

# Persistent Luminescence from $\text{Eu}^{3+}$ in $\text{SnO}_2$ Nanoparticles

## - Supporting Information

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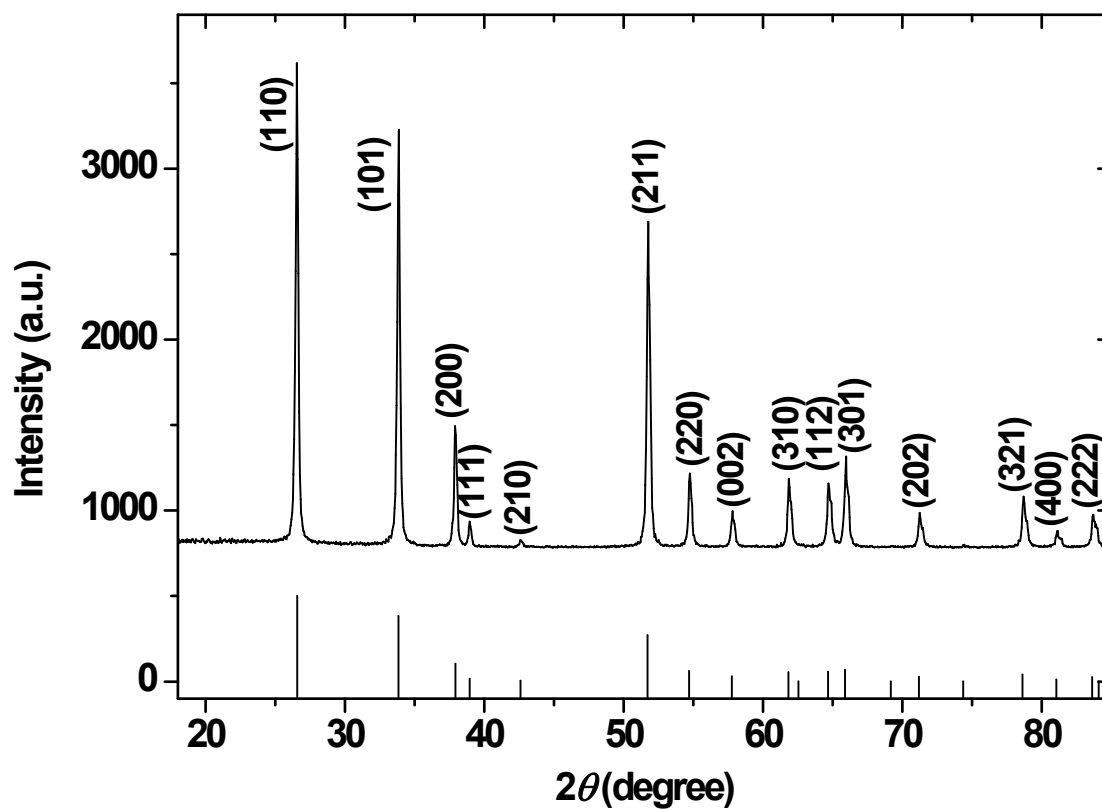
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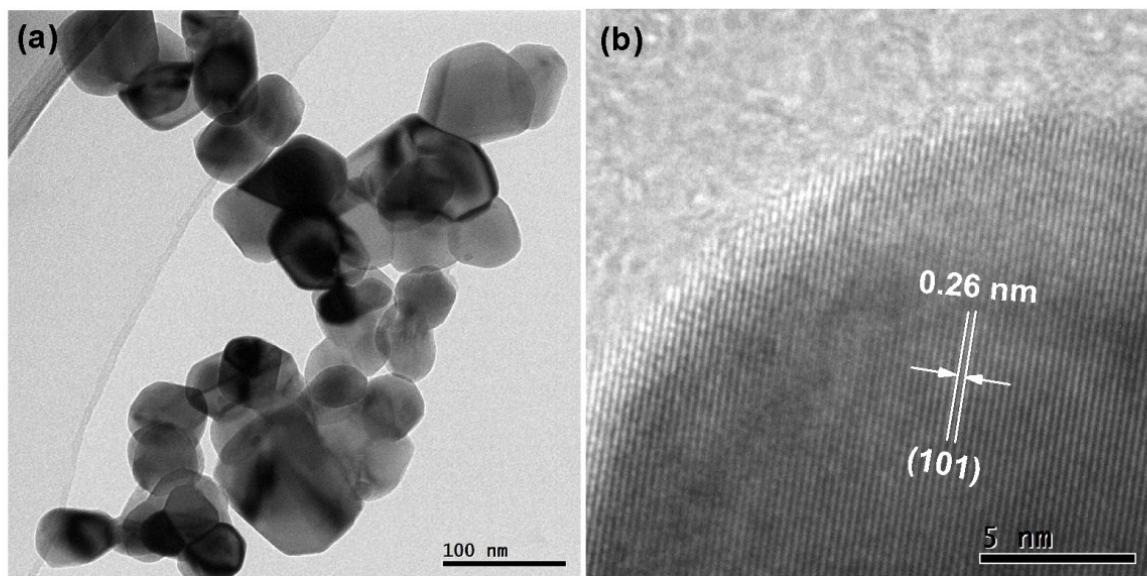
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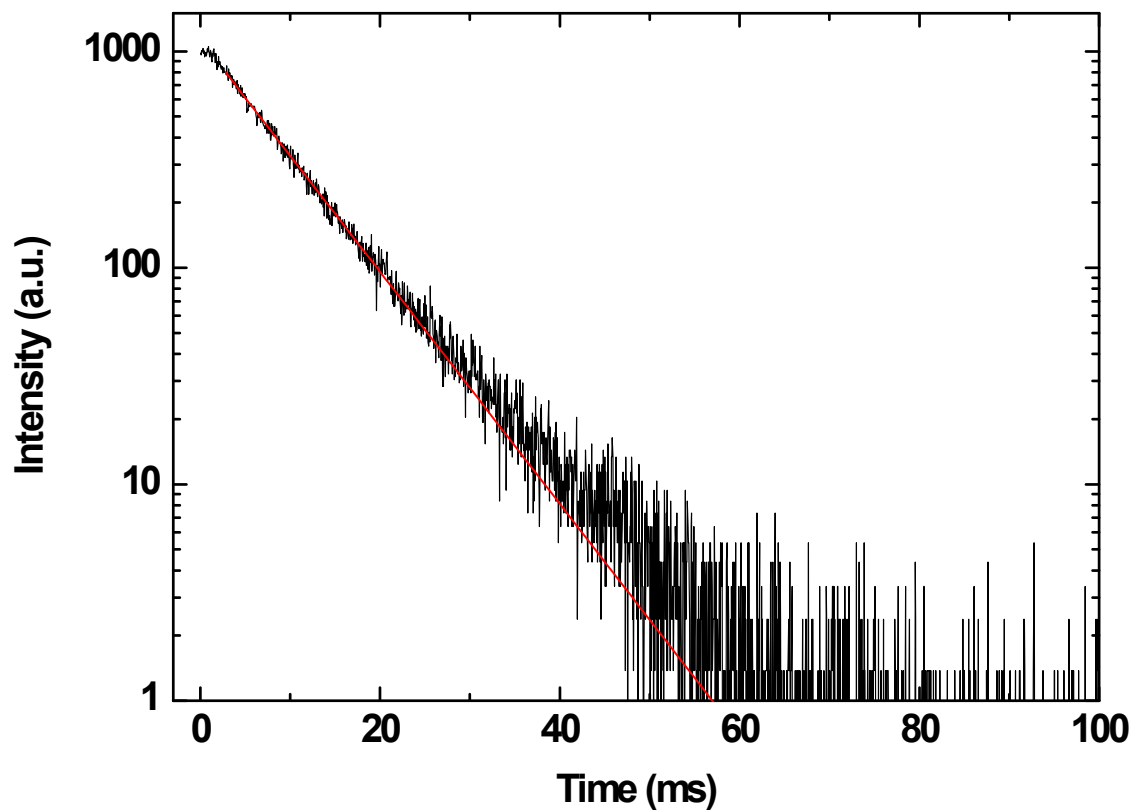
University of Chinese Academy of Sciences, Beijing 100049, China.



