Supplementary Information Section:

Optimizing the Lifetimes of Phenoxonium Cations Derived from Vitamin E *via* Structural Modifications

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1. Synthesis of Modified Forms of Vitamin E

1.1. Experimental Section

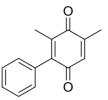
NMR spectra were recorded on Bruker AVANCE 400 NMR instrument and were calibrated using residual non deuterated solvent as an internal reference (CHCl₃ @ 7.26 ppm ¹H-NMR, 77.0 ppm ¹³C-NMR). The following abbreviations were used to explain the multiplicities: s = singlet, brs = broad singlet, d = doublet, t = triplet, dd = doublet of doublets, m = multiplet. Melting points were uncorrected. High-resolution mass spectra (HRMS) were obtained with JEOL MS-700P mass spectrometer, Finnigan MAT 95 XP mass spectrometer (Thermo Electron Corporation), and Q-Tof Premier. Flash column chromatography was performed using Merck silica gel 60 with distilled hexane, and preparative thin-layer chromatography (PTLC) was bought from CEMS MARKETING PTE LTD, PTLC plate (20 cm × 20 cm) was coated with 10 µm, 60 Å particles, 500 µm coating thickness.

1.2. Compound Characterization and Synthetic Steps

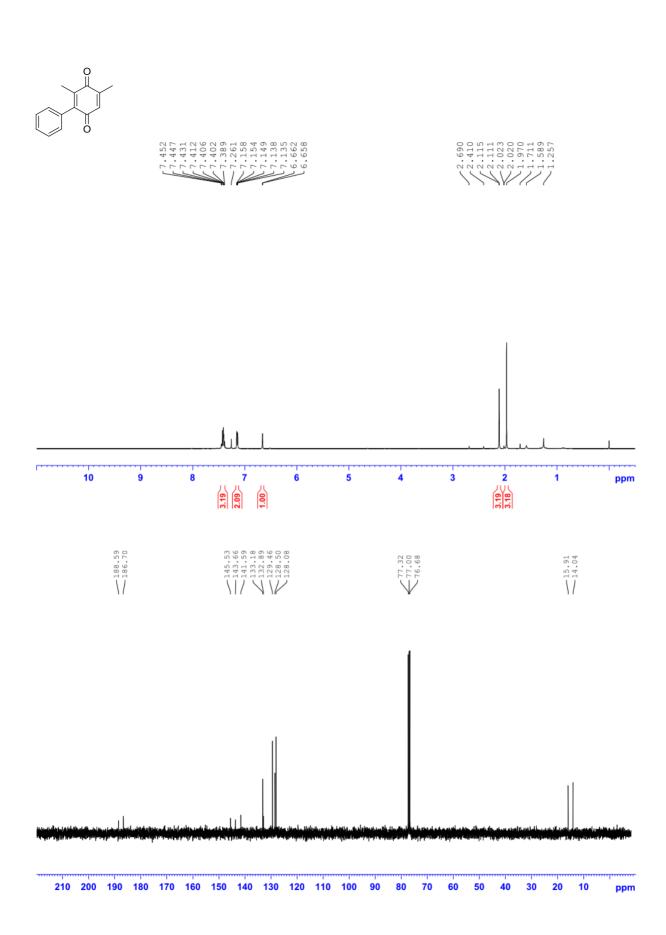
Arylation of 2,6-dimethylcyclohexa-2,5-diene-1,4-dione.^{S1}

To a solution of 2,6-dimethylclohexa-2,5-diene-1,4-dione (4.0 mmol) in trifluorotoluene (20 ml) was added the corresponding boronic acid (6.0 mmol), water (20 ml), silver nitrate (0.8 mmol), and potassium persulfate (12.0 mmol). The solution was then stirred vigorously at room temperature and monitored by thin-layer chromatography of the organic layer. Upon consumption of quinone (6–24 h), the reaction was diluted with dichloromethane (48 ml) and washed with 5% sodium bicarbonate. The layers were separated, and the aqueous layer was extracted with dichloromethane (3×64 ml), dried over MgSO₄, and evaporated *in vacuo*. Purification was performed by silica gel chromatography to give the arylation product in 60–85% yield. (The reaction needed 48 hours to complete because trimethylphenyl boronic acid is a very bulky acid).

