

Supplementary Information for

**One-Pot, Efficient and Green Synthesis of Acridinedione Derivatives using Highly Monodisperse Platinum Nanoparticles
Supported with Reduced Graphene Oxide**

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Materials and Instrumentation

PtCl₄ (99 % Alfa Aesar), tetrahydrofuran (THF) (99.5 %, Merck) and ethanol (99.9 %) were purchased from Merck, 1-octanamine (Sigma Aldrich) were used as received from suppliers. THF was distilled over sodium under argon atmosphere and stored under inert atmosphere. De-ionized water was filtered by Millipore water purification system (18 MΩ) analytical grade. All glassware and Teflon-coated magnetic stir bars were cleaned with aqua regia, followed by washing with distilled water before drying in an oven.

The chemicals used in the synthesis of acridinedione derivatives were obtained from Merck and Aldrich Chemical Company. All chemicals and solvents used for the synthesis were of spectroscopic reagent grade.

Transmission electron microscopy (TEM) images were obtained on a JEOL 200 kV TEM instrument. Sample preparation for TEM analysis involves placement of a drop of 0.5 mg/mL ethanol solution of the prepared catalysts with a carbon support on a carbon covered 400-mesh copper grid; the solvent is then allowed to evaporate. Excess solution was removed with an adsorbent paper and the sample was dried under vacuum at room temperature before analysis. More than 300 particles were calculated to get the integrated information about the overall distribution of Pt-based catalyst sample.

Thermo Scientific spectrometer was used for X-ray Photoelectron Spectroscopy (XPS) measurements and the X-ray source was K α lines of Mg (1253.6 eV, 10 mA). Samples were prepared by depositing the catalyst on Cu double-sided tape (3M Inc.). C 1s line at 284.6 eV was chosen as a reference point and all XPS peaks were fitted using a Gaussian function and the C 1s line at 284.6 eV was used as the reference line.

A Panalytical Empirian diffractometer with Ultima+theta-theta high resolution goniometer, having an X-ray generator (Cu K α radiation, $k = 1.54056 \text{ \AA}$) and operating condition of 45 kV and 40 mA, were employed in XRD analysis.

Raman spectrum was carried out using an in via Raman microprobe (Renishaw Instruments) with 514 nm laser excitation.

Melting points were measured on a Bibby Scientific Stuart Digital, Advanced, and SMP30. Fourier Transform Infrared (FT-IR) spectra were recorded on Bruker Optics, ALPHA FT-IR spectrometer. The ¹H-NMR and ¹³C-NMR spectra were obtained in DMSO-*d*₆ with Bruker DPX-300 as solvents with tetramethylsilane as the internal reference. The mass analyses were

performed on an Agilent Technologies 6530 Accurate-Mass Q-TOF LC/HRMS at the advanced technology research centre of Dumlupınar University (ILTEM).

Characterization of Pt NPs@rGO

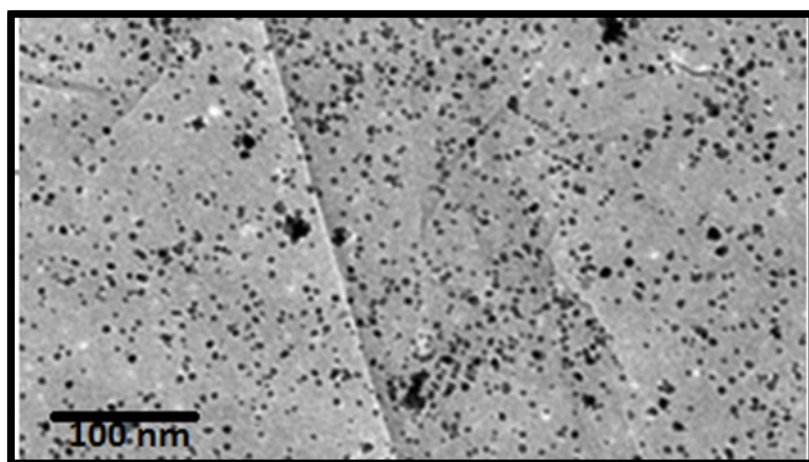


Fig. S1 Low-magnification TEM images of the Pt NPs@rGO

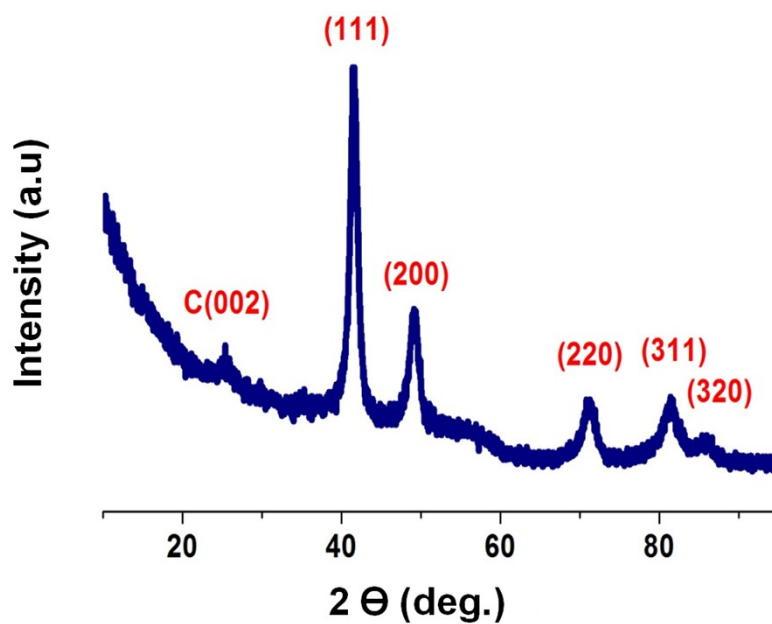


Fig. S2. XRD of Pt NPs@rGO

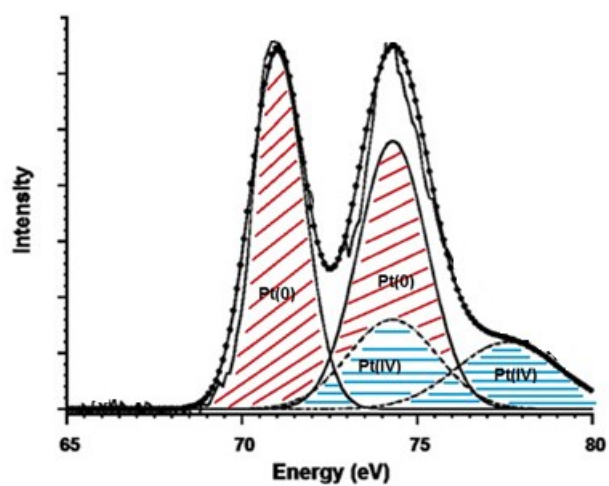


Fig. S3. Pt 4f electron spectra of Pt NPs@rGO

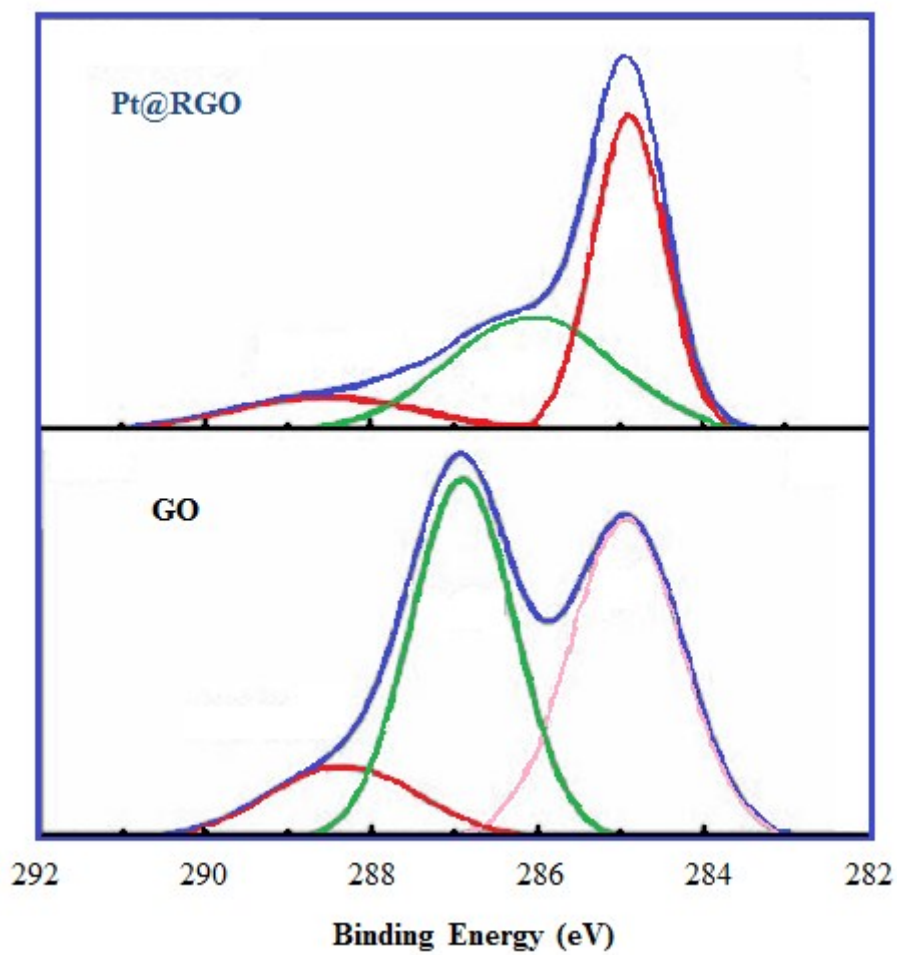


Fig. S4. XPS spectra of C 1s for Pt NPs@rGO and GO

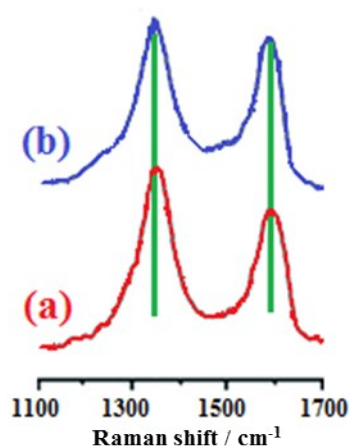


Fig. S5. The Raman spectra of Pt NPs@rGO (a) and RGO (b)

Table S1. The effect of molar ratio (Pt/rGO) on model reaction in the presence of Pt NPs@rGO

Entry	R	R'	Time(min)	Yield (%)	Pt/rGO
4a	-H	4-Cl	60	94	1
4a	-H	4-Cl	60	90	0.7
4a	-H	4-Cl	60	84	0.5

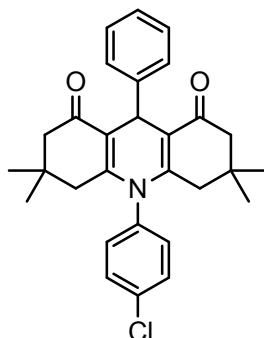
Characterization of acridinedione derivatives

Whereas the infrared (IR) spectra of the novel synthesized 1, 8-dioxoacridine compounds (**4i** and **4j**) aromatic C–H stretching bands are observed between 3056 and 3018 cm^{-1} , the aliphatic C–H stretching bands were observed between 2959 and 2955 cm^{-1} . The compound **4i** of cyan bond was showed at 2223 cm^{-1} and the carbonyl group bonds of compounds **4i** and **4j** were showed sharp peaks between 1632 and 1634 cm^{-1} , respectively.

The ^1H -NMR spectra of the compounds **4i** and **4j** belonging to protons of the methyl groups showed singlet peaks between 0.70–0.90 ppm. The methane ($-\text{CH}_2$) protons of compounds **4i** and **4j** were observed in the region between 1.77–2.21 ppm. The signals for the $-\text{CH}$ protons of all compounds were observed between 4.95–5.10 ppm and the signals for the aromatic protons were observed in the range between 6.61–7.70 ppm. The hydroxyl ($-\text{OH}$) proton of the compound **4j** was showed at 9.05 ppm.

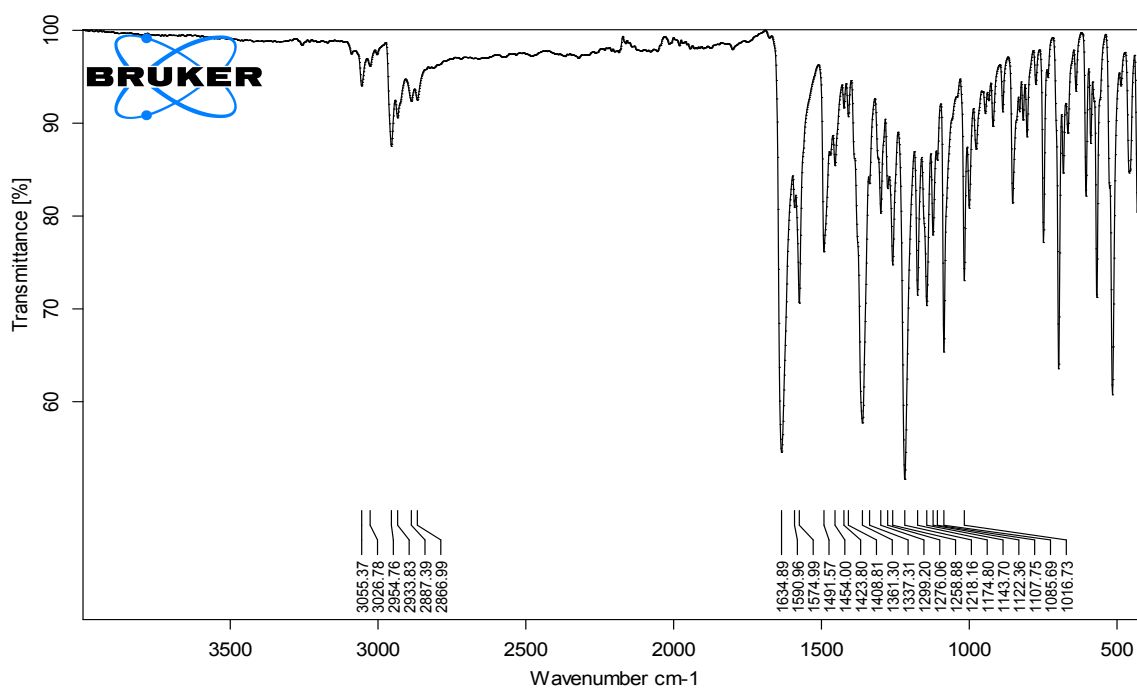
The signals observed in ^{13}C -NMR (APT) spectrums of all novel acridinedione molecules (**4i** and **4j**) are determined to be in line with the recommended molecule structures. Further, when their high resolution mass spectra (HRMS) are examined, the observed molecule ion peaks are in compliance with the recommended structures.

