

Supporting Information for

Self-supported Nanoporous NiCo₂O₄ Nanowires with Cobalt-nickel Layered Oxide Nanosheets for Overall Water Splitting

Jie Yin,^a Panpan Zhou,^a Li An,^a Liang Huang,^a Changwei Shao,^c Jun Wang,^c Hongyan Liu^a and Pinxian Xi^{*a,b}

^aKey Laboratory of Nonferrous Metal Chemistry and Resources Utilization of Gansu Province, State Key Laboratory of Applied Organic Chemistry and The Research Center of Biomedical Nanotechnology, Lanzhou University, Lanzhou, 730000, P. R. China.

E-mail: xipx@lzu.edu.cn

^bKey Laboratory of Advanced Energy Materials Chemistry (Ministry of Education), Nankai University, Tianjin, 300071, P. R. China.

^cAdvanced Ceramic Fibers and Composites laboratory, College of Aerospace Science and Engineering, National University of Defense Technology, Changsha, 410073, P. R. China

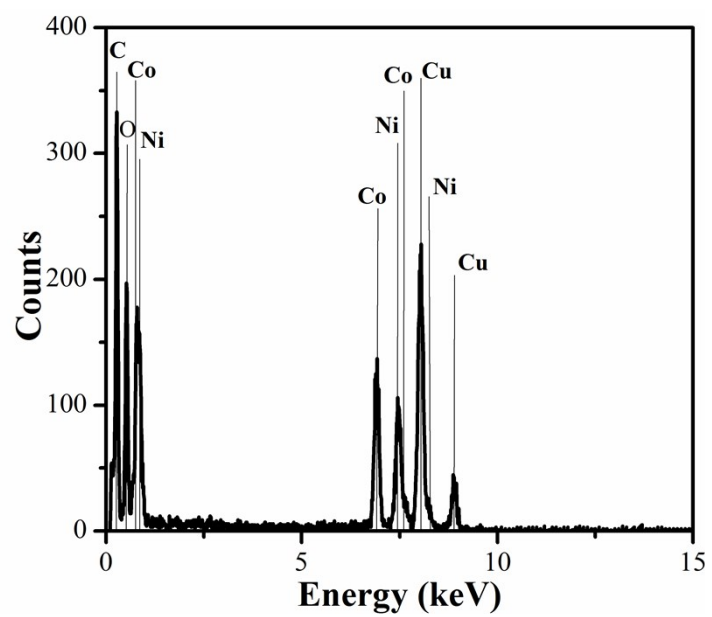


Fig. S1 EDX spectrum for CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs.

