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Supporting Information

Weaving an infinite 3-D supramolecular network *via* Au^I...Au^{III} aurophility and C-H...Cl hydrogen bonding

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General information

All manipulations were performed under a nitrogen atmosphere using standard Schlenk or glovebox techniques. Acetonitrile and dichloromethane were distilled from CaH₂ before use. Diethyl ether was dried by distillation from sodium benzophenone prior to use. All reagents were obtained from commercial sources and used as received without further purification. ¹H NMR and ¹³C NMR spectra were obtained at room temperature using Bruker Avance 500 and 600 MHz spectrometers. IR data were obtained as KBr on a Bruker Alpha-T FTIR spectrometer. Mass spectra were acquired on a Finnigan TSQ 700 spectrometer. Elemental analyses were performed on a HERAEUS CHN-O-S-Rapid elemental analyzer by Instruments Center, National Chung Hsin University. Solid state emission spectra were recorded on a Variable Temperature Photoluminescence Spectrometer (HORIBA JOBIN YVON Fluorolog-3). Molar conductance was carried out on a JENCO_EC3840

X-ray Crystallographic Analysis of 1-5.

Crystal data collection and processing parameters are given in Table S1. Diffraction data were carried out on a Bruker APEX2 CCD diffractometer with graphite monochromated Mo-K α radiation ($\lambda = 0.7107$ Å). The structures 1-5 were solved by direct methods and refined by full-matrix least-squares procedures using the SHELXL-97 program. All of the non-hydrogen atoms were refined with anisotropic temperature factors. All hydrogen atoms were located geometrically and refined in the riding mode. Note that Checkcif /PLATON report for 5 suggests possible pseudo/new *P*nnm space group. We thus try to transform from *P*2₁/n to *P*nnm and to solve from beginning but the structure cannot be refined by using the suggested space group. Additional crystallographic data as CIF files are available as Supporting Information.

	1	2	3	4	5
Empirical Formula	C ₆ H ₁₀ N ₂ AuCl	$C_{12}H_{20}N_5AuO_3$	$C_{12}H_{20}N_4AuBF_4\\$	C ₁₂ H ₂₀ N ₄ AuSbF ₆	$C_{12}H_{20}N_4Au_2Cl_4$
Formula weight	342.58	479.30	504.10	653.04	756.05
<i>T</i> (K)	200(2)	200(2)	200(2)	200(2)	200(2)
λ (Å)	0.71073	0.71073	0.71073	0.71073	0.71073
Crystal system	Orthorhombic	Monoclinic	Triclinic	Triclinic	Monoclinic
Space group	Pbca	$P2_1/m$	Pī	$P\overline{\iota}$	$P2_1/n$
a (Å)	7.3641(7)	8.4029(5)	8.5936(14)	7.450(2)	6.6824(4)
<i>b</i> (Å)	13.8244(19)	20.7542(11)	13.854(2)	7.938(2)	8.2695(5)
<i>c</i> (Å)	16.9502(19)	13.5645(8)	14.342(2)	8.481(3)	17.5884(10)
α (°)	90	90	77.941(2)	89.041(7)	90
β (°)	90	99.439(2)	82.224(3)	84.027(7)	90.010(2)
γ(°)	90	90	77.872(3)	68.263(5)	90
$V(Å^3)$	1725.6(3)	2333.6(2)	1625.0(5)	463.2(2)	971.94(10) Å
Ζ	8	6	4	1	2
Dcalc (Mg m ⁻³)	2.637	2.046	2.060	2.341	2.583
μ (mm ⁻¹)	17.289	9.474	9.093	9.425	15.627
<i>F</i> (000)	1248	1380	960	304	692
θ range for data collection	2.95-25.00°	2.65-25.05°	1.46-25.00°	2.42-25.10°	2.72-25.02°
Reflections collected	8170	15858	13731	3791	6807
Independent reflections (R_{int})	1510 (0.0585)	4207 (0.0457)	5656 (0.0295)	1621(0.0532)	1709 (0.0349)
Data / restraints / parameters	1510 / 0 / 91	4207 / 0 / 325	5656 / 0 / 403	1621 / 0 / 112	1709 / 0 / 103
Goodness-of-fit on F^2	1.021	1.117	1.082	1.158	1.037
R_1, wR_2	0.0357, 0.0629	0.0323, 0.0767	0.0314, 0.0768	0.0531, 0.1368	0.0288, 0.0517
CCDC No.	1008362	1427451	1427452	1427453	1427454