10-(4-Chlorophenyl)-3,3,6,6-tetramethyl-9-phenyl-3,4,6,7,9,10-hexahydroacridine-1,8(2H,5H)-dione (4a)

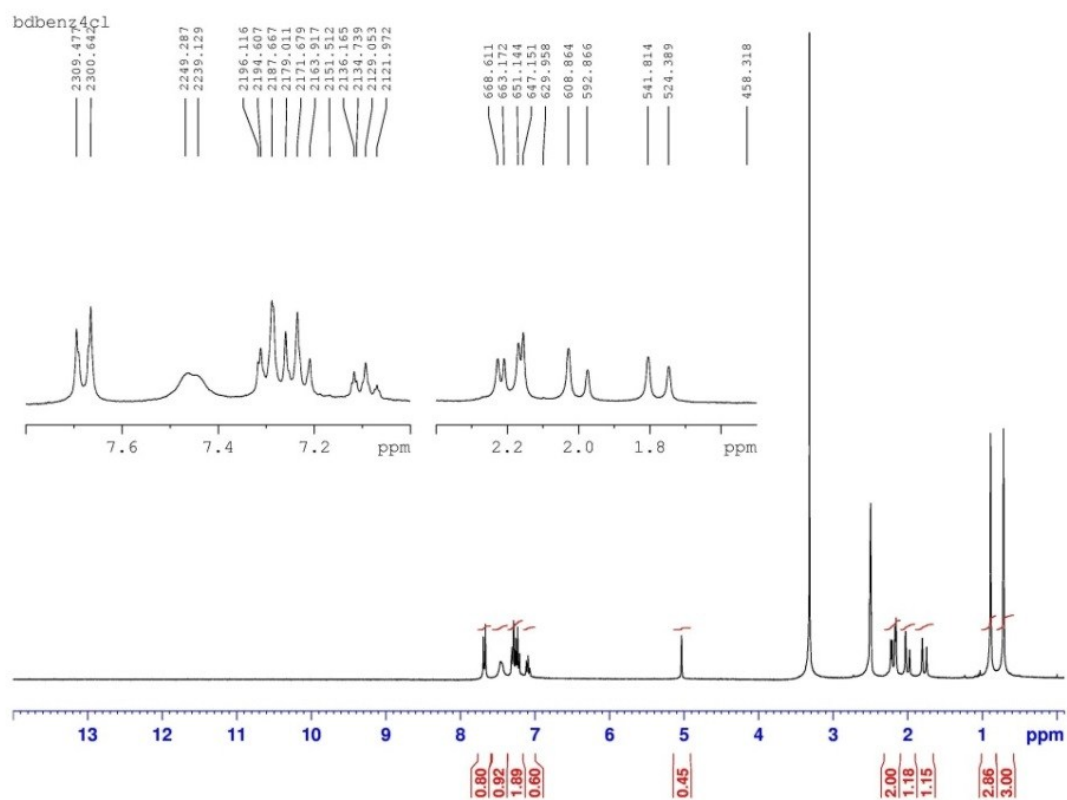


As yellow crystals, mp. (300-302 °C) [30] (ethanol). $^1\text{H-NMR}$ (300 MHz, $\text{DMSO-}d_6$) δ (ppm): 0.72 (s, 6H, 2x- CH_3), 0.88 (s, 6H, 2x- CH_3), 1.78 (d, 2H, $J = 17.43$ Hz, $-\text{CH}_2$), 2.00 (d, 2H, $J = 16.00$ Hz, $-\text{CH}_2$), 2.16-2.22 (m, 4H, $-\text{CH}_2$), 5.05 (s, 1H, $-\text{CH}$), 7.07-7.12 (m, 1H, Ar-H), 7.21-7.32 (m, 4H, Ar-H), 7.48 (d, 2H, $J = 5.44$ Hz, 10.58, Ar-H), 7.68 (d, 2H, $J = 8.84$ Hz, Ar-H); $^{13}\text{C-NMR}$ (75 MHz, $\text{DMSO-}d_6$) δ (ppm): 26.50, 29.72, 32.33, 32.44, 41.35, 50.03, 113.55, 126.26, 127.97, 128.38, 130.56, 134.37, 137.79, 146.57, 150.58, 195.54; IR (cm^{-1}): 3026 w (Ar-H), 2954 s ($-\text{CH}$), 1634 s ($\text{C}=\text{O}$), 1590 s ($\text{C}=\text{C}$); HRMS (QTOF-ESI): m/z $\text{C}_{29}\text{H}_{30}\text{ClNO}_2$: 459.1965; found: 460.2061 ($[\text{M}+\text{H}]^+$).

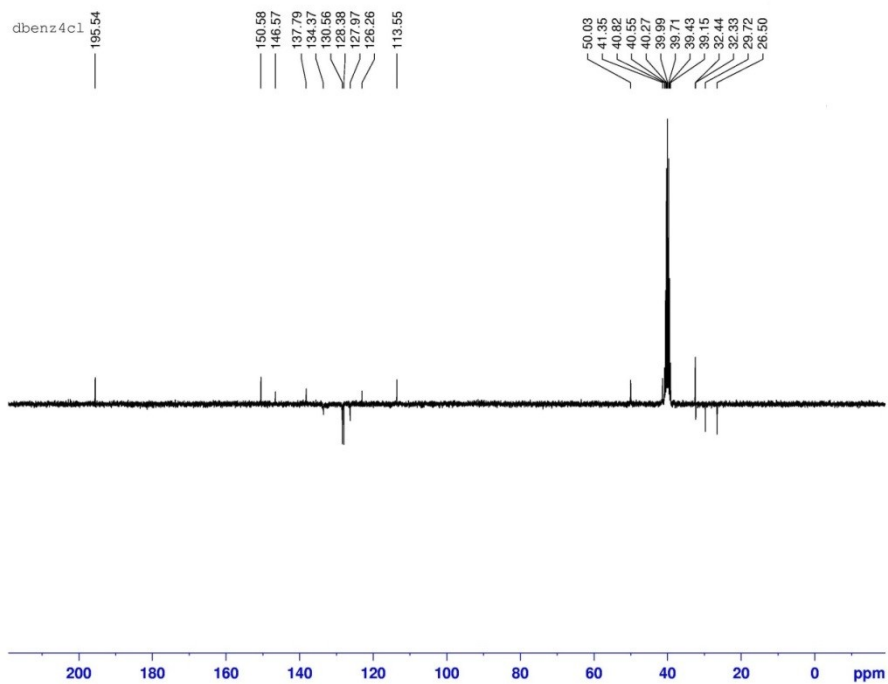
a)



b)



c)



d)

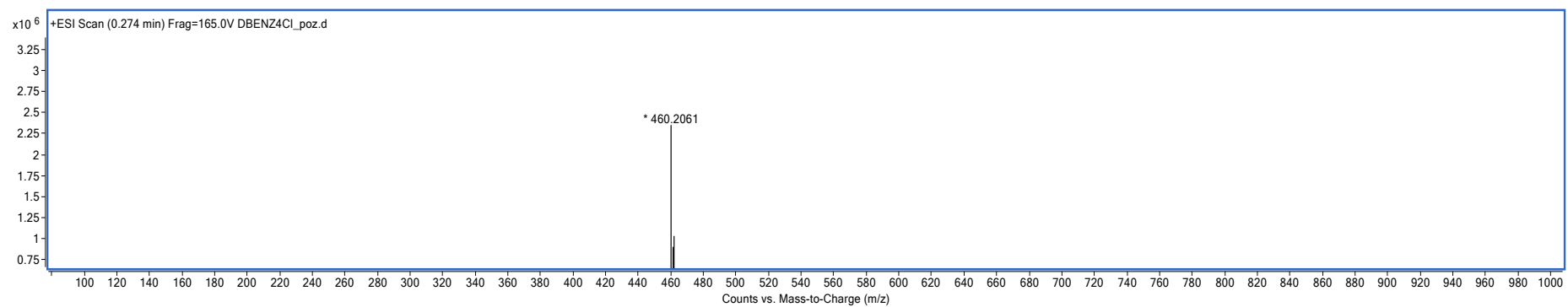
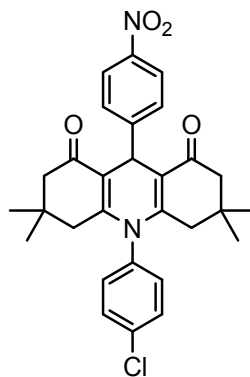


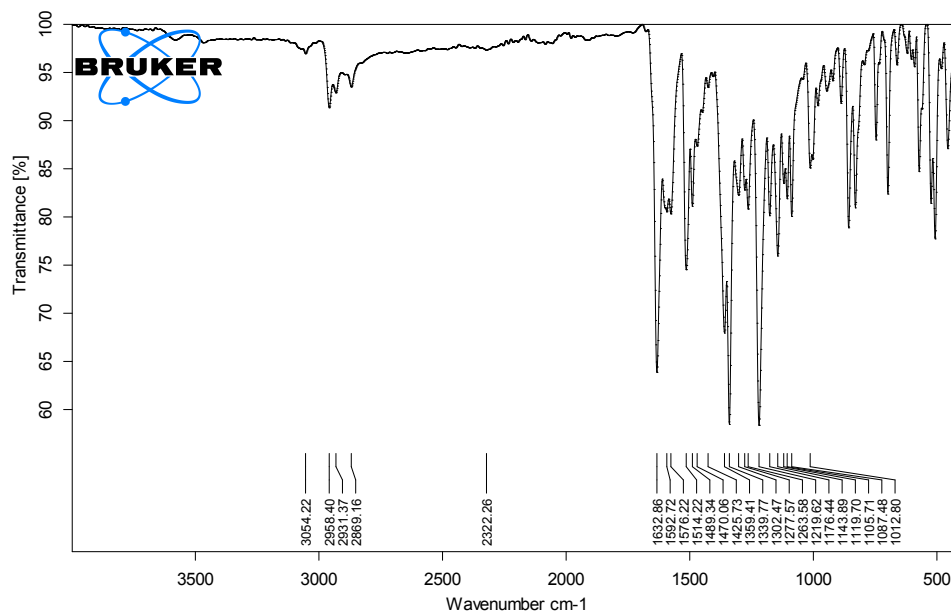
Fig. S6. a) FT-IR, b) ¹H NMR, c) ¹³C NMR (APT), d) Q-TOF LC/HRMS of 4a

10-(4-Chlorophenyl)-3,3,6,6-tetramethyl-9-(4-nitrophenyl)-3,4,6,7,9,10 hexahydroacridine-1,8(2H,5H)-dione (4b)

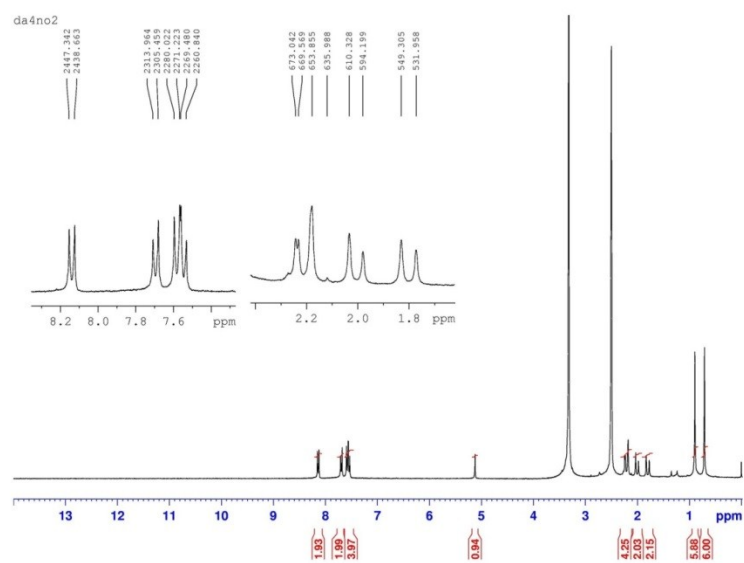


As yellow crystals, mp. (315-317 °C) [31] (ethanol). $^1\text{H-NMR}$ (300 MHz, $\text{DMSO-}d_6$) δ (ppm): 0.70 (s, 6H, 2x- CH_3), 0.90 (s, 6H, 2x- CH_3), 1.80 (d, 2H, $J = 17.35$ Hz, $-\text{CH}_2$), 2.01 (d, 2H, $J = 16.13$ Hz, $-\text{CH}_2$), 2.18-2.24 (m, 4H, $-\text{CH}_2$), 5.10 (s, 1H, $-\text{CH}$), 7.54-7.60 (m, 4H, Ar-H), 7.70 (d, 2H, $J = 8.51$ Hz, Ar-H), 8.14 (d, 2H, $J = 8.68$ Hz, Ar-H); $^{13}\text{C-NMR}$ (75 MHz, $\text{DMSO-}d_6$) δ (ppm): 26.57, 29.61, 32.47, 33.35, 41.39, 49.86, 112.51, 119.43, 123.76, 129.43, 134.55, 137.55, 146.16, 151.32, 154.01, 195.53; IR (cm^{-1}): 3054 w (Ar-H), 2958 w ($-\text{CH}$), 1632 s (C=O), 1592 w (C=C); HRMS (QTOF-ESI): m/z calcd. For $\text{C}_{29}\text{H}_{29}\text{ClN}_2\text{O}_4$: 504.1816; found: 527.1729 ($[\text{M}+\text{Na}]^+$).

