

Supporting Information
**Development of a Cr(III) ion selective fluorescence probe using
organic nanoparticles and its real time applicability**

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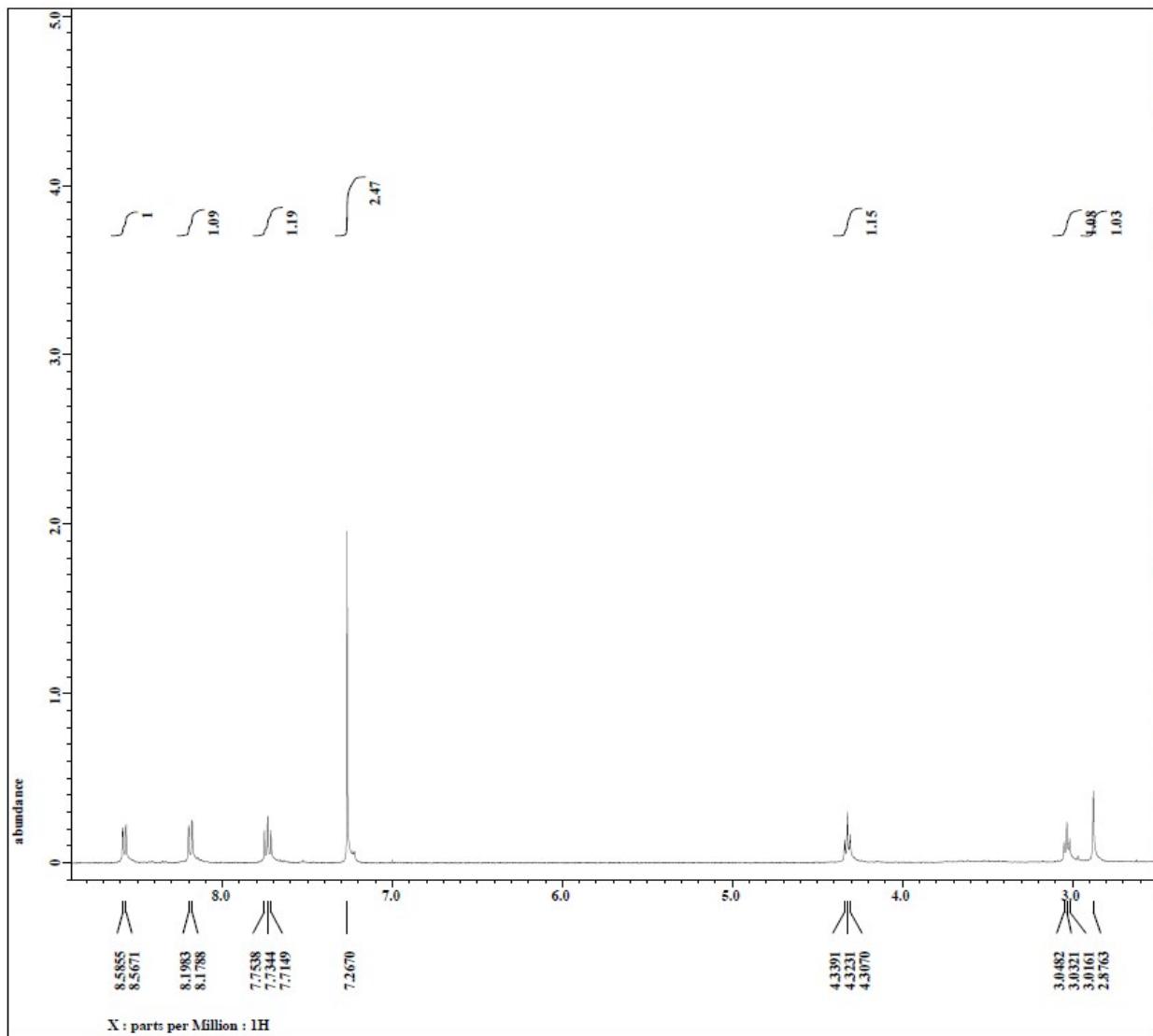


