

**Synthesis and structural chemistry of bicyclic hexaaza-dithia macrocycles
containing pendant donor groups**

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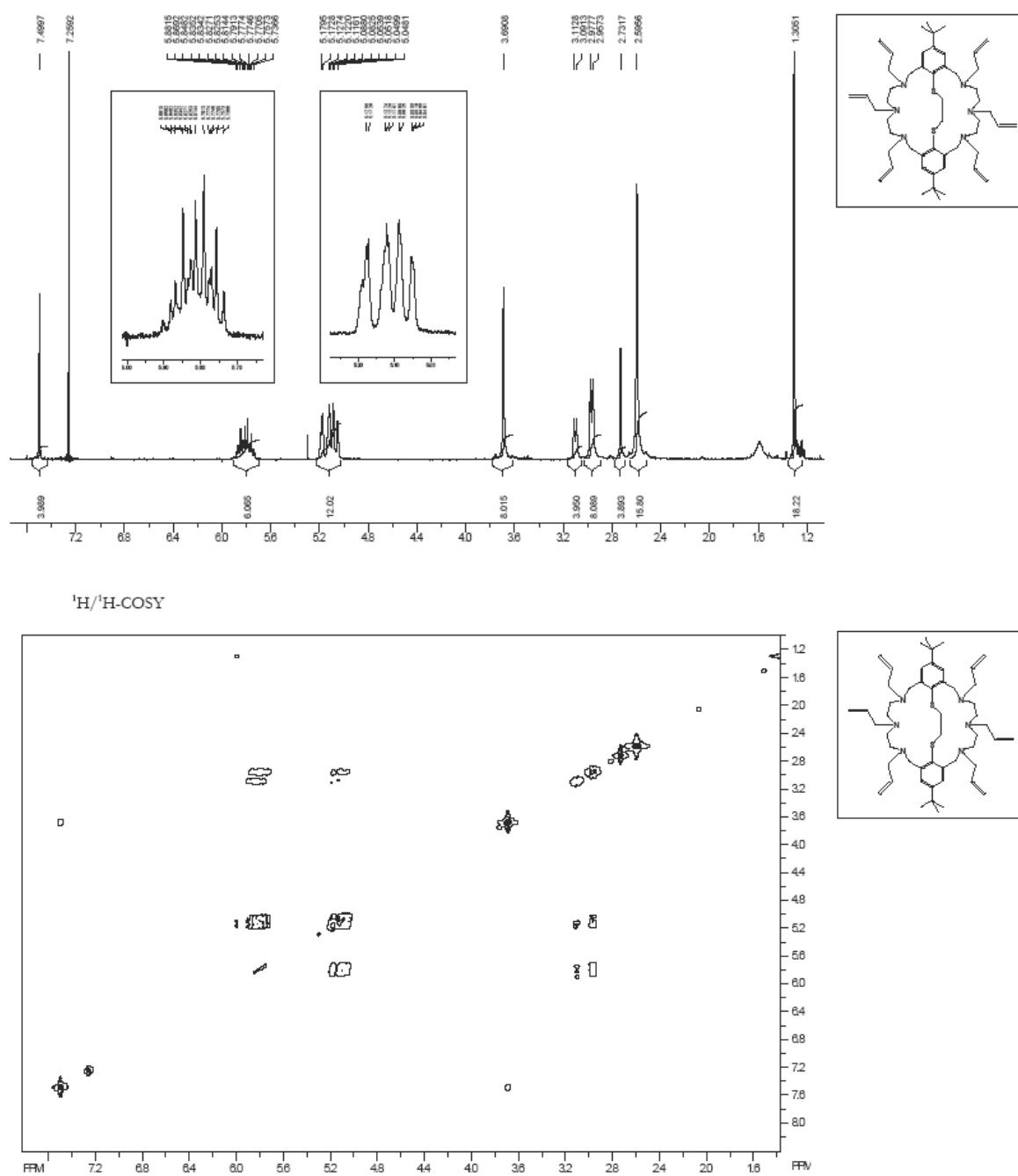
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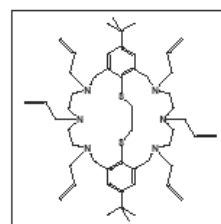
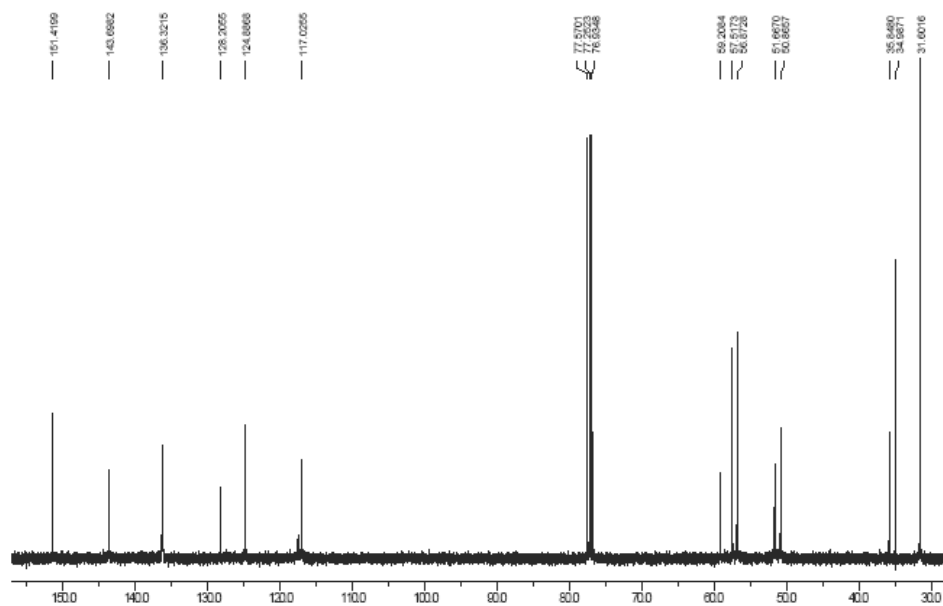
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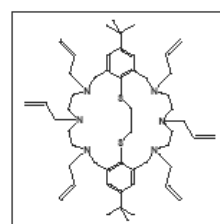
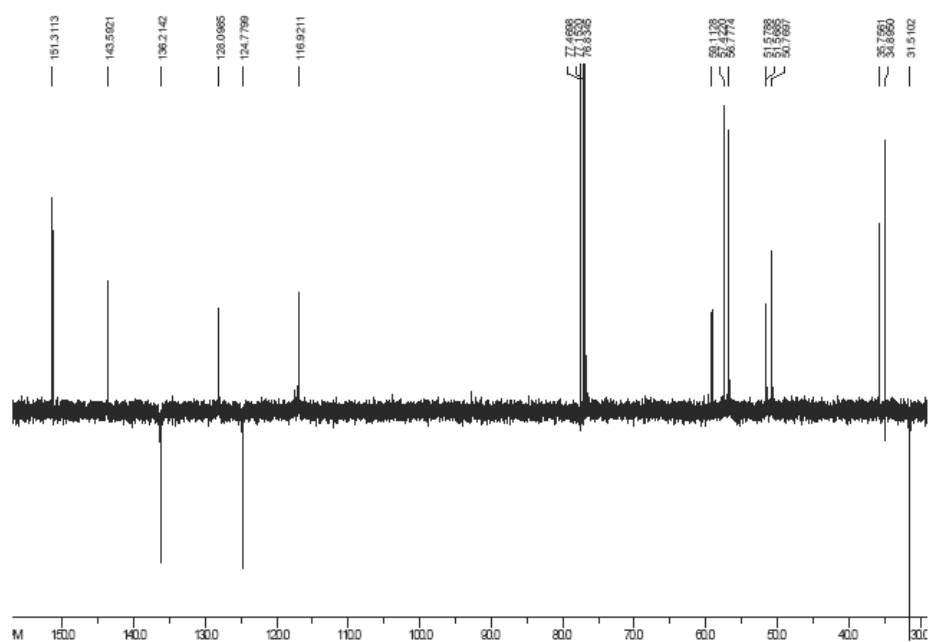
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1. Spectra for hexaallylated aza-thioether **8**.

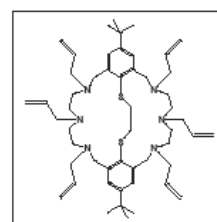
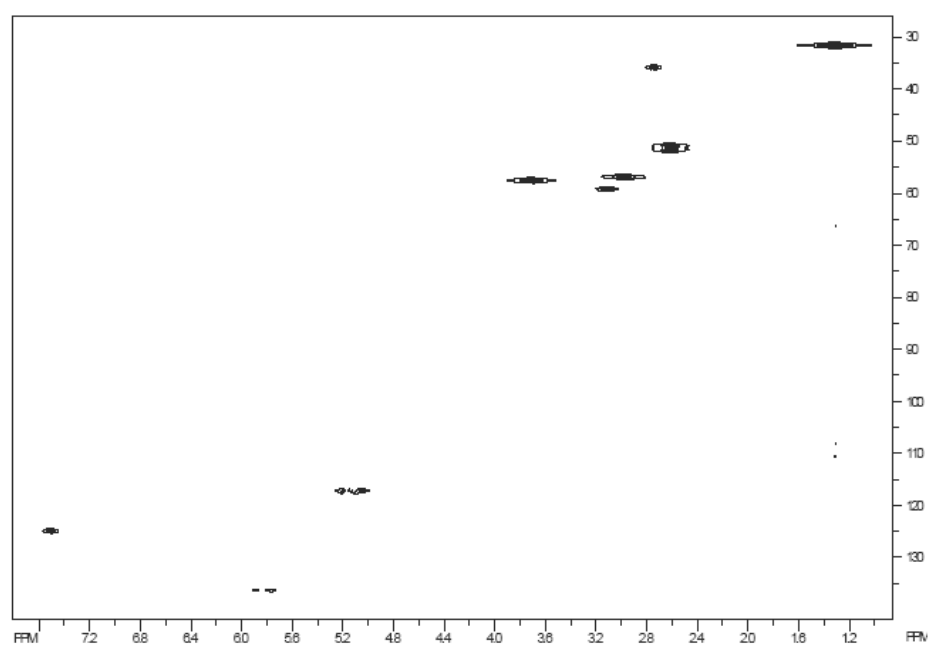




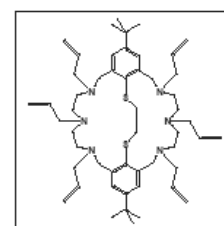
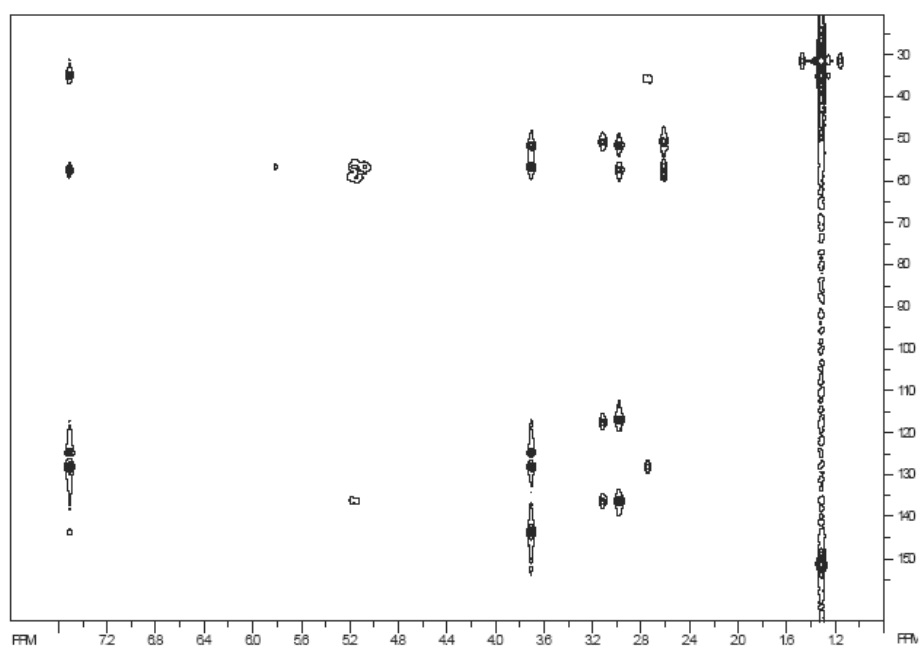
APT



HSQC

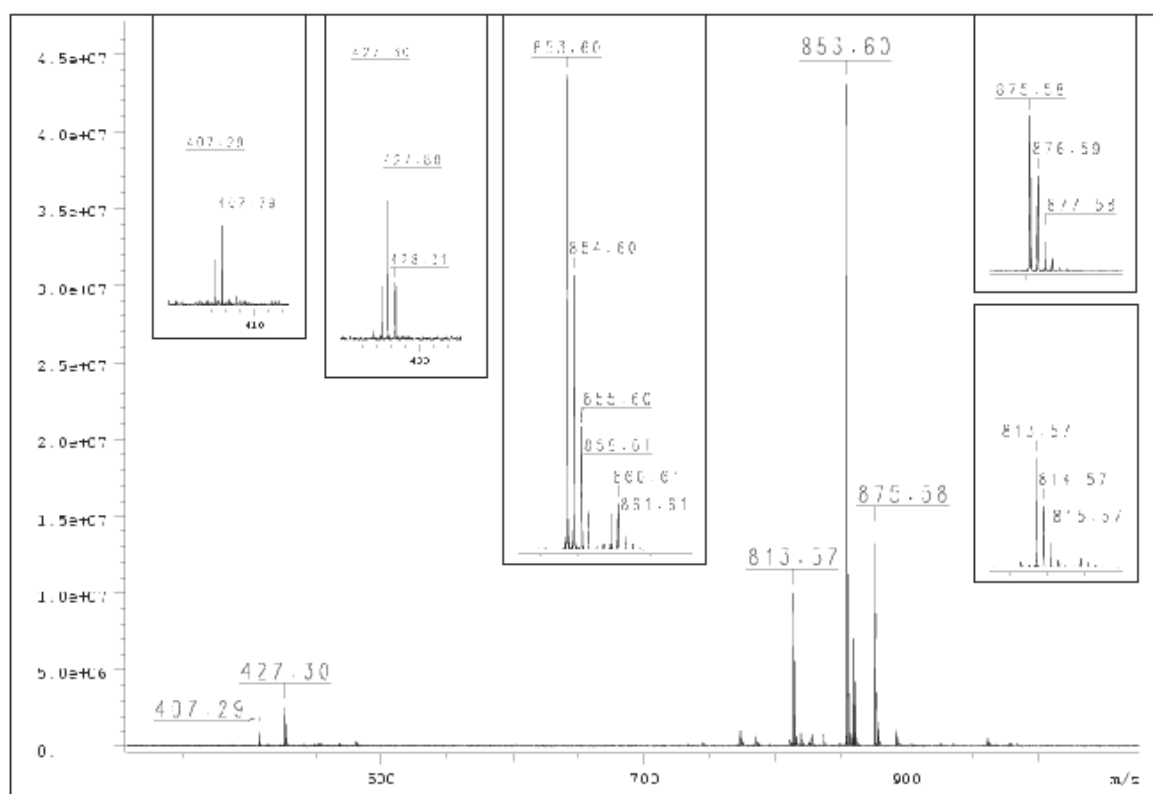
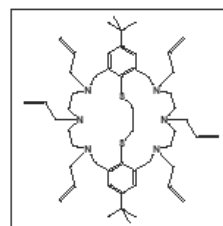


HMBC

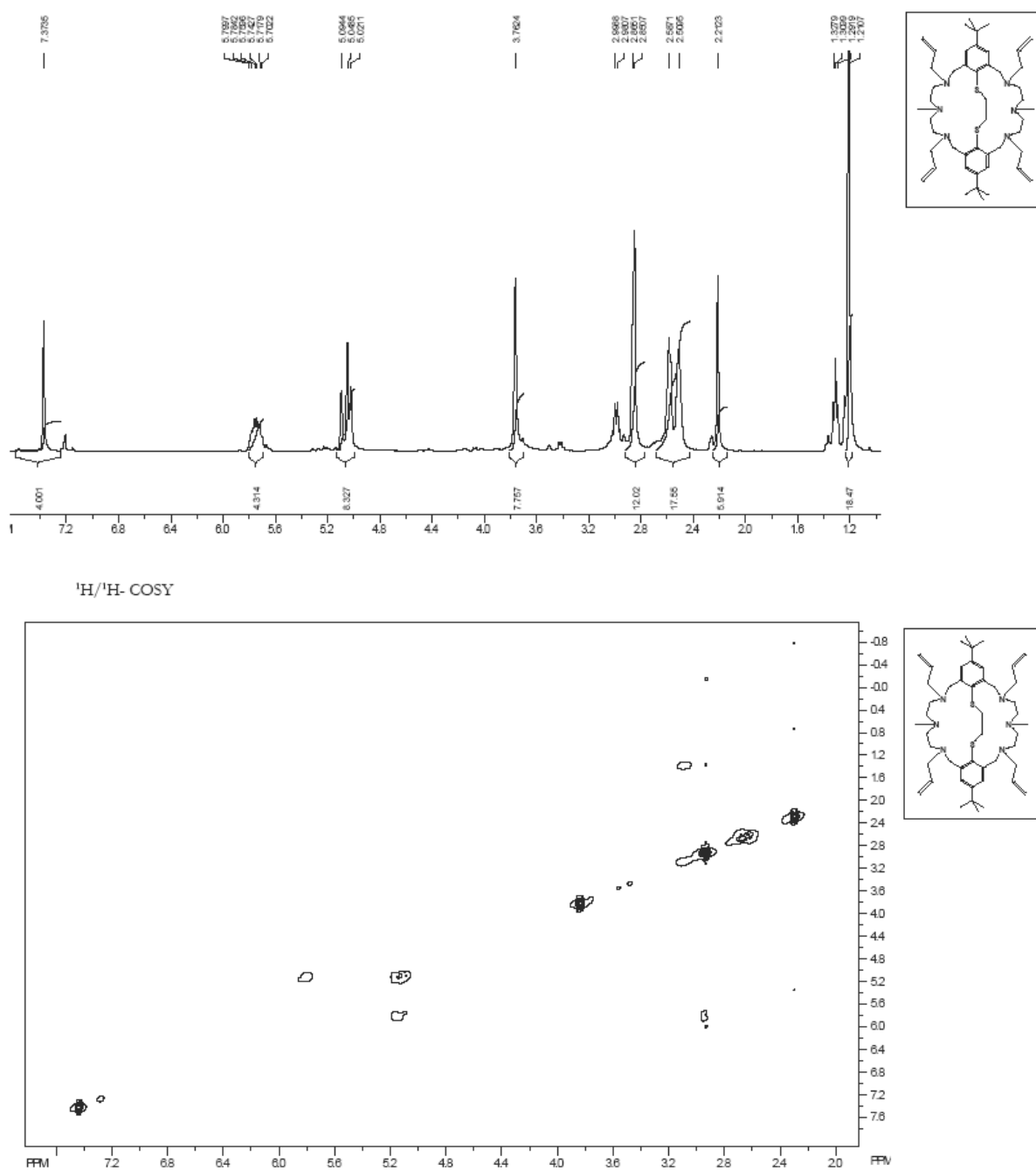


IR spectrum showing Transmittance [%] versus Wavenumber cm^{-1} . The spectrum displays characteristic absorption bands, including a broad peak around 3400 cm^{-1} and several sharp peaks in the fingerprint region below 1500 cm^{-1} .