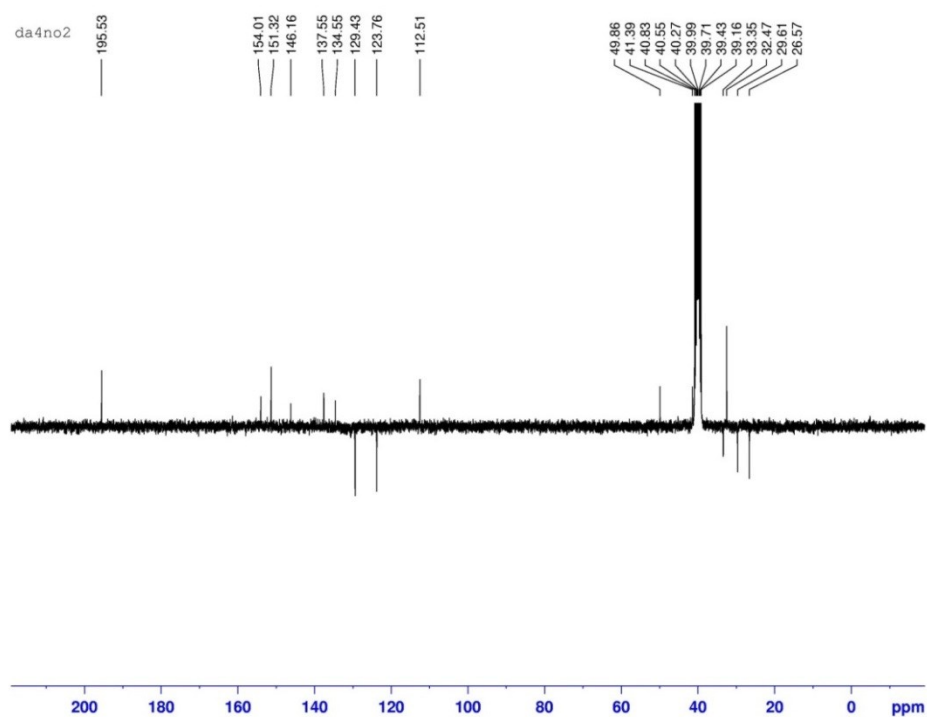
a)



b)



c)



d)

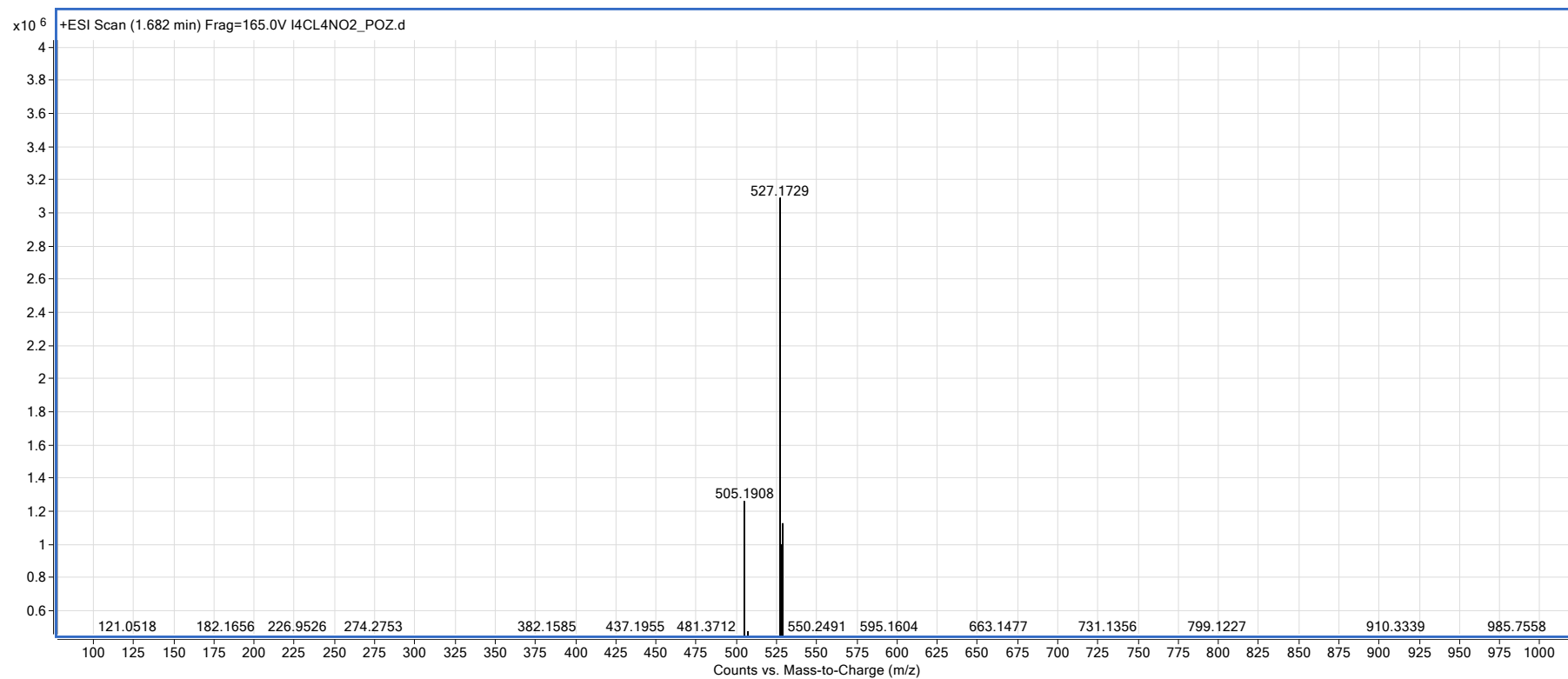
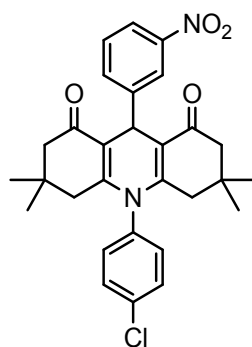


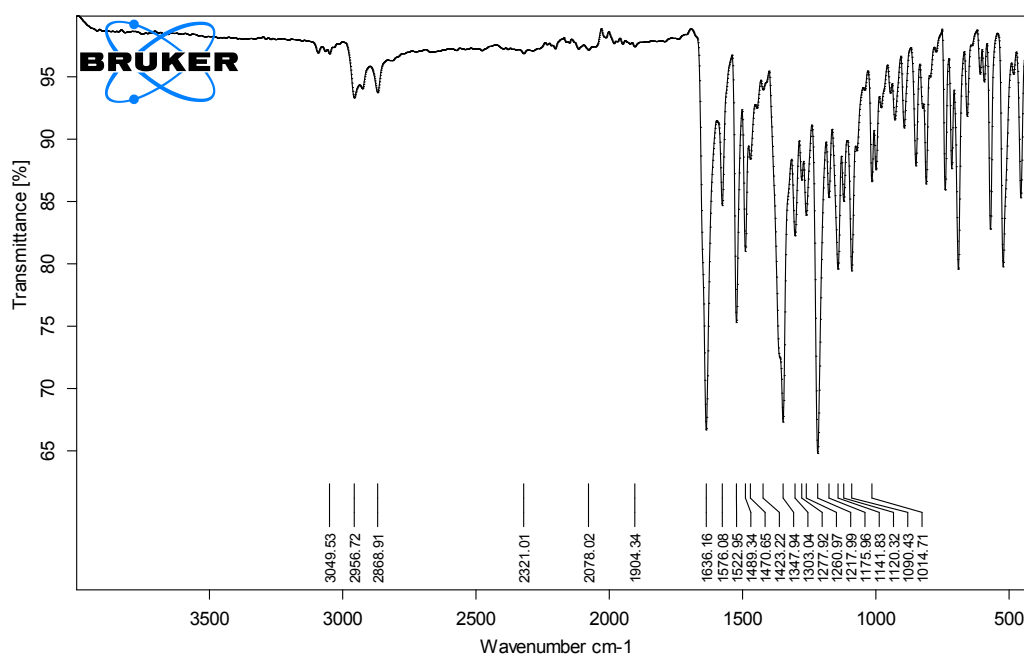
Fig. S7. a) FT-IR, b) ^1H NMR, c) ^{13}C NMR (APT), d) Q-TOF LC/HRMS of **4b**

10-(4-Chlorophenyl)-3,3,6,6-tetramethyl-9-(3-nitrophenyl)-3,4,6,7,9,10 hexahydroacridine-1,8(2H,5H)-dione (4c)

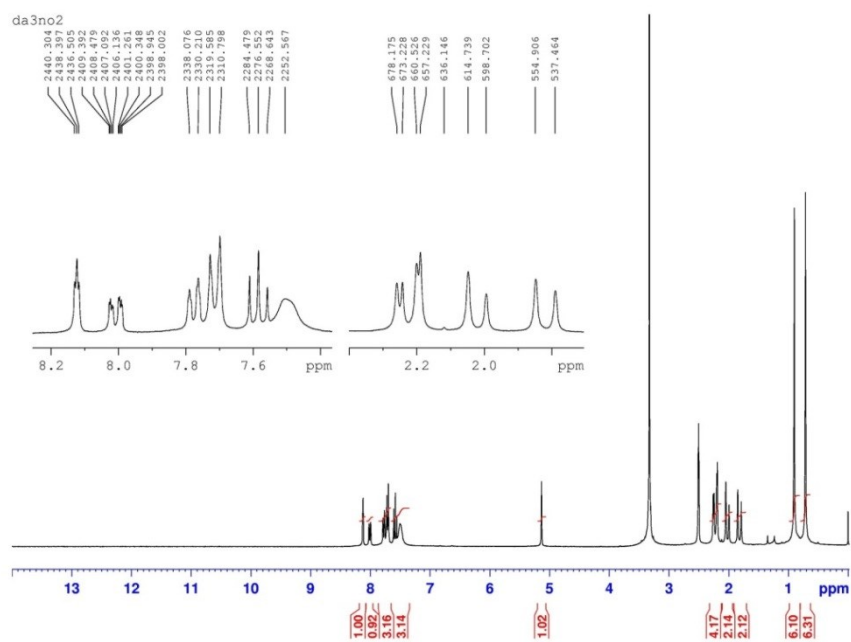


As yellow crystals, mp. (285-287 °C) [32] (ethanol). $^1\text{H-NMR}$ (300 MHz, $\text{DMSO-}d_6$) δ (ppm): 0.72 (s, 6H, 2x- CH_3), 0.88 (s, 6H, 2x- CH_3), 1.82 (d, 2H, $J=17.44$ Hz, $-\text{CH}_2$), 2.02 (d, 2H, $J=16.04$ Hz, $-\text{CH}_2$), 2.20 (d, 2H, $J=3.30$ Hz, $-\text{CH}_2$), 2.25 (d, 2H, $J=4.95$ Hz, $-\text{CH}_2$), 5.15 (s, 1H, $-\text{CH}$), 7.51-7.61 (m, 3H, Ar-H), 7.72 (d, 2H, $J=8.79$ Hz, Ar-H), 7.78 (d, 1H, $J=7.87$ Hz, Ar-H), 7.99-8.03 (m, 1H, Ar-H), 8.12-8.13 (m, 1H, Ar-H); $^{13}\text{C-NMR}$ (75 MHz, $\text{DMSO-}d_6$) δ (ppm): 26.47, 29.62, 32.52, 32.92, 41.34, 49.84, 112.72, 121.50, 122.56, 130.21, 130.65, 134.58, 134.75, 137.52, 147.88, 148.60, 151.39, 195.63; IR (cm^{-1}): 3049 w (Ar-H), 2956 w ($-\text{CH}$), 1636 s ($\text{C}=\text{O}$), 1576 s ($\text{C}=\text{C}$); HRMS (QTOF-ESI): m/z calcd. For $\text{C}_{29}\text{H}_{29}\text{ClN}_2\text{O}_4$: 504.1816; found: 527.1655 ($[\text{M}+\text{Na}]^+$).

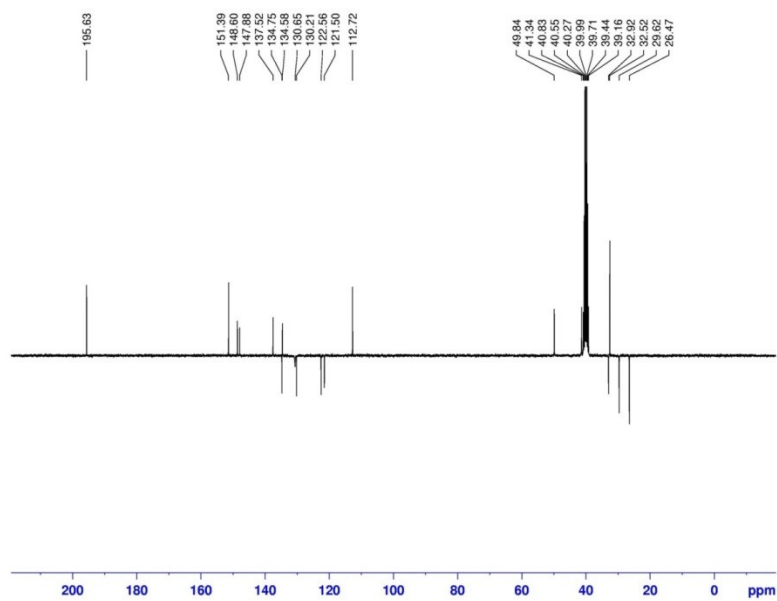
a)



b)



c)



d)

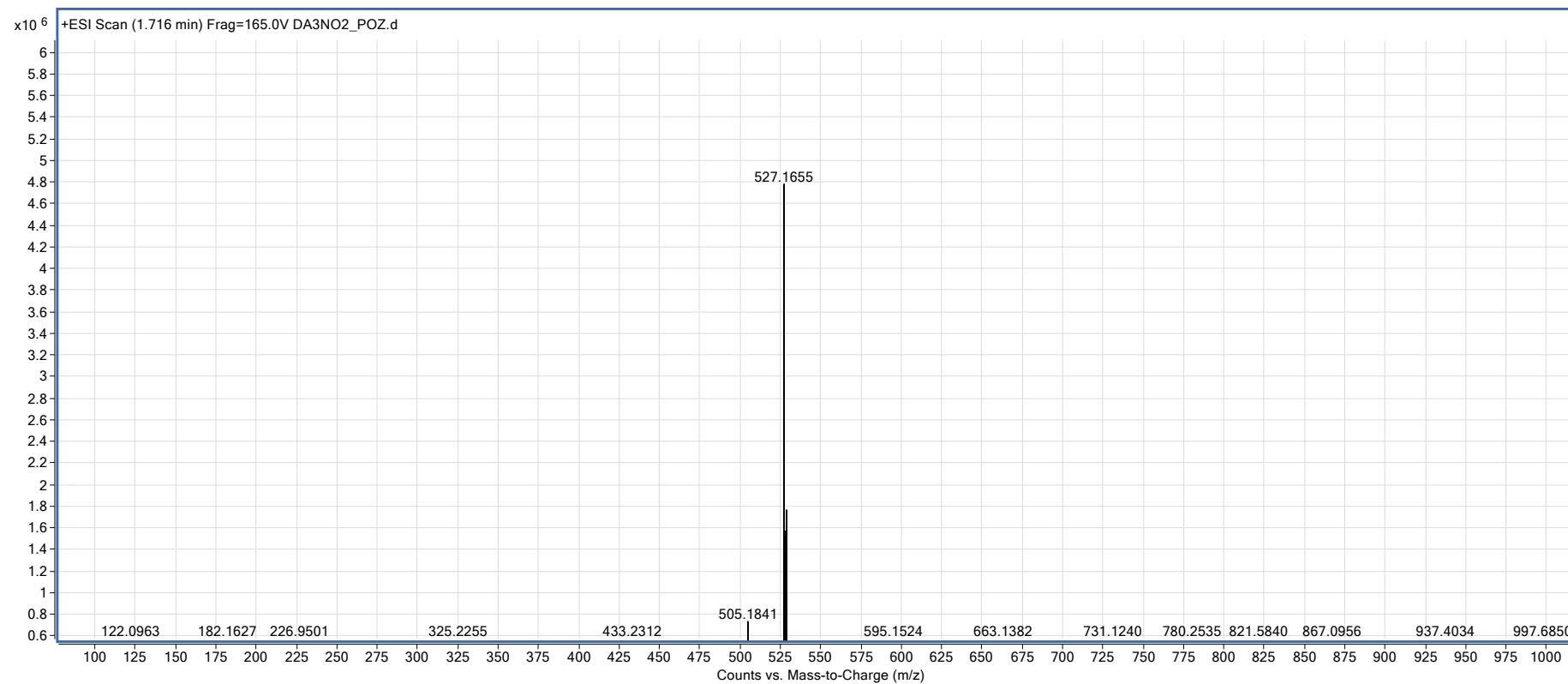
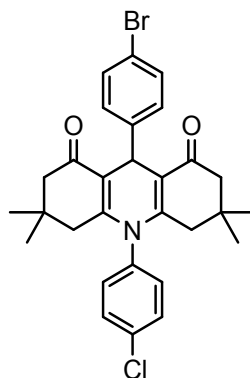


