Controlled Multiple Functionalization of Mesoporous Silica Nanoparticles: Homogeneous Implementation of pairs of Functionalities Communicating through Energy or Proton Transfers.

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Supporting Information

Experimental section:

Materials:

 α -cyclodextrin, 6-hydroxyquinoline, cetyltrimethylammonium bromide (CTAB), sodium ascorbate, rhodamine B and doxorubicin were purchased from Sigma Aldrich. Perfluorooctanoic acid (PFOA), copper sulphate pentahydrate, sodium *N*,*N*-diethyldithiocarbamate trihydrate from Alfa Aesar. (3-Azidopropyl)triethoxysilane (AzPTES)¹, *N*-propargyl bis(triethoxysilylpropyl)amine (Prec-Alk)², *N*-propargyl 4-aminonaphthalimide³, 1-azidomethylpyrene⁴, *N*,*N*-(bispropargyl)anisidine⁵ were synthesized according to literature.

Synthesis of the clickable photoacid N-(2-azidoethyl)-6-hydroxyquinolinium iodide (AHQI): In a two-necked round bottom flask equipped with a condenser, 6-hydroxyquinoline (1.0 g, 6.9 mmol) was suspended in 10 mL acetonitrile. 1-Azido 2-iodoethane (3.0 g, 15 mmol) was then added; the mixture was then stirred at 80 °C for 4 days (The reaction progress was monitored by ¹H NMR). The acetonitrile was removed under reduced pressure and the product was washed with diethyl ether. After filtration and drying, AHQI was obtained as a dark green solid in 95 % yield. ¹H NMR (400 MHz, DMSO-d₆) δ = 11.2 (s,1H, OH), 9.25 (d, 5.4 Hz, 1H), 9.11 (d, 8.5 Hz, 1H), 8.53 (d, 8.5 Hz, 1H), 8.08 (m, 1H), 7.75 (dd, 8.5 Hz and 1.5 Hz; 1H), 7.63 (d, 1.5 Hz, 1H), 5.20 (br, 2H), 4.06 (br, 2H). ¹³C NMR (101 MHz, DMSO-d₆) δ = 158.0, 146.7, 145.7, 132.2, 131.9, 127.6, 121.8, 120.9, 110.7, 56.0, 49.4. HRMS: calculated for C₁₁H₄N₄O⁺: 215.0933 found 215.0935

Synthesis of bis(clickable) mesoporous silica nanoparticles (MSN):

To a solution of sodium hydroxide (2 M, 1.0 mL, 2.0 mmol) in distilled water (80 mL, 4.4 mol), was added CTAB (0.30 g, 0.80 mmol) (in the case of nanorods, PFOA was added at this step with 6 *weight* % relative to CTAB, 0.04 mmol). The mixture was heated to 80 °C and stirred vigorously (750 rpm) until the surfactant was completely dissolved. The silica source tetraethylorthosilicate (TEOS) ((1-2x) × 9 mmol), together with the *clickable* organosilanes

AzPTES ($x \times 9$ mmol) and Prec-Alk ($x \times 9$ mmol) (x = 1, 2 or 5%) were added dropwise over 2 min. A white precipitate appeared and the reaction mixture was stirred at 80 °C for 2 h. The precipitate was then filtered, washed copiously with water and methanol and dried at RT. The molar ratio of the mixture was: Si/ CTAB/ PFOA/ NaOH/ Water: 1/ 0.09/ 0.004/ 0.22/ 490. The surfactant extraction was achieved by Soxhlet extraction (24 h) of the material with a mixture of 200 mL ethanol and 10 mL HCl 12 M. After filtration, the powder was washed with methanol and then dried at 70 °C for 6 h.

Procedure for successive CuAAC click reactions:

CuAAC 1: Bis(*clickable*) nanoparticles ((1+10 x) × 90 mg, 1.5 mmol) with x azide were firstly incubated with 3 eq of the propargylated partner; the energy acceptor (A) N-propargyl 4-aminonaphthalimide ($x \times 1.12$ g, $x \times 4.5$ mmol) or the stalk (ST) N,Nbispropargylanisidine, ($x \times 0.89$ g, $x \times 4.5$ mmol) in the presence of copper sulphate pentahydrate ($x \times 13$ mg, $x \times 0.3$ mmol, 0.2 eq of N₃) and sodium ascorbate ($x \times 26$ mg, $x \times 13$ mg is a solution of N₃). 0.6 mmol, 0.4 eq of N₃) in 4 mL water/t-butanol mixture (v/v = 1). After sonicating for 5 min, the mixture was stirred vigorously at RT for 24 h. The nanoparticles were recovered by centrifugation (8000 rpm, 10 min) and washed with sodium N,Nwater, diethyldithiocarbamate (0.1 M in methanol, 10 mL), methanol (10 mL) and acetone (10 mL) until the clearness of the supernatant (4 ~ 6 times). The resulting material was dried at 70 $^{\circ}$ C for 6 h.