Figure S1. ^1H NMR spectrum of Ligand 1

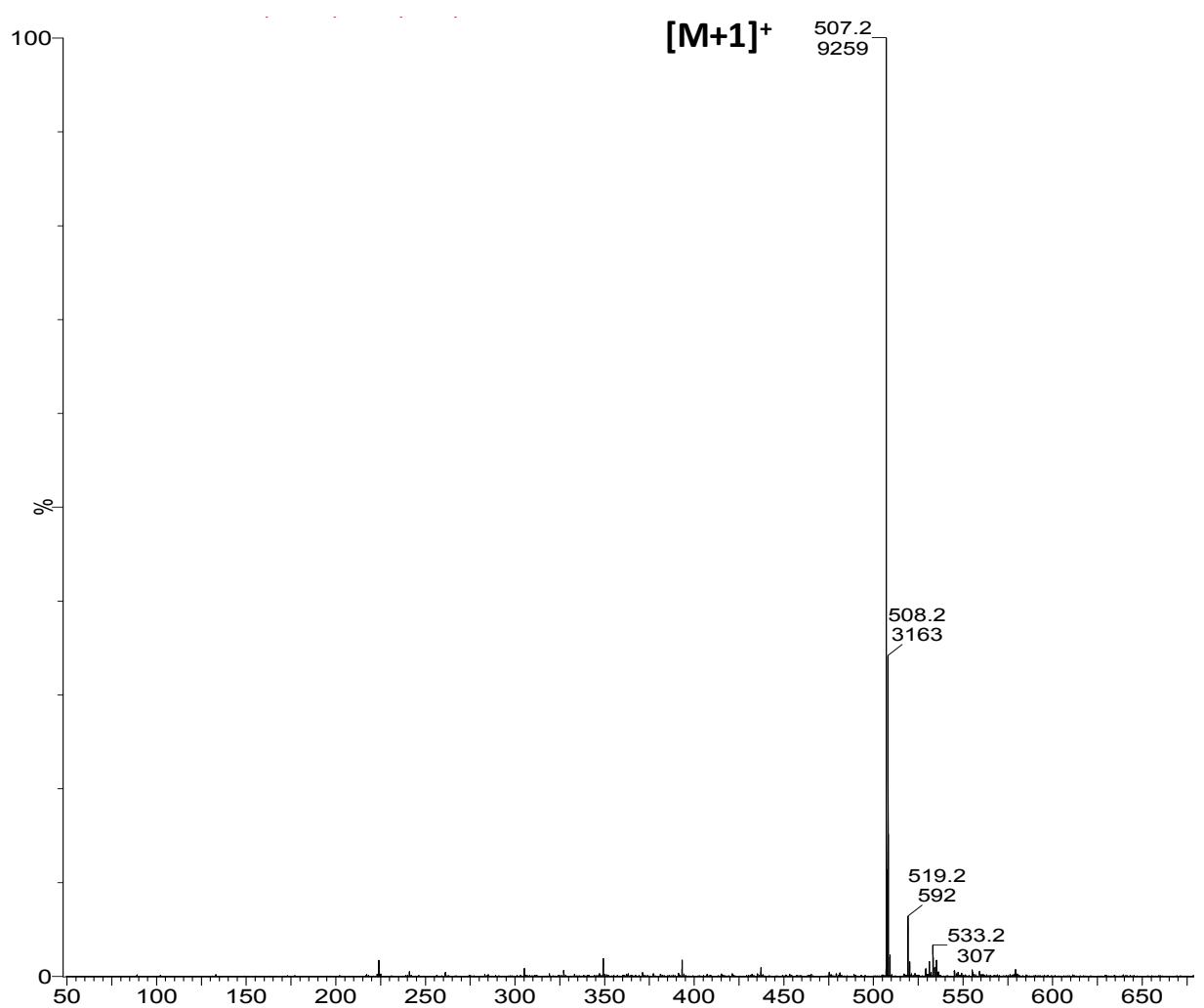


Figure S2. Mass spectrum of Ligand 1

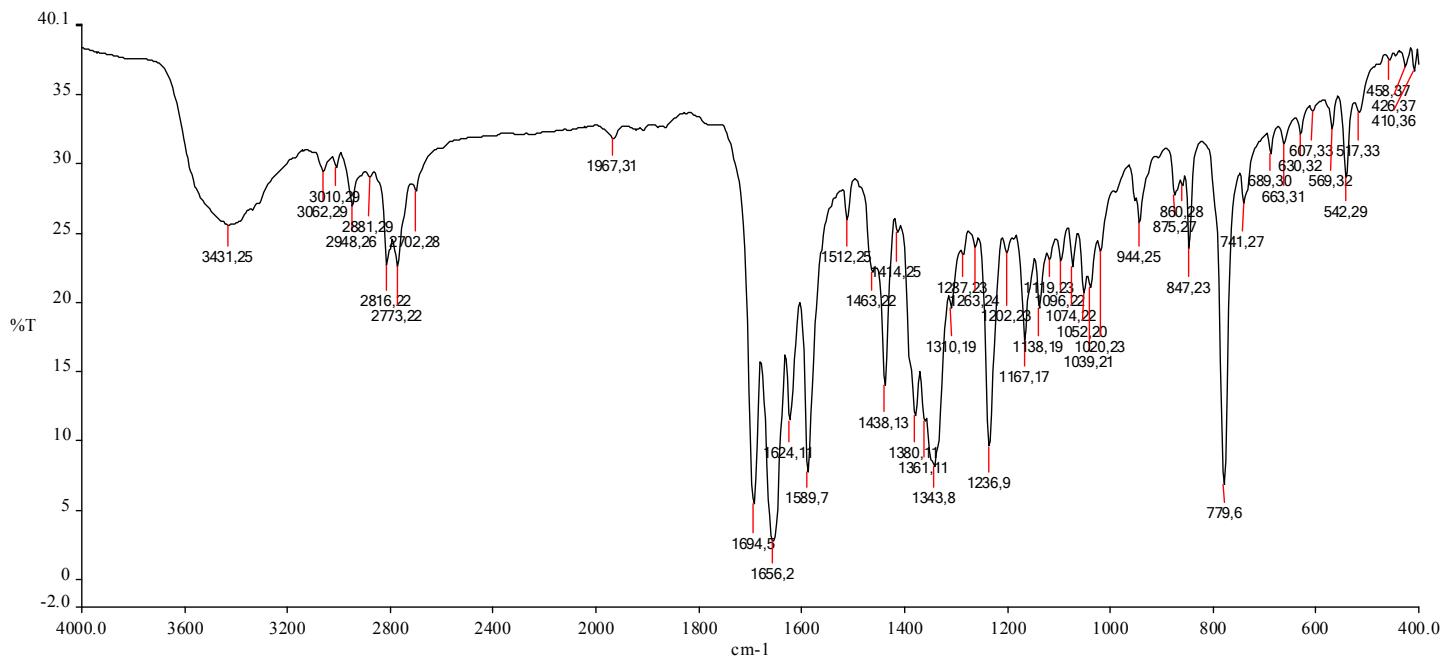


