

Electronic Supporting Information (ESI)

CuBr catalyzed aerobic oxidative coupling of 2-aminopyridines with cinnamaldehydes: Direct access to 3-formyl-2-phenyl-imidazo[1,2-a]pyridines

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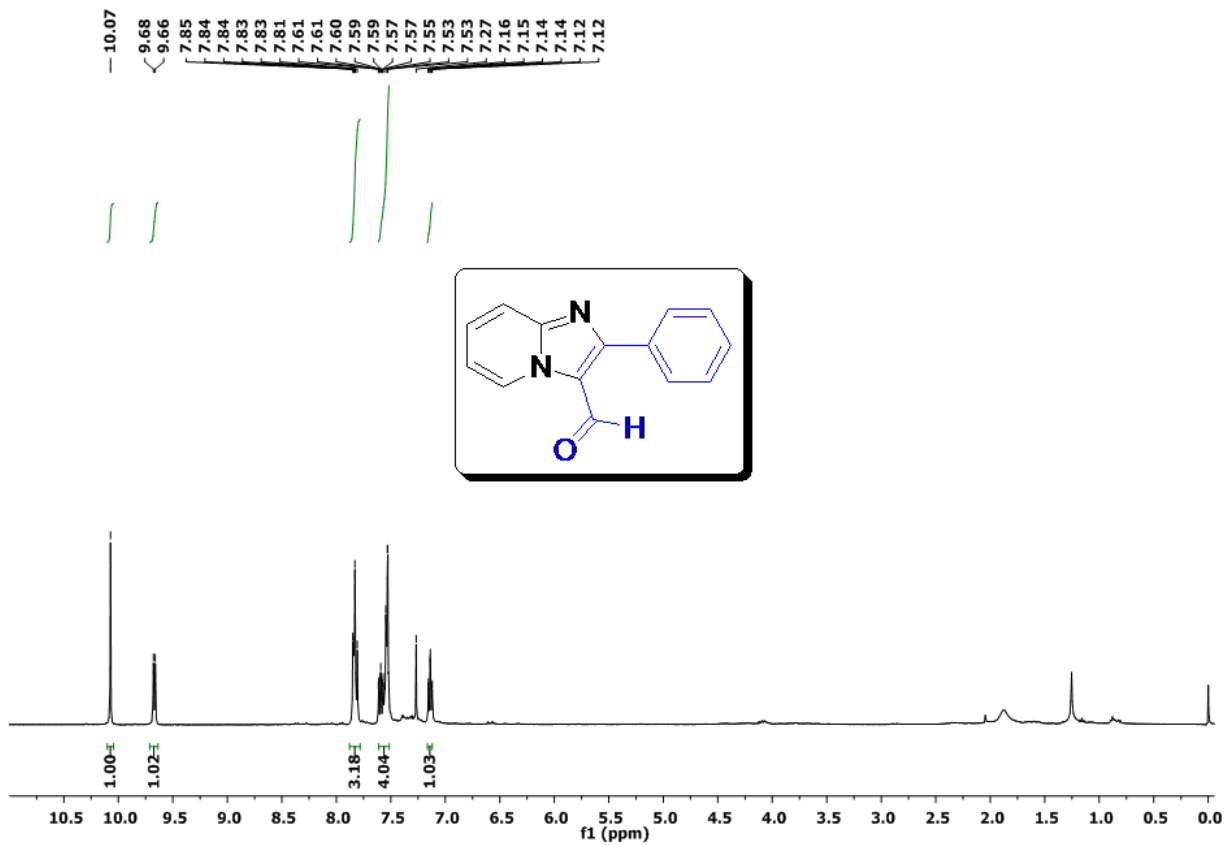
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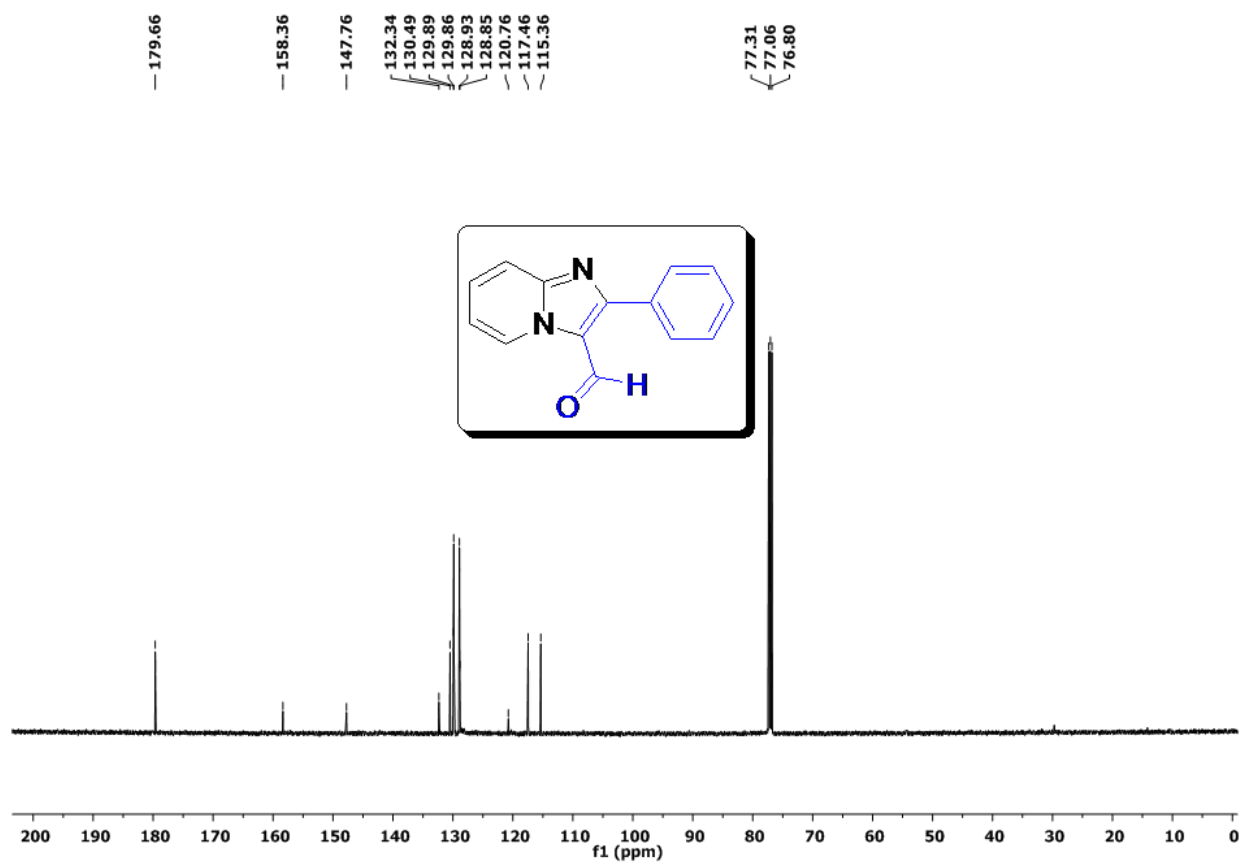
CONTENTS

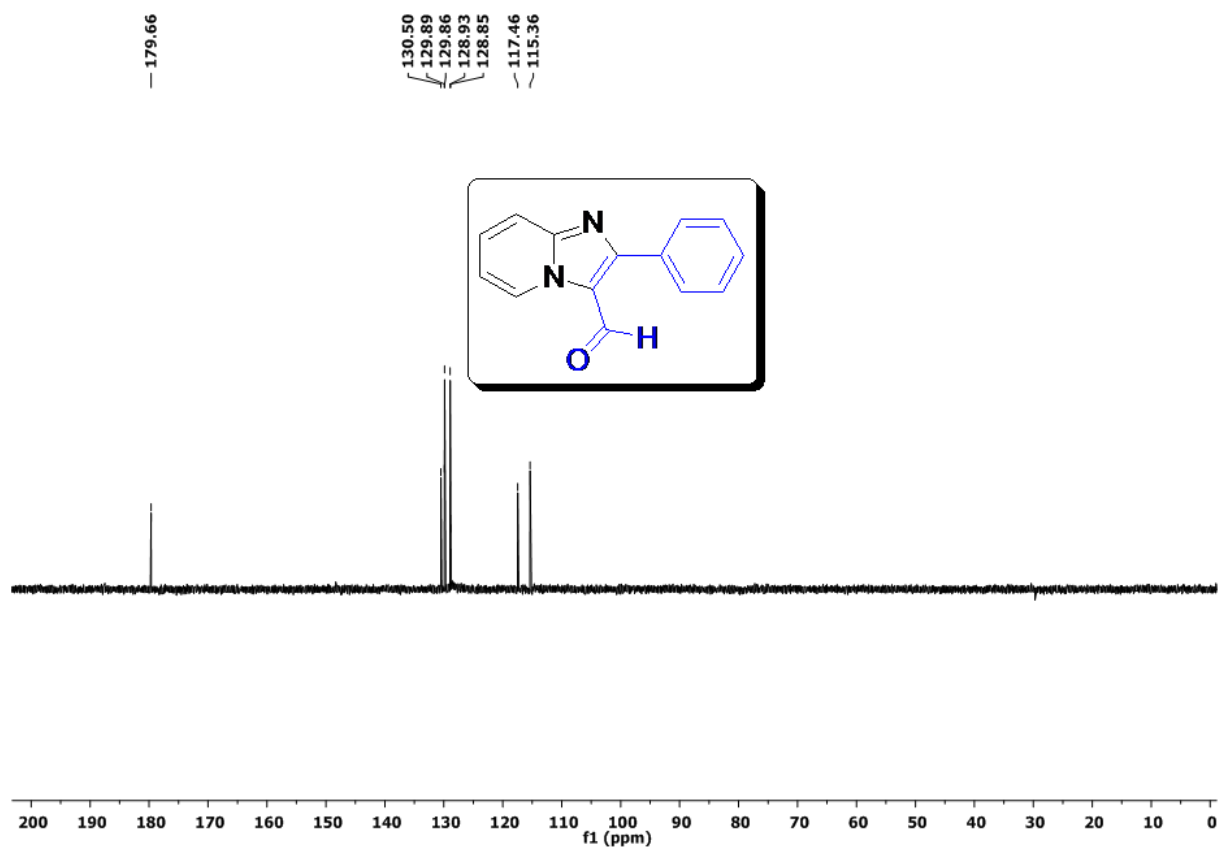
- S1.** NMR spectra scans
- S2.** Reaction monitoring for the color change
- S3.** Table of Gibbs free energy values calculated using B3LYP/6-311+G(d,p) method, for all the reactants, intermediates and products involved in the reaction mechanism

S1. NMR data scans

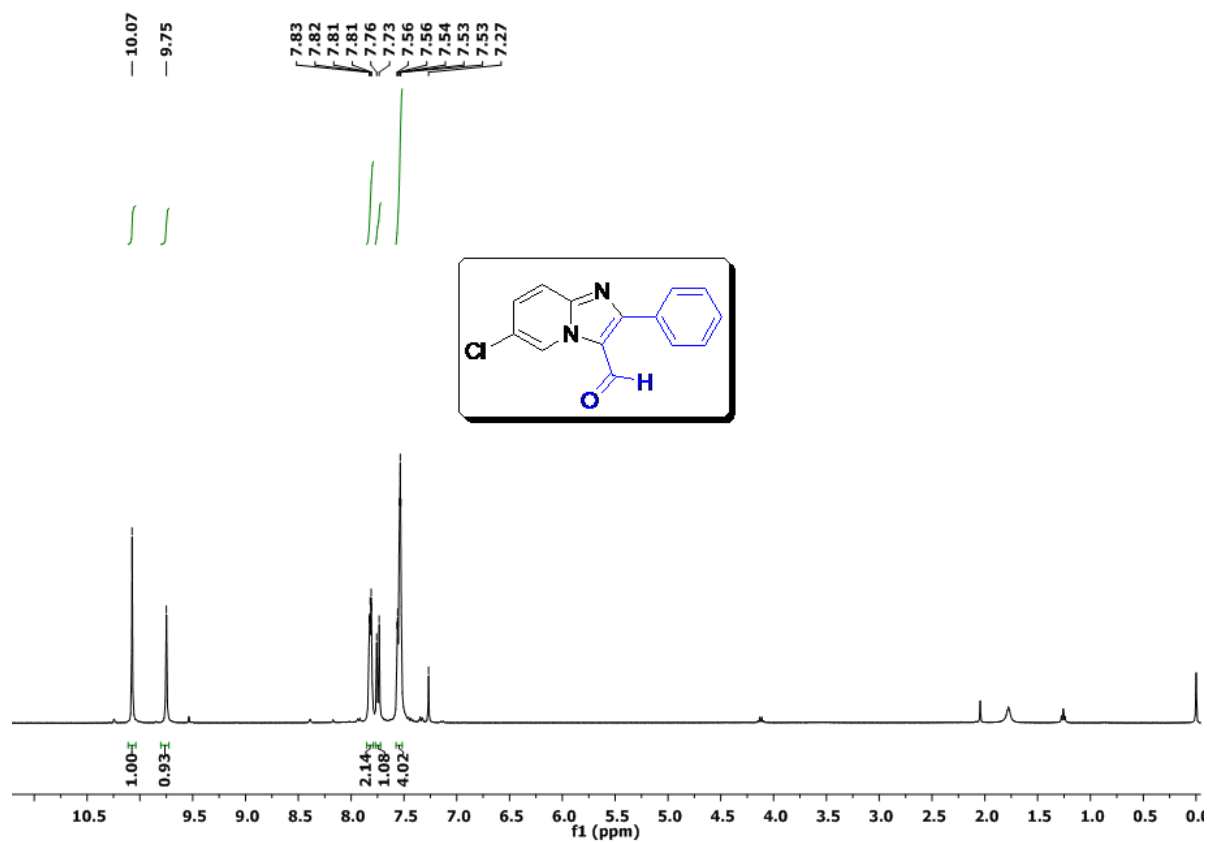
^1H , ^{13}C and DEPT-135 NMR of 2-phenyl-imidazo[1,2-a]pyridine-3-carbaldehyde (3a):

