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Supplementary Information

Hierarchically Porous Titanium Phosphate Monoliths and Its Crystallization Behavior in Ethylene Glycol

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Figure S1 FE-SEM images of TiP monoliths synthesized without (A) and with (B) the addition of DMSO.



Figure S2 Raman spectrum of TiP-0-0.



Figure S3 ³¹P NMR of TiP-0.05-0.08 calcined at 600 °C (black curve: original data, red curve: fitting peaks of the original data, blue curve: peak sum of fitting peaks).



Figure S4 N_2 adsorption desorption isotherms (A) and BJH pore size distributions (B) obtained from the adsorption branch of as-synthesized TiP-0.05-0.08 and those of TiP-0.05-0.08 calcined at different temperatures.



Figure S5 N_2 adsorption desorption isotherms (A) and BJH pore size distribution obtained from the adsorption branch (B) of TiP-0.05-0.08 after solvothermal treatment at different temperatures for 24 h.

Table S1 Pore parameters and compositional information of as-synthesized TiP-0.05-0.08, those of TiP-0.05-0.08 calcined at different temperatures and those of TiP-0.05-0.08 after solvothermal treatment at different temperatures for 24 h.

	S _{BET} /	$V_{ m p}$ /	$D_{\rm p}$ / nm	^a P:Ti	^a S:Ti
	m^2g^{-1}	cm^3g^{-1}		(molar)	(molar)
As-synthesized	320	0.86	21	2.1	0.53
600 °C	120	0.51	21	2.0	0.01
700 °C	104	0.46	24	2.0	0
800 °C	3	0.04	/	2.0	0
EG-100 °C	103	0.73	66	1.3	0
EG-200 °C	40	0.16	/	1.5	0

^a The molar ratios of P:Ti and S:Ti is derived from EDX measurement and is the average value of 5 measurements.