

Electronic Supplementary Information

Fluorescent chiral fluorinated liquid crystalline polymers containing
rare earth complexes

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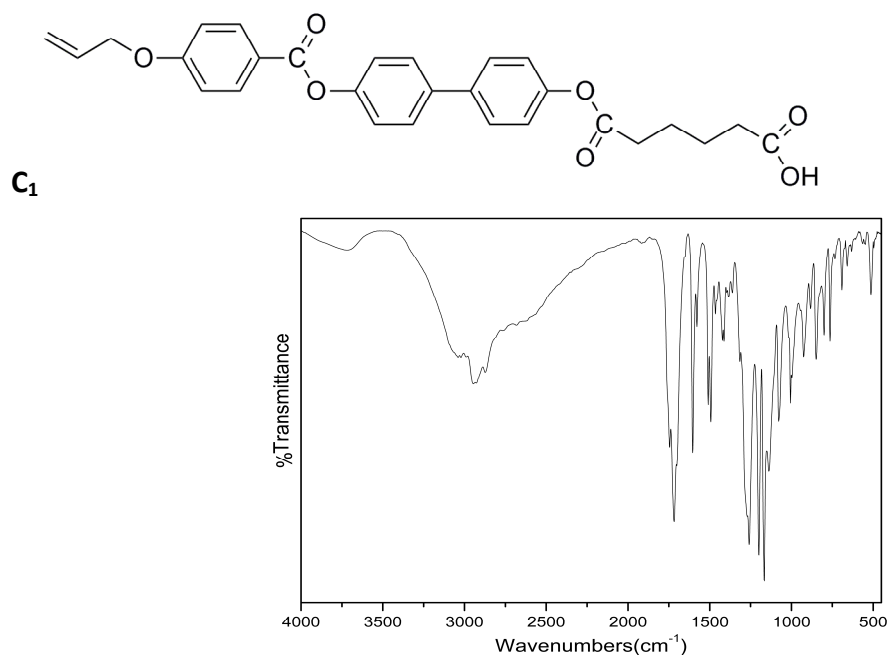


Figure S1. FT-IR spectrogram of **C₁**

Yield: 73%. IR (KBr, cm⁻¹): 3250–2500 (–OH), 2928, 2853 (–CH₃, –CH₂–), 1749, 1718, 1696 (C=O), 1650 (C=C), 1604, 1578 (Ar–)

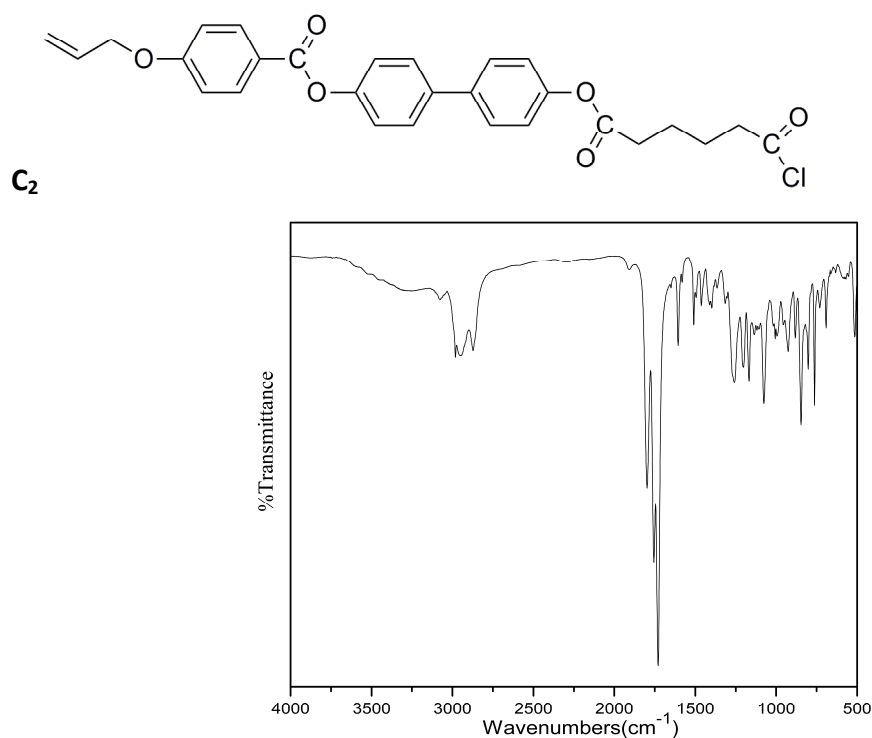


Figure S2. FT-IR spectrogram of **C₂**

Yield: 86%. IR (KBr, cm⁻¹): 2925, 2855 (–CH₃, –CH₂–), 1797, 1753, 1724 (C=O), 1650 (C=C), 1605, 1578 (Ar–)