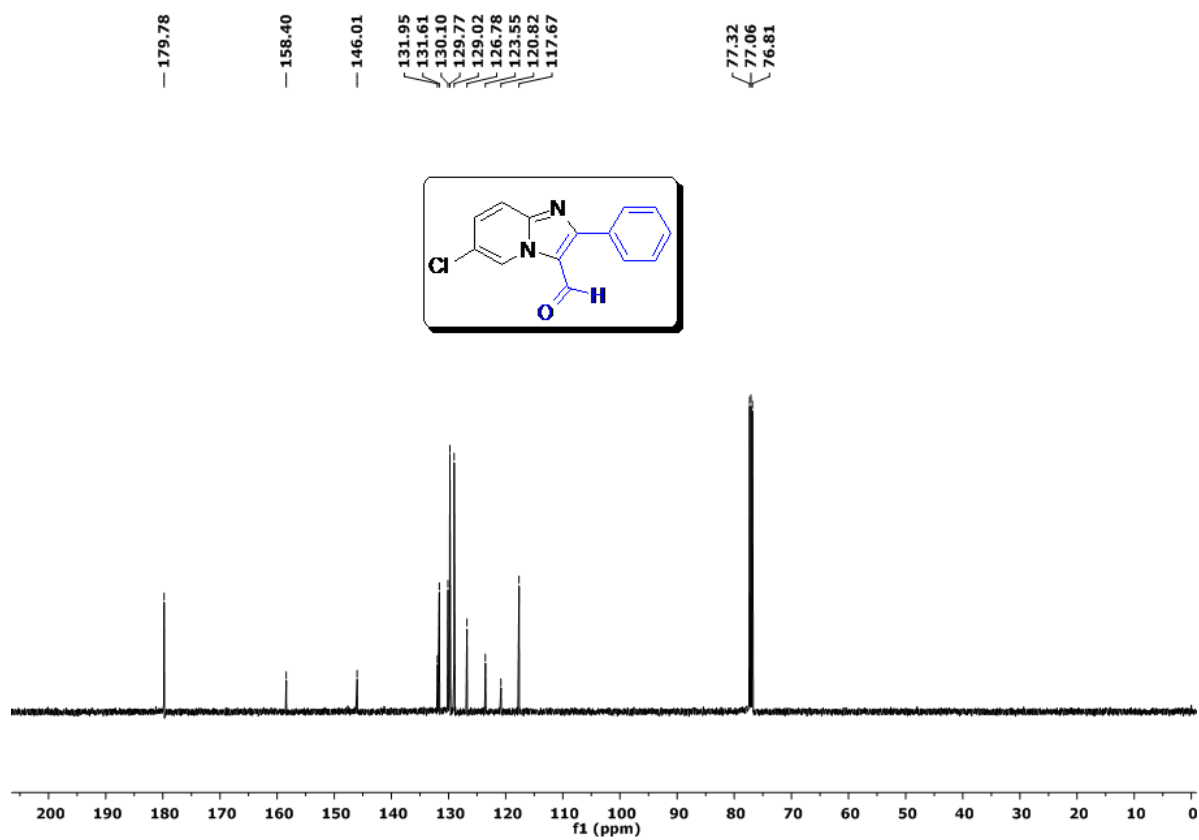


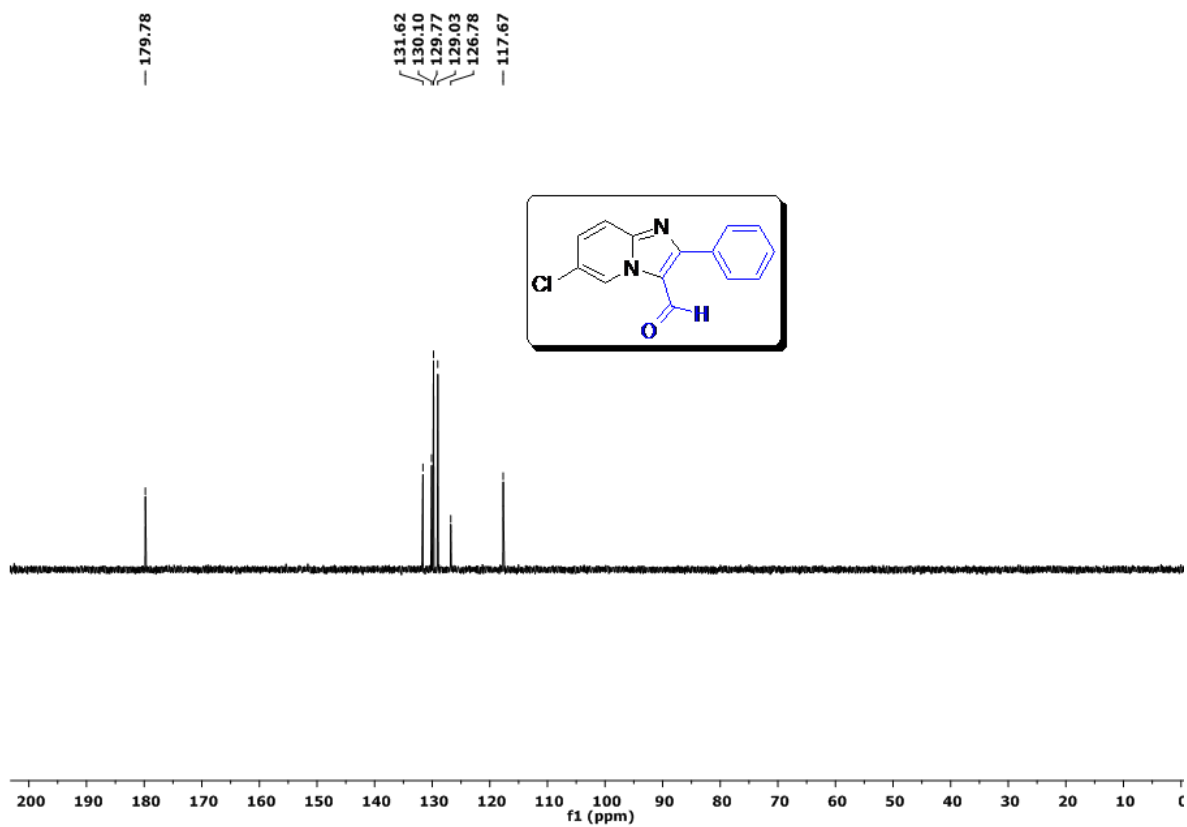




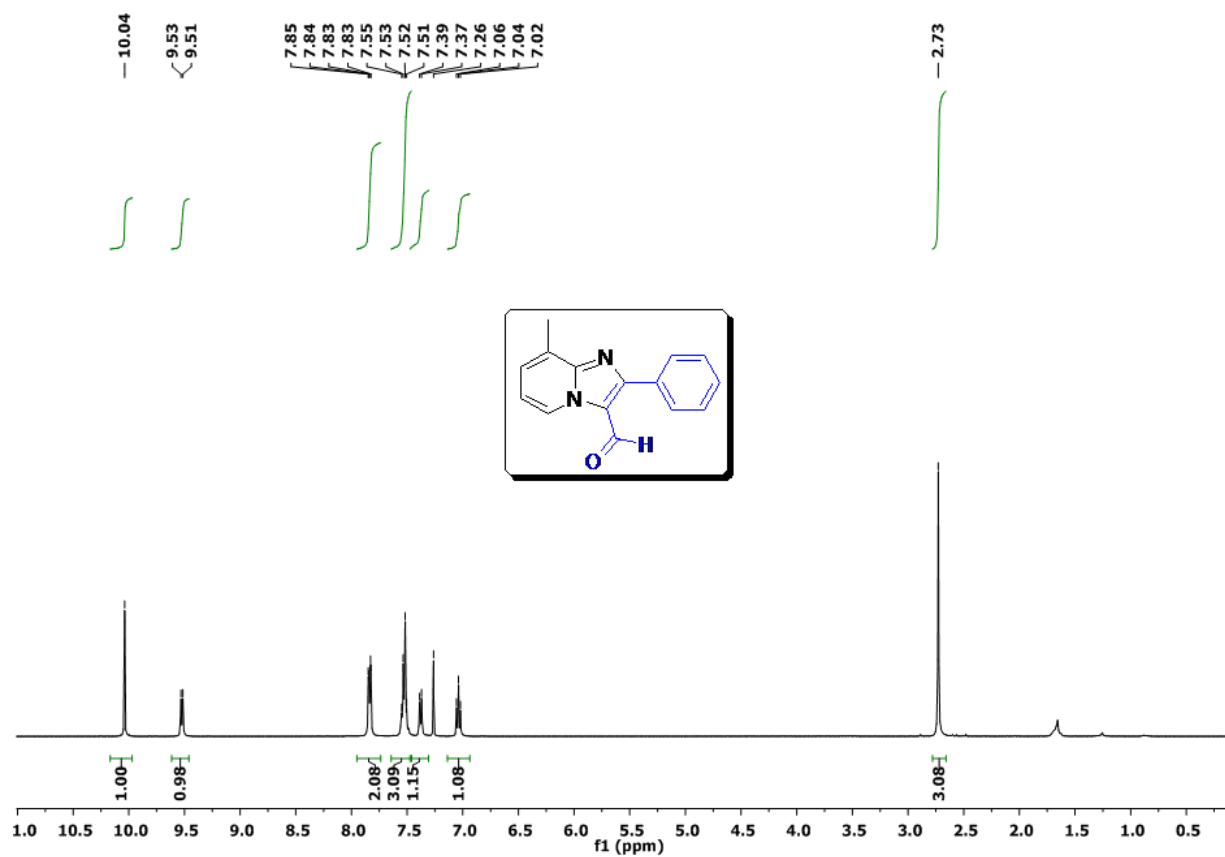
^1H , ^{13}C and DEPT-135 NMR of 7-chloro-2-phenyl-imidazo[1,2-a]pyridine-3-carbaldehyde (3b):

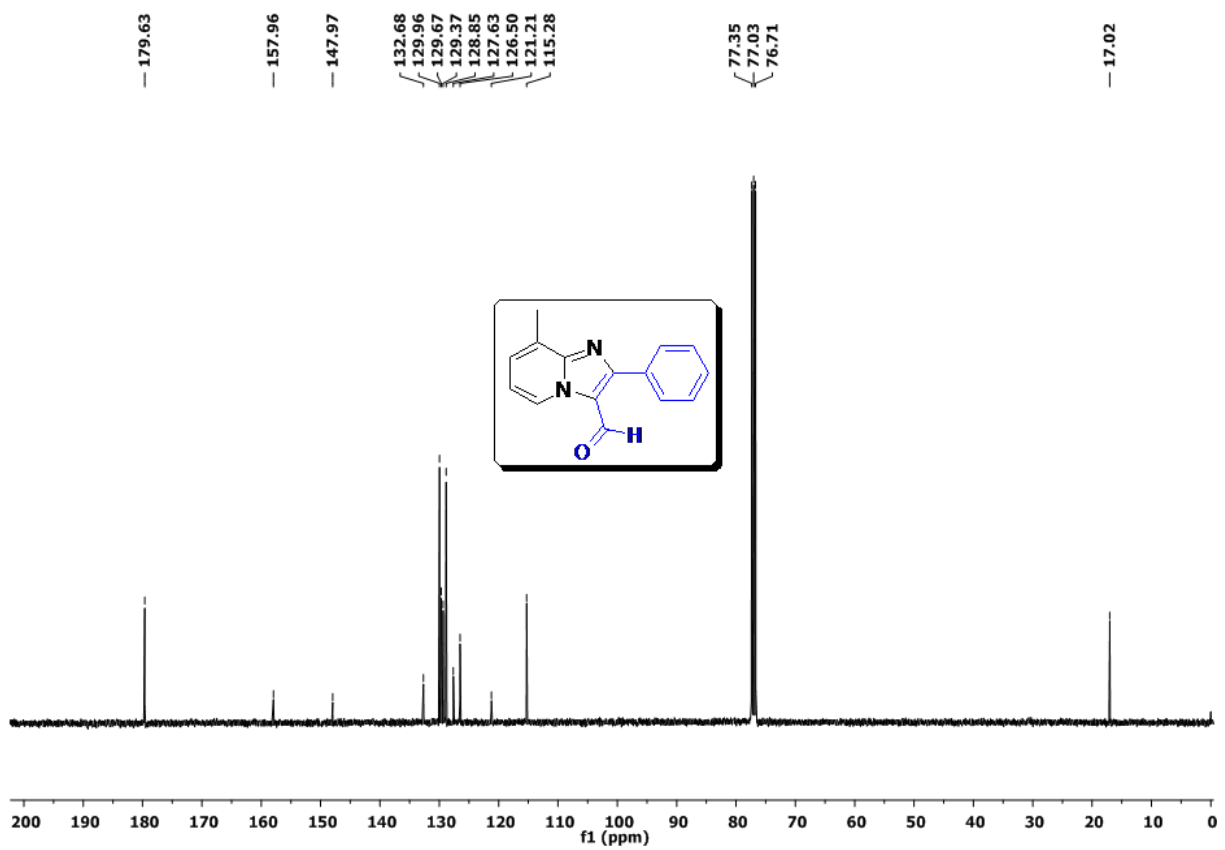


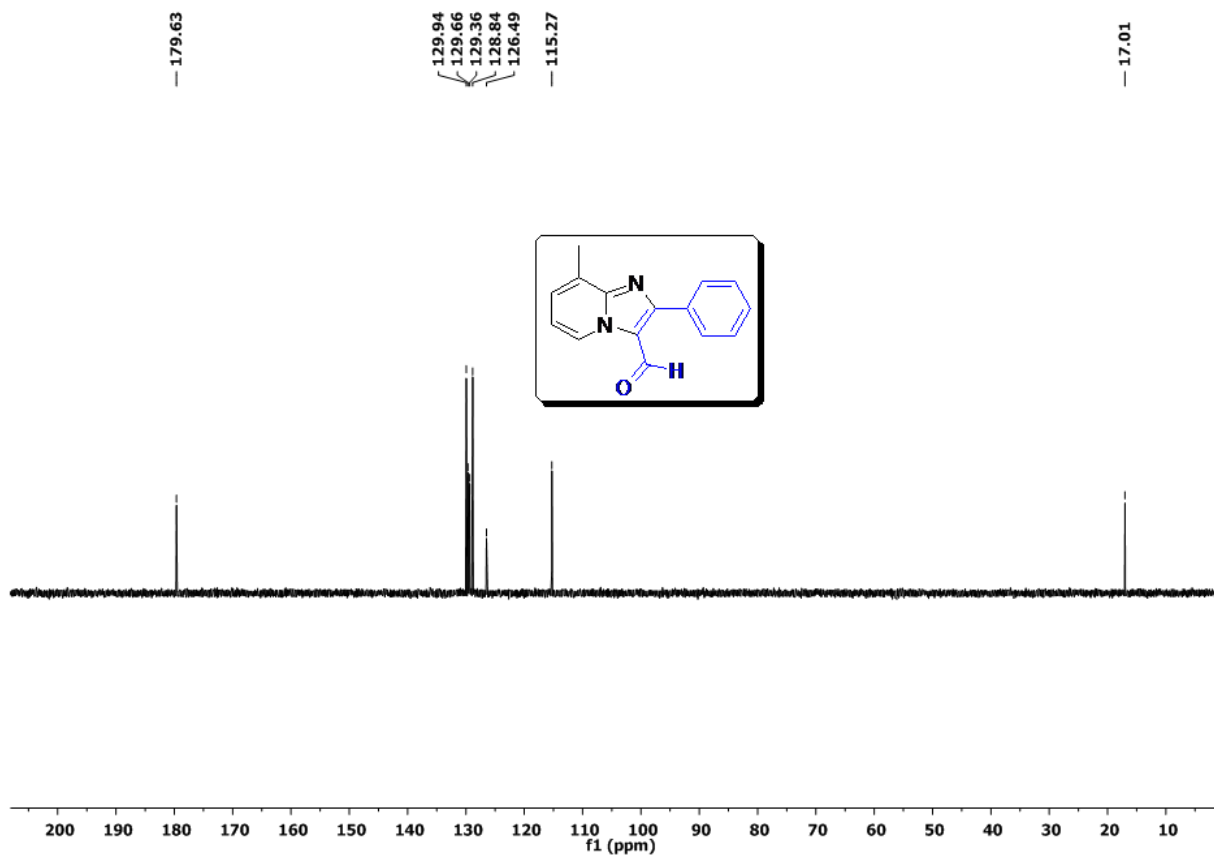




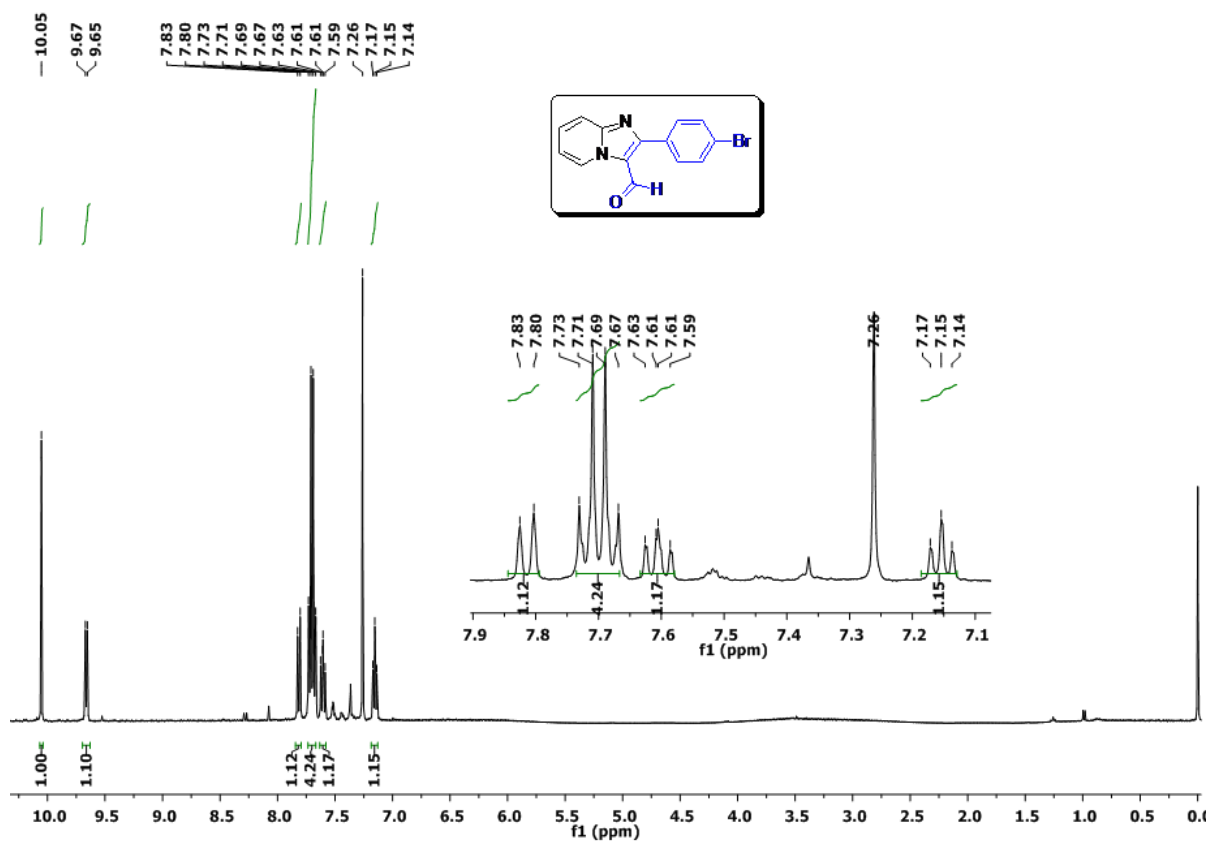
^1H , ^{13}C and DEPT-135 NMR of 8-methyl-2-phenyl-imidazo[1,2-a]pyridine-3-carbaldehyde (3c):

