#### Supporting information for

# Thoils as interfacial modifiers to enhance the performance and stability of perovskite solar cells

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## **Experimental Section**

### Materials synthesis

All chemicals and reagents were used as received from chemical companies without any further purification. The CH<sub>3</sub>NH<sub>3</sub>I was synthesized and purified according to the literature method.<sup>1</sup> To a stirred solution of methylamine in methanol (40 wt.%, 24 mL) was slowly added aqueous hydroiodic acid (57 wt.%, 25 mL) at 0 °C. After 2 h, the precipitate was collected by evaporation at 50 °C for 1 h. The as-obtained product was washed with diethyl ether three times and then finally dried at 60 °C in a vacuum oven for 24 h to afford the desired pure CH<sub>3</sub>NH<sub>3</sub>I as white crystals.

### **Solar Cell Fabrication**

Fluorine-doped Tin Oxide (FTO) glass substrates with dimension of 2.0 cm  $\times$  2.0 cm were patterned by etching with zinc powder and 2 M hydrochloric acid. The substrates were then sequentially washed in ultrasonic baths of acetone, distilled water and ethanol. A compact TiO<sub>2</sub> blocking layer was spin-coated onto the cleaned FTO glass using 0.15 M Titanium tetraisopropanolate in 1-butanol solution at 2000 rpm for 30 s. The substrate was heated at 120 °C for 15 min, and then annealed at 550 °C for 30 min. After cooling to the room temperature, the film was immersed into the 20 mM TiCl<sub>4</sub> solution at 70 °C for 30 min. After dried, a ~200 nm thick mesoporous TiO<sub>2</sub> film was deposited on the pre-treated FTO substrate by spin-coating of the TiO<sub>2</sub> paste (Dyesol DSL 18NR-T) with ethanol (1:3, mass ratio), which was followed by the heating at 550 °C for 30 min. The porous substrate was dipped in 2 mM solution of HOOC-R-SH in ethanol for 30 min for introducing anchor motifs on the titania film, the mesoporous TiO<sub>2</sub> substrate was washed thoroughly by ethanol three times to remove physically absorbed HOOC-R-SH ligands. For the perovskite layer, the as-prepared substrate was infiltrated with a hot solution of PbI<sub>2</sub> in *N*, *N*-dimethylformamide (462 mg/1 mL) by spin-coating at 5000 rpm. for 30 s, and then dried at 120 °C for 5 min. After cooling to room temperature, the film was dipped into the solution of CH<sub>3</sub>NH<sub>3</sub>I in 2-propanol (10 mg/mL) for 30 min, which was spun at 3000 rpm for 20 s and then heated at 100 °C for 5 min. After the deposition of the perovskite layer, the substrate was dipped into a 100 mg/mL solution of HS-R' in CH<sub>2</sub>Cl<sub>2</sub> for 30 s, followed by baking at 100 °C for 5 min. After that, the hole transport layer solution were coating via solution process at 4000 rpm for 30 s, where Spiro-OMeTAD/ chlorobenzene (90 mg/1 mL) solution was employed with the additives containing 17.5 µL Li-TFSI)/acetonitrile (520 mg/1 mL) and 28.8 µl TBP. Finally, a 80 nm thick Au counter electrode was deposited by thermal evaporation under reduced pressure of  $2 \times 10^{-7}$  Torr. The active area was 0.12 cm<sup>2</sup>.

#### **Device Characterization**

Current-voltage characteristics were recorded from a solar simulator equipped with a Keithley 2400 source meter and 300 W collimated Xenon lamp (Newport) calibrated with the light intensity to 100 mW cm<sup>-2</sup> at AM 1.5 G solar light condition by the certified silicon solar cell. Incident photon-to-electron conversion efficiency (IPCE) was measured on a computer-controlled IPCE system (Newport) containing a Xenon lamp, a monochromator and a Keithley multimeter. The system was calibrated with the certified silicon solar cell and the IPCE data were collected at DC mode. XRD patterns were analyzed by an X-ray diffractometer (Rigaku, RINT-2500) with a CuKa radiation source. The surface morphology of were recorded via a SEM-4800 field-emission scanning electron microscope (SEM). The UV-vis spectra were measured with the perovskite infiltrated mesoscopic TiO<sub>2</sub> films supported by FTO glass using a cary-5000 UV-vis spectrophotometer. The Time-resolved photoluminescence (PL) spectra were measured using an Edinburgh Instruments FLS920 spectrometer.

