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Synthesis, Relaxation Properties and *In vivo* Assessment of Carborane-GdDOTA-Monoamide Conjugate as MRI Blood Pool Contrast Agent

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Scheme 1 Synthesis of **CB-GdDOTA-MA**. Reagents and conditions: a) Et₂S, Toluene, 85 °C, 2 h. b) Acetonitrile, 65 °C, 12 h, 41%. c) DO3A-*t*-Bu-ester, *N*, *N*-diisopropylethylamine, 85 °C for 12 h, 60%. d) Formic acid, 60 °C, 12 h, 82%. e) GdCl₃.6H₂O, H₂O-MeOH (1:1), pH 6.5, RT, 12 h, 73%. f) DyCl₃.6H₂O, H₂O-MeOH (1:1), pH 6.5, RT, 12 h, 68%.







































CB-GdDOTA-MA

Analytical HPLC analysis of CB-GdDOTA-MA
Instrument: Alliance Waters 2695.
Column: Agilent Zorbax SB-C18, 5 μm (4.6 x 250 mm).
Eluent: 1 mL/min; gradient: B 0% to 100% over 45 min, B 100% 45-60 min (A: water, B: acetonitrile).
Detector: Evaporative Scattering Detector (SEDEX 75, operated at 45 °C and 3.5 bars using N₂).



Fig. S1 IR spectrums of compound 4 and CB-GdDOTA-MA.



Fig. S2 HPLC trace of CB-GdDOTA-MA.



Fig. S3 Determination of Hydration number (q) for CB-DyDOTA-MA.

Determination of hydration number (q) for CB-DyDOTA-MA: Varying concentrations of **CB-DyDOTA-MA** and $DyCl_3.6H_2O$ over the range 10-80 mmol dm⁻³ were prepared in 80% D_2O-H_2O and the pH of the solutions was adjusted to pH 7.0. The ¹⁷O NMR experiments were performed using a Bruker Ascend 400 MHz NMR instrument at RT with the deuterium signal locked. A graph was plotted between the $\Delta\delta$ and the concentration for both **CB-DyDOTA-MA** and the DyCl₃.6H₂O solutions and slope was obtained. The $\Delta\delta$ value for a complex with the general formula, Dy(ligand)_n(H₂O)_q, is given by the following relation;

 $(\Delta \delta) = q\Delta [Dy(ligand)_n (H_2O)_n]/[H_2O]$

(I)

The slope of a plot of the $\Delta\delta$ versus the Dy³⁺ concentration is proportional to the **q** value of the complex (Figure). The **q** value was obtained by linearly fitting the $\Delta\delta$ value and was found to be 2 for complex **CB-DyDOTA-MA**. From the graph, which fit well to a straight line, the slope was calculated. From relation I, the slope of the graph can be equated with the following; Slope = $q\Delta/[H2O]$, From the graph, the slope was calculated as, Slope = -381 ppm dm³ mol⁻¹ or, $q\Delta/[H2O] = -381 \text{ ppm dm}^3 \text{ mol}^{-1}$ For DyCl₃, the value of **q** was assumed to be 9, because the coordination number of Dy(III) in such complexes known to be 9. Hence, the value of $\Delta/[H2O]$ is calculated as follows: $\Delta/[H2O] = -381 \text{ ppm dm}^3 \text{ mol}^{-1}/9 \text{ or}$ $\Delta/[H2O] = -42.3 \text{ ppm } \text{dm}^3 \text{ mol}^{-1}$ From this calculation, a value of q for the DOTA-Dy³⁺ complex **CB-DyDOTA-MA** is calculated as follows: For complex **CB-DyDOTA-MA**, Slope = $q\Delta/[H2O]$, or $q = slope/(\Delta/[H2O])$ or q = -50.4/-42.3, hence 21 q = 1.2



Fig. S4 Representative *in vivo* T1-weighted MRA scans of a mouse at various time points p.i. of Omniscan at a Gd dose of 1.2 mmol/kg.