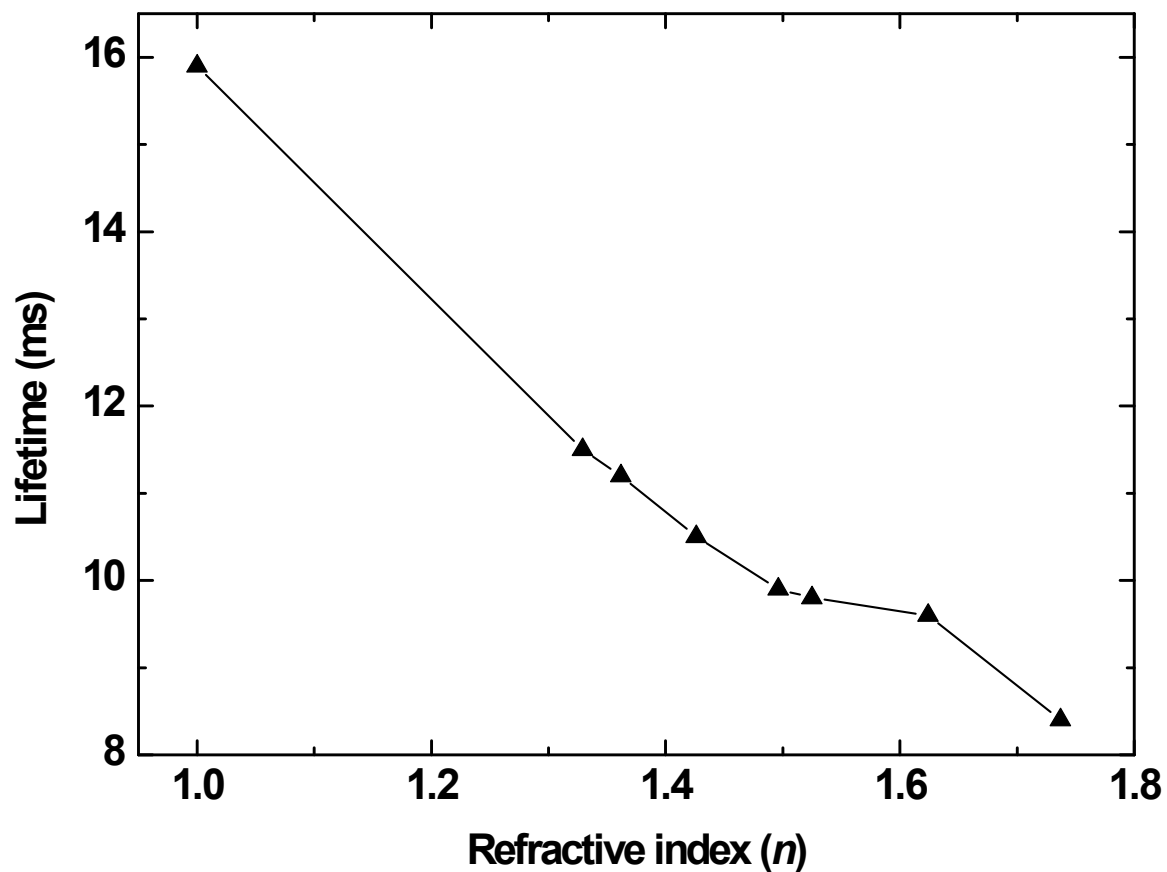
**Figure S1.** XRD pattern of SnO<sub>2</sub>:Eu<sup>3+</sup> nanoparticles (NPs). The bars at the bottom represent the standard diffraction lines of rutile-phase SnO<sub>2</sub> (JCPDS No. 77-0449).



