## SUPPLEMENTARY INFORMATION

Fluorine containing amino acids: Synthesis and peptide coupling of aminoacids containing the all-cis tetrafluorocyclohexyl motif
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## 1. General Experimental

All reactions were carried out in oven-dried glassware under an argon atmosphere using a double vacuum manifold with the inert gas passing through a bed of silica gel and molecular sieves. Petrol refers to the petroleum ether fraction with a boiling point between $40-60{ }^{\circ} \mathrm{C}$. All chemicals were used as supplied. All NMR spectra were recorded using a Bruker Avance III 500, Bruker Avance II 400, Bruker Avance 300 or 500 spectrometers. The deuterated solvent was used for an internal deuterium lock. ${ }^{1} \mathrm{H}$ NMR spectra were recorded at either 300,400 or $500 \mathrm{MHz} .{ }^{13} \mathrm{C}$ NMR spectra were recorded using UDEFT pulse sequence and broadband proton decoupling at either 75,100 or $126 \mathrm{MHz} .{ }^{19} \mathrm{~F}$ NMR spectra were recorded at 282,376 or 470 MHz . All chemical shifts, $\delta$, are stated in units of parts per million (ppm), relative to a standard, for ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR the reference point is TMS ( $\delta_{\mathrm{H}}$ and $\delta_{\mathrm{C}}: 0.00 \mathrm{ppm}$ ). For ${ }^{19} \mathrm{~F}$ NMR the reference point is $\mathrm{CCl}_{3} \mathrm{~F}\left(\delta_{\mathrm{F}}: 0.00\right.$ ppm). Melting points were determined using a Griffin MPA350 or a Electrothermal 9100 melting point apparatus and are uncorrected. High and low resolution mass spectra were obtained by atmospheric pressure chemical ionisation (APCI), electospray ionization (ESI) and electron ionization (EI). ESI-MS spectra were recorded on a Waters Micromass LCT spectrometer in positive mode or negative mode. EI-MS spectra were recorded on a Waters Micromass GCT spectrometer. Values are reported as a ratio of mass to charge ( $\mathrm{m} / \mathrm{z}$ ).

## 2. Experimental Details and Analytical Data

## 1- Iodination of cis-1,2,4,5-tetrafluoro-3-phenylcyclohexane 3

Iodine ( $600 \mathrm{mg}, 2.36 \mathrm{mmol}$ ) was added to a solution of all cis-1,2,4,5-tetrafluoro-3phenylcyclohexane 3 ( $500 \mathrm{mg}, 2.15 \mathrm{mmol}$ ) in acetic acid ( 50 ml ), periodic acid $50 \%(\mathrm{w} / \mathrm{w})$ ( $0.123 \mathrm{~mL}, 0.43 \mathrm{mmol}$ ), conc. $\mathrm{H}_{2} \mathrm{SO}_{4} 95 \%(0.28 \mathrm{~mL}, 5.4 \mathrm{mmol})$, and water ( 10 mL ). The solution was heated for 16 h at $70{ }^{\circ} \mathrm{C}$ and then the mixture was left to cool to room temperature. The reaction was quenched by a adding of solution of saturated sodium bisulfite ( 30 mL ), then the mixture was washed with ethyl acetate ( $3 \times 50 \mathrm{~mL}$ ). The organic layers were combined and dried over sodium sulfate, filtered and the solvent evaporated
under reduced pressure ( $710 \mathrm{mg} 92 \%$ overall yield). The product was purified by flash column chromatography using diethyl ether / petrol (1:2), as an eluent. This gave 4:5:6:7 in a ratio of 1:5:15:4 respectively.

## All cis -1,2,4,5-tetrafluoro-3-(2-iodophenyl)-cyclohexane (4)



Colorless solid (28 mg, 3.6 \%) . mp 169-170 ${ }^{\circ} \mathrm{C}$; ${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{H}} 7.96(1 \mathrm{H}$, dd, J 7.9, 1.6 Hz, CH-6'), 7.90 (1H, dd, J 8.0, 1.3 Hz, CH-3'), 7.41 (1H, td, J 7.6, 1.3 Hz CH-5'), 7.05 (1H, td, J 7.6, 1.6 Hz, CH-4'), $5.11-4.91$ (2H, m, CHF-3), $4.82-4.57$ (2H, m, CHF-2), $3.13(1 \mathrm{H}, \mathrm{tt}, J 36.8,1.6 \mathrm{~Hz}, \mathrm{CH}-4), 2.84-2.69\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}-1\right), 2.55-2.46\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}^{-}}\right.$ 1); ${ }^{13} \mathbf{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{C}} 139.7,137.9,131.1(\mathrm{t}, J 6.6 \mathrm{~Hz}), 130.1,129.2,101.3$ (C-2'), 89.6 - 87.7 (m, CHF-3), 88.7 - 85.9 (m, CHF-2), 47.4 (m, CH-4), 27.3 (tt, J 22.1, 2.4 Hz, CH21); ${ }^{19}{ }^{\mathbf{F}}\left\{{ }^{1} \mathbf{H}\right\}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{F}}-189.9$ ( $2 \mathrm{~F}, \mathrm{dd}, J 7.7,5.6 \mathrm{~Hz}, \mathrm{CHF}-2$ ), -210.9 ( $2 \mathrm{~F}, \mathrm{dd}, J$ 7.7, $5.5 \mathrm{~Hz}, \mathrm{CHF}-3$ ); $\left(\right.$ ESI $\left.^{+}\right)[2 \mathrm{M}+6 \mathrm{H}]^{+}$calcd for $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{IF}_{4}{ }^{+}$: 721.9684 found: 722.2411

## All cis -1,2,4,5-tetrafluoro-3-(2, 4-diiodophenyl)-cyclohexane (7)



Colorless solid ( $122 \mathrm{mg}, 15.8 \%$ ). mp $203-204{ }^{\circ} \mathrm{C} ;{ }^{\mathbf{1}} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{H}} 8.24(1 \mathrm{H}$, d, J $2.2 \mathrm{~Hz}, \mathrm{CH}-3$ '), 7.59 (1H, d, J 8.4, CH-6'), 7.38 (1H, dd, J 8.4, 2.2 Hz CH-5'), $5.10-4.84$ (2H, m, CHF-3), 4.87 - 4.55 (2H, m, CHF-2), 3.04 (1H, tt, J 36.2, 1.6 Hz, CH-4), 2.82 - 2.68 $\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}-1\right), 2.54-2.46\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathrm{B}}-1\right)$; ${ }^{\mathbf{1 3}} \mathbf{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{C}} 140.9,139.8$, 139.8, 139.2, 100.5, 99.9, 89.3 - 87.6 (m, CHF-3), 87.6 - 85.7 (m, CHF-2), 47.1 (m, CH-4),

All cis -1,2,4,5-tetrafluoro-3-(3'-iodophenyl)-cyclohexane 5, all cis -1,2,4,5-tetrafluoro-3-(4'-iodophenyl)-cyclohexane 6 (560 mg $72 \%$ ) were isolated as an inseparable mixture, and were not fully characterized at this stage, in ratio (1:3) respectively.

## General Procedure for the preparation of 9 and 10.

A flame dried, three-necked, round bottomed flask ( 25 mL ) equipped with an argon inlet adapter, reflux condenser, rubber septum, and magnetic stir bar was charged with zinc dust ( $190 \mathrm{mg}, 3.0 \mathrm{mmol}$ ) and iodine ( $38 \mathrm{mg}, 0.15 \mathrm{mmol}$ ). The flask is evacuated and flushed with argon three times and then DMF ( 1 mL ) a solution of iodoalanine ( 330 mg , 1.00 mmol ) in DMF ( 1 mL ) was added dropwise via syringe at $0^{\circ} \mathrm{C}$. The reaction was kept stirred at $0^{\circ} \mathrm{C}$ for 30 min to generate a solution of the zinc reagent. The ice bath is removed and the aryl iodide ( $300 \mathrm{mg}, 0.84 \mathrm{mmol}$ ), tris(dibenzylideneacetone)dipalladium ( 11 mg , 0.0125 mmol ), and Sphos ( $11 \mathrm{mg}, 0.025 \mathrm{mmol}$ ) were added and the reaction mixture stirred at $60^{\circ} \mathrm{C}$ for 16 h . The resulting mixture was poured into a conical flask containing water ( 10 mL ). Citric acid solution ( 5 ml of $10 \%$ ) was added in order to break up the black emulsion. The aqueous mixture was extracted into DCM ( $2 \times 60 \mathrm{~mL}$ ), and the combined organic layers are washed with of water ( 30 mL ) and of brine ( 30 mL ). The organic fractions were dried and filtered. Concentration under vacuum gave the product which was purified over silica gel using petrol /ethyl acetate/DCM, (7:2:1) as an eluent

Methyl-2S-2-(tert-Butoxycarbonylamino)-3-(4-(all-cis-2,3,5,6-tetrafluoro cyclohex-1-yl) phenyl)propanoate (9)


Colorless solid ( 216 mg , 59 \%). mp $186-187^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{20}+22.0\left(\mathrm{c}=1 \times 10^{-3}, \mathrm{CHCl}_{3}\right.$ ); ${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{H}} 7.39(2 \mathrm{H}, \mathrm{d}, J 7.6 \mathrm{~Hz}, \mathrm{CH}-2 '), 7.13(2 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{CH}-3 '), 5.02-4.82(3 \mathrm{H}$, m, CHF-3, NHBoc), $4.780-4.46$ ( $3 \mathrm{H}, \mathrm{m}, \mathrm{CHF}-2$, CHNHBoc), 3.71 (3H, s, COOCH ${ }_{3}$ ), 3.17-2.97 $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{PhCH}_{2}\right), 2.79-2.38\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}-\mathbf{4}, \mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathrm{B}} \mathbf{- 1}\right), 1.39\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right) ;{ }^{\mathbf{1 3}} \mathbf{C}$ NMR (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{C}} 172.3,155.1,136.0,134.3,129.7,129.4,90.1$ - 88.0 (m, CHF-3), 88.0 - 86.1 ( $\mathrm{m}, \mathbf{C H F}-2$ ), 80.0, 54.3, 52.3, 43.7, 38.0, 28.2, 27.1 (tt, $J 22.3,2.2 \mathrm{~Hz}, \mathbf{C H}_{2}-1$ ); ${ }^{19}{ }^{\mathbf{F}}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}$ NMR ( $282 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{F}}-190.7$ (2F, dd, J 7.7, $5.5 \mathrm{~Hz}, \mathrm{CHF}-2$ ), -210.3 ( $2 \mathrm{~F}, \mathrm{dd}, J 8.0,5.1 \mathrm{~Hz}$, CHF3); (ESI+) $m / z[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~F}_{4} \mathrm{NO}_{4}{ }^{+}: 456.1774$ found: 456.1762

Methyl-2S-(tert-Butoxycarbonylamino)-3-(3-(all-cis-2,3,5,6-tetrafluorocyclohex-1yl) phenyl) propanoate. (10)


Colorless solid ( $88 \mathrm{mg}, 24 \%$ ). mp $164-165{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{20}+35.0\left(\mathrm{c}=1 \times 10^{-3}, \mathrm{CHCl}_{3}\right.$ ); $\mathbf{}^{\mathbf{1}} \mathbf{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{H}} 7.33-7.27$ ( $3 \mathrm{H}, \mathrm{m}, \mathrm{CH}-2^{\prime}, \mathrm{CH}-4{ }^{\prime}, \mathrm{CH}-6 '$ ), $7.16-7.09\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}-5^{\prime}\right.$ '), $5.10-4.87$ (3H, m, CHF-3, NHBoc), 4.78 - 4.47 (3H, m, CHF-2, CHNHBoc), 3.72 (3H, s, $\mathrm{COOCH}_{3}$ ), $3.19-3.02\left(2 \mathrm{H}, \mathrm{m}, \mathrm{PhCH}_{2}\right), 2.83-2.40\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}-4, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}-1\right), 1.41(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right) ;{ }^{\mathbf{1 3}} \mathbf{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{C}} 172.1,155.0,136.7,135.8,130.2,129.2,128.9$, 127.8, 89.8 - 88.2 (m, 2C, CHF-3), 87.8 - 86.3 (m, 2C, CHF-2), 79.9, 54.3, 52.3, 43.8 (tt, 1 C J $17.5,5.8 \mathrm{~Hz} \mathbf{C H}-4), 38.3$ ), $28.2,27.1\left(\mathrm{tt}, J 22.3,2.2 \mathrm{~Hz}, \mathbf{C H}_{2}-1\right) ;{ }^{\mathbf{1 9}} \mathbf{F}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\} \mathbf{N M R}(282 \mathrm{MHz}$,
$\mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{F}}-190.7$ (2F, dd, J 7.1, $5.1 \mathrm{~Hz}, \mathrm{CHF}-2$ ), -210.0 (2F, dd, J 7.2, 5.0 Hz, CHF-3); (ESI+) $\mathrm{m} / \mathrm{z}[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~F}_{4} \mathrm{NO}_{4}{ }^{+}$: 456.1774 found: 456.1759
(2S)-2-(aminohydrochloride)-3-(4-(all- cis -2,3,4,5-tetrafluorocyclohex-1-yl) phenyl)propanoic (11)