Figure S3. IR spectrum of Ligand 1

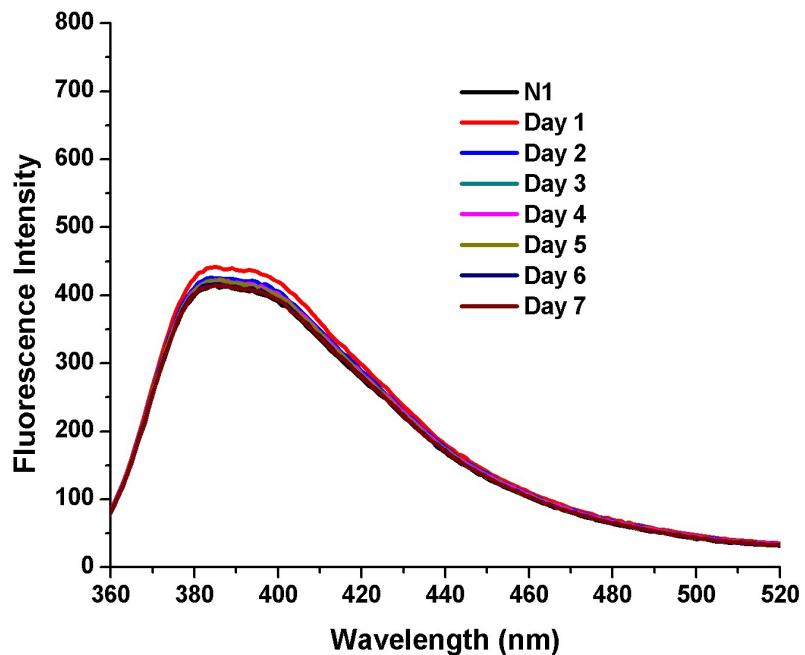


Figure S4. Fluorescence emission profile showing the stability of organic nanoparticles N1.