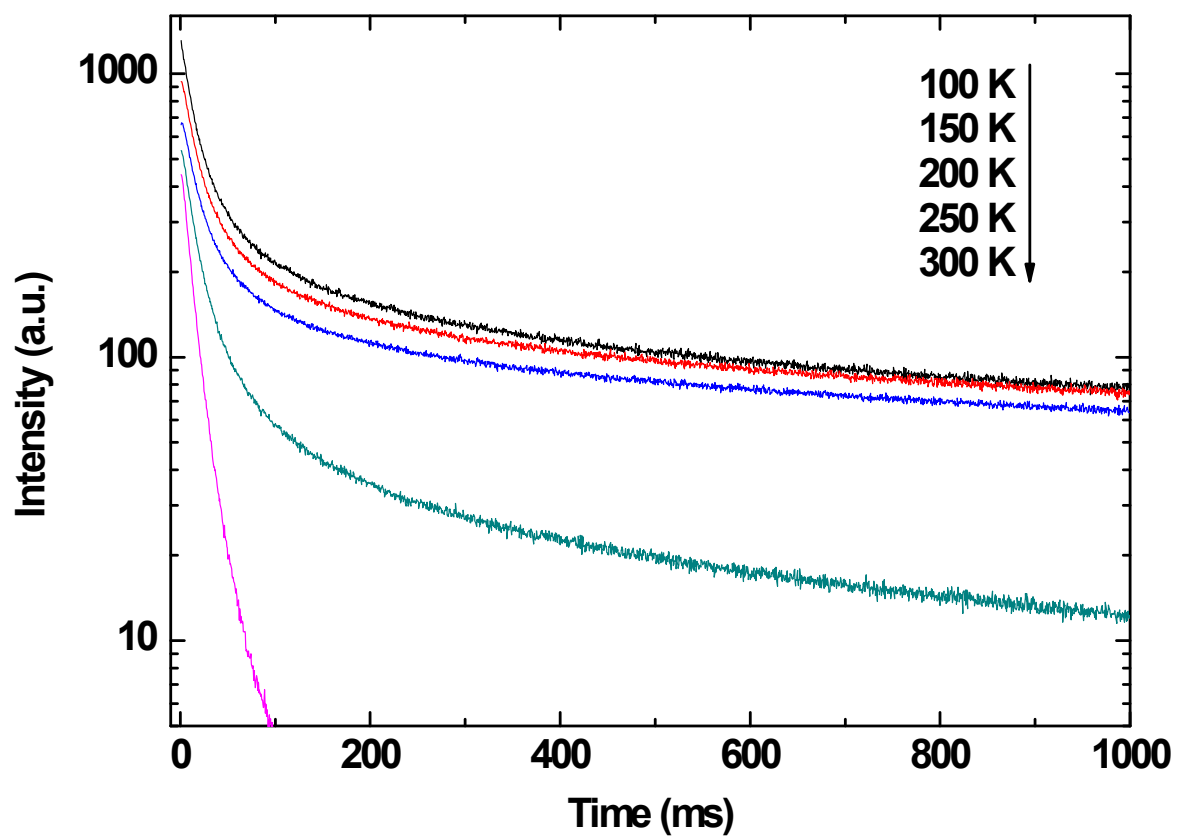
**Figure S2.** (a) TEM and (b) HRTEM images of  $\text{SnO}_2:\text{Eu}^{3+}$  NPs. The HRTEM image shows clear lattice fringes for an individual NP, with an observed  $d$  spacing of 0.26 nm, which is in good agreement with the lattice spacing of the (101) plane of rutile  $\text{SnO}_2$ .