A solution of 9 ( $70 \mathrm{mg}, 0.161 \mathrm{mmol}$ ) in $\mathrm{HCl} 6 \mathrm{M}: 1,4$-dioxane ( $1: 1$ ) ( 4 mL ) and anisole ( 26 $\mathrm{mg}, 0.24 \mathrm{mmol}$ ) was stirred at $70^{\circ} \mathrm{C}$ for 48 h , until TLC showed that the substrate had been consumed. The reaction mixture was diluted with water ( 10 mL ) and the aqueous washed with ethyl acetate ( $2 \times 15 \mathrm{ml}$ ). The aqueous was the evaporated under reduced pressure, to afford the hydrochloride salt $\mathbf{1 1}(52 \mathrm{mg}, 91 \%)$ as colorless needles. mp $273-274{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}{ }^{20}+60.0\left(\mathrm{c}=2 \times 10^{-4}, \mathrm{DMSO}\right) ;{ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO) $\delta_{\mathrm{H}} 8.36\left(3 \mathrm{H}, \mathrm{bs}, \mathrm{NH}_{3} \mathrm{Cl}\right)$, 7.45 (2H, d, J $7.8 \mathrm{~Hz}, \mathrm{CH}-2$ ',), 7.29 ( $2 \mathrm{H}, \mathrm{d}, \mathrm{J} 7.9 \mathrm{~Hz}, \mathrm{CH}-3^{\prime}$ ), 5.19 - 4.83 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{C} \underline{\mathrm{HF}}-3, \mathrm{CHF}-2$ ), $4.17\left(1 \mathrm{H}, \mathrm{t}, \mathrm{J} 6.0 \mathrm{~Hz}, \mathrm{CHNH}_{3} \mathrm{Cl}\right), 3.25-3.5\left(3 \mathrm{H}, \mathrm{m}, \mathrm{PhCH}_{2}, \mathrm{CH}-4\right), 2.46-2.29\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathrm{B}^{-}}\right.$ 1); ${ }^{13}$ C NMR ( $126 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO) $\delta_{\mathrm{C}} 170.8,136.1,134.5,130.0,129.4,90.8-89.1$ (m, CHF3), 88.7 - 88.3 (m, CHF-2), 53.7, 35.8, $42.0,27.4\left(\mathrm{tt}, J 22.1,3.0 \mathrm{~Hz}, \mathrm{CH}_{2}-1\right.$ ); ${ }^{19} \mathbf{F}\left\{{ }^{1} \mathbf{H}\right\}$ NMR (376 MHz, d ${ }_{6}$-DMSO) $\delta_{\mathrm{F}}-189.4$ (2F, dd, J 8.1, $4.6 \mathrm{~Hz}, \mathrm{CHF}-2$ ), -209.7 (2F, dd, J 8.1, 5.8 Hz , CHF-3); (ESI+) $m / z[\mathrm{M}-\mathrm{HCl}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{ClF}_{4} \mathrm{NO}_{2}{ }^{+}: 320.1273$ found 320.1263;
(2S)-2-(Aminohydrochloride)-3-(3-(all-cis-2,3,5,6-tetrafluorocyclohex-1-yl) phenyl)propanoic (12)


A solution of 10 ( $50 \mathrm{mg}, 0.115 \mathrm{mmol}$ ) $\mathrm{HCl} 6 \mathrm{M}: 1,4$-dioxane (1:1) (4 mL) and anisole (21 $\mathrm{mg}, 0.20 \mathrm{mmol}$ ) was stirred at $70^{\circ} \mathrm{C}$ for 48 hr , until TLC showed that the substrate was consumed. The reaction mixture was diluted with water ( 10 mL ) and washed with ethyl acetate ( $2 \times 15 \mathrm{ml}$ ). The aqueous was collected and evaporated under vacuum, to afford the hydrochloride salt 12 ( $38 \mathrm{mg}, 92 \%$ ) as colorless solid. mp $223-224^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{20}+42.0$ (c=2 $\times 10^{-4}$, DMSO); ${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO) $\delta_{\mathrm{H}} 8.41$ ( $3 \mathrm{H}, \mathrm{bs}, \mathrm{NH}_{3} \mathrm{Cl}$ ), $7.24-7.17$ ( $4 \mathrm{H}, \mathrm{m}, \mathrm{CH}-$ 2', CH-4', CH-5', CH-6'), 5.19 - 4.87 (4H, m, CHF-3, CHF-2), 4.13 (1H, bs, $\mathrm{CHNH}_{3} \mathrm{Cl}$ ), 3.14 $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{PhCH}_{2}\right), 2.44-2.36\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}-\mathbf{4}, \mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathrm{B}} \mathbf{- 1}\right),{ }^{\mathbf{1 3}}{ }^{\mathbf{C}} \mathbf{~ N M R}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{C}} 170.8$, 137.5, 135.6, 130.5, 129.1, 128.8, 128.1, 90.9 - 89.2 (m, 2C, CHF-3), 88.1 - 86.7 (m, 2 C, CHF-2), 67.7, 53.7, 36.3.0, 27.5 (bt, $\mathbf{C H}_{2}-1$ ); ${ }^{\mathbf{1 9}}{ }^{\mathbf{F}}\left\{{ }^{1} \mathbf{H}\right\}$ NMR ( $470 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO) $\delta_{\mathrm{F}}-189.4$ (2F, bs, CHF-2), -209.4 (2F, m, CHF-3); (ESI+) m/z [M-HCl + H] ${ }^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{ClF}_{4} \mathrm{NO}_{2}{ }^{+}$: 320.1273 found 320.1266; (ESI-) $m / z \quad[\mathrm{M}-\mathrm{H}]^{+}$calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{ClF}_{4} \mathrm{NO}_{2}{ }^{+}: 354.0962$ found 354.0890;
(2S)-2-(tert-Butoxycarbonylamino)-3-(4-(all cis-2,3,5,6-tetrafluorocyclohex-1-yl) phenyl)propanoic. (13)


Di-tert-butyl dicarbonate ( $45 \mathrm{mg}, 0.2 \mathrm{mmol}$ ), and sodium bicarbonate ( $46 \mathrm{mg}, 0.54 \mathrm{mmol}$ ) was added to a solution of $\mathbf{1 1}(60 \mathrm{mg} .0 .17 \mathrm{mmol})$ in mixture of water and THF 1:1 ( 3 mL ). The reaction was stirred at $0^{\circ} \mathrm{C}$ for 1 h then left to come to ambient over 16 h . The reaction
mixture was then extracted into diethyl ether ( $2 \times 20 \mathrm{~mL}$ ) and the aqueous layer was acidified to pH 2 with HCl 1 M , and then extracted into ethyl acetate $(2 \times 30)$. The organic layers was dried over sodium sulfate, filtered and evaporated under vaccum to afford N boc amino acid 13 ( $66 \mathrm{mg}, 93 \%$ ). mp $189-190^{\circ} \mathrm{C} .[\alpha]_{\mathrm{D}}{ }^{20}+50\left(\mathrm{c}=2 \times 10^{-4}, \mathrm{CHCl}_{3}\right.$ ); ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{H}} 7.45(2 \mathrm{H}, \mathrm{d}, J 7.7 \mathrm{~Hz}, \mathrm{CH}-2 '), 7.20(2 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{CH}-3$ '), $5.09-4.88$ (3H, m, CHF-3, NHBoc), $4.74-4.48$ (3H, m, CHF-2, CHNHBoc), 3.23-3.05 (2H, m, PhCH $)_{2}$, 2.78 $2.49\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}-\mathbf{4}, \mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathbf{B}} \mathbf{- 1}\right), 1.42\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right) ;{ }^{\mathbf{1 3}} \mathbf{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{C}} 175.8$, 155.3, 135.7, 134.5, 129.9, 129.4, 90.2 - 88.1 (m, CHF-3), 88.1 - 86.1 (m, CHF-2), 85.3, 54.3, 43.6, 37.2, 28.3, 27.1 ( $\mathrm{tt}, \mathrm{J} 22.3,2.2 \mathrm{~Hz}, \mathrm{CH}_{2}-1$ ); ${ }^{\mathbf{1 9}}{ }^{\mathbf{F}}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\} \mathbf{N M R}\left(470 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) $\delta_{\mathrm{F}}-190.1$ (2F, bt, CHF-2), -209.4 (2F, bt, CHF-3); (ESI+) m/z [M+Na] calcd for $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{~F}_{4} \mathrm{NO}_{4}{ }^{+}$: 442.1618 found 442.1605;

Methyl-(2S)-2 [N-((2S)-2-(tert-Butoxycarbonylamino)-3-(4-(all cis-2,3,5,6-tetrafluorocyclohex-1-yl)phenyl)propanoyl]-2-amino-3-phenylpropanoate. (14)


EDCI hydrochloride ( $30 \mathrm{mg}, 0.15 \mathrm{mmol}$ ) and NMM ( $\mathrm{mg}, 0.48 \mathrm{mmol}$ ) were added to a solution of N -Boc amino acid $13(50 \mathrm{mg}, 0.119 \mathrm{mmol})(1 \mathrm{~mL})$ and HOBt ( $20 \mathrm{mg}, 0.148$ mmol ) in dry DMF and the solution was stirred for 5 min at $0^{\circ} \mathrm{C}$. L-Phenylalanine OMe hydrochloride 14 ( $31 \mathrm{mg}, 0.144 \mathrm{mmol}$ ) was the added and the solution was stirred at room temperature for 16 h . The reaction was diluted by sat ammonium chloride solution (10 mL ), stirred for 1 h and then extracted into ethyl acetate ( $2 \times 30 \mathrm{~mL}$ ). The organic layer was washed with $\mathrm{NaHCO}_{3} 10 \%(20 \mathrm{~mL})$, and brine ( 10 mL ), dried and then the organic
solvent was evaporated under reduced pressure. The product was purified over silica gel using by ethyl acetate / petrol (1:1) as an eluent, to afford peptide 14 as a white solid (60 $\mathrm{mg}, 87 \%$ ). mp 136-137 ${ }^{\circ} \mathrm{C} .[\alpha]_{\mathrm{D}}{ }^{20}+20\left(\mathrm{c}=2 \times 10^{-4}, \mathrm{CHCl}_{3}\right.$ ); ${ }^{1} \mathbf{H} \mathbf{N M R}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{H}}$ 7.39 ( $2 \mathrm{H}, \mathrm{d}, J 7.7 \mathrm{~Hz}, \mathrm{CH}-2$ '), , 7.26-7.18 (5H, m, CH`,CH`), 6.98 ( $2 \mathrm{H}, \mathrm{d}, J 8.0 \mathrm{~Hz}, \mathrm{CH}^{\prime}, \mathrm{CH}^{`}$ ), 6.26 (1H, bd, C-1"-NH), 5.02 - 4.89 (3H, m, CHF-3, NHBoc), 4.78 (1H, bs, H-1`), \(4.70-4.52\) (2H, m, CHF-2), 4.34 (1H, bs, H-6"), 3.68 (3H, s, COOMe), \(3.11-3.00\) (4H, m, CH2-5`, CH2$\left.2^{\prime \prime}\right), 2.80-2.70\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}-\mathbf{1}\right), 2.56(1 \mathrm{H}, \mathrm{t}, J 37.1 \mathrm{~Hz}, \mathbf{H}-4), 2.48-2.43\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}-\mathbf{1}\right)$, $1.40\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$; ${ }^{13} \mathbf{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{C}} 171.3,170.5,155.2,136.5,135.5$, 134.3, 129.7, 129.5, 129.2, 128.5, 127.1, 89.7 - 88.2 (m, CHF-3), 87.8 - 86.2 (m, CHF-2), 55.7, $53.2,52.3,43.8,37.9,29.7,28.2,27.1\left(\mathrm{t}, J 22.0 \mathrm{~Hz}, \mathrm{CH}_{2}-1\right) ;{ }^{19} \mathbf{F}\left\{{ }^{1} \mathbf{H}\right\}$ NMR ( 470 MHz , $\mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{F}}-190.2$ (2F, m, CHF-2), -209.7 (2F, m, CHF-3); (ESI+) $m / z[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{30} \mathrm{H}_{36} \mathrm{~F}_{4} \mathrm{~N}_{2} \mathrm{O}_{5}{ }^{+}$: 603.2458 found 603.2440 .

Methyl-2S-2 [N-((S)-2-(aminotrifluoroacetic)-3-(4-(all cis-2,3,5,6-tetrafluorocyclohex-1-yl)phenyl)propanoyl]-2-amino-3-phenylpropanoate (15)


A solution of 14 ( $20 \mathrm{mg}, 0.0 .034 \mathrm{mmol}$ ) in a mixture of DCM and TFA ( $4: 1$ ) ( 2 mL ) was stirred at RT for 4 h , until TLC showed the consumption of starting. The reaction mixture was then diluted with water ( 10 mL ) and extracted into diethyl ether ( $2 \times 15 \mathrm{ml}$ ), and the organic layer was washed with water ( 20 mL ). The aqueous layers were collected and evaporated under vacuum and the product was purified using a C-18 coated silica cartridge
with water/methanol (1:1) as the eluent to afford trifluoroacetate salt 15 ( $18 \mathrm{mg}, 94 \%$ ) as colorless solid. mp $240^{\circ} \mathrm{C} .[\alpha]_{\mathrm{D}}{ }^{20}+45$ ( $\mathrm{c}=2 \times 10^{-4}, \mathrm{DMSO}$ ); ${ }^{1} \mathbf{H} \mathbf{N M R}\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{COCD}_{3}\right)$ $\delta_{\mathrm{H}} 7.46$ - 7.15 (9H, m, CH-2', CH-3', H-5`, H-6", H-7`), 5.21 - 4.79 ( $6 \mathrm{H}, \mathrm{m}, \mathrm{CHF}-2$, CHF-3, H$\mathbf{6}^{`}$, NH ), 4.68 ( $1 \mathrm{H}, \mathrm{m}, \mathbf{H}-1^{`}$ ), 3.69 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{COOMe}$ ), $3.40-3.15$ ( $4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}-5^{\prime}, \mathrm{CH}_{2}-2^{`}$ ), 3.12 - $2.99\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathbf{- 1}\right), 2.59-2.41$ ( $2 \mathrm{H}, \mathrm{m}, \mathbf{H}-4, \mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathbf{B}} \mathbf{- 1}$ ); ${ }^{\mathbf{1 3}} \mathbf{C}$ NMR ( 100 MHz , $\left.\mathrm{CD}_{3} \mathrm{COCD}_{3}\right) \delta_{\mathrm{C}} 170.2,167.0,159.9,138.5,137.3,134.6,129.8,129.6,129.4,129.3,129.1$, 128.3, 126.7, 126.7, 126.6, 115.7, 91.0 - 88.6 (m, 2C, CHF-3), 88.2 - 86.0 (m, 2C, CHF-2), 65.3, 57.1, 51.6, 42.6, 36.3, 36.1, 27.3 - $26.8\left(\mathrm{~m}, \mathrm{CH}_{2}-1\right) ;{ }^{19} \mathbf{F}\left\{{ }^{1} \mathbf{H}\right\} \quad$ NMR $(470 \mathrm{MHz}$, $\mathrm{CD}_{3} \mathrm{COCD}_{3}$ ) $\delta_{\mathrm{F}}-76.3\left(3 \mathrm{~F}, \mathrm{~s}, \mathrm{CF}_{3} \mathrm{COO}\right),-191.3(2 \mathrm{~F}, \mathrm{CHF}-2),-210.8(2 \mathrm{~F}, \mathrm{~m}, \mathrm{CHF}-3)$; (ESI+) $m / z[\mathrm{M}-\mathrm{TFA}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{27} \mathrm{H}_{29} \mathrm{~F}_{7} \mathrm{~N}_{2} \mathrm{O}_{5}{ }^{+}: 481.2036$ found 481.2094

