

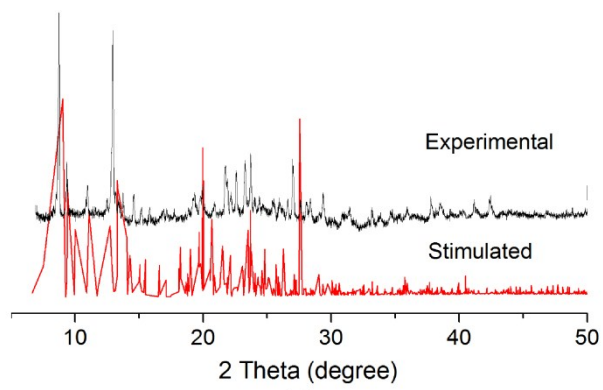
## Supporting information

# Influence of the Protonic State of an Imidazole-Phenanthroline Ligand on the Luminescence Properties of Copper(I) Complexes: Experimental and Theoretical Research

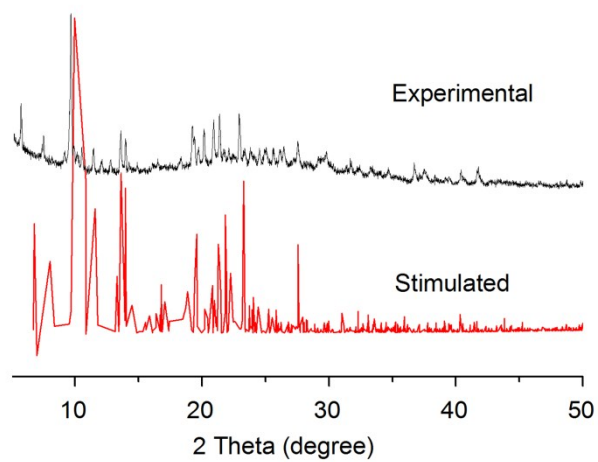
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**Table S1.** Selected Mulliken and NBO charges of **2a** and **2b**

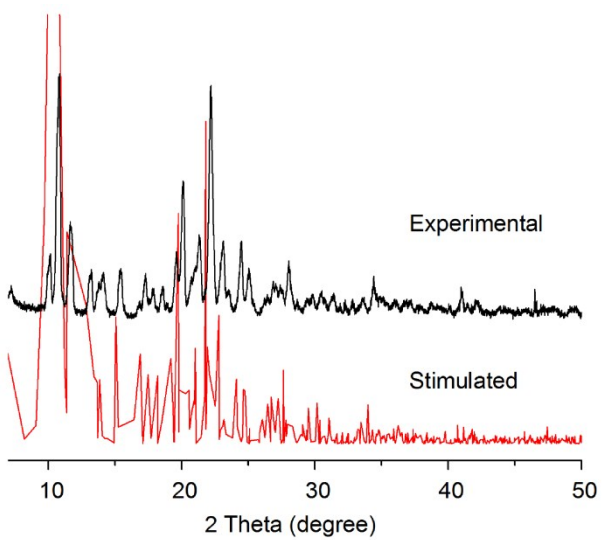
Atom	Complex <b>2a</b>		Complex <b>2b pph3</b>	
	Mulliken	NBO	Mulliken	NBO
Cu	-0.121	0.247	-0.182	0.350
P1	0.564	0.971	0.554	0.948
P2	0.541	0.972	0.537	0.949
N1	-0.482	-0.538	-0.477	-0.523
N2	-0.488	-0.538	-0.478	-0.516



(a)

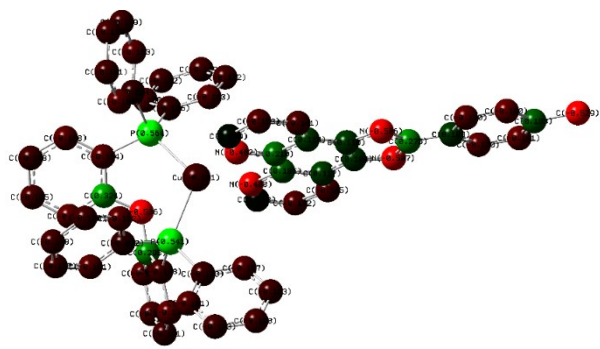


(b)