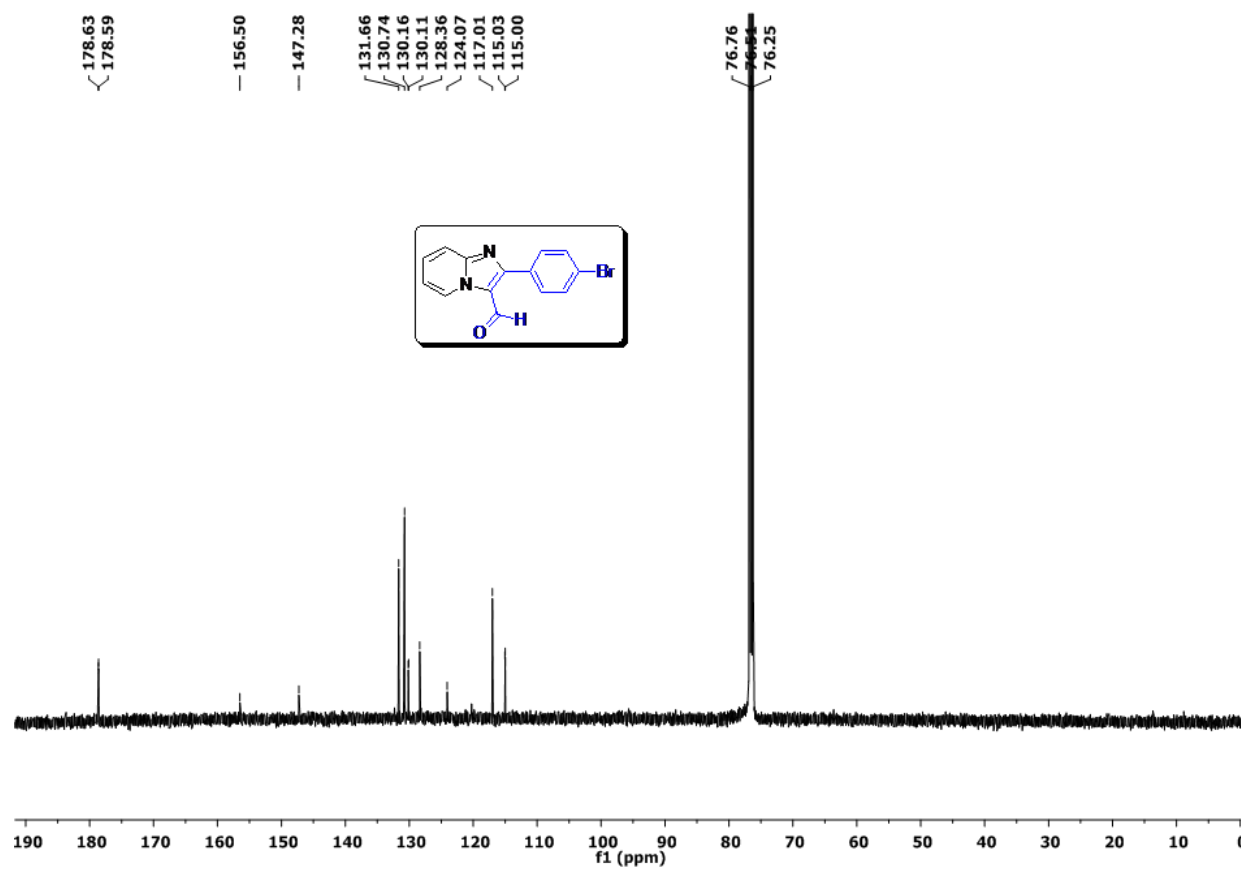


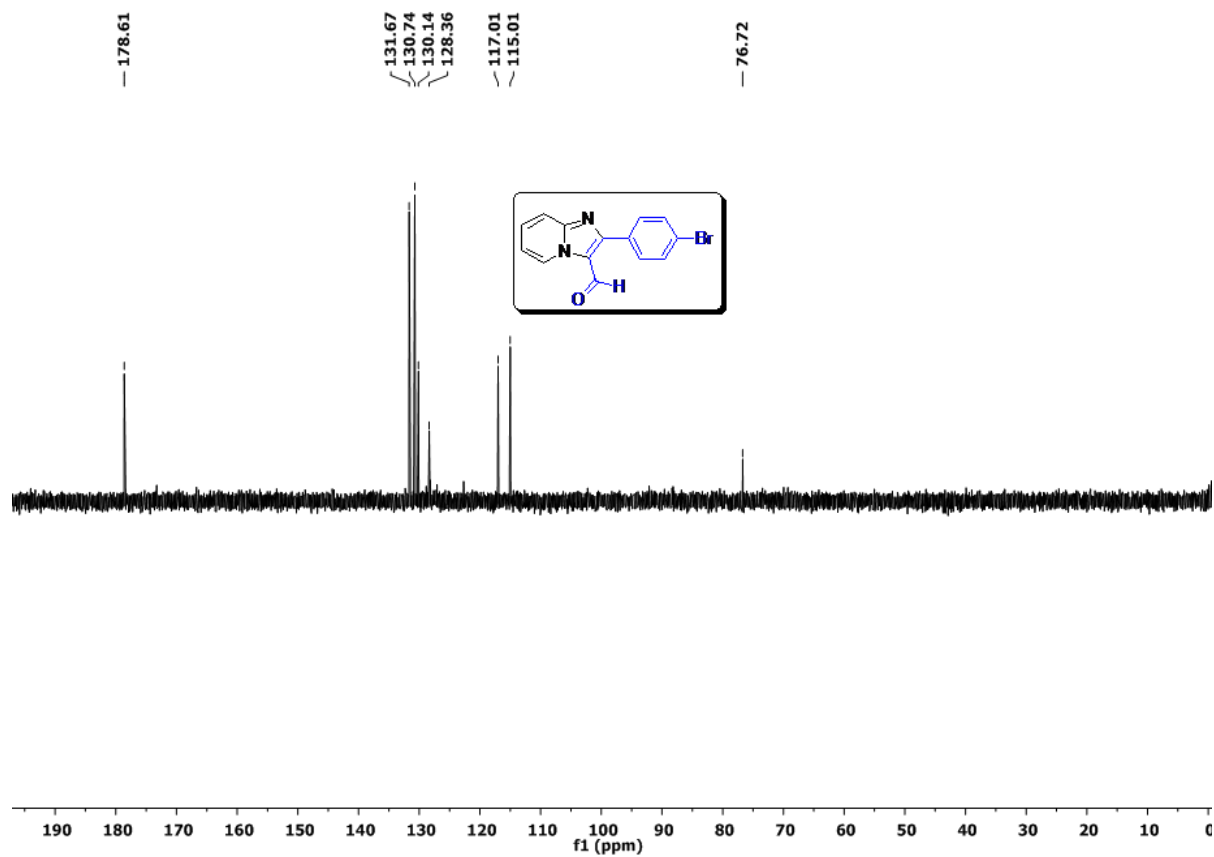




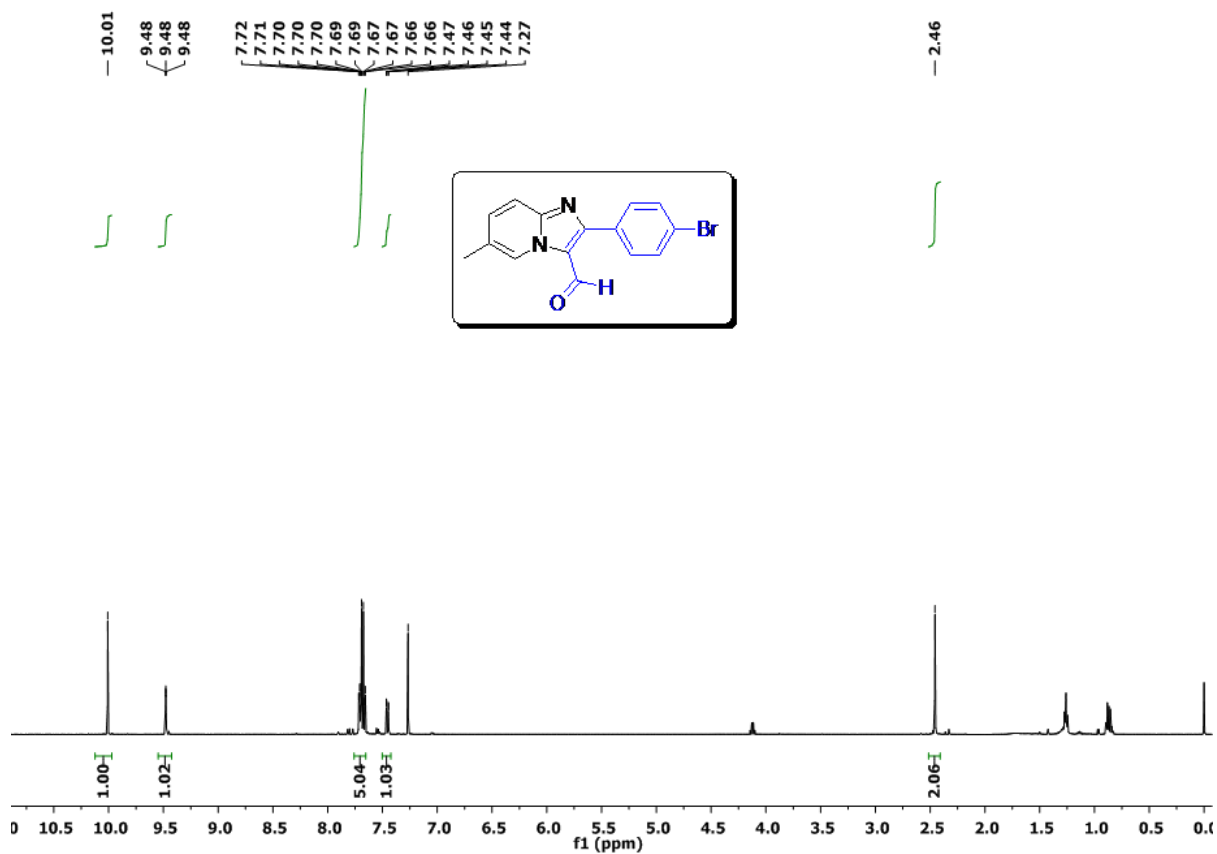
^1H , ^{13}C and DEPT-135 NMR of 2-(4-bromophenyl)imidazo[1,2-a]pyridine-3-carbaldehyde (3d):