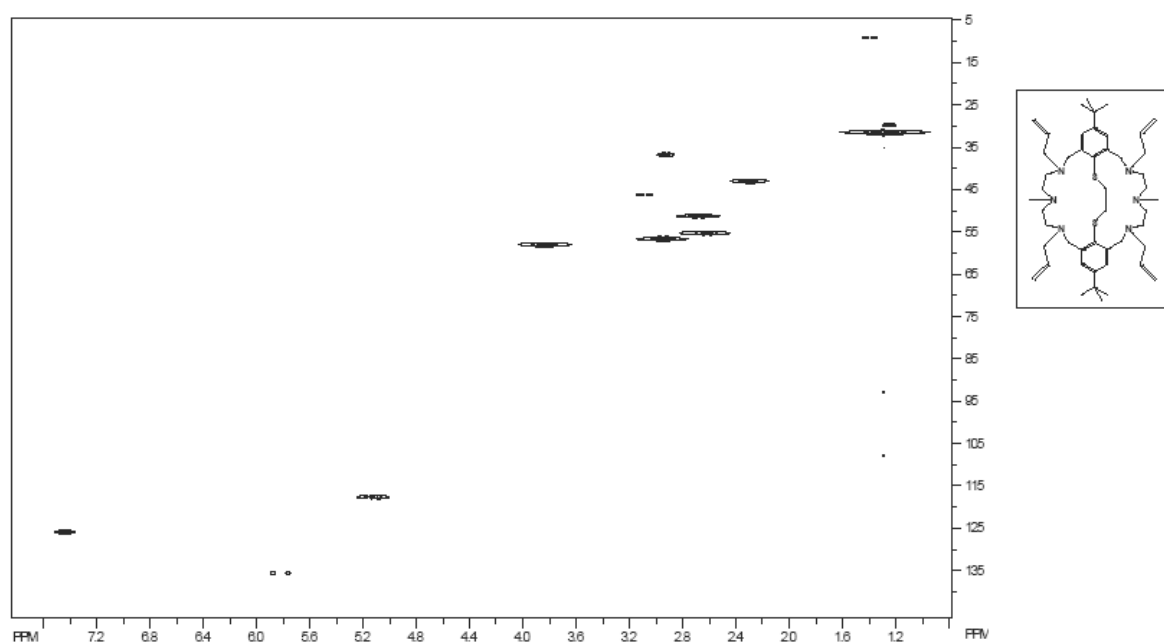
Wavenumber (cm^{-1})
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3077.01
3003.99
2959.31
2927.95
2904.69
2868.32
2811.38
2712.11
1642.59
1593.13
1475.95
1441.43
1416.82
1402.05
1358.03
1280.08
1151.46
1101.21
1023.95
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470.48
448.13
430.29
425.99
412.59



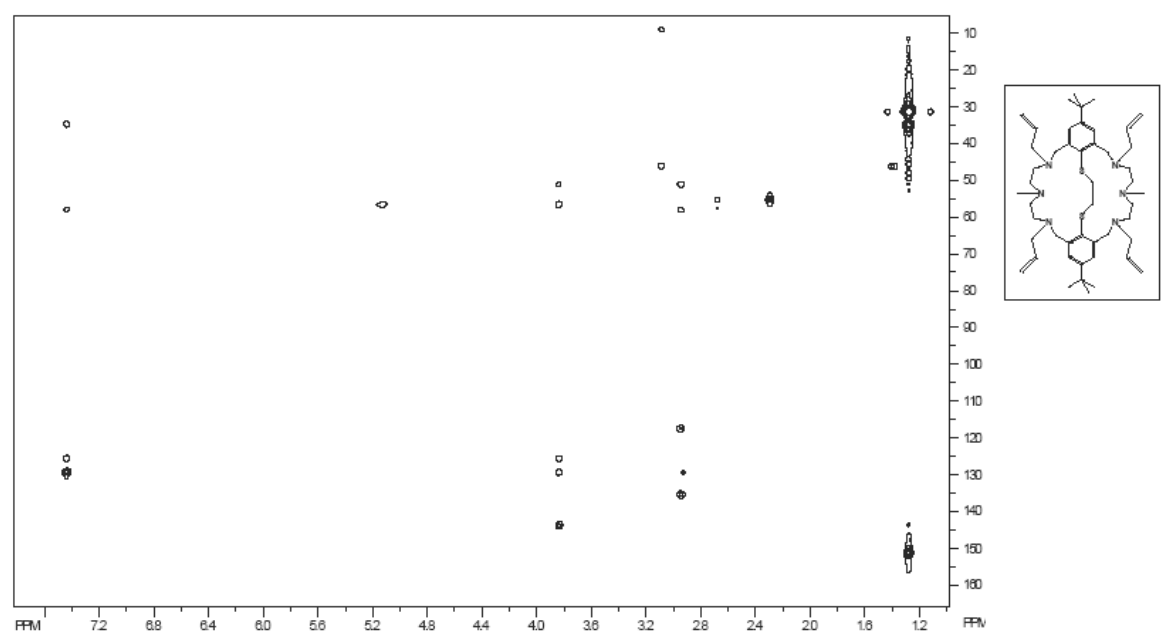
2. Spectra for tetraallylated aza-thioether **9**.



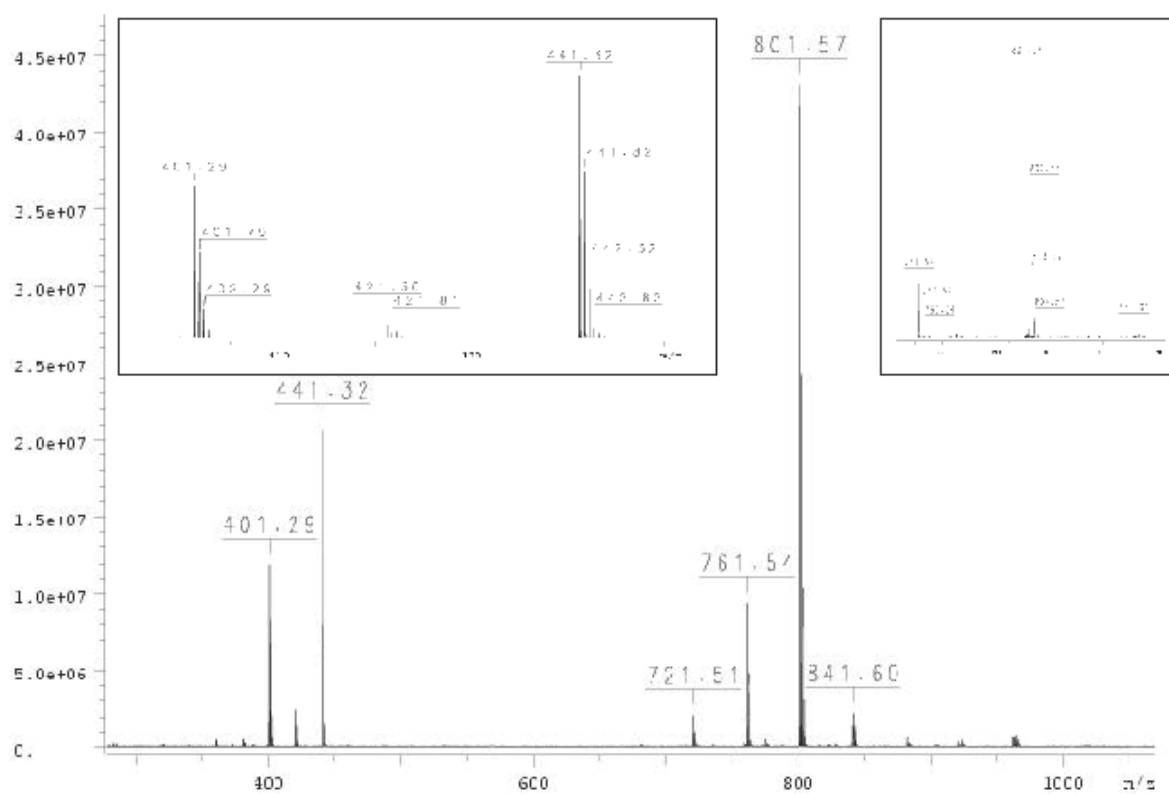
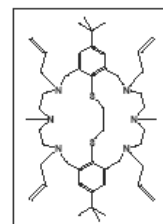
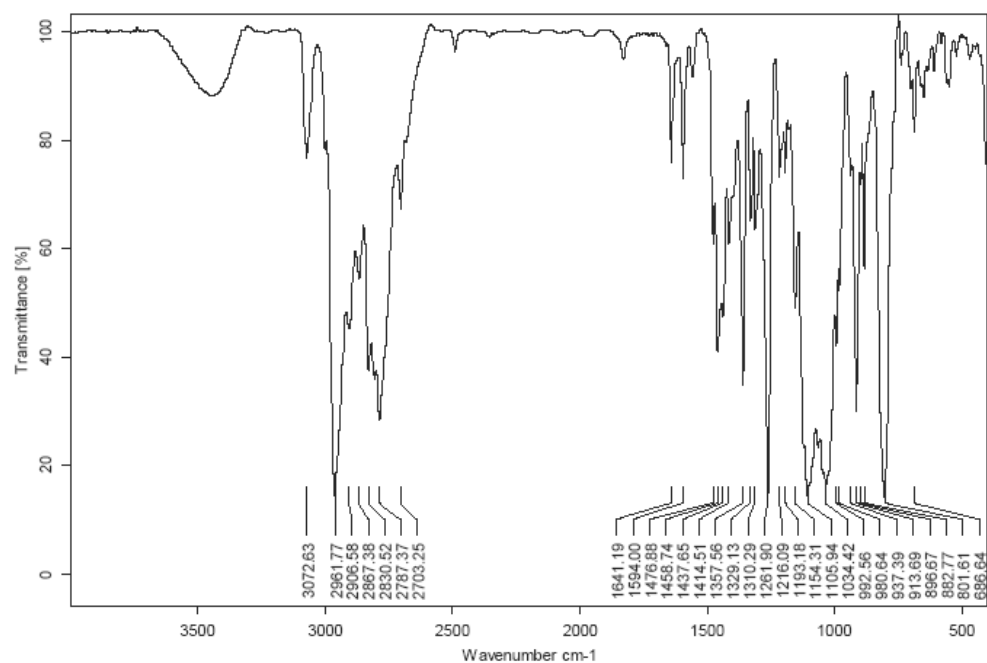
HSQC



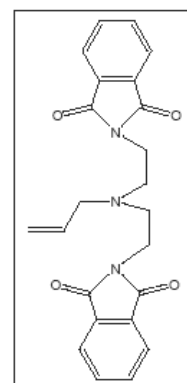
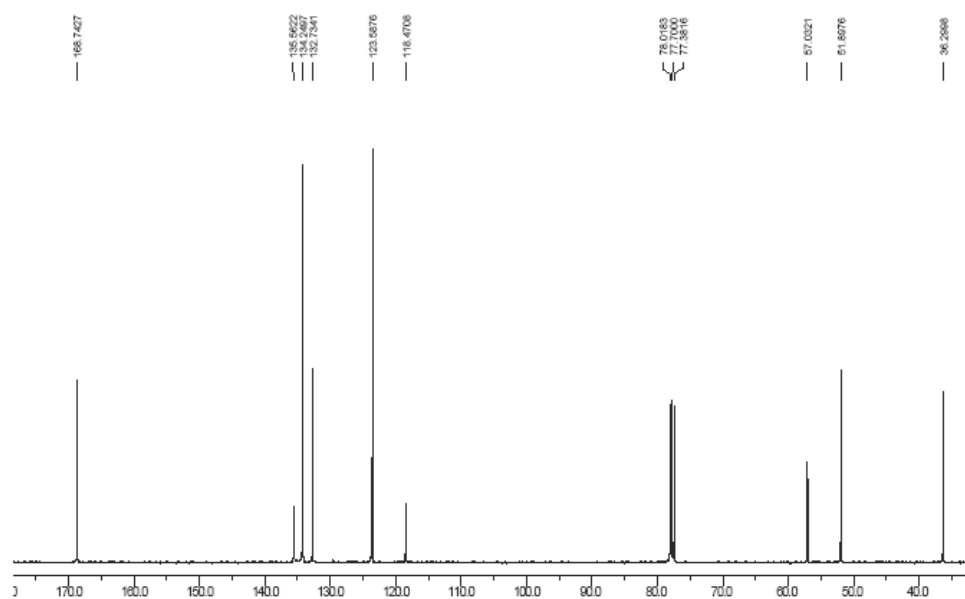
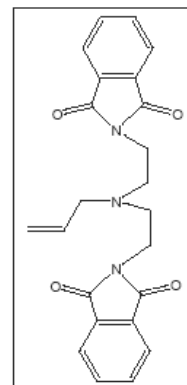
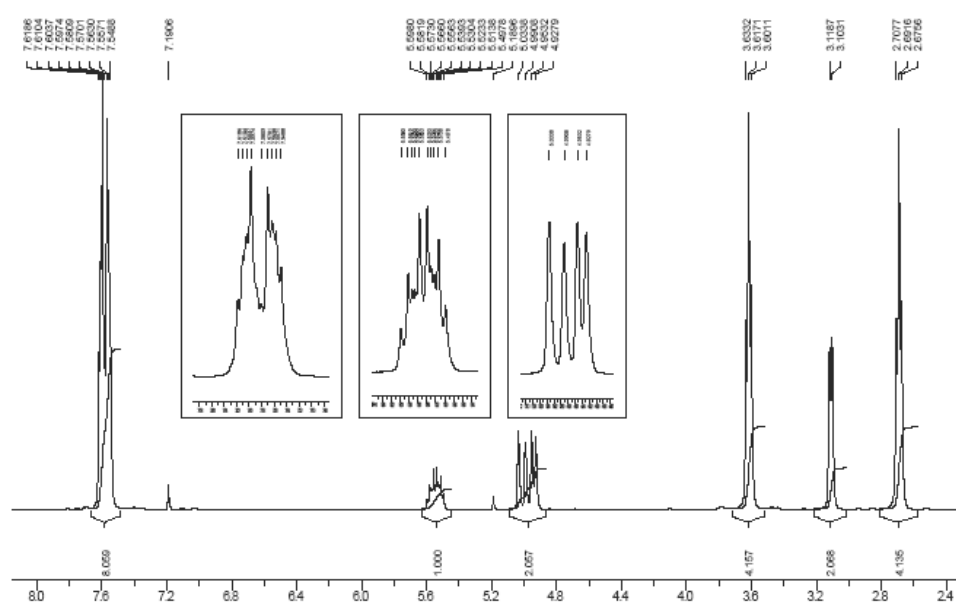
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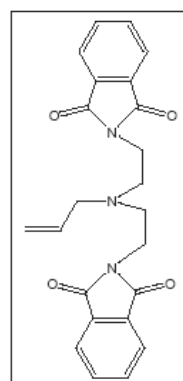
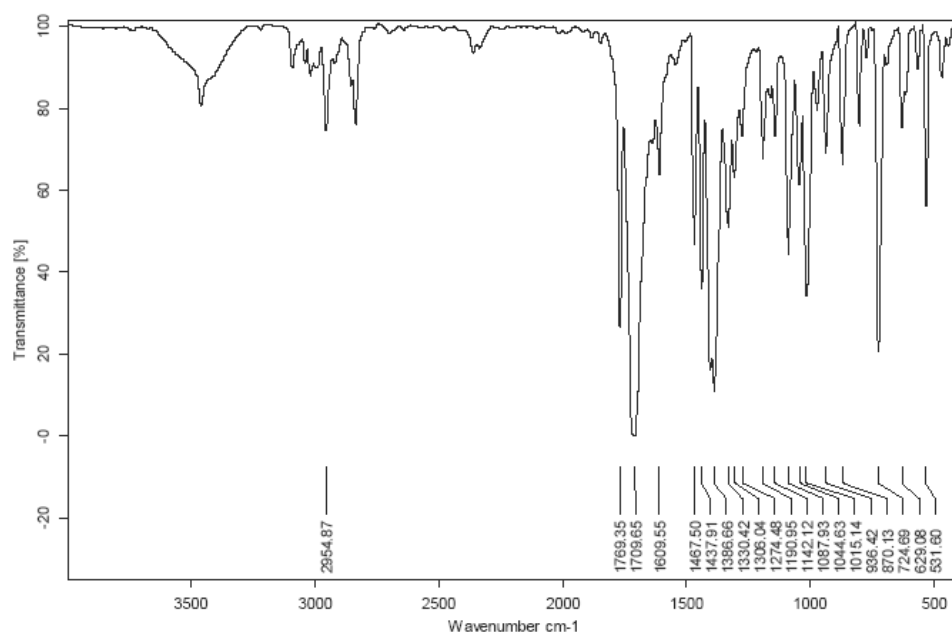
IR- Spektrum



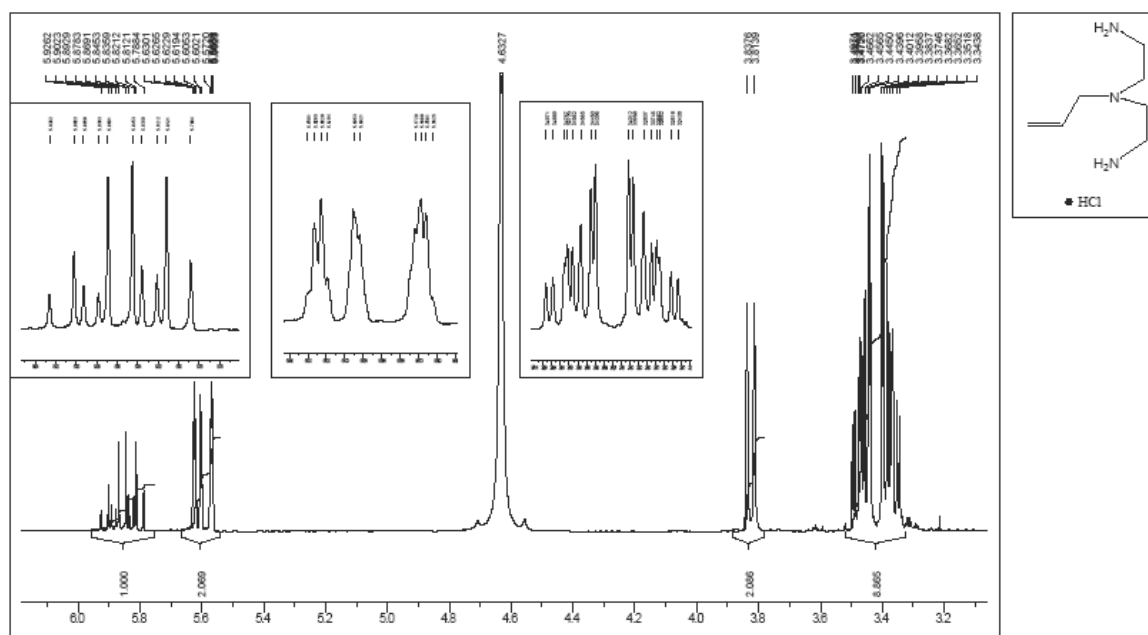
3. Spectra for N-Allyl-bis(2-phthalimidoethyl)amine **12**.