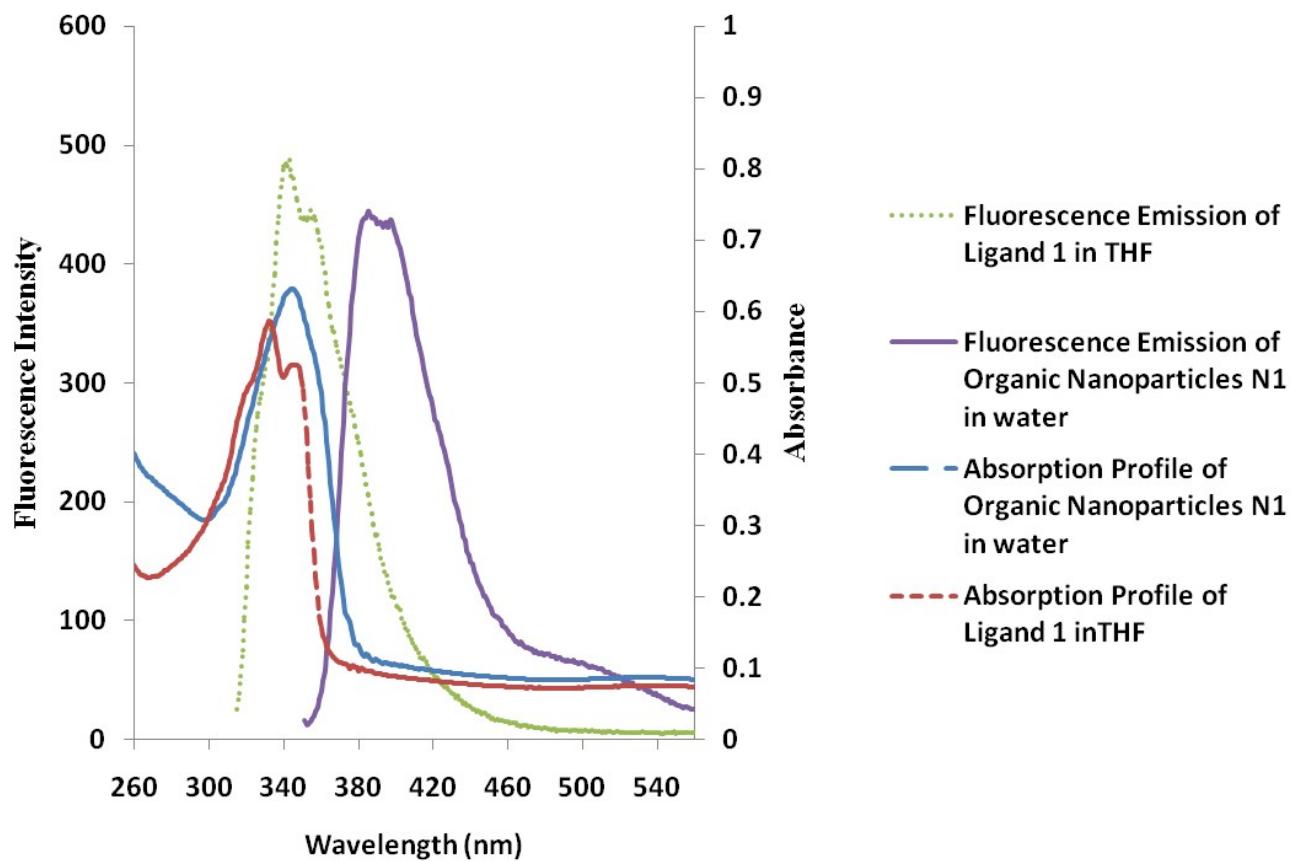


Figure S5. Fluorescence emission and absorption spectra of ligand 1 (0.492 μ M) in THF and organic nanoparticles of ligand 1 (N1) in aqueous medium.