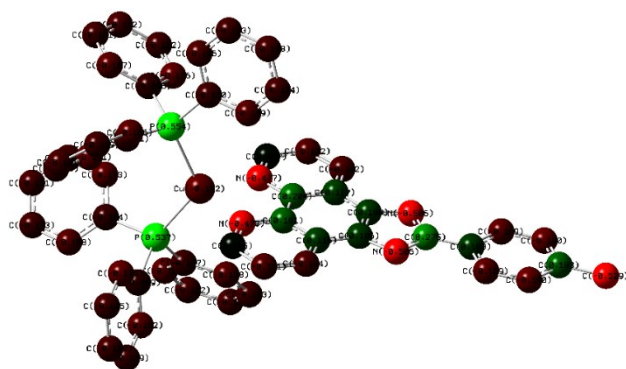


(c)

Figure S1. Comparing the X-ray powder diffraction patterns of microcrystalline powders and stimulated for **1b** (a); **2a** (b); **2b** (c).



2a



2b

Figure S2. Mulliken atom charges of complexes 2a and 2b

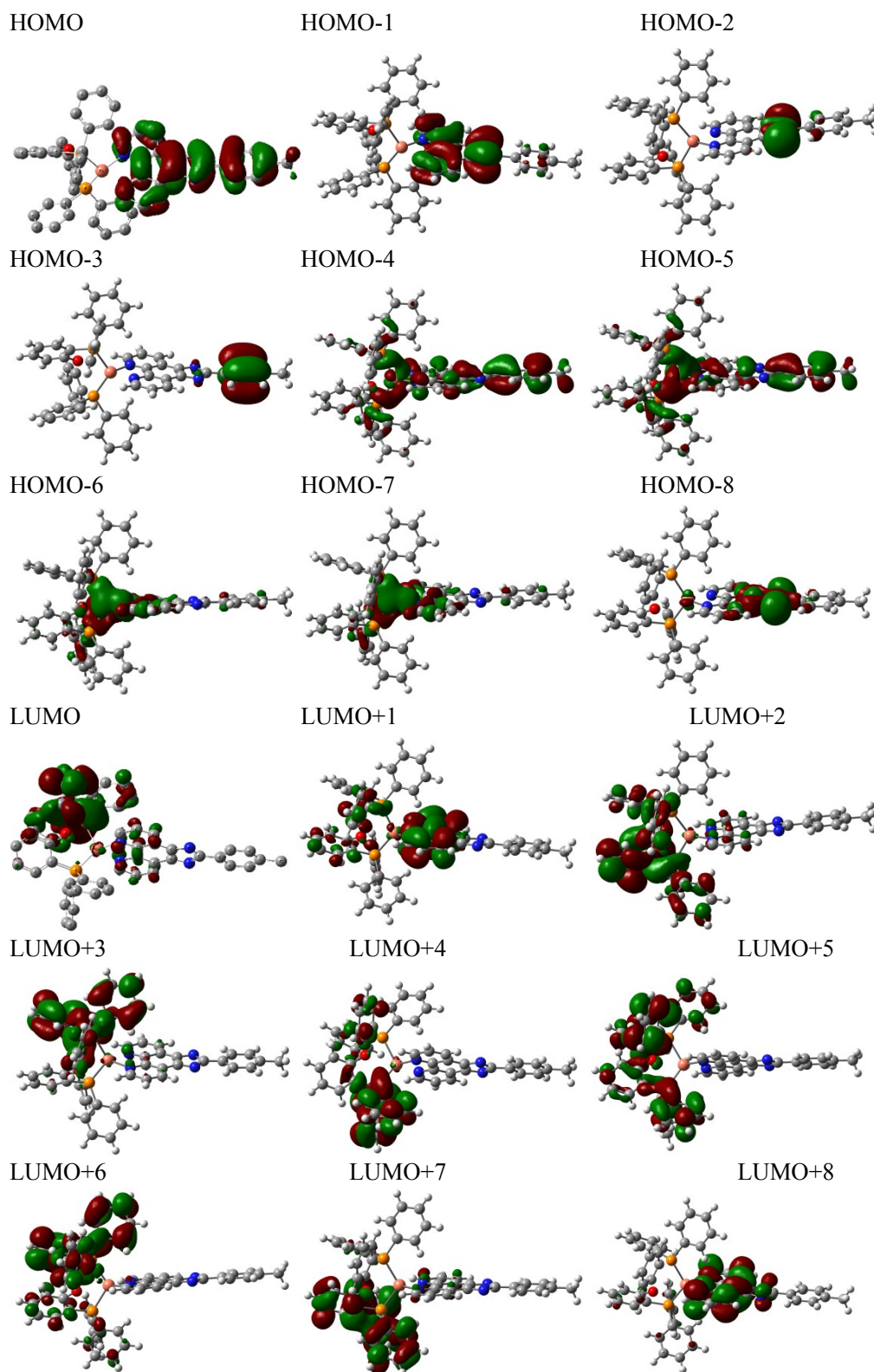


Figure S3. Frontier orbital of complex **2a**

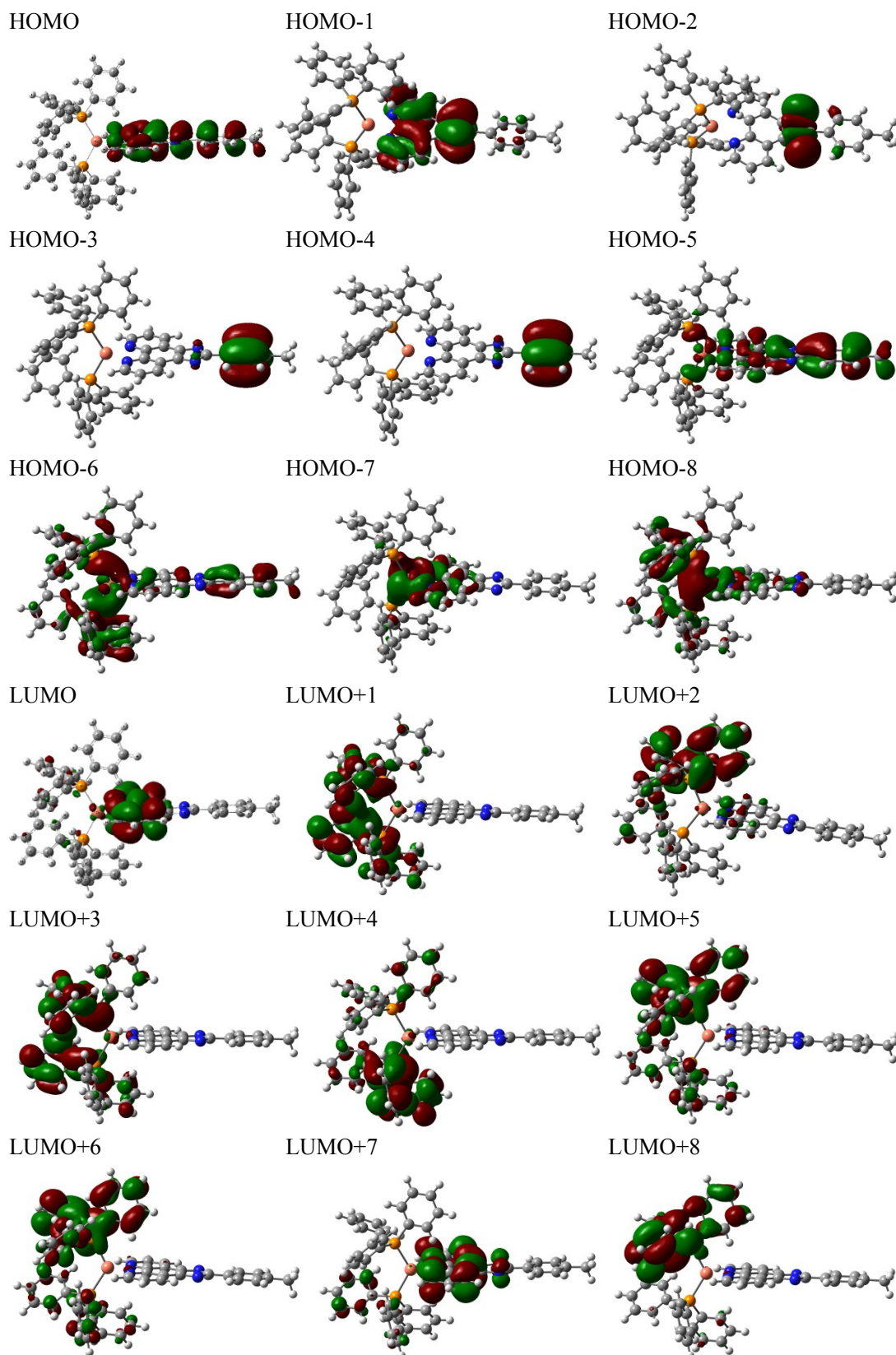