#### Reference

1. J.-H. Im, C.-R. Lee, J.-W. Lee, S.-W. Park and N.-G. Park, Nanoscale, 2011, 3, 4088-4093.

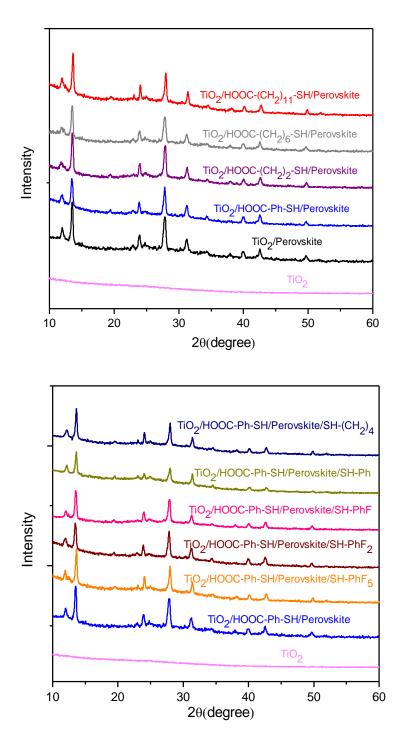


Fig. S1 The XRD spectra of the films.

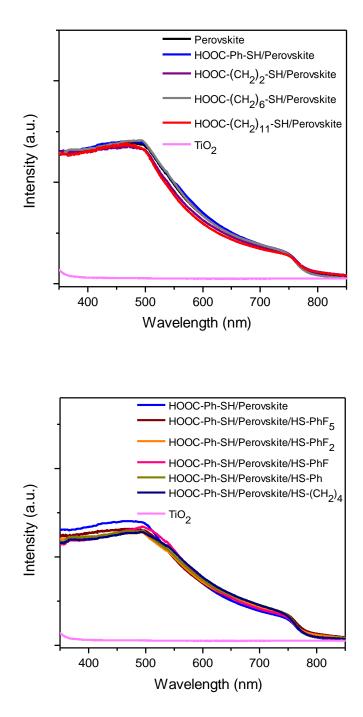
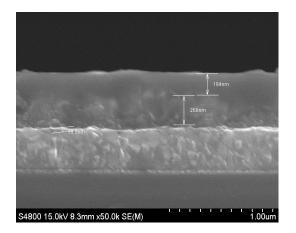
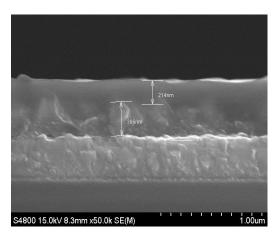


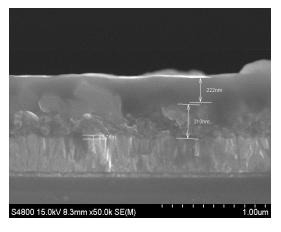
Fig. S2 The UV-vis spectra of the films.



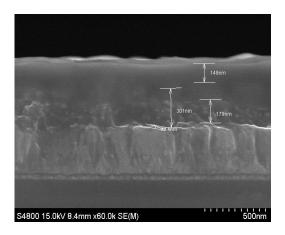
 $TiO_2/Perovskite/Spiro-OMeTAD$ 



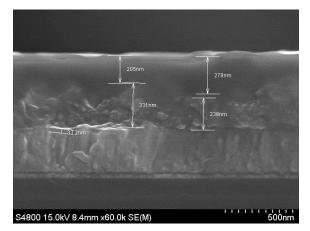
TiO2/HOOC-Ph-SH/perovskite/Spiro-OMeTAD



TiO<sub>2</sub>/HOOC-(CH<sub>2</sub>)<sub>2</sub>-SH/perovskite/Spiro-OMeTAD

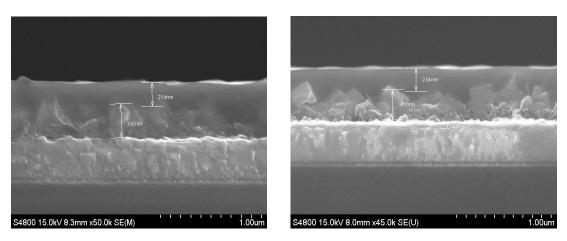


TiO<sub>2</sub>/HOOC-(CH<sub>2</sub>)<sub>6</sub>-SH/perovskite/Spiro-OMeTAD



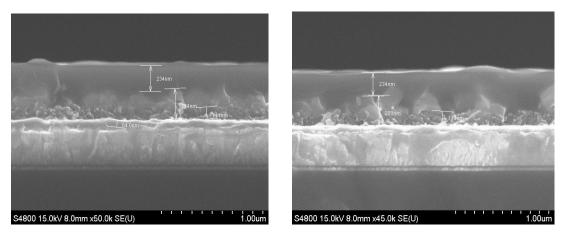
 $TiO_2/HOOC\text{-}(CH_2)_{11}\text{-}SH/perovskite/Spiro-OMeTAD$ 

Fig. S3 Cross-sectional SEM images of the photovoltaic devices.