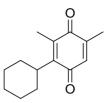
3,5-dimethyl-2-phenylcyclohexa-2,5-diene-1,4-dione



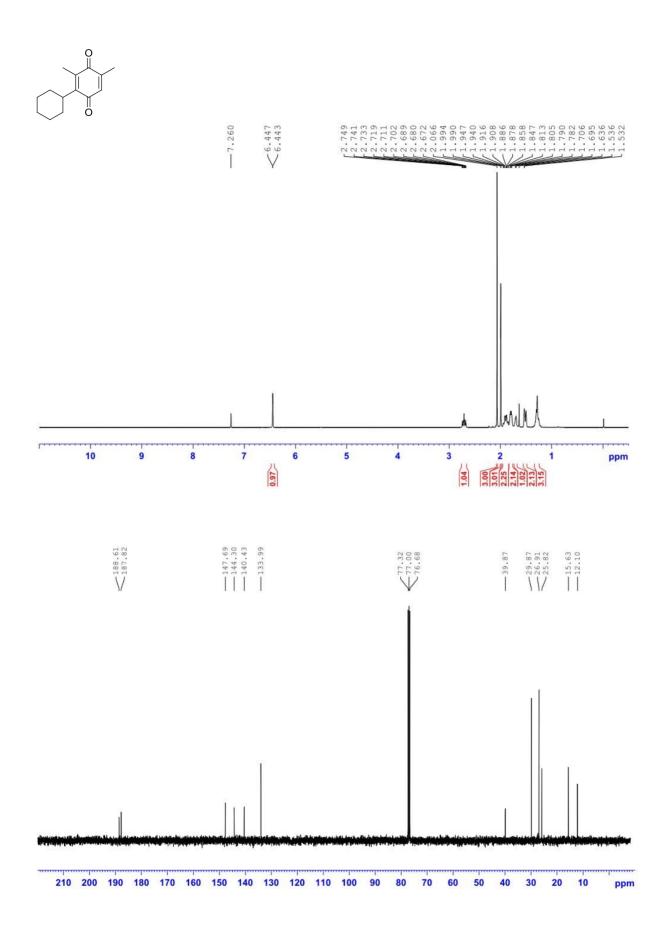
79%; Yellow solid; ¹H NMR (400 MHz, CDCl₃) δ : 1.97 (s, 6H), 2.11 (d, 3H, *J* = 1.6 Hz), 6.66 (d, 1H, *J* = 1.6 Hz), 7.13–7.16 (m, 2H), 7.39–7.45 (m, 3H); ¹³C NMR (100 MHz, CDCl₃) δ : 14.0, 15.9, 128.1, 128.5, 129.5, 132.9, 133.2, 141.6, 143.7, 145.5, 186.7, 188.6; m.p. 66–69 °C (EtOAc); ESI-HRMS: *m/z* calcd for C₁₄H₂₃O₂ [M+H]⁺: 223.1698; found: 223.1691.



2-cyclohexyl-3,5-dimethylcyclohexa-2,5-diene-1,4-dione



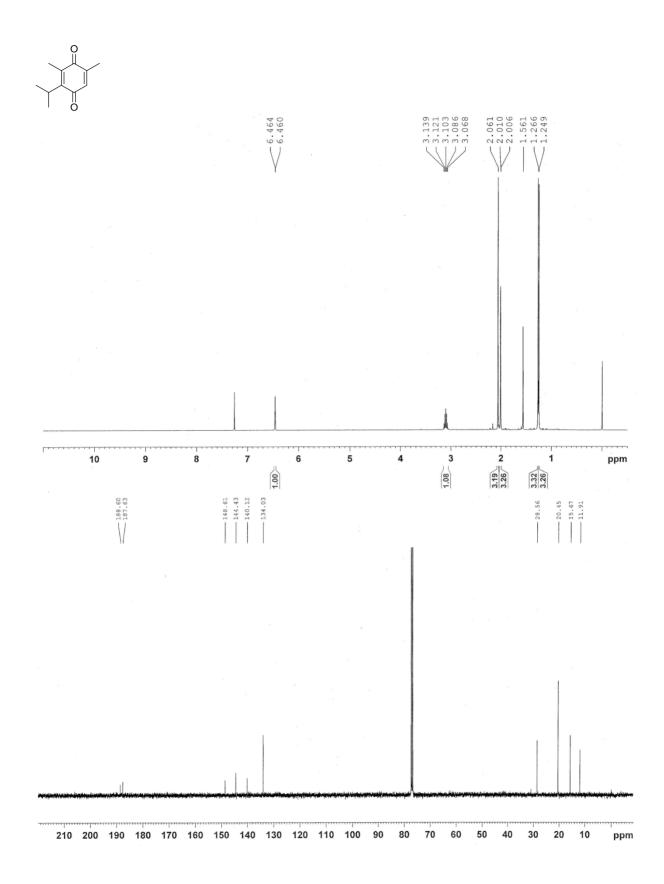
75%; Yellow liquid; ¹H NMR (400 MHz, CDCl₃) δ : 1.28–1.30 (m, 3H), 1.50–1.54 (m, 2H), 1.69–1.71 (m, 1H), 1.78–1.86 (m, 2H), 1.88–1.95 (m, 2H), 1.99 (d, 3H, J = 1.6 Hz), 2.07 (s, 3H), 2.67–2.75 (m, 1H), 6.44 (d, 1H, J = 1.6 Hz); ¹³C NMR (100 MHz, CDCl₃) δ : 12.1, 15.6, 25.8, 26.9, 29.9, 39.9, 134.0, 140.4, 144.3, 147.7, 187.8, 188.6; ESI-HRMS: m/z calcd for C₁₄H₁₉O₂ [M+H]⁺: 219.1385; found: 219.1385.



