

Electronic Supplementary Information

Rapid Flame Synthesis of Internal Mo⁶⁺ Doped TiO₂ Nanocrystals *In-Situ* Decorated with Highly Dispersed MoO₃ Clusters for Lithium Ion Storage

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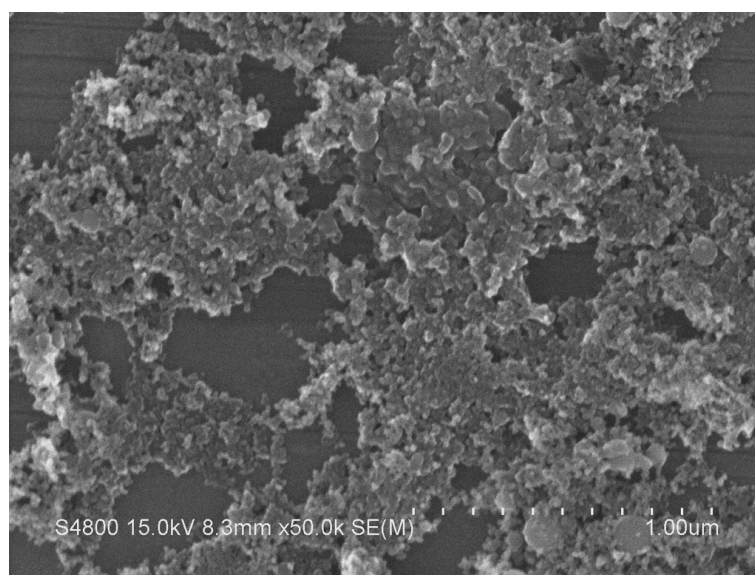


Figure S1 SEM image of the $\text{Mo}^{6+}\text{-TiO}_2/\text{MoO}_3$ NHs.

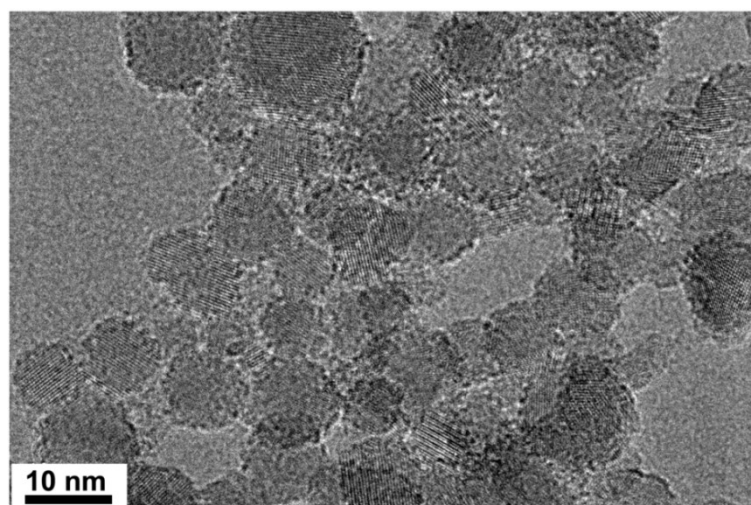


Figure S2 TEM image of the $\text{Mo}^{6+}\text{-TiO}_2/\text{MoO}_3$ NHs.

