

Electronic Supplementary Information

Facile synthesis of 3D few-layer MoS₂ coated TiO₂ nanosheet core-shell nanostructures for stable and high-performance lithium-ion batteries

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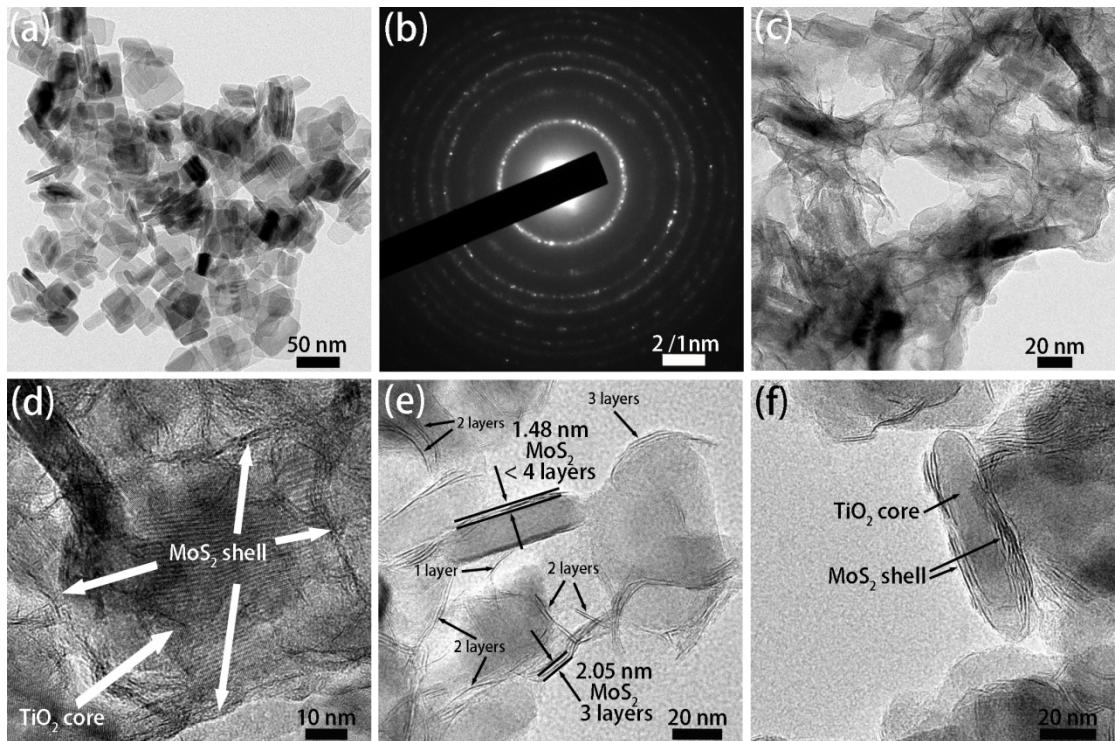


Fig. S1 (a) TEM image and (b) corresponding SAED pattern of F-TiO₂, (c, d) TEM images of 3D FL-MoS₂@TiO₂ before thermal-treated, (e, f) High-resolution TEM images of 3D FL-MoS₂@TiO₂.

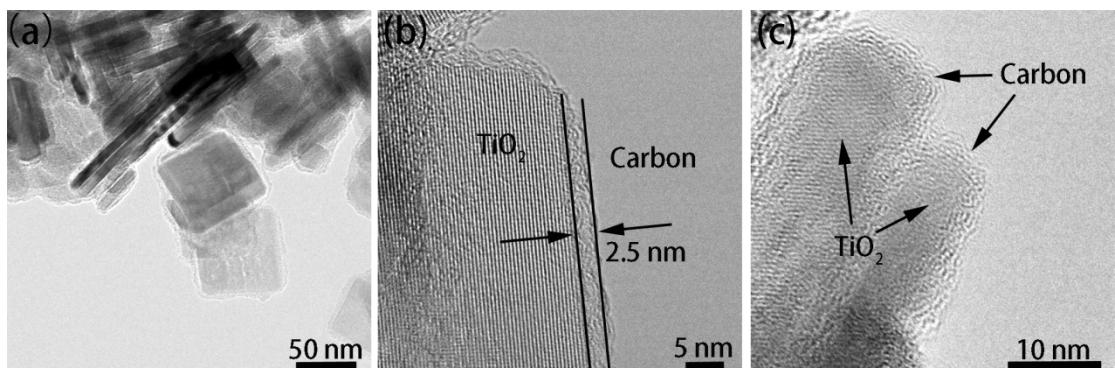


Fig. S2 (a-c) TEM images of hydrothermal products of the complex of F-TiO₂/glucose without Na₂MoO₄-CN₂H₄S.

