

## Supplementary Information

### TiO<sub>2</sub>/Vanadates (Sr<sub>10</sub>V<sub>6</sub>O<sub>25</sub>, Ni<sub>3</sub>V<sub>2</sub>O<sub>8</sub>, Zn<sub>2</sub>V<sub>2</sub>O<sub>7</sub>) Heterostructure Photocatalyst with Enhanced Photocatalytic Activity on Photoreduction of CO<sub>2</sub> into CH<sub>4</sub>

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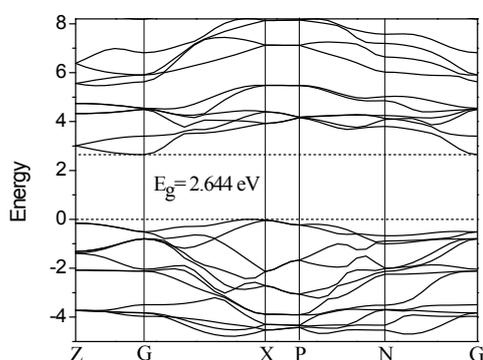


Figure S1. Band structure for pure TiO<sub>2</sub>.

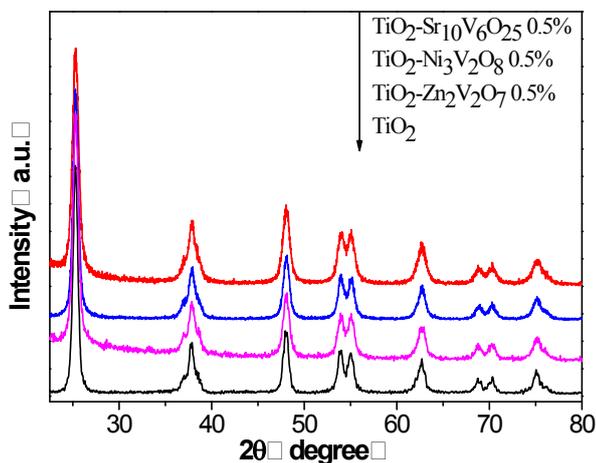


Figure S2. XRD patterns of pure TiO<sub>2</sub>, TiO<sub>2</sub>-Sr<sub>10</sub>V<sub>6</sub>O<sub>25</sub> 0.5%, TiO<sub>2</sub>-Ni<sub>3</sub>V<sub>2</sub>O<sub>8</sub> 0.5% and TiO<sub>2</sub>-Zn<sub>2</sub>V<sub>2</sub>O<sub>7</sub> 0.5% samples.

Table S1. Cell Parameters, Crystallite Size, and Specific Surface Area of Pure TiO<sub>2</sub>, TiO<sub>2</sub>-Sr<sub>10</sub>V<sub>6</sub>O<sub>25</sub> 0.5%, TiO<sub>2</sub>-Ni<sub>3</sub>V<sub>2</sub>O<sub>8</sub> 0.5% and TiO<sub>2</sub>-Zn<sub>2</sub>V<sub>2</sub>O<sub>7</sub> 0.5% samples.

Samples	Lattice parameter		cell volume (Å <sup>3</sup> )	crystal size (nm)	S <sub>BET</sub> (m <sup>2</sup> g <sup>-1</sup> )
	(Å)				
	a=b	c			
TiO <sub>2</sub>	3.780	9.481	135.9	14.51	63.1
TiO <sub>2</sub> -Zn <sub>2</sub> V <sub>2</sub> O <sub>7</sub> 0.5%	3.780	9.504	135.8	13.71	70.8
TiO <sub>2</sub> -Ni <sub>3</sub> V <sub>2</sub> O <sub>8</sub> 0.5%	3.780	9.485	135.8	13.19	76.3
TiO <sub>2</sub> -Sr <sub>10</sub> V <sub>6</sub> O <sub>25</sub> 0.5%	3.778	9.484	135.3	12.15	79.1