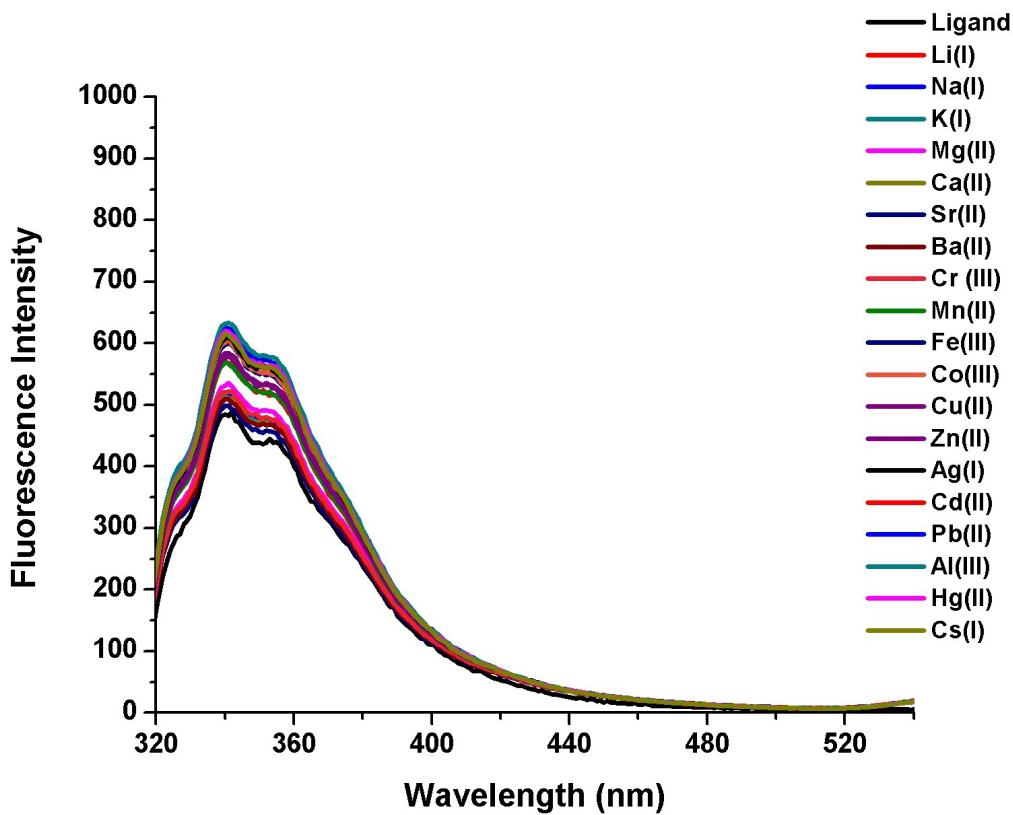


Figure S6. Changes in emission profile of ligand 1 (0.492 μM) upon addition of 20 μM of a particular metal ion.