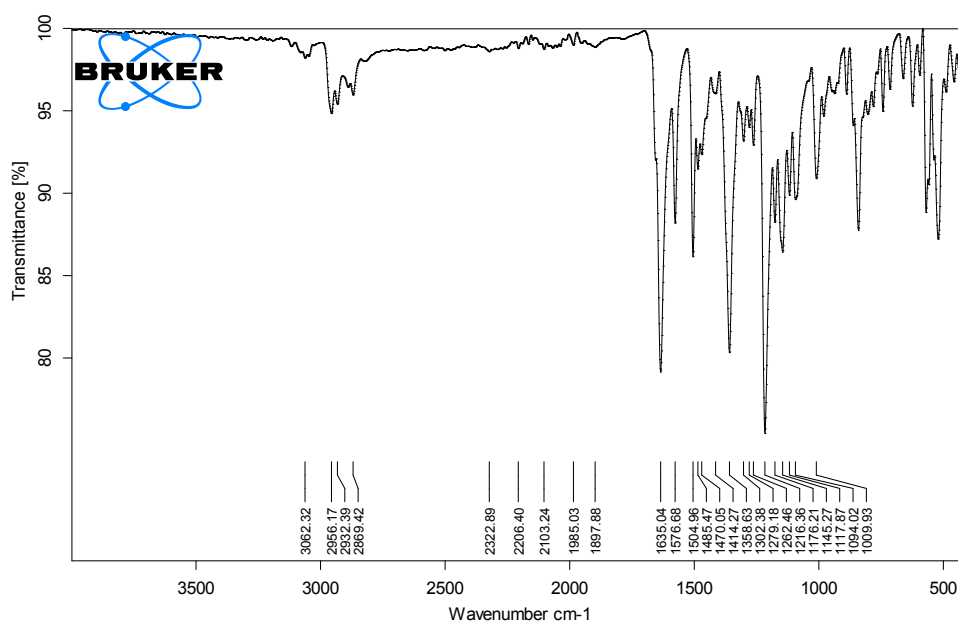
Fig. S8. a) FT-IR, b) ^1H NMR, c) ^{13}C NMR (APT), d) Q-TOF LC/HRMS of **4c**

9-(4-Bromophenyl)-10-(4-chlorophenyl)-3,3,6,6-tetramethyl-3,4,6,7,9,10 hexahydroacridine-1,8(2H,5H)-dione (4d)

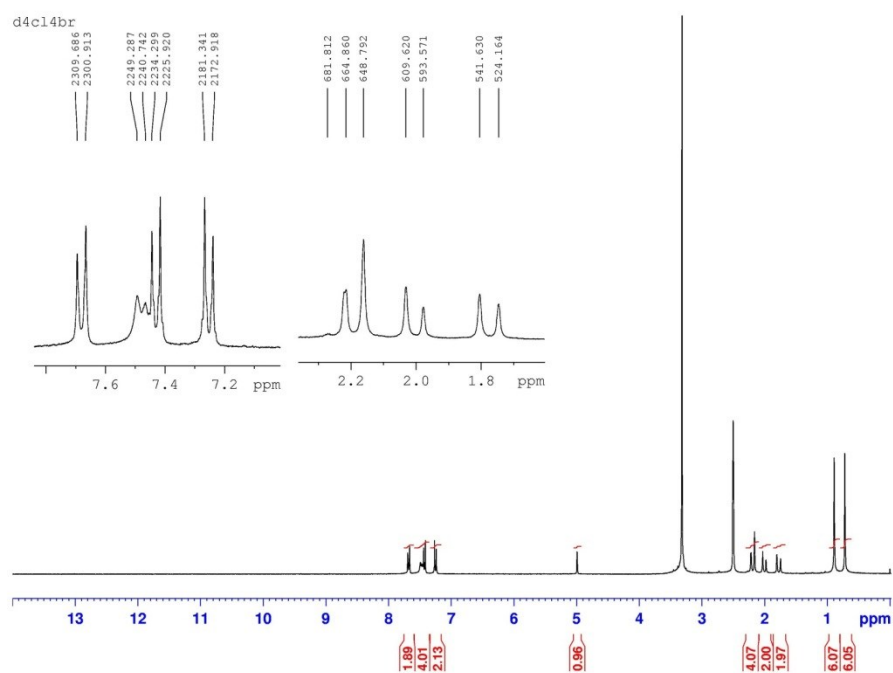


As yellow crystals, mp. (304-305 °C) [31] (ethanol). $^1\text{H-NMR}$ (300 MHz, $\text{DMSO-}d_6$) δ (ppm): 0.70 (s, 6H, 2x- CH_3), 0.90 (s, 6H, 2x- CH_3), 1.78 (d, 2H, $J = 17.47$ Hz, $-\text{CH}_2$), 2.01 (d, 2H, $J = 16.05$ Hz, $-\text{CH}_2$), 2.16-2.19 (m, 4H, $-\text{CH}_2$), 5.00 (s, 1H, $-\text{CH}$), 7.26 (d, 2H, $J = 8.42$ Hz, Ar-H), 7.42-7.50 (m, 4H, Ar-H), 7.68 (d, 2H, $J = 8.77$ Hz, Ar-H); $^{13}\text{C-NMR}$ (75 MHz, $\text{DMSO-}d_6$) δ (ppm): 26.56, 29.66, 32.27, 32.45, 41.35, 49.96, 113.08, 119.25, 130.33, 130.54, 131.26, 134.44, 137.67, 145.96, 150.81, 195.54; IR (cm^{-1}): 3062 w (Ar-H), 2956 w ($-\text{CH}$), 1635 s (C=O), 1576 s (C=C); HRMS (QTOF-ESI): m/z calcd. For $\text{C}_{29}\text{H}_{29}\text{BrClNO}_2$: 537.1070; found: 538.1123 ($[\text{M}+\text{H}]^+$).

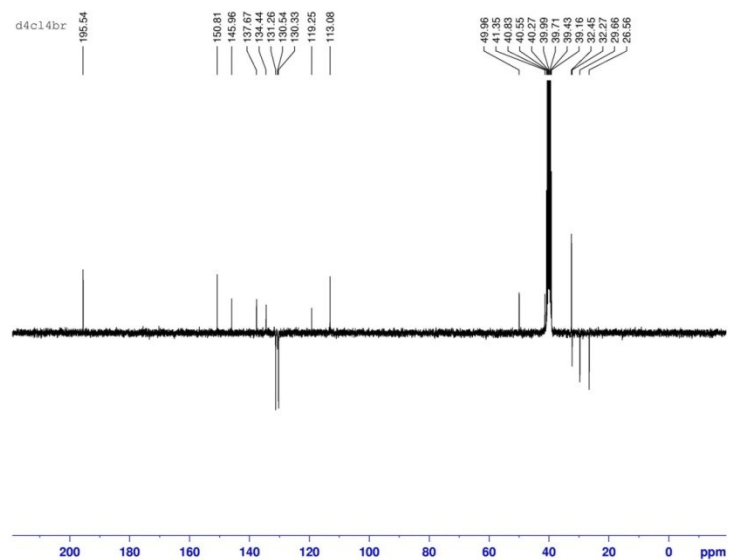
a)



b)



c)



d)

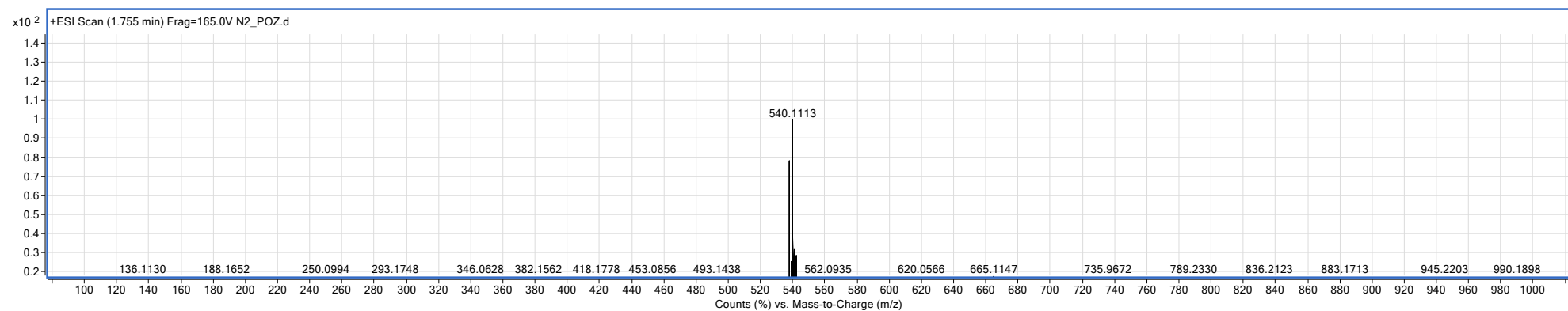
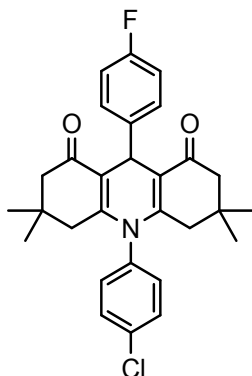


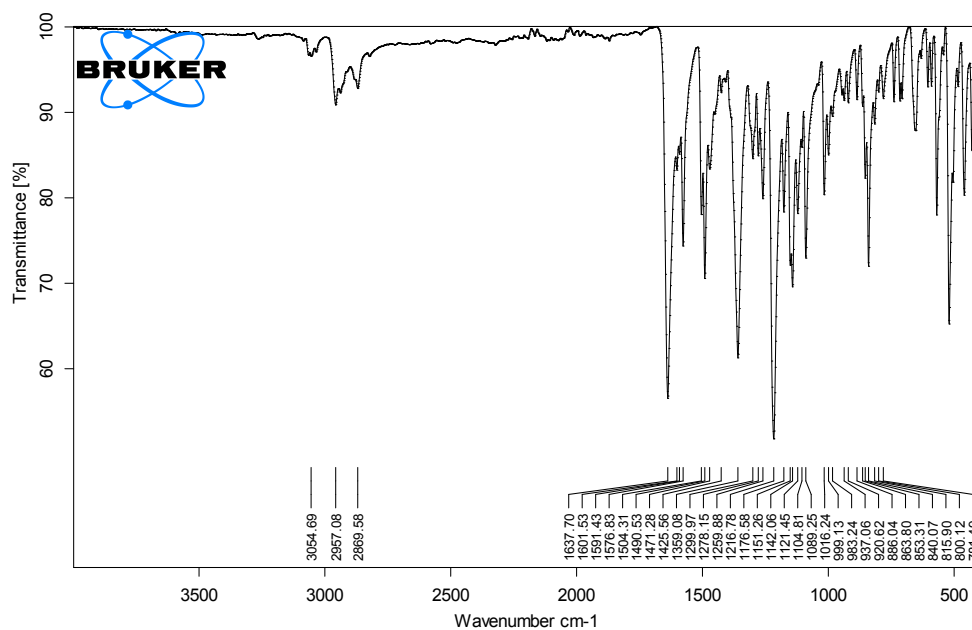
Fig. S9. a) FT-IR, b) ^1H NMR, c) ^{13}C NMR (APT), d) Q-TOF LC/HRMS of **4d**

10-(4-Chlorophenyl)-9-(4-fluorophenyl)-3,3,6,6-tetramethyl-3,4,6,7,9,10-hexahydroacridine-1,8 (2H,5H)-dione (4e)

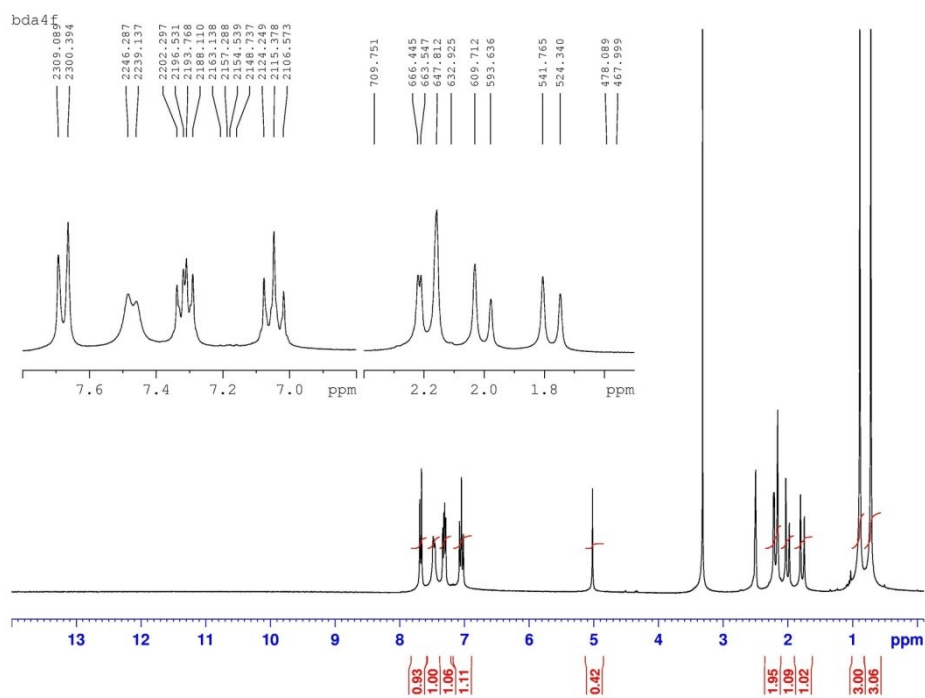


As yellow crystals, mp. (280-282 °C) [33] (ethanol). $^1\text{H-NMR}$ (300 MHz, $\text{DMSO-}d_6$) δ (ppm): 0.70 (s, 6H, 2x- CH_3), 0.90 (s, 6H, 2x- CH_3), 1.78 (d, 2H, $J = 17.43$ Hz, $-\text{CH}_2$), 2.01 (d, 2H, $J = 16.08$ Hz, $-\text{CH}_2$), 2.16-2.22 (m, 4H, $-\text{CH}_2$), 5.05 (s, 1H, $-\text{CH}$), 7.05 (t, 2H, $J = 8.84$ Hz, Ar-H), 7.29-7.34 (m, 2H, Ar-H), 7.48 (d, 2H, $J = 7.15$ Hz, Ar-H), 7.68 (d, 2H, $J = 8.70$ Hz, Ar-H); $^{13}\text{C-NMR}$ (75 MHz, $\text{DMSO-}d_6$) δ (ppm): 26.52, 29.68, 31.84, 32.45, 41.35, 49.98, 113.42, 114.86, 129.79, 130.53, 134.42, 137.72, 142.80, 150.64, 159.29, 195.55; IR (cm^{-1}): 3054 w (Ar-H), 2957 s ($-\text{CH}$), 1637 s (C=O), 1576 s (C=C); HRMS (QTOF-ESI): m/z calcd. For $\text{C}_{29}\text{H}_{29}\text{ClFNO}_2$: 477.1871; found: 478.1956 ($[\text{M}+\text{H}]^+$).