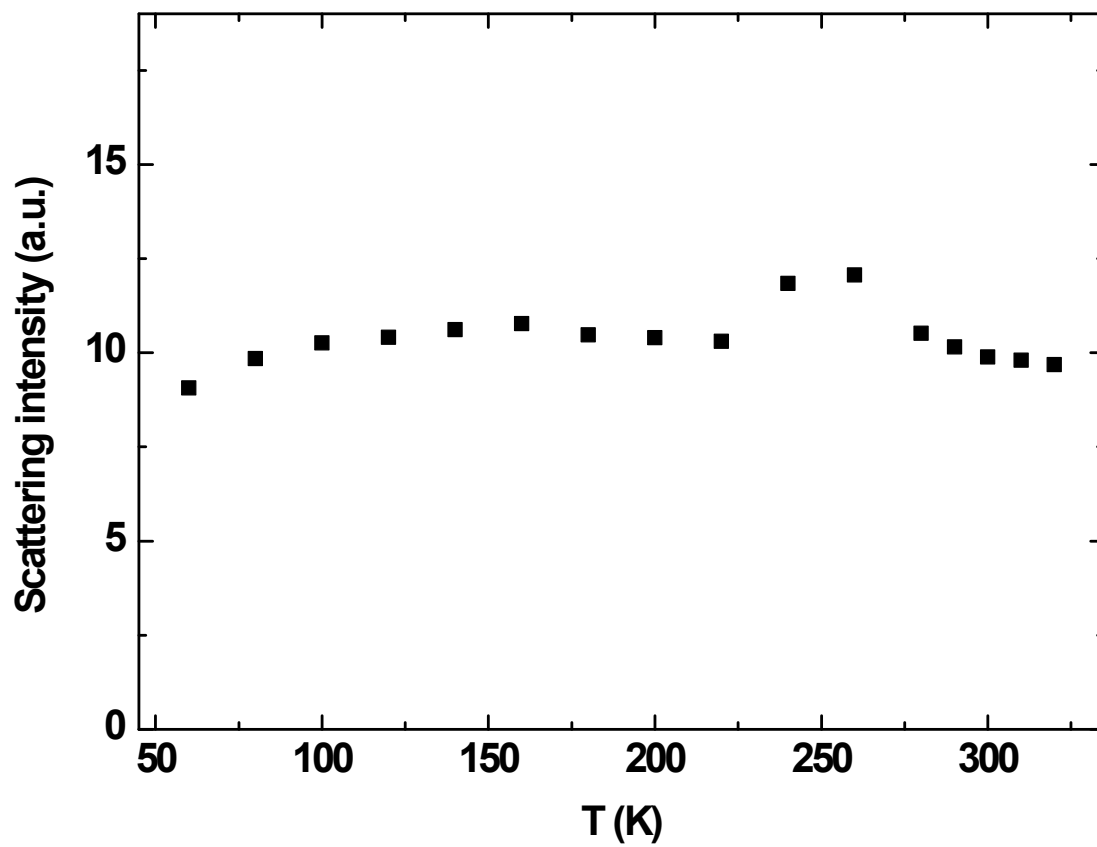
**Figure S3.** PL decay curve of SnO<sub>2</sub>:Eu<sup>3+</sup> microparticles (MPs) by monitoring the Eu<sup>3+</sup> emission at 588.0 nm at 300 K. By fitting the decay curve with a single-exponential function (red line), the <sup>5</sup>D<sub>0</sub> lifetime of Eu<sup>3+</sup> was determined to be 8.1 ms.



