Supporting Information

Catalytic chemodosimeteric approach for detection of nanomolar cyanide ions in water, blood serum and live cell imaging

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Figure SI 1. 1 H and 13 C NMR spectra of compound 1



Figure SI 2. ¹H and ¹³C NMR spectra of chemodosimeter CD-1



Figure SI 3. 1 H and 13 C NMR spectra of chemodosimeter CD-2



Figure SI 4. ¹H and ¹³C NMR spectra of compound 4



Figure SI 5. 1 H and 13 C NMR spectra of compound 5



HRMS spectra of 1, CD-1, CD-2, 4 and 5

Figure SI 6. HRMS spectrum of 1



Figure SI 7. HRMS spectrum of CD-1



Figure SI 8. HRMS spectrum of CD-2



Figure SI 9. HRMS spectrum of compound 4



Figure SI 10. HRMS spectrum of compound 5



Figure SI 11. (a) Effect of various anions (100 μ M) on absorption spectrum of chemodosimeter CD-1 (5 μ M), (b) Effect of various anions (100 μ M) on absorption spectrum of chemodosimeter CD-2 (10 μ M) in HEPES buffer - 5% DMSO (pH 7.4).



Figure SI 12. Effect of various anions (100 μ M) on emission spectra (λ_{ex} 300 nm) of chemodosimeter **CD-2** (10 μ M) in HEPES buffer - 5% DMSO (pH 7.4).



Figure SI 13. Bar diagrams showing the effect of various anions on the emission intensities (λ_{ex} 320 nm) of (a) CD-1 (5 μ M) and (b) CD-2 (10 μ M) in the presence of CN⁻ and ten thousand times concentration of other anions (100 μ M), A (CD+CN⁻), B (CD + AcO⁻ + CN⁻), C (CD + F⁻ + CN⁻), D (CD + Cl⁻ + CN⁻), E (CD + Br⁻ + CN⁻), F (CD + I⁻ + CN⁻), G (CD + ClO₄⁻ + CN⁻), H (CD + H₂PO₄⁻ + CN⁻), I (CD + HSO₄⁻ + CN⁻), J (CD + SO₄²⁻ + CN⁻), K (CD + NO₃⁻ + CN⁻), L (CD + SCN⁻ + CN⁻) and M (CD + OH⁻ + CN⁻) in HEPES buffer - 5% DMSO solution (pH 7.4).

CCDC Numbers

CD-1 CCDC 1042262 Formula: C32 H22 N3 O2 1+,F6 P1 1-,2(C2 H3 N1) Unit Cell Parameters: a 9.895(3) b 11.505(5) c 14.869(4) P-1

Compound 4 : CCDC 1042263 Formula: C32 H23 N3 O3 Unit Cell Parameters: a 9.816(4) b 11.381(4) c 13.411(6) P-1



Figure SI 14. Shows lone pair... π interactions of acetonitrile molecules with anthracene and imidazole rings in **CD-1**(a) stick diagram (b) space filled representation.



Figure SI 15. The formation of a pseudo-square cavity bound by two centrosymmetric molecules (shown in red and green), the solvent (blue and grey space filled) and anions (yellow and orange balls) held by various non-covalent interactions. $\pi \dots \pi$ interactions between two anthracene rings of two such consecutive dimers form linear chains.



Figure SI 16. Shows the formation of linear H-bonded tapes in compound 4



Figure SI 17. DLS (a, b), SEM (c, d) and TEM (e, f) studies of CD-2 (5 μ M, H₂O – DMSO 5%) in the absence (a, c, e) and presence (b, d, f) of CN⁻



Figure SI 18. Blue diamonds show increase in FI of CD-1 (5μ M) + NaCN (5 nM) solution with time and red circles show no change in FI of CD-1 alone during this time period.



Figure SI 19. Comparison of emission spectra of (a) CD-1 (5 μ M) + CN⁻ solution with formamide 4 and (b) CD-2 (10 μ M) + CN⁻ solution with formamide 5.