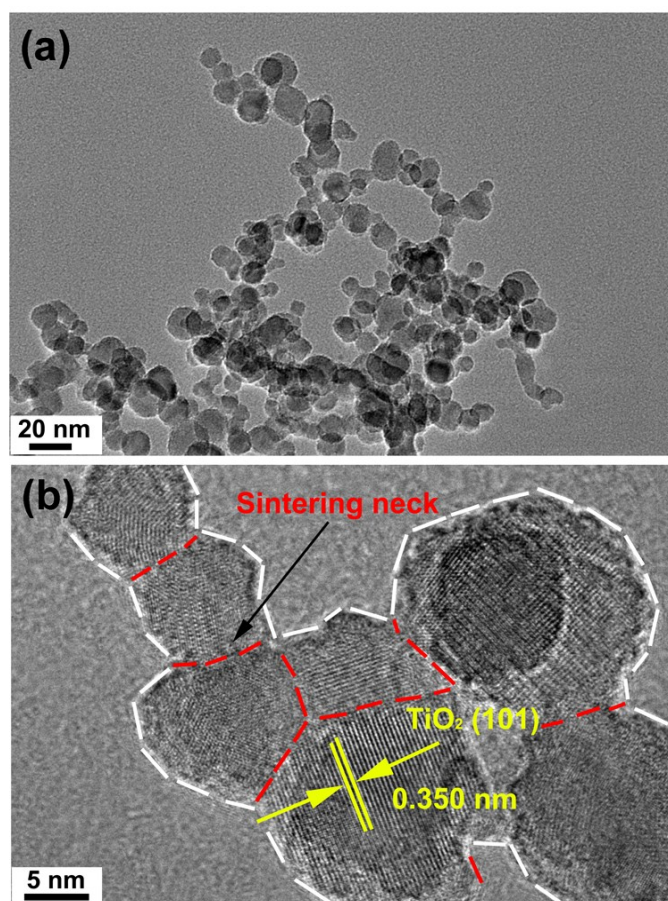


Figure S3 (a) TEM image and (b) HRTEM image of flame made pristine TiO₂ particles.

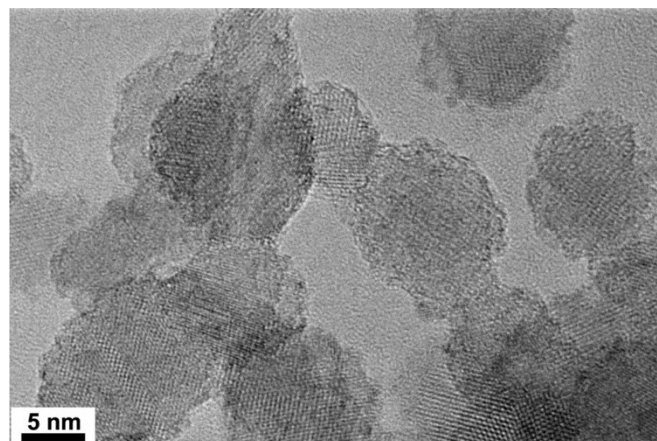


Figure S4 TEM image of Mo⁶⁺-TiO₂/MoO₃ hybrid with designed 15 at.% Mo.

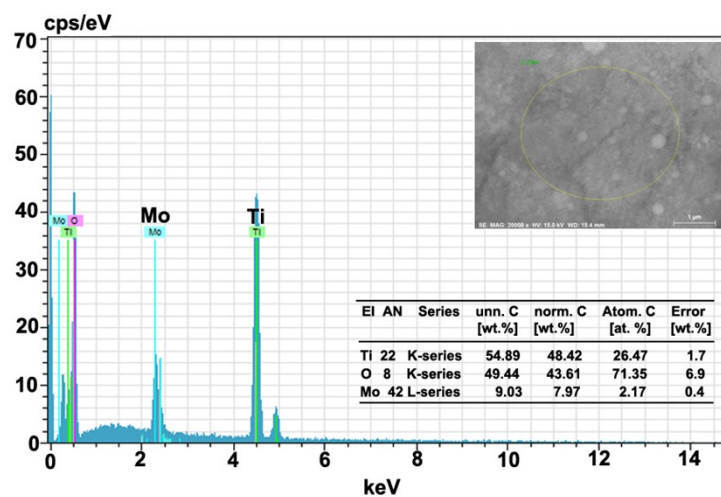


Figure S5 EDX of $\text{Mo}^{6+}\text{-TiO}_2/\text{MoO}_3$ hybrid with designed 10 at.% Mo.