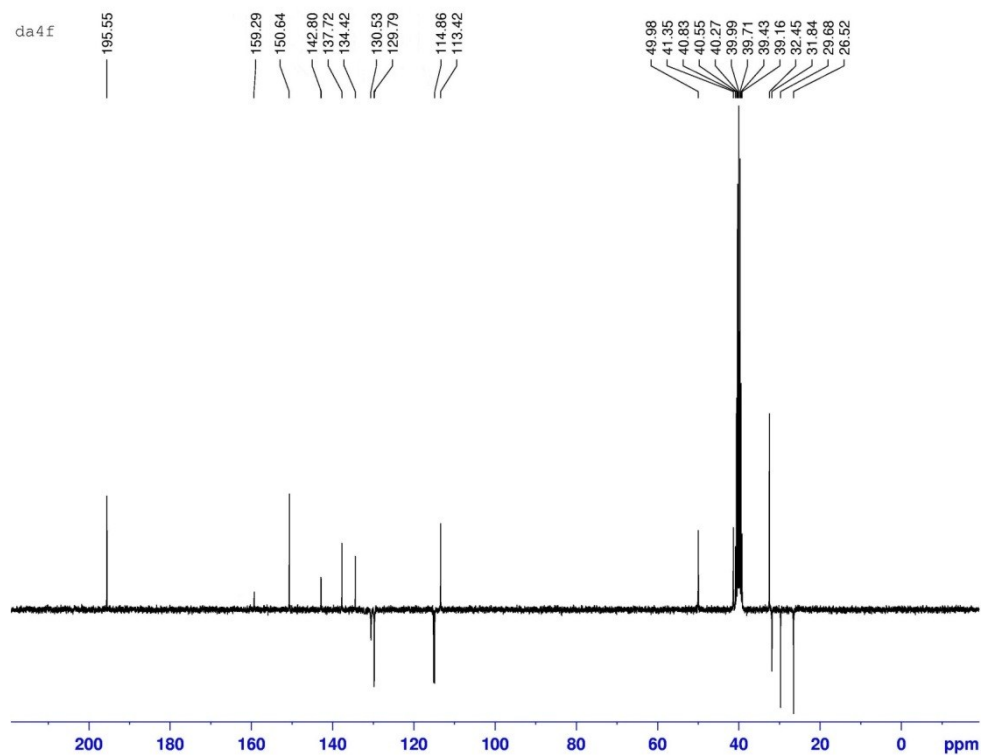
a)



b)



c)



d)

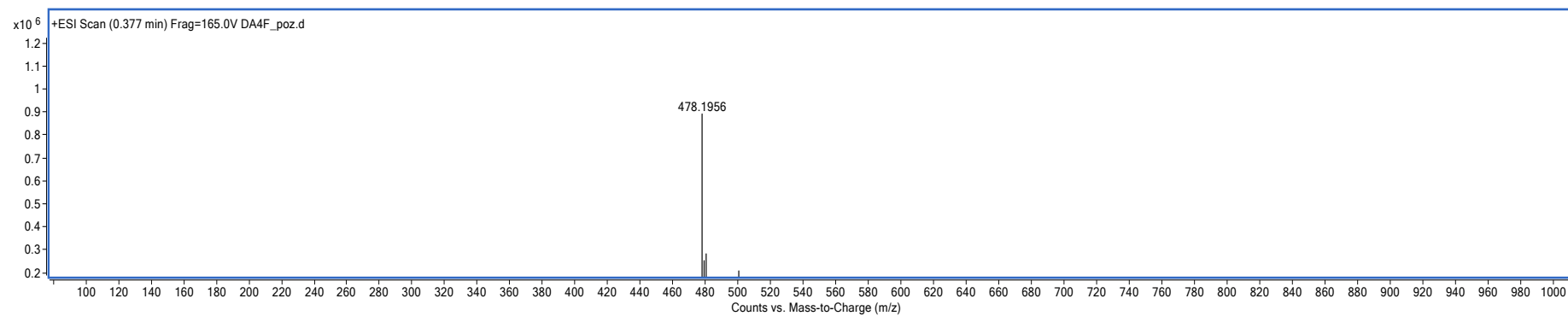
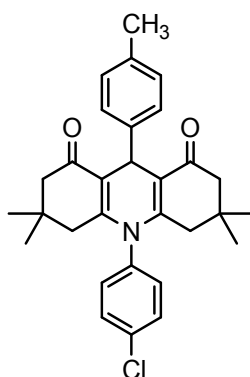


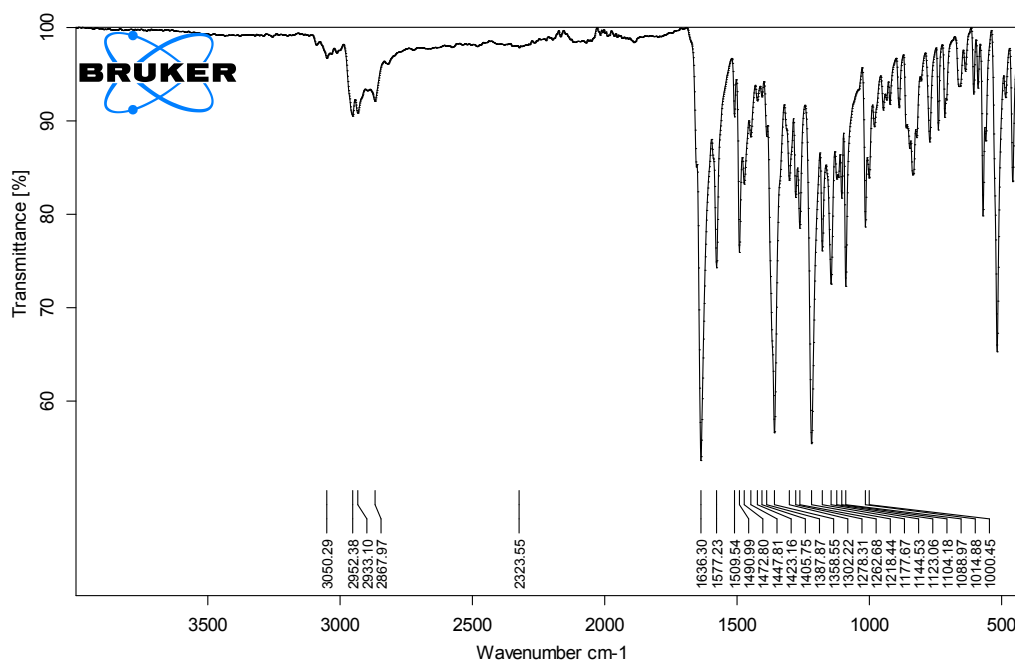
Fig. S10. a) FT-IR, **b)** ^1H NMR, **c)** ^{13}C NMR (APT), **d)** Q-TOF LC/HRMS of **4e**

10-(4-Chlorophenyl)-3,3,6,6-tetramethyl-9-(*p*-tolyl)-3,4,6,7,9,10-hexahydroacridine-1,8(2H,5H)-dione (4f)



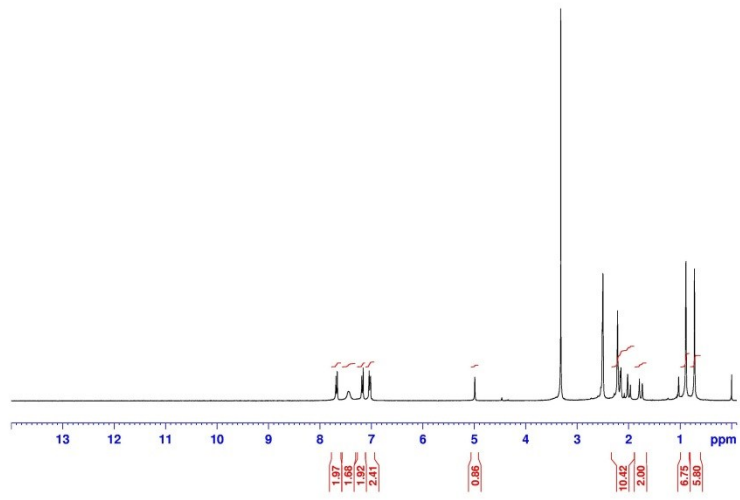
As yellow crystals, mp. (262-265 °C) [32] (ethanol). $^1\text{H-NMR}$ (300 MHz, $\text{DMSO-}d_6$) δ (ppm): 0.75 (s, 6H, 2x- CH_3), 0.80 (s, 6H, 2x- CH_3), 1.76 (d, 2H, $J = 17.29$ Hz, $-\text{CH}_2$), 1.97-2.22 (m, 9H, $-\text{CH}_2$ and $-\text{CH}_3$), 5.00 (s, 1H, $-\text{CH}$), 7.04 (d, 2H, $J = 8.05$ Hz, Ar-H), 7.18 (d, 2H, $J = 8.00$ Hz, Ar-H), 7.35-7.55 (m, 2H, Ar-H), 7.68 (d, 2H, $J = 8.82$ Hz, Ar-H); $^{13}\text{C NMR}$ (75 MHz, $\text{DMSO-}d_6$) δ (ppm): 26.53, 29.15, 31.83, 32.32, 32.43, 41.33, 50.05, 113.69, 127.87, 128.40, 128.96, 134.34, 135.10, 137.83, 143.69, 150.43, 195.54; IR (cm^{-1}): 3050 w (Ar-H), 2952 w ($-\text{CH}$), 1636 s (C=O), 1577 s (C=C); HRMS (QTOF-ESI): m/z calcd. For $\text{C}_{30}\text{H}_{32}\text{ClNO}_2$: 473.2122; found: 474.2215 ($[\text{M}+\text{H}]^+$).

a)

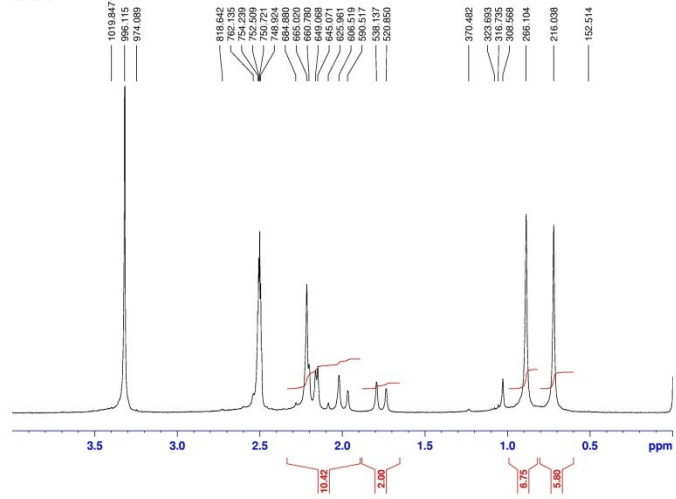


b)

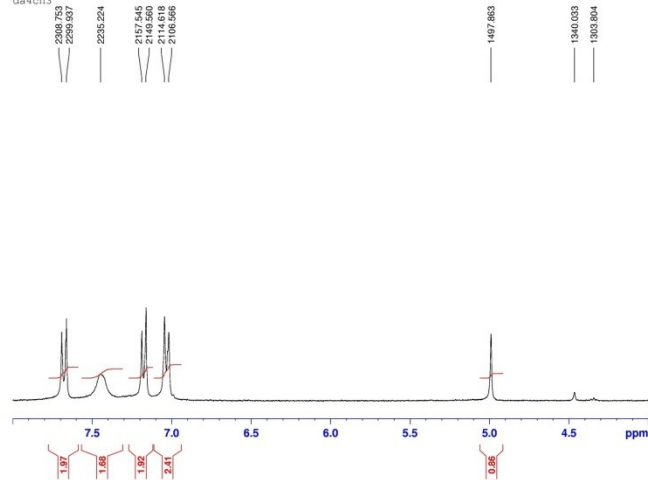
da4ch3



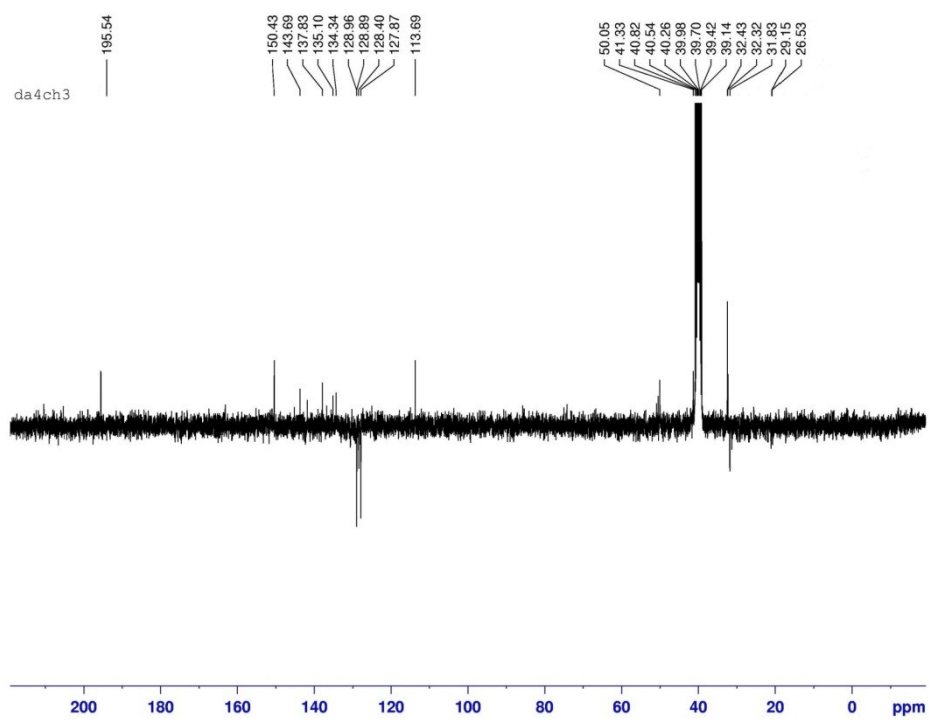
da4ch3



da4ch3



c)



d)