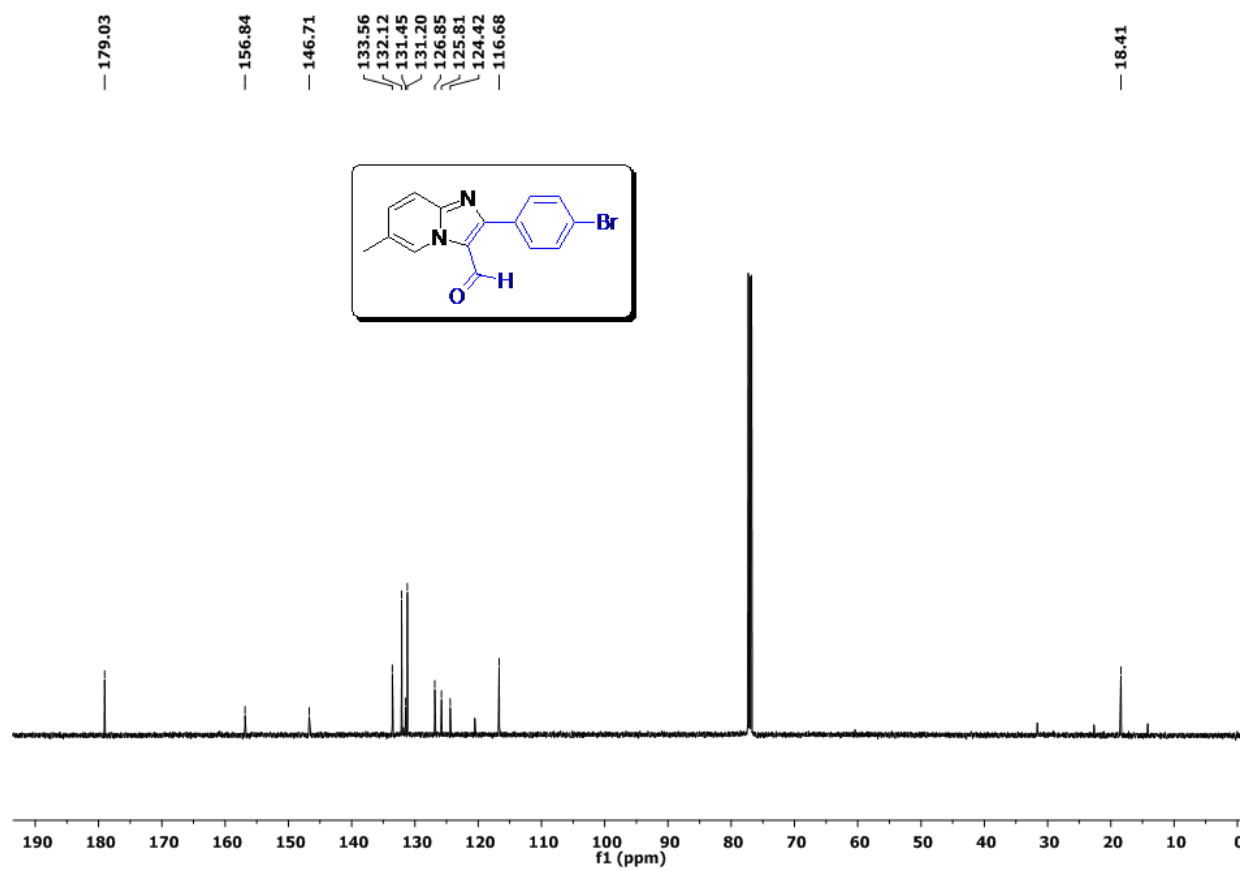


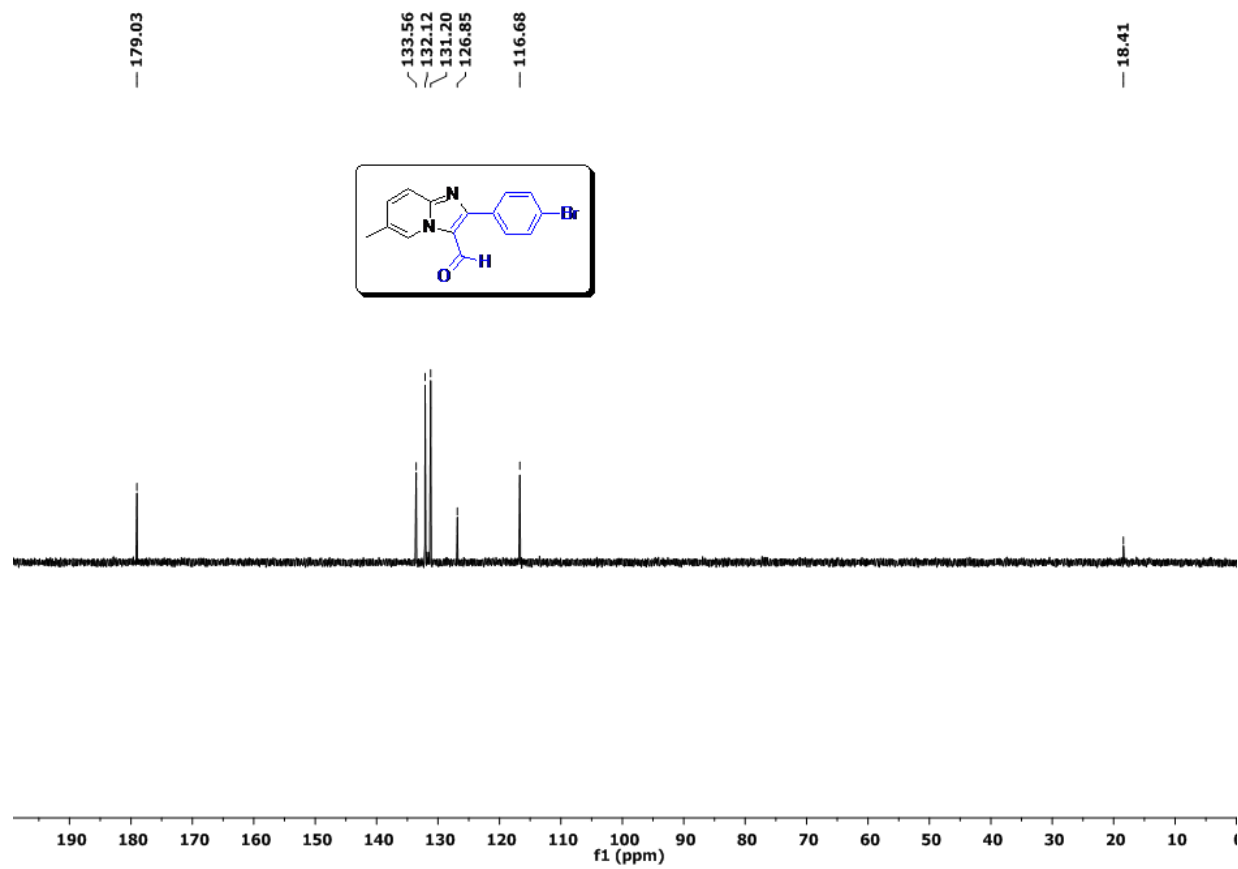




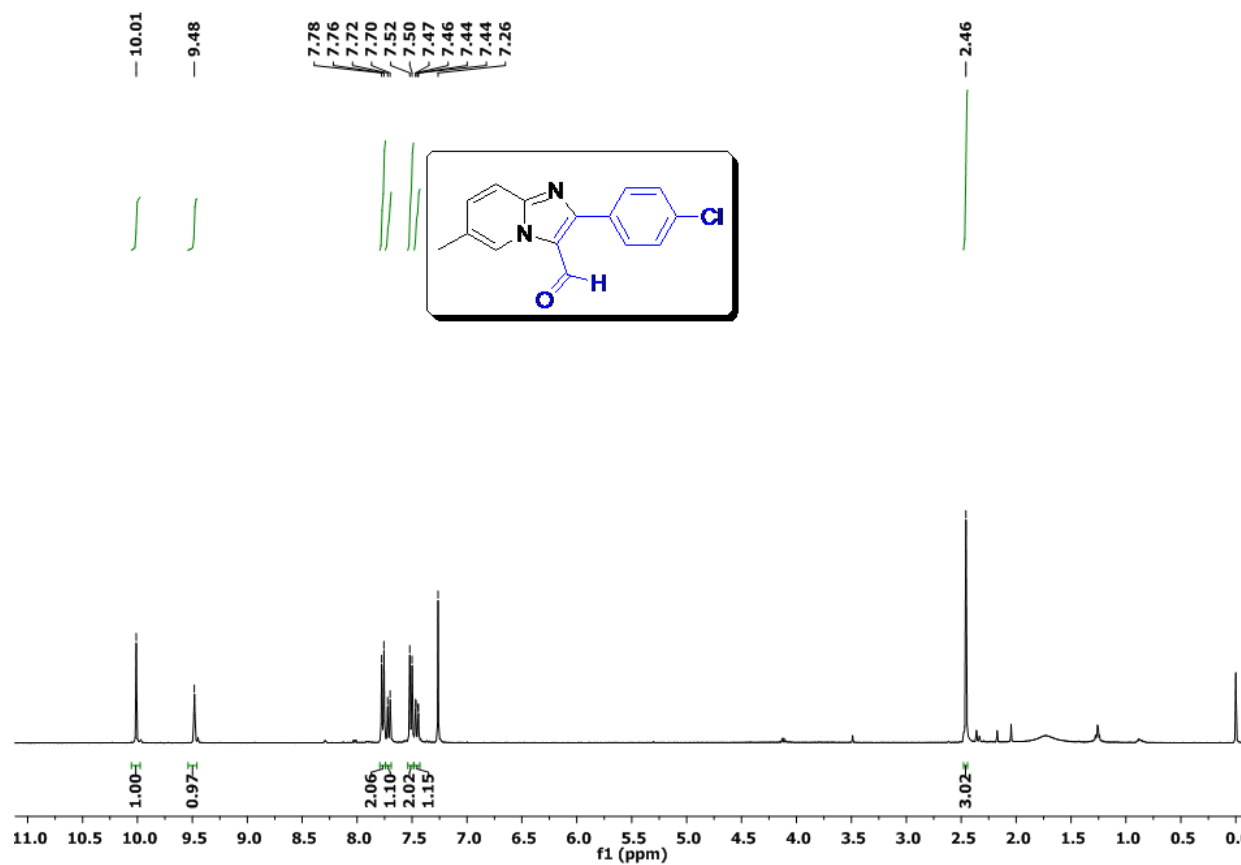
^1H , ^{13}C and DEPT-135 NMR of 2-(4-bromophenyl)-6-methylimidazo[1,2-a]pyridine-3-carbaldehyde (3e):

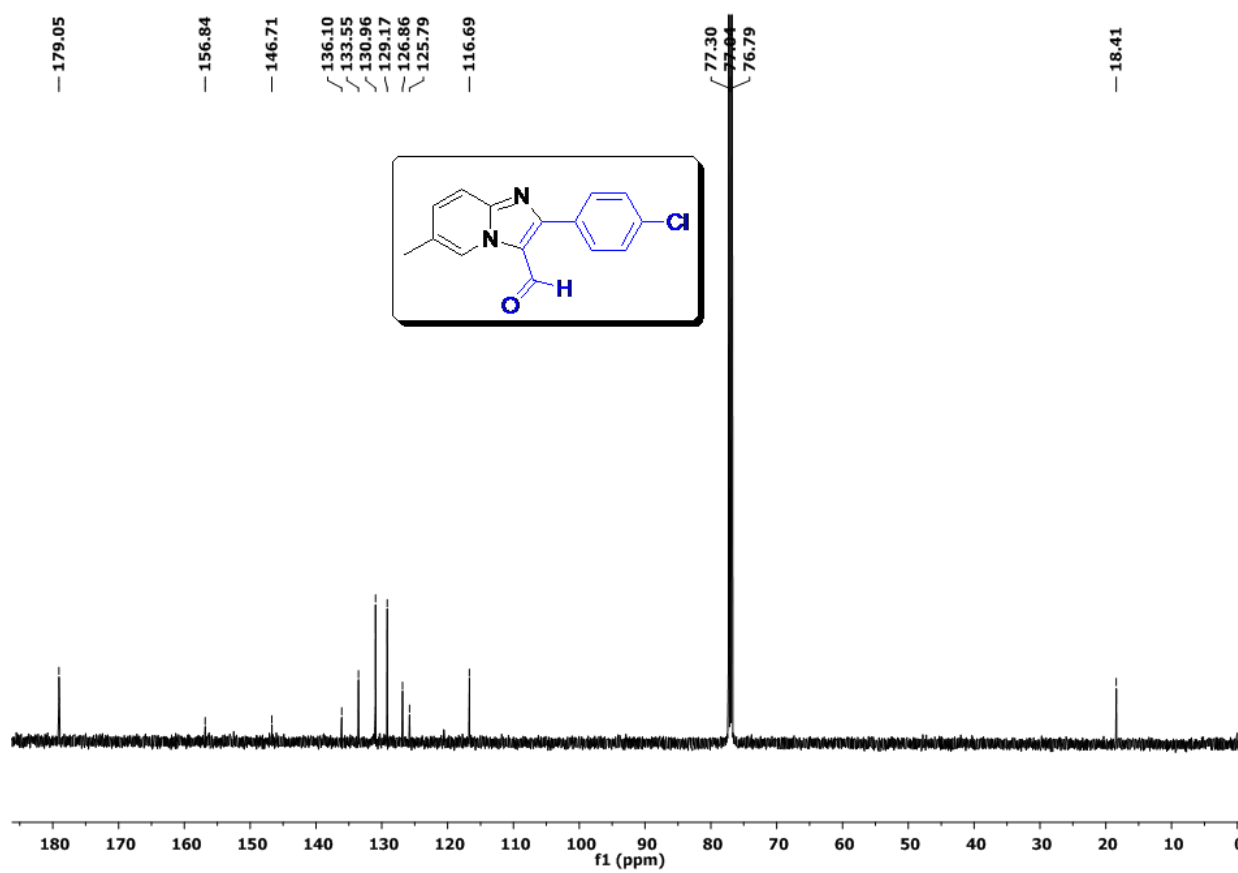


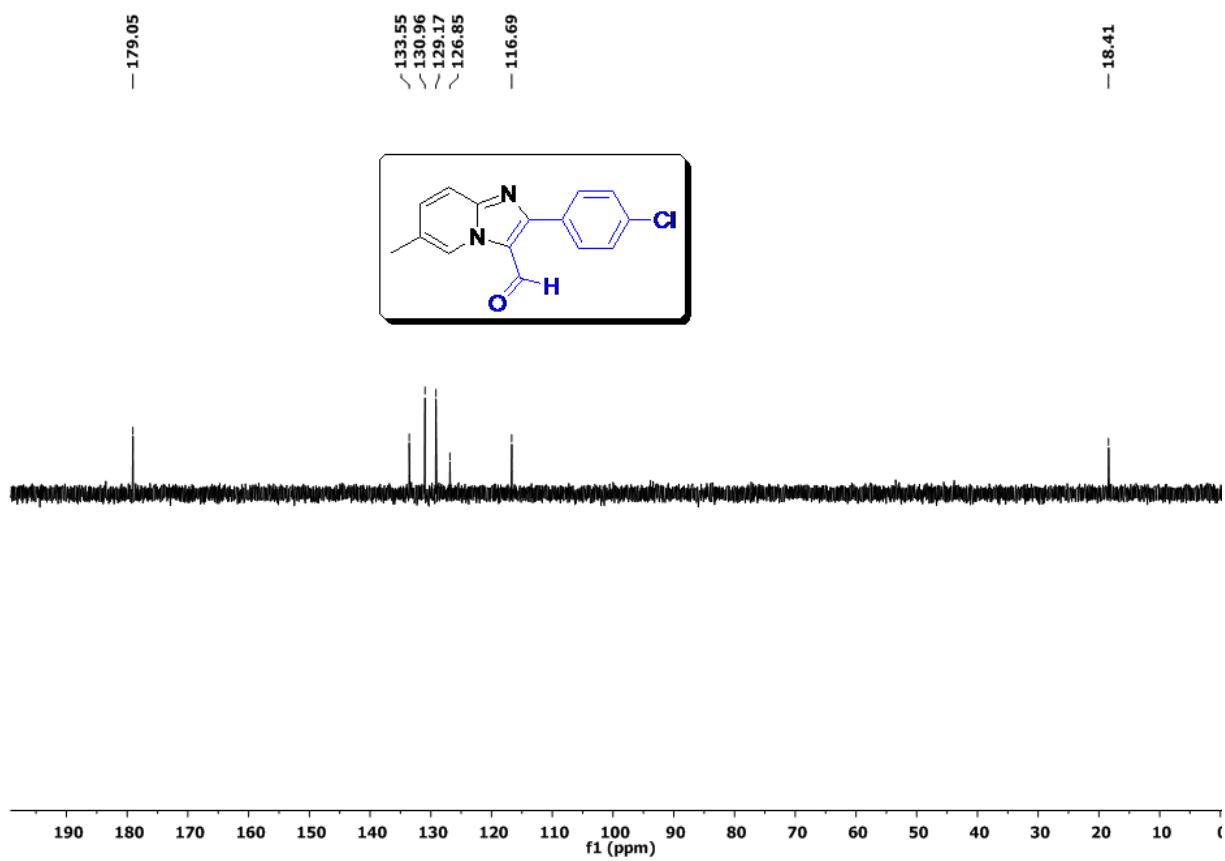




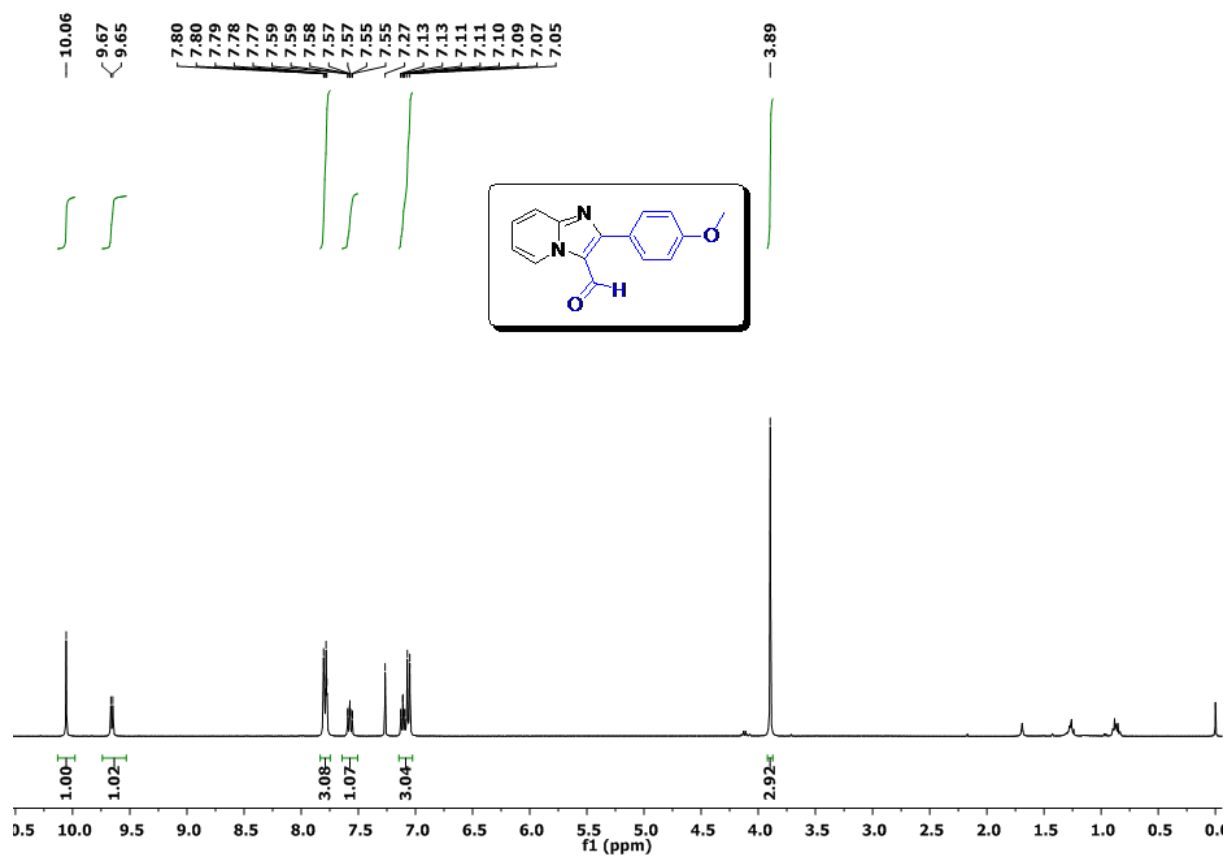
^1H , ^{13}C and DEPT-135 NMR of 2-(4-chlorophenyl)-6-methylimidazo[1,2-a]pyridine-3-carbaldehyde (3f):