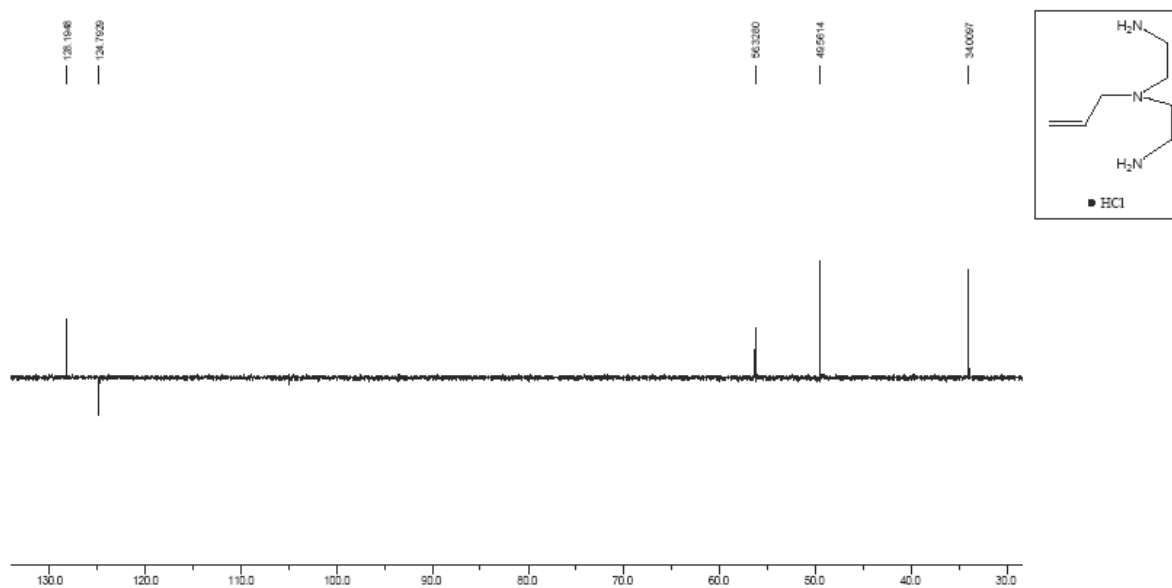
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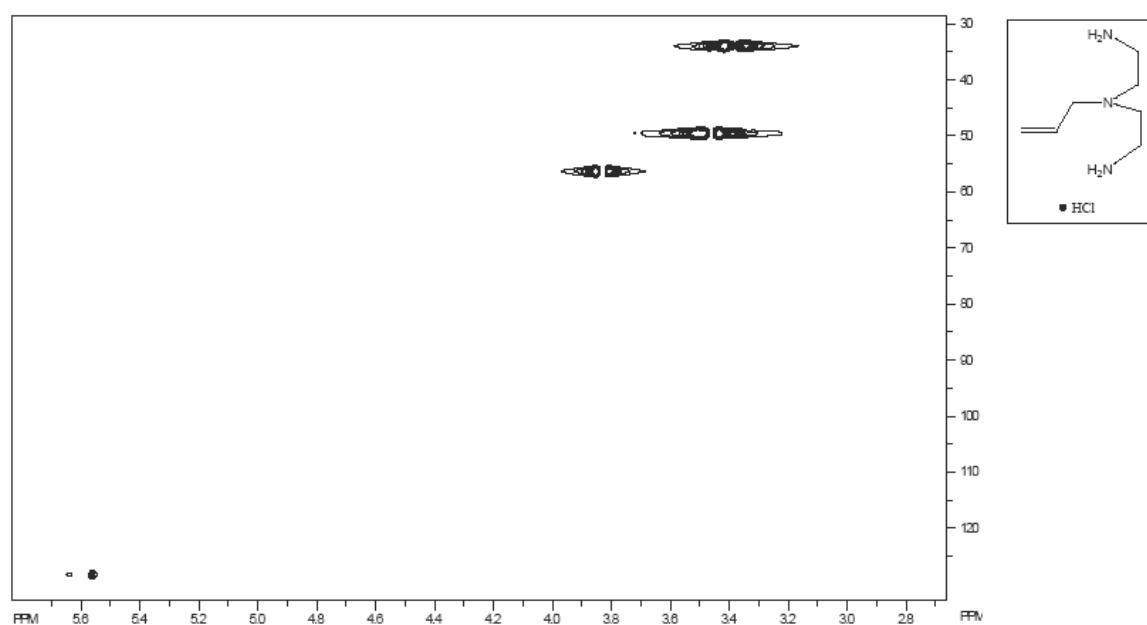
4. Spectra for N-Allyl-bis(2-aminoethyl)amine trihydrochloride (**13**·3HCl).



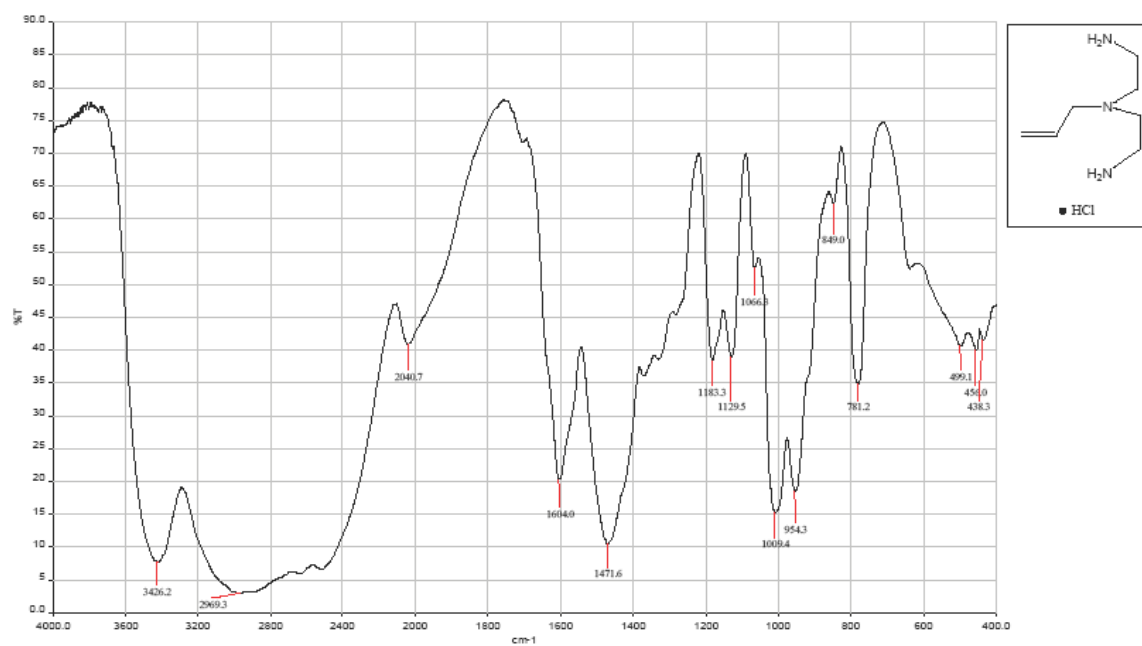
APT (100 MHz, D₂O): δ (ppm) = 34.0 [s, 2 C, (H₂NCH₂CH₂)₂NCH₂CHCH₂], 49.6 [s, 2 C, (H₂NCH₂CH₂)₂NCH₂CHCH₂], 56.3 [s, 1 C, (H₂NCH₂CH₂)₂NCH₂CHCH₂], 124.8 [s, 1 C, (H₂NCH₂CH₂)₂NCH₂CHCH₂], 128.2 [s, 1 C, (H₂NCH₂CH₂)₂NCH₂CHCH₂].



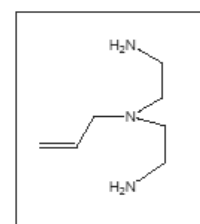
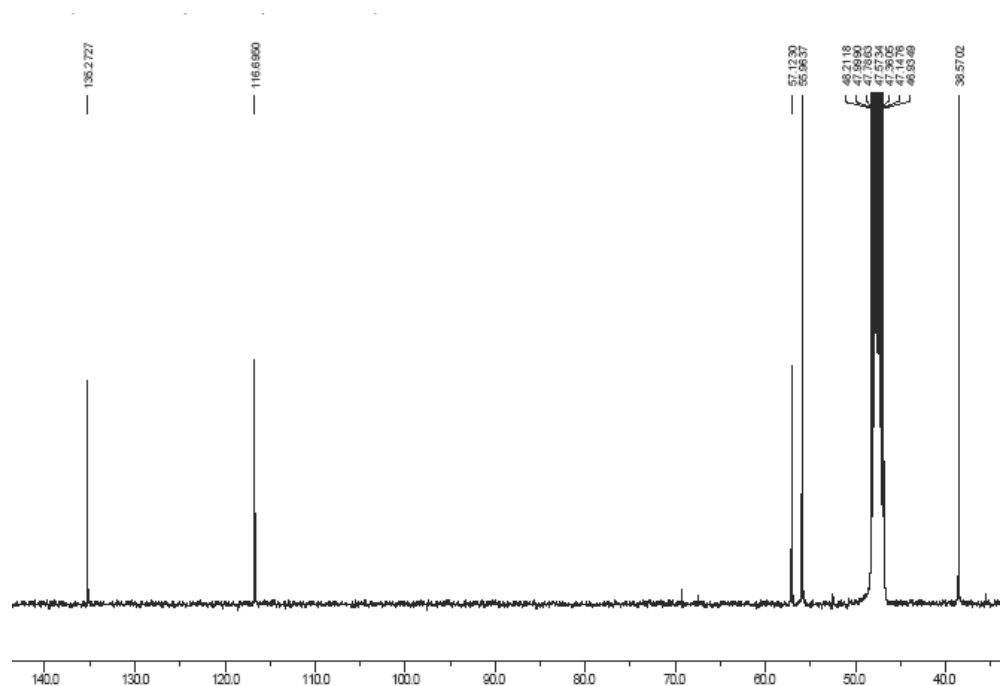
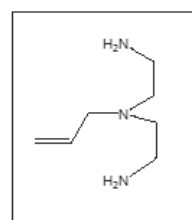
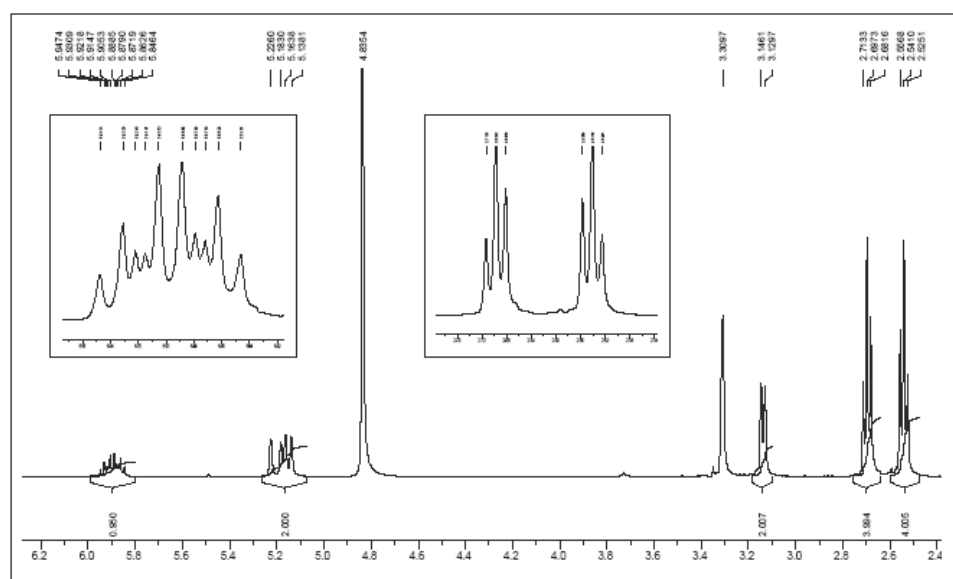
HSQC



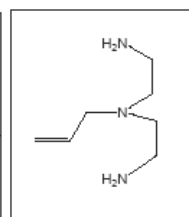
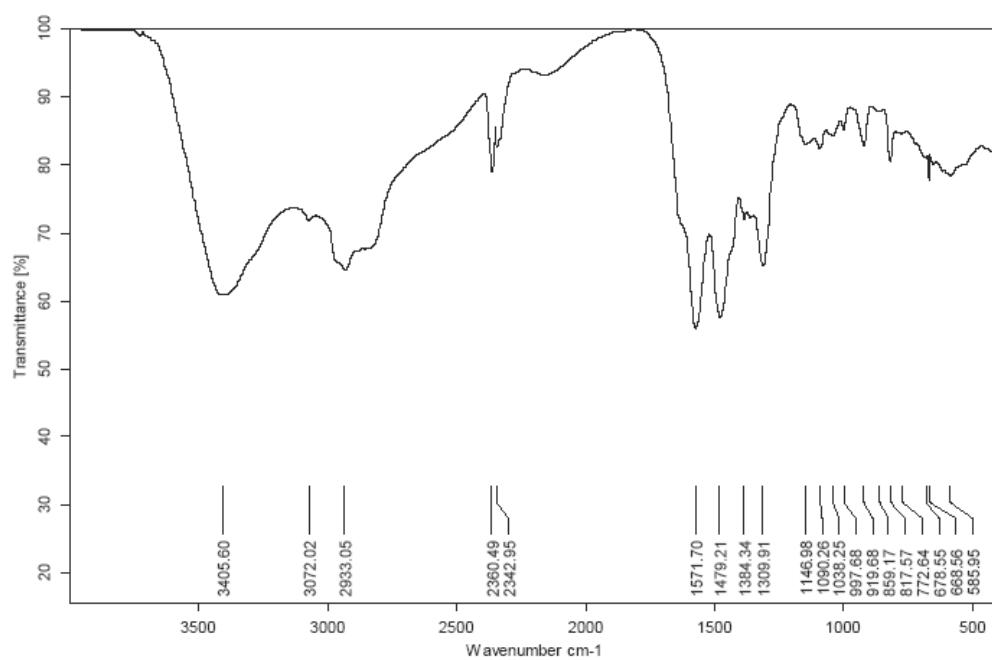
IR-Spektrum



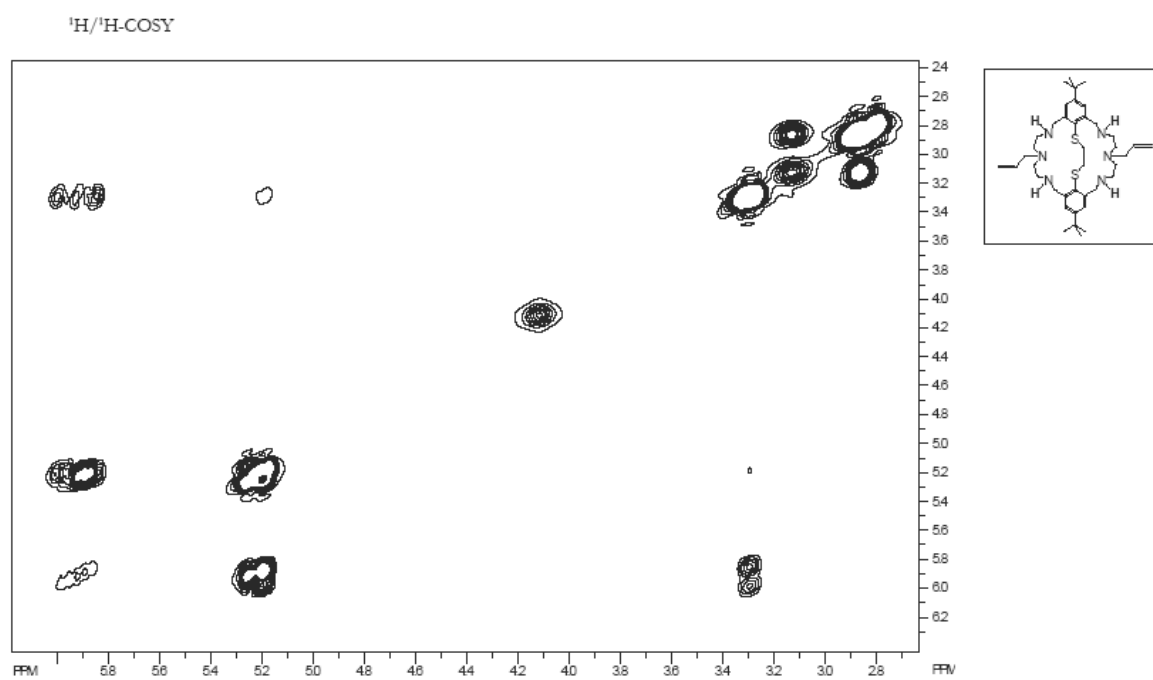
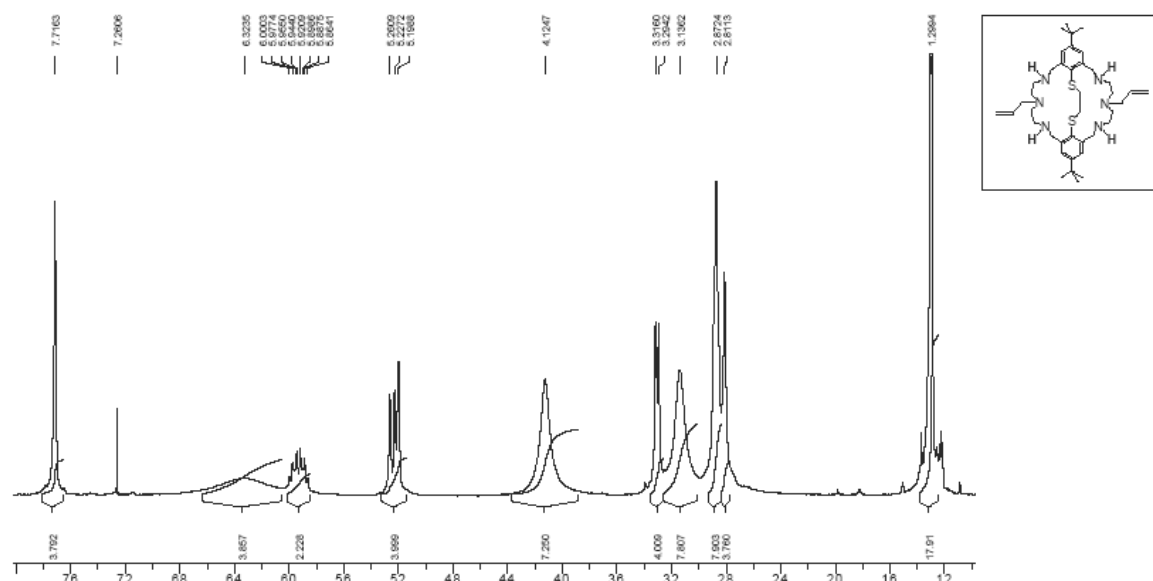
5. Spectra for N-allyl-bis(2-aminoethyl)amine **13**·0.5H₂O.