## Methyl-2S-2- (aminohydrochloride)-3-(4-(all cis-2,3,5,6-tetrafluorocyclohex-1yl))phenylpropanoate. (16)



A solution of protected amino acid $9(90 \mathrm{mg}, 0.208 \mathrm{mmol})$ in mixture of HCl 4 M and ethyl acetate (1:1) (3mL) was stirred for 24 hr at room temperature, until the startingmaterial was consumed. The reaction mixture was the extracted into ethyl acetate ( $2 \times 30 \mathrm{ml}$ ), and the organic layers washed with water ( 20 mL ). The aqueous layers were collected and evaporated at reduced pressure, to afford hydrochloride salt 16 without further purification ( $74 \mathrm{mg}, 96 \%$ ) as colorless solid mp $240-241{ }^{\circ} \mathrm{C}[\alpha]_{\mathrm{D}}{ }^{20}+70.0$ (c= $2 \times 10^{-4}$, DMSO); ${ }^{1} \mathbf{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{D}_{2} 0$ ) $\delta_{\mathrm{H}} 7.45\left(2 \mathrm{H}, \mathrm{d}, J 8.1 \mathrm{~Hz}, \mathrm{CH}-2^{\prime}\right.$,), 7.21 ( $2 \mathrm{H}, \mathrm{d}, J$ 8.2, CH-3'), $5.25-5.03(2 \mathrm{H}, \mathrm{m}, \mathrm{CHF}-3), 4.97-4.77(2 \mathrm{H}, \mathrm{m}, \mathrm{CHF}-2), 4.33\left(1 \mathrm{H}, \mathrm{dd}, J 7.2,6.0 \mathrm{~Hz} \mathrm{CHNH}_{3} \mathrm{Cl}\right)$, $3.71\left(3 \mathrm{H}, \mathrm{s}, \mathrm{COOCH}_{3}\right), 3.28-3.04\left(3 \mathrm{H}, \mathrm{m}, \mathrm{PhCH}_{2}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}-\mathbf{1}\right.$ ), 2.49-2.35 (2H, m, CH-4,
$\mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathrm{B}} \mathbf{- 1}$ ); ${ }^{\mathbf{1 3}} \mathbf{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO) $\delta_{\mathrm{C}} 169.8,136.2,134.2,129.9,129.5,90.9-89.1$ (m, CHF-3), 88.5 - 86.3 (m, CHF-2), 53.5, $53.041 .9,35.8,27.3\left(\mathrm{t}, J 21.3, \mathrm{~Hz}^{2}, \mathbf{C H}_{2}-1\right.$ ); ${ }^{\mathbf{1 9}}{ }^{\mathbf{F}}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}$ NMR (282 MHz, $\mathrm{D}_{2}$ O) $\delta_{\mathrm{F}}-190.9(2 \mathrm{~F}, \mathrm{dd}, J 7.7,5.5 \mathrm{~Hz}, \mathrm{CHF}-2),-210.4(2 \mathrm{~F}, \mathrm{dd}, J 8.3,4.4 \mathrm{~Hz}$, CHF-3); (ESI+) m/z [M-HCl+Na] ${ }^{+}$calcd for $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{ClF}_{4} \mathrm{NO}_{2}{ }^{+}$: 356.1250 found 356.1237

Methyl-2S-2N [(S-2-(tert-Butoxycarbonylamino)-3-phenylpropanoyl]-3-(4-(all cis-2,3,5,6-tetrafluorocyclohex-1-yl)phenyl)-2-amino-propanoate (17)


HOBt ( $19 \mathrm{mg}, 0.14 \mathrm{mmol}$ ), EDCI hydrochloride ( $34 \mathrm{mg}, 0.14 \mathrm{mmol}$ ) and DIPEA ( $63 \mathrm{mg}, 0.48$ mmol ) were added to a solution of N-Boc-L-phenylalanine acid ( $34 \mathrm{mg}, 0.128 \mathrm{mmol}$ ) in dry DMF ( 1 mL ) and the solution was stirred for 5 min at $0^{\circ} \mathrm{C}$. Amino acid hydrochloride 16 (40 $\mathrm{mg}, 0.108 \mathrm{mmol}$ ) was then added and the solution was stirred at room temperature for 12 $h$. The reaction was then quenched by the addition of sat. ammonium chloride ( 10 mL ), stirred for a further 1 h and was then extracted into ethyl acetate $(2 \times 30 \mathrm{~mL})$. The organic layer was washed with $\mathrm{NaHCO}_{3} 10 \%(20 \mathrm{~mL})$ and brine ( 20 mL ), dried and the solvent evaporated under reduced pressure. The product was purified over silica gel using ethyl acetate / petrol (1:2) as an eluent to afford the target peptide as a white solid ( $58 \mathrm{mg}, 92$ \%). mp 194-195 ${ }^{\circ} \mathrm{C}[\alpha]_{\mathrm{D}}{ }^{20}+25.0\left(\mathrm{c}=2 \times 10^{-4}, \mathrm{CHCl}_{3}\right.$ ); ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{H}} 7.35$ ( $2 \mathrm{H}, \mathrm{d}, J 7.7 \mathrm{~Hz}, \mathrm{CH}-2$ '), 7.27 (2H, d, J 7.0, CH-3'), $7.26-7.18\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}{ }^{`}\right), 7.0(2 \mathrm{H}, \mathrm{d}, J 8.0 \mathrm{~Hz}$, CH`), 6.38 (1H, d, J 7.7 Hz, C1`-NH), 5.02 - 4.90 (3H, m, CHF-3, NHBoc), 4.77 (1H, bs, H-6`), 4.71 - 4.53 ( \(2 \mathrm{H}, \mathrm{m}, \mathrm{CHF}-2\) ), 4.34 ( \(1 \mathrm{H}, \mathrm{bs}, \mathbf{H - 2 ` )}\) ), 3.67 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{COOMe}$ ), $3.11-2.99$ ( $4 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2}-5^{`}, \mathrm{CH}_{2}-3^{\prime \prime}\right), 2.78-2.68\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}-\mathbf{1}\right), 2.56(1 \mathrm{H}, \mathrm{t}, J 37.1 \mathrm{~Hz}, \mathrm{H}-4), 2.48-2.42(1 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathrm{B}} \mathbf{- 1}$ ), $1.40\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$; ${ }^{\mathbf{1 3}} \mathbf{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{C}} 171.2,170.9,155.4,136.4$, 135.6, 134.4, 129.7, 129.4, 128.7, 127.0, 89.8 - 88.1 (m, CHF-3), 87.7 - 86.3 (m, CHF-2), 55.8, 53.2, $52.343 .6,38.2,37.5,37.0,28.2,27.1\left(\mathrm{tt}, J 22.0,2.3 \mathrm{~Hz}, \mathbf{C H}_{2}-1\right.$ ), $\left.{ }^{\mathbf{1 9}}{ }^{\mathbf{F}\{ }{ }^{1} \mathbf{H}\right\} \mathbf{N M R}$ ( $470 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{F}}-190.2$ (2F, m, CHF-2), -209.7 (2F, m, CHF-3); (ESI+) $m / z[\mathrm{M}+\mathrm{Na}]^{+}$ calcd for $\mathrm{C}_{30} \mathrm{H}_{36} \mathrm{~F}_{4} \mathrm{~N}_{2} \mathrm{O}_{5}{ }^{+}$: 603.2458 found 603.2444 .

Methyl-2S-2N [(S-2-(ammoniumtrifluoracetate)-3-phenylpropanoyl]-3-(4-(all cis-2,3,5,6-tetrafluorocyclohex-1-yl)phenyl)-2-amino-propanoate (19a)


A solution of 17 ( $15 \mathrm{mg}, 0.025 \mathrm{mmol}$ ) in a mixture of DCM and TFA ( $4: 1$ ) ( 2 mL ) was stirred at room temperature and then the reaction mixture was diluted by water ( 10 mL ) and extracted into diethyl ether ( $2 \times 15 \mathrm{ml}$ ). The organic layer was then washed with water ( 20 mL ) the aqueous layer collected and evaporated under vacuum, to afford the trifluoroacetate salt 19a ( $13 \mathrm{mg}, 89 \%$ ) as a colorless solid. $\mathbf{~ m p}$ decompose at $230^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{20}$ +40.0 ( $c=2 \times 10^{-4}$, DMSO); ${ }^{1} \mathbf{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{COCD}_{3}$ ) $\delta_{\mathrm{H}} 7.50\left(2 \mathrm{H}, \mathrm{d}, J 7.6 \mathrm{~Hz}, \mathrm{CH}-2^{\prime}\right)$, 7.35 (2H, d, J 8.6, CH-3'), 7.30-7.10 (5H, m, H-5`, H-6", H-7`), 5.22 - 4.77 (6H, m, CHF-2, CHF-3, H-2", C1"-NH), 4.68 (1H, dd, J 9.8, $4.7 \mathrm{~Hz}, \mathrm{H}^{\prime} \mathbf{6}^{`}$ ), 3.70 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{COOMe}$ ), $3.37-3.14$ ( $4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathbf{2}}-5^{\prime}, \mathrm{CH}_{\mathbf{2}}-\mathbf{3}^{\prime \prime}$ ), $3.09-3.01\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathbf{- 1}\right.$ ), $2.55-2.40\left(2 \mathrm{H}, \mathrm{m}, \mathbf{H}-4, \mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathrm{B}} \mathbf{- 1}\right.$ ),
${ }^{13}$ C NMR ( $126 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{COCD}_{3}$ ) $\delta_{\mathrm{C}} 171.2,166.6,159.4$ (1C, q. J 36.0, Hz, CF $\mathrm{CO}_{3} \mathrm{COO}$ ), 136.4, 135.6, 135.1, 129.4, 129.4, 129.2, 128.6, 127.3, 116.2 (q, 1C, J $288.9 \mathrm{~Hz}_{2} \mathrm{CF}_{3} \mathrm{COO}$ ), ( 90.6 89.0 (m, 2C, CHF-3), 87.8 - 86.3 (m, 2C, CHF-2), 63.9, 54.0, 51.7, 42.7, 37.0, 36.1, 27.0 (tt, J 21.7, $3.5 \mathrm{~Hz}, \mathrm{CH}_{2}-1$ ); ${ }^{19} \mathbf{F}\left\{{ }^{1} \mathbf{H}\right\}$ NMR ( $282 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{COCD}_{3}$ ) $\delta_{\mathrm{F}}-76.3$ (3F, s, $\mathrm{CF}_{3} \mathrm{COO}$ ), -191.3 (2F, dd, J 6.9, 1.8 Hz ,CHF-2), -210.7 (2F, dd, J 8.7, $2.6 \mathrm{~Hz}, \mathrm{CHF}-3$ ); (ESI+) m/z [M-TFA+H] ${ }^{+}$ calcd for $\mathrm{C}_{27} \mathrm{H}_{29} \mathrm{~F}_{7} \mathrm{~N}_{2} \mathrm{O}_{5}{ }^{+}: 481.2036$ found 481.2095.

## Methyl-2S-2N[((2S)-2-(9-Fluorenylmethoxycarbonylamino)-3-phenylpropanoyl]-3-(4-(all cis-2,3,5,6-tetrafluorocyclohex-1-yl)phenyl propanoate (18).



HBTU ( $27 \mathrm{mg}, 0.071 \mathrm{mmol}$ ), NMM ( $11 \mathrm{mg}, 0.10 \mathrm{mmol}$ ) and HOBt ( $10 \mathrm{mg}, 0.074 \mathrm{mmol}$ ) were added to a solution of N-Fmoc-phenylalanine ( $25 \mathrm{mg}, 0.064 \mathrm{mmol}$ ) $\mathbf{X}$ in dry DMF $(1 \mathrm{~mL})$ and the solution was stirred for 5 min at $0^{\circ} \mathrm{C}$. Amino acid hydrochloride $\mathbf{1 6}(20 \mathrm{mg}$, 0.054 mmol ) was the added and the solution was stirred at room temperature for 6 h when a saturated solution of ammonium chloride ( 10 mL ) was added and the reaction mixture was stirred for a further 10 min . The product was extracted into ethyl acetate ( $2 \times 30 \mathrm{~mL}$ ) and the organic layer was washed with $\mathrm{NaHCO}_{3} 10 \%(20 \mathrm{~mL})$ and brine ( 10 mL ) and dried over $\mathrm{MgSO}_{4}$. The product was purified by over silica gel using ethyl acetate / petrol (2:1) as an eluent to afford dipeptide 18 as white solid ( $31 \mathrm{mg}, 81 \%$ ). mp 210-212 ${ }^{\circ} \mathrm{C} .[\alpha]_{\mathrm{D}}{ }^{20}+$ $40.0\left(\mathrm{c}=2 \times 10^{-4}, \mathrm{DMSO}\right) ;{ }^{\mathbf{1}} \mathbf{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{COCD}_{3}\right) \delta_{\mathrm{H}} 7.84(2 \mathrm{H}, \mathrm{d}, J 7.6 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H})$,
7.61 (3H, bd, Ar-H, NHFmoc), 7.43-7.17 (13H, m, Ar-H), 6.67 (1H, d, J 8.6 Hz, NHCO), 5.17 4.70 ( $5 \mathrm{H}, \mathrm{m}, \mathrm{CHF}-3, \mathrm{CHF}-2$, CH-2`), $4.54-4.46(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ ) ), $4.32-4.24$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CH}-$ 9Fmoc), $4.20-4.10\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right.$-Fmoc), 3.67 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{COOMe}$ ), $3.20-3.00\left(4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}-5\right.$ ), $\left.\mathrm{CH}_{\mathbf{2}}-\mathbf{3}^{\prime \prime}\right), 2.95-2.88\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathbf{- 1}\right), 2.57-2.38\left(2 \mathrm{H}, \mathrm{m}, \mathbf{H}-4, \mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathbf{B}} \mathbf{- 1}\right) ;{ }^{\mathbf{1 3}} \mathbf{C}$ NMR (126 $\left.\mathrm{MHz}, \mathrm{CD}_{3} \mathrm{COCD}_{3}\right) \delta_{\mathrm{C}} 171.4,171.2,155.9,144.1,141.1,137.7,136.1,135.4,129.4,129.3$, 129.2, 128.2, 127.6, 127.0, 126.3, 125.3, 125.2, 119.9, 90.5 - 89.0 (2C, m, CHF-3), 87.8 86.4, 66.3, $56.1,53.6,51.5,47.0,42.6,37.7,36.8,27.0\left(\mathrm{tt}, J 22.2,2.7 \mathrm{~Hz}, \mathbf{C H}_{2}-1\right.$ ), ${ }^{19} \mathbf{F}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}$ NMR (282 MHz, $\mathrm{CD}_{3} \mathrm{COCD}_{3}$ ) $\delta_{\mathrm{F}}-191.3$ (2F, dd, J 6.3, $4.8 \mathrm{~Hz} \mathrm{CHF}-2$ ), -210.7 (2F, dd, J 7.7, 5.3 $\mathrm{Hz}, \mathbf{C H F}-3$ ); (ESI + ) $m / z[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{40} \mathrm{H}_{38} \mathrm{~F}_{4} \mathrm{~N}_{2} \mathrm{O}_{5}{ }^{+}$: 725.2615 found 725.2603.