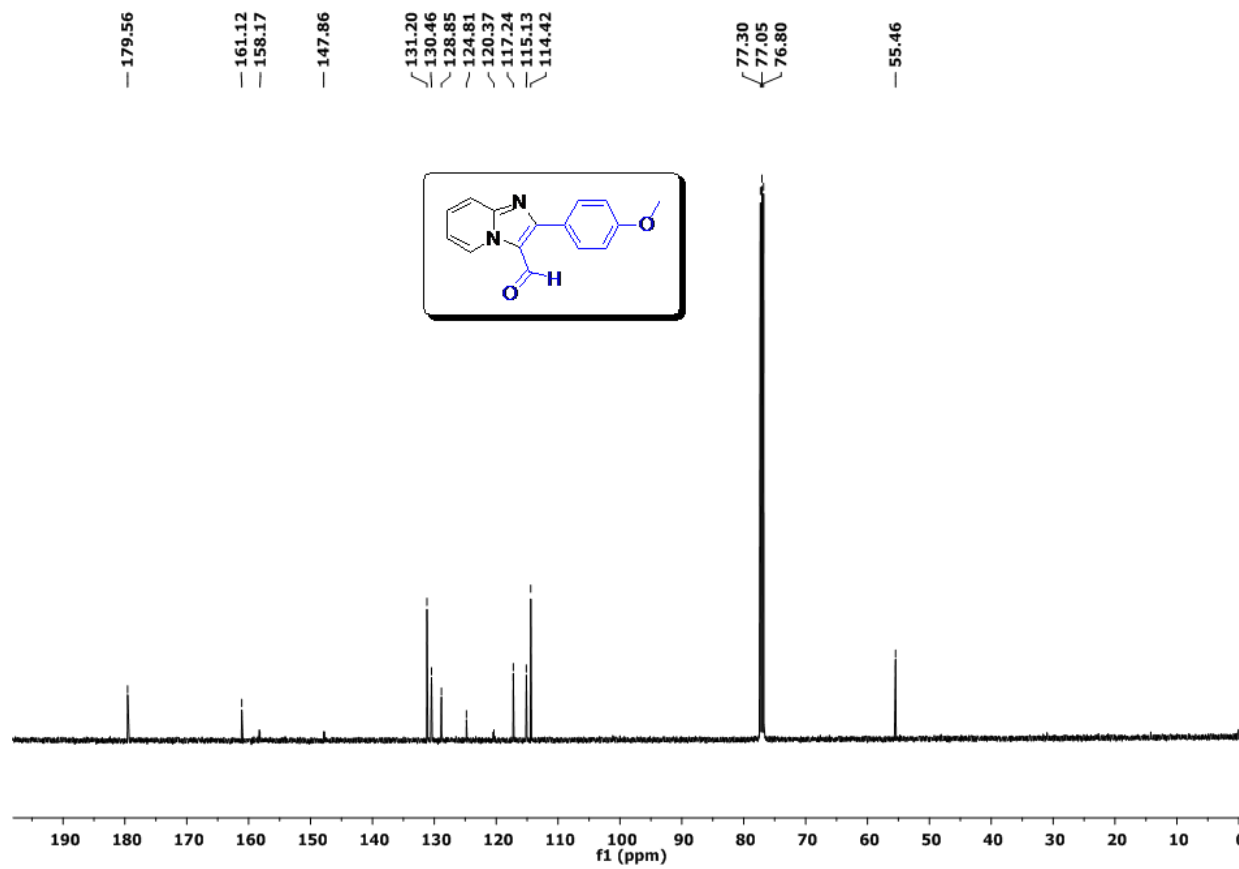


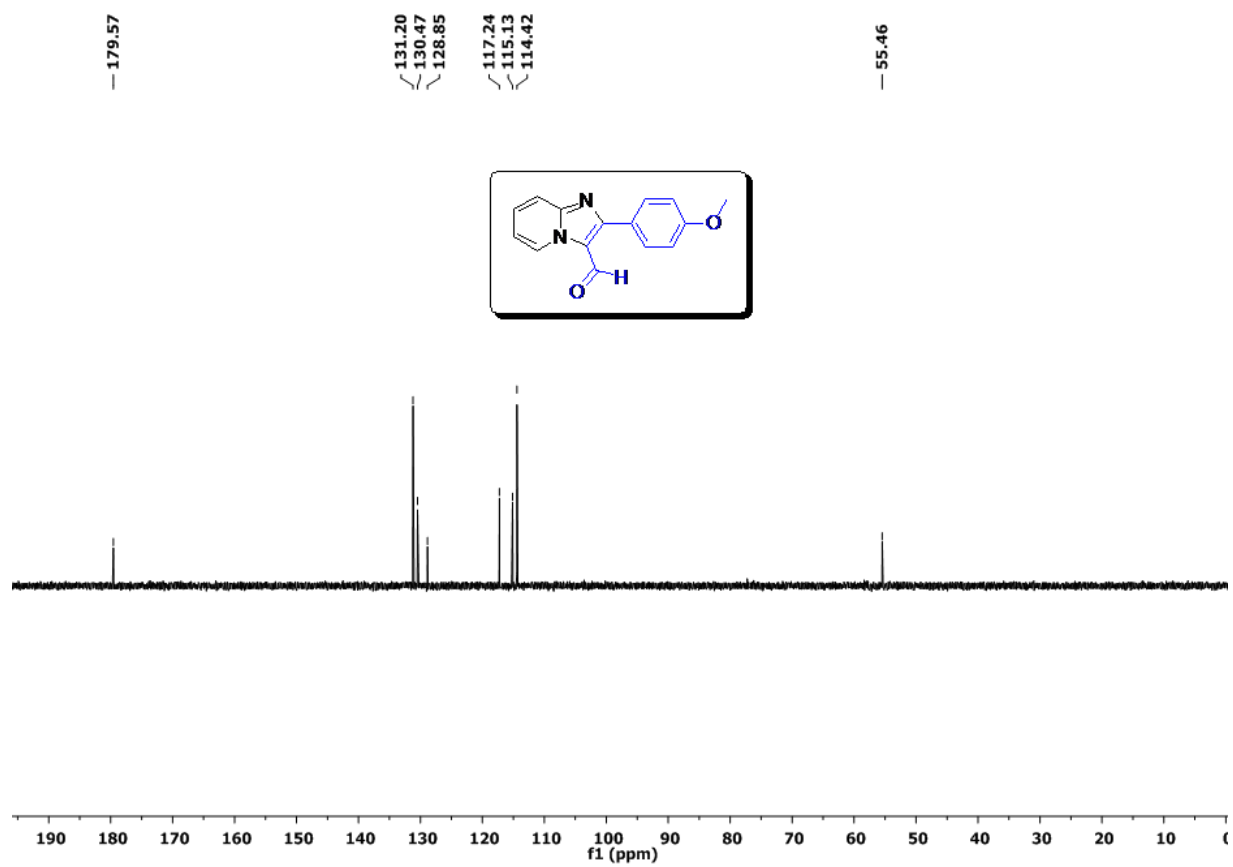




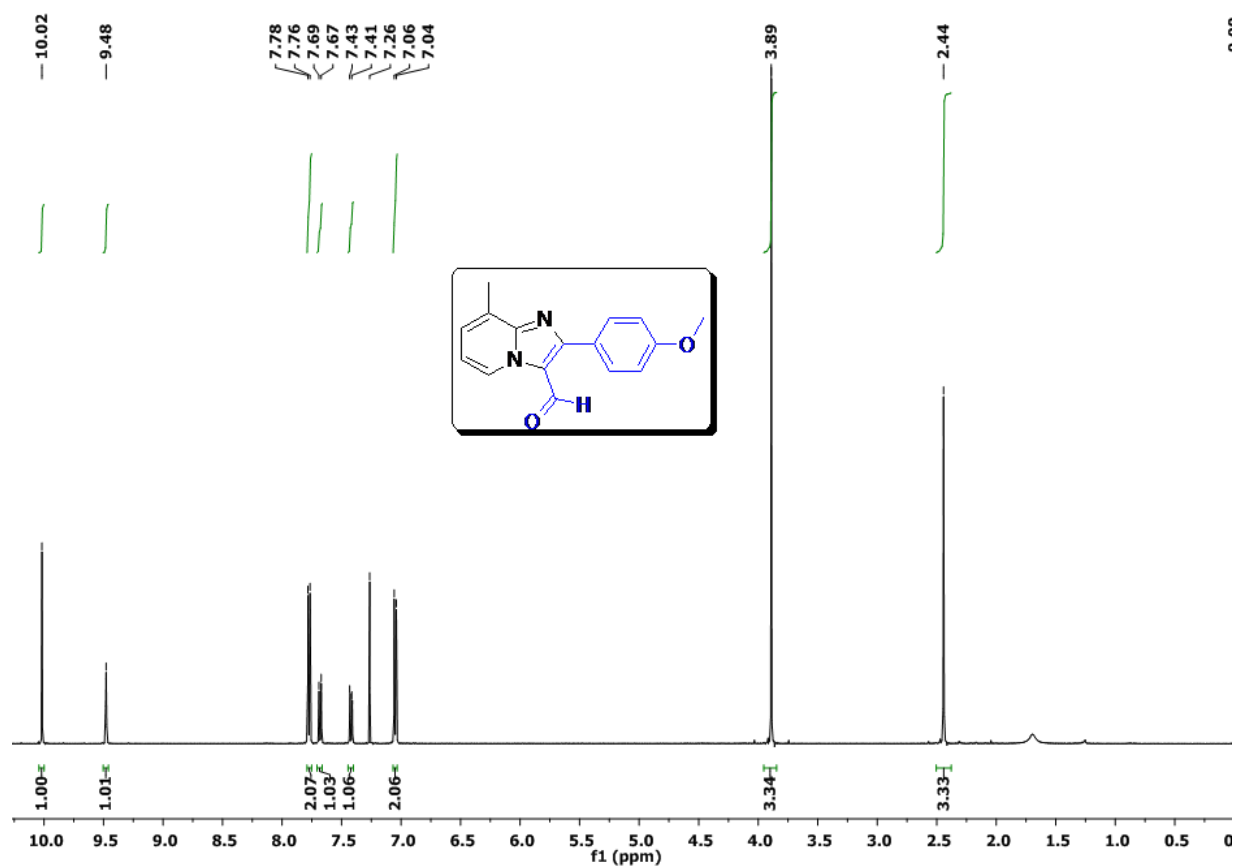
^1H , ^{13}C and DEPT-135 NMR of 2-(4-methoxyphenyl)imidazo[1,2-a]pyridine-3-carbaldehyde (3g):

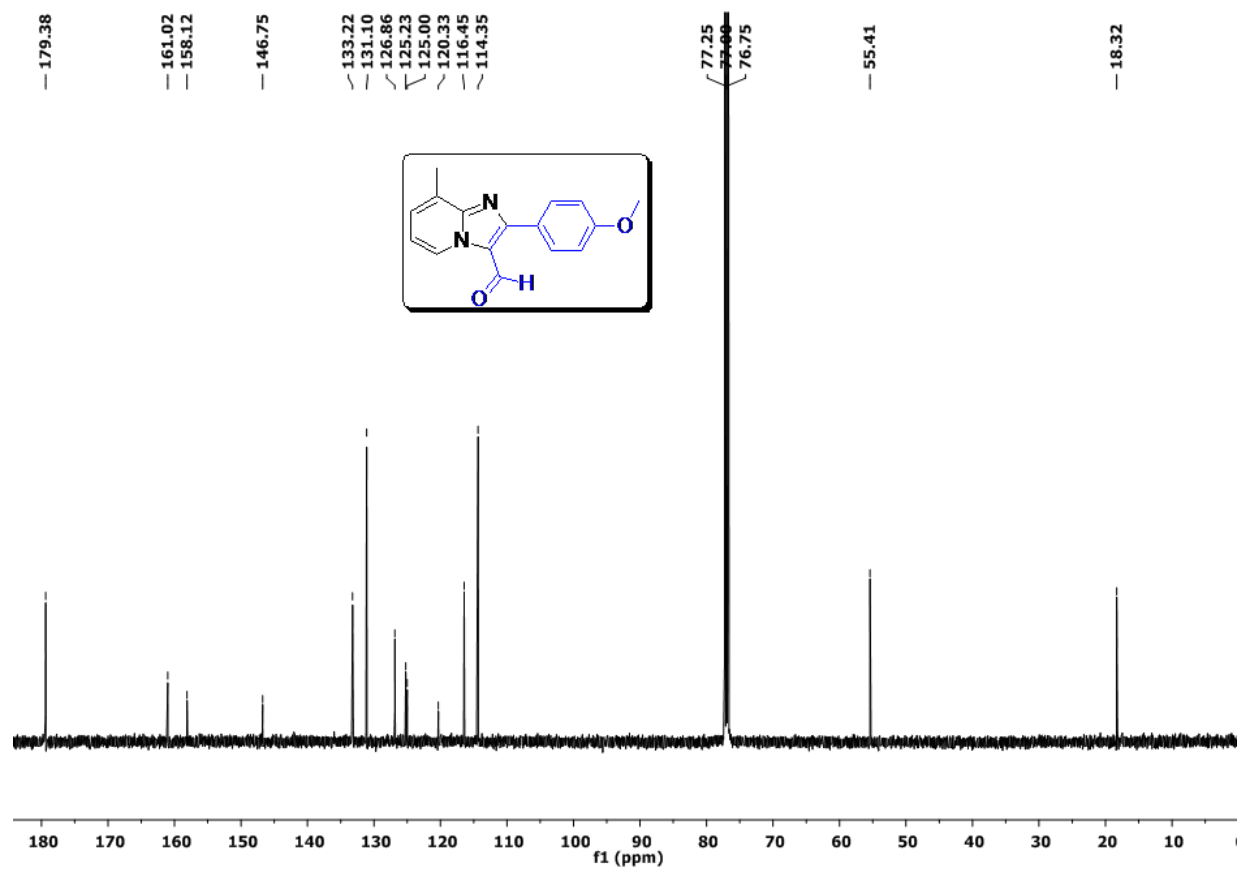


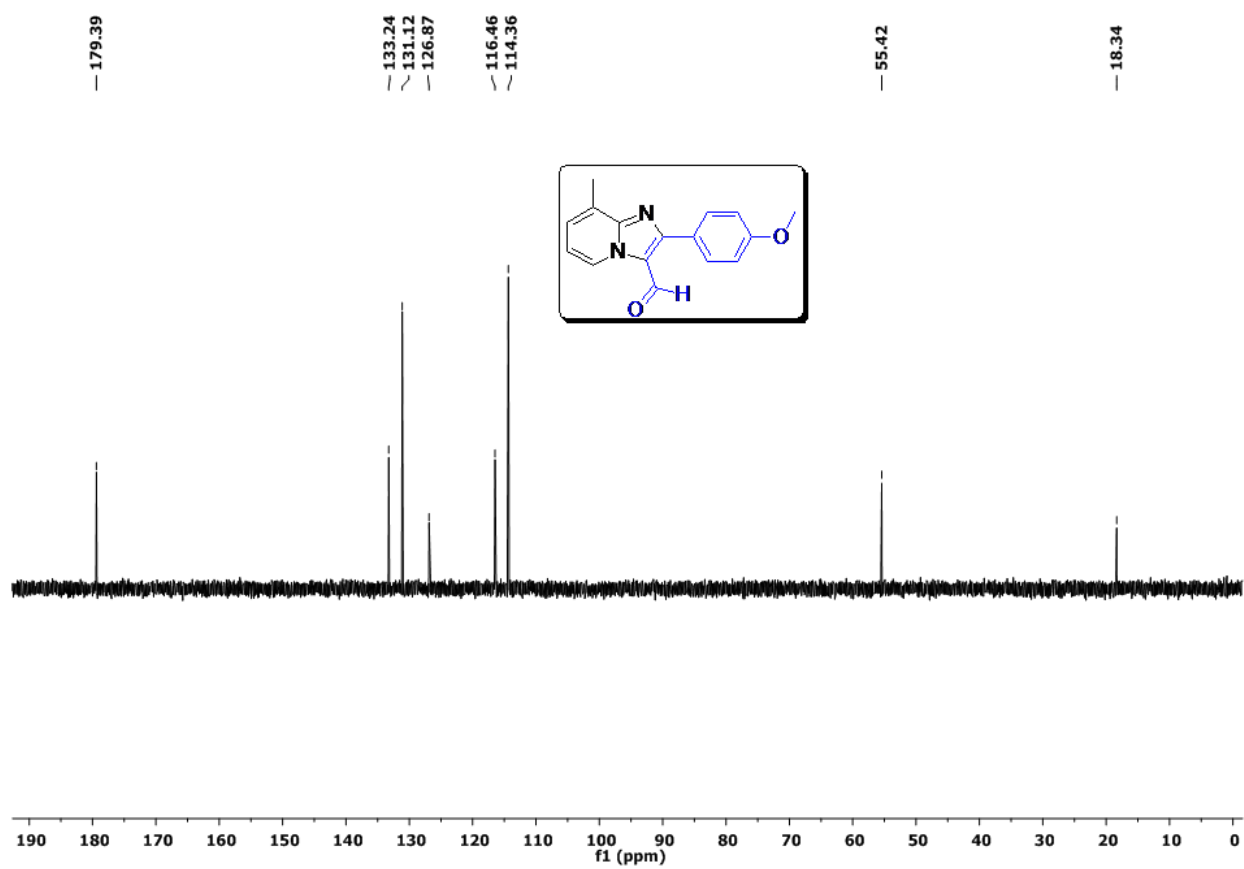




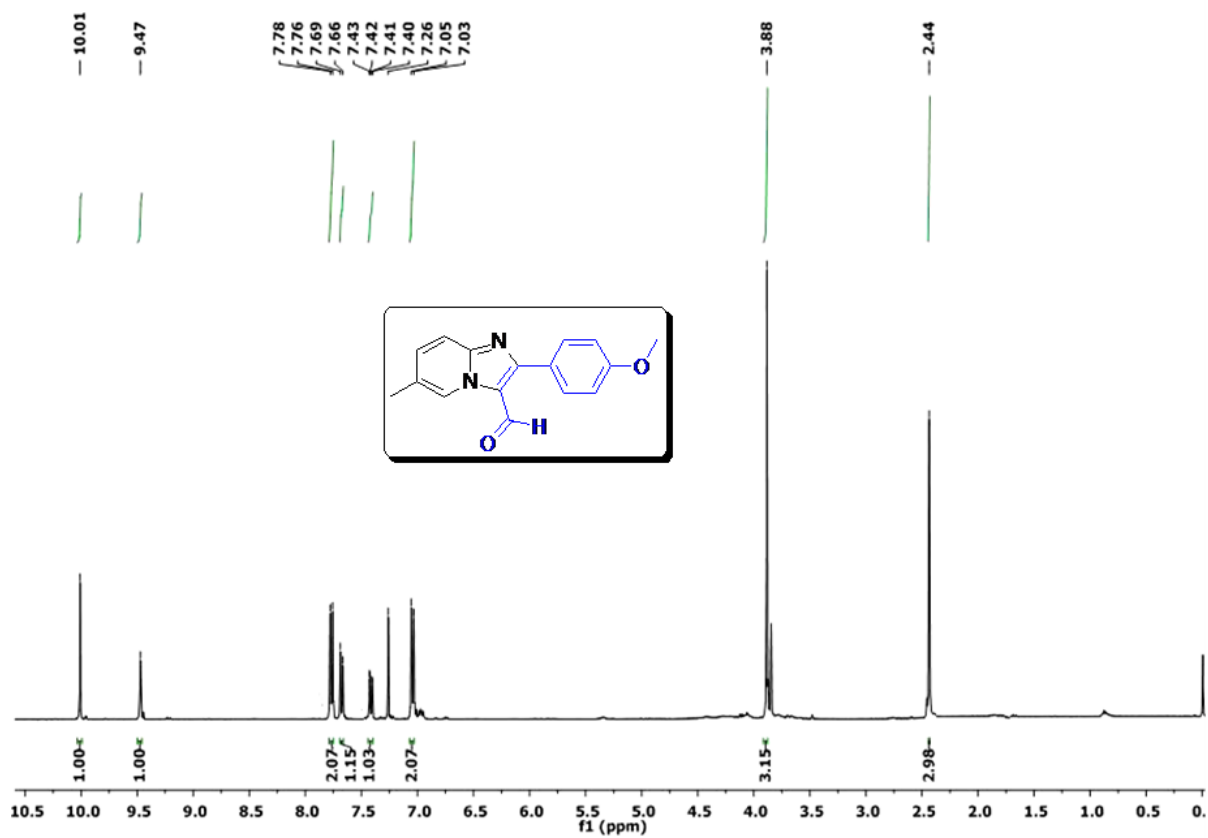
^1H , ^{13}C and DEPT-135 NMR of 2-(4-methoxyphenyl)-8-methylimidazo[1,2-a]pyridine-3-carbaldehyde (3h):