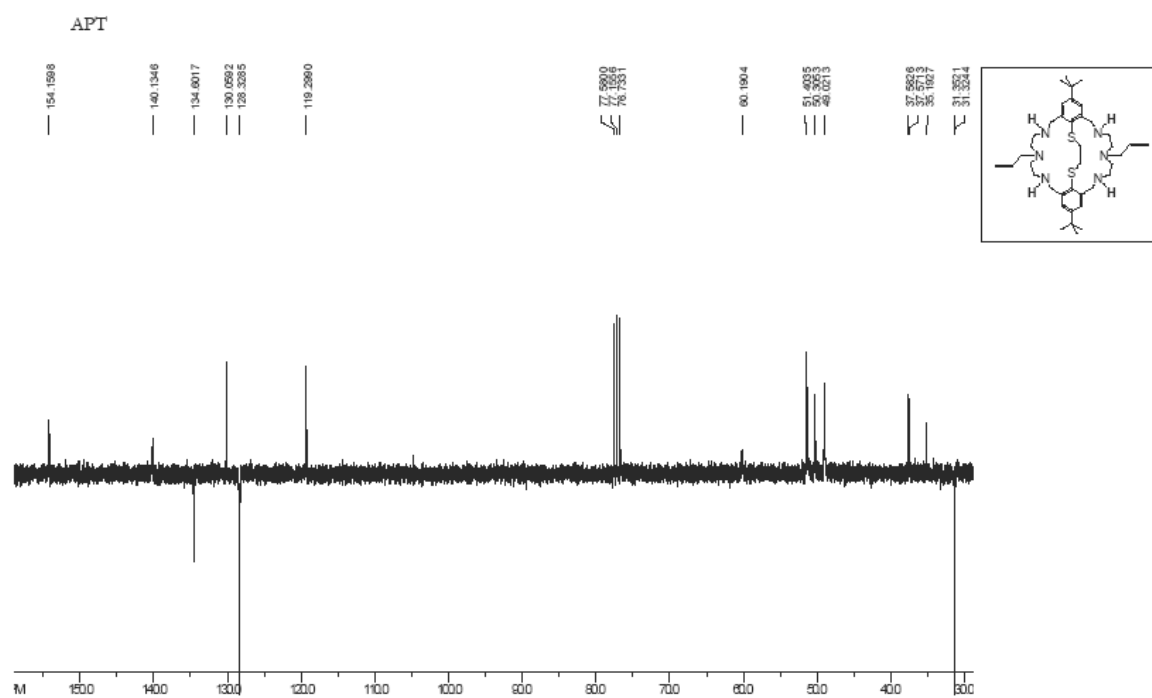
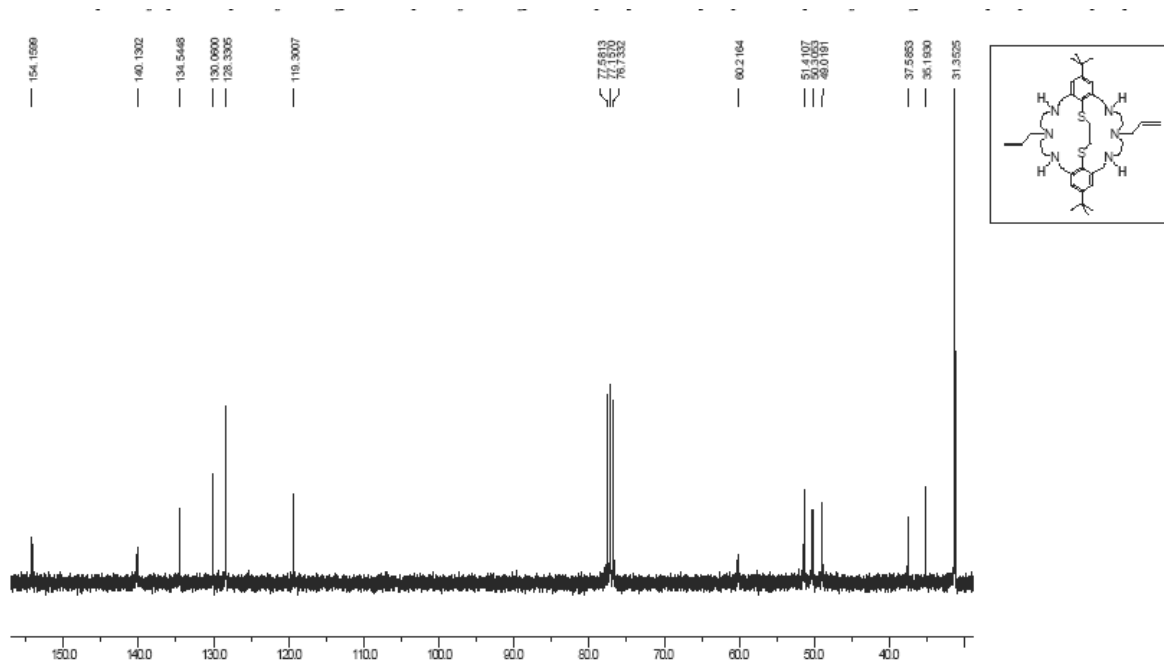


IR-Spektrum

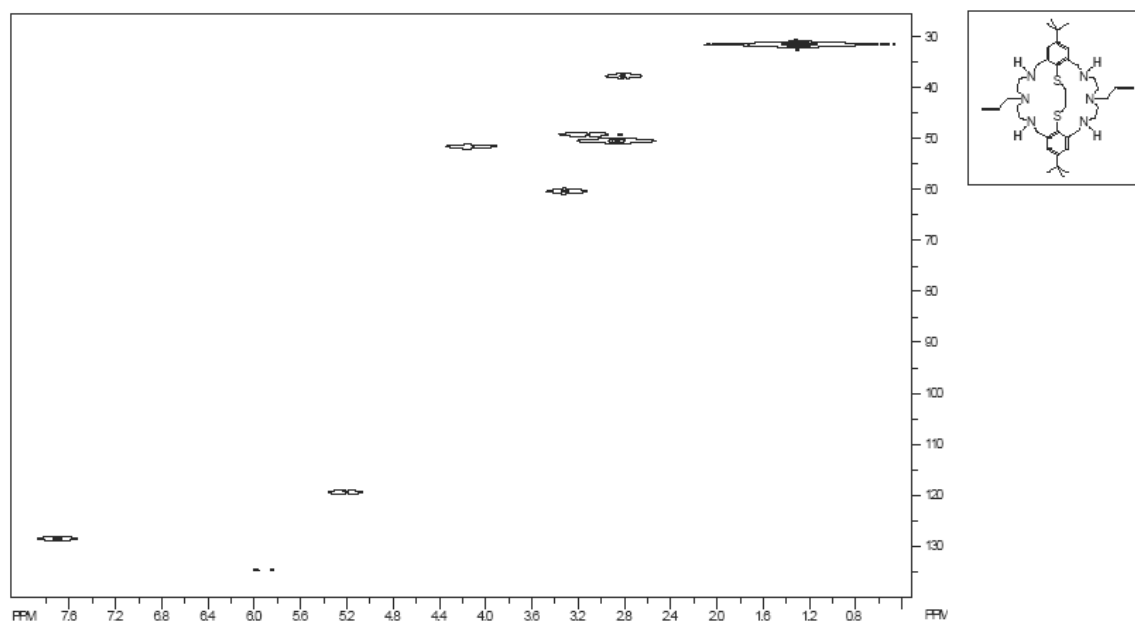


6. Spectra for bis-allylated aza-thioether **14**.

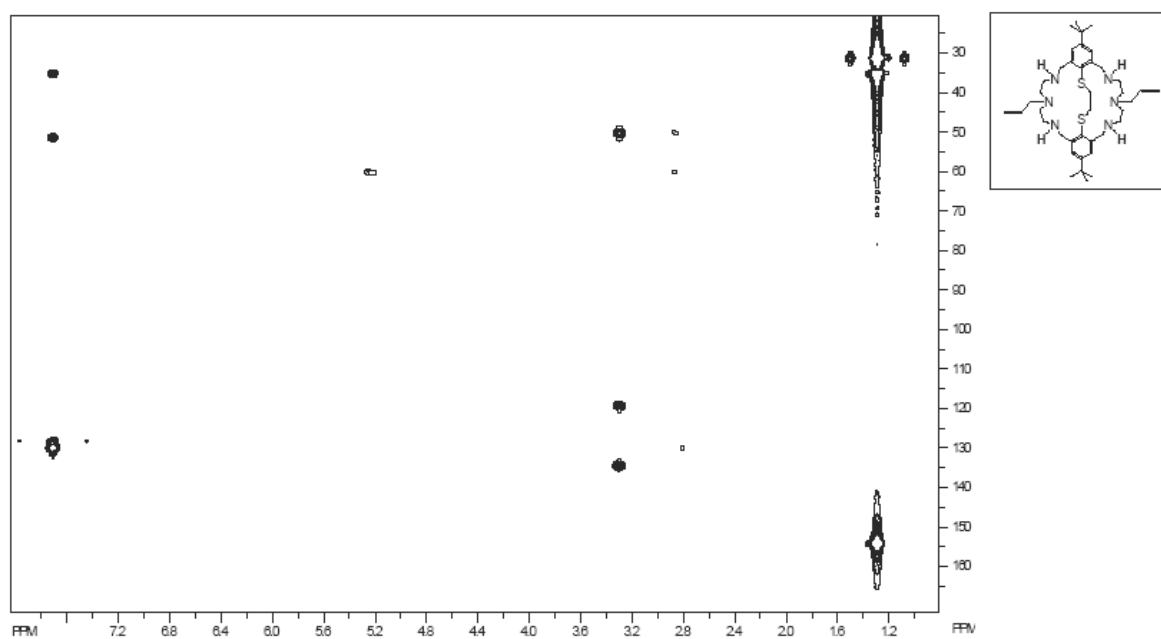




HSQC

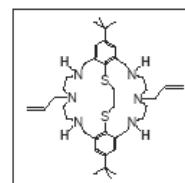


HMBC



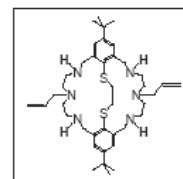
IR spectrum showing Transmittance [%] versus Wavenumber cm^{-1} . The spectrum displays characteristic absorption bands for the polymer structure, including a broad peak around 3400 cm^{-1} , sharp peaks in the 2800-3000 cm^{-1} region, and a complex fingerprint region below 1500 cm^{-1} .

Wavenumber (cm^{-1})
3424.20
3281.30
2955.41
2866.92
2803.30
1588.60
1577.41
1451.22
1417.86
1365.83
1261.54
1230.47
1205.07
1149.08
1094.33
1053.73
1028.54
921.36
897.31
804.94

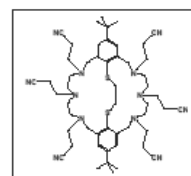
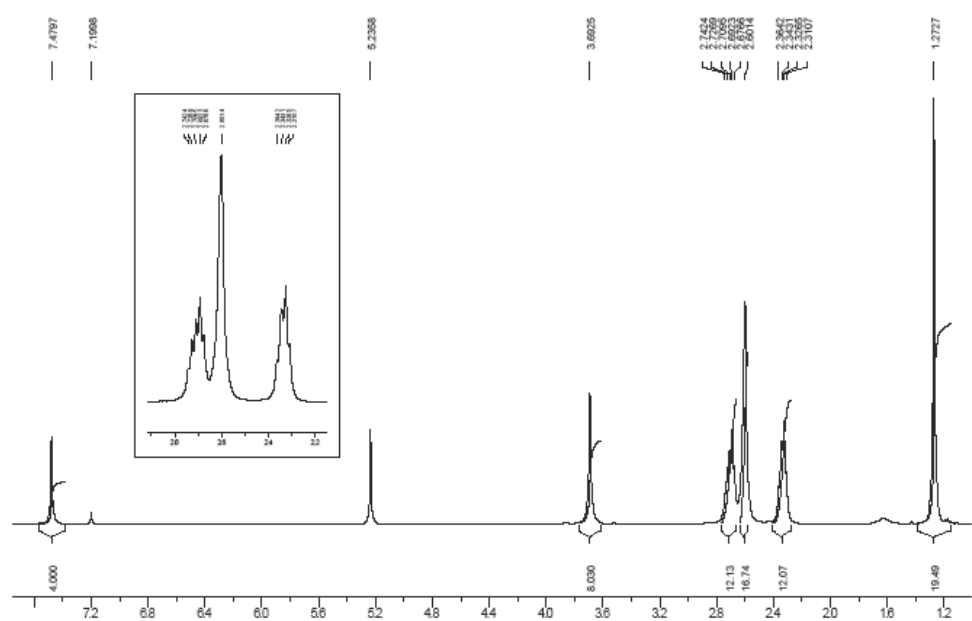


13C NMR spectrum of poly(2-vinylpyridine) in CDCl₃. The spectrum displays four main peaks corresponding to the vinylidene carbons, the backbone carbons, the solvent, and the pyridine ring carbons.

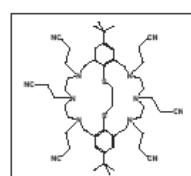
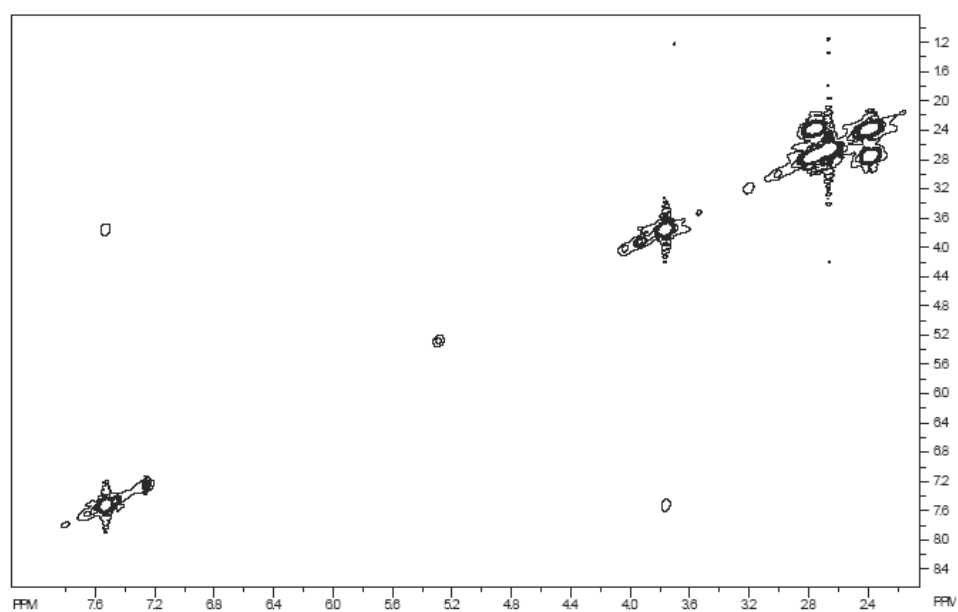
Chemical Shift (ppm)	Assignment
231.18	Vinylidene carbons (CH ₂ =C-)
347.24	Backbone carbons (CH ₂ -CH-)
593.47	Solvent (CDCl ₃)
693.47	Pyridine ring carbons

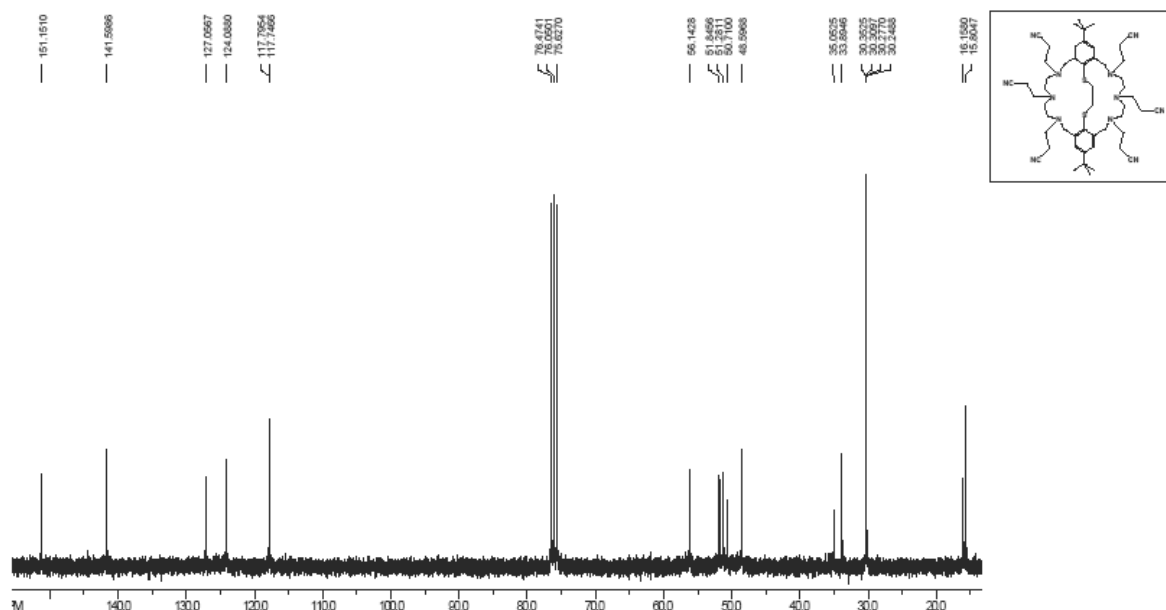


7. Spectra for hexa(cyanoethylated)aza-thioether **15**.

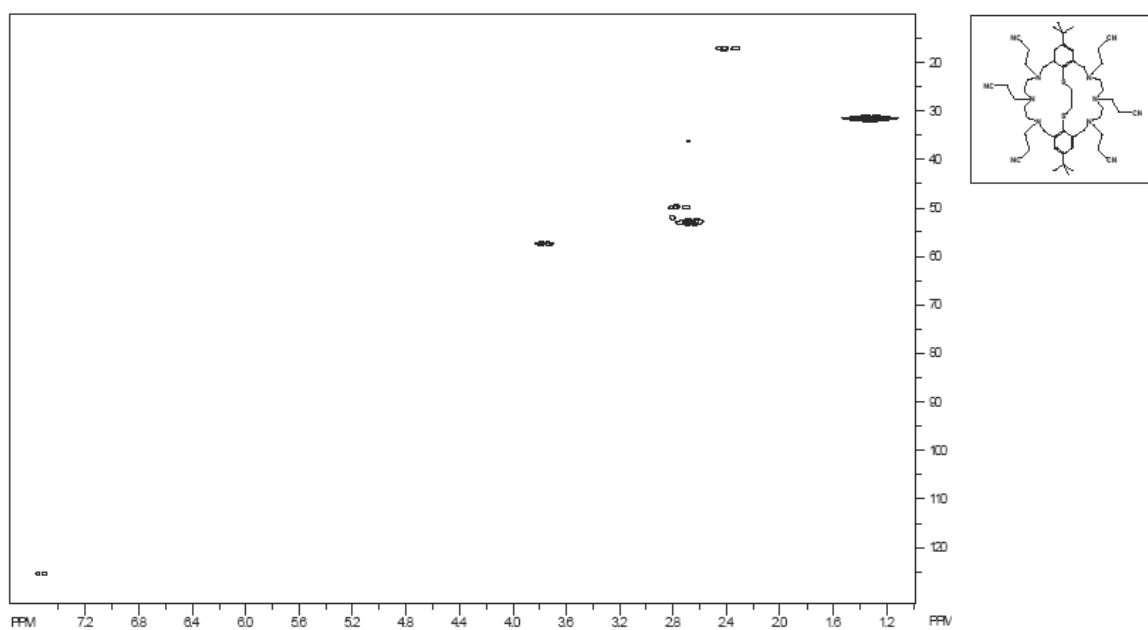


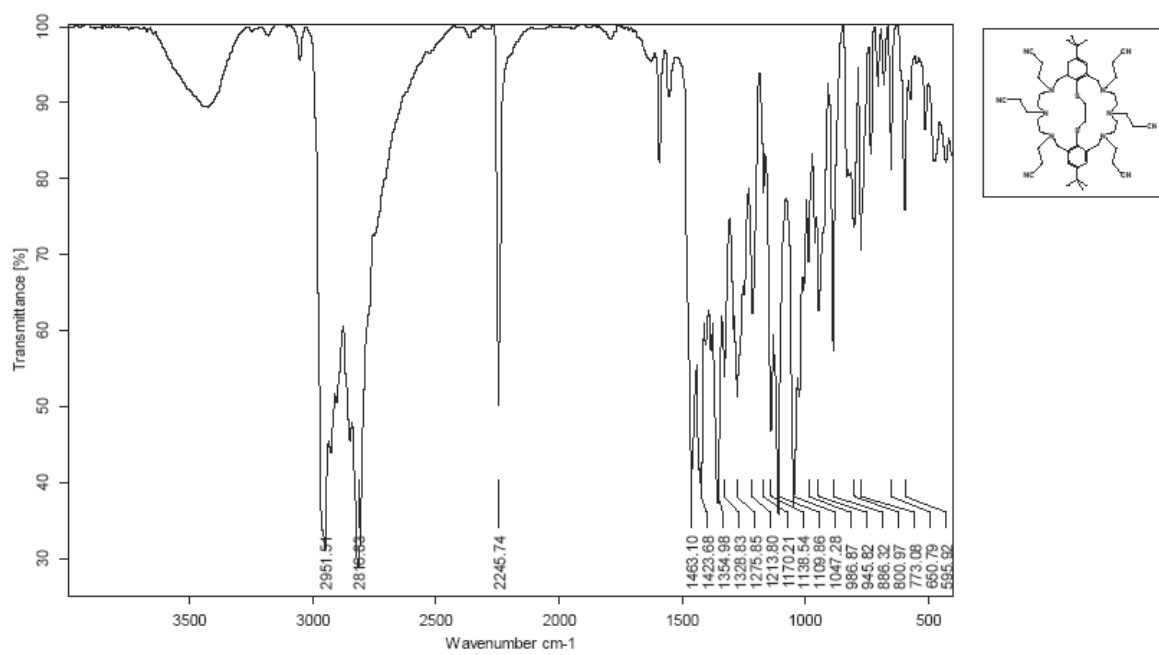
$^1\text{H}/^1\text{H}$ -COSY

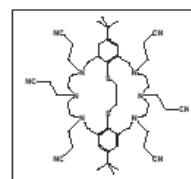
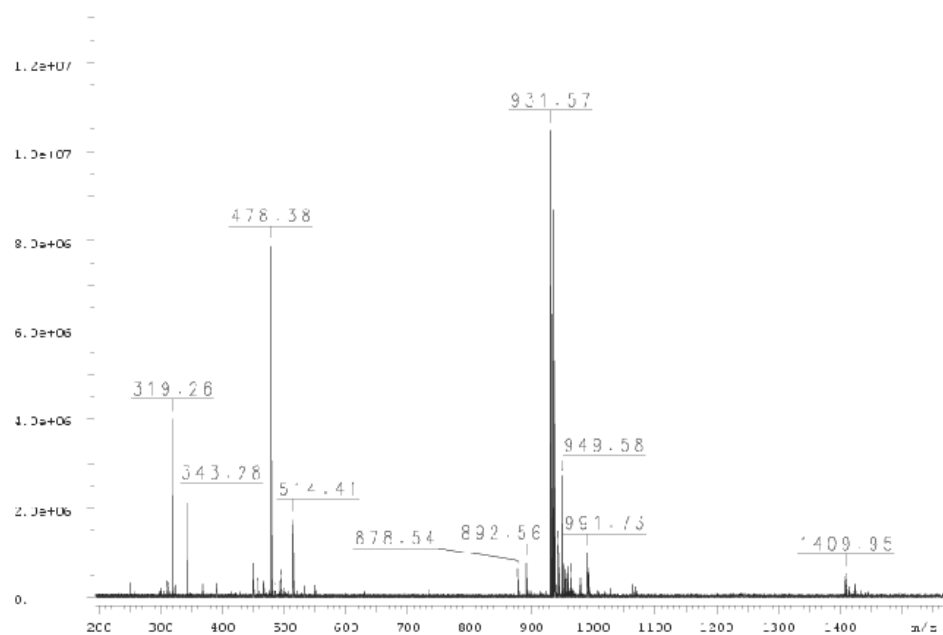