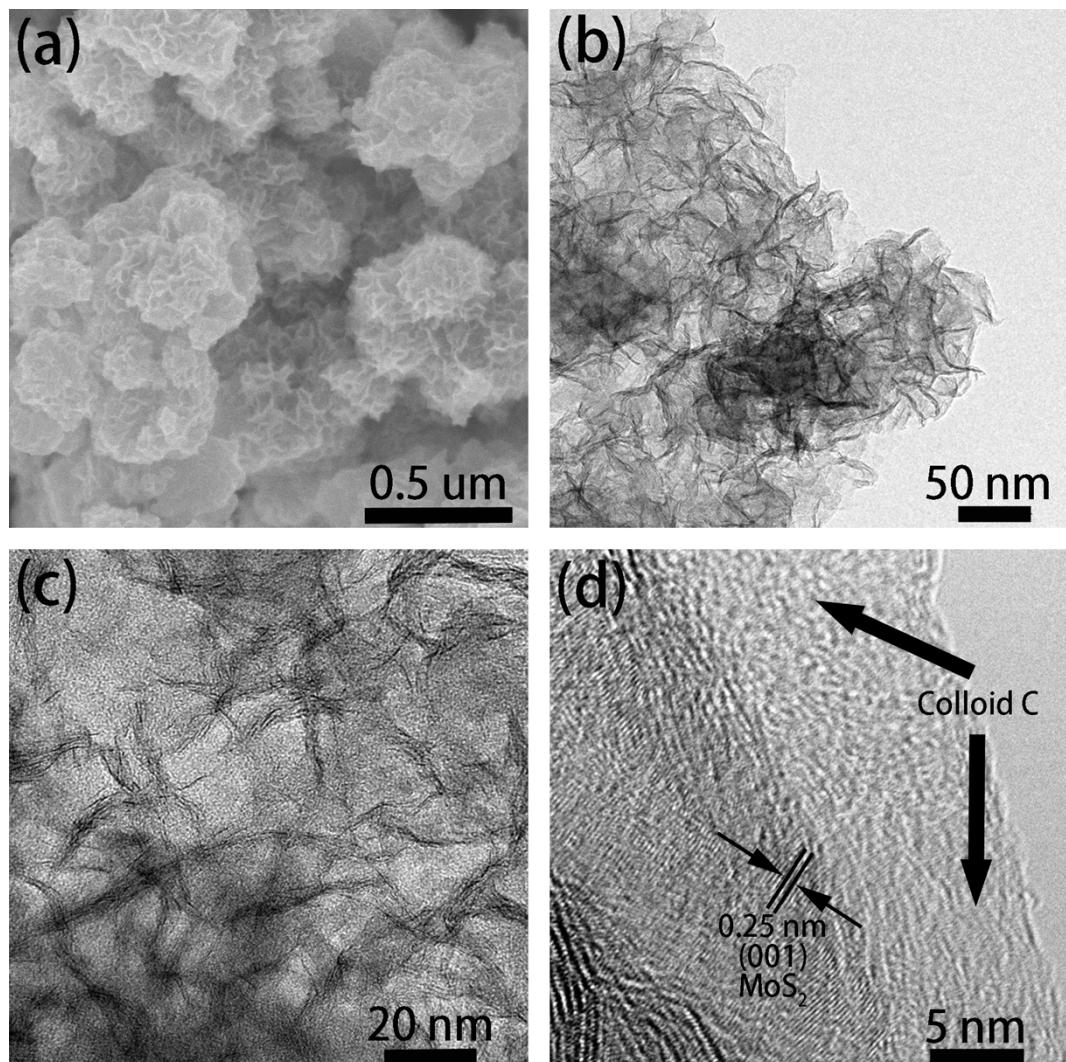


Fig. S3 (a) SEM image and (b-d) TEM images of hydrothermal products of the mixture of Na_2MoO_4 - $\text{CN}_2\text{H}_4\text{S}$ /glucose without F- TiO_2 .