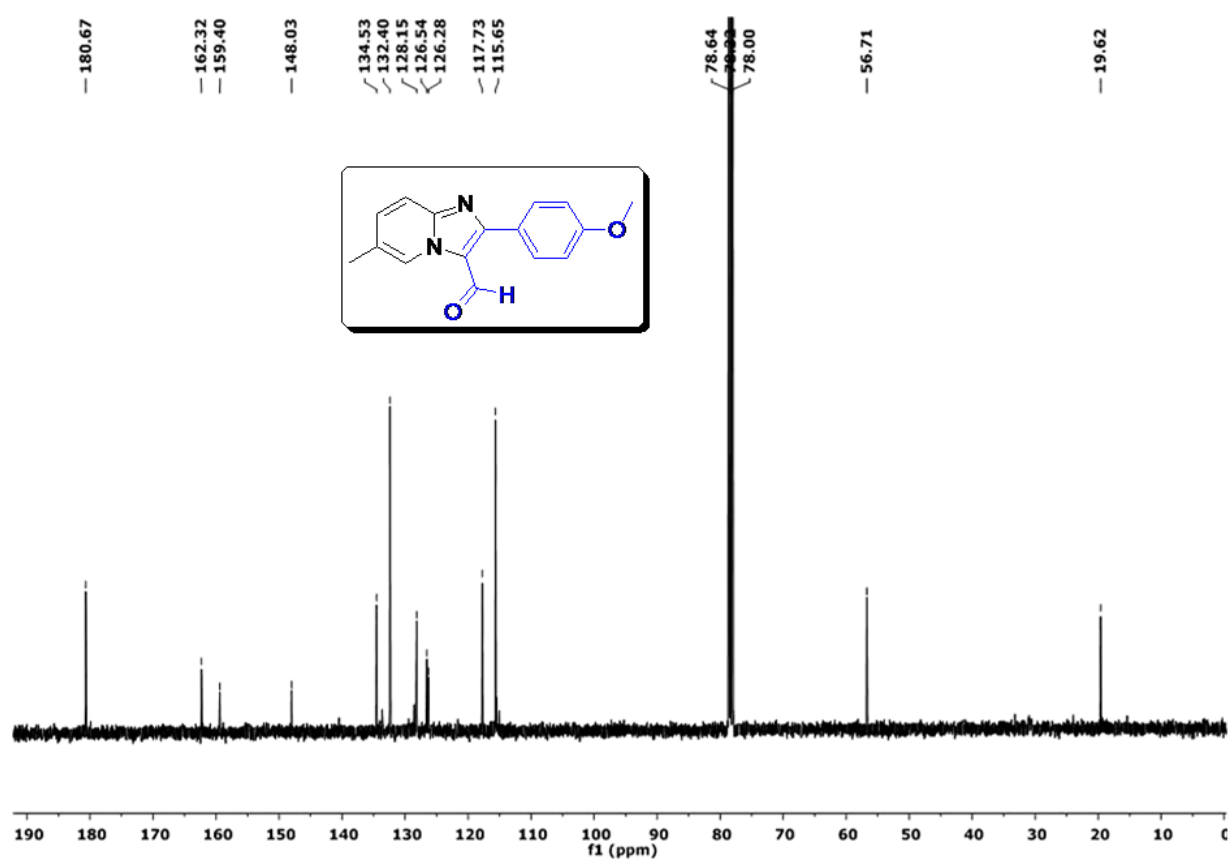


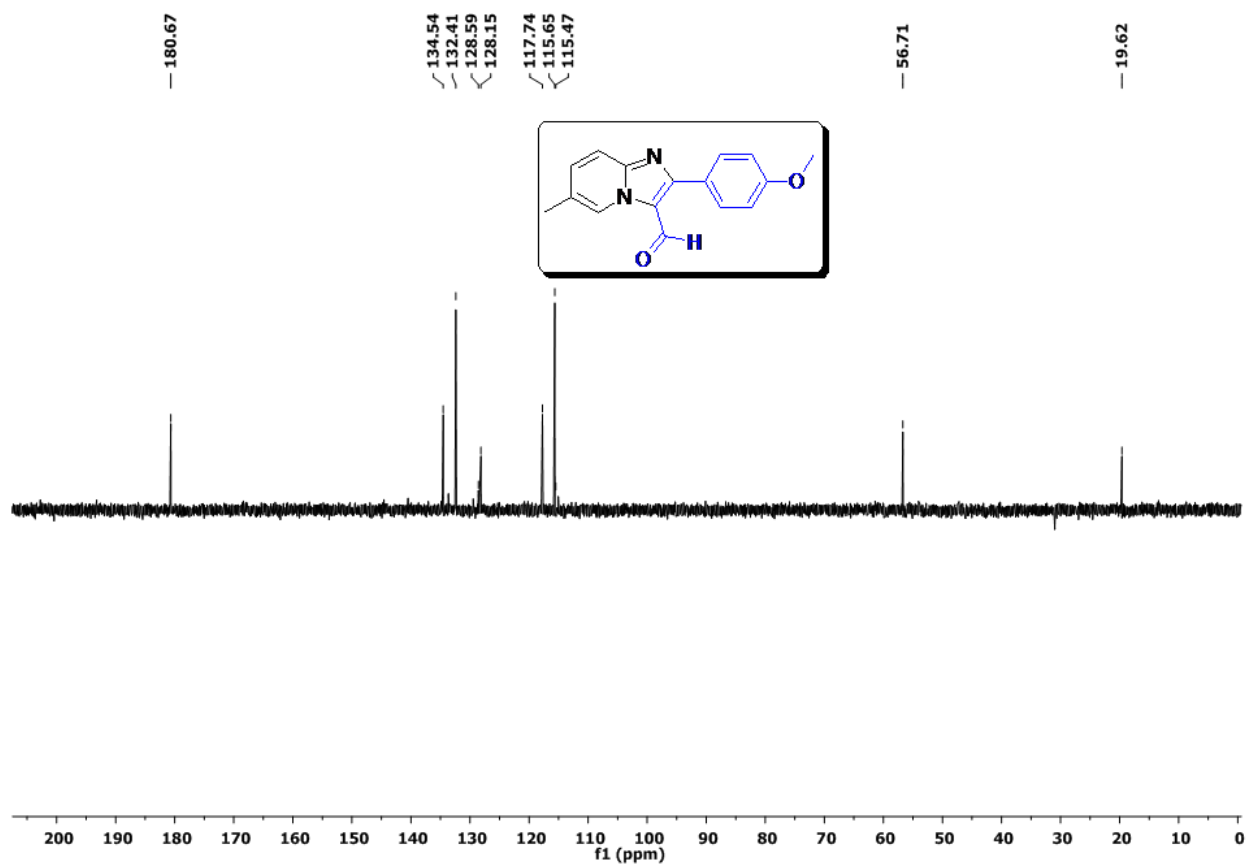




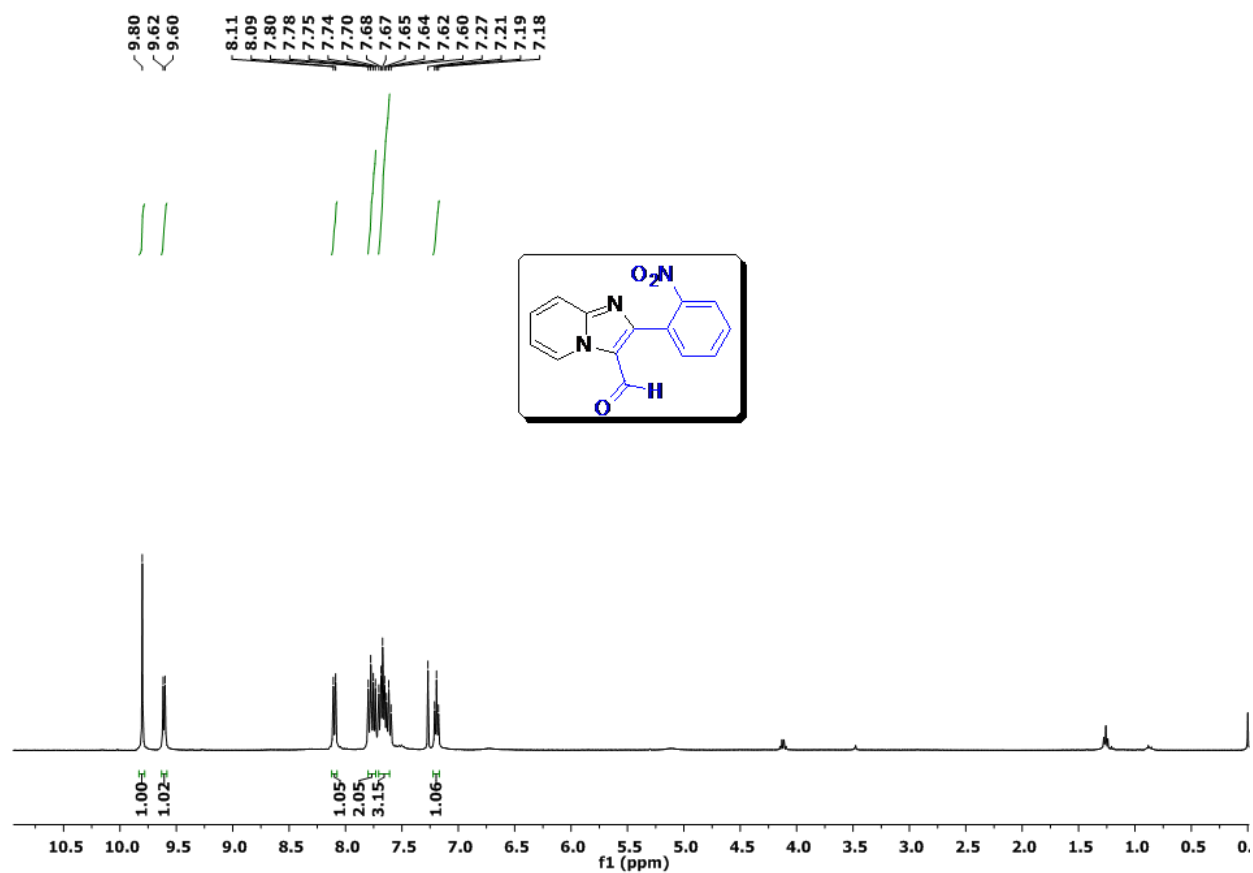
^1H , ^{13}C and DEPT-135 NMR of 2-(4-methoxyphenyl)-6-methylimidazo[1,2-a]pyridine-3-carbaldehyde (3i):