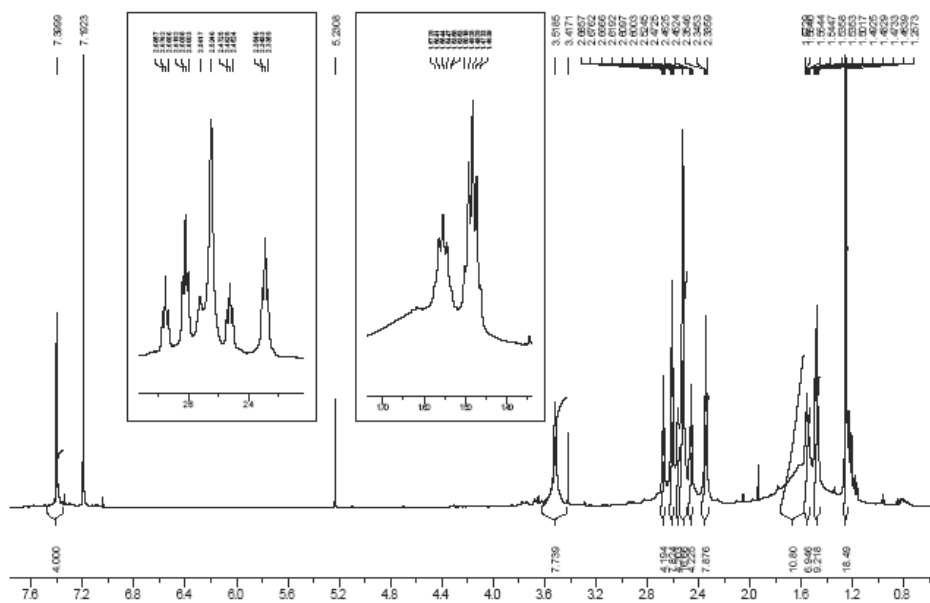
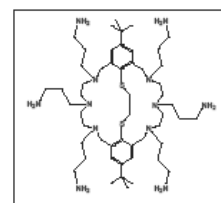
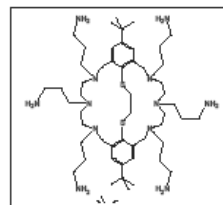
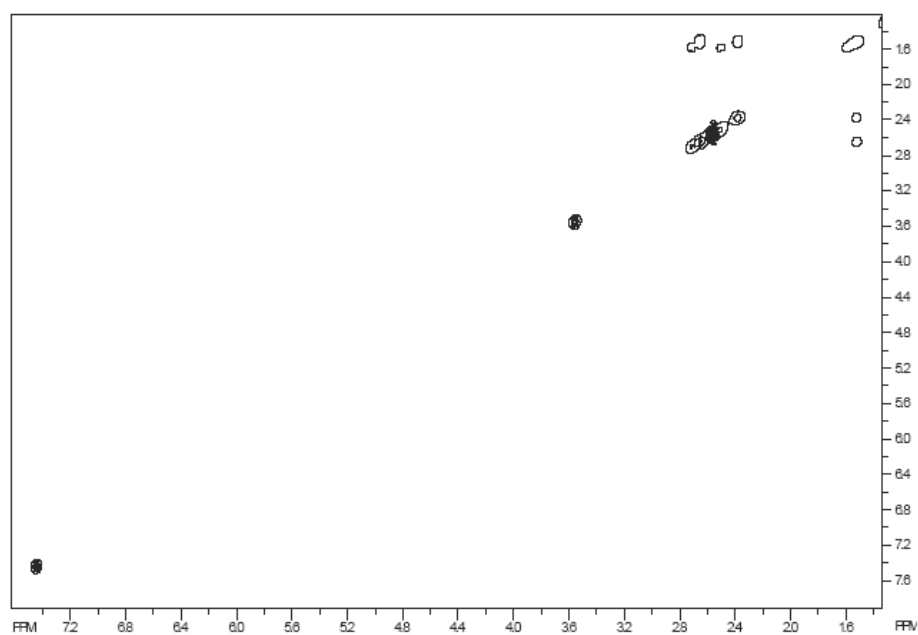
HSQC

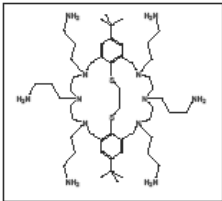




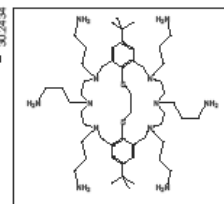


8. Spectra for hexa(3-aminopropylated)aza-thioether **16**.

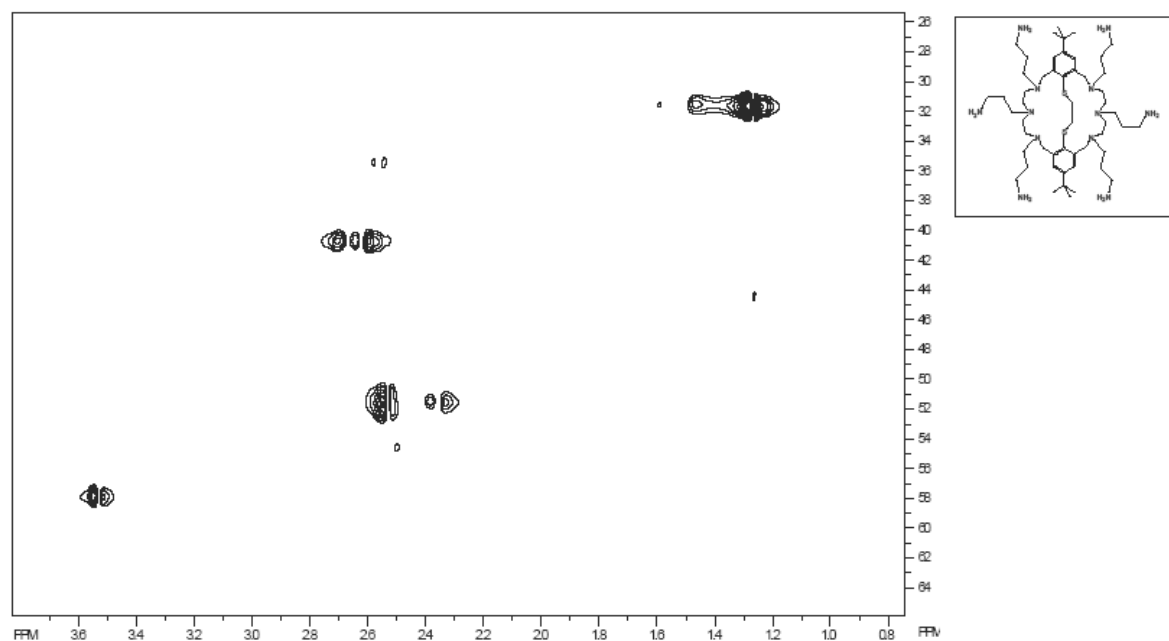
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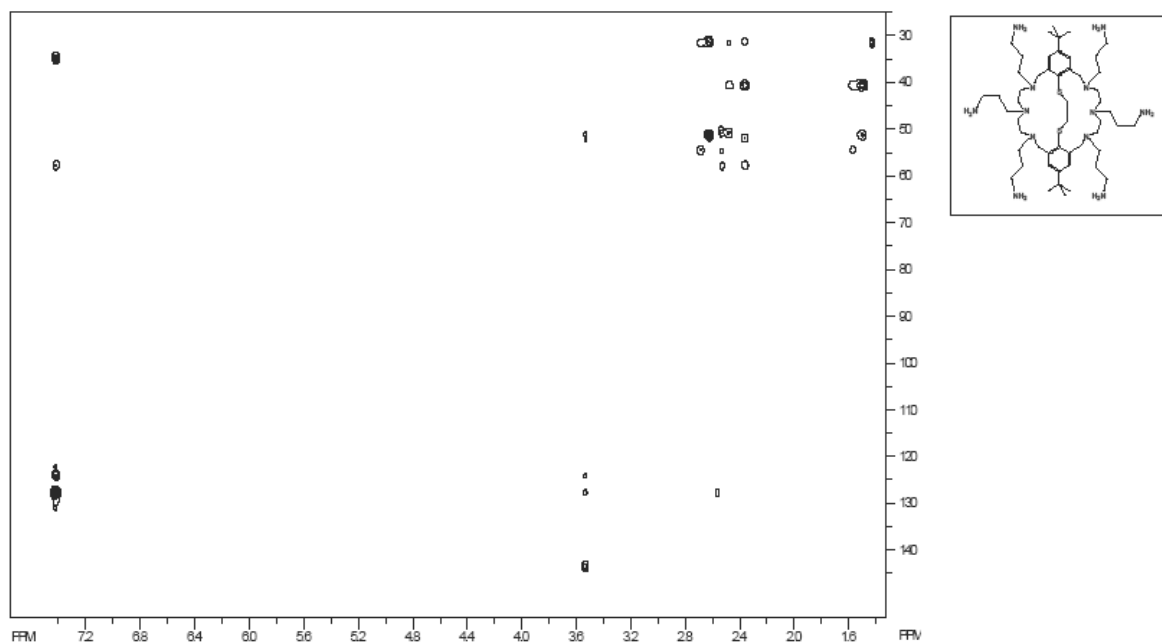
150.1225
142.4538
128.6021
123.2521
76.6126
76.1634
75.6644
56.6508
53.1729
52.7241
50.2635
39.5940
38.4805
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36.5612
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33.2136
32.7644
31.3152
30.8660
30.4168



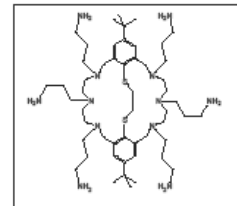
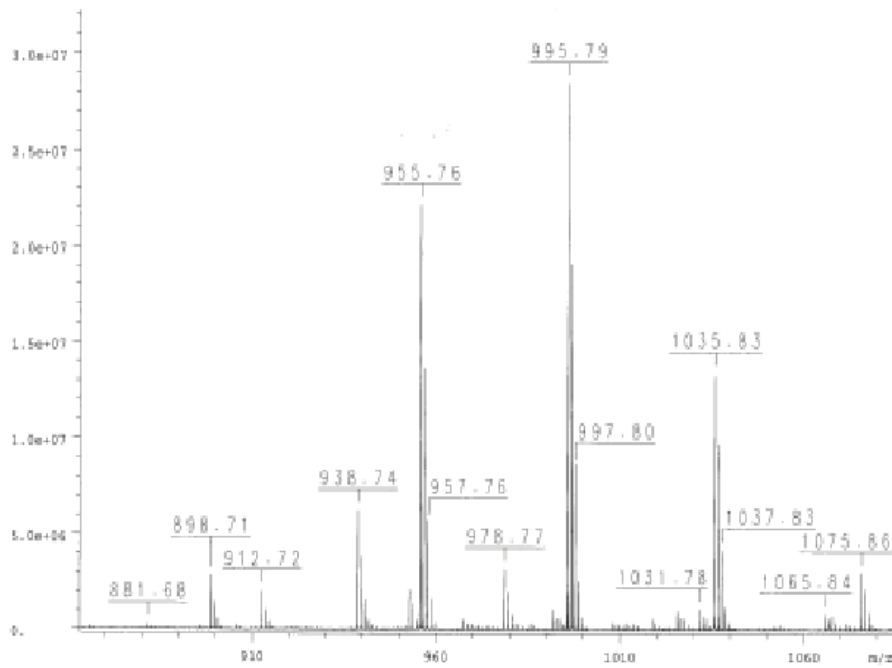
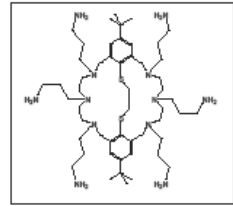
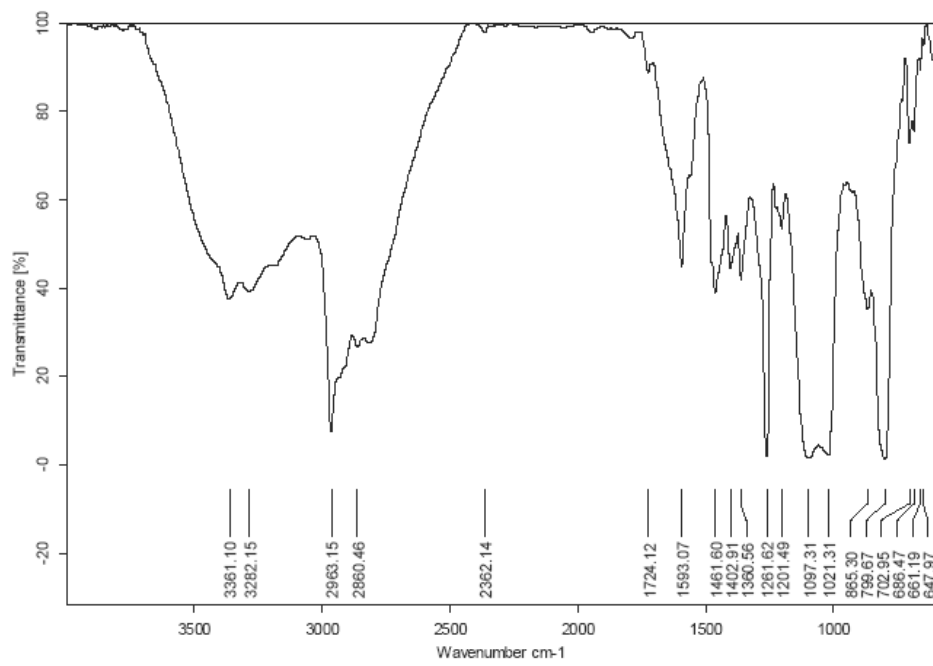
HSQC



HMBC



IR- Spektrum



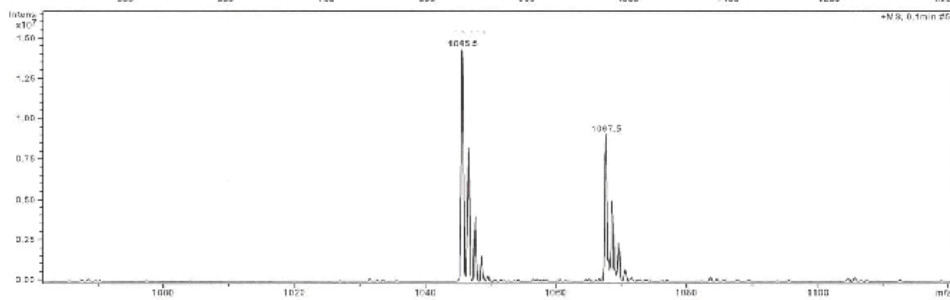
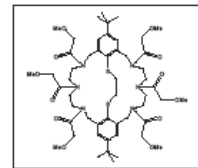
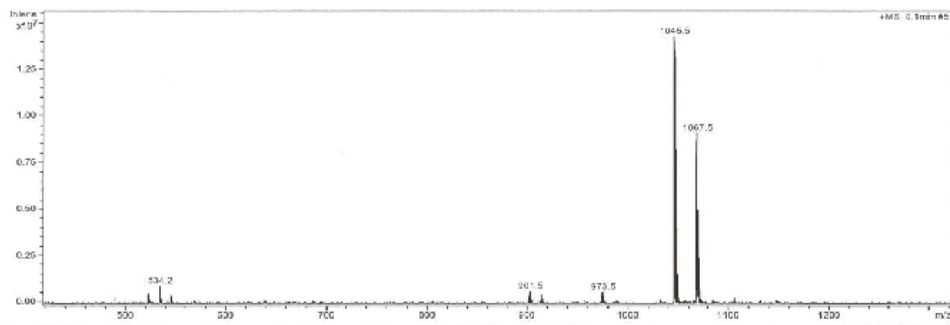
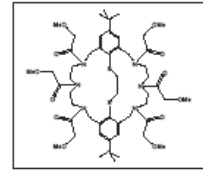
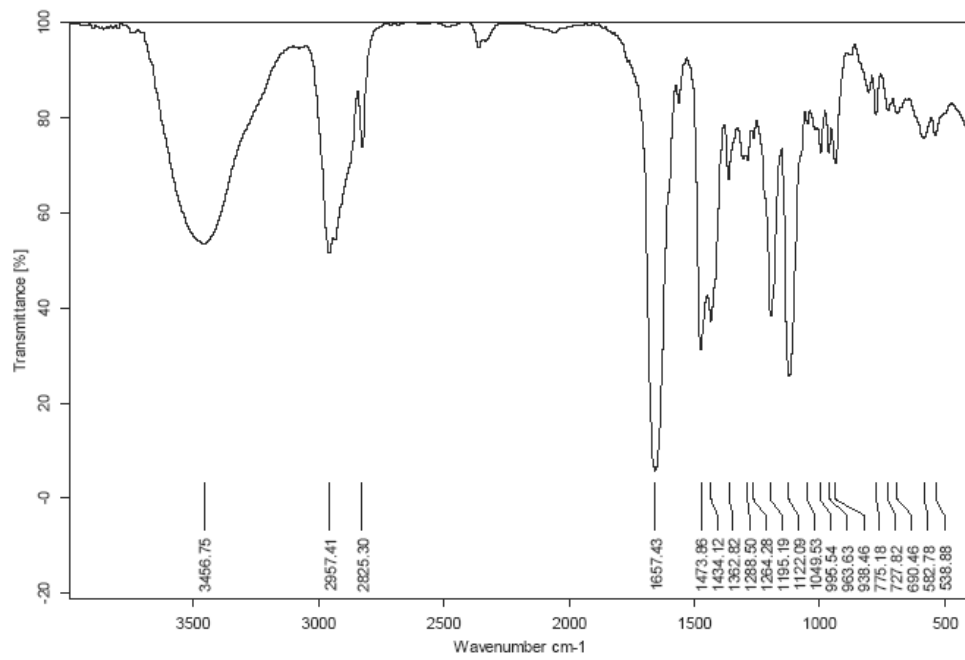
The figure displays the ^1H and ^{13}C NMR spectra of compound 10, along with its chemical structure.

