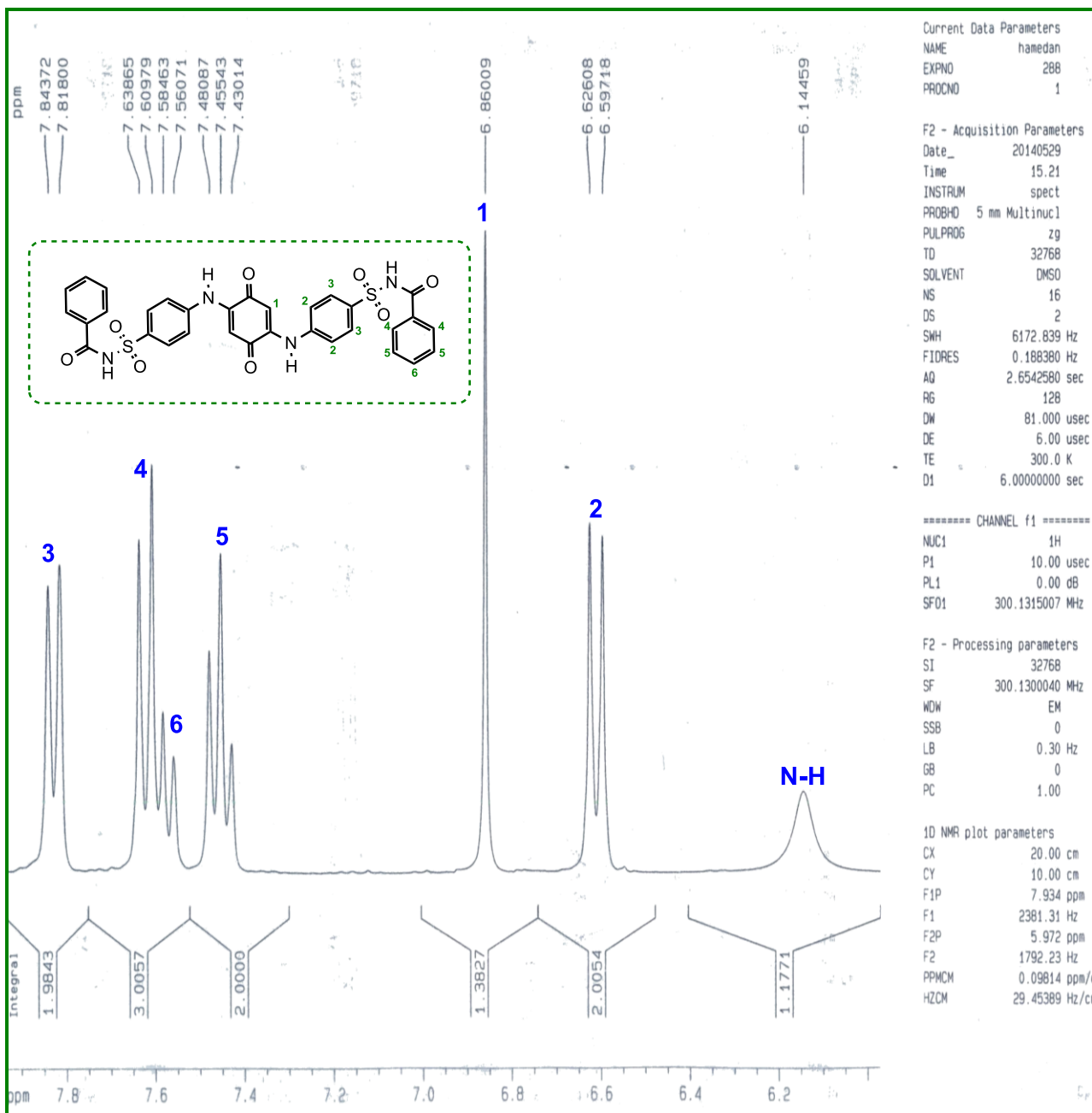
IR spectrum of 5



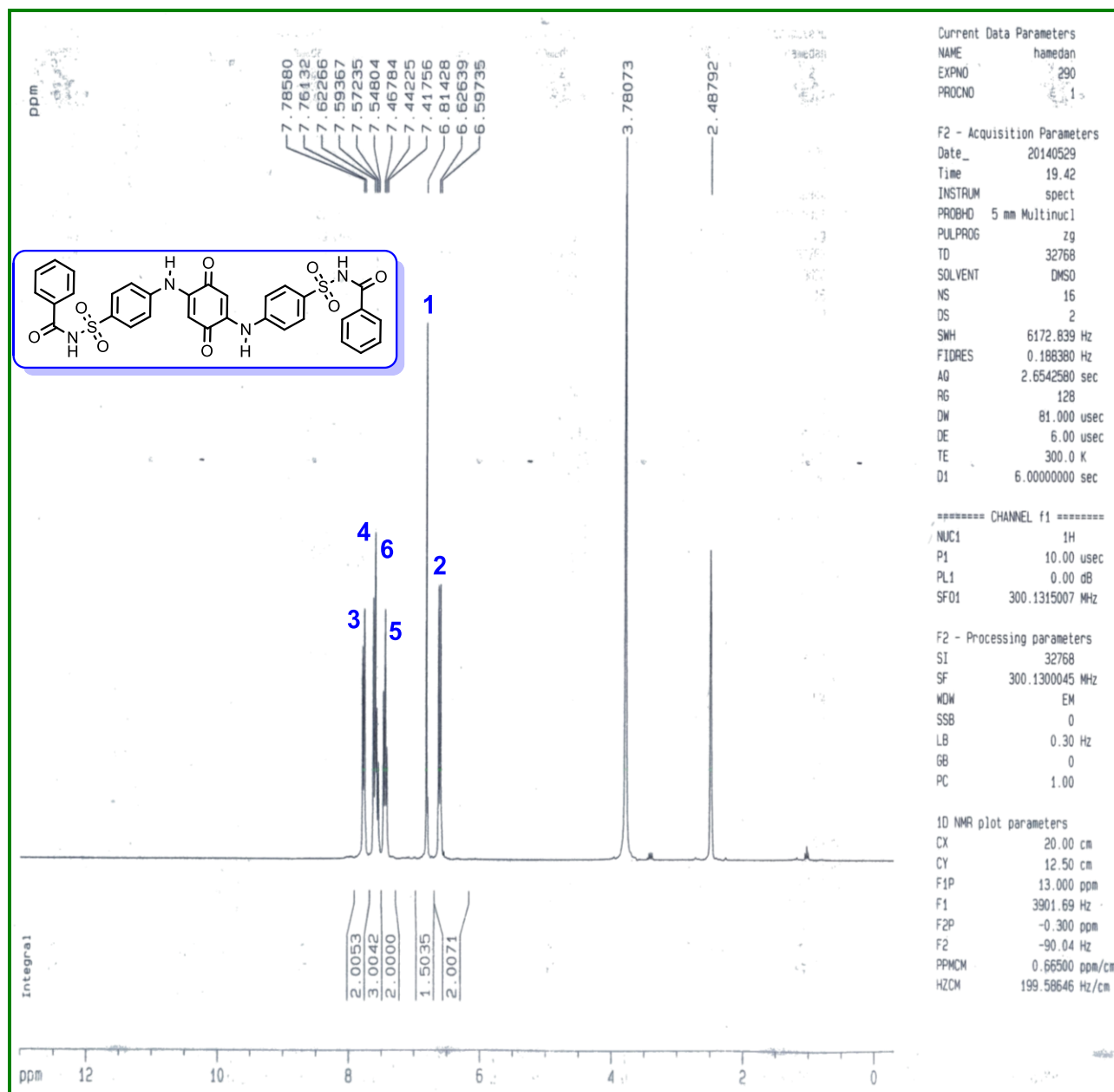