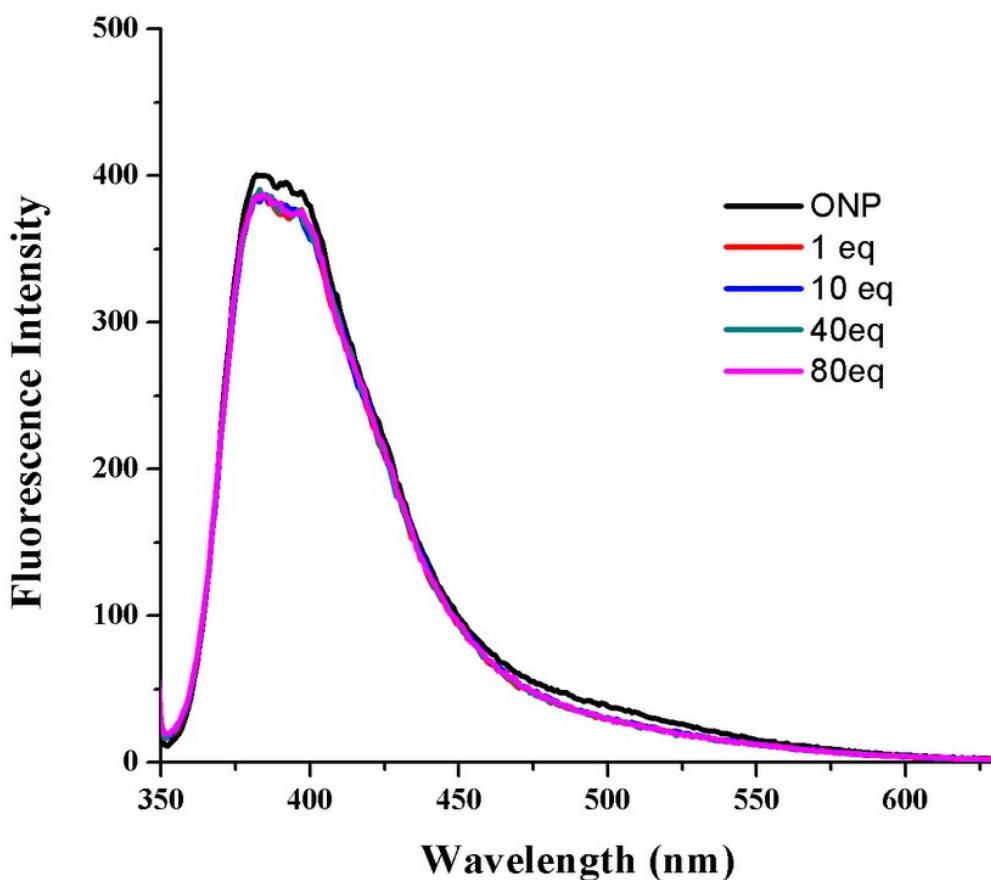


Figure S7. Effect of ionic strength on ONP N1 upon addition of 0-80 equiv. of TBA salt of perchlorate.

Determination of the detection limit.

The limit of detection of organic nanoparticles **N1** of **1** for Cr^{3+} ion was determined from the following equation:

$$\text{Limit of Detection} = (3.3 * \sigma) / M$$

σ : Standard Deviation = 0.438

M: slope of the calibration curve = 22.45

Table S1. A comparison of present work with reported sensors of Cr³⁺ ion in literature

S. No .	Name of Journal	Medium of studies	Detection Limit	Detection Range	Respon se time	Mode of Detection	Real Sample Analysis
1.	Turn-on fluorogenic and chromogenic detection of Fe ³⁺ and Cr ³⁺ in a completely water medium with polyacrylamide covalently bonding to rhodamine B using diethylenetriamine as a linker, RSC Advances, 2014, 4 , 46332- 46339.	Aqueous Medium	9.9×10^{-6} mol L ⁻¹	10×10^{-5} - 45×10^{-5} mol L ⁻¹	----	Fluorogenic and Chromogenic Detection	-----
2.	A single molecular probe for multi-analyte (Cr ³⁺ , Al ³⁺ and Fe ³⁺) detection in aqueous medium and its biological application, ChemComm, 2014, 50 , 12258-12261.	Water-Ethanol (8:2)	0.2 μM	-----	-----	Fluorescence Determination	Cell imaging in living cells
3.	A new highly sensitive and selective fluorescence chemosensor for Cr ³⁺ based on rhodamine B and a 4,13-diaza-18-crown 6-ether conjugate, RSC Advances, 2014, 4 , 2563-2567.	Methanol-Water (3:2)	0.144 μM	-----	---	Fluorescence Determination	<i>In vitro</i> cell studies
4.	A cell permeable Cr ³⁺ selective chemosensor and its application in living cell imaging, RSC Advances, 2013, 3 , 19978-19984.	Ethanol-Water (1:5) HEPES Buffer (0.1 M, pH 7.4)	3.6×10^{-7} mol L ⁻¹	3.6×10^{-7} – 4.5×10^{-5} mol L ⁻¹	15-20 sec	Fluorescence Determination	Sensor applicability in biological system