Methyl-2S-2N [((2S)-2-(amino)-3-phenylpropanoyl]-3-(4-(all-syn-2,3,5,6-tetrafluorocyclohex-1-yl)phenyl-2-aminopropanoate (19b)


Diethylamine was added via syringe ( $0.012 \mathrm{~mL}, 0.11 \mathrm{mmol}$ ) to a solution of Fmocdipeptide 18 ( $28 \mathrm{mg}, 0.039 \mathrm{mmol}$ ) in DMF ( 0.5 mL ). The reaction was stirred for 2 hr at room temperature and then the solvent was evaporated under reduced pressure. The residue was washed with diethyl ether ( $3 \times 10 \mathrm{ml}$ ) and the insoluble product was collected and dried under vaccum, to afford free amine 19b without further purification ( $18 \mathrm{mg}, 94$ $\%$ ) as colorless solid. $\mathbf{m p}$ decompose $270{ }^{\circ} \mathrm{C} .[\alpha]_{\mathrm{D}}{ }^{20}+55.0\left(\mathrm{c}=2 \times 10^{-4}\right.$, DMSO)

## Alternative method:

A solution of piperidine in DMF ( $20 \%, 0.5 \mathrm{~mL}$ ) was added to Fmoc-dipeptide 18 ( 15 mg , 0.021 mmol ) at $0^{\circ} \mathrm{C}$ and the reaction was stirred for 2 h at room temperature. The reaction mixture was the diluted with water and evaporated underreduced pressure. The insoluble residue was washed with diethyl ether ( $3 \times 10 \mathrm{ml}$ ) and the product dried under vacuum, to afford the free amine $\mathbf{1 9 b}$ without further purification ( $10 \mathrm{mg}, 97 \%$ ) as colorless solid. $\mathbf{m p}$ decompose $270^{\circ} \mathrm{C} .[\alpha]_{\mathrm{D}}{ }^{20}+55.0$ ( $\mathrm{c}=2 \times 10^{-4}$, DMSO); ${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO) $\delta_{\mathrm{H}} 7.95$ ( $2 \mathrm{H}, \mathrm{d}, \mathrm{J} 11.6 \mathrm{~Hz}, \mathrm{NH}_{2}$ ), 7.39 - 7.01 ( $9 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$ ), 5.76 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{NHCO}$ ), $5.07-4.80$ ( $4 \mathrm{H}, \mathrm{m}$, CHF-3, CHF-2,), $4.05-3.95$ (2H, m, H-6`, H-2"), 3.40 (3H, s, COOMe), \(3.10-2.88\) ( \(4 \mathrm{H}, \mathrm{m}\), \(\left.\mathrm{CH}_{\mathbf{2}}-\mathrm{S}^{`}, \mathrm{CH}_{2}-3^{\prime \prime}\right), 2.67-2.62\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}-\mathbf{1}\right), 2.39-2.34(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4), 2.18-2.12(1 \mathrm{H}, \mathrm{m}\), $\mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathrm{B}} \mathbf{- 1}$ ); ${ }^{13} \mathbf{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO) $\delta_{\mathrm{C}} 166.8,137.0,136.1,135.4,130.4,130.2,129.2$, 128.6, 127.0, 90.7 - 89.2 (2C, m, CHF-3), 88.0 - 86.6 (2C, m, CHF-2), 66.3, 60.2, 55.9, 55.7, 55.3, 42.0, $27.3\left(\mathrm{~m}, \mathbf{C H}_{2}-1\right) ;{ }^{19} \mathbf{F}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\}$ NMR ( $470 \mathrm{MHz}, \mathrm{d}_{6}$-DMSO) $\delta_{\mathrm{F}}-189.4(2 \mathrm{~F}, \mathrm{~m}, \mathrm{CHF}-2$ ), -
209.7 (2F, m, CHF-3); (ESI + ) $m / z[\mathrm{M}+\mathrm{Na}]^{+}$calcd for $\mathrm{C}_{25} \mathrm{H}_{28} \mathrm{~F}_{4} \mathrm{~N}_{2} \mathrm{O}_{3}{ }^{+}$: 503.1934 found 503.1916.

Benzyl (2S, 5R, 6R)-6- [(N-((2S)-2-(tert-Butoxycarbonylamino)-3-(4-(all cis-2,3,5,6-tetrafluorocyclohex-1-yl)phenyl)propanoyl)amino)]-3,3-dimethyl-7-oxo-4-thia-1-azabicyclo[3.2.0]heptane-2-carboxylate (21).


HOBt ( $40 \mathrm{mg}, 0.3 \mathrm{mmol}$ ), EDCI hydrochloride ( $73 \mathrm{mg}, 0.38 \mathrm{mmol}$ ) and N -methylmorpholine ( $58 \mathrm{mg}, 0.57 \mathrm{mmol}$ ) were added to a solution of N -Boc amino acid 13 ( $80 \mathrm{mg}, 0.191 \mathrm{mmol}$ ) in dry DMF ( 1 mL ). The solution was stirred for 5 min at $0^{\circ} \mathrm{C}$, and then 6 -APA ptoluenesulfonate ( $100 \mathrm{mg}, 0.21 \mathrm{mmol}$ ) was added and the solution stirred at room temperature 12 h . The reaction was diluted with sat. ammonium chloride solution ( 10 mL ) and the stirred for 1 h . The product was extracted into ethyl acetate ( $2 \times 30 \mathrm{~mL}$ ) and the organic layer washed with $\mathrm{NaHCO}_{3} 10 \%(20 \mathrm{~mL})$ and brine ( 20 mL ) and then dried over $\mathrm{MgSO}_{4}$. Solvent removal gave the product which was purified over silica gel using ethyl acetate / petrol (1:1) as an eluent. This gave peptide 18 as a white solid ( $81 \mathrm{mg}, 60 \%$ ). mp $153-154{ }^{\circ} \mathrm{C} .[\alpha]_{\mathrm{D}}{ }^{20}+140.0\left(\mathrm{c}=2 \times 10^{-4}, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathbf{H} \mathbf{N M R}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta_{\mathrm{H}} 7.41(2 \mathrm{H}, \mathrm{d}, \mathrm{J}$ $7.8 \mathrm{~Hz}, \mathrm{CH}-2$ '), 7.38 - 7.34 (5H, m, CH2 Ph ), 7.21 ( $2 \mathrm{H}, \mathrm{d}, J 7.6 \mathrm{~Hz}$ CH-3`), 6.74 ( \(1 \mathrm{H}, \mathrm{d}, J 7.6 \mathrm{~Hz}\), C-6"-NH), 5.59 (1H, dd, J \(6.6,3.9 \mathrm{~Hz}, \mathbf{H}-6\) "), 5.49 (1H, d, J \(4.3 \mathrm{~Hz}, \mathbf{H}-5 \times\) ), 5.20 - 5.14 (2H, m, PhCH 2 ), 5.08 - 4.90 ( \(3 \mathrm{H}, \mathrm{m}, \mathrm{CHF}-3\), NHBoc), 4.78 (1H, bs, H-1`), $4.71-4.53$ (2H, m, CHF-2),

$\mathrm{t}, \mathrm{J} 37.3 \mathrm{~Hz}, \mathrm{H}-4), 2.49-2.42\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}-\mathbf{1}\right), 1.53\left(3 \mathrm{H}, \mathrm{s}, \mathrm{C} 3\right.$ " $\left.-\mathrm{CH}_{3 \mathrm{~A}}\right), 1.40\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, $1.38\left(3 \mathrm{H}, \mathrm{s}, \mathrm{C} 3 `-\mathrm{CH}_{3 \mathrm{~B}}\right) ;{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{C}} 173.1,170.9,167.5,155.4,136.2$, 134.6, 134.3, 129.7, 129.6, 128.79, 128.74, 128.71, 89.9 - 88.1 (m, CHF-3), 87.9 - 86.1 (m, CHF-2), 70.4, 67.7, 67.5, 64.8, 58.5, 55.7, 43.6, 37.6, 31.5, 29.7, 28.8, 28.4, 27.1 (tt, J 22.0, $2.8 \mathrm{~Hz}, \mathbf{C H}_{2}-1$ ), 26.8 ; ${ }^{\mathbf{1 9}}{ }^{\mathbf{F}}\left\{{ }^{1} \mathbf{H}\right\}$ NMR ( $470 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{F}}-190.2(2 \mathrm{~F}, \mathrm{~m}, \mathrm{CHF}-2),-209.7(2 \mathrm{~F}$, m, CHF-3); (ESI+) $m / z[M+N a]^{+}$calcd for $\mathrm{C}_{35} \mathrm{H}_{41} \mathrm{~F}_{4} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{~S}^{+}: 730.2550$ found 730.2531.

Benzyl (2S, 5R, 6R)-6- [(N-((S)-2-(ammonium hydrochloride)-3-(4-(all cis-2,3,5,6 tetrafluorocyclohex-1-yl)phenyl)propanoyl)amino)]-3,3-dimethyl-7-oxo-4-thia-1-azabicyclo[3.2.0]heptane-2-carboxylate. (22).


Boc-dipeptide 18 ( $20 \mathrm{mg}, 0.028 \mathrm{mmol}$ ) was added to mixture of HCl 4 M and 1,4-dioxane $(1: 1)(2 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ and then and the mixture was stirred at room temperature for 1 h . The reaction was diluted with water ( 10 mL ) and extracted into diethyl ether ( $2 \times 15 \mathrm{ml}$ ). The organic layer was washed with water ( 20 mL ) and the aqueous collected and evaporated under vacuum. The product was purified using a C-18 coated silica cartridge using methanol-water as the eluent (1:1) to afford hydrochloride salt 19 ( $15 \mathrm{mg}, 83 \%$ ) as a colourless solid. mp decompose at $130^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{20}+95.0$ ( $\mathrm{c}=2 \times 10^{-4}$, DMSO); ${ }^{\mathbf{1}} \mathbf{H}$ NMR (500 $\mathrm{MHz}, \mathrm{CD}_{3} \mathrm{COCD}_{3}$ ) $\delta_{\mathrm{H}} 7.47-7.28$ ( $8 \mathrm{H}, \mathrm{m}$ Ar-H), 7.15 ( $1 \mathrm{H}, \mathrm{dd}, \mathrm{J} 8.1,1.6 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}$ ), 5.27 - 5.16 (3H, m, PhCH2, NH), 5.27 - 4.82 (4H, m, CHF-2, CHF-3), 4.75 - 4.70 (1H, m, H-6'), 4.14 $4.03\left(1 \mathrm{H}, \mathrm{m}, \mathbf{H}-6^{\prime}\right), 3.83-3.64\left(3 \mathrm{H}, \mathrm{m}, \mathrm{NH}_{3}\right), 3.43-3.34(1 \mathrm{H}, \mathrm{m}, \mathbf{H}-5 `), 3.30(2 \mathrm{H}, \mathrm{bs}$,
$\mathrm{PhCH}_{2}-5^{`}$ ), $3.14\left(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-3^{\prime \prime}\right)$ 2.68-2.33 (3H, m, $\mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathbf{- 1}, \mathrm{CH}_{\mathrm{A}} \mathbf{H}_{\mathrm{B}} \mathbf{- 1}, \mathbf{H}-4$ ), 1.57 (3H, s, C3"$\mathrm{CH}_{3} \mathrm{~A}$ ), 1.18-1.14 (3H, m, C3"-CH3B); ${ }^{13}$ C NMR ( $126 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{COCD}_{3}$ ) $\delta_{\mathrm{C}} 172.2,169.0$, 168.7, 137.6, 136.8, 135.1, 129.9, 128.9, 128.7, 128.4, 128.3, 90.7 - 89.0 (m, CHF-3), 88.1 86.3 (m, CHF-2), 74.1, 67.0, 66.5, 65.6, 44.8, 42.6, 41.5, 40.1, $29.4-28.4\left(3 \mathrm{C}, \mathrm{CH}_{2}-1, \mathrm{CH}_{3 A}\right.$, $\mathbf{C H}_{3 \mathrm{~B}}$ under the solvent peak) ; ${ }^{\mathbf{1 9}} \mathbf{F}\left\{{ }^{\mathbf{1}} \mathbf{H}\right\} \mathbf{N M R}$ ( 282 MHz , d6-Acetone) $\delta_{\mathrm{F}}-191.3$ (2F, m, CHF2), -210.9 (2F, m, CHF-3); HRMS (ESI+) m/z $[\mathrm{M}+\mathrm{H}]^{+}$calcd for $\mathrm{C}_{30} \mathrm{H}_{34} \mathrm{ClF}_{4} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{~S}^{+}$: 644.1895 found 644.2524 .

## Crystallographic Details for single X-ray structure of 11

## Data Collection

A colorless needle crystal of $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{ClF}_{4} \mathrm{NO}_{2}$ having approximate dimensions of $0.300 \times 0.030 \times 0.010 \mathrm{~mm}$ was mounted in a loop. All measurements were made on a Rigaku XtaLAB P100 diffractometer using multi-layer mirror monochromated $\mathrm{Cu}-\mathrm{K} \alpha$ radiation.