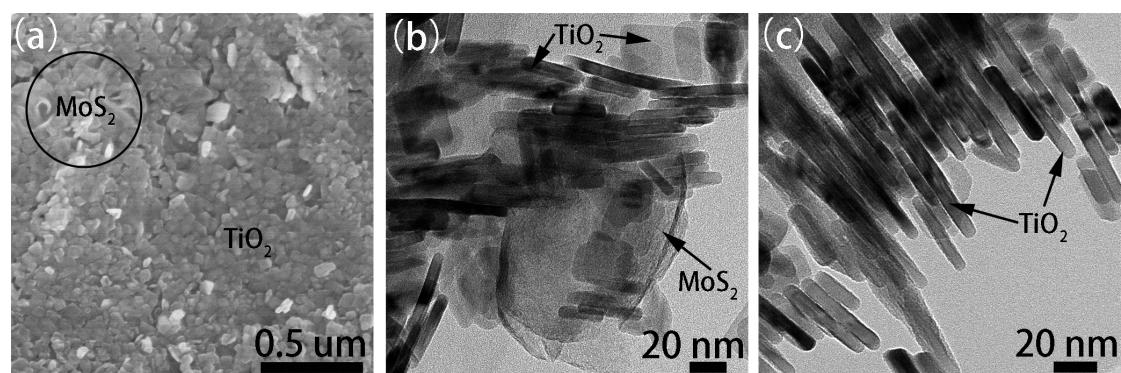


Fig. S4 (a) SEM image and (b, c) TEM images of hydrothermal products of the hybrid of Na_2MoO_4 - $\text{CN}_2\text{H}_4\text{S}/\text{F}-\text{TiO}_2$ without glucose.