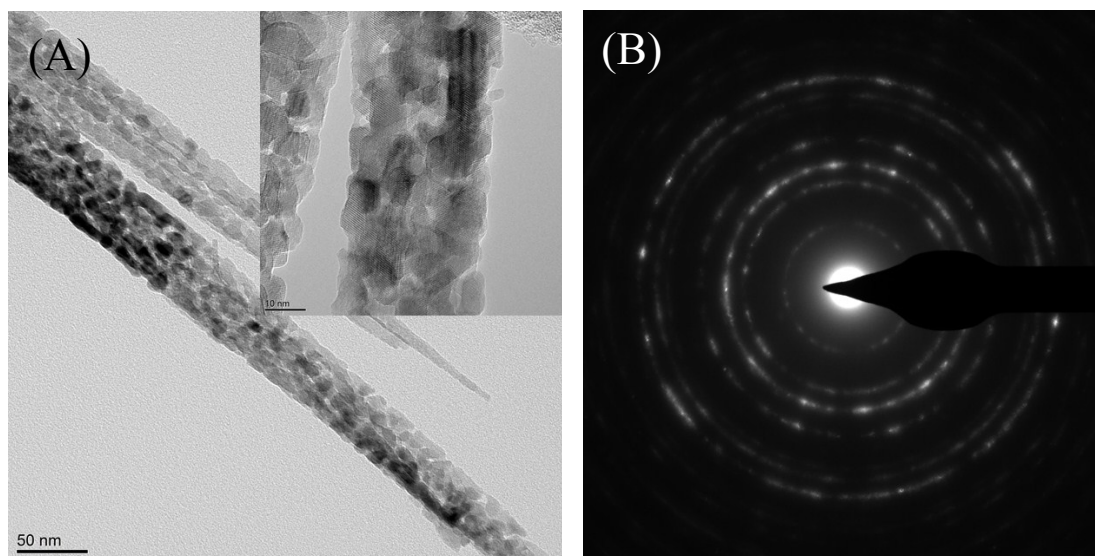


Fig. S2 HR-TEM and SEAD pattern of the NiCo_2O_4 NWs.

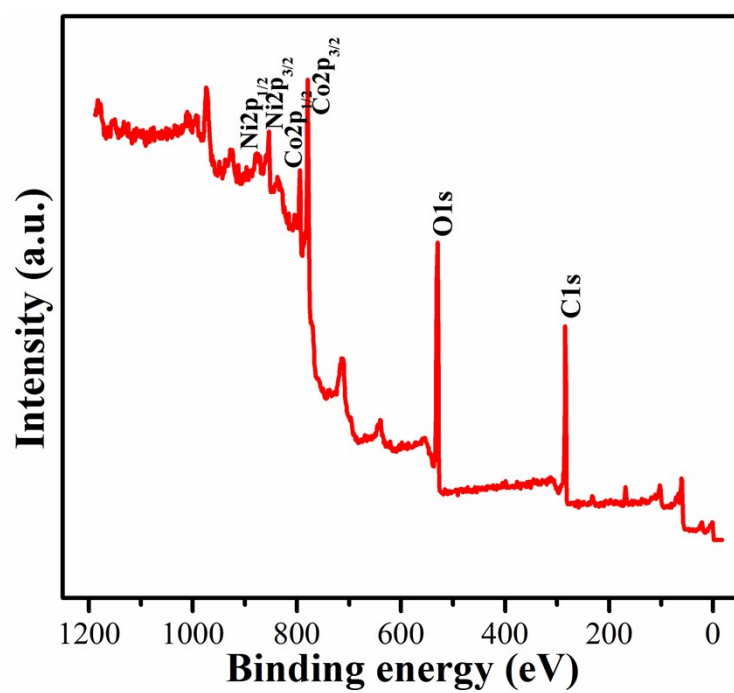


Fig. S3 XPS spectrum for CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs.

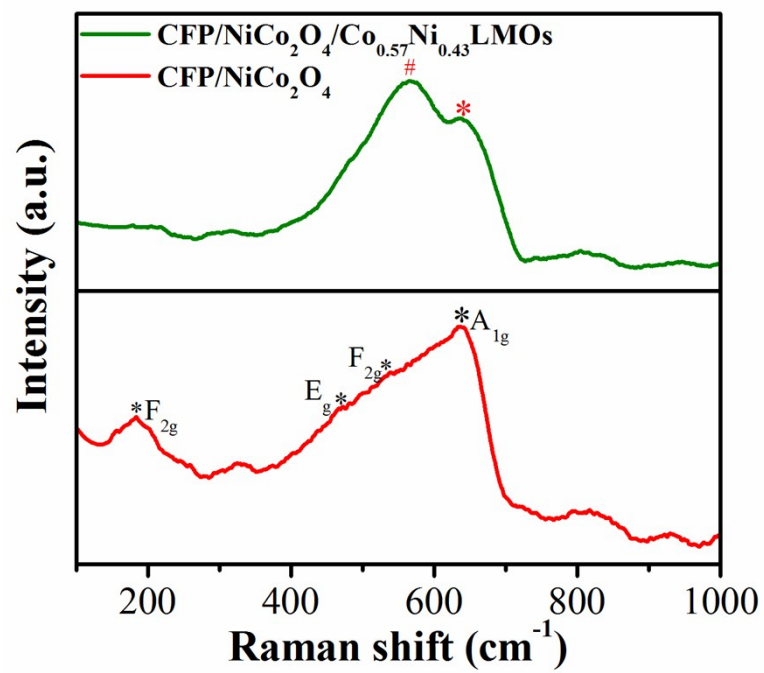


Fig. S4 The Raman spectrum of the CFP/NiCo₂O₄ and CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs.

