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## Short and Highly Efficient Synthesis of Lipid Peroxidation Inhibitor Pyrrolostatin and Some Analogues Thereof

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## **Materials and Methods**

All reagents were used as purchased from commercial suppliers. All dry solvents were purified using a solvent purification system. All reactions were performed under an atmosphere of dry nitrogen unless otherwise mentioned. Reactions were monitored by thin layer chromatography using silica or aluminium oxide precoated aluminium plates and stained with vanillin [vanillin (1 g), H<sub>2</sub>SO<sub>4 conc</sub> (10 mL), AcOH (20 mL) ethanol (170 mL)] or ceric ammonium molybdate [phosphomolybdic acid (25 g), Ce $(SO_4)_2 \cdot 2 H_2O$  (10 g), H<sub>2</sub>SO<sub>4 conc</sub> (60 mL), H<sub>2</sub>O (940 mL)]. Chromatographic purification was performed as flash chromatography on silica gel (particle size 0.040-0.063 mm) or aluminium oxide (particle size 0.060–0.200 mm). Yields refer to chromatographically purified and spectroscopically pure compounds. NMR spectra were recorded on a 300 MHz (300 MHz for <sup>1</sup>H and 75 MHz for <sup>13</sup>C acquisitions), a 400 MHz (400 MHz for <sup>1</sup>H and 100 MHz for <sup>13</sup>C acquisitions), a 500 MHz (500 MHz for <sup>1</sup>H and 125 MHz for <sup>13</sup>C acquisitions) and a 600 MHz (600 MHz for <sup>1</sup>H and 150 MHz for <sup>13</sup>C acquisitions) spectrometer. Chemical shifts  $\delta$  are reported in ppm with tetramethylsilane or the solvent resonance as the internal standard. Coupling constants Jare given in Hertz (Hz). Multiplicities are classified as follows: s = singlet, d = doublet, t =triplet, q = quartet, sept = septet and combinations thereof, or m = multiplet or br = broad signal. Two-dimensional NMRs (H-COSY, HSQC, HMBC) were used for the assignment of all final compounds. High resolution mass spectra were obtained on an ESI-TOF massspectrometer. IR spectra were recorded on a FT-IR spectroscope by attenuated total reflection (ATR). Absorbance frequencies  $\tilde{v}$  are reported in reciprocal centimeters (cm<sup>-1</sup>). The reported melting points are uncorrected. For simplicity, the numbering of the carbon atoms of a given structure does not follow the IUPAC rules.



3-Bromo-1(triisopropylsilyl)pyrrole (2.27 g, 7.52 mmol, 1.00 equiv.) in dry tetrahydrofuran (40 mL) was cooled to -78 °C. *Tert*-butyllithium (1.9 M in heptane, 7.90 mL, 15.0 mmol, 2.00 equiv.) was added dropwise

and the yellowish solution was stirred for 5 min. Freshly prepared trans, trans-geranyl bromide (1.96 g, 9.03 mmol, 1.20 equiv.) in dry tetrahydrofuran (5 mL) was added dropwise over 5 min and the mixture was stirred for 4 h with warming to -50 °C. The reaction was quenched with brine (40 mL), the aqueous layer was extracted with ethyl acetate (3 x 40 mL), the combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub> and the solvents were removed under reduced pressure. The residue was dissolved in tetrahydrofuran (50 mL) and tetra-nbutylammonium fluoride trihydrate (2.85 g, 9.02 mmol, 1.20 equiv.) was added at rt. The reaction was stirred at rt for 30 min and the solvent was removed under reduced pressure. Flash chromatography (hexanes/ethyl acetate/ $Et_3N = 85:10:5$ , silica) of the residue gave 3geranyl pyrrole (1.45 g, 7.14 mmol, 95%) together with some impurities as a yellowish oil. The crude product was dissolved in dry diethyl ether (20 mL) and trichloroacetyl chloride (796 µL, 7.14 mmol, 1.00 equiv.) was added dropwise over 2 min. After stirring for 2 h at rt, the reaction mixture was quenched with brine (20 mL). The layers were separated and the aqueous layer was extracted with ethyl acetate (3 x 10 mL). The combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub> and the solvent was removed under reduced pressure. The residue was dissolved in tetrahydrofuran (10 mL), NaOH (2 M, 10 mL) was added and the residue heated at 75 °C for 2 h. The residue was acidified with NaHSO<sub>4</sub> (1 M, 30 mL) and extracted with ethyl acetate (3 x 30 mL). Flash chromatography (hexanes/ethyl acetate/AcOH = 85:10:5, silica) of the residue gave pyrrolostatin (3) along with its regioisomer as a brown residue. Precipitation from dichloromethane with hexane (1:50, 20 mL) gave pyrrolostatin (3) (898 mg, 48%) as a colorless amorphous powder.