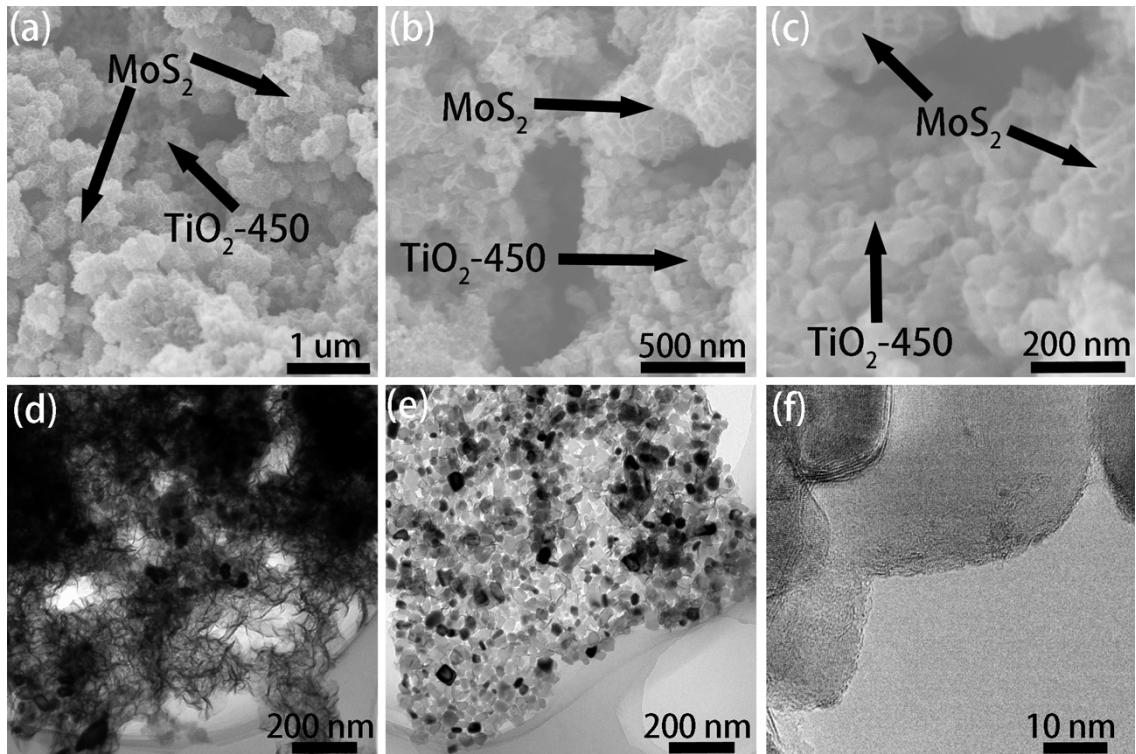


Fig. S5 (a-c) SEM images of MoS₂/TiO₂ composite before thermal-treated, (d-f) TEM images of MoS₂/TiO₂ composite after thermal-treated.

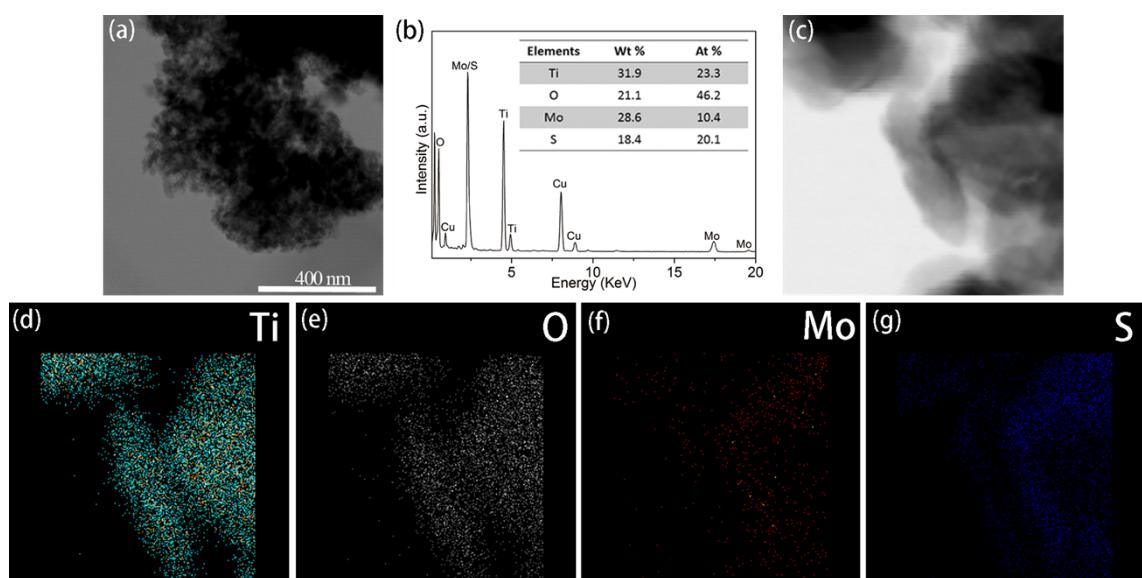


Fig. S6 (a) TEM image and (b) EDX spectra of a typical 3D FL-MoS₂@TiO₂. (c) STEM image of a detailed 3D FL-MoS₂@TiO₂, (d-g) corresponding EDX element mapping images of Ti, O, Mo, and S, respectively.