Figure S4. Frontier orbital of complex **2b**

Figure S5.  $^1\text{H}$  NMR of complex **1a**

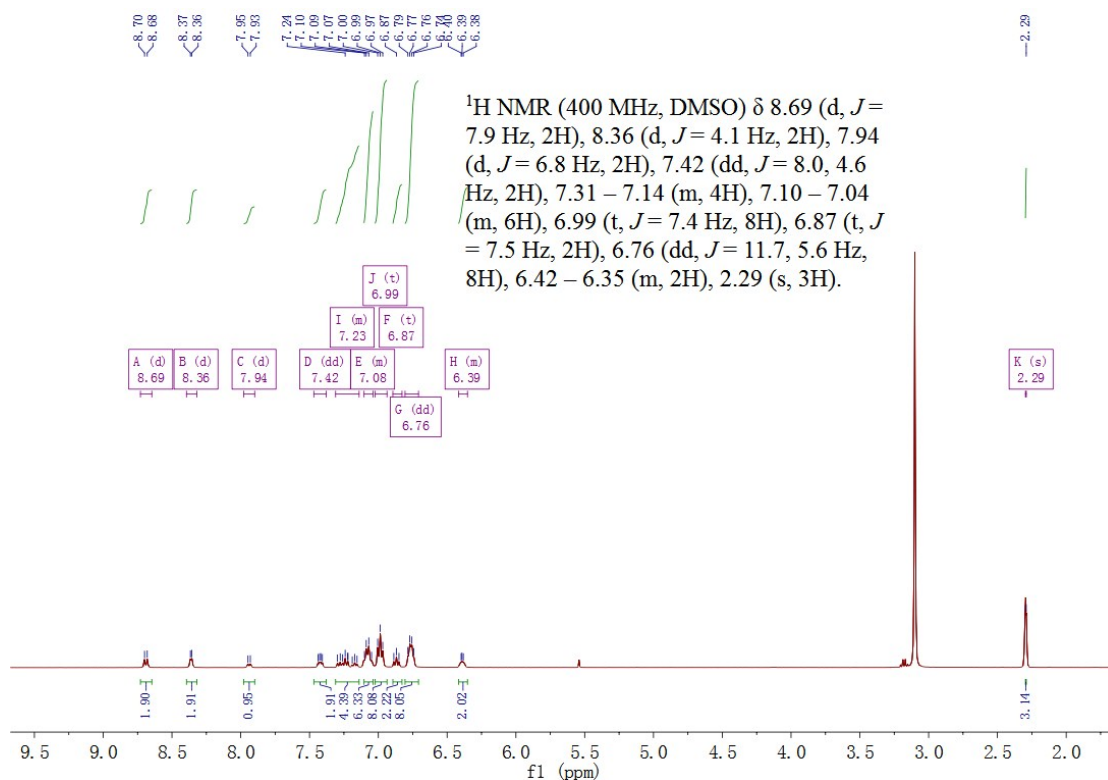


Figure S6.  $^1\text{H}$  NMR of complex **1b**