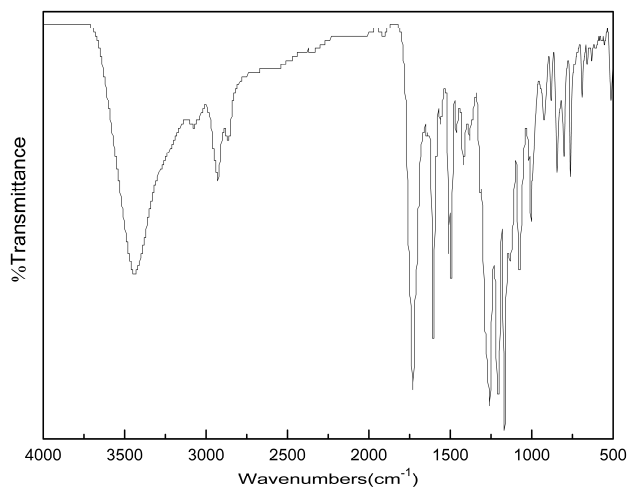
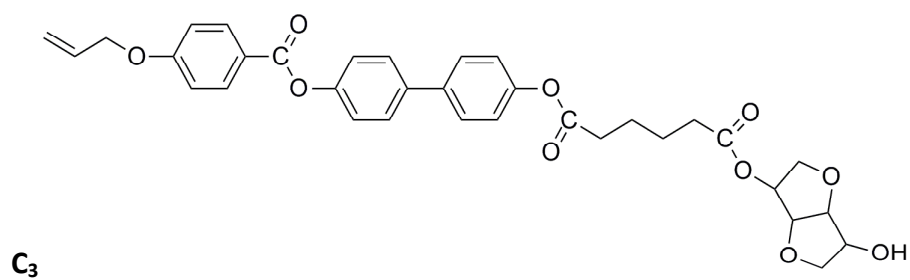


Figure S3. FT-IR spectrogram of C₃

Yield: 67%. IR (KBr, cm⁻¹): 3450 (-OH), 2937, 2865 (-CH₃, -CH₂-), 1755, 1741 1725 (C=O), 1650 (C=C), 1605, 1580 (Ar-)

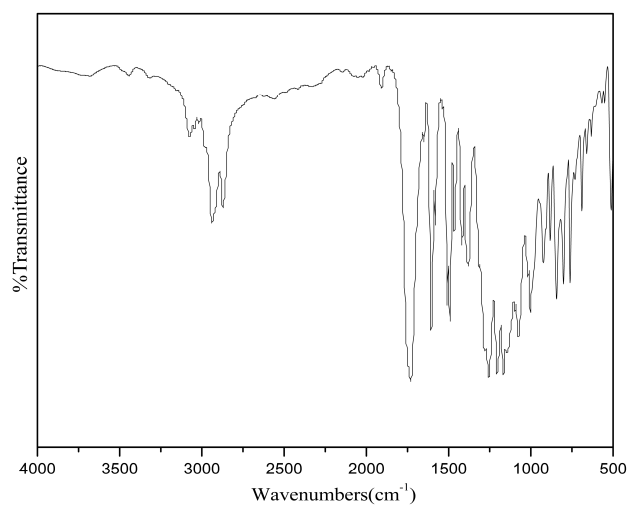
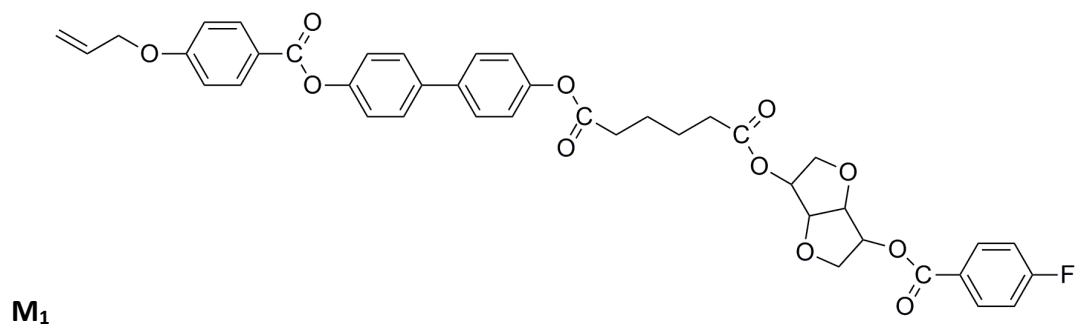