**mp:** 121 °C; <sup>1</sup>**H NMR** (400 MHz, MeOH- $d_4$ ):  $\delta = 6.70$  (br s, 1 H, H-5), 6.66 (br s, 1 H, H-3), 5.35–5.28 (m, 1 H, H-11), 5.15–5.07 (m, 1 H, H-7), 3.17 (d, J = 7.3 Hz, 2 H, H-6), 2.15–2.02 (m, 4 H, H-9, H-10), 1.68 (s, 3 H, H-15), 1.67 (s, 3 H, H-13), 1.60 (s, 3 H, H-14) ppm; <sup>13</sup>C **NMR** (100 MHz, MeOH- $d_4$ ):  $\delta = 164.5$  (C-16), 136.1 (C-8), 132.2 (C-12), 126.2 (C-2), 125.4 (C-7), 125.0 (C-11), 123.5 (C-4), 122.3 (C-5), 116.3 (C-3), 40.8 (C-9), 27.7 (C-10), 26.3 (C-6), 25.9 (C-13), 17.8 (C-14), 16.0 (C-15) ppm; **IR** (ATR):  $\tilde{\nu} = 3345$ , 2968, 2913, 2854, 2641, 1672, 1485, 1434, 1379, 1347, 1289, 1279, 1211, 1114, 965, 872, 836, 762, 698,

631, 599, 582, 563 cm<sup>-1</sup>; **HRMS** (ESI-TOF)  $m/_{Z}$ : [M+H]<sup>+</sup> calcd for C<sub>15</sub>H<sub>23</sub>NO<sub>2</sub> 248.1645; found: 248.1648.

All of these data are in accord with previously published data.<sup>1,2</sup>



Pyrrolostatin (3) (124 mg, 0.50 mmol, 1.00 equiv.) in

uronium hexafluorophosphate (228 mg, 0.60 mmol, 1.20 equiv.) was added and the solution stirred for 30 min. N.N-diisopropylethylamine (209 µL, 1.20 mmol, 2.40 equiv.) and ammonia (7 M in methanol, 5 mL) were added and the reaction stirred for 15 h. After complete conversion of the starting material all volatiles were removed under reduced pressure. Flash chromatography (hexanes/ethyl acetate = 90:10, silica) of the residue gave product 12 (80 mg, 65%) as a white amorphous powder.

**mp:** 80 °C; <sup>1</sup>**H NMR** (400 MHz, CDCl<sub>3</sub>):  $\delta = 9.40$  (br s, 1 H, N-H), 6.73 (br s, 1 H, H-5), 6.44 (br s, 1 H, H-3), 5.65 (br s, 2 H, NH<sub>2</sub>), 5.37–5.28 (m, 1 H, H-11), 5.15–5.07 (m, 1 H, H-7), 3.19 (d, J = 7.2 Hz, 2 H, H-6), 2.14–2.00 (m, 4 H, H-9, H-10), 1.68 (br s, 6 H, H-13, H-15), 1.61 (s, 3 H, H-14) ppm; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 163.0$  (C-16), 135.9 (C-8), 131.6 (C-12), 125.7 (C-2), 124.7 (C-7), 124.4 (C-11), 123.2 (C-4), 120.2 (C-5), 110.5 (C-3), 39.8 (C-9), 26.8 (C-10), 25.9 (C-6), 25.5 (C-13), 17.9 (C-14), 16.1 (C-15) ppm; IR (ATR):  $\tilde{v} = 3394, 3335, 3178, 2968, 2913, 2853, 1636, 1608, 1485, 1430, 1125, 837, 821, 560 \text{ cm}^{-1};$ **HRMS** (ESI-TOF)  ${}^{m/_{Z_{:}}}$  [M+H]<sup>+</sup> calcd for C<sub>15</sub>H<sub>23</sub>N<sub>2</sub>O 247.1805; found: 247.1811.