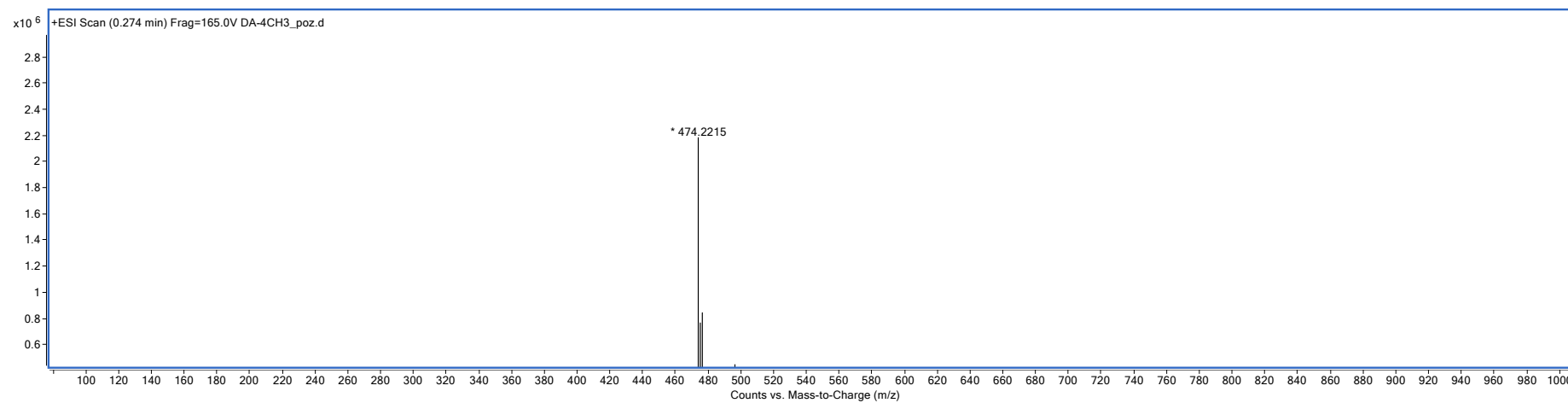
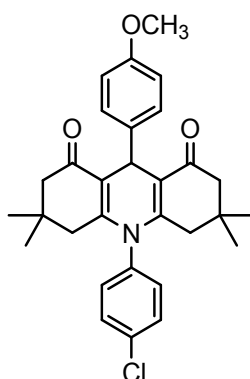


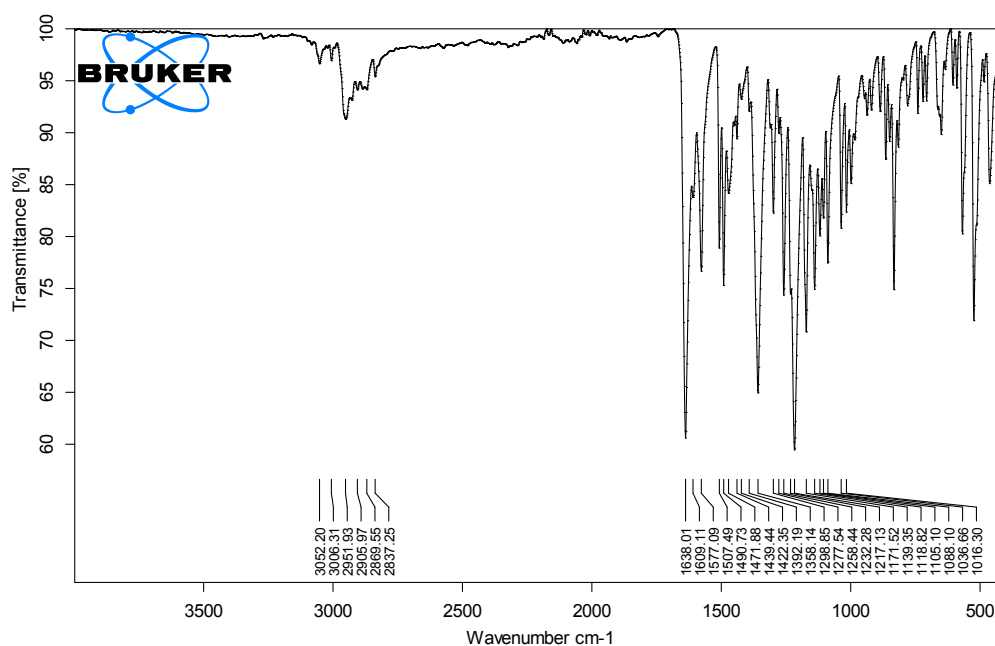
Fig. S11. a) FT-IR, b) ^1H NMR, c) ^{13}C NMR (APT), d) Q-TOF LC/HRMS of **4f**

10-(4-Chlorophenyl)-9-(4-methoxyphenyl)-3,3,6,6-tetramethyl-3,4,6,7,9,10-hexahydro acridine-1,8(2H,5H)-dione (4g)



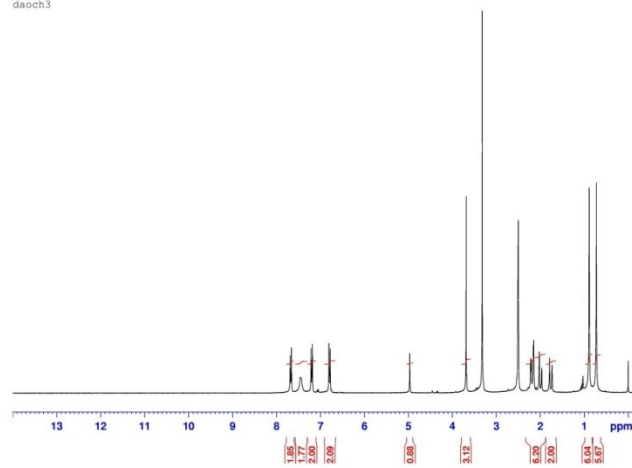
As yellow crystals, mp. (255-257 °C) [34] (ethanol). $^1\text{H-NMR}$ (300 MHz, $\text{DMSO-}d_6$) δ (ppm): 0.75 (s, 6H, 2x- CH_3), 0.85 (s, 6H, 2x- CH_3), 1.76 (d, 2H, $J = 17.40$ Hz, $-\text{CH}_2$), 1.97-2.22 (m, 6H, $-\text{CH}_2$), 3.70 (s, 1H, $-\text{OCH}_3$), 4.95 (s, 1H, $-\text{CH}$), 6.80 (d, 2H, $J = 8.62$ Hz, Ar-H), 7.20 (d, 2H, $J = 8.61$ Hz, Ar-H), 7.40-7.55 (m, 2H, Ar-H), 7.68 (d, 2H, $J = 8.78$ Hz, Ar-H); $^{13}\text{C NMR}$ (75 MHz, $\text{DMSO-}d_6$) δ (ppm): 26.56, 29.73, 31.39, 32.44, 50.07, 55.31, 113.71, 113.83, 128.93, 130.56, 134.34, 137.85, 138.92, 150.28, 157.76, 195.55; IR (cm^{-1}): 3006 w (Ar-H), 2951 w ($-\text{CH}$), 1638 s (C=O), 1577 s (C=C); HRMS (QTOF-ESI): m/z calcd. For $\text{C}_{30}\text{H}_{32}\text{ClNO}_3$: 489.2071; found: 490.2160 ($[\text{M}+\text{H}]^+$).

a)

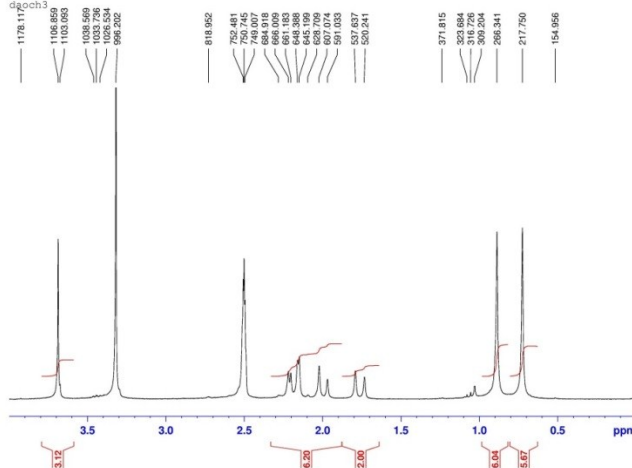


b)

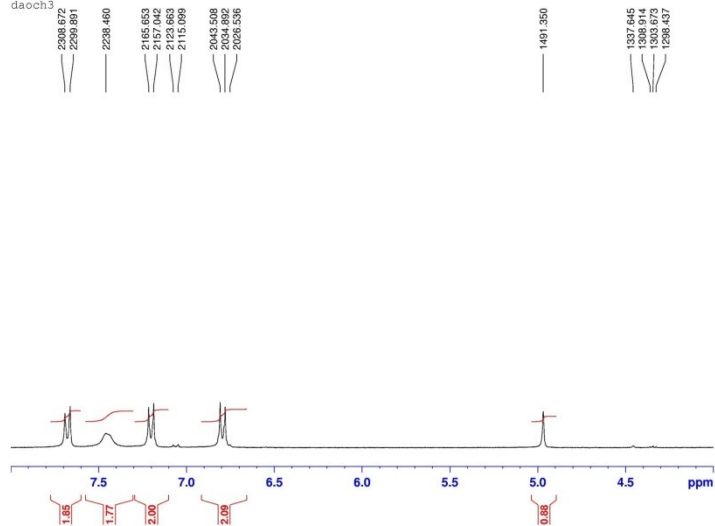
daoeh3



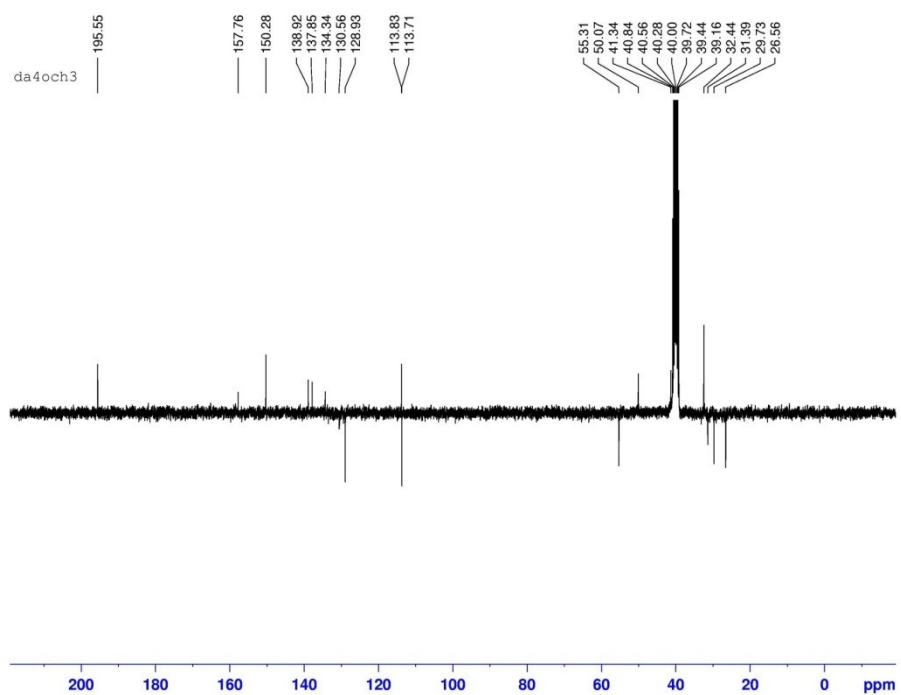
daoeh3



daoeh3



c)



d)

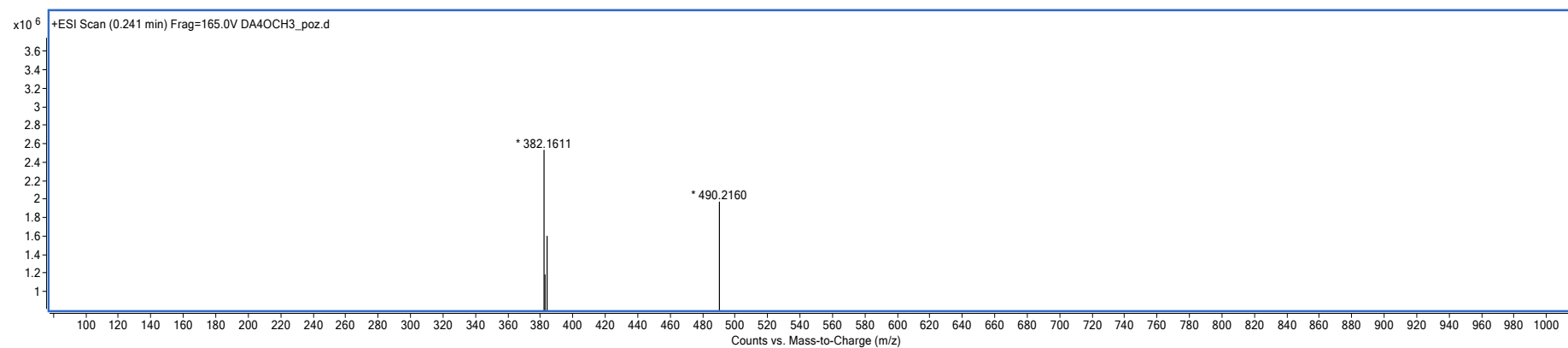
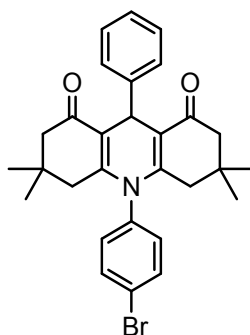