¹H NMR spectrum of 5



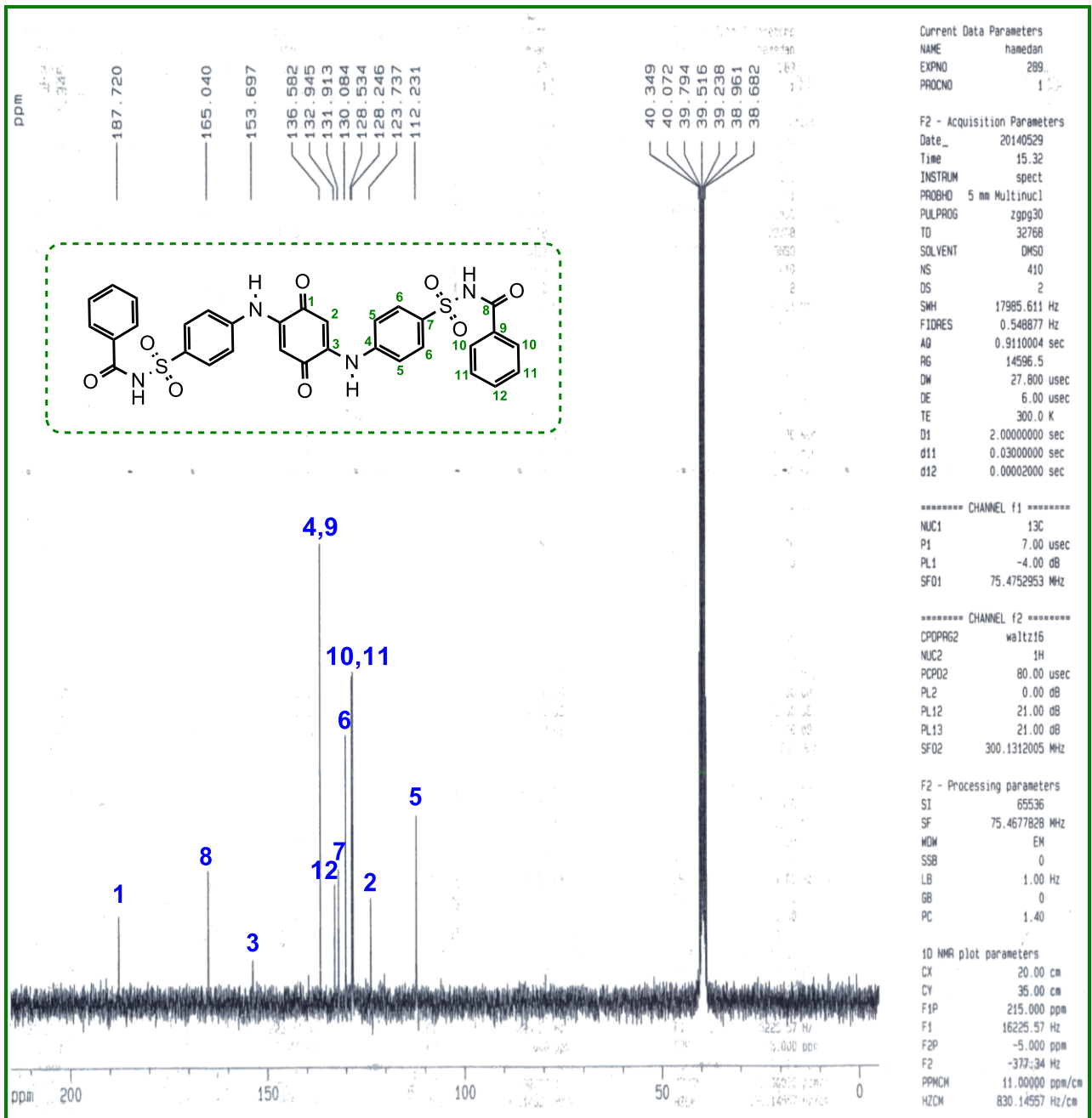
Expanded ¹H-NMR spectrum of 5



¹H NMR spectrum of 5 with D₂O



¹³CNMR spectrum of 5



Mass spectrum of 5