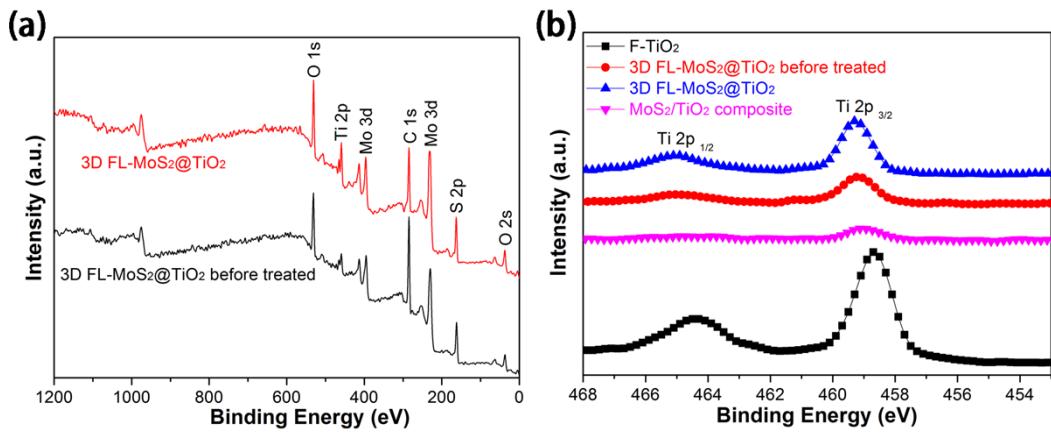


Fig. S7 (a) XPS survey spectra of the 3D FL-MoS₂@TiO₂ before and after calcination. (b) High-resolution XPS spectra of the Ti 2p region of F-TiO₂, MoS₂/TiO₂ composite, 3D FL-MoS₂@TiO₂ before and after calcination.

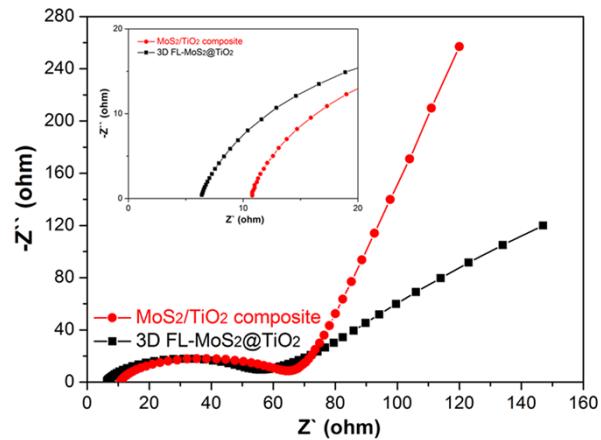


Fig. S8 Nyquist plots of 3D FL-MoS₂@TiO₂ and MoS₂/TiO₂ composite at the state of before cycling over the frequency range from 100 KHz to 100 mHz. The inset is the corresponding magnified high frequency region.

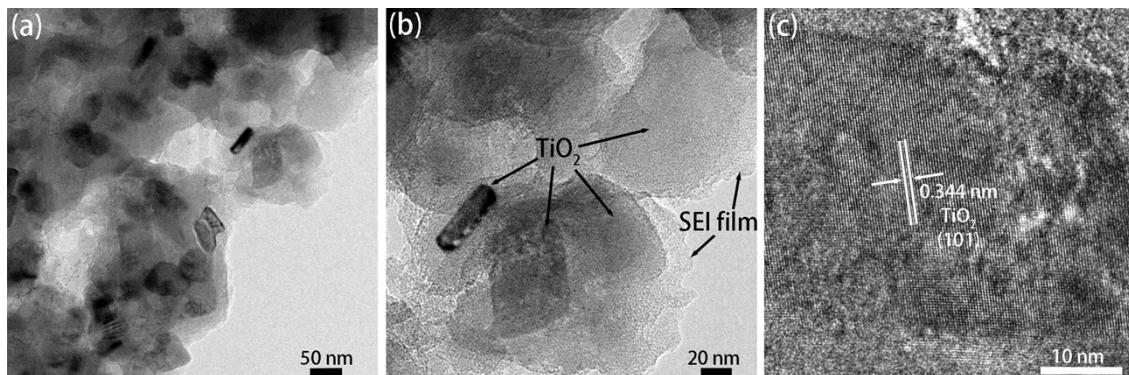


Fig. S9 (a-c) TEM images of the 3D FL-MoS₂@TiO₂ nanocomposite after the 150 electrochemical cycles for the rate cycle performance test in Fig. 6d.

Table S1 Detailed information of rate cycle performance of MoS₂, MoS₂/TiO₂ composite and 3D FL-MoS₂@TiO₂ electrodes corresponding to Fig. 6d.