Pyrrolostatin (3) (62 mg, 0.25 mmol, 1.00 equiv.) was 7 9 10 12 13 14 13 13 15 14 13 15 100 equiv.) was dissolved in dry toluene/methanol (7:1, 8 mL) and stirred at rt. (trimethylsilyl)diazomethane (2.0 M in stirred at rt. (trimethylsilyl)diazomethane (2.0 M in diethyl ether, 150 µL, 0.30 mmol, 1.20 equiv.) was

added and the solution stirred for 2 h. After complete conversion of the starting material all volatiles were removed under reduced pressure. Flash chromatography (hexanes/ethyl acetate = 90:10, silica) of the residue gave product 13 (65 mg, 99%) as a colorless oil.

<sup>1</sup>**H NMR** (600 MHz, CDCl<sub>3</sub>):  $\delta = 8.98$  (br s, 1 H, NH), 6.79–6.68 (m, 2 H, H-3, H-5), 5.37– 5.27 (m, 1 H, H-11), 5.16–5.06 (m, 1 H, H-7), 3.83 (s, 3 H, H-17), 3.19 (d, J = 7.2 Hz, 2 H, H-6), 2.15–2.00 (m, 4 H, H-9, H-10), 1.69 (br s, 3 H, H-13), 1.68 (br s, 3 H, H-15), 1.61 (s, 3 H, H-14) ppm; <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  = 161.8 (C-16), 135.9 (C-8), 131.6 (C-12), 126.0 (C-2), 124.4 (C-7), 123.1 (C-11), 122.4 (C-4), 120.8 (C-5), 115.1 (C-3), 51.5 (C-17),

39.8 (C-9), 26.7 (C-10), 25.8 (C-6), 25.5 (C-13), 17.8 (C-14), 16.1 (C-15) ppm; IR (ATR):  $\tilde{v} = 3314, 2915, 2854, 1682, 1480, 1437, 1395, 1204, 1104, 998, 971, 836, 768 \text{ cm}^{-1}$ ; **HRMS** (ESI-TOF)  $m/_{Z}$ : [M+H]<sup>+</sup> calcd for C<sub>16</sub>H<sub>24</sub>NO<sub>2</sub> 262.1802; found: 262.1805.



 $6^{7} \times 9^{10} \times 11^{14}$  1.20 mmol, 1.20 equiv. was added to a solution of methyl ester 13 (261 mg 1 00 m 1 00 m)DMF (10 mL) at 0 °C and the solution was stirred for

1 h. Methyl iodide (93 µL, 1.50 mmol, 1.50 equiv.) was added and the solution was stirred overnight. After complete conversion of the starting material diethyl ether (20 mL) and H<sub>2</sub>O (20 mL) were added. The phases were separated and the organic phase was washed with H<sub>2</sub>O (3 x 10 mL). The organic phase was dried over Na<sub>2</sub>SO<sub>4</sub> and all volatiles were removed under pressure. The residue dissolved in reduced crude was а mixture of tetrahydrofuran/methanol/H2O (9 mL/6 mL/3 mL), saturated LiOHaq (0.5 mL) was added and the mixture stirred at rt overnight. After complete conversion the reaction mixture was acidified with NaHSO<sub>4</sub> (1 M, 5 mL) and extracted with diethyl ether (3 x 20 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub> and all volatiles were removed. Flash chromatography (hexanes/ethyl acetate/AcOH = 85:10:5, silica) of the residue gave product 14 (139 mg, 0.53 mmol 53%) as a colorless oil.