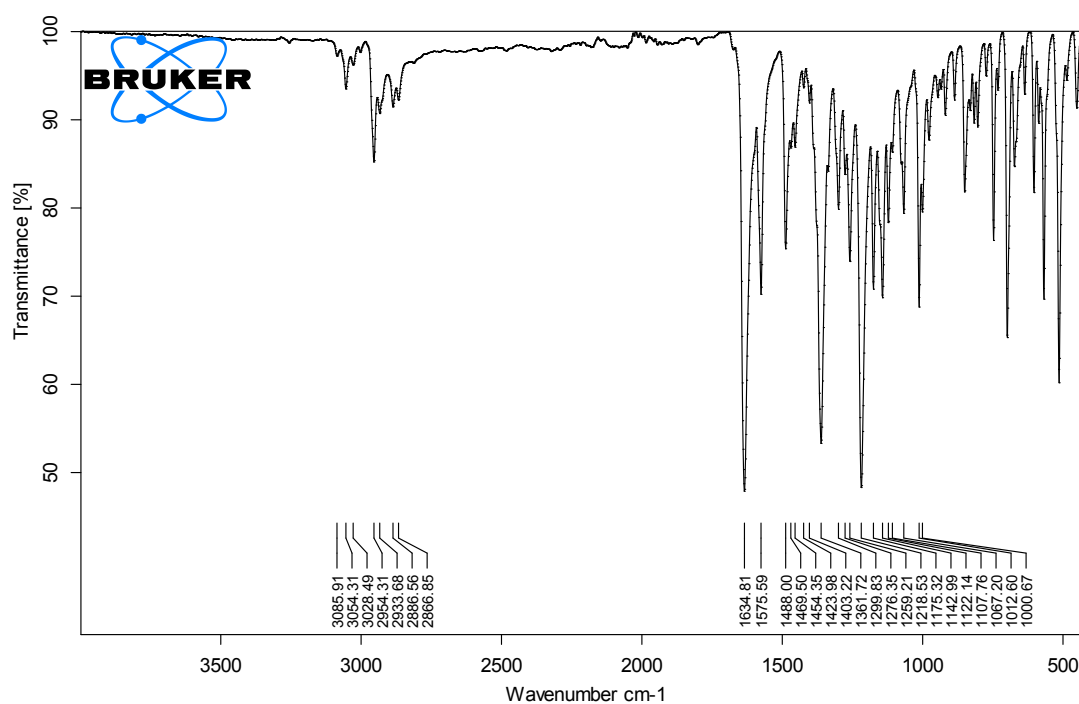
Fig. S12. a) FT-IR, b) ^1H NMR, c) ^{13}C NMR (APT), d) Q-TOF LC/HRMS of **4g**

10-(4-Bromophenyl)-3,3,6,6-tetramethyl-9-phenyl-3,4,6,7,9,10-hexahydroacridine-1,8(2H,5H)-dione (4h)

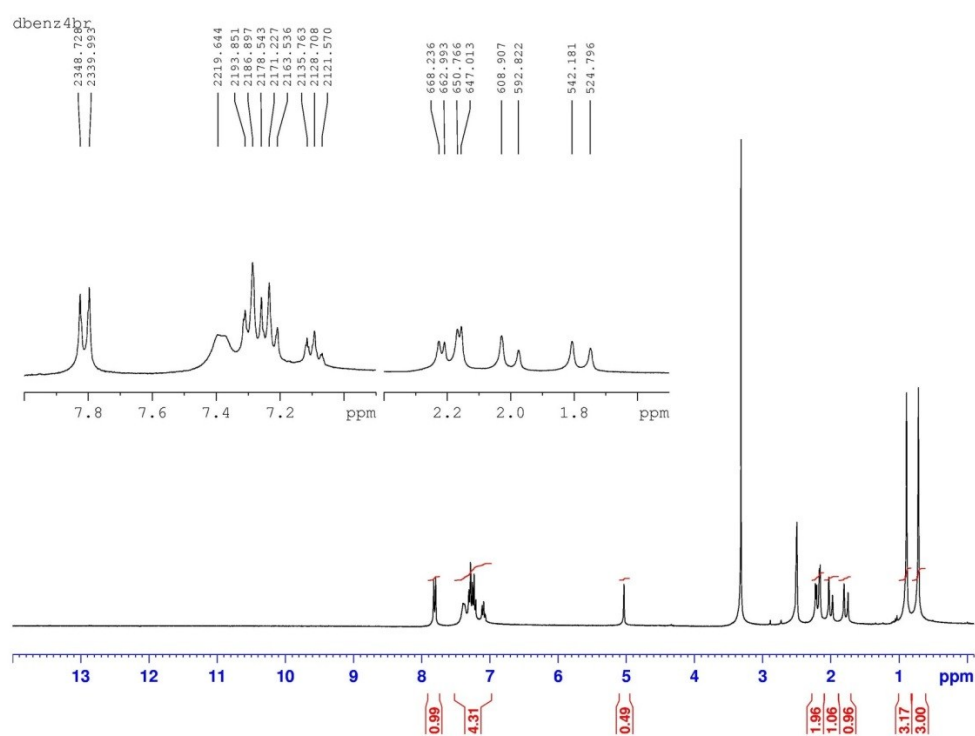


As yellow crystals, mp. (303-305 °C) [35] (ethanol). $^1\text{H-NMR}$ (300 MHz, $\text{DMSO-}d_6$) δ (ppm): 0.70 (s, 6H, 2x- CH_3), 0.90 (s, 6H, 2x- CH_3), 1.78 (d, 2H, $J = 17.39$ Hz, $-\text{CH}_2$), 2.00 (d, 2H, $J = 16.90$ Hz, $-\text{CH}_2$), 2.16-2.23 (m, 4H, $-\text{CH}_2$), 5.05 (s, 1H, $-\text{CH}$), 7.07-7.40 (m, 7H, Ar-H), 7.81 (d, 2H, $J = 8.74$ Hz, Ar-H); $^{13}\text{C-NMR}$ (75 MHz, $\text{DMSO-}d_6$) δ (ppm): 26.50, 29.71, 32.32, 32.45, 41.35, 50.03, 113.54, 123.01, 126.27, 127.97, 128.37, 133.54, 138.22, 146.57, 150.51, 195.54; IR (cm^{-1}): 3028 s (Ar-H), 2954 s ($-\text{CH}$), 1634 s (C=O), 1575 s (C=C); HRMS (QTOF-ESI): m/z calcd. For $\text{C}_{29}\text{H}_{30}\text{BrNO}_2$: 503.1460; found: 504.1529 ($[\text{M}+\text{H}]^+$).

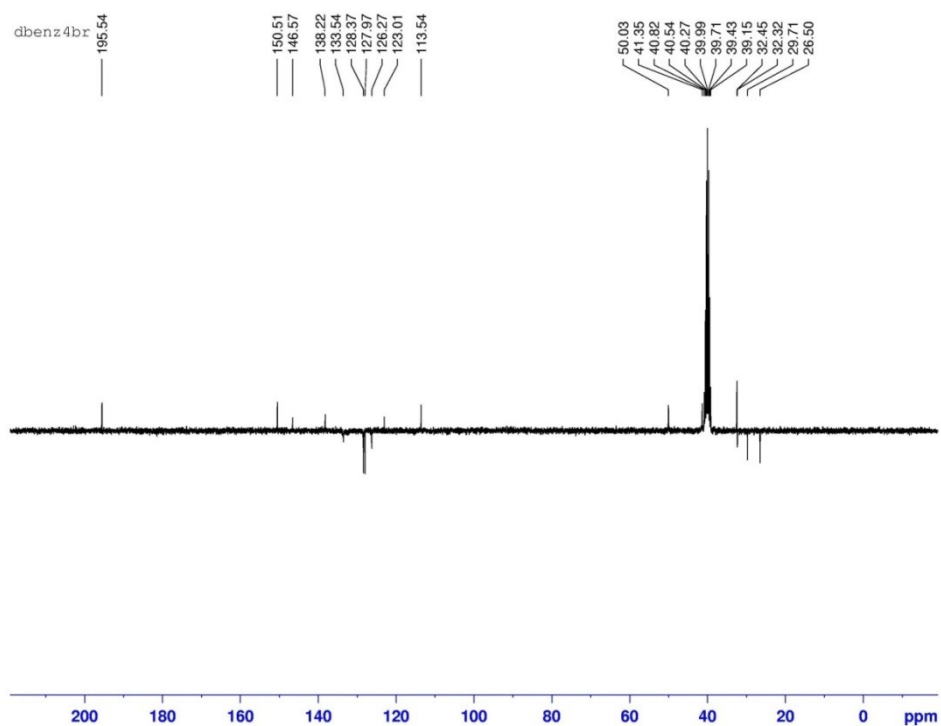
a)



b)



c)



d)

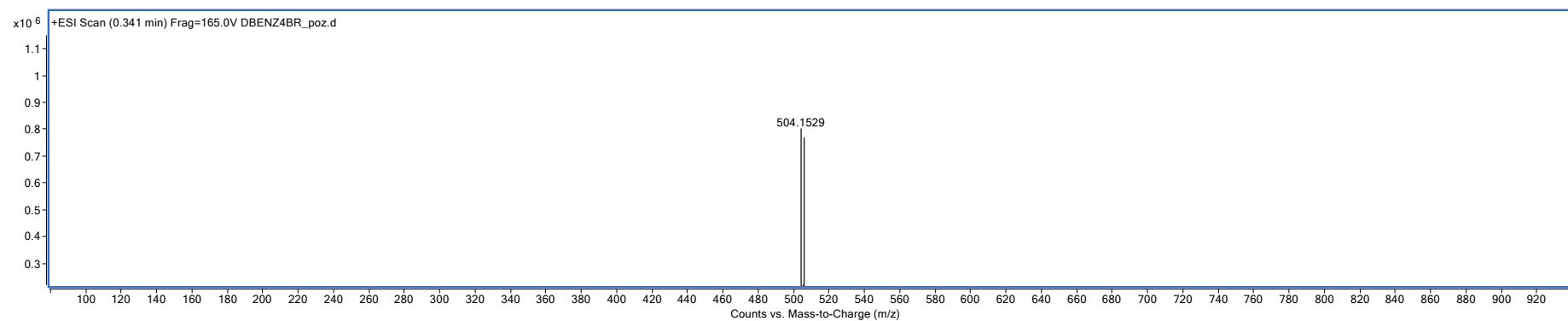
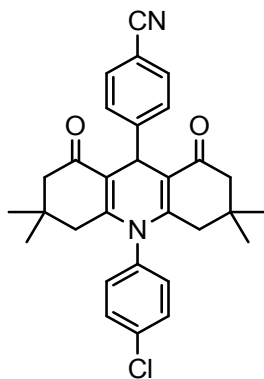
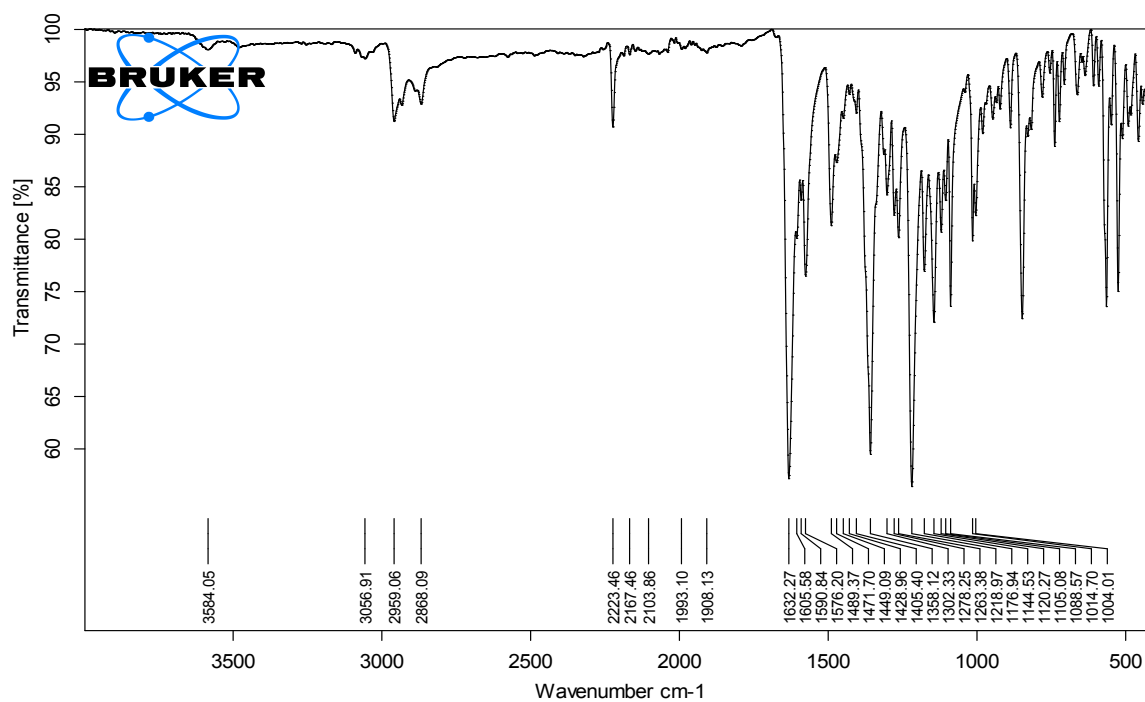


Fig. S13. a) FT-IR, **b)** ^1H NMR, **c)** ^{13}C NMR (APT), **d)** Q-TOF LC/HRMS of **4h**

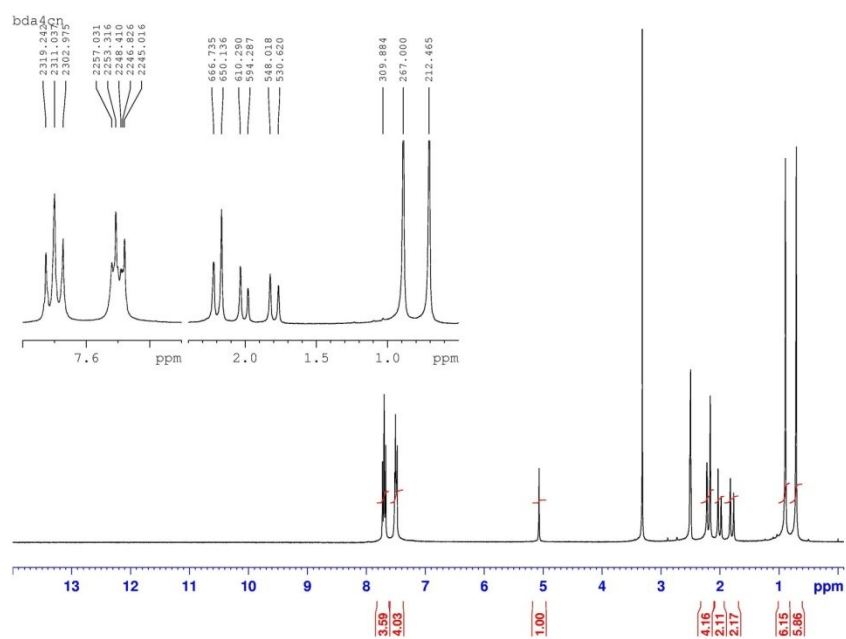
4-(10-(4-Chlorophenyl)-3,3,6,6-tetramethyl-1,8-dioxo-1,2,3,4,5,6,7,8,9,10-decahydro acridin-9-yl)benzonitrile (4i)