Figure S4. FT-IR spectrogram of M₁

ABA

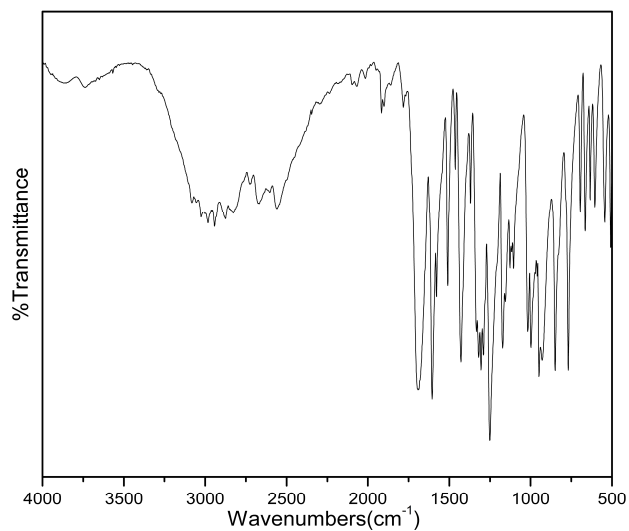
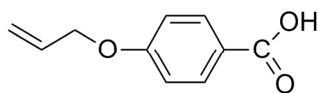


Figure S5. FT-IR spectrum of ABA

Tb-M₂

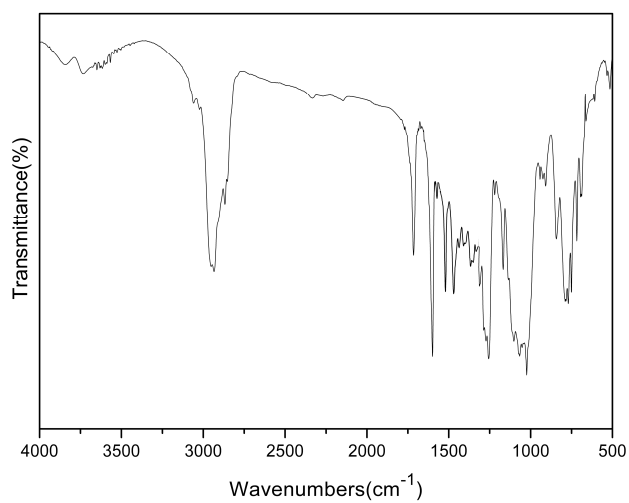
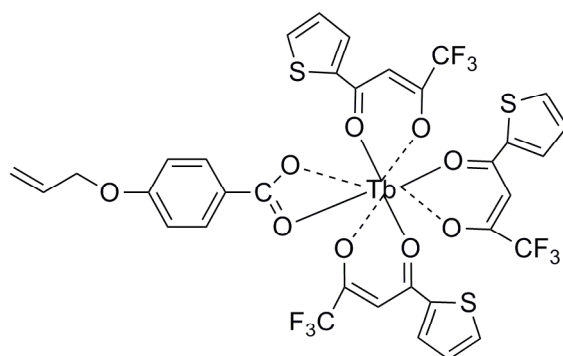


Figure S6. FT-IR spectrum of Tb-M₂

Sm-M₂

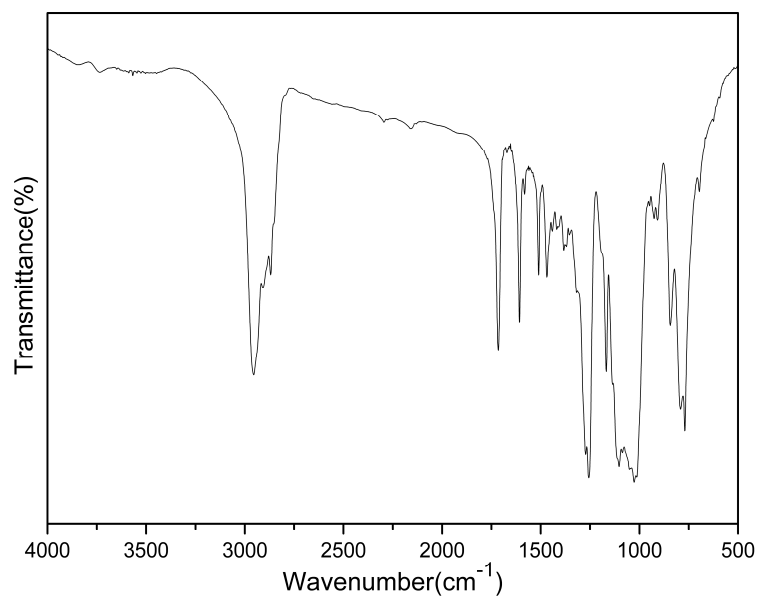
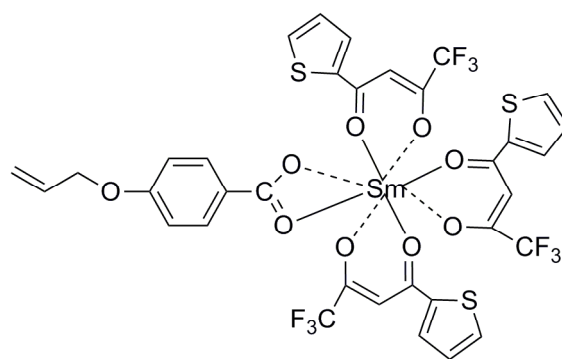