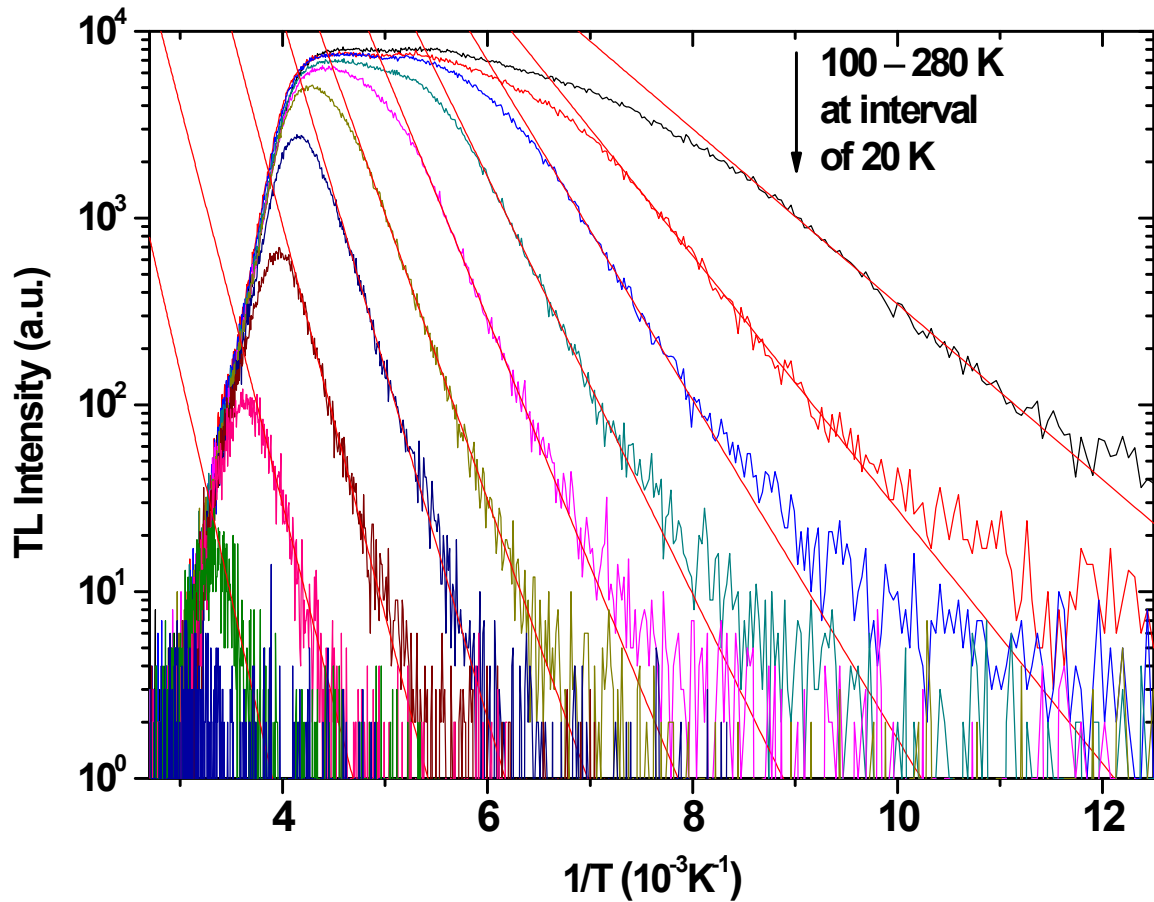
**Figure S4.** Dependence of the  $^5D_0$  PL lifetime of  $\text{Eu}^{3+}$  in  $\text{SnO}_2$  NPs on the refractive index ( $n$ ) of the surrounding media. The media are, in turn, air ( $n = 1$ ), methanol (1.329), ethanol (1.362), cyclohexane (1.426), toluene (1.496), chlorobenzene (1.525), carbon disulfide (1.624), and diiodomethane (1.737).