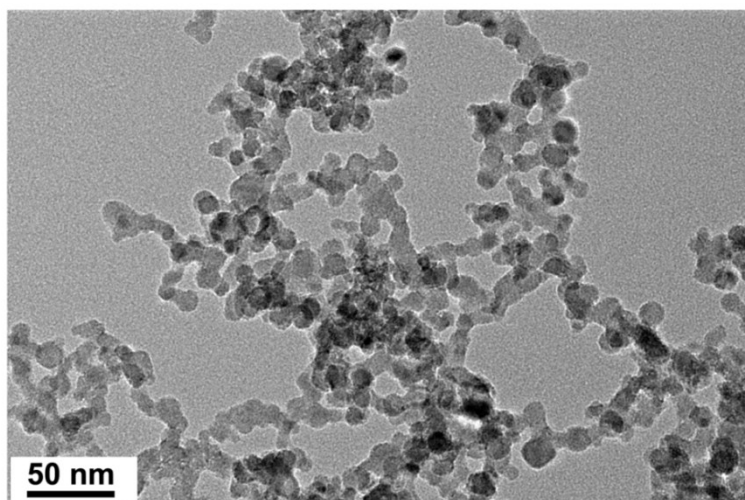


Figure S6 TEM image of the $\text{Mo}^{6+}\text{-TiO}_2/\text{MoO}_3$ NHs after sulfuration with thiourea at 400°C for 3 h.

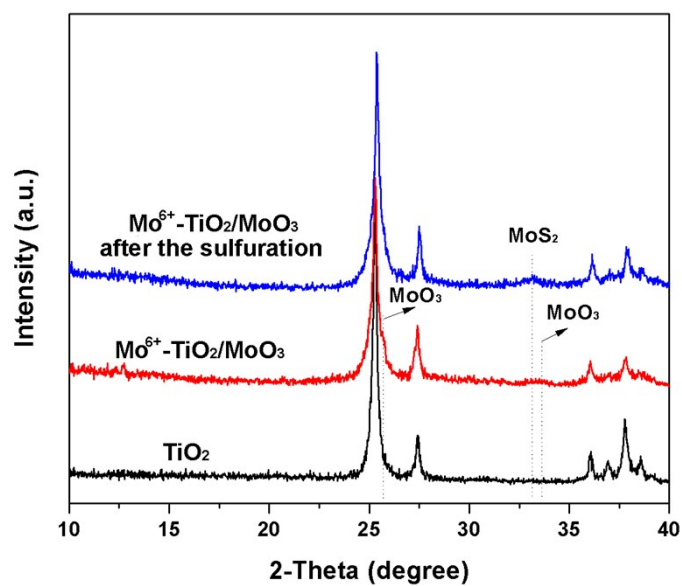


Figure S7 XRD patterns of TiO₂, and Mo⁶⁺-TiO₂/MoO₃ NHs before and after the sulfuration.

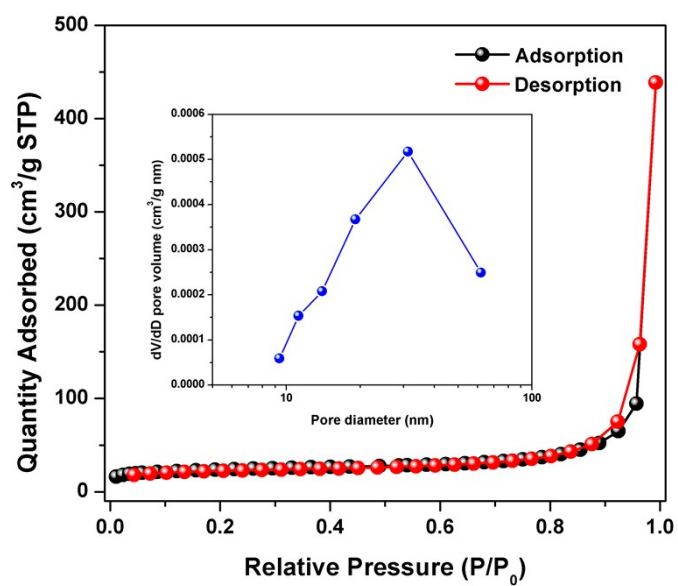


Figure S8 Nitrogen adsorption/desorption isotherm of the Mo⁶⁺-TiO₂/MoO₃ NHs (inset: the corresponding pore distribution profiles).