^1H NMR Spectrum (Top): The spectrum shows peaks in the aromatic region (6.8–7.0 ppm), a broad peak around 4.8 ppm, a multiplet between 3.5 and 4.2 ppm, and a sharp peak at 1.954 ppm. Integration values are provided below the peaks: 3.996, 7.997, 4.145, 8.009, 15.55, 11.97, 4.023, and 1.954.

^{13}C NMR Spectrum (Bottom): The spectrum shows peaks in the aromatic region (121–169 ppm), a cluster of peaks between 33 and 40 ppm, and a peak at 29.809 ppm. The chemical shift values are listed on the left: 168.6921, 168.6194, 151.6901, 140.6993, 125.8198, 121.1653, 70.2054, 70.1176, 57.7328, 57.6837, 48.4922, 44.8992, 44.8592, 39.7517, 39.7517, 39.5947, 39.5947, 39.3971, 39.3971, 33.3617, 33.3617, and 29.8099.

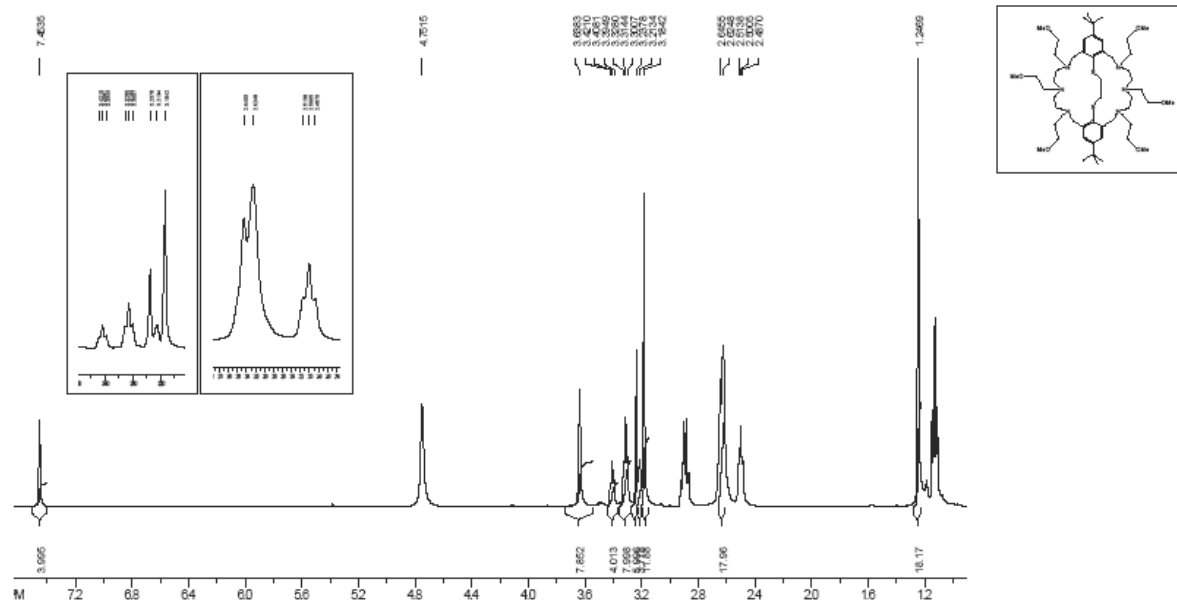
Chemical Structure (Right): The structure of compound 10 is shown, featuring a central core with multiple substituents, including a long alkyl chain and a terminal group.

IR-Spektrum

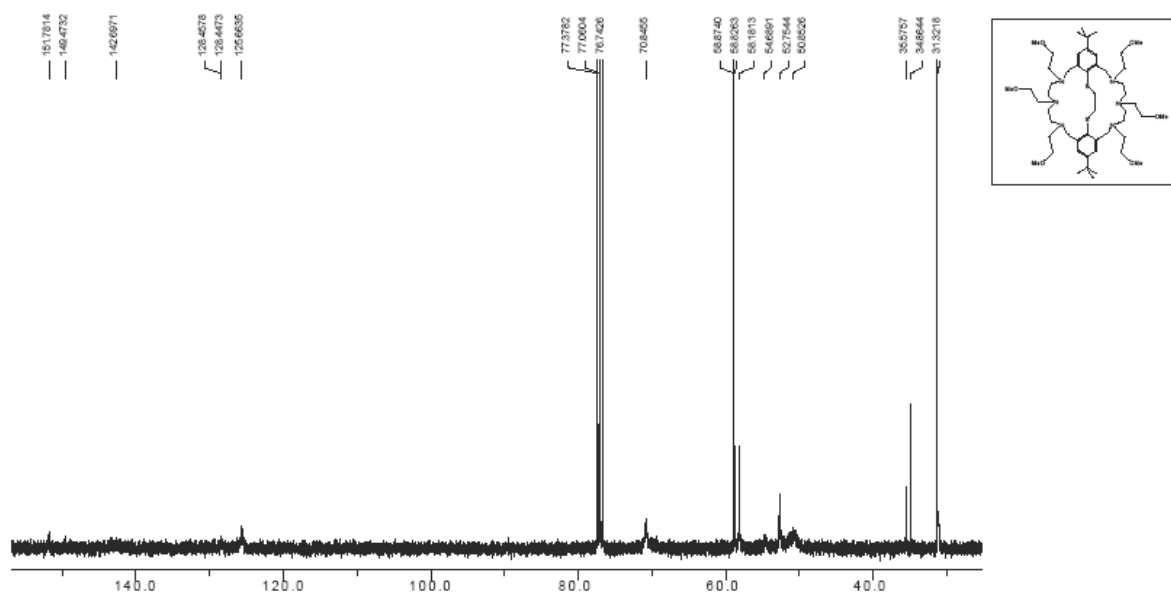
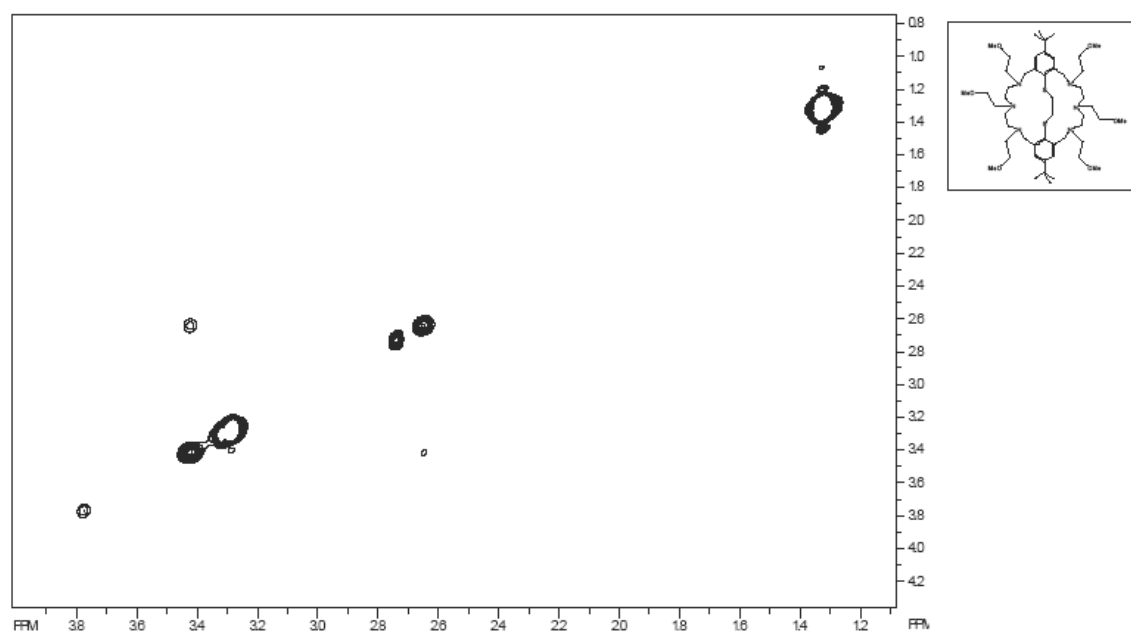


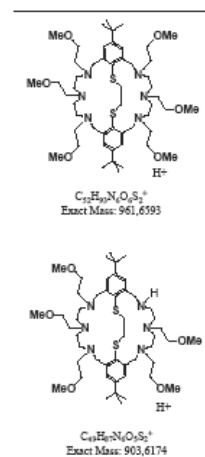
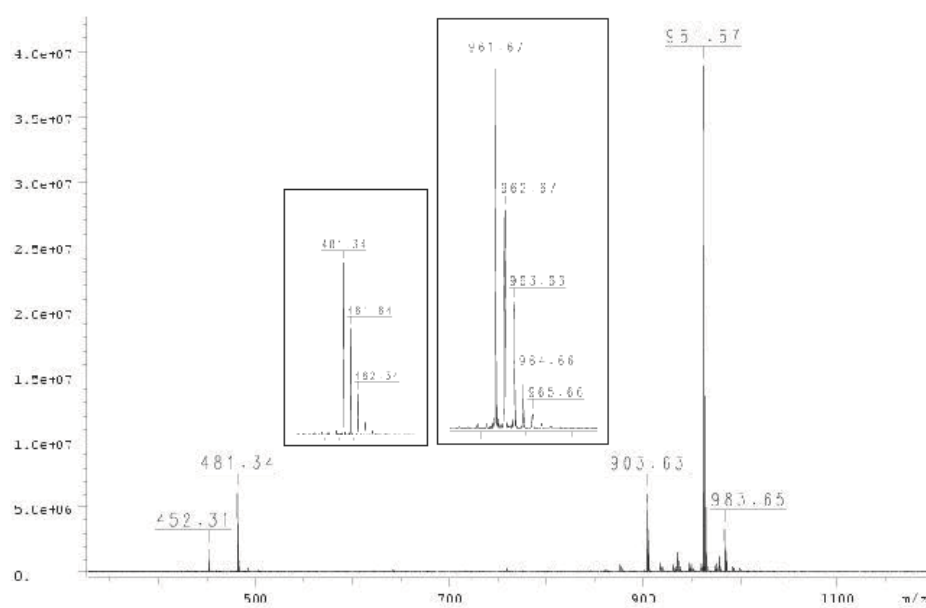
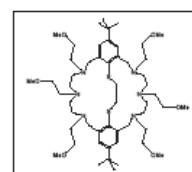
10. Spectra for hexa(2-methoxyethylated)aza-thioether **18**.

^1H NMR (400 MHz, CDCl_3)

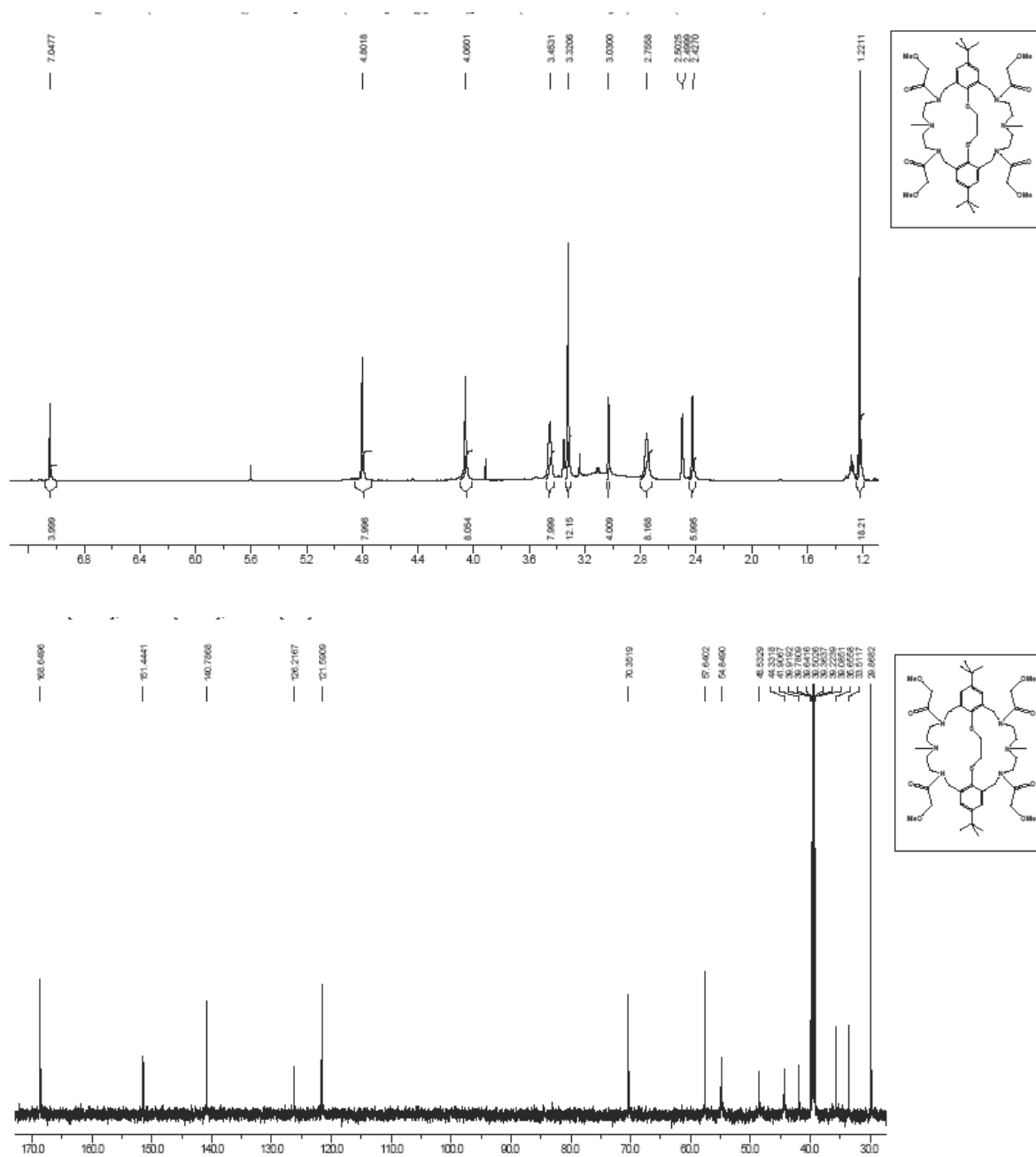


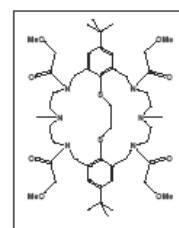
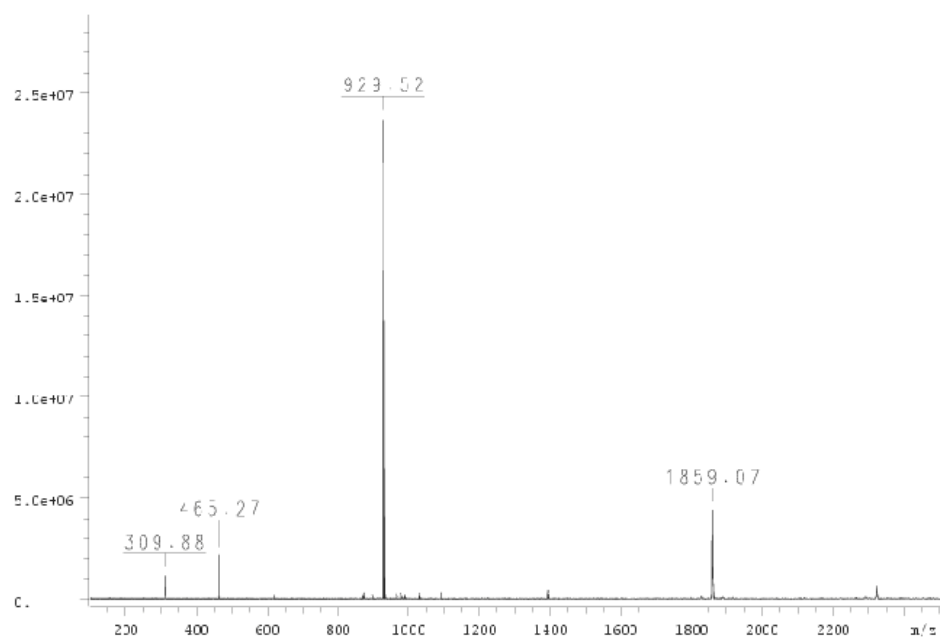
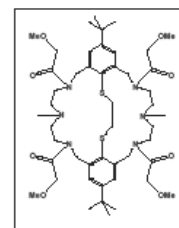
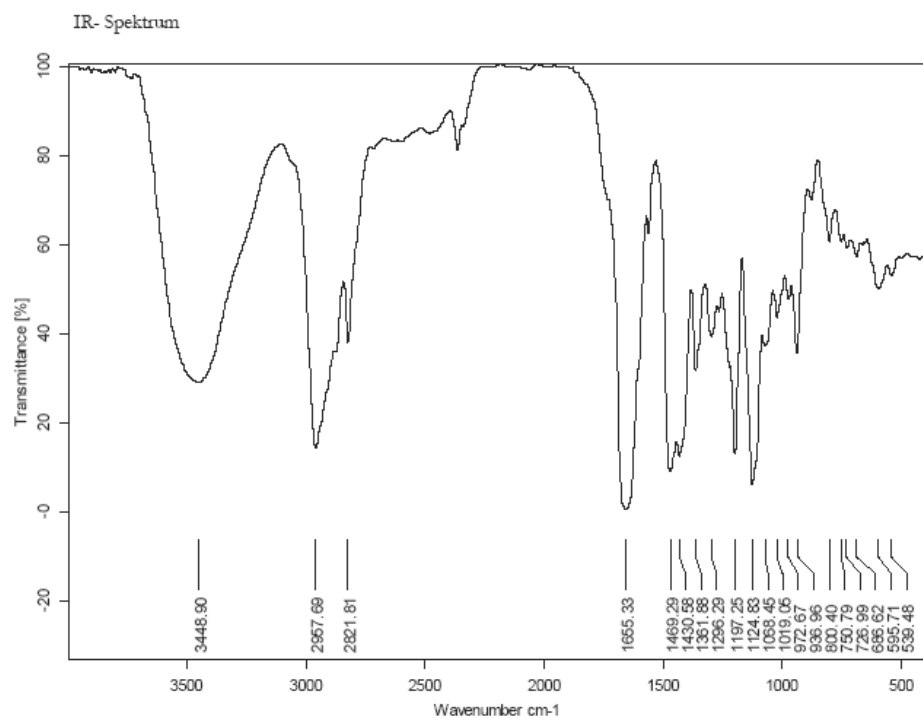
$^1\text{H}/^1\text{H}$ -COSY