The crystal-to-detector distance was 30.10 mm .
Cell constants and an orientation matrix for data collection corresponded to a primitive monoclinic cell with dimensions:

$$
\begin{aligned}
& a=5.3827(19) \AA \\
& b=7.108(2) \AA \quad \beta=97.301(6)^{0} \\
& c=20.990(7) \AA \\
& V=796.6(4) \AA^{3}
\end{aligned}
$$

For $Z=2$ and F.W. $=355.76$, the calculated density is $1.483 \mathrm{~g} / \mathrm{cm}^{3}$. Based on the reflection conditions of:

OkO: $\mathrm{k}=2 \mathrm{n}$
packing considerations, a statistical analysis of intensity distribution, and the successful solution and refinement of the structure, the space group was determined to be:

$$
\text { P2 } 1 \text { (\#4) }
$$

The data were collected at a temperature of $-100 \pm 1^{\circ} \mathrm{C}$ to a maximum $2 \theta$ value of $136.4^{\circ}$. A total of 3983 oscillation images were collected. A sweep of data was done using $\phi$ scans from 0.0 to $200.0^{\circ}$ in $0.50^{\circ}$ step, at $\omega=0.0^{\circ}$ and $\chi=0.0^{\circ}$. The exposure rate was 40.0 [sec./O]. The detector swing angle was $-30.73^{\circ}$. A second sweep was performed using $\omega$ scans from -72.0 to $1.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=-90.0^{\circ}$. The exposure rate was 40.0 [sec. $/ 0$ ]. The detector swing angle was -30.730 . Another sweep was performed using $\omega$ scans from -113.0 to $-71.0^{\circ}$ in 0.500 step, at $\chi=0.00$ and $\phi=-45.0^{\circ}$. The exposure rate was 40.0 [sec./ O ]. The detector swing angle was -30.73 O . Another sweep was performed using $\phi$ scans from 0.0 to $200.0^{\circ}$ in $0.50^{\circ}$ step, at $\omega=0.0^{\circ}$ and $\chi=0.0^{\circ}$. The exposure rate was 40.0 [sec./ 0 ]. The detector swing angle
was $-67.73^{\circ}$. Another sweep was performed using $\omega$ scans from -85.0 to $-13.0^{0}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=90.0^{\circ}$. The exposure rate was 40.0 [sec./O]. The detector swing angle was -67.730 . Another sweep was performed using $\omega$ scans from -133.0 to -93.00 in $0.50^{\circ}$ step, at $\chi=0.00$ and $\phi=0.00$. The exposure rate was 40.0 [sec./0]. The detector swing angle was $-67.73^{0}$. Another sweep was performed using $\omega$ scans from 85.0 to $-12.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=-90.0^{\circ}$. The exposure rate was 40.0 [sec./O]. The detector swing angle was $-67.73^{\circ}$. Another sweep was performed using $\omega$ scans from -126.0 to $-86.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=-180.0^{\circ}$. The exposure rate was 40.0 [sec. $/ 0$ ]. The detector swing angle was -67.730 . Another sweep was performed using $\phi$ scans from 0.0 to $200.0^{\circ}$ in $0.50^{\circ}$ step, at $\omega=-25.0^{\circ}$ and $\chi=0.0^{\circ}$. The exposure rate was 40.0 [sec./ 0 ]. The detector swing angle was -110.730 . Another sweep was performed using $\omega$ scans from -128.0 to $-20.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=-135.0^{\circ}$. The exposure rate was 40.0 [sec./ ${ }^{\circ}$ ]. The detector swing angle was 110.730. Another sweep was performed using $\omega$ scans from -128.0 to $-20.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=45.0^{\circ}$. The exposure rate was 40.0 [ $\mathrm{sec} . /{ }^{\circ} \mathrm{O}$. The detector swing angle was $-110.73^{0}$. Another sweep was performed using $\omega$ scans from -115.0 to $20.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=-45.0^{\circ}$. The exposure rate was 40.0 [sec. $/^{\circ}$ ]. The detector swing angle was $-110.73^{\circ}$. Another sweep was performed using $\omega$ scans from -113.0 to $-23.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=135.0^{\circ}$. The exposure rate was 40.0 [sec. $/ 0$ ]. The detector swing angle was -110.730 . Another sweep was performed using $\omega$ scans from -141.0 to $-46.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=-90.0^{\circ}$. The exposure rate was 40.0 [ $\mathrm{sec} . /{ }^{\circ} \mathrm{O}$. The detector swing angle was $-110.73^{\circ}$. Another sweep was performed using $\omega$ scans from -141.0 to $-46.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=-$ $180.0^{0}$. The exposure rate was 40.0 [sec./ 0 ]. The detector swing angle was -110.730 . Another sweep was performed using $\omega$ scans from -141.0 to -59.00 in 0.500 step, at $\chi=0.0^{\circ}$ and $\phi=90.0^{\circ}$. The exposure rate was 40.0 [sec. $/^{\circ}$ ]. The detector swing angle was -110.730 . Another sweep was performed using $\omega$ scans from -141.0 to $-55.0^{0}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=0.0^{\circ}$. The exposure rate was 40.0 [ $\mathrm{sec} . / \mathrm{O}$ ]. The detector swing angle was $-110.73^{\circ}$. Another sweep was performed using $\omega$ scans from -141.0 to $-59.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=45.0^{\circ}$. The exposure rate was 40.0 [sec./ 0 ]. The detector swing angle was -110.730 . Another sweep was performed using $\omega$ scans from -141.0 to $-78.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=-45.0^{\circ}$. The exposure rate was 40.0 [sec./O]. The detector swing angle was -110.730 . Another sweep was performed using $\omega$ scans from -140.0 to $-77.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=-135.0^{\circ}$. The exposure rate was 40.0 [sec. $/ 0$ ]. The detector swing angle was -110.730 . Another sweep was performed using $\omega$ scans from -140.0 to $-56.0^{\circ}$ in $0.50^{\circ}$ step, at $\chi=0.0^{\circ}$ and $\phi=135.0^{0}$. The exposure rate was 40.0 [sec./O]. The detector swing angle was 110.730. Readout was performed in the 0.172 mm pixel mode.

## Data Reduction

Of the 7844 reflections were collected, where 2667 were unique ( $R_{i n t}=0.1436$ ); equivalent reflections were merged. Data were collected and processed using CrystalClear (Rigaku). 1

The linear absorption coefficient, $\mu$, for $\mathrm{Cu}-\mathrm{K} \alpha$ radiation is $26.058 \mathrm{~cm}^{-1}$. An empirical absorption correction was applied which resulted in transmission factors ranging from 0.214 to 0.974 . The data were corrected for Lorentz and polarization effects. A correction for secondary extinction ${ }^{2}$ was applied (coefficient $=0.038050$ ).

## Structure Solution and Refinement

The structure was solved by direct methods ${ }^{3}$ and expanded using Fourier techniques.

The crystal is a non-merohedral twin with twin law:

$$
\begin{array}{rrr}
-1.00000 & 0.00000 & 0.00000 \\
0.00000 & -1.00000 & 0.00000 \\
0.99100 & 0.00000 & 1.00000
\end{array}
$$

Twin component \#1 comprises $28.50 \%$ of the crystal.
The non-hydrogen atoms were refined anisotropically. Some hydrogen atoms were refined isotropically, and the rest were refined using the riding model. The final cycle of full-matrix least-squares refinement ${ }^{4}$ on $\mathrm{F}^{2}$ was based on 2667 observed reflections and 222 variable parameters and converged (largest parameter shift was 0.00 times its esd) with unweighted and weighted agreement factors of:

$$
\begin{gathered}
\mathrm{R} 1=\Sigma| | \mathrm{Fo}|-|\mathrm{Fc}|| / \Sigma|\mathrm{Fo}|=0.0966 \\
\mathrm{wR} 2=\left[\Sigma\left(\mathrm{w}\left(\mathrm{Fo}^{2}-\mathrm{Fc}^{2}\right)^{2}\right) / \Sigma \mathrm{w}\left(\mathrm{Fo}^{2}\right)^{2}\right]^{1 / 2}=0.2711
\end{gathered}
$$

The goodness of fit 5 was 0.98 . Unit weights were used. The maximum and minimum peaks on the final difference Fourier map corresponded to 0.59 and -0.33 $e^{-} / \AA^{3}$, respectively. The final Flack parameter 6 was $-0.03(6)$, indicating that the present absolute structure is correct. 7

Neutral atom scattering factors were taken from International Tables for Crystallography (IT), Vol. C, Table 6.1.1.4 8. Anomalous dispersion effects were
included in Fcalc ${ }^{9}$; the values for $\Delta f^{\prime}$ and $\Delta f^{\prime \prime}$ were those of Creagh and McAuley ${ }^{10}$. The values for the mass attenuation coefficients are those of Creagh and Hubbell ${ }^{11}$. All calculations were performed using the CrystalStructure ${ }^{12}$ crystallographic software package except for refinement, which was performed using SHELXL201313.

## References for the above paragraph

(1) CrystalClear: Data Collection and Processing Software, Rigaku Corporation (1998-2014). Tokyo 196-8666, Japan.
(2) Larson, A.C. (1970), Crystallographic Computing, 291-294. F.R. Ahmed, ed. Munksgaard, Copenhagen (equation 22, with V replaced by the cell volume).
(3) SIR2011: Burla, M. C., Caliandro, R., Camalli, M., Carrozzini, B., Cascarano, G. L., Giacovazzo, C., Mallamo, M., Mazzone, A., Polidori, G. and Spagna, R. (2012). J. Appl. Cryst. 45, 357-361.
(4) Least Squares function minimized: (SHELXL2013)
$\Sigma w\left(F_{0}{ }^{2}-F_{c}{ }^{2}\right)^{2} \quad$ where $w=$ Least Squares weights.
(5) Goodness of fit is defined as:

$$
\left[\Sigma w\left(\mathrm{~F}_{\mathrm{o}}{ }^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2} /\left(\mathrm{N}_{\mathrm{o}}-\mathrm{N}_{\mathrm{v}}\right)\right]^{1 / 2}
$$

where: $\quad N_{0}=$ number of observations
$N_{V}=$ number of variables
(6) Parsons, S. and Flack, H. (2004), Acta Cryst. A60, s61.
(7) Flack, H.D. and Bernardinelli (2000), J. Appl. Cryst. 33, 114-1148.
(8) International Tables for Crystallography, Vol.C (1992). Ed. A.J.C. Wilson, Kluwer Academic Publishers, Dordrecht, Netherlands, Table 6.1.1.4, pp. 572.
(9) Ibers, J. A. \& Hamilton, W. C.; Acta Crystallogr., 17, 781 (1964).
(10) Creagh, D. C. \& McAuley, W.J .; "International Tables for Crystallography", Vol C, (A.J.C. Wilson, ed.), Kluwer Academic Publishers, Boston, Table 4.2.6.8, pages 219-222 (1992).
(11) Creagh, D. C. \& Hubbell, J.H..; "International Tables for Crystallography", Vol C, (A.J.C. Wilson, ed.), Kluwer Academic Publishers, Boston, Table 4.2.4.3, pages

200-206 (1992).
(12) CrystalStructure 4.1: Crystal Structure Analysis Package, Rigaku Corporation (2000-2014). Tokyo 196-8666, Japan.
(13) SHELXL2013: Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.

## EXPERIMENTAL DETAILS

## A. Crystal Data

| Empirical Formula | $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{ClF}_{4} \mathrm{NO}_{2}$ |
| :---: | :---: |
| Formula Weight | 355.76 |
| Crystal Color, Habit | colorless, needle |
| Crystal Dimensions | $0.300 \times 0.030 \times 0.010 \mathrm{~mm}$ |
| Crystal System | monoclinic |
| Lattice Type | Primitive |
| Lattice Parameters | $\begin{aligned} & a=5.3827(19) \AA \\ & b=7.108(2) \AA \\ & c=20.990(7) \AA \\ & \beta=97.301(6) \mathrm{O} \\ & V=796.6(4) \AA^{3} \end{aligned}$ |
| Space Group | P21 (\#4) |
| $Z$ value | 2 |
| $\mathrm{D}_{\text {calc }}$ | $1.483 \mathrm{~g} / \mathrm{cm}^{3}$ |
| $\mathrm{F}_{000}$ | 368.00 |
| $\mu(\mathrm{CuK} \alpha)$ | $26.058 \mathrm{~cm}^{-1}$ |


| Diffractometer | XtaLAB P100 |
| :---: | :---: |
| Radiation | $\operatorname{CuK} \alpha(\lambda=1.54187 \AA)$ multi-layer mirror monochromated |
| Voltage, Current | 40kV, 30mA |
| Temperature | $-100.0^{\circ} \mathrm{C}$ |
| Detector Aperture | $83.8 \times 33.5 \mathrm{~mm}$ |
| Data Images | 3983 exposures |
| $\phi$ oscillation Range ( $\omega=0.0, \chi=0.0$ ) | 0.0-200.00 |
| Exposure Rate | $40.0 \mathrm{sec} . / \mathrm{O}$ |
| Detector Swing Angle | -30.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=-90.0$ ) | -72.0-1.00 |
| Exposure Rate | 40.0 sec./o |
| Detector Swing Angle | -30.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=-45.0$ ) | -113.0--71.00 |
| Exposure Rate | $40.0 \mathrm{sec} . / \mathrm{O}$ |
| Detector Swing Angle | -30.730 |
| $\phi$ oscillation Range ( $\omega=0.0, \chi=0.0$ ) | 0.0-200.00 |
| Exposure Rate | $40.0 \mathrm{sec} . /{ }^{\text {a }}$ |
| Detector Swing Angle | -67.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=90.0$ ) | -85.0--13.00 |
| Exposure Rate | 40.0 sec./o |


| Detector Swing Angle | -67.730 |
| :---: | :---: |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=0.0$ ) | -133.0--93.00 |
| Exposure Rate | 40.0 sec./o |
| Detector Swing Angle | -67.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=-90.0$ ) | -85.0--12.00 |
| Exposure Rate | 40.0 sec./o |
| Detector Swing Angle | -67.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=-180.0$ ) | -126.0--86.00 |
| Exposure Rate | 40.0 sec./o |
| Detector Swing Angle | -67.730 |
| $\phi$ oscillation Range ( $\omega=-25.0, \chi=0.0$ ) | 0.0-200.00 |
| Exposure Rate | 40.0 sec./0 |
| Detector Swing Angle | -110.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=-135.0$ ) | -128.0--20.00 |
| Exposure Rate | 40.0 sec./O |
| Detector Swing Angle | -110.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=45.0$ ) | -128.0--20.00 |
| Exposure Rate | 40.0 sec./o |
| Detector Swing Angle | -110.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=-45.0$ ) | -115.0--20.00 |
| Exposure Rate | 40.0 sec./O |


| Detector Swing Angle | -110.730 |
| :---: | :---: |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=135.0$ ) | -113.0--23.00 |
| Exposure Rate | 40.0 sec./O |
| Detector Swing Angle | -110.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=-90.0$ ) | -141.0--46.00 |
| Exposure Rate | 40.0 sec./0 |
| Detector Swing Angle | -110.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=-180.0$ ) | -141.0--46.00 |
| Exposure Rate | 40.0 sec./o |
| Detector Swing Angle | -110.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=90.0$ ) | -141.0--59.00 |
| Exposure Rate | $40.0 \mathrm{sec} . / \mathrm{O}$ |
| Detector Swing Angle | -110.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=0.0$ ) | -141.0--55.00 |
| Exposure Rate | 40.0 sec./o |
| Detector Swing Angle | -110.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=45.0$ ) | $-141.0-59.00$ |
| Exposure Rate | 40.0 sec./0 |
| Detector Swing Angle | -110.730 |
| $\omega$ oscillation Range ( $\chi=0.0, \phi=-45.0$ ) | -141.0--78.00 |
| Exposure Rate | 40.0 sec./o |
| Detector Swing Angle | -110.730 |

