

Supporting Information

Self-assembly synthesis of Co_3O_4 / multi walled carbon nanotubes

composites: An efficient enzyme-free glucose sensor

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Fig. S1

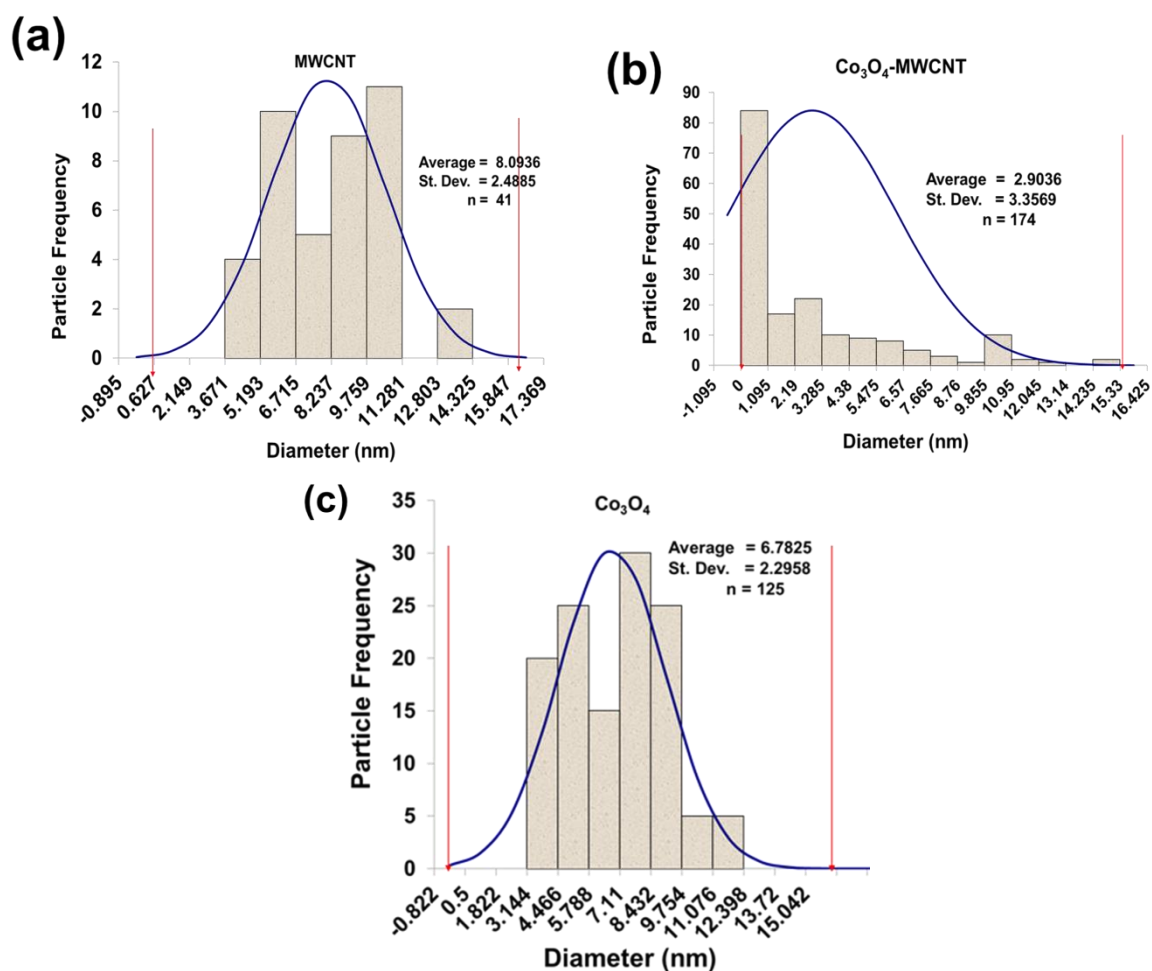


Fig. S1 The size distribution histograms and corresponding Gaussian fits. (a) MWCNT (b) Co_3O_4 -MWCNT composite and (c) Co_3O_4 .