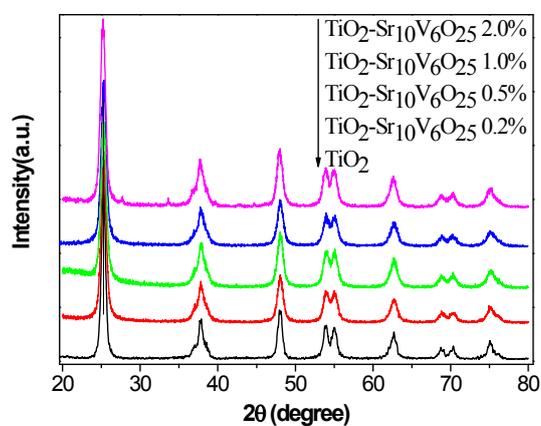


Figure S3. XRD patterns of pure TiO<sub>2</sub> and TiO<sub>2</sub>-Sr<sub>10</sub>V<sub>6</sub>O<sub>25</sub> x% samples.

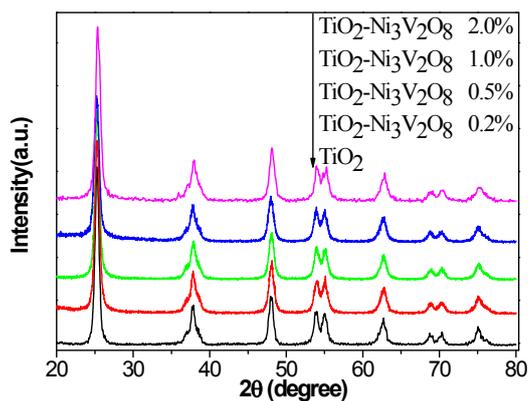


Figure S4. XRD patterns of pure TiO<sub>2</sub> and TiO<sub>2</sub>-Ni<sub>3</sub>V<sub>2</sub>O<sub>8</sub> x% samples.

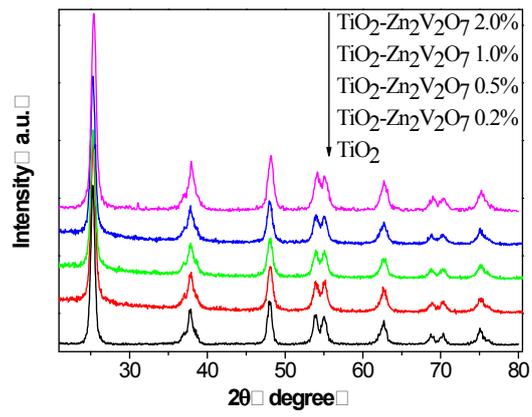
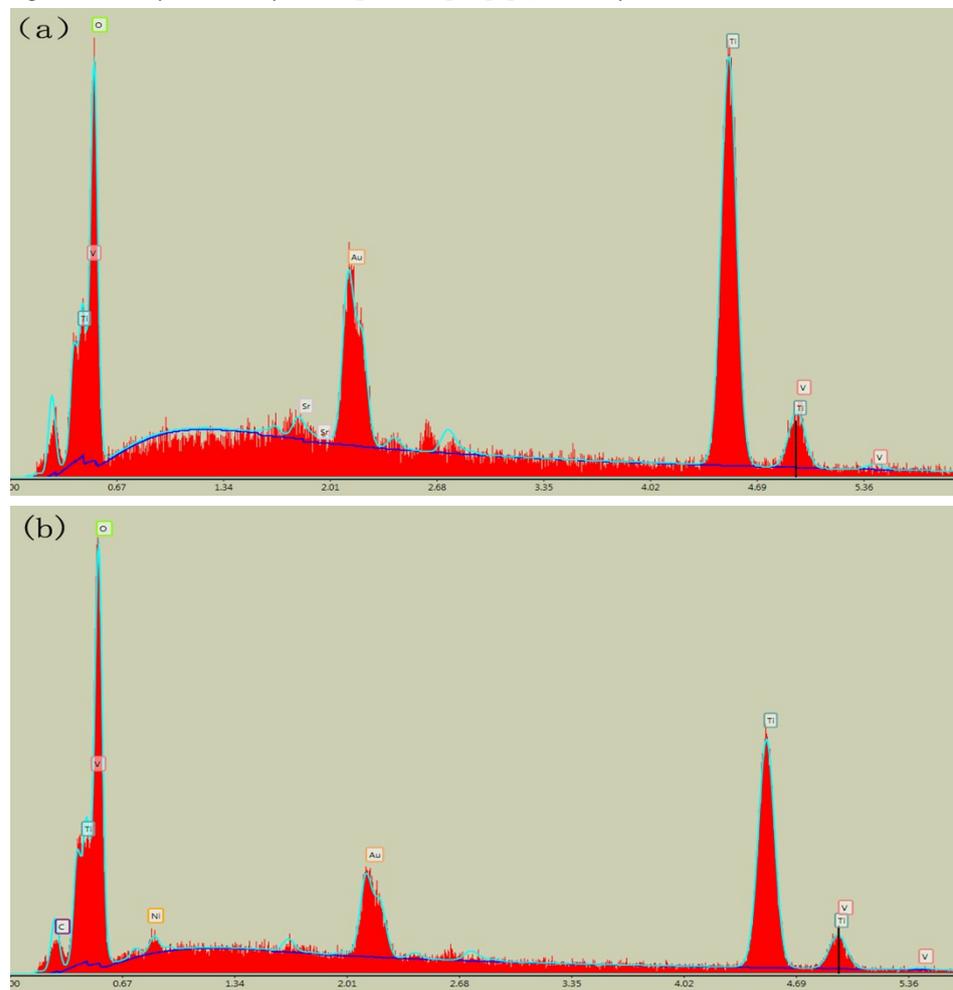