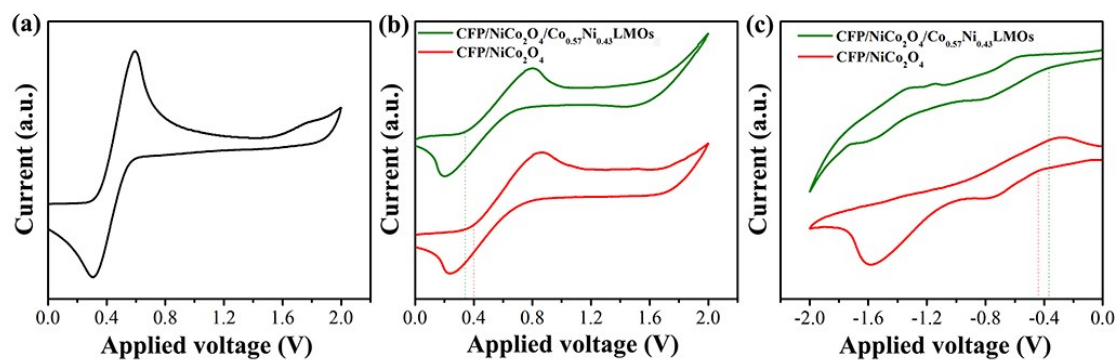


Fig. S5 (a) The cyclic voltammograms (CV) of the oxidation potential of ferrocene as the internal standard to calibrate the measurements, and (b) oxide and (c) reduction CV of CFP/NiCo₂O₄ and CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs.

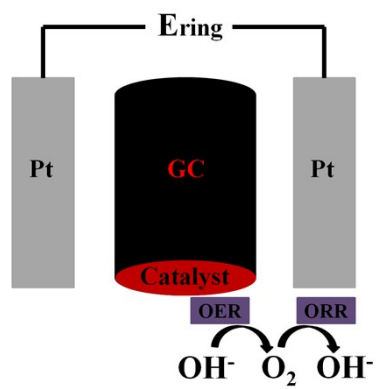


Fig. S6 Schematic illustration of the continuous OER (disk electrode) to ORR (ring electrode) process initiated on a RRDE.