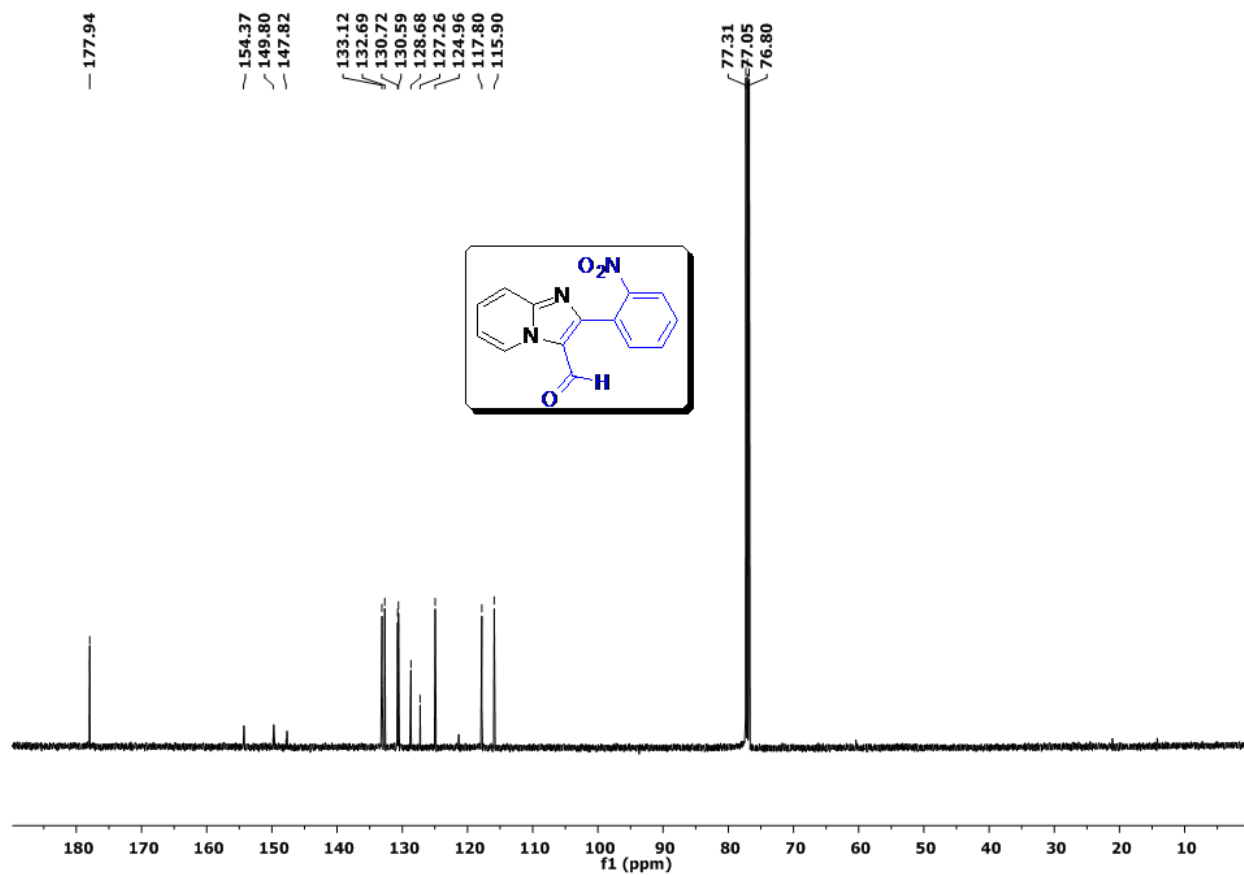


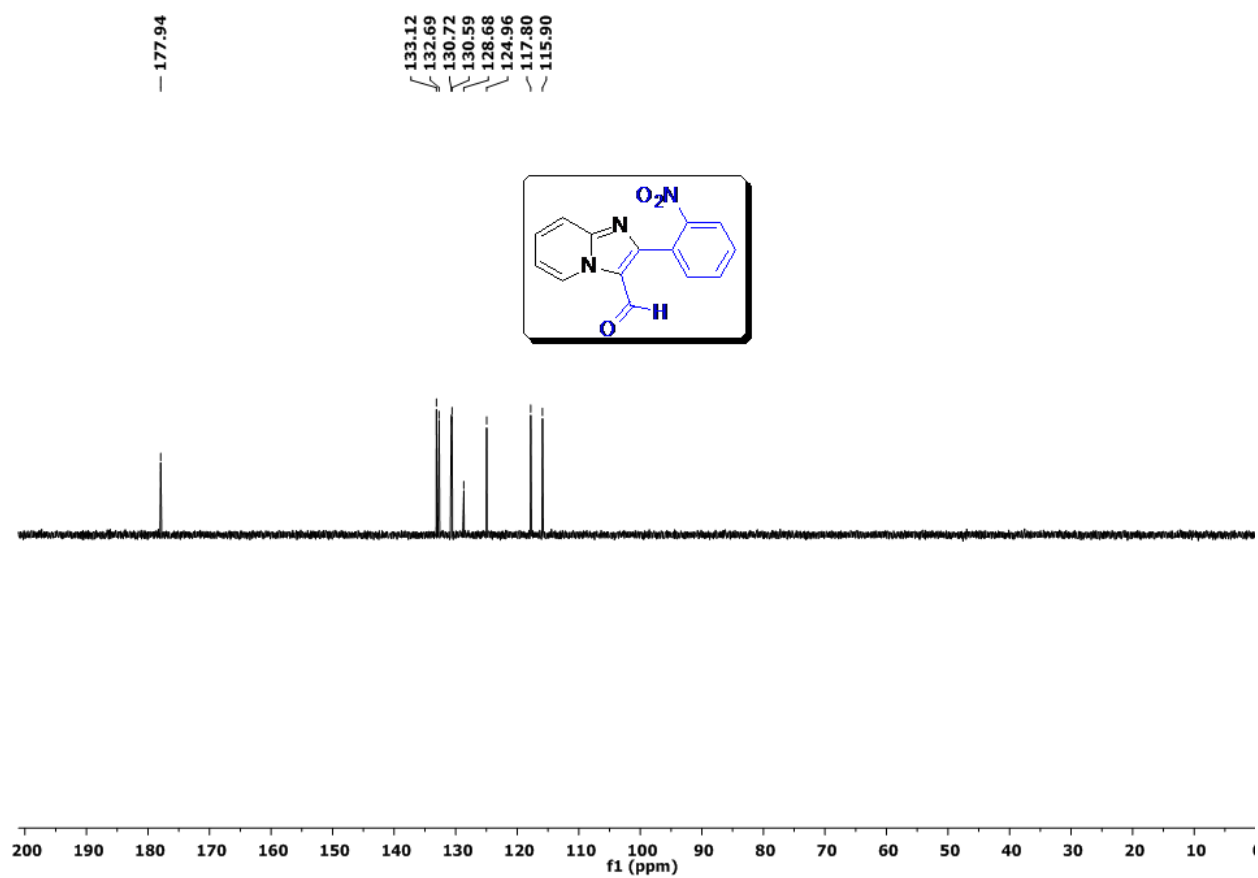




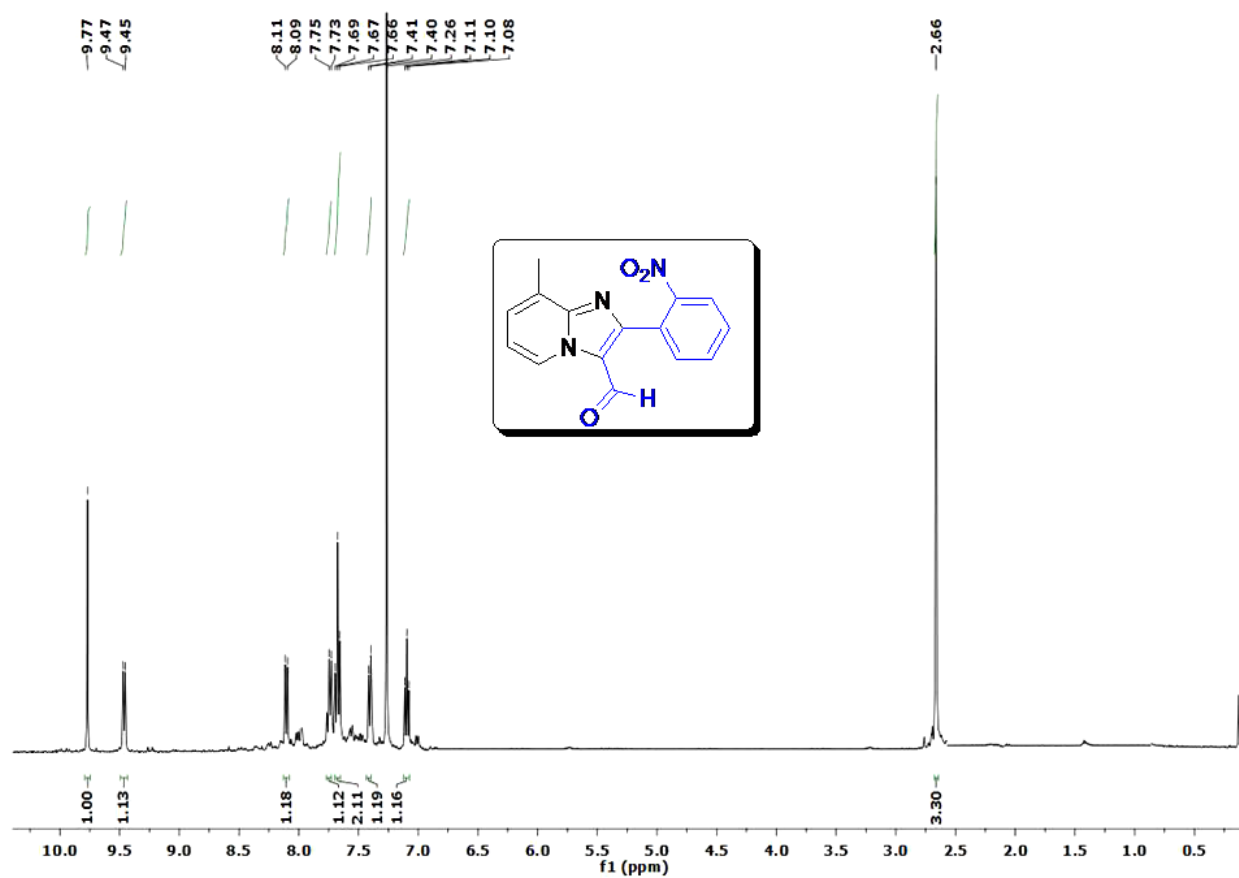
^1H , ^{13}C and DEPT-135 NMR of 2-(2-nitrophenyl)imidazo[1,2-a]pyridine-3-carbaldehyde (3j):

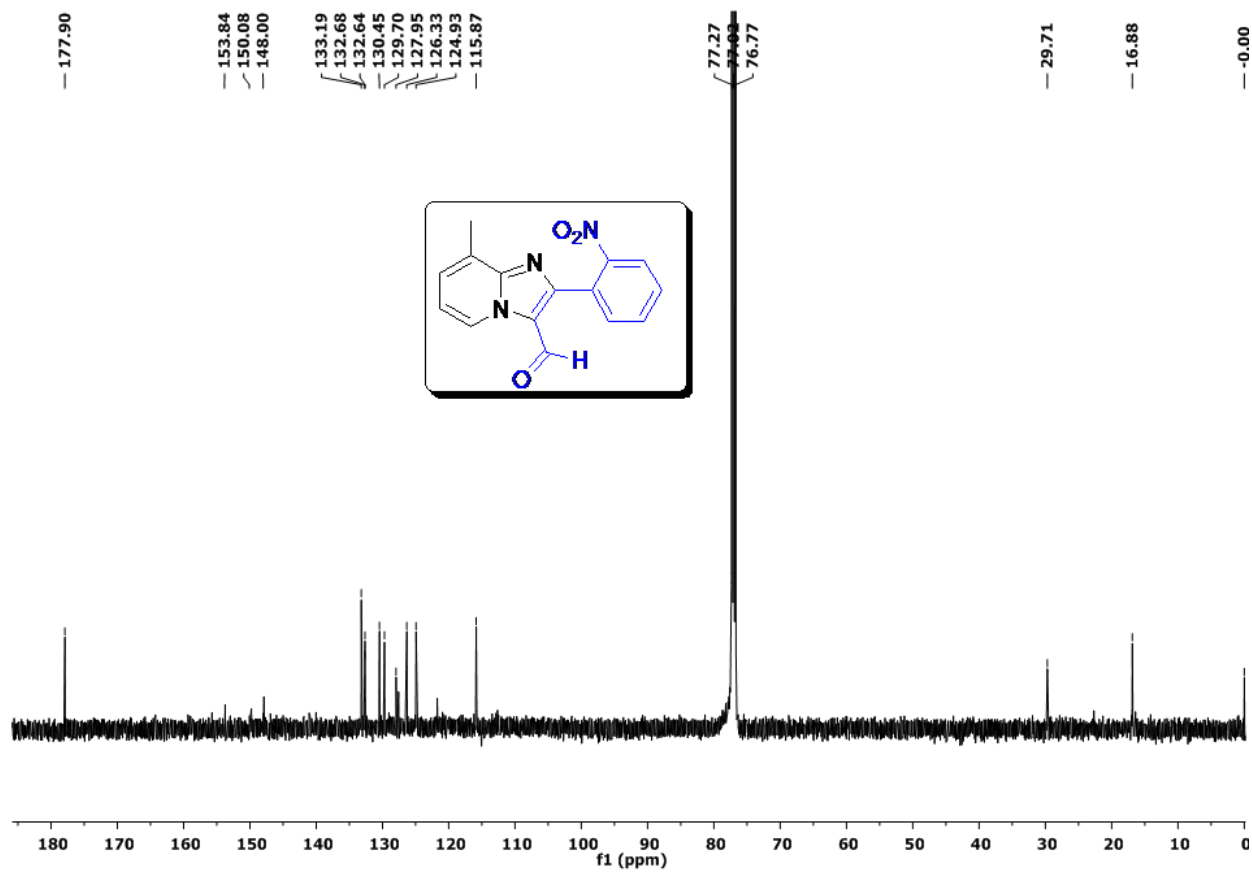


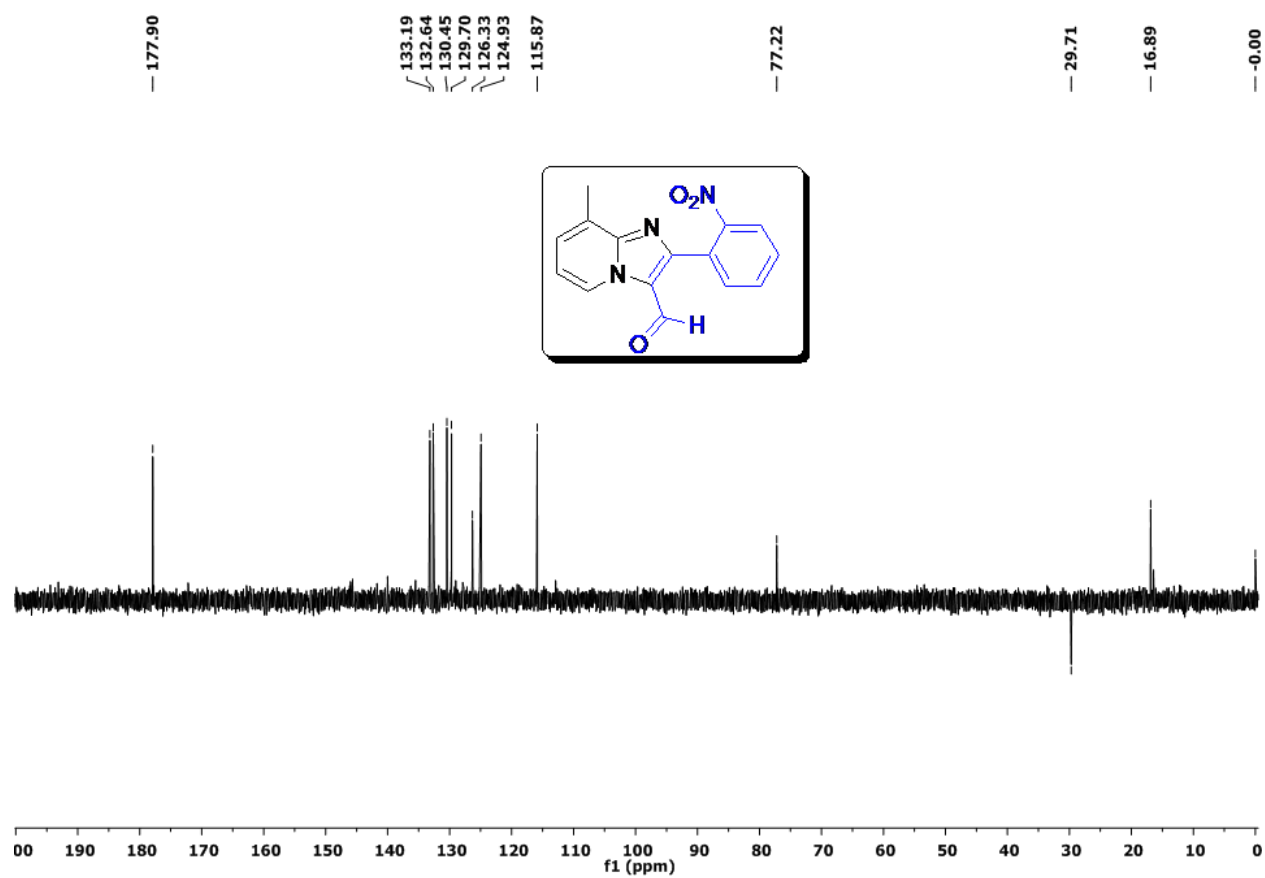




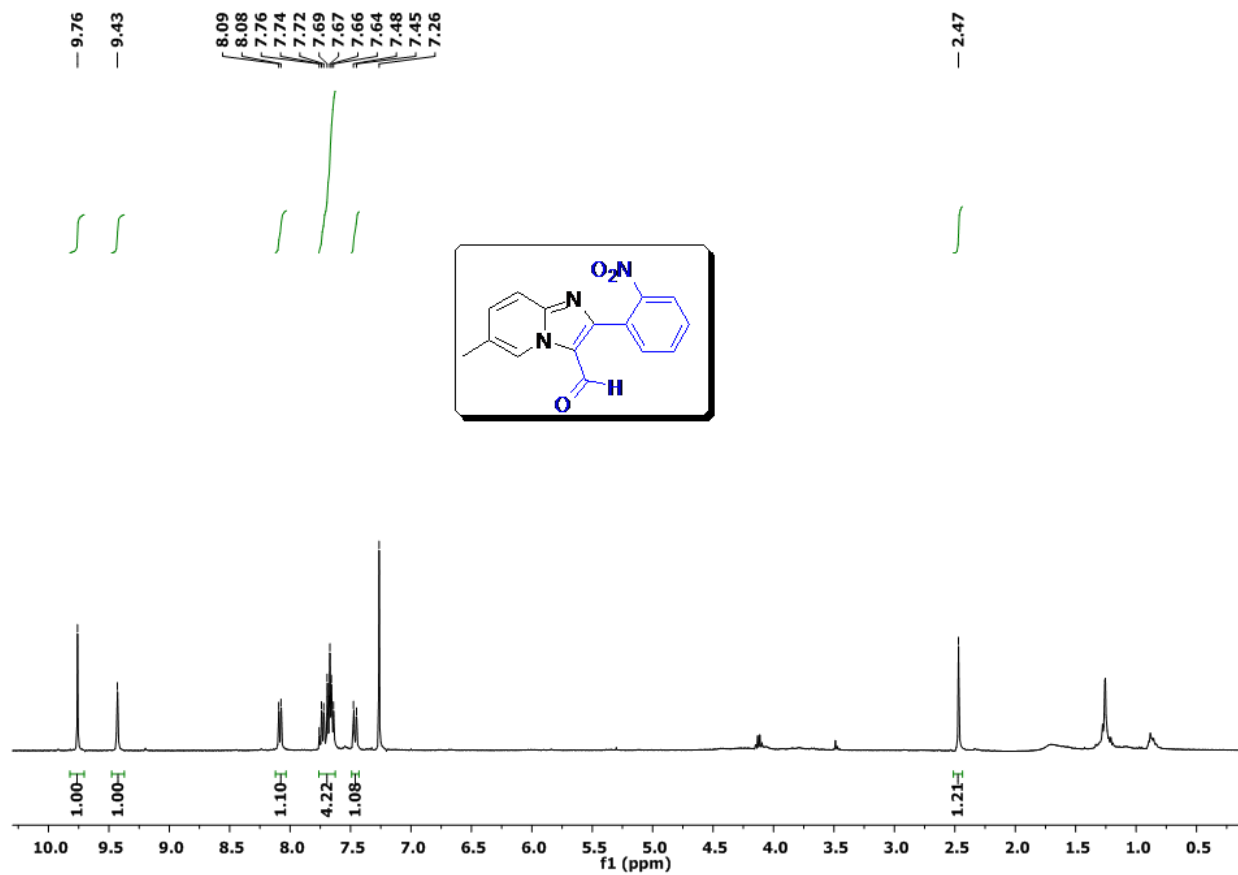
^1H , ^{13}C and DEPT-135 NMR of 2-(2-nitrophenyl)imidazo[1,2-a]pyridine-carbaldehyde (3k):