Figure S5. XRD patterns of pure TiO<sub>2</sub> and TiO<sub>2</sub>-Zn<sub>2</sub>V<sub>2</sub>O<sub>7</sub> x% samples.



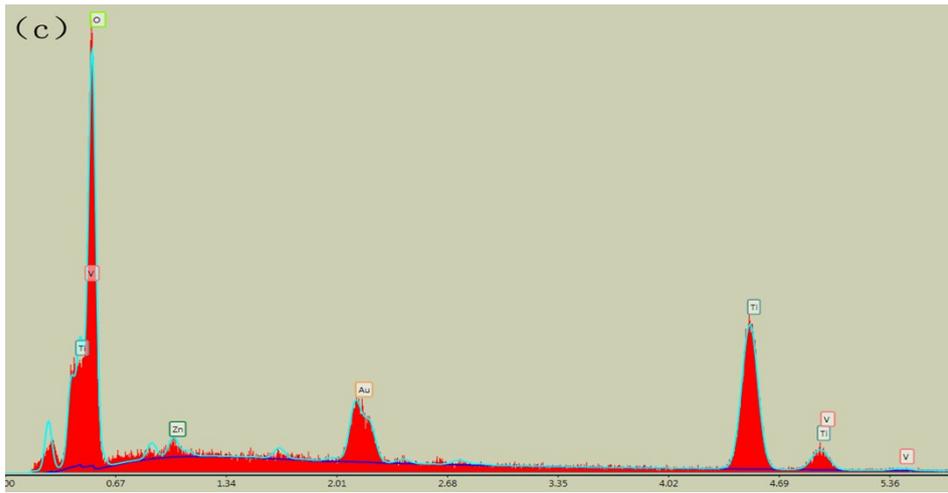


Figure S6. EDX spectrum of (a)  $\text{TiO}_2\text{-Sr}_{10}\text{V}_6\text{O}_{25}$  2.0%, (b)  $\text{TiO}_2\text{-Ni}_3\text{V}_2\text{O}_8$  2.0% and (c)  $\text{TiO}_2\text{-Zn}_2\text{V}_2\text{O}_7$  2.0% samples.