<sup>1</sup>**H NMR** (500 MHz, CDCl<sub>3</sub>):  $\delta = 6.89$  (br s, 1 H, H-5), 6.61 (br s, 1 H, H-3), 5.34–5.27 (m, 1 H, H-11), 5.14–5.07 (m, 1 H, H-7), 3.86 (s, 3 H, H-17), 3.14 (d, J = 7.2 Hz, 2 H, H-6), 2.14-2.00 (m, 4 H, H-9, H-10), 1.69 (s, 3 H, H-13), 1.67 (s, 3 H, H-15), 1.61 (s, 3 H, H-14) ppm; <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta = 165.6$  (C-16), 135.9 (C-8), 131.6 (C-12), 129.0 (C-3), 124.4 (C-7), 123.7 (C-2), 123.1 (C-11), 121.2 (C-4), 119.3 (C-5), 39.8 (C-9), 37.0 (C-17), 26.8 (C-10), 25.9 (C-13), 25.3 (C-6), 17.9 (C-14), 16.2 (C-15) ppm; IR (ATR):  $\tilde{v} = 2965, 2914, 2853, 1657, 1448, 1270, 1137, 1104, 1056, 832, 727, 613, 561 \text{ cm}^{-1}; \text{HRMS}$ (ESI-TOF)  $m/_{Z}$ : [M+H]<sup>+</sup> calcd for C<sub>16</sub>H<sub>24</sub>NO<sub>2</sub> 262.1802; found: 262.1804.



Pd/C (10%, 42 mg, 0.025 mmol, 0.1 equiv.) was added  $7 = 9 = 10^{-11} = 13^{-12}$  to a stirred solution of pyrrolostatin (3) (62 mg, 0.25 mmol, 1.00 equiv.) in methanol at rt The solution was stirred under a hydrogen atmosphere (1 bar) for

3 h. The reaction mixture was filtered over  $Na_2SO_4$  and all volatiles were removed. The pure product 15 (65 mg, 99%) was isolated as a colorless amorphous solid.

**mp:** 132 °C; <sup>1</sup>**H NMR** (500 MHz, CDCl<sub>3</sub>):  $\delta = 8.98$  (br s, 1 H, N-H), 6.89 (s, 1 H, H-5), 6.80 (s, 1 H, H-3), 2.60–2.38 (m, 2 H, H-6), 1.67–1.08 (m, 10 H, H-7 to H-12), 0.91 (s, 3 H, H-13/H-14/H-15), 0.87 (s, 3 H, H-13/H-14/H-15), 0.86 (s, 3 H, H-13/H-14/H-15) ppm; <sup>13</sup>C **NMR** (125 MHz, CDCl<sub>3</sub>):  $\delta = 165.4$  (C-16), 127.7 (C-4), 122.0 (C-2), 121.4 (C-5), 117.0 (C-3), 39.5, 38.4, 37.3, 32.5, 28.1, 24.9, 24.3, 22.9, 22.8, 19.7 (C-6 to C-15) ppm; **IR** (ATR):  $\tilde{\nu} = 3365, 3075, 2954, 2922, 1681, 1488, 1439, 1343, 1279, 1207, 1129, 1113, 965, 883, 835, 818, 764, 681, 632, 610, 574 cm<sup>-1</sup>;$ **HRMS**(ESI-TOF) <math>m/z: [M+H]<sup>+</sup> calcd for C<sub>15</sub>H<sub>26</sub>NO<sub>2</sub> 252.1958; found: 252.1962.



tetrahydrofuran (50 mL) was cooled to -78 °C. *Tert*-butyllithium (1.9 M in heptane, 10.5 mL, 20.0 mmol, 2.00 equiv.) was added dropwise and the yellowish solution was stirred for 5 min. Freshly distilled *trans,trans*-geranyl bromide (2.61 g, 12.0 mmol, 1.20 equiv.) in dry tetrahydrofuran (5 mL) was added dropwise over 5 min and the mixture was stirred for 4 h with warming to