**Figure S5.** PL decay curves of SnO<sub>2</sub>:Eu<sup>3+</sup> MPs in the temperature range of 100-300 K. The long-lasting decay tails at temperatures below 250 K indicate the persistent luminescence feature of SnO<sub>2</sub>:Eu<sup>3+</sup> MPs.

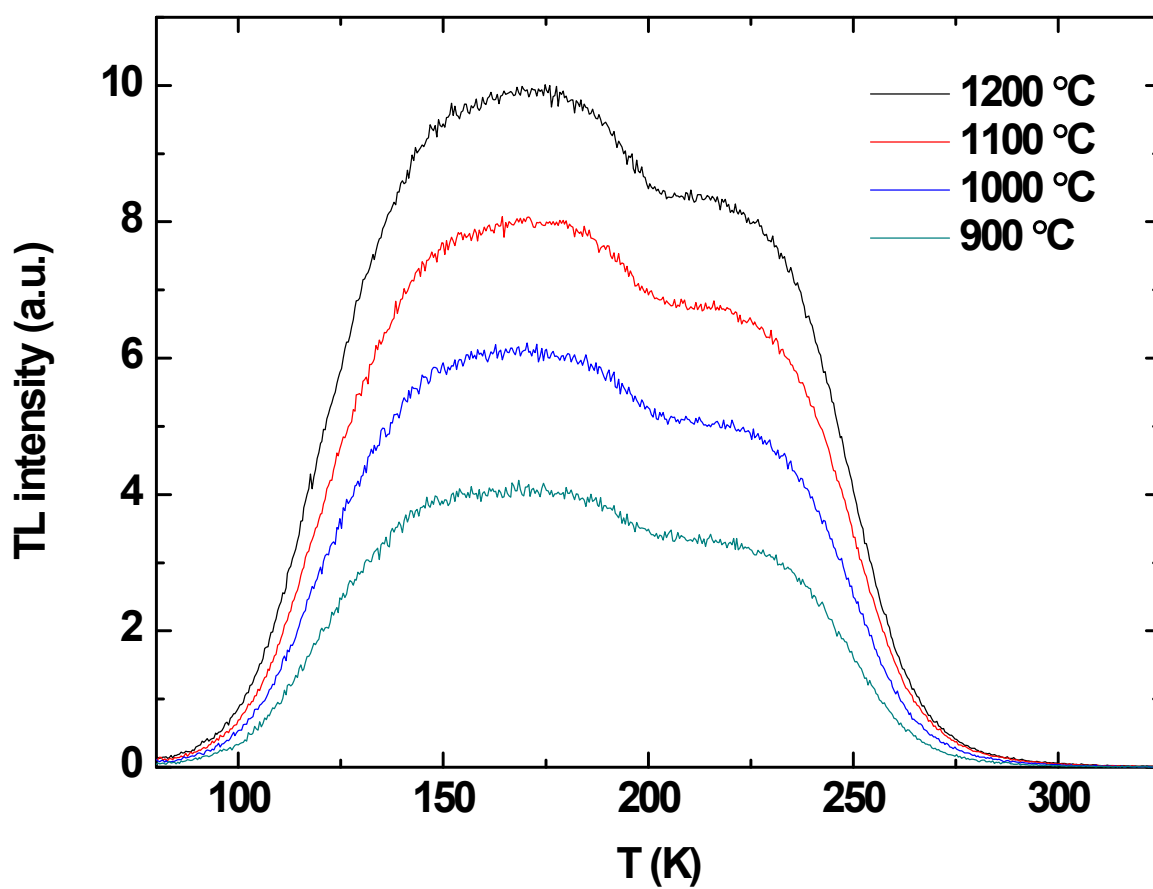


**Figure S6.** The scattering of SnO<sub>2</sub>:Eu<sup>3+</sup> NPs to the excitation light at 300 nm as a function of temperature.