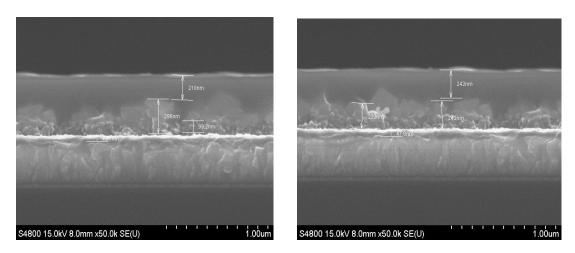


TiO2/HOOC-Ph-SH/perovskite/Spiro-OMeTAD

TiO<sub>2</sub>/HOOC-Ph-SH/perovskite/HS-PhF<sub>5</sub>/Spiro-OMeTAD



 $TiO_2/HOOC-Ph-SH/perovskite/HS-PhF_2/Spiro-OMeTAD \ TiO_2/HOOC-Ph-SH/perovskite/HS-PhF/Spiro-OMeTAD \ TiO_2/HOOC-Ph-SH/perovskite/HS-PhF/Spiro-P$ 



 $TiO_2/HOOC-Ph-SH/perovskite/HS-Ph/Spiro-OMeTAD TiO_2/HOOC-Ph-SH/perovskite/HS-(CH_2)_4/Spiro-OMeTAD TiO_2/HOOC-Ph-SH/po-SH/PADA TIO_2/HOOC-Ph-SH/PADA TIO_2/HOOC-Ph-SH/PADA TIO_2/HOOC-Ph-SH/PADA TIO_2/HOOC-Ph-SH/PADA TIO_2/HOOC-Ph-SH/PADA TIO_2/HOOC-Ph-SH/PADA TIO_2/HOOC-Ph-SH/PADA TIO_2/HOOC-Ph-SH/PADA TIO_2/HOOC-Ph-SH/PADA TIO_2/HOOC-Ph-SH/PADA$ 

Fig. S4 Cross-sectional SEM images of the photovoltaic devices.

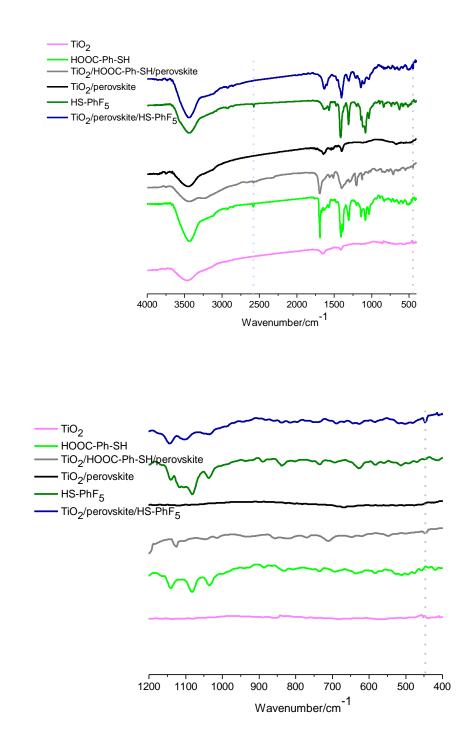


Fig. S5 The FTIR spectra of the films with different components and modifications.

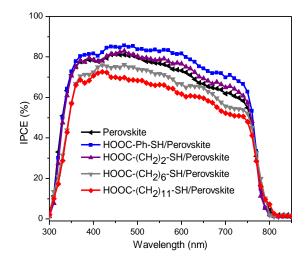
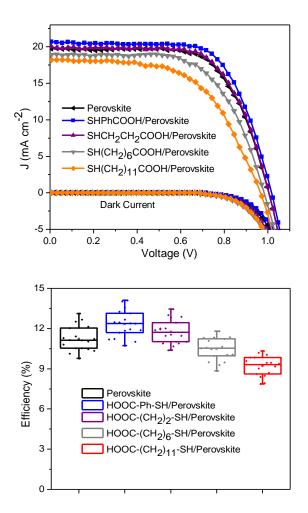
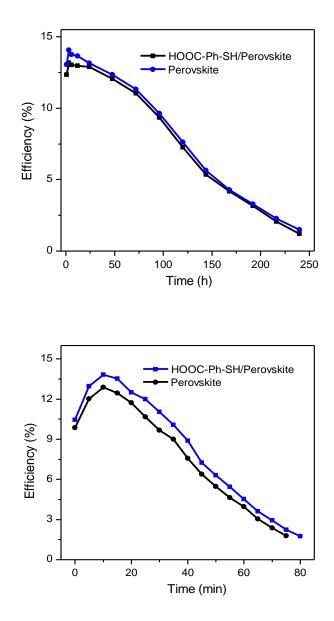