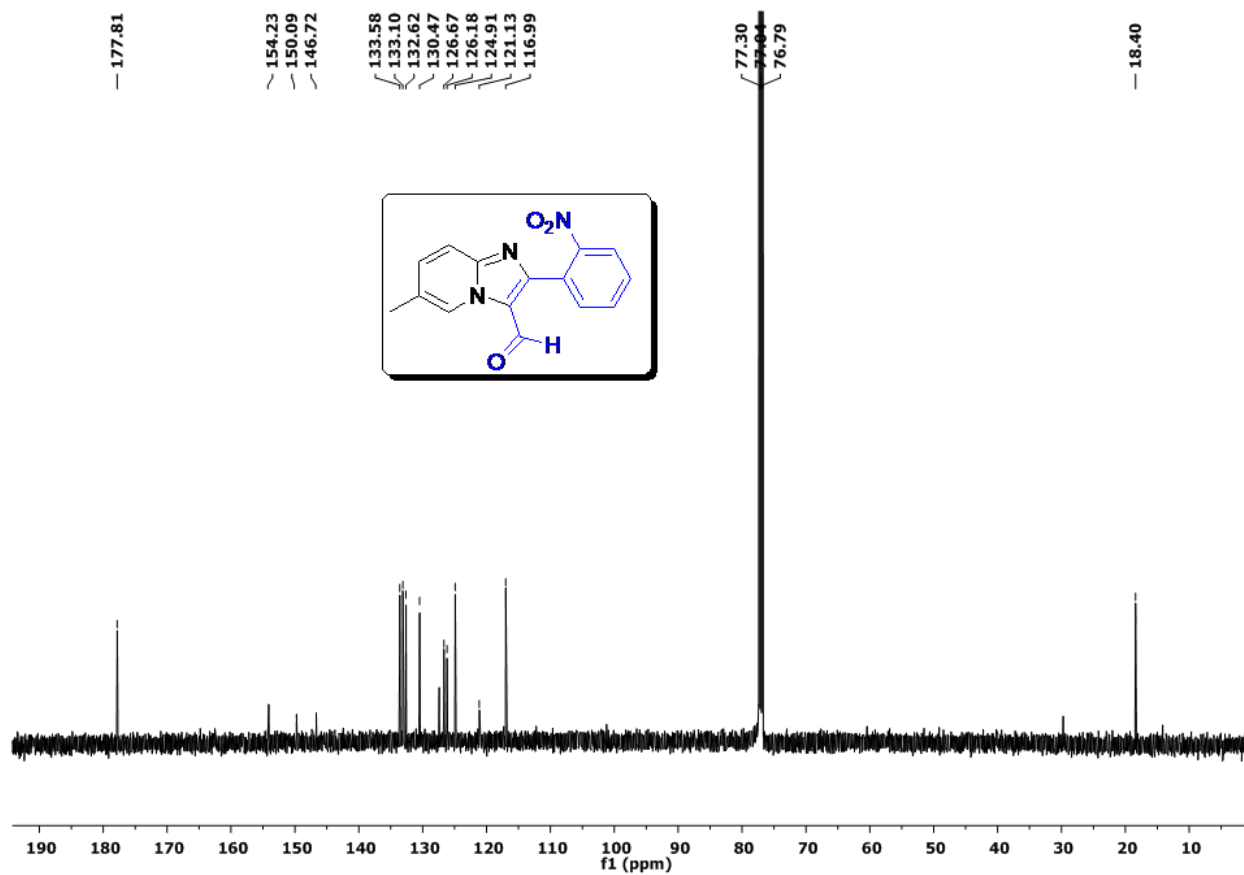


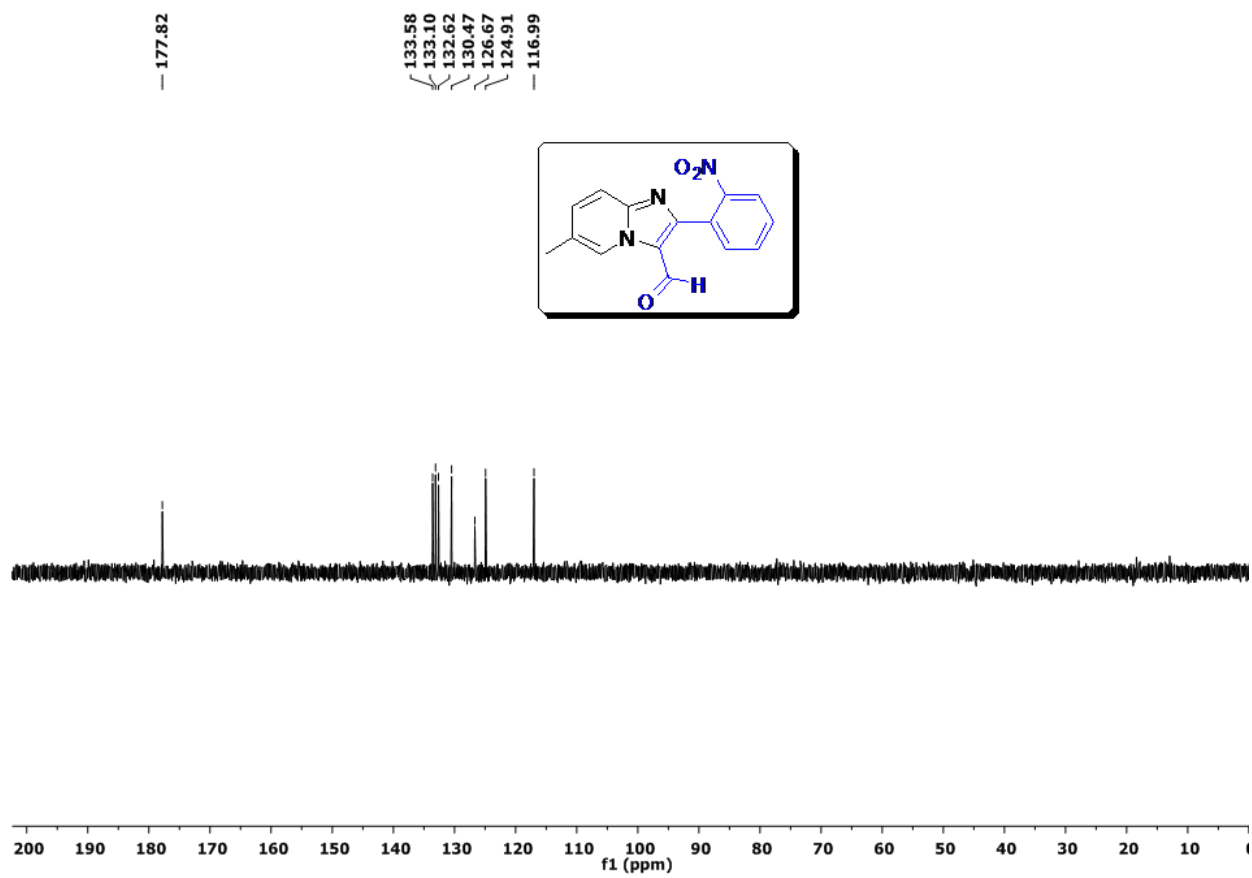




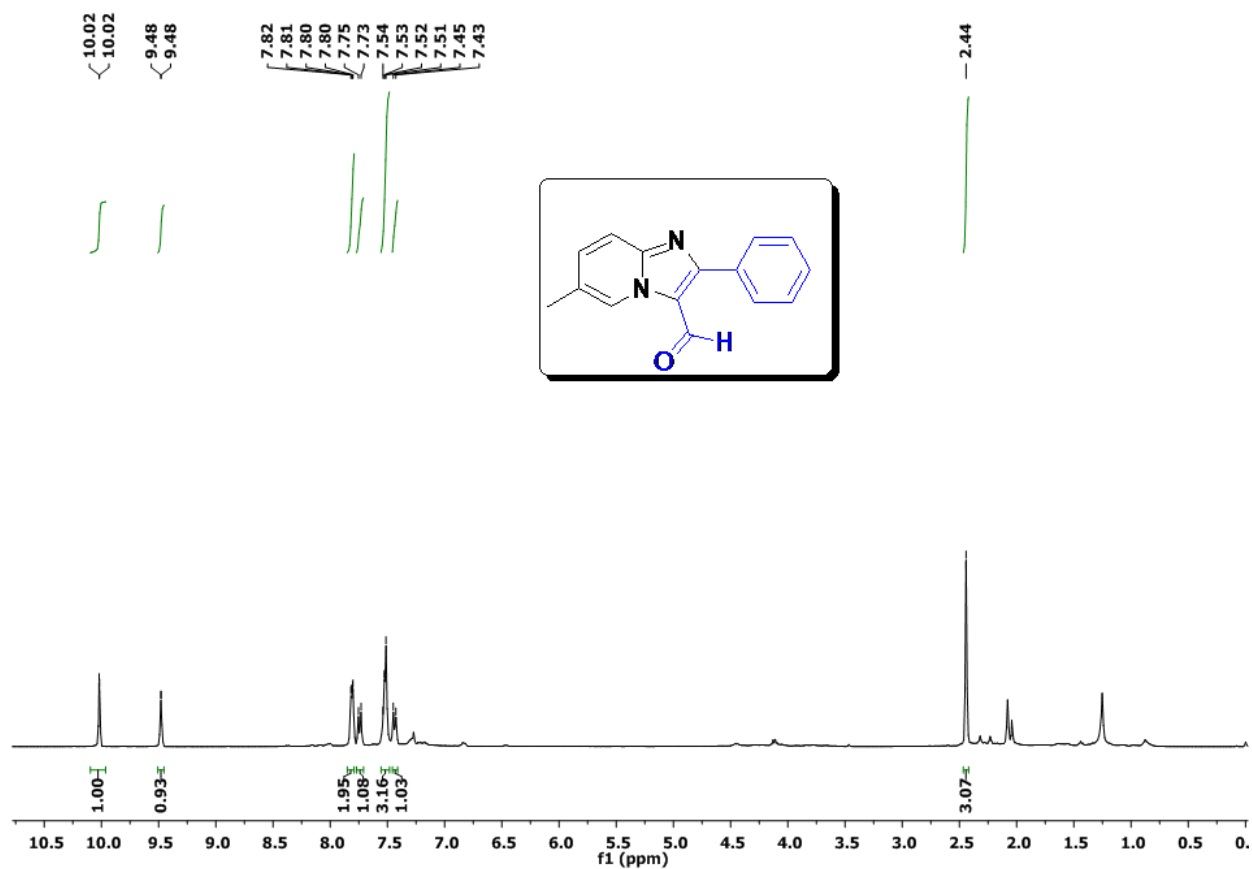
^1H , ^{13}C and DEPT-135 NMR of 6-methyl-2-(2-nitrophenyl)imidazo[1,2-a]pyridine-3-carbaldehyde (3l):

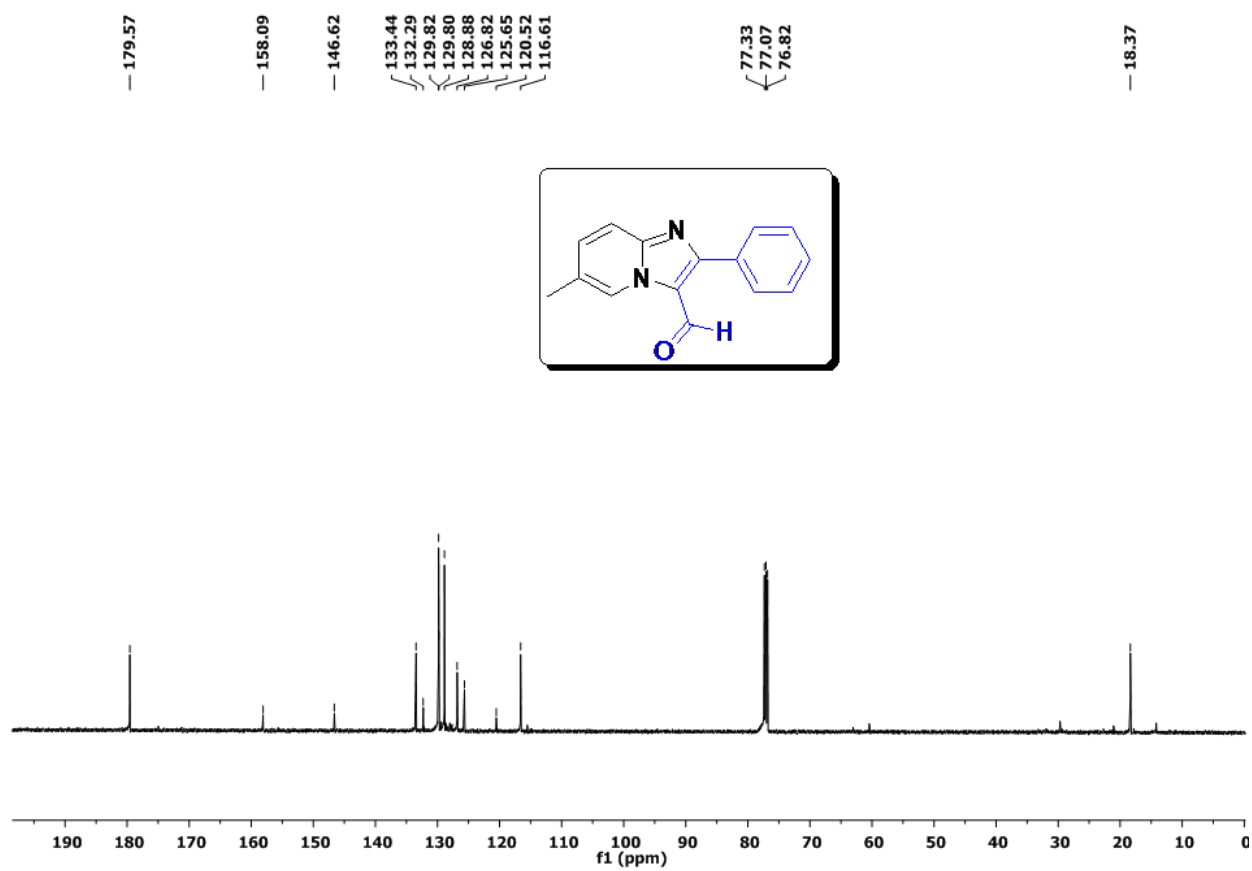


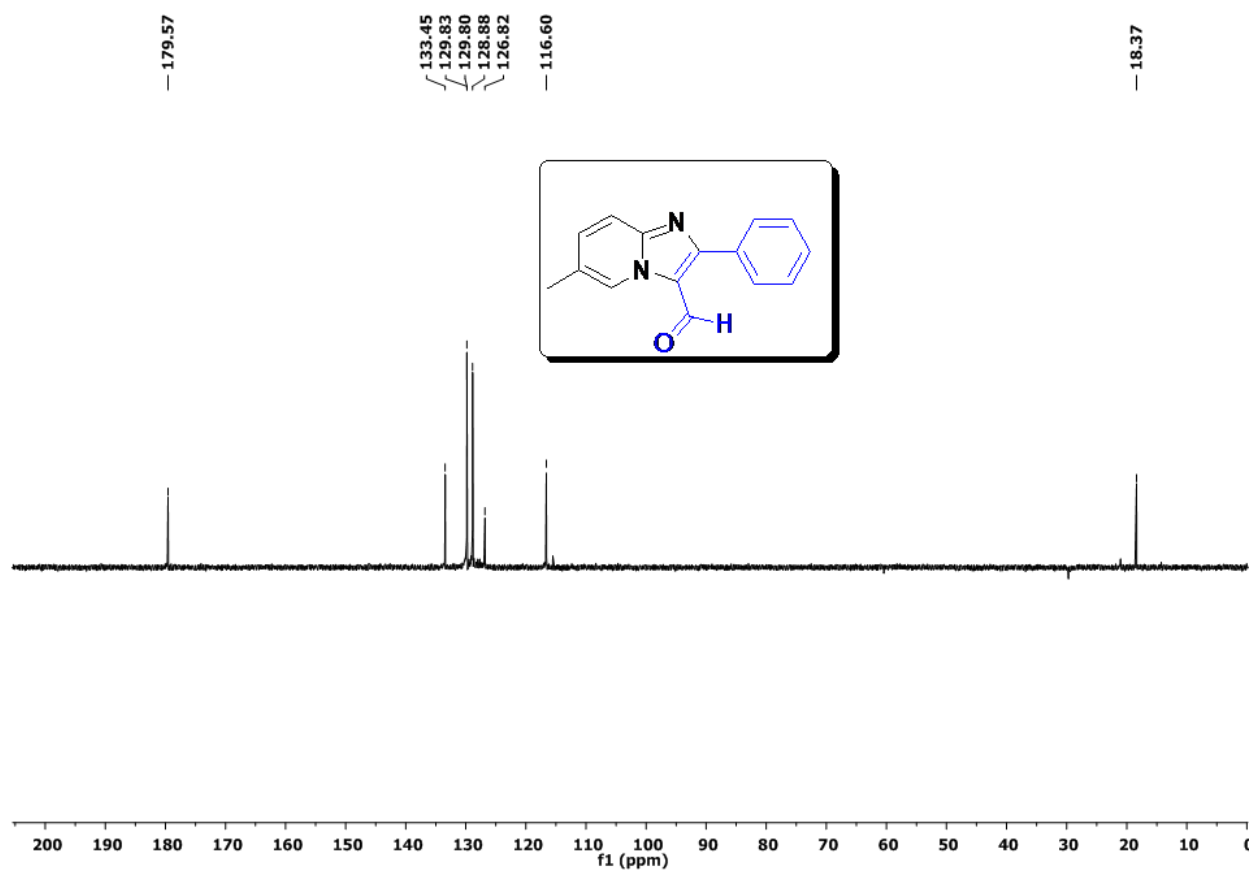




^1H , ^{13}C and DEPT-135 NMR of 6-methyl-2-phenylimidazo[1,2-a]pyridine-3-carbaldehyde (3m):

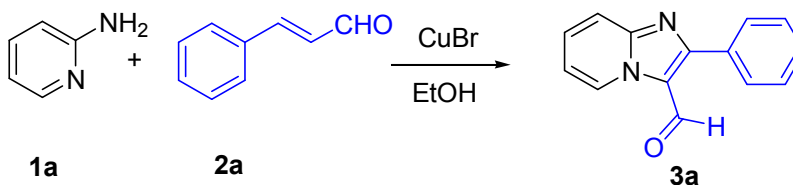






S2. Monitoring of reaction color change

To the mixture of 2-aminopyridine (**1a**, 1.0 equiv.) and cinnamaldehyde (**2a**, 1.2 equiv.) in ethanol was added 10 mol% CuBr and reaction mixture was stirred at 60 °C for 8 h. The initial green color of the reaction was changed to red at the completion (after 8 h), indicating the conversion of CuBr (green) to Cu₂O (red).



Reaction mixture at start of the reaction

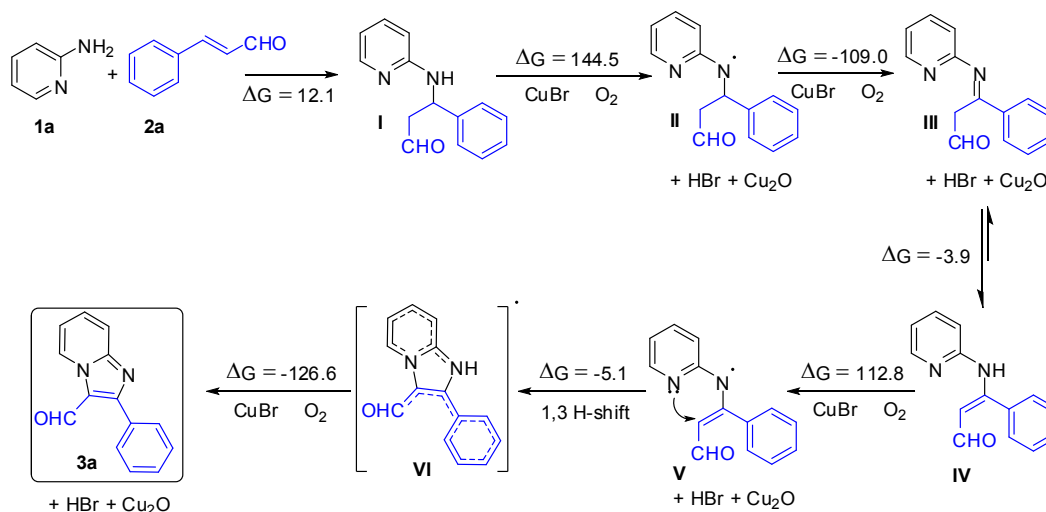


Reaction mixture after 8 h



S3. Table of Gibbs free energy values calculated using B3LYP/6-311+G(d,p) method, for all the reactants, intermediates and products involved in the reaction mechanism

No.	Compound	Gibbs free energy values (Hartrees)
1	1a	-303.662219
2	2a	-422.984557
3	3a	-724.261612
4	I	-726.627541
5	II	-725.977056
6	III	-725.427576
7	IV	-725.433823
8	V	-724.795972
9	VI	-724.8041
10	CuBr	-4214.719201
11	O₂	-150.324188
12	HBr	-2574.76639
13	Cu₂O	-3356.253546



Plausible mechanism for the preparation of 3-formyl-2-phenyl-imidazo[1,2-a]pyridine **3a**. The Gibbs free energy values at each step are given in kcal/mol.