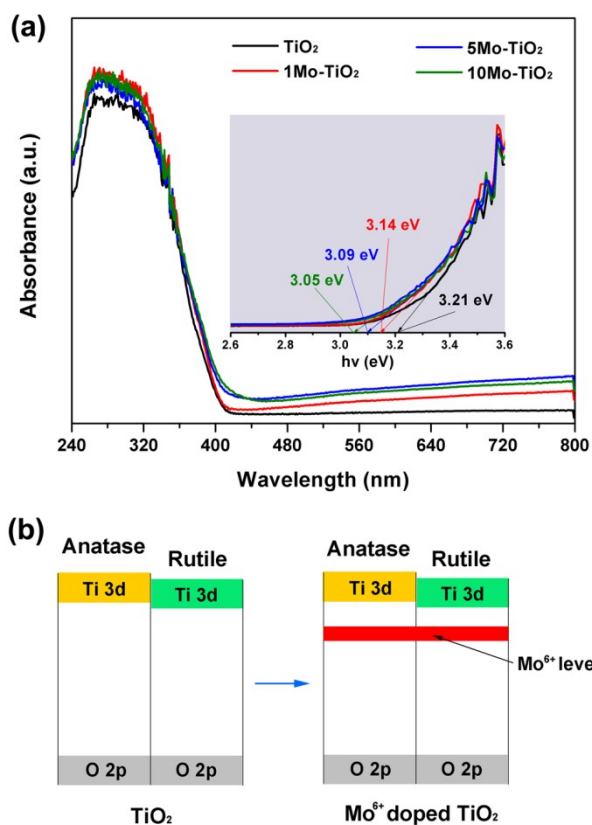


Figure S9 (a) UV-vis absorption spectra of the TiO₂, 1Mo-TiO₂, 5Mo-TiO₂ and 10Mo-TiO₂ samples (the inset curves show the corresponding band gap, (b) schematic of the band structures of TiO₂ and Mo⁶⁺ doped TiO₂.

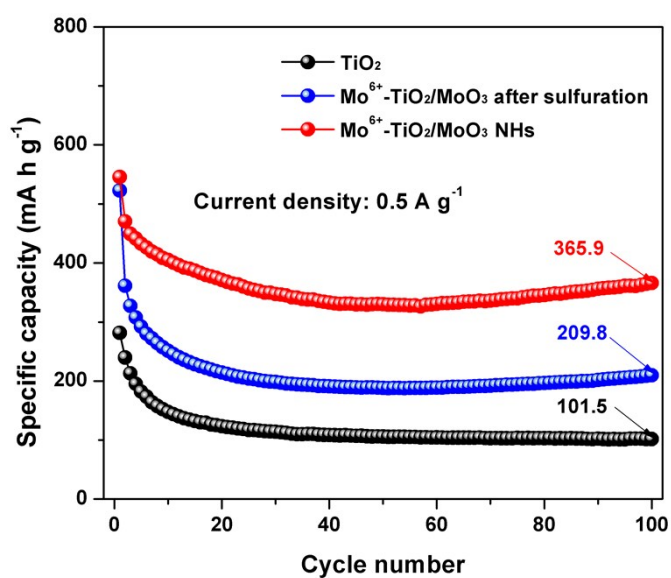


Figure S10 Cycling performance of TiO₂, and Mo⁶⁺-TiO₂/MoO₃ NHs before and after the sulfuration at a current density of 0.5 A g⁻¹.