Fig. S2

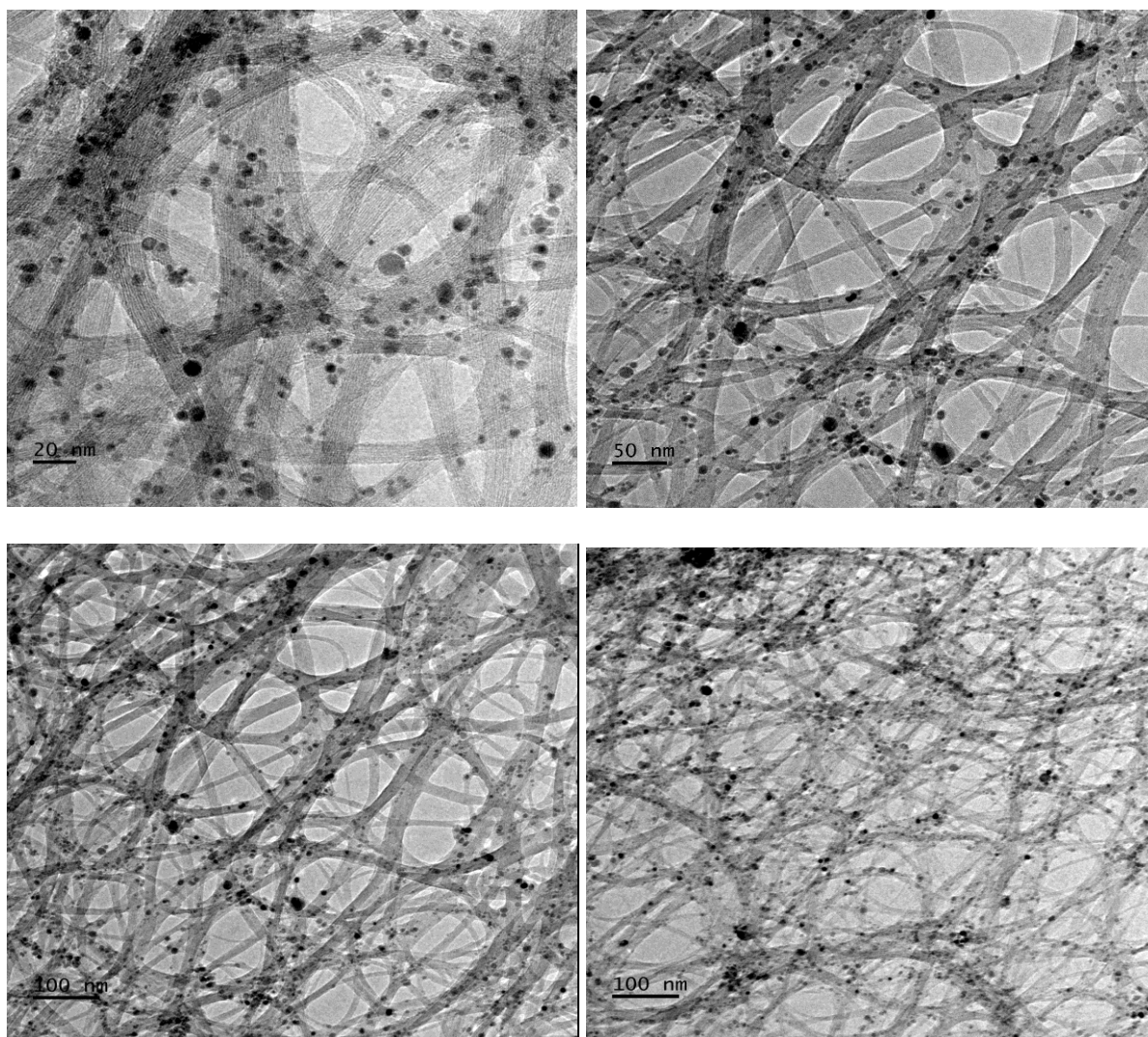


Fig. S2 The TEM images of Co_3O_4 -MWCNT composite at different magnification (20 nm, 50 nm and 100 nm).

Fig. S3

Glucose concentration studies

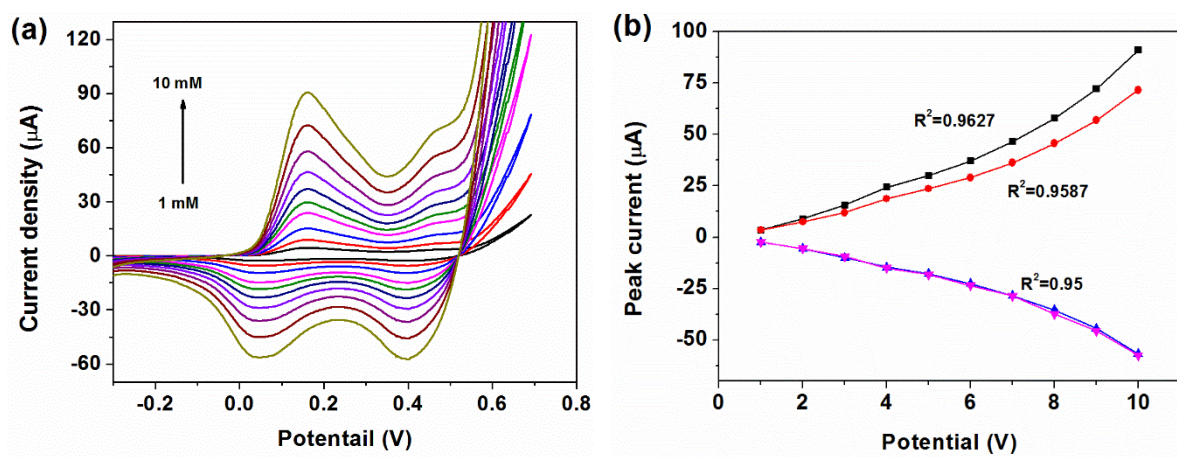


Fig. S3 (a) CV spectra of Co₃O₄-MWCNT/GCE for different glucose concentration (1 mM to 10 mM) in 0.2 M NaOH solution at 50 mV s⁻¹ and (b) plot of peak current vs. potential with the linear regression values calculated.

Fig. S3 shows the CV spectra of Co₃O₄-MWCNT/GCE for various glucose concentrations. It is observed that the magnitude of the electrochemical response current of the Co₃O₄-MWCNT/GCE increases with increasing concentration of glucose (Fig. S3 (a)). It can be attributed to the catalytic action of Co₃O₄ bound on the MWCNT surface that acts as a high surface area matrix and facilitates higher electron transfer rate. From the linear relationship measurements, the regression coefficient values of anodic and cathodic peaks are found to be $R^2 = 0.96, 0.9587, 0.95$ and 0.95 respectively (Fig. S3 (b)). These results suggest that the synergistic effect of Co₃O₄ and MWCNT will effectively enhance the direct electrocatalytic oxidation of glucose.