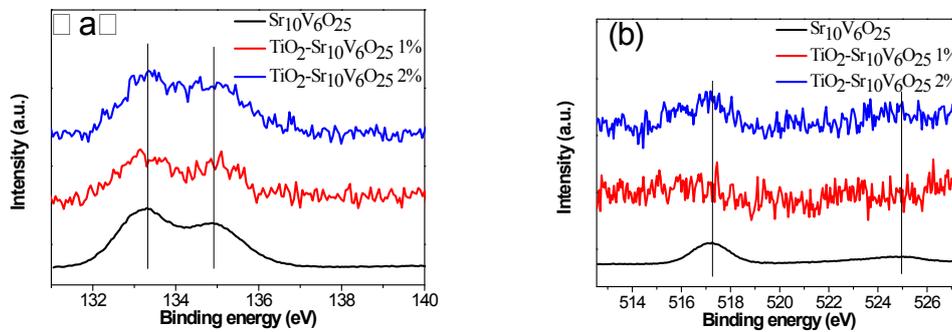


Figure S7. Sr 3d (a) XPS spectra of  $\text{Sr}_{10}\text{V}_6\text{O}_{25}$  and  $\text{TiO}_2\text{-Sr}_{10}\text{V}_6\text{O}_{25}$  x%; V 2p (b) XPS spectra of  $\text{Sr}_{10}\text{V}_6\text{O}_{25}$  and  $\text{TiO}_2\text{-Sr}_{10}\text{V}_6\text{O}_{25}$  x%.

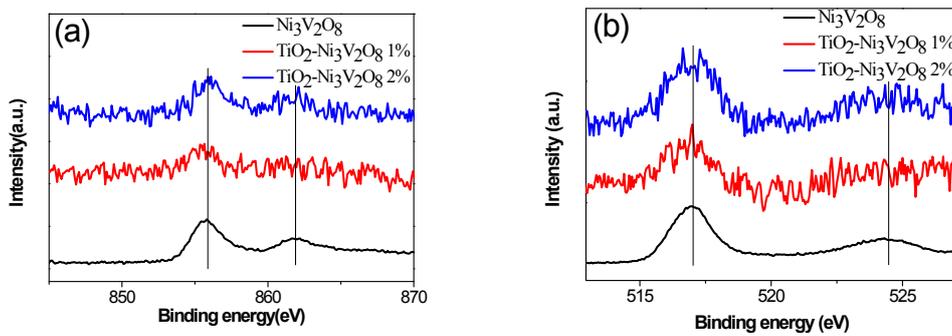


Figure S8. Ni 2p (a) XPS spectra of  $\text{Ni}_3\text{V}_2\text{O}_8$  and  $\text{TiO}_2\text{-Ni}_3\text{V}_2\text{O}_8$  x%; V 2p (b) XPS spectra of  $\text{Ni}_3\text{V}_2\text{O}_8$  and  $\text{TiO}_2\text{-Ni}_3\text{V}_2\text{O}_8$  x%.

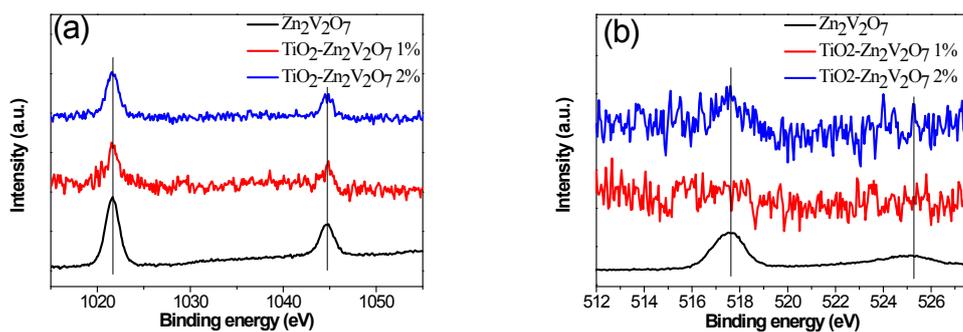


Figure S9. Zn 2p (a) XPS spectra of  $Zn_2V_2O_7$  and  $TiO_2-Zn_2V_2O_7$  x%; V 2p (b) XPS spectra of  $Zn_2V_2O_7$  and  $TiO_2-Zn_2V_2O_7$  x%.