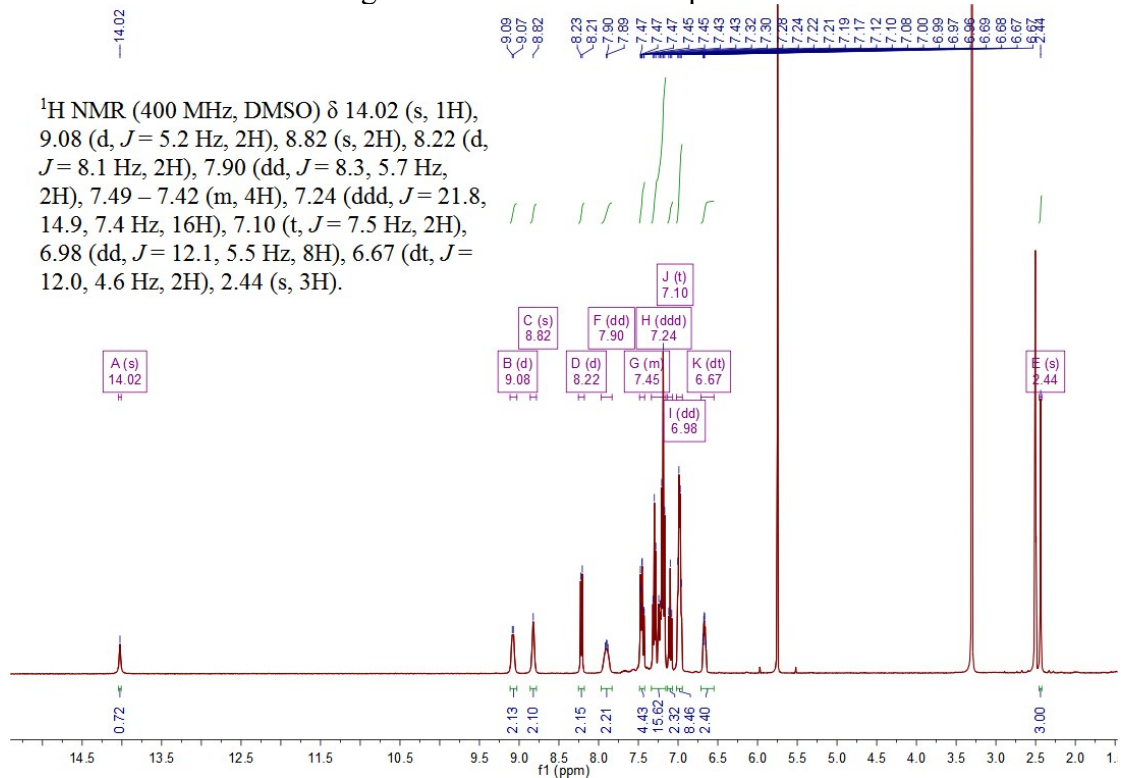


Figure S7.  $^1\text{H}$  NMR of complex **2a**

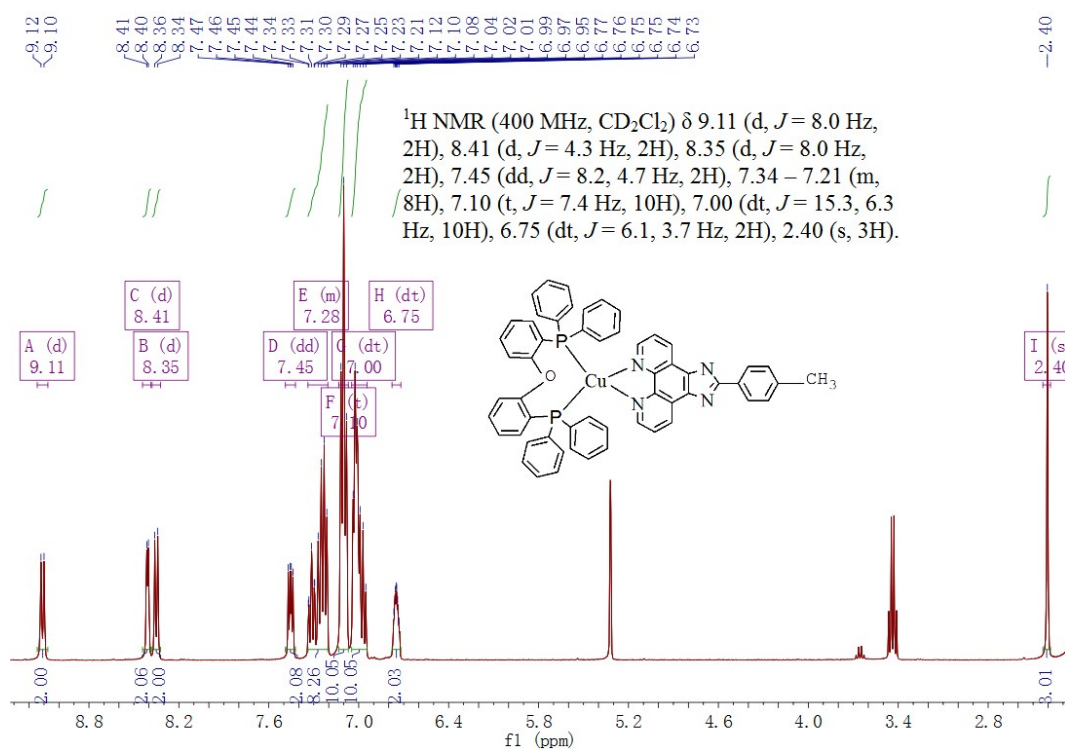


Figure S8.  $^1\text{H}$  NMR of complex **2b**