CuAAC 2: The resulting material with reactive propargyl groups, was incubated with 3 eq of the azide-*clickable* partner: the energy donor (D) azidomethylpyrene ($x \times 1.15$ g, $x \times 4.5$ mmol) or the photoacid generator (PAG) AHQI ($x \times 1.54$ g, $x \times 4.5$ mmol) under the same conditions as for the first CuAAC reaction. The afforded bis*clicked* nanoparticles are denoted NS*x*-DA (D and A are the donor and the acceptor) or NR*x* -PAG-ST (PAG and ST are the photoacid generator and the stalk respectively) while x = 1%, 2% or 5%.

Characterization:

The SEM images were obtained with a Hitachi S-4800 apparatus after platinum metallization. TEM micrographs were obtained using a JEOL 1200 EX2 apparatus equipped with a SIS Olympus Quemesa 11 Mpixel camera. FTIR spectra were recorded using a Perkin100 spectrometer equipped with a mono internal reflection ATR module. Raman spectra were recorded with a LabRAM ARAMIS (Horiba) spectrometer using a HeNe laser (633 nm). Absorption spectra were recorded using an Agilent 8453 UV-visible Spectroscopy System. X-ray diffraction was performed on X'PERT Pro MPD PAnalytical with a power of 45 kV x 20 mA and a radiation of CuK α 1.5418 Å, data acquisition was realized in Bragg Brentano mode with slots of divergence 1/16 and anti-diffusion of 1/32°. N₂ adsorption–desorption isotherms were obtained using a Micromeritics ASAP 2020 apparatus after outgassing the samples for 18 h at 40 °C. The specific surface areas were determined from the linear part of the BET transform of the adsorption isotherms in the range of 0.05 <p/p°< 0.2. NMR spectra were recorded with a 400 MHz Bruker spectrometer in dry CDCl₃ at 298 K. ¹H and ¹³C chemical shifts are reported in ppm relative to Me₄Si. Mass spectrometry was carried out in the platform of physico-chemical analyses of the IBMM, University of Montpellier.

Loading and release procedures:

In order to monitor the ability of the designed system to induce light-controlled activation, the nanoparticles were loaded with a fluorescent dye (rhodamine B). In vitro tests were then carried out on the nanoparticles loaded by doxorubicin. These tests were performed with nanorods because of their higher loading capacity.

Loading: A solution of rhodamine B (5 mM) or doxorubicin (3 mM) in water (1 mL) was added to 3 mg of functionalized nanoparticles. The mixture was sonicated for 10 min and stirred vigorously at room temperature for 24 h in the dark. After 2 washings, α -cyclodextrin (6 mg) was added in water (1 mL) and the medium was sonicated for 5 min and stirred in the dark for 48 h. After successive washings with water (about 10 times centrifugation 22000 rpm, 5 min) to eliminate the dye which is not retained in the pores, the particles were kept overnight at RT for drying.

Release procedure: The nanoparticles loaded by rhodamine B were placed in a *cuvette* and slightly ground with the spatula to improve the diffusion of the dye. Distilled water was carefully added to ensure that no particles are floating into the aqueous phase. After the baseline recording, the *cuvette* was placed under 365 nm irradiation (See details in the Supporting Information, Figure S6) and UV-vis spectra of the solution were regularly recorded at different time intervals to track the released rhodamine B at 553 nm.

In vitro experiments on breast cancer cells, imaging, cytotoxicity and drug delivery:

Human breast cancer cells (MCF-7) were purchased from ATCC (American Type Culture Collection, Manassas, VA) and cultured in Dulbecco's Modified Eagle's Medium (DMEM-F12) supplemented with 10% fetal bovine serum and 50 μ g mL⁻¹ gentamycin. Cells were allowed to grow in a humidified atmosphere at 37°C under 5% CO₂.

The day prior to the experiment, MCF7 cells were seeded onto the bottom of sterile glass dishes (World Precision Instrument, Stevenage, UK) at a density of 10^6 cells cm⁻². After 24 h, the cells were washed (once) and incubated in 1 mL medium containing the nanorods loaded with rhodamine B at a concentration of 50 µg mL⁻¹ for 20 h. Before visualization, cells were washed gently with phenol red-free DMEM. Cells were then scanned with a LSM 780 LIVE confocal microscope (Carl Zeiss, Le Pecq, France), at 405 nm with a slice depth (Z stack) of 0.62 µm.