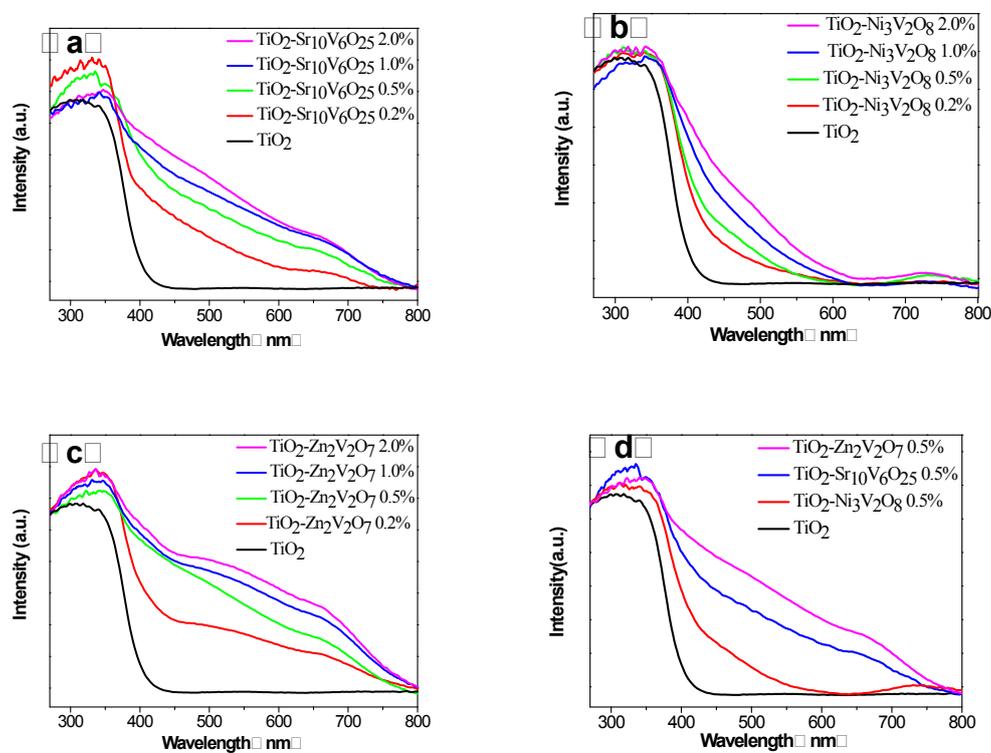


Figure S10. Diffuse reflectance UV-Vis spectra of (a)  $TiO_2$  and  $TiO_2-Sr_{10}V_6O_{25}$  x%; (b)  $TiO_2$  and  $TiO_2-Ni_3V_2O_8$  x%; (c)  $TiO_2$  and  $TiO_2-Zn_2V_2O_7$  x%. (d)  $TiO_2$ ,  $TiO_2-Sr_{10}V_6O_{25}$  0.5%,  $TiO_2-Ni_3V_2O_8$  0.5% and  $TiO_2-Zn_2V_2O_7$  0.5% samples.

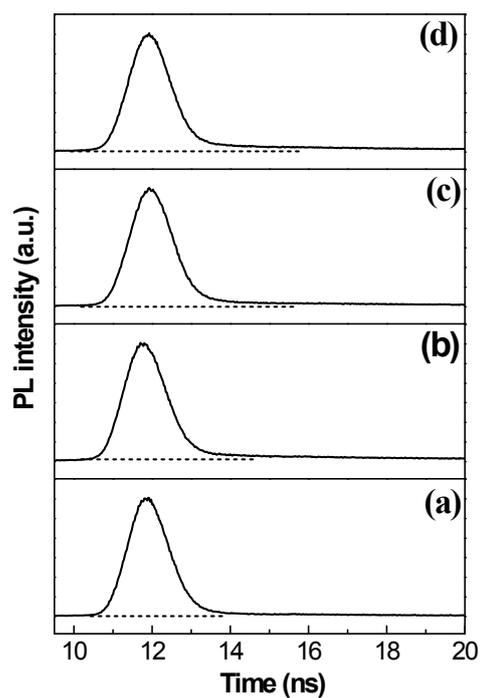


Figure S11. Time-resolved PL decay curves for (a) pure  $\text{TiO}_2$ , (b)  $\text{TiO}_2\text{-Zn}_2\text{V}_2\text{O}_7$  0.5%, (c)  $\text{TiO}_2\text{-Ni}_3\text{V}_2\text{O}_8$  0.5% and (d)  $\text{TiO}_2\text{-Sr}_{10}\text{V}_6\text{O}_{25}$  0.5% samples.