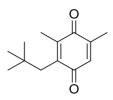
3,5-dimethyl-2-(propan-2-yl)cyclohexa-2,5-diene-1,4-dione



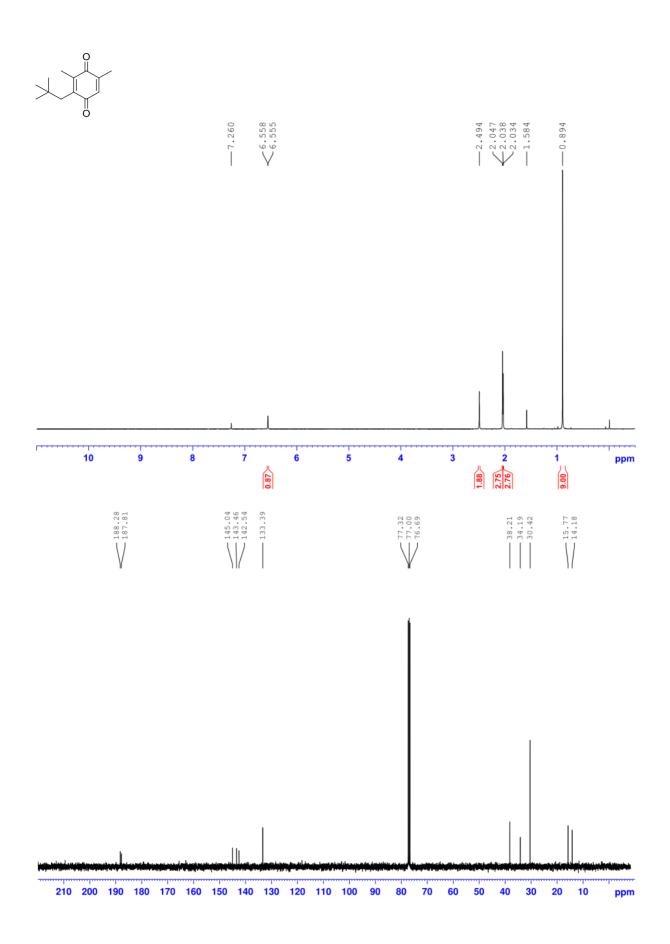
85%; Yellow liquid; ¹H NMR (400 MHz, CDCl₃) δ: 1.25 (s, 3H), 1.27 (s, 3H), 2.00 (s, 3H), 2.06 (s, 3H), 3.07–3.14 (m, 1H), 6.46 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 11.9, 15.7, 20.4, 28.6, 134.0, 140.1, 144.4, 148.6, 187.6, 188.6; ESI-HRMS: *m/z* calcd for C₁₁H₁₅O₂ [M+H]⁺: 179.1072; found: 179.1073.



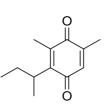
2-(2,2-dimethyl propyl)-3,5-dimethylcyclohexa-2,5-diene-1,4-dione



