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Practical Isolation of Polygodial from *Tasmannia lanceolata*: a Viable Scaffold for Synthesis

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Supporting Information

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I. Supplementary Figures and Tables



Figure S1. PHWE extract (left) and isolated polygodial (right).



Figure S2. Maceration extract (left) and polygodial isolated after chromatography from maceration (right) (Table 1, entry 7).

 $EtOH/H_2O(\% v/v)$ Yield, $\% w/w^{[a]}$ 15 25 35 0.11 (0.12) 70 0.19(0.20)0.42 (0.45) Temp. (°C) 90 0.15 (0.17) 0.29 (0.31) 0.46 (0.49) 0.13 (0.17) 0.42 (0.46) 110 0.30(0.35)130 0.36(0.46)0.15(0.24)0.18(0.25)

Table S1. Effect of temperature and solvent composition on polygodial yield (500 psi).

[a] Value in parentheses refers to the combined yield of polygodial and epi-polygodial.

Table S2. Effect of tem	perature and solvent	composition on	polygodial vield	(1500 psi).
	peracare and sorreite	composition on	por, Sound , iere	

Yield, $\% \text{ w/w}^{[a]}$		EtOH/H ₂ O (% v/v)			
		0	15	25	35
	70	-	0.10 (0.11)	0.18 (0.19)	0.43 (0.46)
(°C)	90	-	0.14 (0.15)	0.28 (0.30)	0.46 (0.49)
ıp. (110	-	0.12 (0.15)	0.25 (0.29)	0.46 (0.51)
lem	130	-	0.15 (0.23)	0.27 (0.37)	0.21 (0.26)
L .	150	0.04 (0.08)	-	-	-

[a] Value in parentheses refers to the combined yield of polygodial and epi-polygodial.



Figure S3. Effect of temperature and solvent composition on polygodial yield. Experiments were performed using a Dionex ASE200 at 1500 psi (~35 bar). Yields were determined *via* ¹H NMR spectroscopy with the aid of an internal standard.



polygodial CDCl3



polygodial 13C



drimendiol CDCl3



ОН

 13	7	•	0

-127.1

	67	.3
	61	.2
	54	.5
	49	.5
\nearrow	42	.1
\sim	39	.4
\sim	35	6.6
$\overline{}$	33	3.3
\sim	33	8.0
~	23	6
/	2 -	•••



drimenol 1H CDC13

S--8









Polygodial Wittig derivative (6) ¹H-¹³C HSQCme NMR Spectrum





Polygodial Diels–Alder derivative (7) ¹³C NMR Spectrum



Polygodial Diels–Alder derivative (7) ¹H-¹³C HSQCme NMR Spectrum











euryfuran 13C CDCl3

