$\omega$ oscillation Range ( $\chi=0.0, \phi=-135.0$ )
Exposure Rate

Detector Swing Angle
$\omega$ oscillation Range ( $\chi=0.0, \phi=135.0$ )
Exposure Rate

Detector Swing Angle
$\omega$ oscillation Range ( $\chi=0.0, \phi=0.0$ )
Exposure Rate

Detector Swing Angle
Detector Position
Pixel Size
$2 \theta_{\text {max }}$
No. of Reflections Measured
Parsons quotients (Flack x parameter): 359
Corrections
$-140.0-77.0^{0}$
40.0 sec./o
$-110.730$
$-140.0--56.0^{0}$
40.0 sec./o
$-110.730$
$-91.0--90.50$
$4.0 \mathrm{sec} . /{ }^{\circ}$
$-30.73^{0}$
90.10 mm
0.172 mm
$136.4^{\circ}$
Total: 7844
Unique: $2667\left(\mathrm{R}_{\mathrm{int}}=0.1436\right)$

Lorentz-polarization
Absorption
(trans. factors: 0.214-0.974)
Secondary Extinction
(coefficient: 3.80500e-002)

## C. Structure Solution and Refinement

| Structure Solution | Direct Methods (SIR2011) |
| :---: | :---: |
| Refinement | Full-matrix least-squares on F2 |
| Function Minimized | $\Sigma \mathrm{w}\left(\mathrm{Fo}^{2}-\mathrm{Fc}^{2}\right)^{2}$ |
| Least Squares Weights | $\begin{aligned} & \mathrm{w}=1 /\left[\sigma^{2}\left(\mathrm{Fo}{ }^{2}\right)+(0.1611 \cdot \mathrm{P})^{2}\right. \\ & +0.0000 \cdot \mathrm{P}] \\ & \text { where } \mathrm{P}=\left(\operatorname{Max}\left(\mathrm{Fo}^{2}, 0\right)+2 \mathrm{Fc}^{2}\right) / 3 \end{aligned}$ |
| $2 \theta_{\text {max }}$ cutoff | 136.40 |
| Anomalous Dispersion | All non-hydrogen atoms |
| No. Observations (All reflections) | 2667 |
| No. Variables | 222 |
| Reflection/Parameter Ratio | 12.01 |
| Residuals: R1 (I>2.00б(I)) | 0.0966 |
| Residuals: R (All reflections) | 0.1193 |
| Residuals: wR2 (All reflections) | 0.2711 |
| Goodness of Fit Indicator | 0.978 |
| Flack parameter (Parsons' quotients $=359$ ) | -0.03(6) |
| Max Shift/Error in Final Cycle | 0.000 |
| Maximum peak in Final Diff. Map | $0.59 \mathrm{e}^{-/} / \AA^{3}$ |
| Minimum peak in Final Diff. Map | $-0.33 e^{-/ /} \AA^{3}$ |

Table 1. Atomic coordinates and $\mathrm{B}_{\text {iso }} / \mathrm{B}_{\text {eq }}$

| atom | x | y | z | $\mathrm{B}_{\text {eq }}$ |
| :--- | :--- | :--- | :--- | :--- |
| Cl1 | $0.7416(7)$ | $0.4443(6)$ | $0.9258(2)$ | $6.55(9)$ |
| F8 | $0.5256(17)$ | $0.6947(12)$ | $0.6644(5)$ | $6.09(18)$ |
| F9 | $0.9512(19)$ | $0.8566(13)$ | $0.6395(6)$ | $7.4(2)$ |
| F11 | $0.6905(19)$ | $0.3659(16)$ | $0.4920(5)$ | $7.2(2)$ |
| F12 | $0.3734(15)$ | $0.4077(15)$ | $0.5814(4)$ | $6.3(2)$ |
| O15 | $0.5873(19)$ | $-0.0674(18)$ | $0.9385(6)$ | $6.9(2)$ |
| O16 | $0.336(2)$ | $-0.2767(19)$ | $0.8827(7)$ | $7.2(3)$ |
| N17 | $0.228(2)$ | $0.1933(18)$ | $0.9441(7)$ | $5.9(2)$ |
| C1 | $0.522(2)$ | $0.303(2)$ | $0.7156(9)$ | $5.5(3)$ |
| C2 | $0.398(3)$ | $0.415(2)$ | $0.7561(8)$ | $6.1(3)$ |
| C3 | $0.228(3)$ | $0.332(2)$ | $0.7949(7)$ | $5.6(3)$ |
| C4 | $0.203(3)$ | $0.138(2)$ | $0.7964(8)$ | $5.6(3)$ |
| C5 | $0.333(3)$ | $0.029(2)$ | $0.7567(10)$ | $6.9(4)$ |
| C6 | $0.488(3)$ | $0.110(2)$ | $0.7161(10)$ | $6.2(3)$ |
| C7 | $0.697(3)$ | $0.3812(19)$ | $0.6689(8)$ | $5.3(3)$ |
| C8 | $0.743(3)$ | $0.593(2)$ | $0.6780(8)$ | $5.5(3)$ |
| C9 | $0.921(3)$ | $0.660(2)$ | $0.6314(9)$ | $6.0(3)$ |
| C10 | $0.830(3)$ | $0.614(3)$ | $0.5623(10)$ | $6.9(4)$ |
| C11 | $0.788(3)$ | $0.404(2)$ | $0.5560(8)$ | $5.5(3)$ |
| C12 | $0.611(3)$ | $0.331(2)$ | $0.6011(8)$ | $6.0(3)$ |
| C13 | $0.028(3)$ | $0.052(2)$ | $0.8403(7)$ | $5.5(3)$ |
| C14 | $0.154(3)$ | $0.010(2)$ | $0.9093(8)$ | $5.7(3)$ |
| C15 | $0.389(3)$ | $-0.1135(19)$ | $0.9114(9)$ | $6.2(3)$ |

$B_{e q}=8 / 3 \pi^{2}\left(U_{11}\left(a^{*}\right)^{2}+U_{22}\left(b b^{*}\right)^{2}+U_{33}\left(\mathrm{cc}^{*}\right)^{2}+2 \mathrm{U}_{12}\left(a a^{*} b b^{*}\right) \cos \gamma+2 U_{13}\left(a a^{*} c c^{*}\right) \cos \beta+2 U_{23}\left(b b^{*} c c^{*}\right) \cos \alpha\right)$

Table 2. Atomic coordinates and $\mathrm{B}_{\text {iso }}$ involving hydrogen atoms

| atom | $x$ | $y$ | $z$ | $B_{\text {iso }}$ |
| :--- | :--- | :---: | :--- | :--- |
| H16 | $0.46(4)$ | $-0.38(3)$ | $0.888(12)$ | 10.8258 |
| H17A | $0.27(5)$ | $0.15(3)$ | $0.988(4)$ | 8.8163 |
| H17B | $0.11(3)$ | $0.30(2)$ | $0.933(10)$ | 8.8163 |
| H17C | $0.38(2)$ | $0.24(3)$ | $0.930(10)$ | 8.8163 |
| H2 | 0.42637 | 0.54665 | 0.75783 | 7.368 |
| H3 | 0.13214 | 0.40909 | 0.81956 | 6.691 |
| H5 | 0.31536 | -0.10441 | 0.75724 | 8.291 |
| H6 | 0.57142 | 0.03250 | 0.68854 | 7.398 |
| H7 | 0.86264 | 0.31831 | 0.68033 | 6.404 |
| H8 | 0.81885 | 0.61802 | 0.72320 | 6.588 |
| H9 | 1.08745 | 0.59851 | 0.64333 | 7.177 |
| H10A | 0.95597 | 0.65384 | 0.53453 | 8.330 |
| H10B | 0.67170 | 0.68107 | 0.54833 | 8.330 |
| H11 | 0.95327 | 0.33873 | 0.56586 | 6.642 |
| H12 | 0.60066 | 0.19090 | 0.59731 | 7.220 |
| H13A | -0.04170 | -0.06680 | 0.82080 | 6.658 |
| H13B | -0.11395 | 0.13911 | 0.84285 | 6.658 |
| H14 | 0.03015 | -0.05643 | 0.93298 | 6.786 |

Table 3. Anisotropic displacement parameters

| atom | $\mathrm{U}_{11}$ | $\mathrm{U}_{22}$ | $\mathrm{U}_{33}$ | $\mathrm{U}_{12}$ | $\mathrm{U}_{13}$ | $\mathrm{U}_{23}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :---: |
| Cl1 | $0.071(2)$ | $0.0630(18)$ | $0.115(3)$ | $0.0164(17)$ | $0.0100(18)$ | $-0.000(2)$ |
| F8 | $0.060(5)$ | $0.056(5)$ | $0.117(6)$ | $0.004(4)$ | $0.021(4)$ | $0.001(4)$ |
| F9 | $0.076(6)$ | $0.070(6)$ | $0.135(8)$ | $-0.020(5)$ | $0.021(5)$ | $-0.010(5)$ |
| F11 | $0.073(6)$ | $0.109(8)$ | $0.095(6)$ | $-0.011(5)$ | $0.017(5)$ | $-0.008(5)$ |
| F12 | $0.054(4)$ | $0.087(6)$ | $0.100(6)$ | $-0.004(4)$ | $0.011(4)$ | $-0.013(5)$ |
| O15 | $0.057(6)$ | $0.060(5)$ | $0.148(9)$ | $-0.005(6)$ | $0.023(6)$ | $0.002(7)$ |
| O16 | $0.072(7)$ | $0.069(7)$ | $0.130(9)$ | $0.009(6)$ | $0.002(6)$ | $-0.001(6)$ |
| N17 | $0.067(7)$ | $0.058(6)$ | $0.096(8)$ | $0.007(6)$ | $0.004(7)$ | $-0.010(6)$ |
| C1 | $0.046(7)$ | $0.050(7)$ | $0.114(11)$ | $0.001(5)$ | $0.010(7)$ | $0.000(7)$ |
| C2 | $0.072(9)$ | $0.049(8)$ | $0.112(11)$ | $0.009(7)$ | $0.011(8)$ | $0.002(7)$ |
| C3 | $0.063(8)$ | $0.058(8)$ | $0.092(9)$ | $-0.005(6)$ | $0.013(7)$ | $0.002(7)$ |
| C4 | $0.057(8)$ | $0.055(7)$ | $0.103(10)$ | $0.011(7)$ | $0.015(7)$ | $0.004(7)$ |
| C5 | $0.062(9)$ | $0.049(8)$ | $0.153(16)$ | $-0.011(7)$ | $0.019(9)$ | $-0.001(9)$ |
| C6 | $0.062(8)$ | $0.058(8)$ | $0.116(12)$ | $0.001(7)$ | $0.017(8)$ | $-0.014(8)$ |
| C7 | $0.054(8)$ | $0.049(7)$ | $0.101(10)$ | $0.004(6)$ | $0.010(6)$ | $-0.003(6)$ |
| C8 | $0.060(8)$ | $0.066(9)$ | $0.083(9)$ | $-0.006(7)$ | $0.010(6)$ | $-0.012(6)$ |
| C9 | $0.059(8)$ | $0.058(8)$ | $0.112(11)$ | $-0.008(6)$ | $0.019(8)$ | $-0.010(7)$ |
| C10 | $0.040(8)$ | $0.086(12)$ | $0.138(15)$ | $0.004(7)$ | $0.017(8)$ | $0.007(10)$ |
| C11 | $0.064(8)$ | $0.058(9)$ | $0.091(8)$ | $-0.002(7)$ | $0.020(6)$ | $-0.005(6)$ |
| C12 | $0.057(8)$ | $0.075(10)$ | $0.099(10)$ | $0.008(7)$ | $0.019(7)$ | $-0.007(8)$ |
| C13 | $0.061(8)$ | $0.066(9)$ | $0.082(9)$ | $-0.003(6)$ | $0.003(7)$ | $0.001(7)$ |
| C14 | $0.064(8)$ | $0.063(7)$ | $0.089(8)$ | $-0.002(7)$ | $0.011(7)$ | $-0.003(7)$ |
| C15 | $0.051(8)$ | $0.047(7)$ | $0.141(13)$ | $-0.010(6)$ | $0.018(8)$ | $0.005(7)$ |

The general temperature factor expression: $\exp \left(-2 \pi^{2}\left(a^{*} U_{11} h^{2}+b^{*} U_{22} k^{2}+c^{*} U_{33} 1^{2}\right.\right.$ $\left.\left.+2 a^{*} b^{*} U_{12} h k+2 a^{*} c^{*} U_{13} h l+2 b^{*} c^{*} U_{23} k l\right)\right)$

Table 4. Fragment Analysis
fragment: 1 $\mathrm{Cl}(1)$
fragment: 2

| $\mathrm{F}(8)$ | $\mathrm{F}(9)$ | $\mathrm{F}(11)$ | $\mathrm{F}(12)$ | $\mathrm{O}(15)$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{O}(16)$ | $\mathrm{N}(17)$ | $\mathrm{C}(1)$ | $\mathrm{C}(2)$ | $\mathrm{C}(3)$ |
| $\mathrm{C}(4)$ | $\mathrm{C}(5)$ | $\mathrm{C}(6)$ | $\mathrm{C}(7)$ | $\mathrm{C}(8)$ |
| $\mathrm{C}(9)$ | $\mathrm{C}(10)$ | $\mathrm{C}(11)$ | $\mathrm{C}(12)$ | $\mathrm{C}(13)$ |
| $\mathrm{C}(14)$ | $\mathrm{C}(15)$ |  |  |  |