-50 °C. The reaction was quenched with brine (50 mL), the aqueous layer was extracted with ethyl acetate (3 x 50 mL), the combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub> and the solvents were removed under reduced pressure. The residue was dissolved in tetrahydrofuran (50 mL) and tetra-*n*-butylammonium fluoride trihydrate (3.79 g, 12.0 mmol, 1.20 equiv.) was added at rt. The reaction was stirred for 30 min at rt and the solvent was removed under reduced pressure. Flash chromatography (hexanes/ethyl acetate/Et<sub>3</sub>N = 85:10:5, silica) of the residue gave the sensitive 3-geranylated pyrrole. Immediately after the purification of geranylpyrrole it was dissolved in nitroethane (30 mL) and cooled to 0 °C. In a separate flask HNO<sub>3</sub> (>90%, 417  $\mu$ L, 10.0 mmol, 1.00 equiv.) was added to cooled acetic anhydride (5 mL) and stirred for 5 min. The nitronium acetate solution was added dropwise to the nitroethane solution of geranylpyrrole at 0 °C and stirred for 1 h. After complete conversion toluene (100 mL) was added and all volatiles were removed under reduced pressure. Flash chromatography (hexanes/ethyl acetate = 95:5, silica) of the raw product gave 2-nitro-4-geranylpyrrole (16)

(247 mg, 1.00 mmol, 10 %) and 2-nitro-3-geranylpyrrole (17) (294 mg, 1.18 mmol, 12 \%) as yellowish oils.

Analytical data for 2-nitro-4-geranyl pyrrole (16): <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 9.30$  (br s, 1 H, N-H), 6.98–6.93 (m, 1 H, H-5), 6.78–6.71 (m, 1 H, H-3), 5.34–5.23 (m, 1 H, H-11), 5.14–5.04 (m, 1 H, H-7), 3.18 (d, J = 7.2 Hz, 2 H, H-6), 2.18–2.00 (m, 4 H, H-9 to H-10), 1.69 (s, 3 H, H-15), 1.67 (s, 3 H, H-13), 1.61 (s, 3 H, H-14) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 164.5$  (C-16), 137.1 (C-8), 131.6 (C-12), 127.1 (C-2), 124.1 (C-7), 122.3 (C-4), 121.5 (C-11), 121.0 (C-5), 110.8 (C-3), 39.6 (C-9), 26.5 (C-10), 25.7 (C-6), 25.3 (C-13), 17.7 (C-14), 16.0 (C-15) ppm; <sup>3</sup> IR (ATR):  $\tilde{\nu} = 3314$ , 2915, 2854, 1682, 1480, 1437, 1395, 1204, 1104, 998, 971, 836, 768 cm<sup>-1</sup>; HRMS (ESI-TOF) m/z: [M+H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>21</sub>N<sub>2</sub>O<sub>2</sub> 249.1598; found: 249.1597.

Analytical data for 2-nitro-3-geranyl pyrrole (17): <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta = 9.45$  (br s, 1 H, N-H), 6.86 (t, J = 3.0 Hz, 1 H, H-2), 6.23–6.16 (m, 1 H, H-3), 5.39–5.29 (m, 1 H, H-11), 5.15–5.05 (m, 1 H, H-7), 3.60 (d, J = 7.3 Hz, 2 H, H-6), 2.17–2.00 (m, 4 H, H-9, H-10), 1.69 (s, 6 H, H-13, H-15), 1.60 (s, 3 H, H-14) ppm; <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta = 137.8$  (C-8), 131.7 (C-12), 129.6 (C-2), 124.3 (C-4), 122.1 (C-11), 120.4 (C-7), 112.5 (C-3), 39.8 (C-9), 26.7 (C-10), 25.9 (2 C, C-6 and C-13), 17.8 (C-14), 16.3 (C-15) ppm;<sup>3</sup> IR (ATR):  $\tilde{v} = 3325, 2917, 2859, 1685, 1483$  1429, 1200, 1115, 993, 968 cm<sup>-1</sup>; HRMS (ESI-TOF)  $\frac{m}{z}$ :

 $[M+H]^+$  calcd for  $C_{14}H_{21}N_2O_2$  249.1598; found: 249.1595.