Table S1 X-ray crystallographic data for complexes 1 - 5.

1. Synthesis of [Au(L)Cl] (1)

A 4 ml CH₂Cl₂ solution of AuCl (0.146 g, 0.628 mmol) and 1,3,5-trimethyl-1Hpyrazole (0.069 g, 0.628 mmol) was stirred at room temperature for 1 h under N₂. After the addition of 10 ml hexane, the white precipitates formed was filtered and washed by hexane for two times. The resulting white solid was dried under vacuum to yield 0.201 g (93.1 %). Single crystals of **1** were grown from layer CH₂Cl₂/hexane at room temperature. ¹H NMR (500 MHz, CD₂Cl₂, 298 K): δ 5.99 (s, H, CH), δ 3.79 (s, 3H, CH₃), 2.25 (s, 3H, CH₃), 2.23 (s, 3H, CH₃) ppm. ¹³C NMR (150.91 MHz, CD₂Cl₂, 300 K): δ 150.42, 143.40, 107.05, 37.51, 14.76, 12.49 ppm. Anal. Calc. for C₆H₁₀N₂AuCl (**1**): C, 21.04; H, 2.99; N, 8.18. Found: C, 20.88; H, 2.80; N, 8.16. Molar conductance, $A_{\rm M}$ (CH₂Cl₂, 298 K) = 6.3 Ω^{-1} cm²mol⁻¹.

2. Synthesis of [Au(L)₂](NO₃) (2)

A 10 ml CH₂Cl₂ solution of **1** (0.109 g, 0.318 mmol) and AgNO₃ (0.054 g, 0.318 mmol), was stirred for 1h under N₂. The white precipitates formed was filtered off and the resulting solution was dried under vacuum to yield 0.120 g (76.9 %). Single crystals of **2** were grown from layer CH₂Cl₂/hexane at room temperature. Elemental analysis: Anal. Calc. for C₁₂H₂₀AuO₃N₅: C, 30.07; H, 4.21; N, 14.61. Found: C, 30.36; H, 4.32; N, 14.88. Positive ESI/MS: 417.14 ([M–NO₃]⁺, 100%). ¹H NMR (600MHz, 300K, CD₂Cl₂): δ 6.12 (s, 2H, 2CH), 3.95 (s, 6H, 2CH₃), 2.39 (s, 6H, 2CH₃), 2.36 (s, 6H, 2CH₃) ppm. ¹³C NMR (150.91MHz, 300K, CD₂Cl₂): δ 151.19, 145.26, 107.71, 14.73, 12.53 ppm. $\Lambda_{\rm M}$ (CH₂Cl₂, 298 K) = 60.9 Ω^{-1} cm²mol⁻¹.