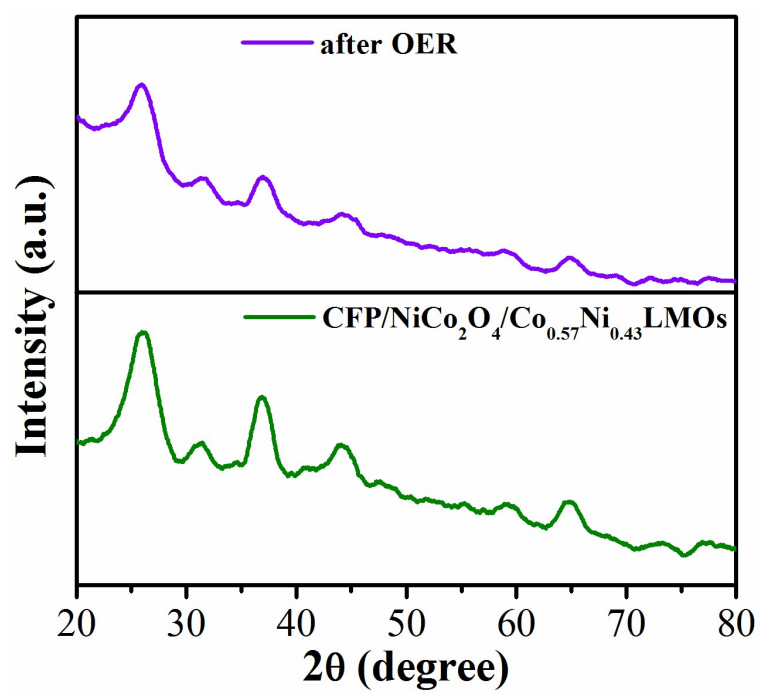


Fig. S7 XRD spectra of CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs before and after the OER stability measurements.

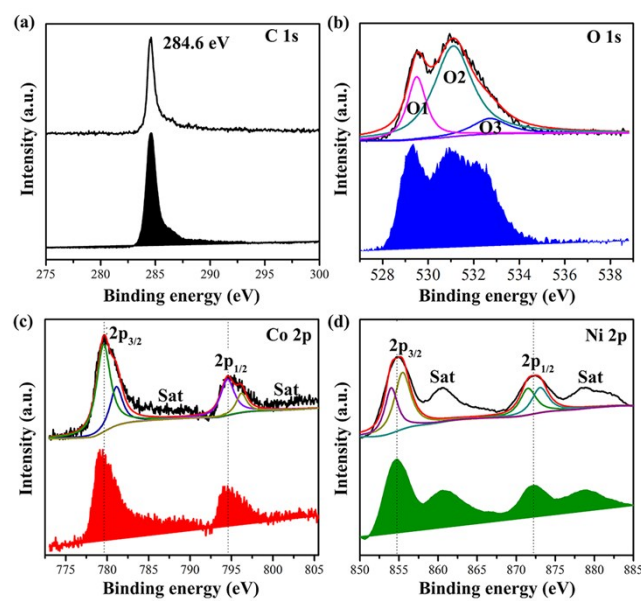


Fig. S8 XPS spectra of (a) C 1s, (b) Co 2p, (c) Ni 2p and (d) O 1s of CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs before and after the OER stability measurements.