For photo-induced drug release, MCF-7 cells were seeded into 384-well plates at 10^3 cells per well in 100 µL culture medium and allowed to grow for 24 h. Then cells were incubated for 20 h with nanorods at a concentration of 50 µg mL^{-1.} After incubation, the cells were submitted to laser irradiation for 15 sec (405 nm, 18 mW cm⁻², 9 J cm⁻²). After two days, a cell proliferation assay was performed (MTS) to quantify the cell death.



Figure S1: FTIR and Raman spectra of parent bisclickable (left) nanorods and (right) nanospheres



Figure S2: Images obtained by (left) SEM and (right) TEM of a) NR2 b) NS2 and c) NS5 SEM shows nanospheres with narrow sizes (100-150 nm). TEM shows ordered pore channels and the inset in b) is the FFT of the selected part of the image indicating a 2D-hexagonal structure.



Figure S3: (left) X-ray diffractograms and (right) N₂ physisorption of parent bisclickable nanospheres. (inset: BJH plot)



Figure S4: Spectral overlap between the emission of the donor and the excitation of the acceptor



Figure S5: SEM of NS2-DA



In order to confirm that the energy transfer is intraparticular (interaction of donor and acceptor within the same nanoparticles) instead of interparticle interaction at high concentration, one of the samples (NS1-DA) was diluted from 10 to 50 times (from 300 to 30, 15, 10 and 6 mg/L) and the emission spectra were recorded for the samples. The ratio "r" of emission intensities maxima between pyrene (395 nm) and naphthalimide (525 nm) does not decrease with the concentration confirming that an intraparticular FRET is occurring.

Figure S6: Emission spectra of NS1-DA recorded after several dilutions. (Excitation wavelength = 343 nm)



Figure S7: (Above) Molecular structure of the photoacid and the stalk and (below) light excitation setup.



Figure S8: Absorption spectra of (red) NR5-PAG-ST and (black) NR2-PAG-ST. Dashed spectra are (orange) the corresponding anisidine mono*clicked* nanoparticles NR2-ST, (green) PAG in ethanol and (brown) ST in ethanol.



Figure S9: (left) FTIR spectra of bis*clicked* nanoparticles: (blue) 2% and (orange) 5% nanospheres with (red) 2% and (green) 5% nanorods. Dashed lines correspond to the (green) AHQI (PAG) and (brown) Alk-anisidine (ST) precursors. Inset: zoom on N_3 peak in the zone 2050-2150 cm⁻¹ (right) UV-vis spectra of the *clicked* nanospheres.



Figure S10: N₂ sorption isotherms of the (black) parent and (red) 2% (left) and 5% (right) bis*clicked* nanospheres



Figure S11: Evolution of the (left) specific surface area and (right) total pore volume in (blue cones) parent and (red cones) bis*clicked* nanoparticles



Figure S12: Absorption spectra of the photoacid PAG and the stalk ST with the used wavelength corresponding to release experiments with rhodamine B and in vitro tests with doxorubicin



Figure S13: Release profiles of rhodamine B: (a) in pure water with bulk acidification for control NR2-ST; sequential on/off cycles (b) in pure water and (c) in Tris buffer. Red, black and green traces correspond to NR2-PAG-ST, NR5-PAG-ST and NR2-ST, respectively.



Figure S14: Cytotoxicity of nanorods loaded with doxorubicin, without photo-activation. MCF-7 cells were treated with NR2-PAG-ST (blue) and NR5-PAG-ST (pink) with a concentration range of 1 to 100 μ g.mL⁻¹. Three days after treatment, a cell proliferation assay was performed (MTS).



Scheme S1: Synthetic route for obtaining bifunctional nanorods from original precursors *via* CuAAC click reaction

	d ₁₀	Cell parameter a = $(2d_{10}) 3^{-1/2}$	BET surface area (m^2/g)	Uptake at saturation (cm ³ /g)	BJH most probable pore diameter (Å)	Total pore volume (cm ³ /g)
NS2	3.86	4.46	1104	476	25	0.76
NS5	3.86	4.46	1087	375	23	0.67

Table S1: textural parameters of parent bisclickable nanospheres

Functionalization (mmol/g) (conversion %)

	Stalk		Photoacid	
$\approx (M \text{ cm})^{-1}$ 4800 ($\lambda = 261 \text{ nm}$))	13200 ($\lambda = 265 \text{ nm}$)	
X	2	5	2	5
NR <i>x</i> -PAG-ST	0.16 (35%)	0.34 (29%)	0.05 (16%)	0.11 (13%)
NS <i>x</i> -PAG-ST	0.29 (65%)	0.41 (38%)	0.10 (30%)	0.14 (17%)

Table S2: Concentrations of *clicked* photoacid and stalk on nanoparticles and the corresponding functionalization rate.

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