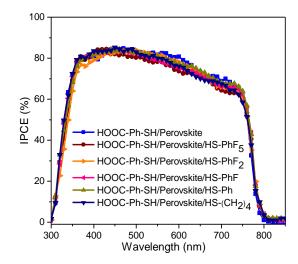
Fig. S6 The IPCE spectra of the devices.

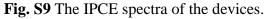


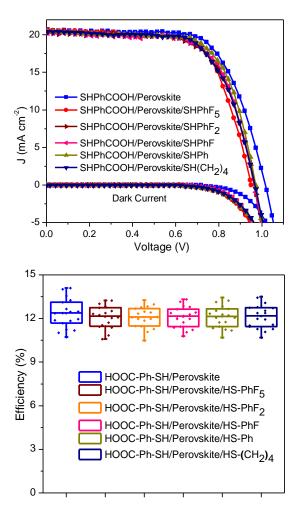
**Fig. S7** The best I-V characteristics (top), comparison of the performance distributions of 18 individual devices (bottom) of perovskite and HOOC-R-SH/perovskite solar cells.



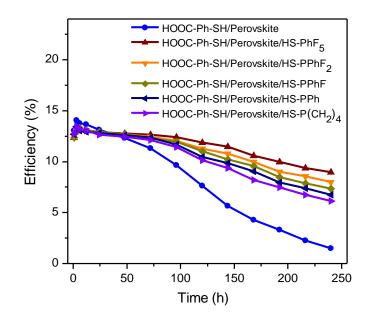
**Fig. S8** The efficiency variation of the devices stored in air at room temperature with the humidity of 45% (top) and under illumination at AM 1.5 G (bottom).







**Fig. S10** The best I-V characteristics (top), comparison of the performance distributions of 18 individual devices (bottom) of HOOC-Ph-SH/perovskite and HOOC-Ph-SH/perovskite/HS-R'.



**Fig. S11** The efficiency variation of the devices stored in air at room temperature with a humidity of about 45%.

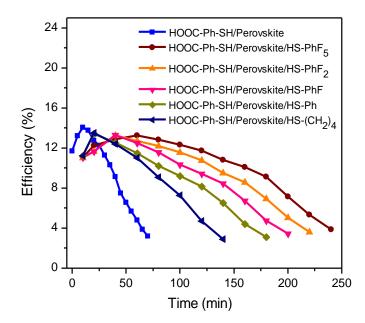


Fig. S12 The variation of the device efficiency under illumination at AM 1.5 G.

Dye	$J_{sc}/\mathrm{mA}\cdot\mathrm{cm}^{-2}$	$V_{oc}/V$	<i>FF</i> /%	η/%	$Rs/\Omega\cdot cm^{-2}$	$Rsh/\Omega\cdot cm^{-2}$
Perovskite	19.81	1.00	66.43	13.18	8.95	1021.10
HOOC-Ph-SH/Perovskite	20.66	1.02	66.54	14.11	8.49	923.10
HOOC-(CH <sub>2</sub> ) <sub>2</sub> -SH/Perovskite	20.08	1.01	66.38	13.45	8.64	1060.90
HOOC-(CH <sub>2</sub> ) <sub>6</sub> -SH/Perovskite	19.04	0.98	63.18	11.80	9.77	1833.17
HOOC-(CH <sub>2</sub> ) <sub>11</sub> -SH/Perovskite	18.28	0.97	57.71	10.32	11.59	734.01
HOOC-Ph-SH/Perovskite/HS-PhF5	20.66	0.97	65.79	13.21	10.08	639.26
HOOC-Ph-SH/Perovskite/HS-PhF <sub>2</sub>	20.58	0.97	65.85	13.22	9.18	733.38
HOOC-Ph-SH/Perovskite/HS-PhF	20.56	0.97	65.97	13.28	8.89	723.46
HOOC-Ph-SH/Perovskite/HS-Ph	20.56	0.97	66.07	13.30	9.54	493.05
HOOC-Ph-SH/Perovskite/HS-(CH <sub>2</sub> ) <sub>4</sub>	20.58	0.97	66.11	13.32	9.39	996.53

**Table S1.** Summary of photovoltaic parameters of cells with different modifications.