Electronic Supplementary Information

Electrocatalytic Activity of Alkyne-Functionalized AgAu Alloy Nanoparticles for Oxygen Reduction in Alkaline Media

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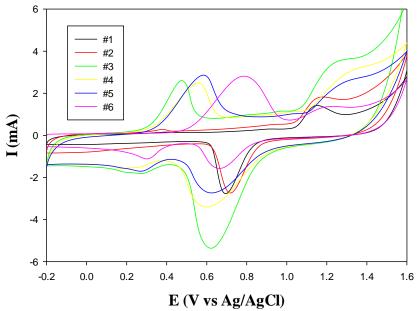


Figure S1. Cyclic voltommograms of glassy carbon electrodes modified with AgAu nanoparticles (same as in Figure 5) in 0.1 M NaOH. The effective electrochemical surface areas as summarized below were determined by oxygen adsorption on Ag (210 μ C/cm²) and Au (390 μ C/cm²), in which the Ag oxidation peaks and gold oxide reduction peaks were used to obtain the amount of charge. ^{1,2}

Samples	#1	#2	#3	#4	#5	#6
$A (cm^2)$	6.07	5.56	2.46	4.44	4.33	4.04

- (1) Trasatti, S.; Petrii, O. A. *J Electroanal Chem* **1992**, *327*, 353.
- (2) Motheo, A. J.; Machado, S. A. S.; Vankampen, M. H.; Santos, J. R. J Brazil Chem Soc 1993, 4, 122.

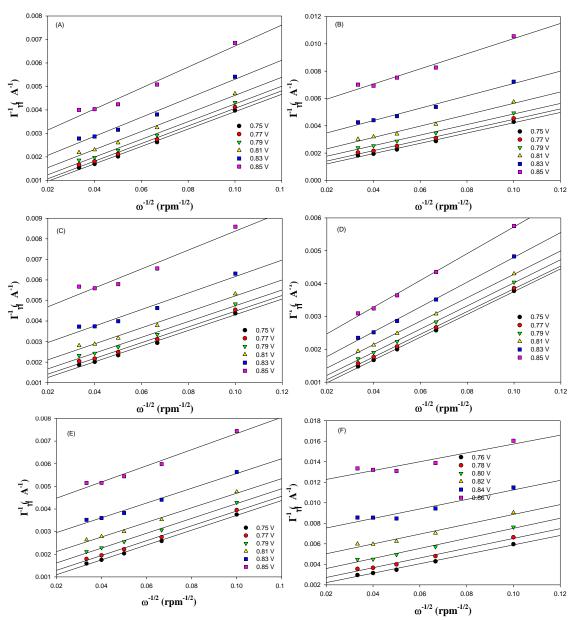


Figure S2. Koutecky-Levich plots of the six AgAu nanoparticles in oxygen-saturated 0.1 M KOH at varied rotation rates (specified in figure legends). Symbols are experimental data from Figure 5 and lines are linear regressions.