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Ultrafast switching of electrochromic device based on layered double

hydroxide/Prussian blue multilayer films

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Supplementary Figures



Fig. S1 XRD patterns of the MgAl(CO₃)-LDH and MgAl(NO₃)-LDH.



Fig. S2 UV-vis absorption spectrum of PB aqueous solution.



Fig. S3 The zeta potential of a) LDH nanosheets suspension and b) PB NPs colloid.



Fig. S4 TEM images of the $(LDH/PB)_n$ film scratched from the substrate.



Fig. S5 Morphology of the (LDH/PB)_n (n=20–100) films: (a) top-view SEM images (inset: side-view images); (b) tapping-mode AFM topographical images (2 µm×2 µm). From 1 to 5: n=20, 40, 60, 80, 100, respectively.



Fig. S6 XPS spectra of Fe 2p for: a) $(LDH/PB)_n$ film after applying 0.6 and -0.2 V voltage; b) $K_4[Fe^{II}(CN)_6]$ and $K_3[Fe^{III}(CN)_6]$ as reference compounds.



Fig. S7 FT-IR spectra of the (LDH/PB)n film applying a bias voltage of 0.6 and -0.2 V, respectively.



Fig. S8 Optical transmittance spectra of the (LDH/PB)₁₀₀ and (LDH/PB)₁₂₀ films at the colored and bleached states, respectively.



Fig. S9 Cyclic voltammograms of the LBL assembled (LDH/PB)₂₀ film, LBL assembled (PDDA/PB)₆₀ film and spin-coated PB film.



Fig. S10 SEM image of the (LDH/PB)₆₀ film after 400 switching cycles..



Fig. S11 The optical transmittance change of the (LDH/PB)₆₀ film within 1000 switching cycles.



Fig. S12 Optical transmittance of the (LDH/PB)₆₀-ITO/0.1 M KCl electrolyte/ITO-glass ECD after withdrawing applied potential for 24 h.