Table S2. Photocatalytic activity of pure  $\text{TiO}_2$ ,  $\text{TiO}_2\text{-Sr}_{10}\text{V}_6\text{O}_{25}$  0.5%,  $\text{TiO}_2\text{-Ni}_3\text{V}_2\text{O}_8$  0.5% and  $\text{TiO}_2\text{-Zn}_2\text{V}_2\text{O}_7$  0.5% under UV light irradiation for 8 h.

sample	$\text{CH}_4$ generation amount ( $10^{-6}\text{mol}$ )	Specific photocatalytic activity <sup>a</sup> ( $10^{-6}\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$ )	CO generation amount ( $10^{-6}\text{mol}$ )	specific photocatalytic activity <sup>b</sup> ( $10^{-6}\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$ )
Blank	0.399	0.499	1.699	2.12
$\text{TiO}_2$	0.409	0.512	2.143	2.68
$\text{TiO}_2\text{-Sr}_{10}\text{V}_6\text{O}_{25}$ 0.5%	3.369	4.21	2.788	3.49
$\text{TiO}_2\text{-Ni}_3\text{V}_2\text{O}_8$ 0.5%	2.184	2.73	2.786	3.48
$\text{TiO}_2\text{-Zn}_2\text{V}_2\text{O}_7$ 0.5%	1.972	2.46	2.652	3.32

<sup>a</sup>specific photocatalytic activity of  $\text{CH}_4$ ,  $\text{CH}_4$  generation amount per unit mass catalyst per hour;

<sup>b</sup>specific photocatalytic activity of CO, CO generation amount per unit mass catalyst per hour.

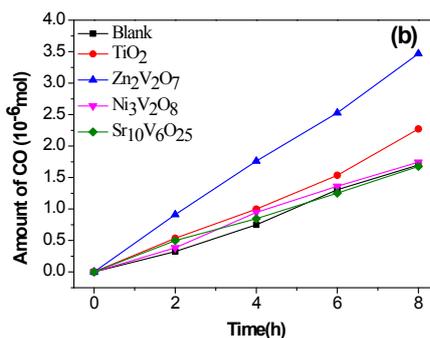
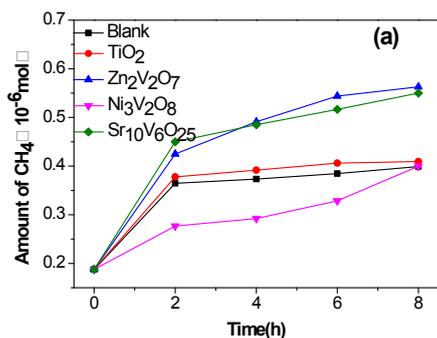


Figure S12. Photocatalytic activity for reduction of CO<sub>2</sub> into CH<sub>4</sub> (a) and CO (b) of TiO<sub>2</sub>, Zn<sub>2</sub>V<sub>2</sub>O<sub>7</sub>, Ni<sub>3</sub>V<sub>2</sub>O<sub>8</sub> and Sr<sub>10</sub>V<sub>6</sub>O<sub>25</sub> under UV light irradiation for 8 h.

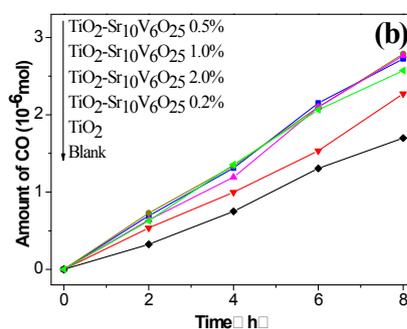
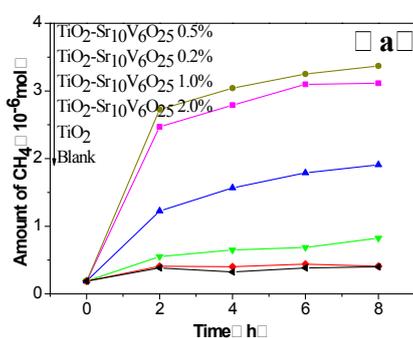


Figure S13. Photocatalytic activity for reduction of CO<sub>2</sub> into CH<sub>4</sub> (a) and CO (b) of TiO<sub>2</sub> and TiO<sub>2</sub>-Sr<sub>10</sub>V<sub>6</sub>O<sub>25</sub> x% under UV light irradiation for 8 h.