Compound **18** was synthesized according to the procedure described for the synthesis of pyrrolostatin (**3**) from 3-bromo-1(triisopropylsilyl)pyrrole (3.66 g, 12.1 mmol, 1.00 equiv.) and 3,3-

 $\stackrel{\text{II}}{\text{O}}$   $\stackrel{\text{II}}{\text{H}}$  dimethylallyl bromide (1.67 mL, 14.5 mmol, 1.20 equiv.). Flash chromatography (hexanes/ethyl acetate/AcOH = 85:10:5, silica) gave the product along with its regioisomer as an off-white solid. Precipitation from dichloromethane with hexane (1:50, 20 mL) gave compound **18** (915 mg, 5.11 mmol, 42%) as a colorless amorphous powder.

**mp:** 180 °C (decomposition); <sup>1</sup>**H NMR** (600 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 12.05 (br s, 1 H, COOH), 11.38 (br s, 1 H, NH), 6.69 (s, 1 H, H-5), 6.51 (s, 1 H, H-3), 5.30–5.22 (m, 1 H, H-7), 3.08 (d, *J* = 4.2 Hz, 2 H, H-6), 1.68 (s, 3 H, H-9/H-10), 1.63 (s, 3 H, H-9/H-10) ppm; <sup>13</sup>C NMR (150 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 161.8 (C-11), 130.6 (C-8), 123.9 (C-2), 123.8 (C-7),

122.5 (C-5), 120.9 (C-4), 114.3 (C-3), 25.4 (C-6), 25.0 (C-9/C-10), 17.5 (C-9/C-10) ppm; IR (ATR):  $\tilde{v} = 3356, 2968, 2920, 1683, 1485, 1438, 1350, 1235, 1115, 967, 857, 816, 763, 689,$ 612 cm<sup>-1</sup>; **HRMS** (ESI-TOF) m/z: [M+H]<sup>+</sup> calcd for C<sub>10</sub>H<sub>14</sub>NO<sub>2</sub> 180.1019; found: 180.1018.



mL, 23.2 mmol, 1.05 equiv.) was added

dropwise. After 2 h reaction time all volatiles were removed and the residue was chromatographically purified (hexanes/ethyl acetate = 90:10, silica). The sensible product was dissolved in tetrahydrofuran (10 mL), NaOH (2 M, 60 mL) was added and the mixture was heated at 75 °C for 2 h. The solution was acidified with NaHSO<sub>4</sub> (1 M, 150 mL) and extracted with ethyl acetate (3 x 50 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub>, all volatiles were removed and the residue was purified by column chromatography (hexanes/ethyl acetate/AcOH = 90:5:5, silica). The regioisomerically pure product **19** (3.13 g, 9.94 mmol, 45 %) was isolated as a colorless amorphous powder after precipitation from dichloromethane with hexane (1:50, 40 mL).

**mp:** 101-102 °C; <sup>1</sup>**H** NMR (600 MHz, CDCl<sub>3</sub>):  $\delta = 9.04$  (br s, 1 H, N-H), 6.88 (br s, 1 H, H-3), 6.78 (br s, 1 H, H-5), 5.33 (t, J = 7.2 Hz, 1 H, H-7), 5.14–5.09 (m, 2 H, H-11/H-15), 3.20 (t, J = 7.2 Hz, 2 H, H-6), 2.14–1.98 (m, 8 H, H-9, H-10, H-13, H-14), 1.69 (s, 3 H, H-20), 1.68 (s, 3 H, H-17), 1.604 (s, 3 H, H-19), 1.598 (s, 3 H, H-18) ppm; <sup>13</sup>C NMR (150 MHz,  $CDCl_3$ ):  $\delta = 165.8$  (C-21), 136.1 (C-8), 135.3 (C-12), 131.5 (C-16), 126.6 (C-4), 124.5 (C-11/C-15), 124.3 (C-11/C-15), 123.0 (C-7), 122.1 (C-5), 121.6 (C-2), 117.1 (C-3), 39.9 (C-9/C-13), 39.8 (C-9/C-13), 26.9 (C-10/C-14), 26.7 (C-10/C-14), 25.8 (C-17), 25.5 (C-6), 17.8 (C-18), 16.2 (2 C, C-19/C-20) ppm; **IR** (ATR):  $\tilde{v} = 3347, 2967, 2913, 2854, 1678, 1487,$ 1442, 1381, 1348, 1289, 1237, 1115, 966, 887, 873, 837, 772, 762, 699, 631 cm<sup>-1</sup>; HRMS (ESI-TOF)  $m/_{Z}$ : [M+H]<sup>+</sup> calcd for C<sub>20</sub>H<sub>30</sub>NO<sub>2</sub> 316.2271; found: 316.2275.