Table 5. Bond lengths ( $\AA$ )

| atom | atom | distance | atom | atom | distance |
| :--- | :--- | :--- | :--- | :--- | :--- |
| F8 | C8 | $1.376(18)$ | F9 | C9 | $1.415(18)$ |
| F11 | C11 | $1.406(18)$ | F12 | C12 | $1.404(17)$ |
| O15 | C15 | $1.191(19)$ | O16 | C15 | $1.32(2)$ |
| N17 | C14 | $1.52(2)$ | C1 | C2 | $1.39(2)$ |
| C1 | C6 | $1.38(2)$ | C1 | C7 | $1.55(2)$ |
| C2 | C3 | $1.43(2)$ | C3 | C4 | $1.39(2)$ |
| C4 | C5 | $1.39(3)$ | C4 | C13 | $1.53(2)$ |
| C5 | C6 | $1.39(3)$ | C7 | C8 | $1.53(2)$ |
| C7 | C12 | $1.48(2)$ | C8 | C9 | $1.53(2)$ |
| C9 | C10 | $1.51(3)$ | C10 | C11 | $1.51(3)$ |
| C11 | C12 | $1.52(2)$ | C13 | C14 | $1.55(2)$ |
| C14 | C15 | $1.53(2)$ |  |  |  |

Table 6. Bond lengths involving hydrogens ( $\AA$ )

| atom | atom | distance | atom | atom | distance |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 016 | H16 | 1.0(2) | N17 | H17A | 0.99(11) |
| N17 | H17B | 0.98(16) | N17 | H17C | 0.98(17) |
| C2 | H2 | 0.950 | C3 | H3 | 0.950 |
| C5 | H5 | 0.950 | C6 | H6 | 0.950 |
| C7 | H7 | 1.000 | C8 | H8 | 1.000 |
| C9 | H9 | 1.000 | C10 | H10A | 0.990 |
| C10 | H10B | 0.990 | C11 | H11 | 1.000 |
| C12 | H12 | 1.000 | C13 | H13A | 0.990 |
| C13 | H13B | 0.990 | C14 | H14 | 1.000 |

Table 7. Bond angles ( ${ }^{( }$)

| atom | atom | atom | angle | atom | atom | atom | angle |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| C2 | C1 | C6 | $119.1(16)$ | C2 | C1 | C7 | 124.0(13) |
| C6 | C1 | C7 | $116.9(15)$ | C1 | C2 | C3 | $120.3(14)$ |
| C2 | C3 | C4 | $119.7(15)$ | C3 | C4 | C5 | $118.9(15)$ |
| C3 | C4 | C13 | $118.7(14)$ | C5 | C4 | C13 | $122.3(14)$ |
| C4 | C5 | C6 | $121.5(15)$ | C1 | C6 | C5 | $120.4(17)$ |
| C1 | C7 | C8 | $112.1(13)$ | C1 | C7 | C12 | $112.5(12)$ |
| C8 | C7 | C12 | $112.5(13)$ | F8 | C8 | C7 | $111.8(12)$ |
| F8 | C8 | C9 | $106.6(12)$ | C7 | C8 | C9 | $109.4(13)$ |
| F9 | C9 | C8 | $107.7(13)$ | F9 | C9 | C10 | $110.6(14)$ |
| C8 | C9 | C10 | $113.2(13)$ | C9 | C10 | C11 | $109.1(16)$ |
| F11 | C11 | C10 | $107.9(13)$ | F11 | C11 | C12 | $109.9(12)$ |
| C10 | C11 | C12 | $112.4(14)$ | F12 | C12 | C7 | $110.5(13)$ |
| F12 | C12 | C11 | $107.8(12)$ | C7 | C12 | C11 | $111.8(13)$ |
| C4 | C13 | C14 | $114.2(13)$ | N17 | C14 | C13 | $110.0(12)$ |
| N17 | C14 | C15 | $108.1(12)$ | C13 | C14 | C15 | $113.6(14)$ |
| O15 | C15 | O16 | $126.1(15)$ | O15 | C15 | C14 | $122.8(14)$ |
| O16 | C15 | C14 | $111.0(13)$ |  |  |  |  |

Table 8. Bond angles involving hydrogens ( ${ }^{( }$)

| atom | atom | atom | angle | atom | atom | atom | angle |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| H16 | O16 | C15 | $118(13)$ | H17A | N17 | H17B | 123(18) |
| H17A | N17 | H17C | $109(18)$ | H17A | N17 | C14 | 100(12) |
| H17B | N17 | H17C | $99(17)$ | H17B | N17 | C14 | $115(10)$ |
| H17C | N17 | C14 | $110(13)$ | C1 | C2 | H2 | 119.9 |
| C3 | C2 | H2 | 119.9 | C2 | C3 | H3 | 120.2 |
| C4 | C3 | H3 | 120.2 | C4 | C5 | H5 | 119.3 |
| C6 | C5 | H5 | 119.3 | C1 | C6 | H6 | 119.8 |
| C5 | C6 | H6 | 119.8 | C1 | C7 | H7 | 106.4 |
| C8 | C7 | H7 | 106.4 | C12 | C7 | H7 | 106.4 |
| F8 | C8 | H8 | 109.7 | C7 | C8 | H8 | 109.7 |
| C9 | C8 | H8 | 109.7 | F9 | C9 | H9 | 108.4 |
| C8 | C9 | H9 | 108.4 | C10 | C9 | H9 | 108.4 |
| C9 | C10 | H10A | 109.9 | C9 | C10 | H10B | 109.9 |
| C11 | C10 | H10A | 109.9 | C11 | C10 | H10B | 109.9 |
| H10A | C10 | H10B | 108.3 | F11 | C11 | H11 | 108.8 |
| C10 | C11 | H11 | 108.8 | C12 | C11 | H11 | 108.8 |
| F12 | C12 | H12 | 108.9 | C7 | C12 | H12 | 108.9 |
| C11 | C12 | H12 | 108.9 | C4 | C13 | H13A | 108.7 |
| C4 | C13 | H13B | 108.7 | C14 | C13 | H13A | 108.7 |
| C14 | C13 | H13B | 108.7 | H13A | C13 | H13B | 107.6 |
| N17 | C14 | H14 | 108.4 | C13 | C14 | H14 | 108.4 |
| C15 | C14 | H14 | 108.4 |  |  |  |  |

Table 9. Torsion Angles $\left({ }^{( }\right)$
(Those having bond angles > 160 or $<20$ degrees are excluded.)

| atom1 | atom2 | atom3 | atom4 |  |  |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| C2 | C1 | C6 | C5 | $-0(2)$ | atom1 |  | atom2 | atom3 | atom4 | angle

Table 10. Possible hydrogen bonds

| Donor | H | Acceptor | D...A | D-H | H...A | D-H...A |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| O16 | H16 | $\mathrm{Cl} 1{ }^{1}$ | 3.004(13) | 1.0(2) | 2.1(2) | 163(21) |  |
| N17 | H17A | $\mathrm{Cl}_{1}{ }^{2}$ | 3.240(15) | 0.99(11) | 2.31(15) |  | 158(17) |
| N17 | H17A | 015 | 2.690(18) | 0.99(11) | 2.6(2) | 84(12) | intramo |
| N17 | H17B | $\mathrm{Cl}_{1}{ }^{3}$ | 3.153(14) | 0.98(16) | 2.25(18) |  | 154(15) |
| N17 | H17C | O 15 | 2.690(18) | 0.98(17) | 2.5(2) | 92(15) | intramol |

Symmetry Operators:
(1) $X, Y-1, Z$
(2) $-X+1, Y+1 / 2-1,-Z+2$
(3) $X-1, Y, Z$

Table 11. Intramolecular contacts less than $3.60 \AA$

| atom | atom | distance | atom | atom | distance <br> F8 |
| :--- | :--- | :--- | :--- | :--- | :--- |
| F9 | $2.673(14)$ | F8 | F12 | $2.739(13)$ |  |
| F8 | C1 | $2.986(17)$ | F8 | C2 | $2.912(19)$ |
| F8 | C10 | $2.92(2)$ | F8 | C11 | $3.501(19)$ |
| F8 | C12 | $2.969(19)$ | F11 | F12 | $2.710(14)$ |
| F12 | C1 | $2.925(19)$ | F12 | C6 | $3.52(2)$ |
| F12 | C8 | $2.962(17)$ | F12 | C9 | $3.495(18)$ |
| F12 | C10 | $2.932(19)$ | O15 | N17 | $2.690(18)$ |
| O15 | C13 | $3.530(18)$ | O16 | C4 | $3.49(2)$ |
| O16 | C5 | $3.42(2)$ | O16 | C13 | $2.94(2)$ |
| N17 | C3 | $3.28(2)$ | C2 | C5 | $3.11(2)$ |
| C1 | C4 | $2.82(2)$ | C3 | C6 | $2.77(2)$ |
| C2 | C8 | $2.92(2)$ | C4 | C15 | $3.79(2)$ |
| C3 | C14 | $3.38(2)$ | C5 | C15 | $3.38(3)$ |
| C5 | C14 | $3.46(3)$ | C7 | C10 | $2.94(3)$ |
| C6 | C12 | $3.02(3)$ | C9 | C12 | $2.90(2)$ |
| C8 | C11 | $2.93(2)$ |  |  |  |

Table 12. Intramolecular contacts less than $3.60 \AA$ involving hydrogens

| atom | atom | distance | atom | atom | distance |
| :--- | :--- | :--- | :--- | :--- | :--- |
| F8 | H2 | 2.347 | H7 | 3.226 |  |
| F8 | H9 | 3.186 | F8 | H10B | 2.654 |
| F9 | H8 | 2.605 | F9 | H10A | 2.636 |
| F9 | H10B | 2.598 | F11 | H10A | 2.589 |
| F11 | H10B | 2.541 | F11 | H12 | 2.635 |
| F12 | H6 | 3.563 | F12 | H7 | 3.203 |
| F12 | H10B | 2.668 | F12 | H11 | 3.217 |
| O15 | H16 | $2.5(2)$ | O15 | H17A | $2.6(2)$ |
| O15 | H17C | $2.5(2)$ | O15 | H14 | 2.987 |
| O16 | H5 | 2.894 | O16 | H13A | 2.714 |
| O16 | H14 | 2.590 | H16 | C14 | $3.3(2)$ |
| H16 | H5 | 3.368 | H16 | H14 | 3.471 |
| N17 | H3 | 3.019 | N17 | H13A | 3.358 |
| N17 | H13B | 2.658 | H17A | C13 | $3.28(11)$ |
| H17A | C15 | $2.58(18)$ | H17A | H13B | 3.460 |
| H17A | H14 | 2.163 | H17 | H17B | C3 |


| H17C | H13B | 3.121 | H17C | H14 | 2.861 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| C1 | H3 | 3.302 | C1 | H5 | 3.261 |
| C1 | H8 | 2.743 | C1 | H12 | 2.692 |
| C2 | H6 | 3.256 | C2 | H7 | 3.204 |
| C2 | H8 | 2.846 | C3 | H5 | 3.248 |
| C3 | H13A | 3.259 | C3 | H13B | 2.596 |
| C4 | H2 | 3.286 | C4 | H6 | 3.281 |
| C4 | H14 | 3.415 | C5 | H3 | 3.254 |
| C5 | H13A | 2.651 | C5 | H13B | 3.283 |
| C6 | H2 | 3.254 | C6 | H7 | 2.683 |
| C6 | H12 | 2.701 | C7 | H2 | 2.771 |
| C7 | H6 | 2.615 | C7 | H9 | 2.717 |
| C7 | H10B | 3.298 | C7 | H11 | 2.726 |
| C8 | H2 | 2.558 | C8 | H10A | 3.384 |
| C8 | H10B | 2.773 | C8 | H11 | 3.280 |

Table 12. Intramolecular contacts less than $3.60 \AA$ involving hydrogens (continued)

| atom | atom | distance | atom | atom | distance |
| :---: | :---: | :---: | :---: | :---: | :---: |
| C8 | H12 | 3.359 | C9 | H7 | 2.671 |
| C9 | H11 | 2.683 | C10 | H7 | 3.235 |
| C10 | H8 | 3.386 | C10 | H12 | 3.365 |
| C11 | H7 | 2.661 | C11 | H9 | 2.666 |
| C12 | H6 | 2.831 | C12 | H8 | 3.354 |
| C12 | H9 | 3.225 | C12 | H10A | 3.366 |
| C12 | H10B | 2.760 | C13 | H3 | 2.649 |
| C13 | H5 | 2.712 | C14 | H3 | 3.399 |
| C14 | H5 | 3.507 | C15 | H5 | 3.211 |
| C15 | H13A | 2.825 | C15 | H13B | 3.410 |
| H2 | H3 | 2.380 | H2 | H7 | 3.430 |
| H2 | H8 | 2.375 | H3 | H13A | 3.511 |
| H3 | H13B | 2.417 | H5 | H6 | 2.331 |
| H5 | H13A | 2.489 | H5 | H13B | 3.553 |
| H6 | H7 | 2.585 | H6 | H12 | 2.244 |
| H7 | H8 | 2.336 | H7 | H9 | 2.505 |
| H7 | H11 | 2.516 | H7 | H12 | 2.285 |
| H8 | H9 | 2.353 | H9 | H10A | 2.337 |
| H9 | H10B | 2.863 | H9 | H11 | 2.505 |
| H10A | H11 | 2.335 | H10B | H11 | 2.866 |
| H11 | H12 | 2.337 | H13A | H14 | 2.338 |
| H13B | H14 | 2.396 |  |  |  |