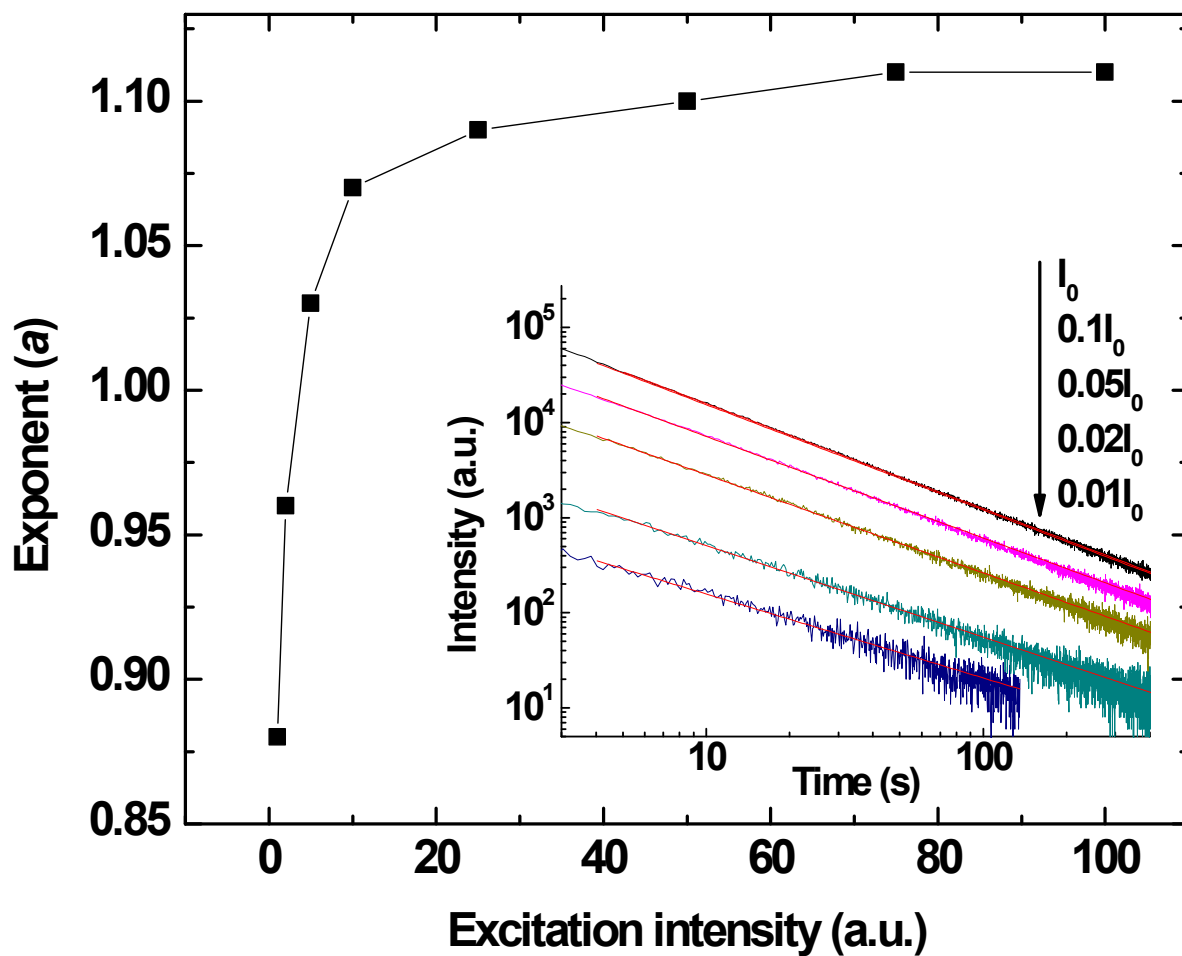


**Figure S7.** Thermoluminescence (TL) glow curves of  $\text{SnO}_2:\text{Eu}^{3+}$  NPs with excitation at 100 K and varying thermal cleaning temperatures from 120 to 280 K in  $\ln(I)$  versus  $1/T$  plot. By utilizing the initial rise analysis, namely, fitting the low-temperature side of the curves with  $I(T) = C \exp(-E_T/k_B T)$ , where  $C$  is a constant and  $k_B$  is Boltzman's constant (K. Van den Eeckhout, A. J. J. Bos, D. Poelman and P. F. Smet, *Phys. Rev. B*, 2013, **87**, 045126), the trap depths ( $E_T$ ) were determined (red lines).