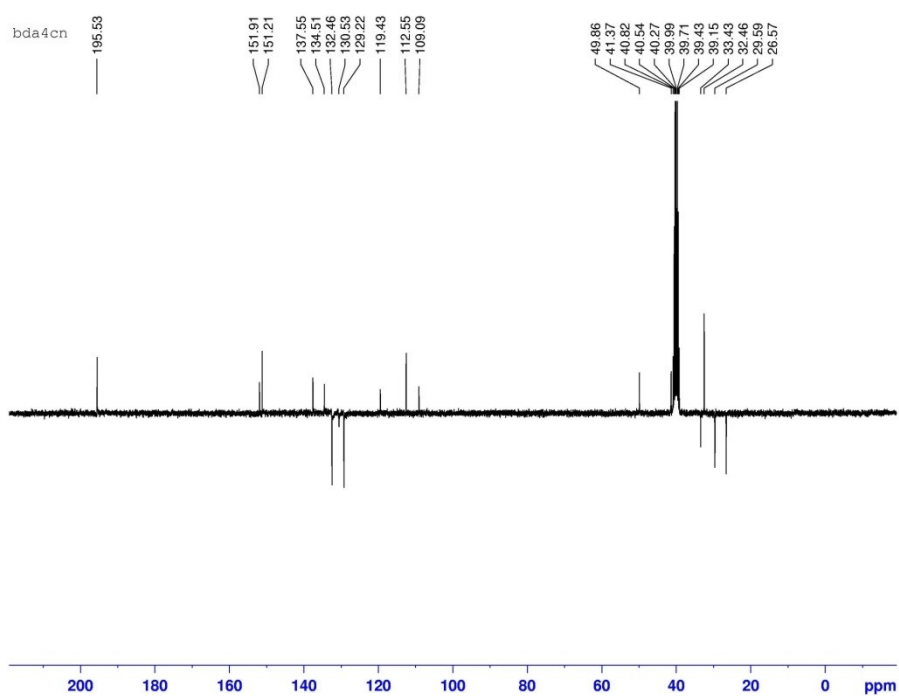
a)



b)



c)



d)

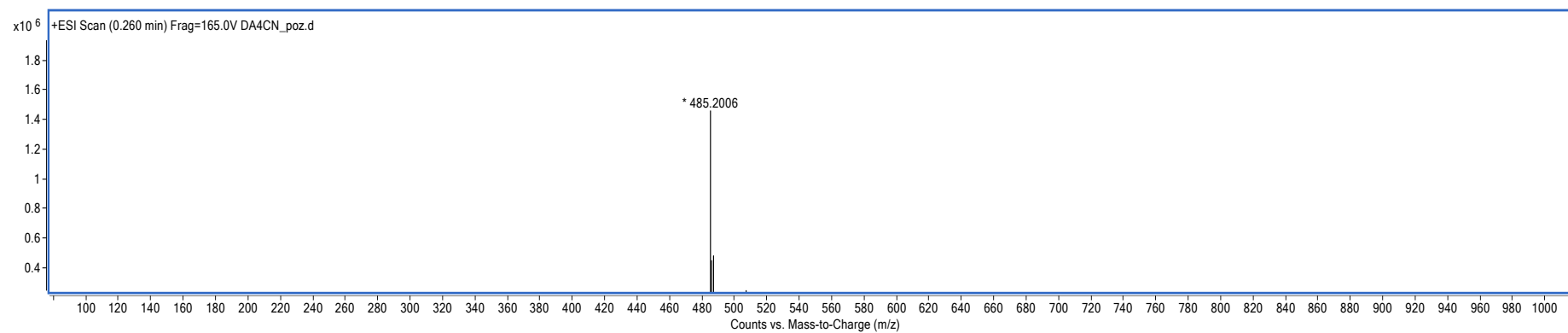
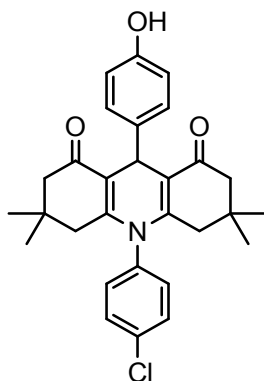
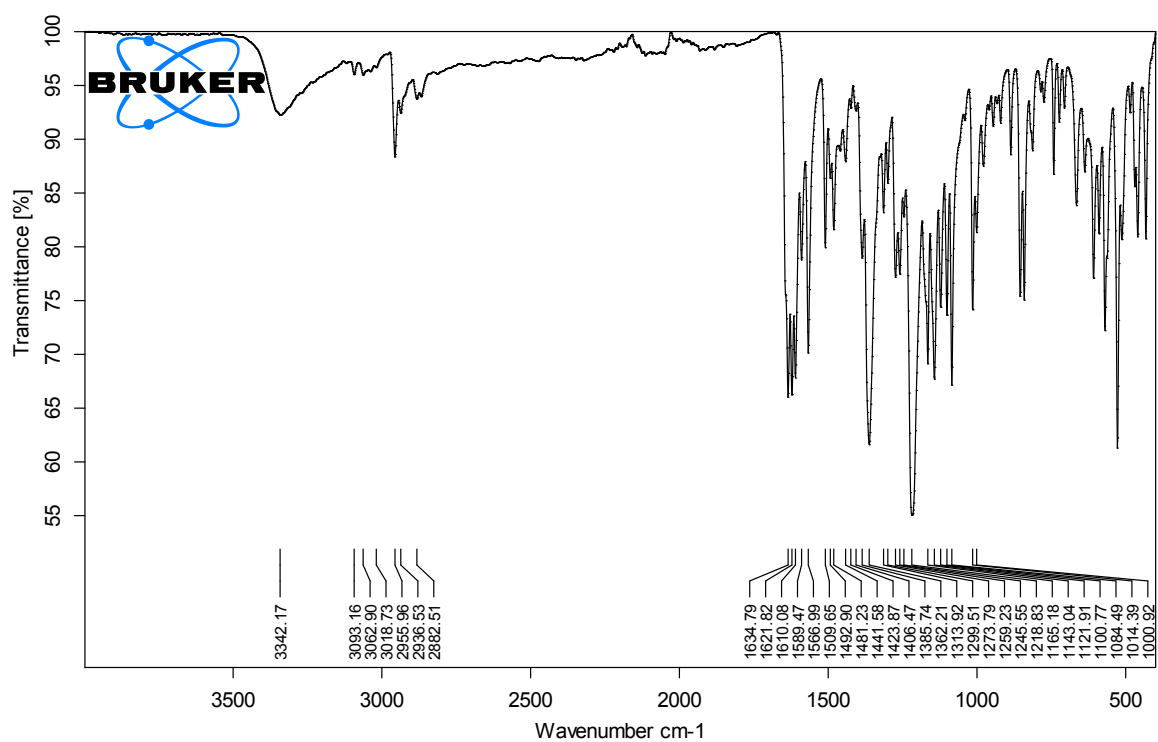


Fig. S14. a) FT-IR, b) ^1H NMR, c) ^{13}C NMR (APT), d) Q-TOF LC/HRMS of **4i**

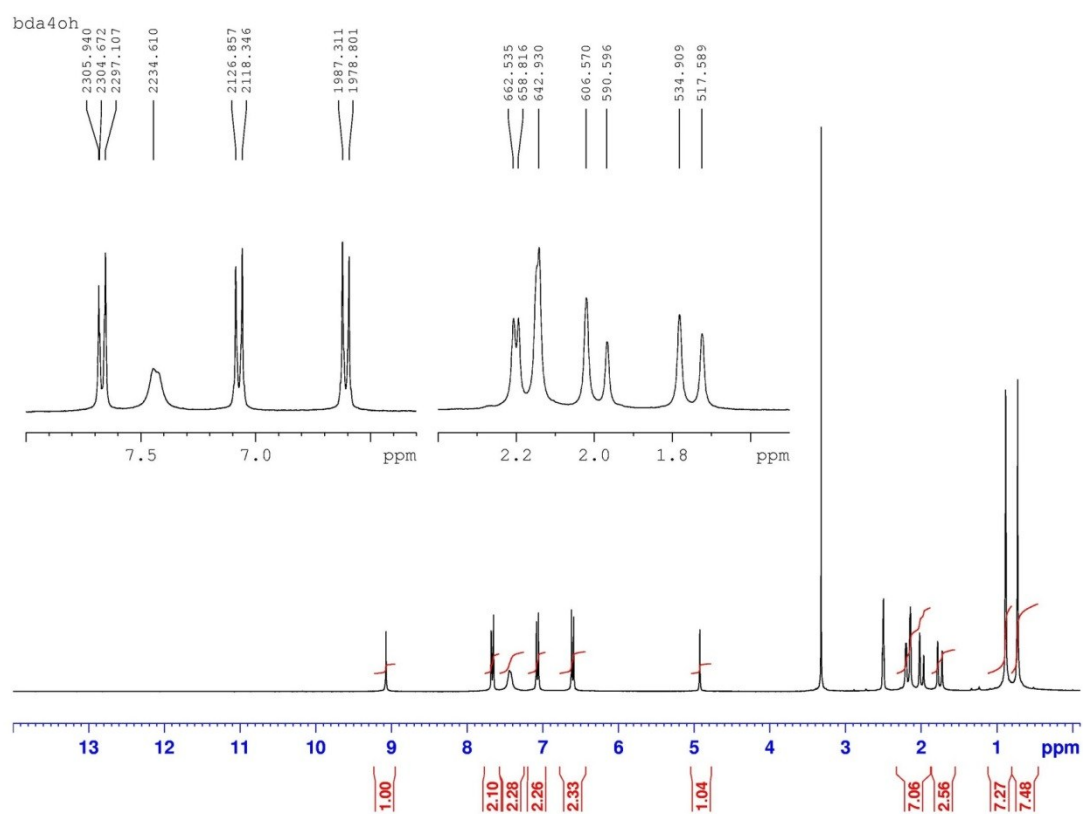
10-(4-Chlorophenyl)-9-(4-hydroxyphenyl)-3,3,6,6-tetramethyl-3,4,6,7,9,10-hexahydro acridine-1,8(2H,5H)-dione (4j)



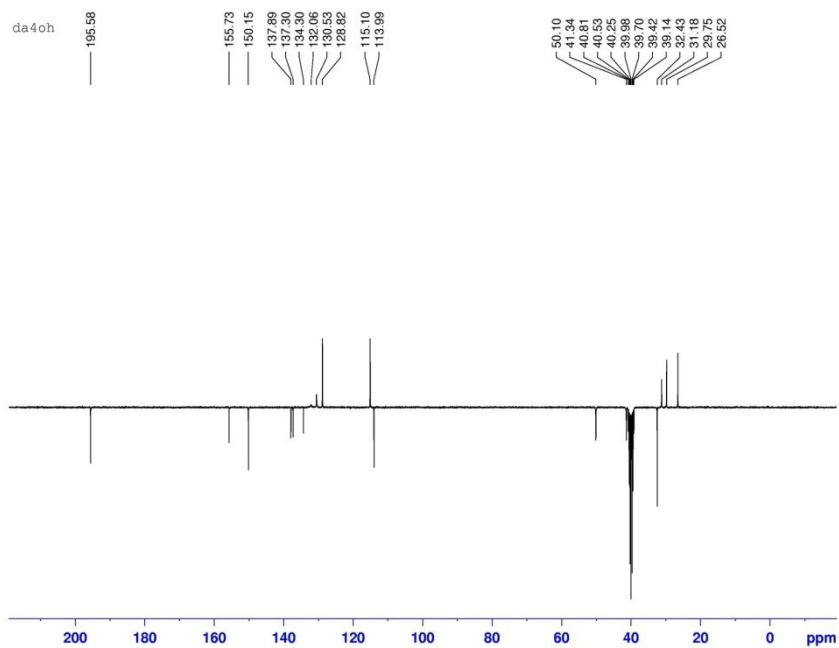
a)



b)



c)



d)

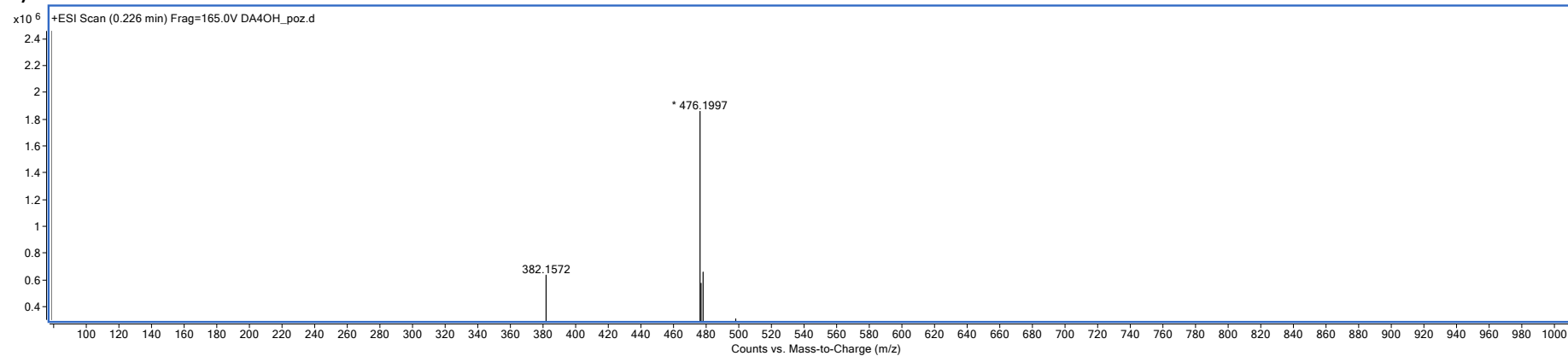


Fig. S15. a) FT-IR, b) ¹H NMR, c) ¹³C NMR (APT), d) Q-TOF LC/HRMS of 4j