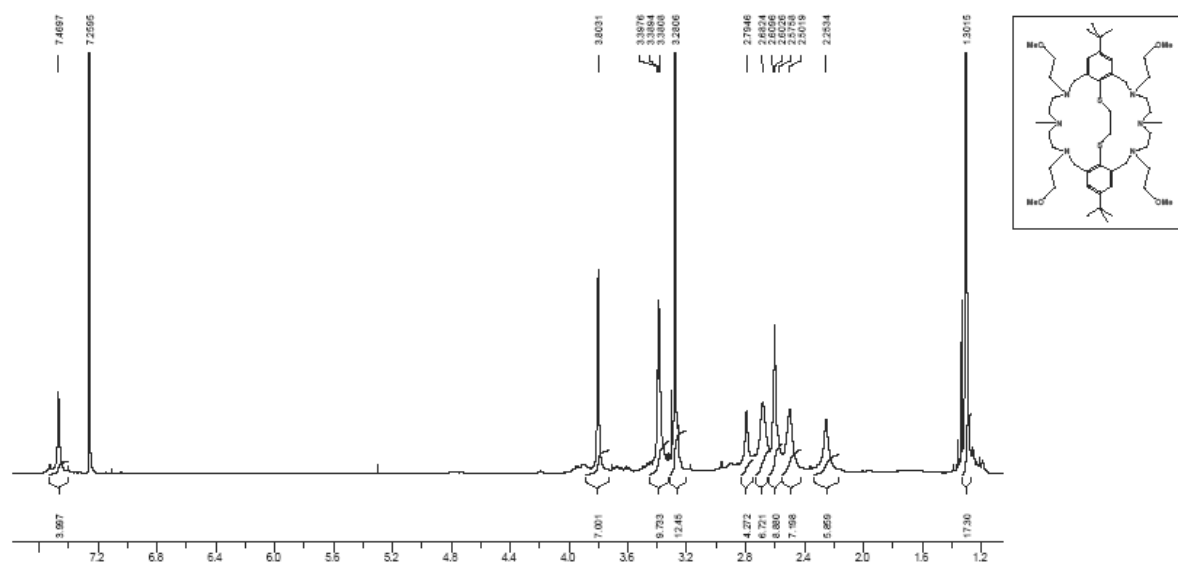


11. Spectra for tetra(2-methoxyacetylated)aza-thioether **19**.

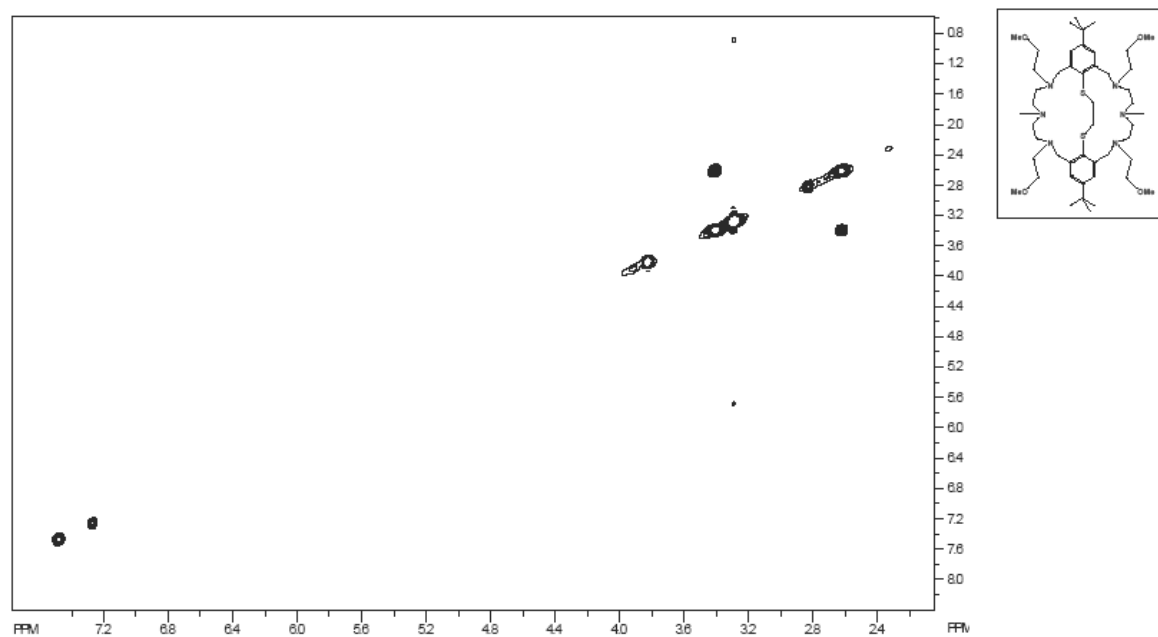




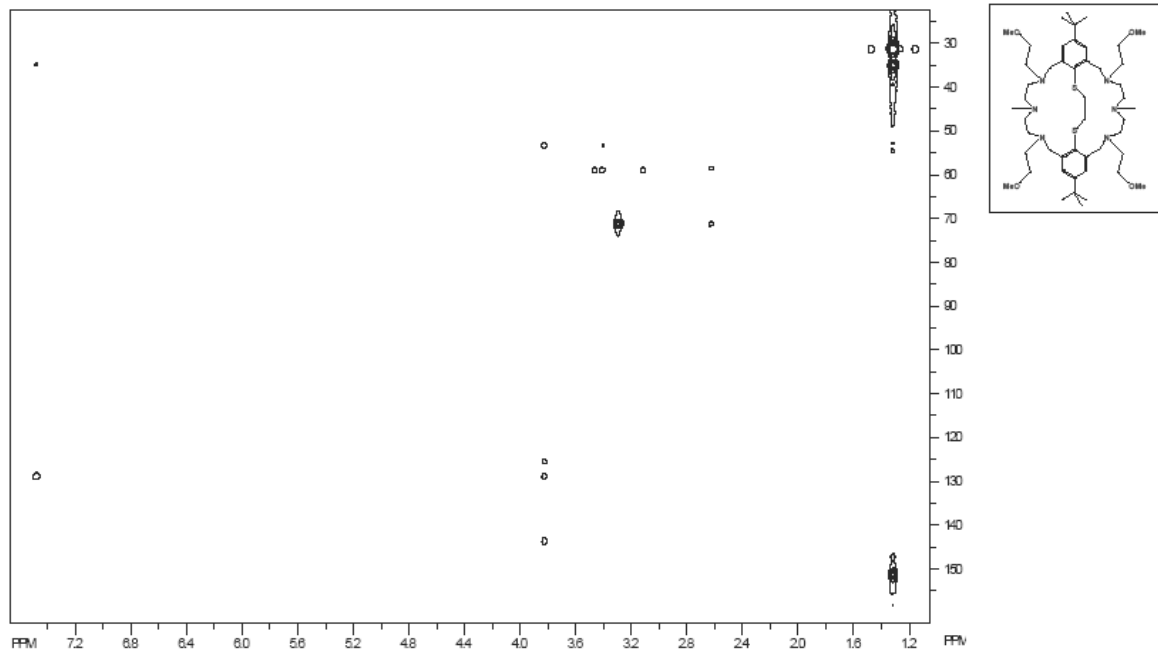
12. Spectra for tetra(2-methoxyethylated)aza-thioether **20**.



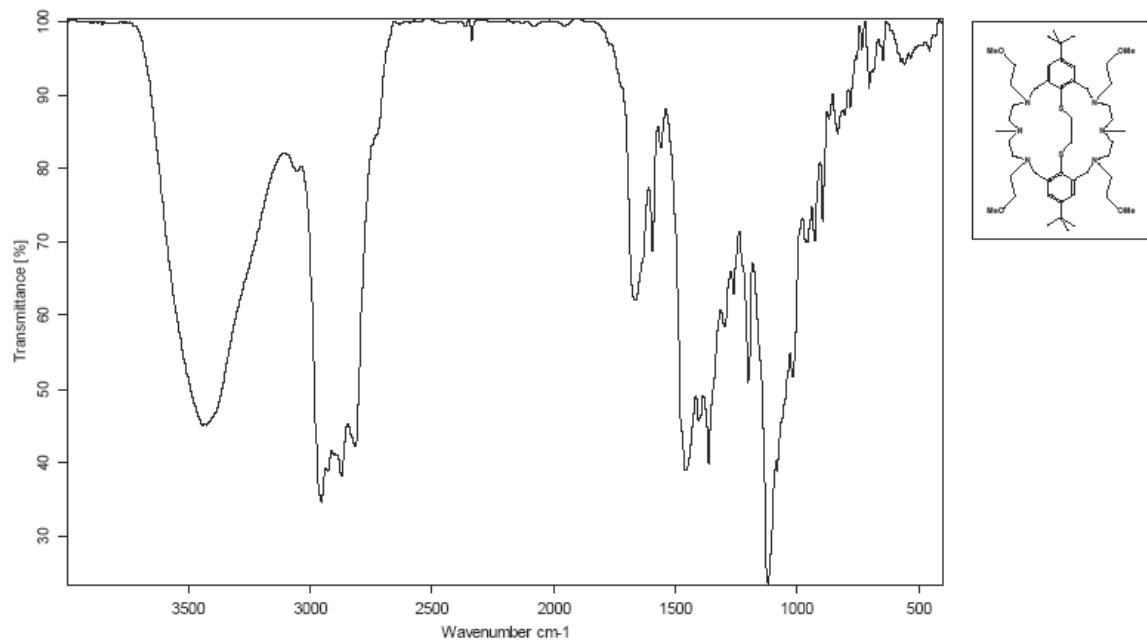
¹H/¹H- COSY

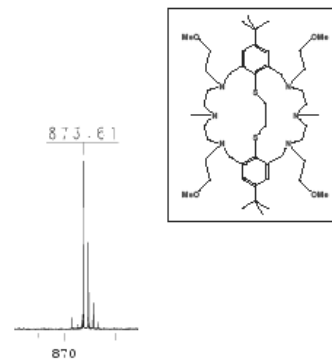
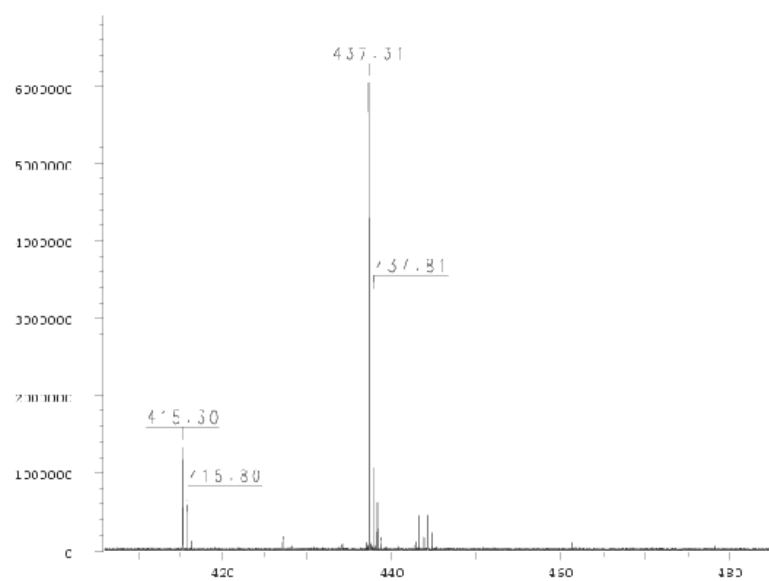


HMBC

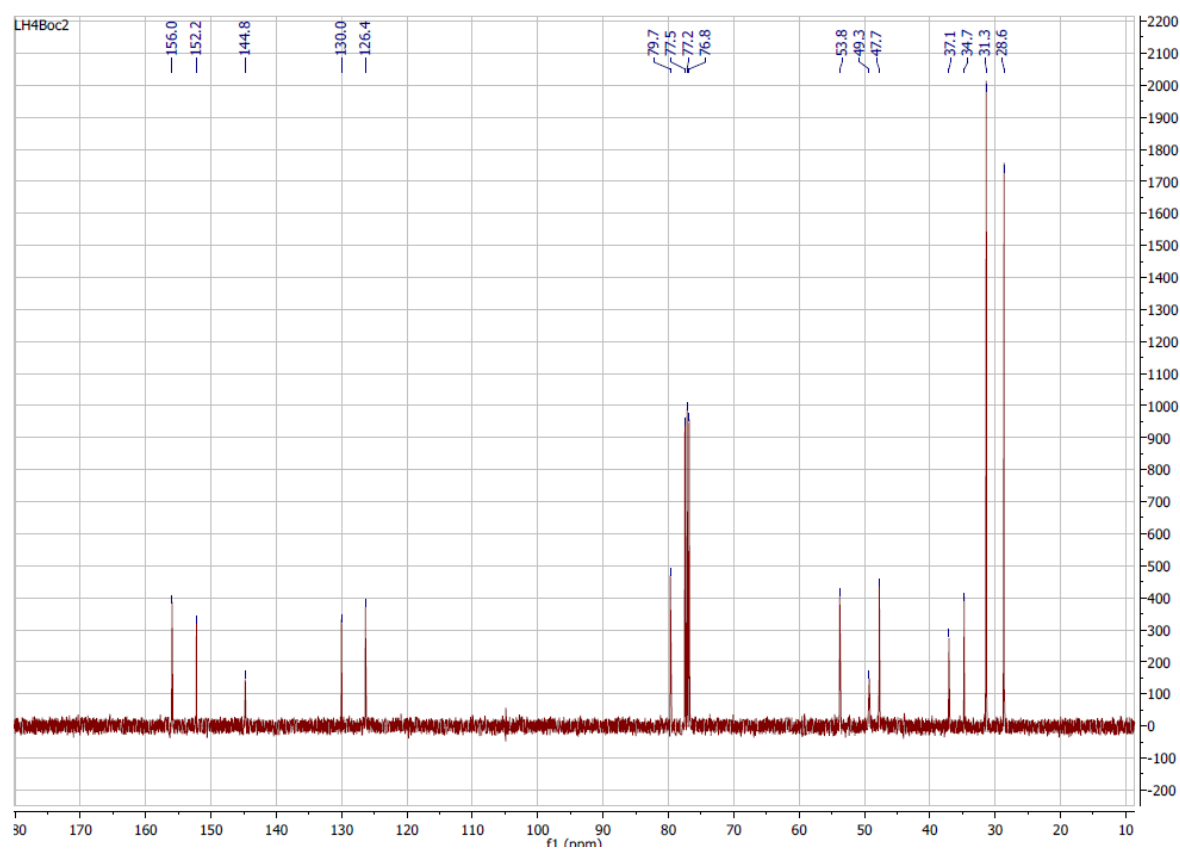
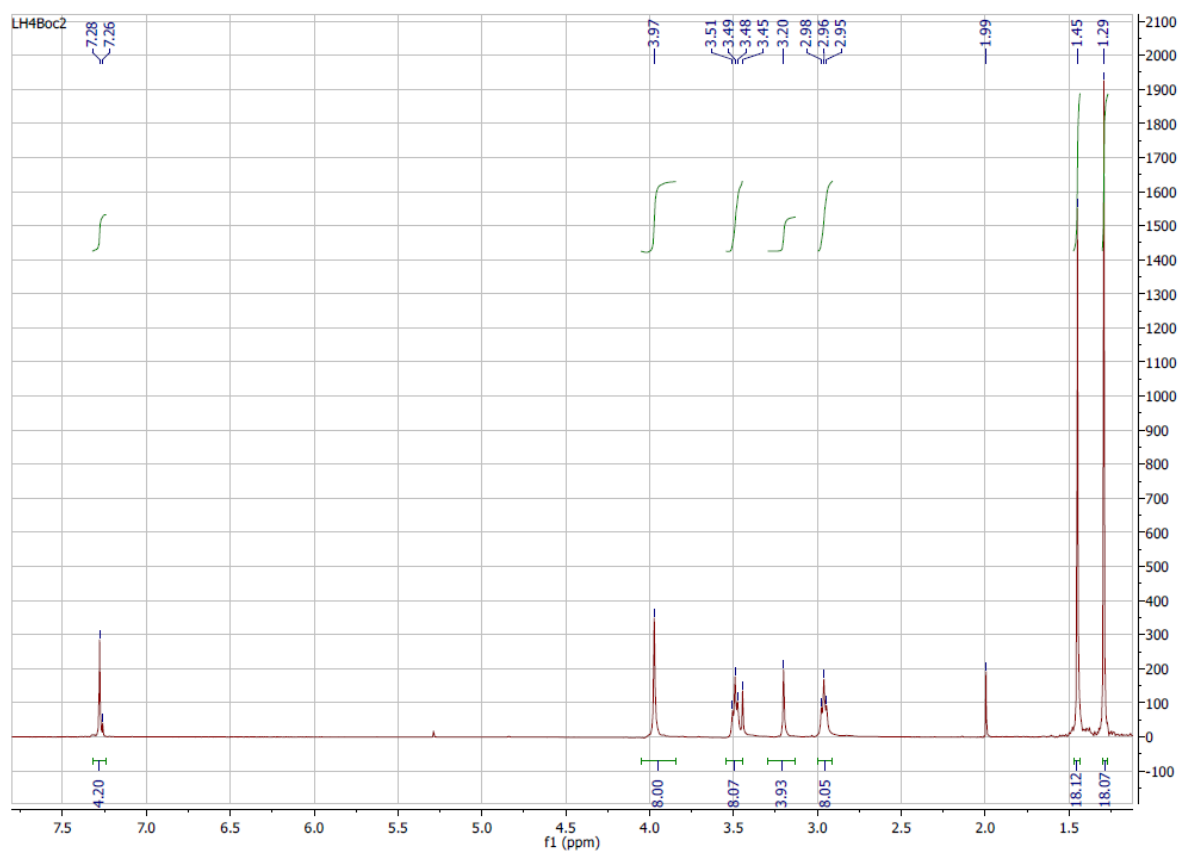