Range of cycle	Charge/discharge rate (A/g)	Average charge capacity of MoS ₂	Average charge capacity of MoS ₂ /TiO ₂ composite	Average charge capacity of 3D FL-MoS ₂ @TiO ₂
1-10	0.1	732.8	683.2	685.2
11-20	0.2	232.0	633.5	666.4
21-30	0.4	97.7	436.7	612.5
31-40	0.6	56.6	329.5	591.8
41-50	0.8	37.6	280.7	562.1
51-60	1.0	30.5	251.9	550.9
61-70	2.0	18.3	176.6	469.7
71-80	1.0	36.8	254.2	543.4
81	0.1	347.0	542.1	724.2
150	0.1	162.6	425.6	739.2

Table S2 Recently reported LIBs systems based on TiO₂ and MoS₂.

Materials	Voltage range (V)	Charge capacity (mA h/g) at Current Density (mA/g) (capacity retention after cycle-index)	Rate capacity	Current Density(mA/g)
3D FL-MoS ₂ @TiO ₂ [this work]	0.005-3.0	713.7 at 100 (95.9% after 100 cycles)	684.2 550.9 469.7 724.2	100 1000 2000 100
3D assembled MoS ₂ [ref. 21]	0.01-3.0	839 at 100 (71.6% after 50 cycles) (discharge)	600 500 (discharge)	1000 5000
MoS ₂ nanosheets [ref. 22]	0.01-3.0	750 at 50 (63% after 50 cycles) (discharge)	820 500 700 (discharge)	50 1000 100
MoS ₂ microspheres [ref. 23]	0.01-3.0	585 at 100 (73% after 70 cycles)	726 353 700	200 1000 100
Ordered mesoporous MoS ₂ [ref. 24]	0.01-3.0	645 at 50 (92% after 20 cycles)	640 220 580	50 1000 50
MoS ₂ nanoflakes [ref. 25]	0.05-3.0	705.8 at 50 (70% after 40 cycles) (discharge)		
Anatase TiO ₂ nanosheets [ref. 47]	1.0-3.0	136 at 850 (80.5% after 100 cycles)	192 95 169	170 3400 170
Anatase TiO ₂ nanosheet [ref. 49]	1.0-3.0	120.2 at 850 (81.7% after 200 cycles)		
TiO ₂ nanotube [ref. 58]	1.0-3.0	227 at 83.75 (100% after 100 cycles) (discharge)	267 176 260 (discharge)	33.5 3350 33.5
MoS ₂ nanosheet@TiO ₂ nanotube [ref. 42]	0.005-3.0	472 at 100 (68.1% after 100) (discharge)	713 461 611 (discharge)	100 1000 100
few-layered MoS ₂ @ TiO ₂ nanobelt [ref. 43]	0.01-3.0	710 at 100 (91.5% after 100)	717 417 710	100 1000 100
MoS ₂ nanosheet@TiO ₂ nanowire [ref. 44]	0.01-3.0	544 at 100 (75.1% after 100) (discharge)	724 414 563 (discharge)	100 1000 100

Mesoporous MoS ₂ -TiO ₂ nanofibers [ref. 45]	1.0-3.0	124 at 6000 (75.2% after 1000) (discharge)	188 177 120 187 (discharge)	1000 2000 40000 1000
TiO ₂ microspheres embedded with MoS ₂ nanosheets [ref. 46]	0.01-3.0	714 at 100 (86.2% after 200) (discharge)	962 450 805 (discharge)	100 1000 100
C@MoS ₂ microspheres [ref. 26]	0.05-3.0	750 at 100 (~74% after 50) (discharge)	~780 500 750 (discharge)	100 1000 100
MoS ₂ /polyaniline nanowires [ref. 27]	0.01-3.0	952.6 at 100 (89.6% after 50)	1006.4 320 900	200 1000 200
MoS _x /CNT Nanocomposites [ref. 29]	0.01-3.0		1119 358 197 1087 (discharge)	50 1000 2000 50
CNT@MoS ₂ [ref. 30]	0.01-3.0	698 at 100 (48.7% after 60) (discharge)	653 389 (discharge)	200 1000
MoS ₂ -MWCNT hybrids [ref. 31]	0.01-3.0	1090 at 100 (89.8% after 30)	~870 ~550 ~1000	200 500 100