Figure S7. FT-IR spectrum of Sm-M₂

RE-LCPs

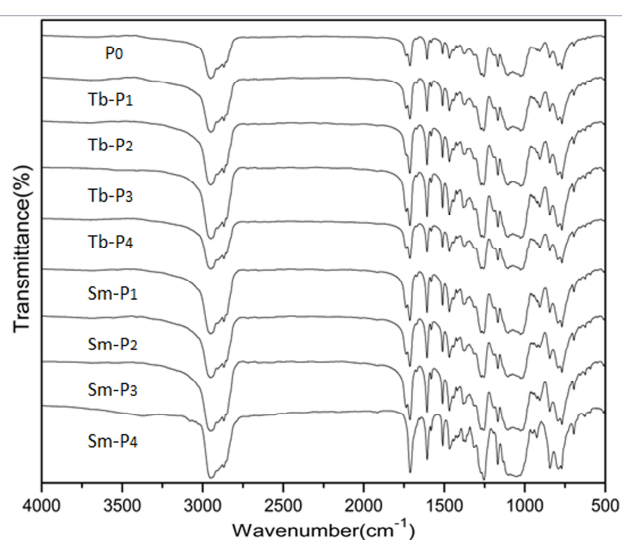


Figure S8. FT-IR spectrogram of RE-LCPs

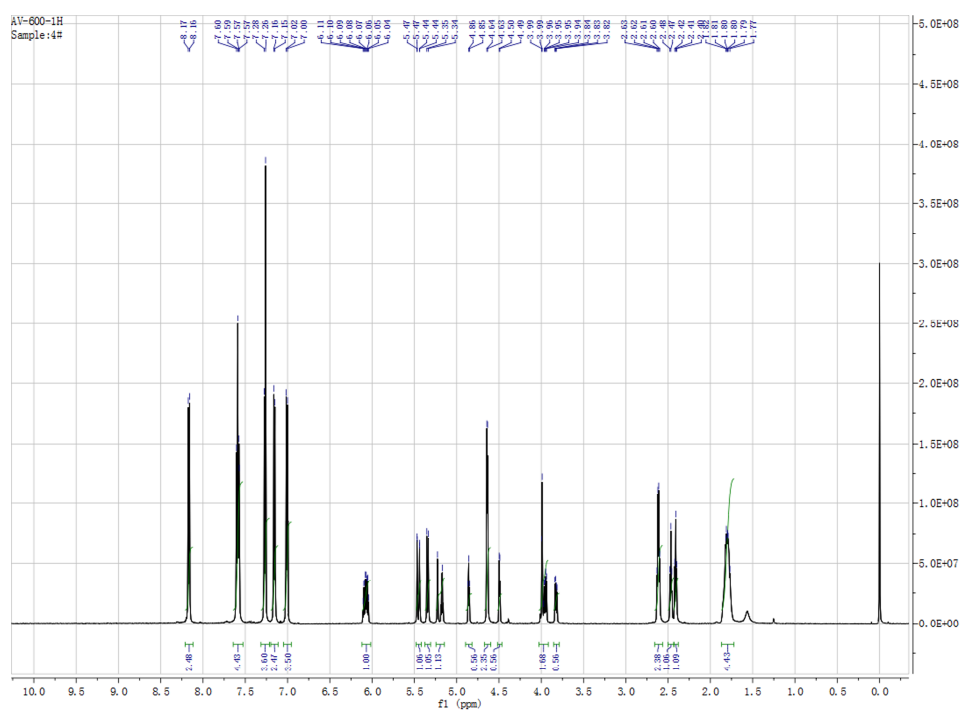


Figure S9. ^1H NMR spectrum of M_1 (600 MHz, CDCl_3).

Table 1 Specific rotations of M_1 and polymers

Sample	Concentration (g/100ml)	Solvent	Reference Fluid	Specific Rotations
M_1	0.1	THF	THF	+19.06
P_0	0.1	THF	THF	+19.42
Tb-P_1	0.1	THF	THF	+18.73
Tb-P_2	0.1	THF	THF	+18.26
Tb-P_3	0.1	THF	THF	+17.94
Tb-P_4	0.1	THF	THF	+17.58
Sm-P_1	0.1	THF	THF	+18.85
Sm-P_2	0.1	THF	THF	+18.32
Sm-P_3	0.1	THF	THF	+17.84
Sm-P_4	0.1	THF	THF	+17.47