Bio-imaging of cyanide ions in live C6 cells

Figure. **SI 20**. Cell viability of C6 glioma cells tested by MTT assay after exposure to 5 μ M, 10 μ M, 20 μ M and 50 μ M concentrations of **CD-1** and **CD-2**, and also NaCN (10 μ M and 50 μ M) for 30 min. C6 cells treated with **CD-1** indicated nearly 86% cell viability while with **CD-2** treatment > 96% C6 cells were viable.



Figure SI 21. Respective images of C6 glioma cells are as brightfield image and under blue and green windows (a) image of untreated C6 glioma cells, (b) fluorescence image of C6 glioma cells treated with chemodosimeter **CD-1**, (c) images of C6 glioma cells incubated with **CD-1** and then with CN⁻ (10 nM) (d) images of C6 glioma cells incubated with **CD-1** and then with CN⁻ (10 nM)



Figure SI 22. Respective images of C6 glioma cells are as brightfield image and under blue and green windows (a) image of untreated C6 glioma cells, (b) fluorescence image of C6 glioma cells treated with chemodosimeter **CD-2**, (c) images of C6 glioma cells incubated with **CD-2** and then with CN⁻ (10 nM) (d) images of C6 glioma cells incubated with **CD-2** and then with CN⁻ (10 nM).

S.No.	Reference	species	TON/rate
	Present work (first one for anion)	CN-	70-360/12.64S ⁻¹
1	M. Pal, K. Parasuraman and K. R. Yeleswarapu, <i>Org. Lett.</i> 2003, 5 , 349.	Pd ^o	NR
2	F. Song, A. L. Garner and K. Koide J. Am. Chem. Soc. 2007, 129 , 12354.	Pd ^o	NR
3	J. Kovács and A. Mokhir, Inorg. Chem. 2008, 47, 1880;	Cu ²⁺	55
4	D. N. Lee, G. J. Kim and H. J. Kim, <i>Tetrahedron Lett.</i> , 2009, 50 , 4766.	Hg ²⁺	2
5	A. L. Garner and K. Koide, Chem. Commun., 2009, 86.	Pd ^o , Pt ^o	/ 3.1 h ⁻¹
6	A. L. Garner and K. Koide Chem. Commun., 2009, 83.	Pt ^o	/1200 h ⁻¹
7	M. Santra, D. Ryu, A. Chatterjee, S. K. Ko, I. Shin and K. H. Ahn, <i>Chem. Commun.</i> , 2009, 2115.	Hg ²⁺ , CH ₃ HgCl	2
8	M. H. Kim, H. H. Jang, S. Yi, S. K. Chang and M. S. Han, <i>Chem. Commun.</i> , 2009, 4838.	Cu ²⁺	NR
9	Q. L. Wang, H. Zhang and Y. B. Jiang, <i>Tetrahedron Lett.</i> , 2009, 50 , 29.	Cu ²⁺	20
10	A. L. Garner, F. Song and K. Koide J. Am. Chem. Soc. 2009, 131 , 5163.	Pd ^o	/1.8 h ⁻¹
11	M. Santra, S. K. Ko, I. Shin and K. H. Ahn, <i>Chem. Commun.</i> , 2010, 46 , 3964.	Pd°, Pd ²⁺	NR
12	J. H. Do, H. N. Kim, J. Yoon, J. S. Kim, and H. J. Kim, Org. Lett., 2010, 12 , 932.	Au ³⁺	/3.31x10 ⁻⁵ S ⁻¹ (0.12h ⁻¹)
13	M. E. Jun and K. H. Ahn, Org. Lett., 2010, 12, 2790.	Pdo	NR
14	Z. Zhou, N. Li and A. Tong, Anal. Chim. Acta., 2011, 702, 81.	Cu ²⁺	NR

Table: List of catalytic chemodosimeters

NR = not reported