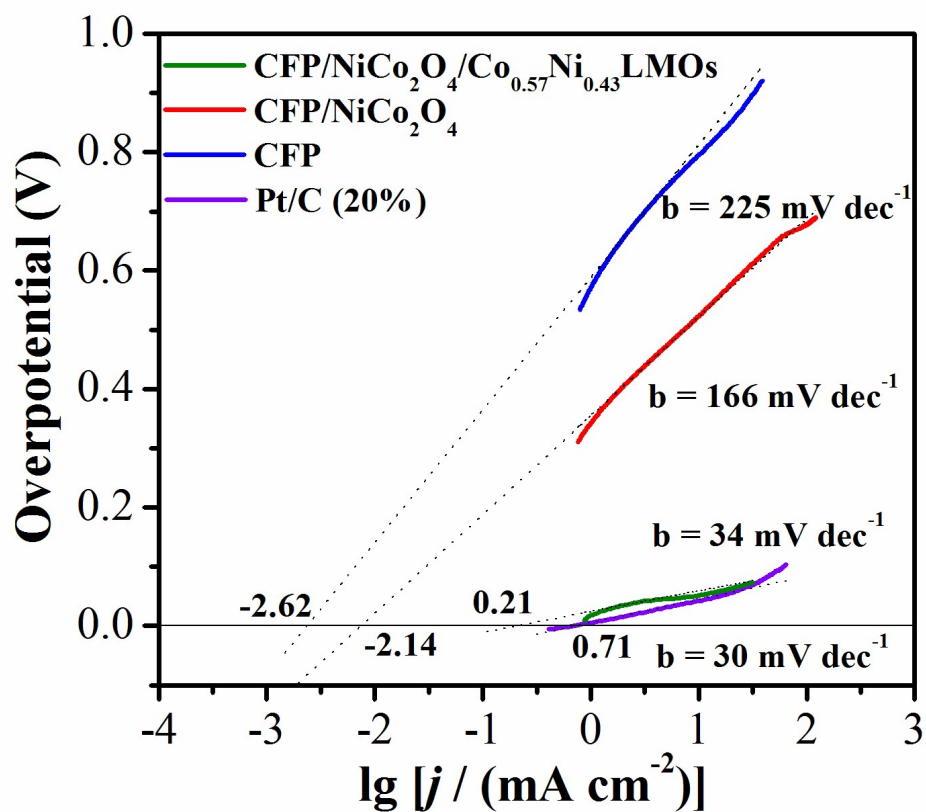


Fig. S9 Calculated exchange current density for CFP, CFP/NiCo₂O₄, CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs and Pt/C in 0.5 M H₂SO₄ by applying extrapolation method to the Tafel plot.

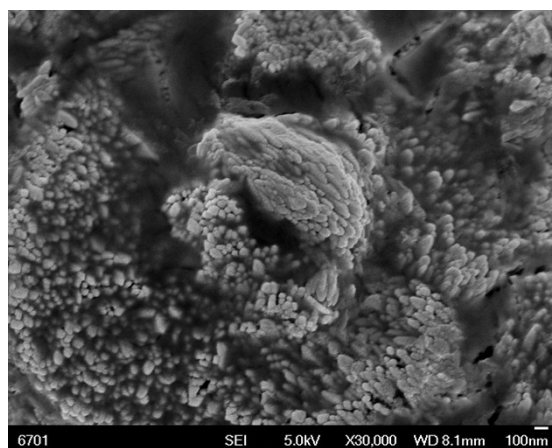


Fig. S10 SEM image of CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs before and after the HER stability measurements.

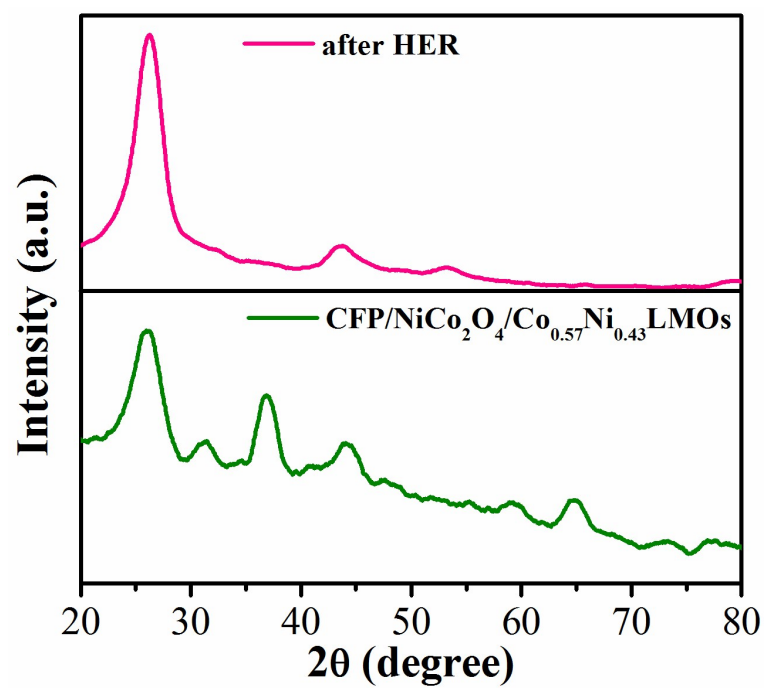


Fig. S11 XRD spectra of CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs before and after the HER stability measurements.