Iodide 20 (251 mg, 1.00 mmol, 1.00 equiv.) was dissolved in dry and degassed dimethylformamide (5 mL). Successively, 1-octin (295 µL, 2.00 mmol, 2.00 equiv.), triethylamine (557 µL, 4.00 mmol, 4.00 equiv.), copper(I) iodide (10 mg, 0.05 mmol, 0.05 equiv.) and Pd(PPh<sub>3</sub>)<sub>4</sub> (58 mg, 0.05 mmol, 0.05 equiv.) were added and the reaction mixture was stirred at 50 °C for 24 h. Then the reaction mixture was diluted with diethyl ether (50 mL) and extracted with H<sub>2</sub>O (3 x 10 mL). The organic phase was dried over Na<sub>2</sub>SO<sub>4</sub>, all volatiles were removed under reduced pressure and the residue was purified by flash chromatography (hexanes/ethyl acetate = 90:10, silica) to yield product 21 (191 mg, 0.82 mmol, 82%) as a colorless oil.

<sup>1</sup>**H NMR** (600 MHz, CDCl<sub>3</sub>):  $\delta = 9.10$  (br s, 1 H, N-H), 7.04–7.03 (m, 1 H, H-5), 6.90–6.89 (m, 1 H, H-3), 3.84 (s, 3 H, H-15), 2.36 (t, J = 7.2 Hz, 2 H, H-8), 1.59–1.55 (m, 2 H, H-10/H-11), 1.45–1.40 (m, 2 H, H-10/H-11), 1.36–1.27 (m, 4 H, H-9, H-12), 0.90 (t, J = 6.9, 3 H, H-13) ppm; <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta = 161.3$  (C-14), 125.8 (C-5), 122.4 (C-2), 118.1 (C-3), 107.7 (C-4), 89.3 (C-7), 73.9 (C-6), 51.8 (C-15), 31.5 (C-12), 29.0, 28.8 (C-10 to C-11), 22.7 (C-9), 19.6 (C-8), 14.2 (C-13) ppm; **IR** (ATR):  $\tilde{v} = 3279, 2955, 2932, 2854, 1679,$ 1487, 1444, 1383, 1267, 1212, 1124, 979, 914, 846, 774, 742, 720, 635, 609 cm<sup>-1</sup>; HRMS (ESI-TOF)  $m/_{Z}$ : [M+H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>20</sub>NO<sub>2</sub> 234.1489; found: 234.1489.





Pd/C (10%, 43 mg, 0.026 mmol, 0.1 equiv.) was  $\begin{array}{c} 3 & 4 & 6 \\ 15 & 0 & 14 \\ 15 & 0 & 14 \\ 15 & 0 & 14 \\ 15 & 0 & 14 \\ 15 & 0 & 14 \\ 15 & 0 & 14 \\ 15 & 0 & 10 \\ 10 & 0 & 0 \\ 10 &$ 0.26 mmol, 1.00 equiv.) in methanol (10 mL) at rt. The solution was stirred under a hydrogen

atmosphere (1 bar) for 3 h. The reaction mixture was filtered over Na<sub>2</sub>SO<sub>4</sub> and all volatiles were removed. Flash chromatography (hexanes/ethyl acetate = 90:10, silica) of the residue gave product 22 (61 mg, 99%) as a colorless amorphous solid.