3. Synthesis of $[Au(L)_2](BF_4)$ (3)

A 10 ml CH₂Cl₂ solution of **1** (0.078 g, 0.228 mmol) and AgBF₄ (0.044 g, 0.228 mmol), was stirred 1h under N₂. The white precipitates formed was filtered off and the resulting solution was dried under vacuum to yield 0.101 g (87.7 %). Single crystals of **3** were grown from layer CH₂Cl₂/hexane at room temperature. Positive ESI/MS: 417.22 ([M–BF₄]⁺, 100 %). Elemental analysis: Anal. Calc. for C₁₂H₂₀N₄AuBF₄: C, 28.59; H, 4.00; N: 11.11. Found: C, 28.78; H, 3.86; N, 11.32. ¹H NMR (600MHz, 300K, CD₂Cl₂): δ 6.12 (s, 2H, 2CH), 3.92 (s, 6H, 2CH₃), 2.40 (s, 6H, 2CH₃), 2.36 (s, 6H, 2CH₃) ppm. ¹³C NMR (150.91MHz, 300K, CD₂Cl₂): δ 151.22, 145.31, 107.73, 14.71, 12.49 ppm. $\Lambda_{\rm M}$ (CH₂Cl₂, 298 K) = 69.4 Ω^{-1} cm²mol⁻¹

4. Synthesis of [Au(L)₂](SbF₆) (4)

A 10ml CH_2Cl_2 solution of **1** (0.203 g, 0.593 mmol) and $AgSbF_6$ (0.125 g, 0.593 mmol mol) was stirred for 0.5 h under N₂. The white precipitates formed was filtered off and resulting solution was dried under vacuum to give light yellow solid 0.382 g

(98.4 %). Single crystals of **4** were grown from layer CH₂Cl₂/hexane at room temperature. ¹H NMR (600MHz, 300K, CD₂Cl₂): δ 6.13 (s, 2H, 2C*H*), δ 3.91 (s, 6H, 2C*H*₃), 2.40 (s, 6H, 2C*H*₃), 2.36 (s, 6H, 2C*H*₃) ppm. ¹³C NMR (150.92MHz, 300K, CD₂Cl₂): δ 151.35, 145.32, 107.8, 37.73, 14.74, 12.53 ppm. Positive ESI/MS: 417.4 ([M–SbF₆]⁺, 100 %). Elemental analysis: Anal. Calc. for C₁₂H₂₀N₄AuSbF₆: C, 22.07; H, 3.09; N, 8.58. Found: C, 22.06; H, 2.67; N, 8.60. $\Lambda_{\rm M}$ (CH₂Cl₂, 298 K) = 67.7 Ω⁻¹cm²mol⁻¹

5. Synthesis of [Au(L)₂](AuCl₄) (5)

A 5 ml CH₂Cl₂ solution of **1** (0.199 g, 0.580 mmol) and Na[AuCl₄]•2H₂O (0.231 g, 0.580 mmol), was stirred for 2 h under N₂. The precipitates formed was filtered off and the filtrate was added 10 ml hexane to give yellow precipitates. The yellow solid was filtered and dried under vacuum to yield 0.381 g (87.0 %). Single crystals of **5** were grown from layer CH₂Cl₂/hexane at room temperature. ¹H NMR (600MHz, 300K, CD₂Cl₂): δ 6.01 (s, 2H, 2CH), δ 3.86 (s, 6H, 2CH₃), 2.34 (s, 6H, 2CH₃), 2.27 (s, 6H, 2CH₃) ppm. ¹³C NMR (150.92MHz, 300K, CD₂Cl₂): δ 150.42, 143.40, 107.05, 37.51, 14.76, 12.50 ppm. Positive ESI-MS: 417.14 ([M–AuCl₄]⁺, 100 %). Elemental analysis: Anal. Calc. for C₁₂H₂₀N₄Au₂Cl₄: C, 19.06; H, 2.67; N, 7.41. Found: C, 18.95; H, 2.60; N, 7.59. $\Lambda_{\rm M}$ (CH₂Cl₂, 298 K) = 69.0 Ω^{-1} cm²mol⁻¹



6. Figure S1 Variable-temperature ¹H NMR spectra of 1



7. Figure S2 1 H & 13 C NMR spectra of 2



8. Figure S3 1 H & 13 C NMR spectra of 3



9. Figure S4 1 H & 13 C NMR spectra of 4



10. Figure S5 1 H & 13 C NMR spectra of **5**



11. Figure S6 Molecular structure of 2



12. Figure S7 Molecular structure of **3**