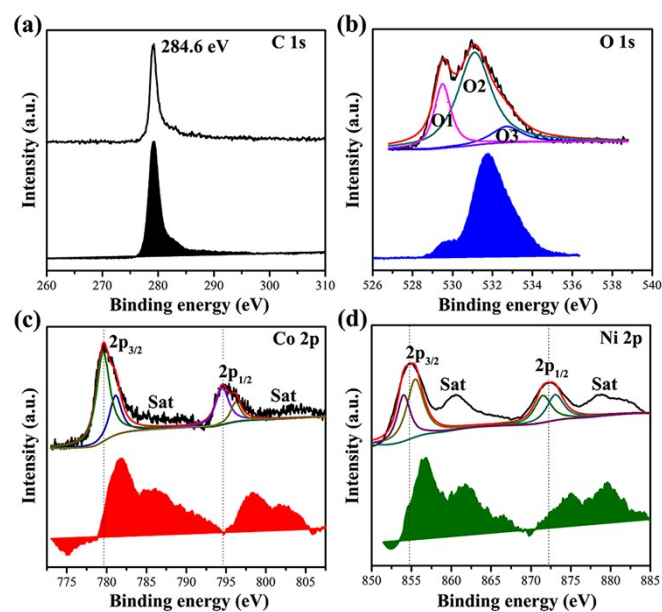


Fig. S12 XPS spectra of (a) C 1s, (b) Co 2p, (d) Ni 2p and (d) O 1s of CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs before and after the HER stability measurements.

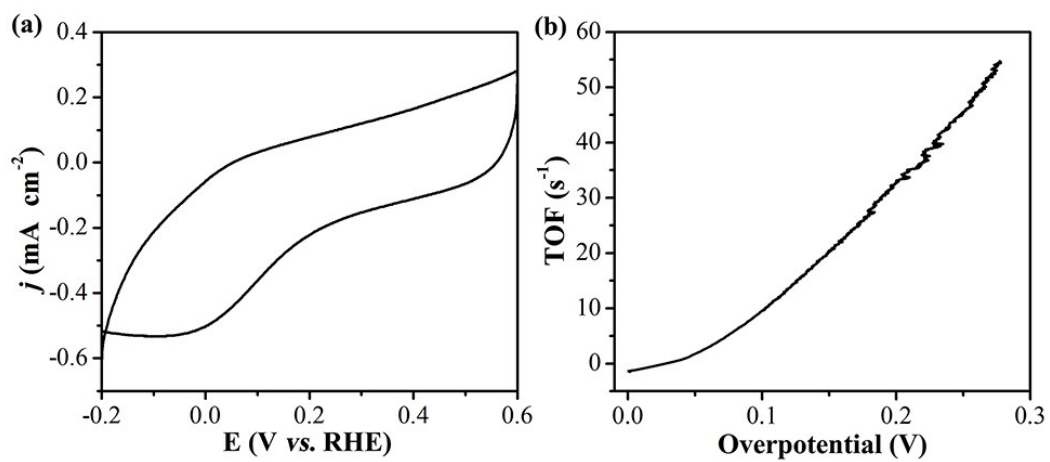


Fig. S13 (a) CV of the CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs in pH = 7 phosphate buffer between -0.2 V and 0.6 V (vs. RHE) with a scan rate of 50 mV s⁻¹. (b) Turnover frequencies of the CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs catalyst in 0.5 M H₂SO₄.

Table S1. Comparison of OER performance in alkaline media of CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs with other OER electrocatalysts