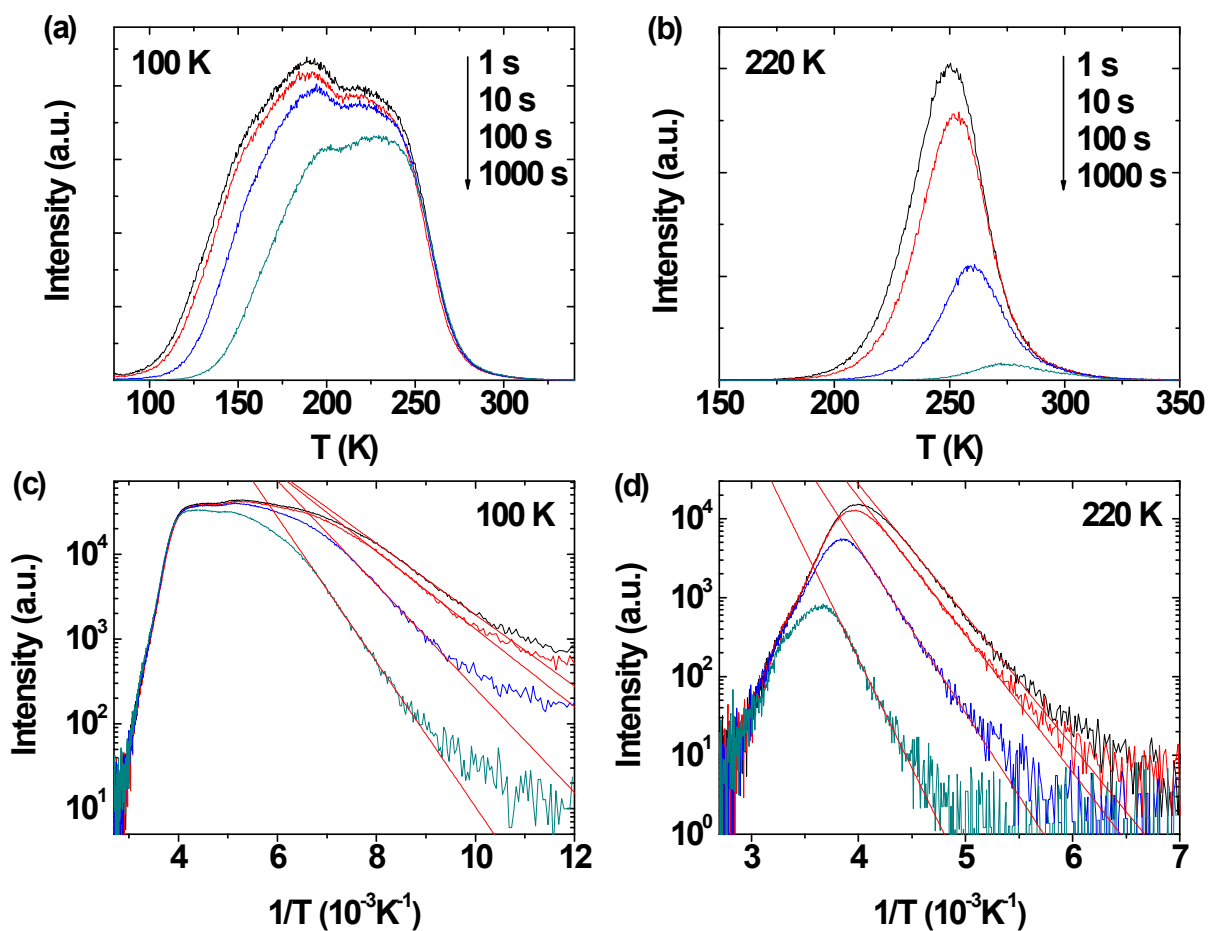




**Figure S8.** TL glow curves of SnO<sub>2</sub>:Eu<sup>3+</sup> NPs synthesized at different annealing temperatures from 900 to 1200 °C. The shapes of the TL glow curves are almost identical, indicating that the traps and their depth distribution are independent of the annealing temperature.



**Figure S9.** Dependence of the inverse power-law exponent on the excitation intensity. The inset shows the persistent luminescence decay curves of SnO<sub>2</sub>:Eu<sup>3+</sup> NPs measured under excitation at 180 K. The exponents were determined by fitting the decay curves with the inverse power law.



**Figure S10.** TL glow curves of  $\text{SnO}_2:\text{Eu}^{3+}$  NPs measured under excitation at (a) 100 K and (b) 220 K by varying the delay time from 1 to 1000 s. (c) and (d) are the corresponding  $\ln(I)$  versus  $1/T$  plot of the TL glow curves in (a) and (b), respectively. The trap depths were determined through the initial rise analysis as described in Figure S7 (red lines).