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Supporting Information

A Novel 3-fold Interpenetration Metal–Organic Framework with

Tunable Luminescence and Selective Adsorption of CO₂ Properties

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Figure S1. Along *c* axis direction, the two-dimensional layers in polymer 1.



Figure S2. Schematic description of the 6-connected α -Po nets.



Figure S3. The CIE chromaticity coordinates of the emissions spectrum excited at 285–380 nm.



Figure S4. Adsorption isotherms of N_2 at 77 K of 1.



Figure S5. Comparison of vapor adsorption isotherms of 1 for water, ethanol and methanol vapour at 293 K.



Figure S6. TG curve of as-synthesized 1.



Figure S7. Powder XRD patterns for 1: (red: simulated one; black: the as-synthesized sample).

	1
Chemical formula	$C_{52}H_{48}N_6O_{16}Zn_2$
$M_{ m r}$	1143.70
Crystal system, space group	Monoclinic, $P2_1/c$
Temperature (K)	296
<i>a</i> , <i>b</i> , <i>c</i> (Å)	14.5740 (7), 23.5585 (11), 15.4557 (7)
β (°)	110.801 (3)
$V(\text{\AA}^3)$	4960.7 (4)
Ζ	4
$\mu (mm^{-1})$	1.531
Crystal size (mm)	$0.30 \times 0.25 \times 0.20$
T_{\min}, T_{\max}	0.7441, 0.8179
R _{int}	0.0465
$(\sin \theta / \lambda)_{\text{max}} (\text{\AA}^{-1})$	0.594
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.0474, 0.1262, 1.051
No. of reflections	8734
No. of parameters	623
No. of restraints	20
$\Delta \rho_{max}, \Delta \rho_{min} (e \text{ Å}^{-3})$	0.498, -0.391

 Table S1. Crystallographic data for 1.

Table S2 Selected geometric parameters (Å, °)

N1—Zn2	2.026 (3)	O9—Zn2 ⁱⁱⁱ	2.475 (2)
N4—Zn1 ⁱ	2.029 (3)	O10—Zn2 ⁱⁱⁱ	1.970 (2)
O1—Zn1	1.991 (3)	Zn1—O4 ^{iv}	1.931 (2)
O2—Zn2	1.954 (2)	Zn1—N4 ^v	2.029 (3)
O4—Zn1 ⁱⁱ	1.931 (2)	Zn2—O10 ^{vi}	1.970 (2)
O6—Zn1	1.927 (3)	Zn2—O9 ^{vi}	2.475 (2)
O7—Zn2	1.964 (2)		
O(7n1) O(4iv)	122(0(12))	$O7$ $7m^2$ $O10vi$	104.17(10)
$00-211-04^{11}$	155.00 (12)	$0/-2112-010^{11}$	104.17 (10)
$06-2n1-04^{19}$ 06-2n1-01	133.60 (12) 103.89 (11)	O2—Zn2—N1	100.25 (11)
06—Zn1—04 ^{iv} 06—Zn1—01 04 ^{iv} —Zn1—01	103.89 (11) 103.50 (11)	O7—Zn2—N1 O7—Zn2—N1	104.17 (10) 100.25 (11) 107.27 (11)
06—Zn1—04 ^{Iv} 06—Zn1—01 04 ^{iv} —Zn1—01 06—Zn1—N4 ^v	103.89 (11) 103.50 (11) 105.99 (13)	O7—Zn2—N1 O7—Zn2—N1 O10 ^{vi} —Zn2—N1	104.17 (10) 100.25 (11) 107.27 (11) 102.54 (11)
06—Zn1—04 ^{IV} 06—Zn1—01 04 ^{iv} —Zn1—01 06—Zn1—N4 ^v 04 ^{iv} —Zn1—N4 ^v	103.89 (11) 103.50 (11) 105.99 (13) 108.43 (12)	O7—Zn2—O10 ¹¹ O2—Zn2—N1 O7—Zn2—N1 O10 ^{vi} —Zn2—N1 O2—Zn2—O9 ^{vi}	104.17 (10) 100.25 (11) 107.27 (11) 102.54 (11) 84.89 (10)
O6—Zn1—O4 ^{IV} O6—Zn1—O1 O4 ^{iv} —Zn1—O1 O6—Zn1—N4 ^v O4 ^{iv} —Zn1—N4 ^v O1—Zn1—N4 ^v	103.89 (11) 103.50 (11) 105.99 (13) 108.43 (12) 94.38 (12)	O7—Zn2—O10 ¹¹ O2—Zn2—N1 O7—Zn2—N1 O10 ^{vi} —Zn2—N1 O2—Zn2—O9 ^{vi} O7—Zn2—O9 ^{vi}	104.17 (10) 100.25 (11) 107.27 (11) 102.54 (11) 84.89 (10) 156.25 (11)
$O6-Zn1-O4^{iv}$ O6-Zn1-O1 $O4^{iv}-Zn1-O1$ $O6-Zn1-N4^{v}$ $O4^{iv}-Zn1-N4^{v}$ $O1-Zn1-N4^{v}$ O2-Zn2-O7	103.89 (11) 103.50 (11) 105.99 (13) 108.43 (12) 94.38 (12) 103.74 (11)	$O7 = Zn2 = O10^{11}$ O2 = Zn2 = N1 O7 = Zn2 = N1 $O10^{vi} = Zn2 = O9^{vi}$ $O7 = Zn2 = O9^{vi}$ $O10^{vi} = Zn2 = O9^{vi}$	104.17 (10) 100.25 (11) 107.27 (11) 102.54 (11) 84.89 (10) 156.25 (11) 57.61 (9)

Symmetry codes: (i) *x*-1, -*y*+3/2, *z*-3/2; (ii) -*x*, *y*+1/2, -*z*+5/2; (iii) *x*+1, *y*, *z*; (iv) -*x*, *y*-1/2, -*z*+5/2; (v) *x*+1, -*y*+3/2, *z*+3/2; (vi)

x−1, *y*, *z*.