Catalyst	Tafel slope (mV dec ⁻¹)	Current density (j_0 , mA cm ⁻²)	η at the corresponding j (mV)	Ref.
NiCo ₂ O ₄	32	285	0.8 (V vs. Ag/AgCl)	<i>Angew. Chem. Int. Ed.</i> , 2015, 54 , 1-7
Co ₃ O ₄	70	10	1.5 (V vs. RHE)	<i>J. Am. Chem. Soc.</i> , 2014, 136 , 13925-13931
Zn _x Co _{3-x} O ₄	51	10	320	<i>Chem. Mater.</i> , 2014, 26 , 1889-1895
NiCo-LDH	40	10	367	<i>Nano Lett.</i> , 2015, 15 , 1421-1427
NiCo ₂ O ₄	63.1	10	320	<i>Nanoenergy</i> , 2014, 11 , 021 333-340
NiCo ₂ O ₄ /FTO	292	5	165	<i>J. Phys. Chem. C</i> , 2014, 118 , 25939-25946
		10	323	
NiFe-LDH	30	10	235	<i>ACS Appl. Mater. Interfaces</i> , 2014, 6 , 7918-7925
α -Ni(OH) ₂	42	10	331	<i>J. Am. Chem. Soc.</i> , 2014, 136 , 7077-7084
Ni-Fe nanoparticles	40	10	280	<i>Langmuir</i> , 2014, 30 , 7893-7901
NG-CoSe ₂	40	10	366	<i>ACS Nano.</i> , 2014, 8 , 3970-3978
Mn _x Co _{3-x} O _{4-δ}	85	10	1.58 (V vs. RHE)	<i>Chem. Eur. J.</i> , 2014, 20 , 1 - 9
CFP/NiCo ₂ O ₄ /Co _{0.53} Ni _{0.47} LMOs	63	10	340	In this work
		20	380	

Table S2 OER Electrochemical Parameters of CFP, CFP/NiCo₂O₄ and CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs.

Catalyst	R_s (Ohm)	R_{ct} (Ohm)
CFP	14.6	330.4
CFP/NiCo ₂ O ₄	19.4	13.5
CFP/NiCo ₂ O ₄ /Co _{0.57} Ni _{0.43} LMOs	15.7	5.2

Table S3. Comparison of HER performance in acid media of CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs with other HER electrocatalysts

Catalyst	Tafel slope (mV dec ⁻¹)	Current density (j_0 , mA cm ⁻²)	η at the corresponding j (mV)	Ref.
WP ₂	57	10	161	<i>ACS Catal.</i> , 2015, 5 , 145-149
NiP ₂ NS/CC	51	10	75	<i>Nanoscale</i> , 2014, 6 , 13440-13445
		100	204	
MoP	45	10	90	<i>Adv. Mater.</i> , 2014, 26 , 5702-5707
		20	105	
MoO ₃ - MoS ₂ /FTO	50-60	10	310	<i>Nano Lett.</i> , 2011, 11 ,4168-4175
CoP/CNT	54	2	70	<i>Angew. Chem. Int. Ed.</i> , 2014, 53 , 6710 –6714
		10	122	
NiMnN _x /C	35.9	2	170	<i>Angew. Chem. Int. Ed.</i> , 2012, 51 , 6131-6135
Ni ₂ P hollow nanoparticles	46	10	116	<i>J. Am. Chem. Soc.</i> , 2013, 135 , 9267-9270
		100	180	
CoP hollow nanoparticles	50	20	85	<i>Angew. Chem. Int. Ed.</i> , 2014, 53 ,5427-5430
bulk Mo ₂ C	56	1	150	<i>Angew. Chem. Int. Ed.</i> , 2012, 51 , 12703-12706
defect-rich MoS ₂	50	13	200	<i>Adv. Mater.</i> , 2013, 25 , 5807-5813
Co-NRCNTs	69	1	140	<i>Angew. Chem. Int. Ed.</i> , 2014, 53 , 4372-4376
CFP/NiCo ₂ O ₄ / Co _{0.57} Ni _{0.43} LMOs	34	10	52	In this work
		100	65.5	

Table S4 HER Electrochemical Parameters of CFP, CFP/NiCo₂O₄ and CFP/NiCo₂O₄/Co_{0.57}Ni_{0.43}LMOs.

Catalyst	R_s (Ohm)	R_{ct} (Ohm)
CFP	2.7	532.9
CFP/NiCo ₂ O ₄	2.7	189.3
CFP/NiCo ₂ O ₄ /Co _{0.57} Ni _{0.43} LMOs	1.5	4.5