Table 13. Intermolecular contacts less than $3.60 \AA$

| atom | atom | distance | atom | atom | distance |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Cl 1 | O15 ${ }^{1}$ | 3.586(13) | Cl 1 | O15 ${ }^{2}$ | 3.539(14) |
| Cl 1 | O16 ${ }^{1}$ | 3.004(13) | Cl 1 | N17 | 3.351(14) |
| Cl 1 | N17 ${ }^{3}$ | 3.153(14) | Cl 1 | N17 ${ }^{2}$ | 3.240(15) |
| Cl 1 | C14 ${ }^{2}$ | 3.467(16) | Cl1 | C15 ${ }^{2}$ | 3.60(2) |
| F8 | F9 ${ }^{4}$ | 3.278(13) | F8 | F115 | 3.554(13) |
| F8 | C51 | 3.31(2) | F8 | C61 | 3.159(18) |
| F8 | C94 | 3.250(18) | F9 | F8 ${ }^{3}$ | 3.278(13) |
| F9 | F11 ${ }^{6}$ | 3.563(16) | F9 | C5 ${ }^{7}$ | 3.24(2) |
| F11 | F88 | 3.554(13) | F11 | F99 | 3.563(16) |
| F11 | C10 ${ }^{8}$ | 3.40(2) | F11 | C10 ${ }^{9}$ | 3.45(2) |
| F12 | C94 | 3.300 (19) | F12 | C104 | 3.248(18) |
| F12 | C114 | 3.126(17) | 015 | $\mathrm{Cl} 1{ }^{10}$ | 3.586(13) |
| 015 | Cl 111 | 3.539(14) | 015 | N17 ${ }^{11}$ | 3.053(18) |
| 015 | C13 ${ }^{3}$ | 3.44(2) | 015 | C14 ${ }^{3}$ | 3.24(2) |
| 016 | $\mathrm{Cl} 1{ }^{10}$ | 3.004(13) | 016 | C2 ${ }^{10}$ | 3.49(2) |
| 016 | C3 ${ }^{10}$ | 3.35(2) | N17 | $\mathrm{Cl} 1{ }^{4}$ | 3.153(14) |
| N17 | Cl 1 | 3.351(14) | N17 | Cl 111 | 3.240(15) |
| N17 | O15 ${ }^{2}$ | 3.053(18) | C2 | O161 | 3.49(2) |
| C3 | O16 ${ }^{1}$ | 3.35(2) | C5 | F8 ${ }^{10}$ | 3.31(2) |
| C5 | F9 ${ }^{12}$ | 3.24(2) | C6 | F8 ${ }^{10}$ | 3.159(18) |
| C9 | F8 ${ }^{3}$ | 3.250(18) | C9 | F12 ${ }^{3}$ | 3.300(19) |
| C10 | F115 | 3.40(2) | C10 | F11 ${ }^{6}$ | 3.45(2) |


| C 10 | $\mathrm{~F} 12^{3}$ | $3.248(18)$ | C 11 | $\mathrm{~F} 12^{3}$ | $3.126(17)$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| C 13 | $\mathrm{O} 15^{4}$ | $3.44(2)$ | C 14 | $\mathrm{Cl1}^{11}$ | $3.467(16)$ |
| C 14 | $\mathrm{O} 15^{4}$ | $3.24(2)$ | C 15 | $\mathrm{Cl1} 11$ | $3.60(2)$ |

Symmetry Operators:
(1) $X, Y+1, Z$
(2) $-X+1, Y+1 / 2,-Z+2$
(3) $X+1, Y, Z$
(4) $X-1, Y, Z$
(5) $-X+1, Y+1 / 2,-Z+1$
(6) $-X+2, Y+1 / 2,-Z+1$
(7) $X+1, Y+1, Z$
(8) $-X+1, Y+1 / 2-1,-Z+1$
(9) $-X+2, Y+1 / 2-1,-Z+1$
(11) $-X+1, Y+1 / 2-1,-Z+2$
(10) $X, Y-1, Z$
(12) $\mathrm{X}-1, \mathrm{Y}-1, \mathrm{Z}$

Table 14. Intermolecular contacts less than $3.60 \AA$ involving hydrogens

| atom | atom | distance | atom | atom | distance |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Cl1 | H16 ${ }^{1}$ | 2.1(2) | Cl 1 | H17A ${ }^{2}$ | 2.31(15) |
| Cl1 | H17B | 3.55(19) | Cl 1 | H17B ${ }^{3}$ | 2.25(18) |
| Cl1 | H17C | 2.41(18) | Cl 1 | H3 ${ }^{3}$ | 3.264 |
| Cl1 | H13B ${ }^{3}$ | 2.948 | Cl 1 | H142 | 3.060 |
| F8 | H5 ${ }^{1}$ | 2.771 | F8 | H61 | 2.460 |
| F8 | H94 | 2.441 | F9 | H5 ${ }^{5}$ | 2.966 |
| F9 | H61 | 2.708 | F9 | H65 | 3.591 |
| F9 | H7 ${ }^{1}$ | 3.440 | F9 | H12 ${ }^{1}$ | 3.093 |
| F11 | H10A ${ }^{6}$ | 2.544 | F11 | H10B ${ }^{7}$ | 2.413 |
| F11 | H12 ${ }^{8}$ | 3.250 | F12 | H94 | 2.530 |
| F12 | H10A ${ }^{4}$ | 2.918 | F12 | H10A ${ }^{7}$ | 3.351 |
| F12 | H10B ${ }^{7}$ | 3.147 | F12 | H114 | 2.296 |
| 015 | H17A ${ }^{9}$ | 2.61(18) | 015 | H17B ${ }^{9}$ | 3.11(19) |
| 015 | H17C ${ }^{9}$ | 3.1(2) | 015 | H13A ${ }^{3}$ | 3.369 |
| 015 | H13B ${ }^{3}$ | 3.098 | 015 | H14 ${ }^{3}$ | 2.403 |
| 016 | H17A ${ }^{9}$ | 3.27(15) | 016 | H17B ${ }^{10}$ | 3.46(17) |
| O16 | H17C ${ }^{10}$ | 3.5(2) | 016 | $\mathrm{H} 2{ }^{10}$ | 3.002 |
| 016 | H3 ${ }^{10}$ | 2.752 | H16 | $\mathrm{Cl} 1{ }^{10}$ | 2.1(2) |
| H16 | N17 ${ }^{10}$ | 3.6(2) | H16 | H17A ${ }^{9}$ | 2.8(3) |
| H16 | H17B ${ }^{10}$ | 3.2(3) | H16 | H17C ${ }^{10}$ | 2.9(3) |
| H16 | C2 ${ }^{10}$ | 3.1(2) | H16 | C3 ${ }^{10}$ | 3.0(2) |
| H16 | $\mathrm{H} 2{ }^{10}$ | 2.767 | H16 | $\mathrm{H} 3{ }^{10}$ | 2.626 |
|  |  |  |  |  |  |


| N17 | H16 ${ }^{1}$ | 3.6(2) | N17 | H1411 | 3.560 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| H17A | $\mathrm{Cl} 1{ }^{9}$ | 2.31(15) | H17A | O15 ${ }^{2}$ | 2.61(18) |
| H17A | O16 ${ }^{2}$ | 3.27(15) | H17A | H16 ${ }^{2}$ | 2.8(3) |
| H17A | C15 ${ }^{2}$ | 3.13(18) | H17A | H14 ${ }^{11}$ | 3.238 |
| H17B | $\mathrm{Cl} 1{ }^{4}$ | 2.25(18) | H17B | Cl 1 | 3.55(19) |
| H17B | O15 ${ }^{2}$ | 3.11(19) | H17B | O16 ${ }^{1}$ | 3.46(17) |
| H17B | H16 ${ }^{1}$ | 3.2(3) | H17B | H14 ${ }^{11}$ | 3.182 |
| H17C | Cl 1 | 2.41(18) | H17C | O15 ${ }^{2}$ | 3.1(2) |
| H17C | O161 | 3.5(2) | H17C | H16 ${ }^{1}$ | 2.9(3) |
| H17C | C15 ${ }^{2}$ | 3.6(2) | H17C | H13B ${ }^{3}$ | 3.539 |
| C1 | H7 ${ }^{4}$ | 3.535 | C1 | H94 | 3.360 |
| C1 | H13B ${ }^{3}$ | 3.316 | C2 | H16 ${ }^{1}$ | 3.1(2) |
| C2 | H5 ${ }^{1}$ | 3.447 | C2 | H7 ${ }^{4}$ | 3.183 |
| C2 | H84 | 3.425 | C2 | H94 | 3.015 |
| C2 | H13B ${ }^{3}$ | 3.584 | C3 | H16 ${ }^{1}$ | 3.0(2) |

Table 14. Intermolecular contacts less than $3.60 \AA$ involving hydrogens (continued)

| atom | atom | distance | atom | atom | distance |
| :---: | :---: | :---: | :---: | :---: | :---: |
| C3 | H7 ${ }^{4}$ | 2.907 | C3 | H84 | 3.226 |
| C4 | H7 ${ }^{4}$ | 3.130 | C5 | $\mathrm{H} 2{ }^{10}$ | 3.462 |
| C5 | H7 ${ }^{4}$ | 3.491 | C5 | H13A ${ }^{3}$ | 3.529 |
| C5 | H13B ${ }^{3}$ | 3.375 | C6 | H13A ${ }^{3}$ | 3.376 |
| C6 | H13B ${ }^{3}$ | 3.203 | C8 | H61 | 3.276 |
| C8 | H94 | 3.511 | C9 | H5 ${ }^{5}$ | 3.587 |
| C9 | H61 | 3.544 | C10 | $\mathrm{H} 11^{12}$ | 3.457 |
| C11 | H10A ${ }^{6}$ | 3.055 | C11 | H10B ${ }^{7}$ | 3.474 |
| C12 | H10B ${ }^{7}$ | 3.475 | C12 | H114 | 3.524 |
| C15 | H17A ${ }^{9}$ | 3.13(18) | C15 | H17C ${ }^{9}$ | 3.6(2) |
| C15 | H143 | 3.448 | H2 | O161 | 3.002 |
| H2 | H16 ${ }^{1}$ | 2.767 | H2 | C5 ${ }^{1}$ | 3.462 |
| H2 | H5 ${ }^{1}$ | 2.551 | H2 | H84 | 3.298 |
| H2 | H94 | 2.851 | H3 | $\mathrm{Cl} 1{ }^{4}$ | 3.264 |
| H3 | O16 ${ }^{1}$ | 2.752 | H3 | H16 ${ }^{1}$ | 2.626 |
| H3 | H74 | 3.160 | H3 | H84 | 2.876 |
| H5 | F810 | 2.771 | H5 | F913 | 2.966 |
| H5 | $\mathrm{C} 2{ }^{10}$ | 3.447 | H5 | C913 | 3.587 |
| H5 | $\mathrm{H} 2{ }^{10}$ | 2.551 | H5 | H8 ${ }^{13}$ | 3.326 |
| H5 | H8 ${ }^{10}$ | 3.499 | H5 | $\mathrm{H} 9^{13}$ | 3.307 |
| H5 | H13A ${ }^{3}$ | 3.557 | H6 | F8 ${ }^{10}$ | 2.460 |
| H6 | F913 | 3.591 | H6 | F910 | 2.708 |
|  |  |  |  |  |  |


| H6 | C8 ${ }^{10}$ | 3.276 | H6 | C910 | 3.544 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| H6 | H8 ${ }^{10}$ | 3.277 | H6 | H13A ${ }^{3}$ | 3.328 |
| H6 | H13B ${ }^{3}$ | 3.541 | H7 | F910 | 3.440 |
| H7 | $\mathrm{C} 1^{3}$ | 3.535 | H7 | $\mathrm{C} 2{ }^{3}$ | 3.183 |
| H7 | C3 ${ }^{3}$ | 2.907 | H7 | $C 4^{3}$ | 3.130 |
| H7 | C5 ${ }^{3}$ | 3.491 | H7 | H3 ${ }^{3}$ | 3.160 |
| H8 | C2 ${ }^{3}$ | 3.425 | H8 | C3 ${ }^{3}$ | 3.226 |
| H8 | $\mathrm{H} 2^{3}$ | 3.298 | H8 | H3 ${ }^{3}$ | 2.876 |
| H8 | H5 ${ }^{1}$ | 3.499 | H8 | H5 ${ }^{5}$ | 3.326 |
| H8 | H61 | 3.277 | H8 | H13A ${ }^{5}$ | 3.064 |
| H9 | $F 8^{3}$ | 2.441 | H9 | F12 ${ }^{3}$ | 2.530 |
| H9 | C1 ${ }^{3}$ | 3.360 | H9 | $\mathrm{C} 2{ }^{3}$ | 3.015 |
| H9 | C83 | 3.511 | H9 | $\mathrm{H} 2{ }^{3}$ | 2.851 |
| H9 | H5 ${ }^{5}$ | 3.307 | H10A | F11 ${ }^{12}$ | 2.544 |
| H10A | F12 ${ }^{3}$ | 2.918 | H10A | F12 ${ }^{8}$ | 3.351 |

Table 14. Intermolecular contacts less than $3.60 \AA$ involving hydrogens (continued)

| atom | atom | distance | atom | atom | distance |
| :---: | :---: | :---: | :---: | :---: | :---: |
| H10A | C1112 | 3.055 | H10A | H11 ${ }^{12}$ | 2.582 |
| H10B | F11 ${ }^{8}$ | 2.413 | H10B | F12 ${ }^{8}$ | 3.147 |
| H10B | C11 ${ }^{8}$ | 3.474 | H10B | C12 ${ }^{8}$ | 3.475 |
| H10B | H11 ${ }^{12}$ | 3.508 | H10B | H12 ${ }^{8}$ | 3.219 |
| H11 | F12 ${ }^{3}$ | 2.296 | H11 | C10 ${ }^{6}$ | 3.457 |
| H11 | C12 ${ }^{3}$ | 3.524 | H11 | H10A ${ }^{6}$ | 2.582 |
| H11 | H10B ${ }^{6}$ | 3.508 | H12 | F910 | 3.093 |
| H12 | F11 ${ }^{7}$ | 3.250 | H12 | H10B ${ }^{7}$ | 3.219 |
| H13A | O15 ${ }^{4}$ | 3.369 | H13A | C54 | 3.529 |
| H13A | C64 | 3.376 | H13A | H54 | 3.557 |
| H13A | H64 | 3.328 | H13A | H8 ${ }^{13}$ | 3.064 |
| H13B | $\mathrm{Cl} 1{ }^{4}$ | 2.948 | H13B | O15 ${ }^{4}$ | 3.098 |
| H13B | H17C ${ }^{4}$ | 3.539 | H13B | C1 ${ }^{4}$ | 3.316 |
| H13B | $\mathrm{C} 2{ }^{4}$ | 3.584 | H13B | C54 | 3.375 |
| H13B | C64 | 3.203 | H13B | H64 | 3.541 |
| H14 | $\mathrm{Cl} 1{ }^{9}$ | 3.060 | H14 | O15 ${ }^{4}$ | 2.403 |
| H14 | N17 ${ }^{14}$ | 3.560 | H14 | H17A ${ }^{14}$ | 3.238 |
| H14 | H17B ${ }^{14}$ | 3.182 | H14 | C154 | 3.448 |

Symmetry Operators:
(1) $X, Y+1, Z$
(2) $-X+1, Y+1 / 2,-Z+2$
(3) $X+1, Y, Z$
(4) $X-1, Y, Z$
(5) $\mathrm{X}+1, \mathrm{Y}+1, \mathrm{Z}$
(6) $-X+2, Y+1 / 2-1,-Z+1$
(7) $-X+1, Y+1 / 2-1,-Z+1$
(8) $-X+1, Y+1 / 2,-Z+1$
(9) $-X+1, Y+1 / 2-1,-Z+2$
(10) $X, Y-1, Z$
(11) $-X, Y+1 / 2,-Z+2$
(12) $-X+2, Y+1 / 2,-Z+1$
(13) $X-1, Y-1, Z$
(14) $-X, Y+1 / 2-1,-Z+2$