5.	Selective and cyclic detection of Cr ³⁺ using poly (methylacrylic acid) monolayer protected gold nanoparticles, New Journal of Chemistry, 2014, 38 , 717-722.	Aqueous Medium	40 µM	----	---	Colorimetric Determination	-----
6.	Visual and near IR (NIR) fluorescence detection of Cr ³⁺ in aqueous media via spirobenzopyran ring opening with application in logic gate and bio-imaging, Dalton Transactions, 2014, 43 , 231–239	CH ₃ CN/ HEPES buffer solution (7/3,v/v)	6.09 µM with UV titration and 4.6 µM with fluorescence titration	----	-----	Visual and NIR fluorescence detection	Applicability of sensor to cell biology
7.	A selective carbazole-based fluorescent probe for chromium (III), Analytical Methods, 2013, 5 , 5549-5554.	Ethanol–water (9:1,v/v)	0.10 µmol L ⁻¹	1.0- 20 µmol L ⁻¹	----	Fluorescence Determination	-----
8.	A highly selective and dual responsive test paper sensor of Hg ^{2+/Cr³⁺ for naked eye detection in neutral water, RSC Advances, 2012, 2, 3714–3721}	Acetonitrile	-----	-----	---	Colorimetric and fluorogenic Determination	-----
9.	Thiophene anchored naphthalene derivative: Cr ³⁺ selective turn-on fluorescent probe for living cell imaging, Analytical Methods, 2012, 4 , 2254-2258.	Water-Methanol (1:9,v/v)	1.5×10 ⁻⁷ M	1.0×10 ⁻⁶ - 52×10 ⁻⁶ M	-----	Fluorescence Determination	Applicability of sensor to living cells
10.	FRET-based sensor for imaging chromium (III) in living cells, ChemComm, 2008, 3387–3389.	Ethanol-Water (2:1,v/v)	-----	-----	-----	Fluorescent Determination	Monitoring of Cr ³⁺ ion in living cells
11.	Recognition of Hg ²⁺ and Cr ³⁺ in Physiological Conditions by a	CH ₃ CN/ HEPES buffer	5.6 ppm	-----	--	UV-Visible and Fluorescence	Cell imaging studies.

	Rhodamine Derivative and Its Application as a Reagent for Cell-Imaging Studies, Inorganic Chemistry, 2012, 51 , 336-345	solution (1 mM; 3:2, v/v)				Determination	
12.	A novel dual-switch fluorescent probe for Cr(III) ion based on PET-FRET processes, Analyst, 2014, 139 , 3607-3613.	CH ₃ CN-HEPES buffer solution	0.14 nM	30 nM to 80 μM	-----	Fluorescence Probe	-----
13.	A ratiometric fluorescent chemosensor for Cr ³⁺ based on monomer-excimer conversion of a pyrene compound, Sensors and Actuators B: Chemical, 2014, 203 , 712-718	Tris-HNO ₃ buffered solution	4 × 10 ⁻⁸ M	2.0 × 10 ⁻⁷ - 1.0 × 10 ⁻⁵ M	Less than two minutes	Fluorescence Chemosensor	Real sample analysis on river and pond water

Therefore the proposed sensor depicts high selectivity towards Cr³⁺ ion in aqueous medium with limit of detection (64 nM) in nanomolar range which is quite efficient among other reported sensors in literature. Moreover the present sensor is efficiently determining the presence of Cr³⁺ ion in water samples taken from different sources of water.