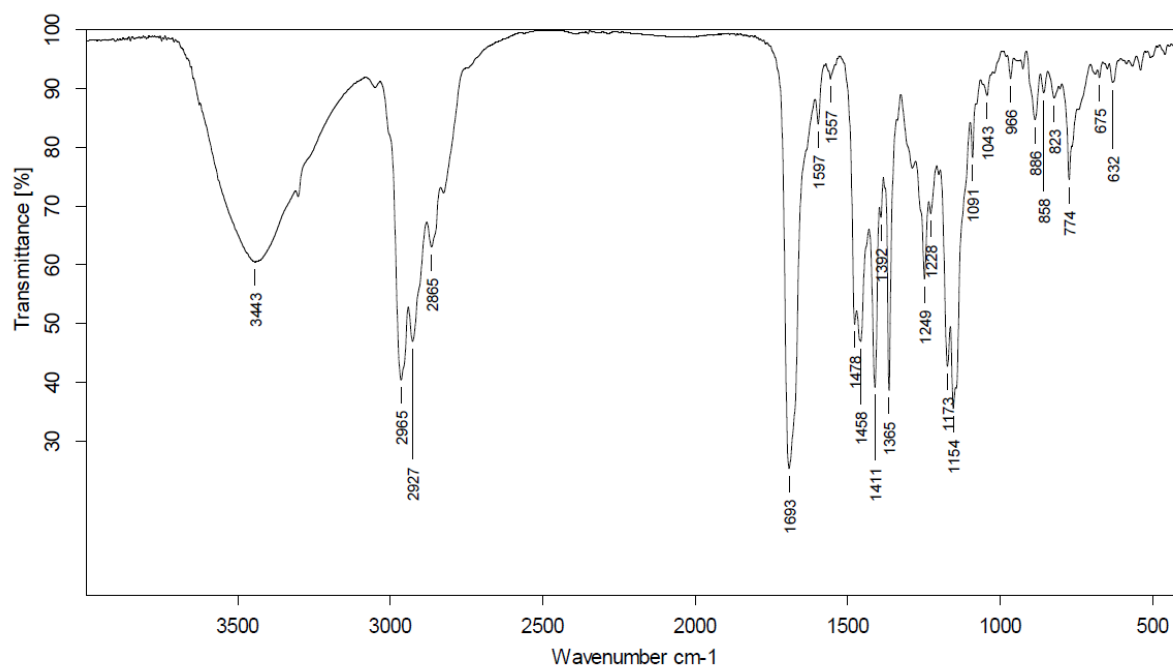
IR- Spektrum





13. Spectra for dicarbamolyated macrobicycle **22**.

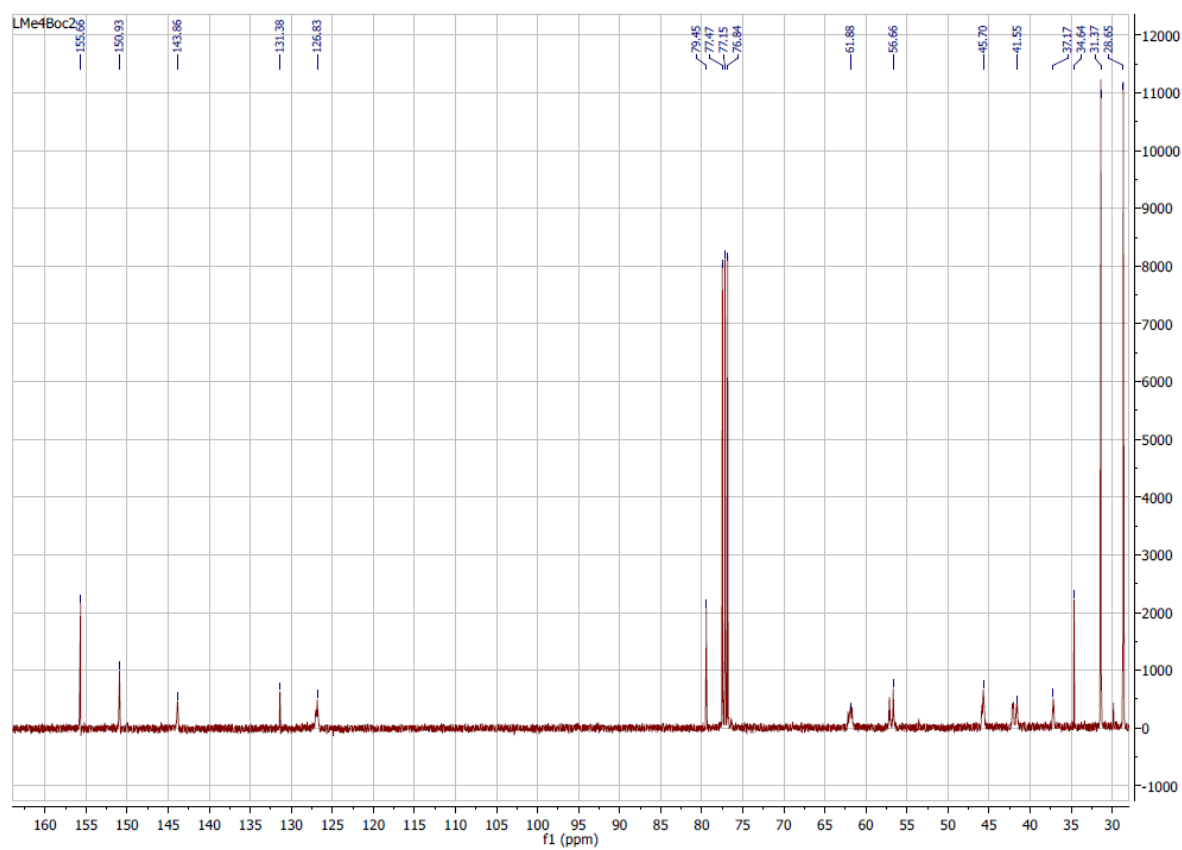
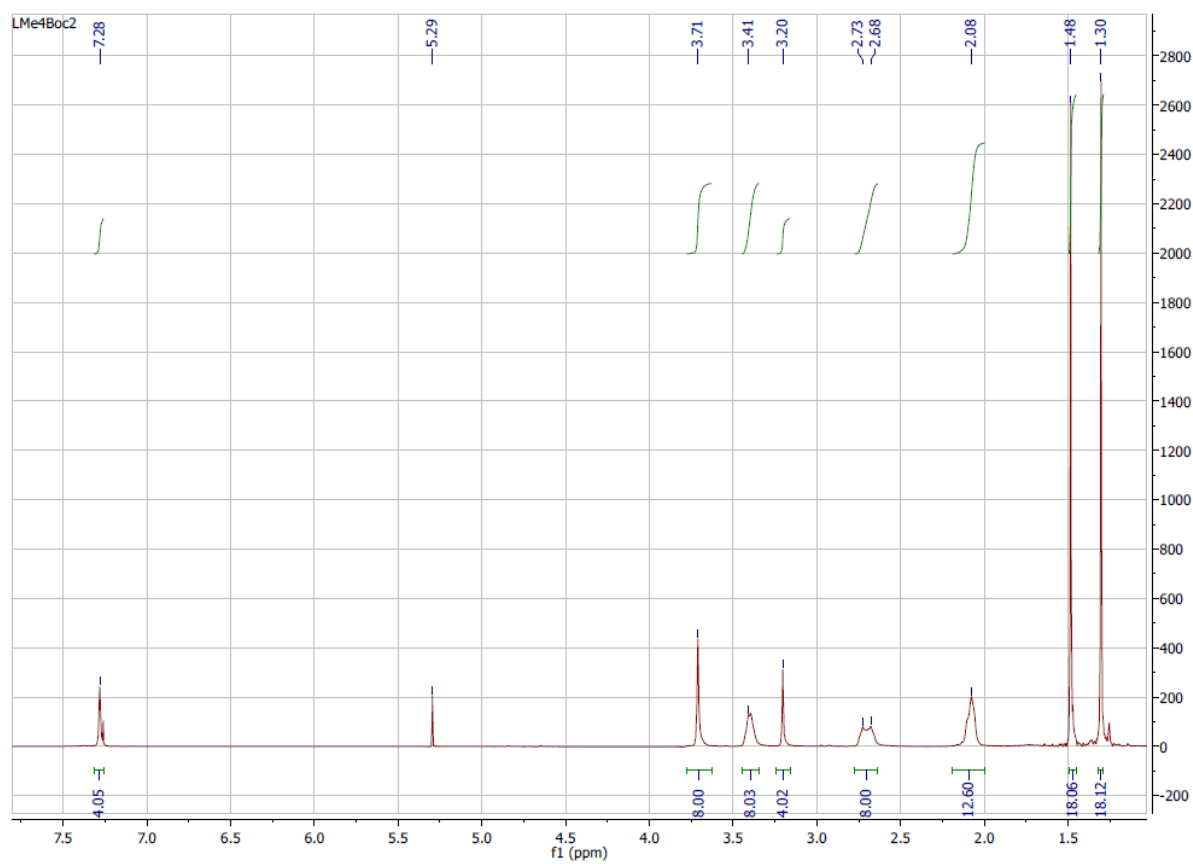


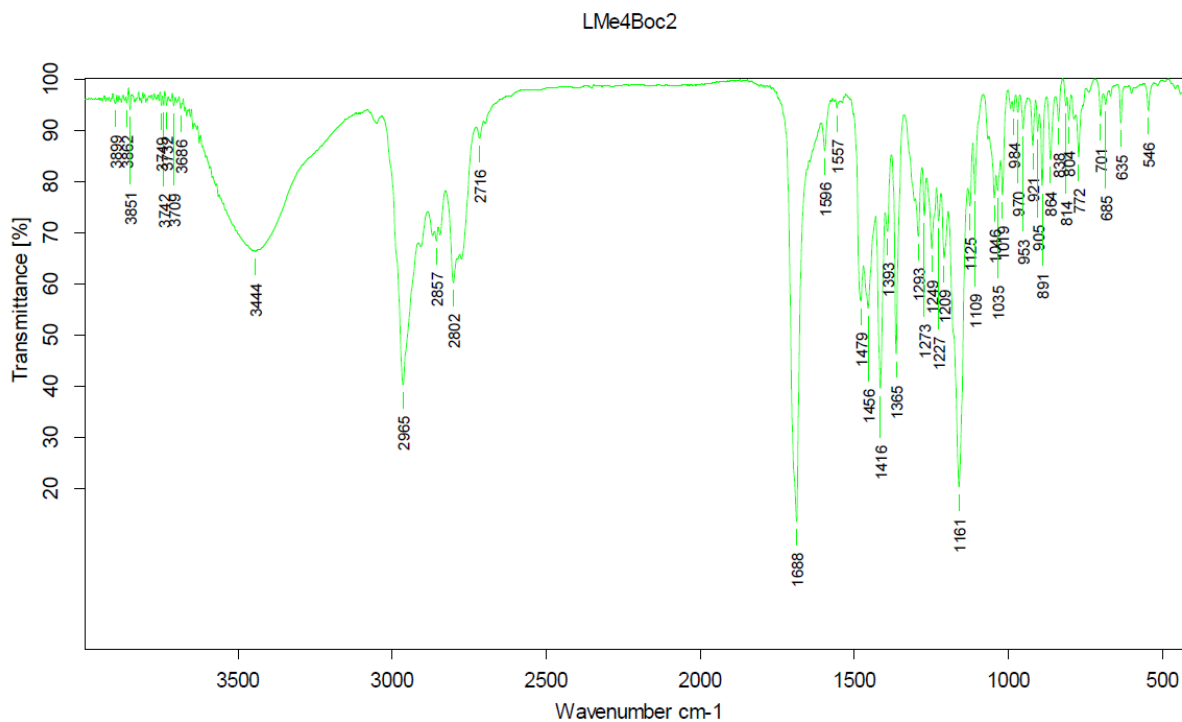


Messxperiment: Standard.XPM

Pfad: C:\IR\Kersting\Ul\BocLigand

14. Spectra for methylated macrobicycle 23.





Messxperiment: Standard.XPM

Pfad: C:\IR\Kersting\Uli\BocLigand\LMe4Boc2

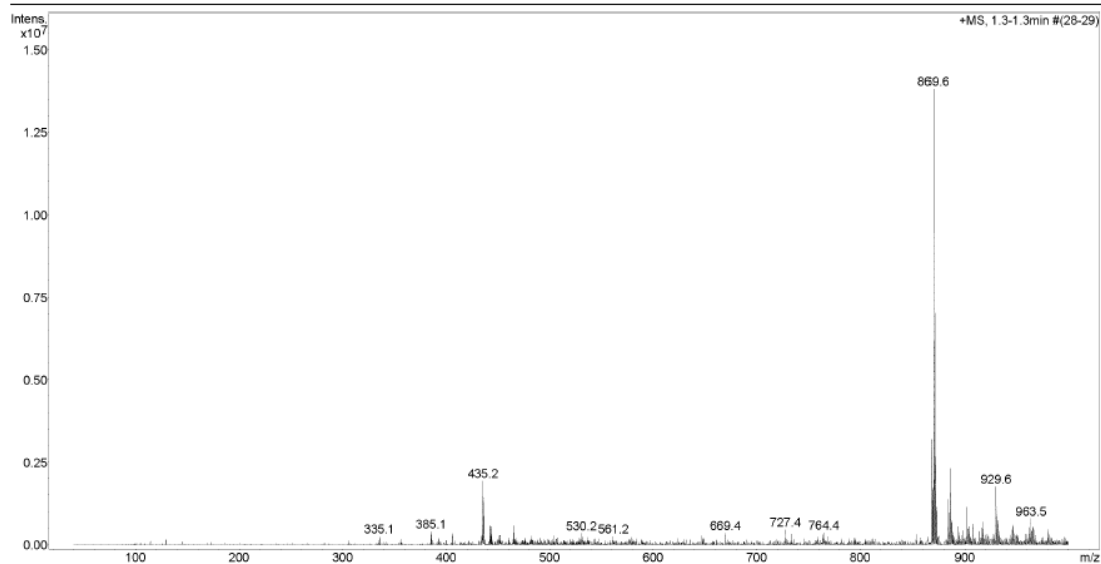
Benutzer: uli

Messung vom: 19/05/2011

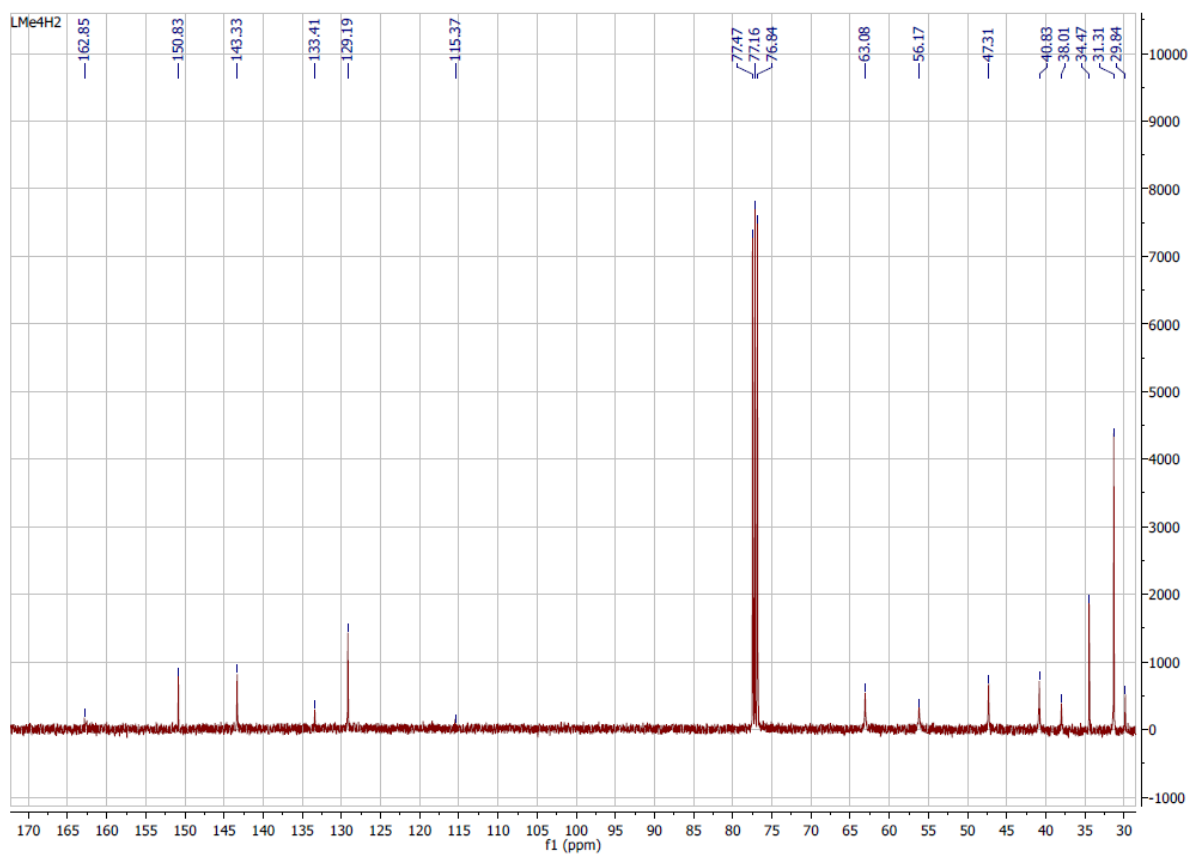
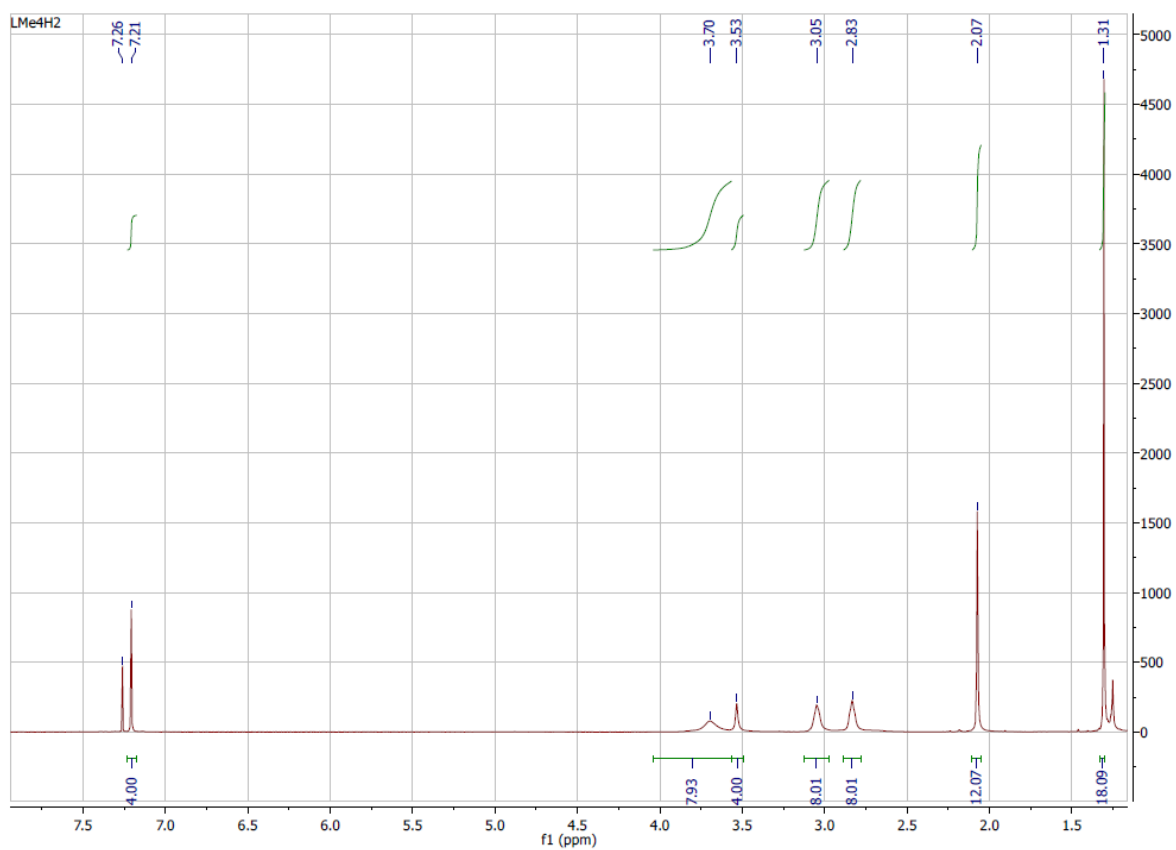
Generic Display Report

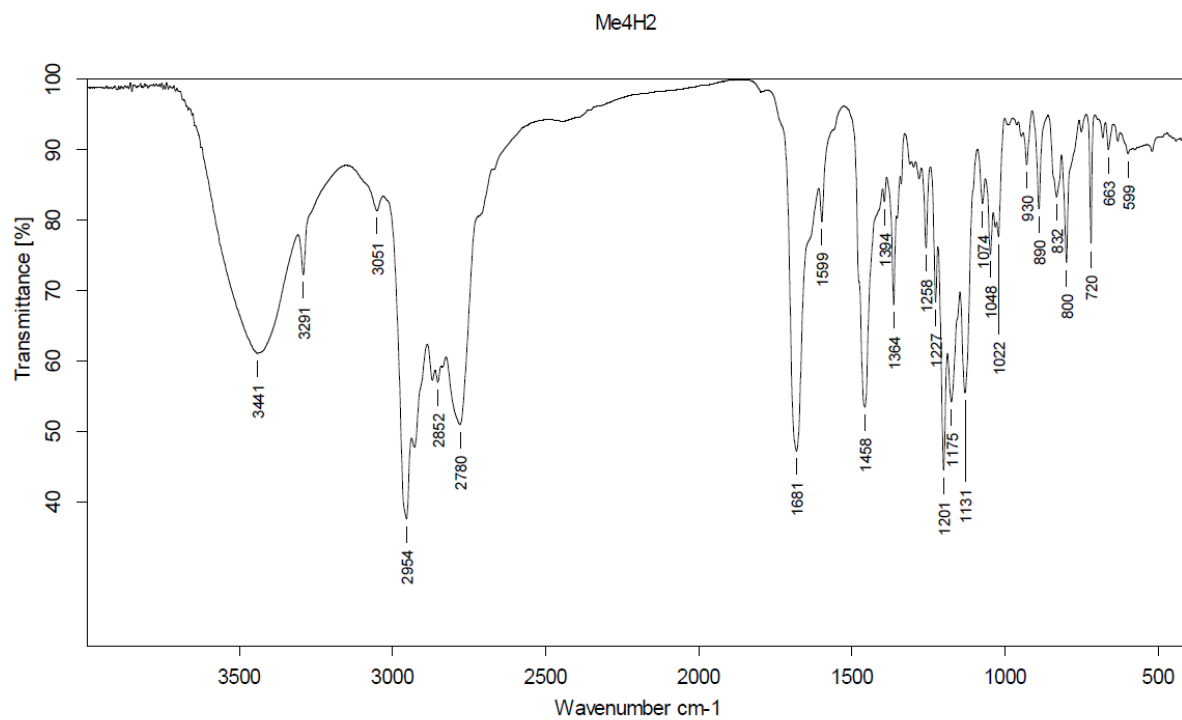
Analysis Info
 Analysis Name D:\Data\Service_Q1_10\Lehmann_MeBocMc.d
 Method service_ESI.m
 Comment Lehmann MeBocMc in MeOH + CHCl3

Acquisition Date 1/7/2010 10:38:01 AM
 Operator AK
 Instrument esquire3000 plus



15. Spectra for deprotected macrobicycle **24**.





Messxperiment: Standard.XPM

Pfad: C:\IR\Kersting\Uli\BocLigand

