Electronic Supplementary Material (ESI) for New Journal of Chemistry. This journal is © The Royal Society of Chemistry and the Centre National de la Recherche Scientifique 2016

## **New Journal of Chemistry**

## Self-regenerative and Self-enhanced Smart Graphene/Ag<sub>3</sub>PO<sub>4</sub> Hydrogel Adsorbent under Visible Light

Jie Ma<sup>1,2\*</sup>, Chunyang Chen<sup>1</sup>, Fei Yu<sup>1,2,3\*</sup>

1 State Key Laboratory of Pollution Control and Resource Reuse, School of Environmental Science and Engineering, Tongji University, 1239 Siping Road, Shanghai 200092, P. R. China. Tel: 86-21-6598 1831; E-mail: jma@tongji.edu.cn

2 Tianjin Key Laboratory of Aquatic Science and Technology, Tianjin Chengjian University, 26 Jinjing Road, Tianjin 300384, China
3 College of Chemistry and Environmental Engineering, Shanghai Institute of Technology, Shanghai 2001418, China; E-mail: fyu@yip.163.com



Fig. S1. The thermal analysis curve of graphite oxide



Fig. S2 The adsorption isotherm and models of MB on  $rGO/Ag_3PO_4$  composite





Fig. S3 The adsorption kinetics curves (a), pseudo first-order adsorption kinetic models (b) and pseudo second-order adsorption kinetic models (c) of MB on  $rGO/Ag_3PO_4$  composite

Absorbent	Adsorbate	Langmuir model			Freundlich model		
		$K_L(l/mg)$	$q_m(mg/g)$	$R^2$	$K_F$	1/n	$R^2$
rGO/Ag3PO4	MB	11.83	109.47	0.908	83.9	0.1389	0.75
composite							

Table S1. Langmuir and Freundlich isotherms parameters of rGO/Ag3PO4 composite

Table S2. Kinetic parameters of pseudo first- and second-order adsorption kinetic models for MB o	n
$rGO/Ag_3PO_4$ composite	

Dye	Initial conc.	$q_{e,exp}$	Pseudo first-order model			Pseudo second-order model		
	(mg/L)	(mg/g)	$k_l(min^{-l})$	$q_{e,cal}$	$R^2$	$k_2(min^{-1})$	$q_{e,cal}$	$R^2$
				(mg/g)			(mg/g)	
MB	50	34.85	-0.3865	76.1351	0.9494	0.0006	82.5763	0.7084