**mp:** 56 °C; <sup>1</sup>**H NMR** (500 MHz, CDCl<sub>3</sub>):  $\delta$  = 8.90 (br s, 1 H, N-H), 6.77–6.71 (m, 2 H, H-3, H-5), 3.83 (s, 3 H, H-15), 2.45 (t, J = 7.6 Hz, 2 H, H-6), 1.59–1.51 (m, 2 H, H-7), 1.36-1.22 (m, 10 H, H-8 to H-12), 0.88 (t, J = 7.0 Hz, 3 H, H-13) ppm; <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta = 161.7$  (C-14), 127.0 (C-4), 122.3 (C-2), 120.7 (C-5), 115.1 (C-3), 51.5 (C-15), 32.0, 31.1, 29.6, 29.39, 29.37, 26.8 (C-6 to C-11), 22.8 (C-12), 14.3 (C-13) ppm; IR (ATR):  $\tilde{v} = 3290, 2954, 2918, 2850, 1688, 1441, 1393, 1214, 1131, 1110, 993, 971, 769, 603 \text{ cm}^{-1};$ **HRMS** (ESI-TOF)  $\frac{m}{Z}$ : [M+H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>24</sub>NO<sub>2</sub> 238.1802; found: 238.1809.

$$HO_{14} \xrightarrow{2}_{O} \xrightarrow{N}_{H} \xrightarrow{2} 23$$

$$Methyl ester 22 (795 mg, 3.35 mmol, 1.00 equiv.) was dissolved in a mixture of tetrahydrofuran/methanol/H2O (9 mL/6 mL/3 mL) and tetra$$

KOH (940 mg, 16.8 mmol, 5.00 equiv.) was added. The resulting mixture was stirred for 20 h at rt. After complete conversion the reaction mixture was acidified with NaHSO<sub>4</sub> (1 M, 20 mL) and extracted with diethyl ether (3 x 15 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub> and all volatiles were removed. Flash chromatography (hexanes/ethyl acetate/AcOH = 88:10:2 in hexanes, silica) of the residue gave product **23** (572 mg, 76%) as a colorless amorphous powder.

**mp:** 153 °C; <sup>1</sup>**H NMR** (400 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 12.02 (br s, 1 H, OH), 11.35 (br s, 1 H, NH), 6.72 (m, 1 H, H-5), 6.54 (m, 1 H, H-3), 2.36 (t, *J* = 7.5 Hz, 2 H, H-6), 1.54–1.42 (m, 2 H, H-7), 1.36–1.15 (m, 10 H, H-8 to H-12), 0.85 (t, *J* = 6.8 Hz, 3 H, H-13) ppm; <sup>13</sup>**C NMR** (100 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 161.8 (C-14), 124.7 (C-4), 122.3 (C-2), 121.0 (C-5), 114.3 (C-3), 31.3, 30.6, 28.8, 28.7 (2 C), 26.2 (C-6 to C-11), 22.1 (C-12), 13.9 (C-13) ppm; **IR** (ATR):  $\tilde{\nu}$  = 3366, 2952, 2918, 2849, 1665, 1574, 1486, 1435, 1341, 1277, 1216, 1129, 1114, 961, 883, 837, 817, 764, 593 cm<sup>-1</sup>; **HRMS** (ESI-TOF) m/z: [M+H]<sup>+</sup> calcd for C<sub>13</sub>H<sub>22</sub>NO<sub>2</sub> 224.1645; found: 224.1647.

Literature:

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- 2 Y. Fumoto, T. Eguchi, H. Uno and N. Ono, J. Org. Chem., 1999, 64, 6518.
- 3 The carbon-atoms attached to the nitro function were not detectable due to low intensity.
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![](_page_19_Figure_0.jpeg)

![](_page_20_Figure_0.jpeg)

![](_page_21_Figure_0.jpeg)

![](_page_22_Figure_0.jpeg)

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