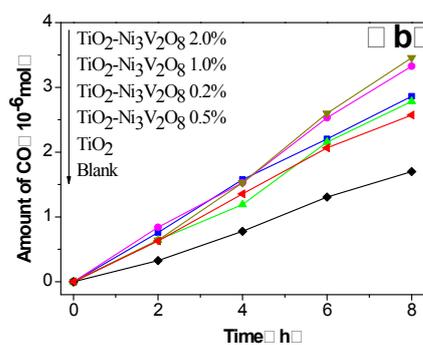
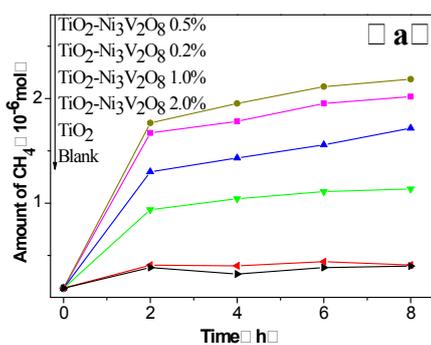


Figure S14. Photocatalytic activity for reduction of CO<sub>2</sub> into CH<sub>4</sub> (a) and CO (b) of TiO<sub>2</sub> and TiO<sub>2</sub>-Ni<sub>3</sub>V<sub>2</sub>O<sub>8</sub> x% under UV light irradiation for 8 h.

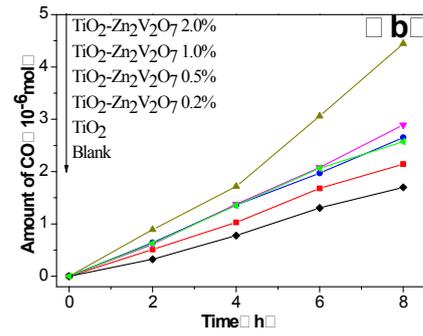
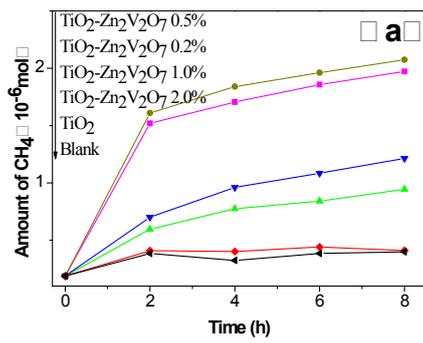


Figure S15. Photocatalytic activity for reduction of CO<sub>2</sub> into CH<sub>4</sub> (a) and CO (b) of TiO<sub>2</sub> and TiO<sub>2</sub>-Zn<sub>2</sub>V<sub>2</sub>O<sub>7</sub> x% under UV light irradiation for 8 h.

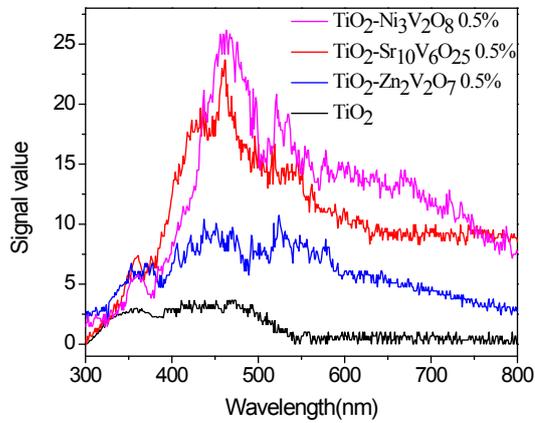


Figure S16. Photocurrent action spectra of pure TiO<sub>2</sub>, TiO<sub>2</sub>-Sr<sub>10</sub>V<sub>6</sub>O<sub>25</sub> 0.5%, TiO<sub>2</sub>-Ni<sub>3</sub>V<sub>2</sub>O<sub>8</sub> 0.5% and TiO<sub>2</sub>-Zn<sub>2</sub>V<sub>2</sub>O<sub>7</sub> 0.5%