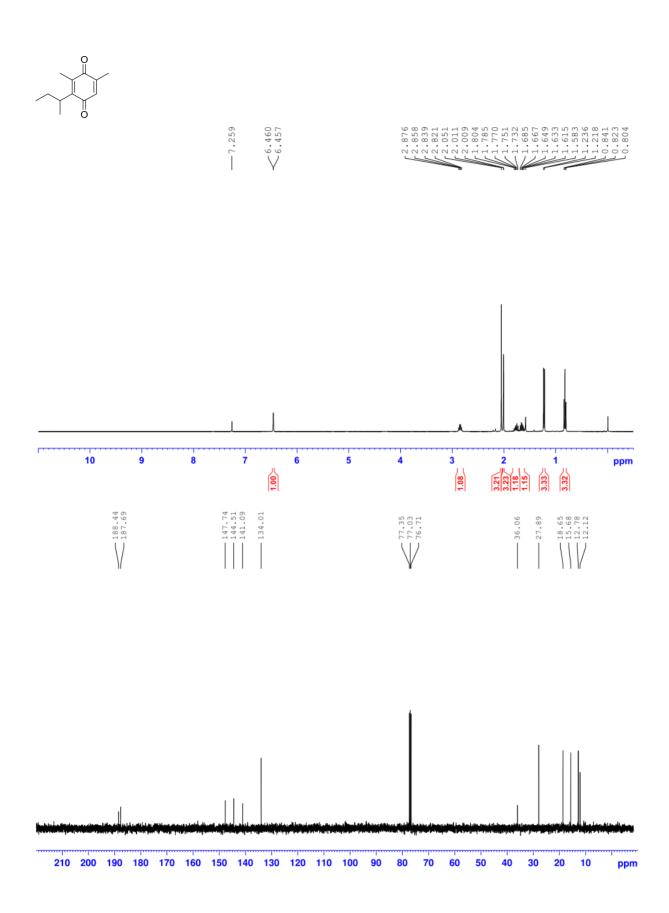
82%; Yellow solid; ¹H NMR (400 MHz, CDCl₃) δ: 0.89 (s, 9H), 2.03(s, 3H), 2.04 (s, 3H), 2.49 (s, 2H), 6.55 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 14.2, 15.8, 30.4, 34.2, 38.2, 133.4, 142.5, 143.5, 145.0, 187.8, 188.3; m.p. 54–56 °C (EtOAc); ESI-HRMS: *m*/*z* calcd for C₁₃H₁₉O₂ [M+H]⁺: 207.1385; found: 207.1382.



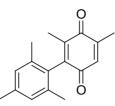
2-(butan-2-yl)-3,5-dimethylcyclohexa-2,5-diene-1,4-dione



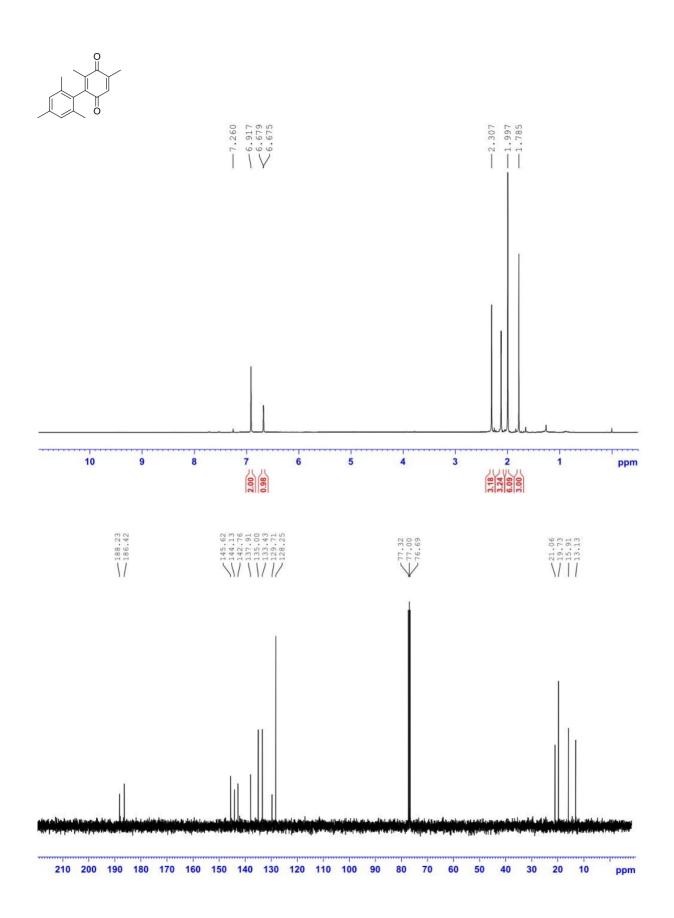
82%; Yellow liquid; ¹H NMR (400 MHz, CDCl₃) δ: 0.82 (t, 3H, *J* = 7.6 Hz), 1.23 (d, 3H, *J* = 7.2 Hz), 1.61–1.68 (m, 1H), 1.73–1.80 (m, 1H), 2.01 (d, 3H, *J* = 0.8 Hz), 2.05(s, 3H), 2.82–2.88 (m, 1H), 6.46 (d, 1H, *J* =1.2 Hz); ¹³C NMR (100 MHz, CDCl₃) δ: 12.1, 12.8, 15.7, 18.6, 27.9, 36.1, 134.0, 141.1, 144.5, 147.7, 187.7, 188.4; ESI-HRMS: *m/z* calcd for C₁₂H₁₇O₂ [M+H]⁺: 193.1229; found: 193.1229.



3,5-dimethyl-2-(2,4,6-trimethylphenyl)cyclohexa-2,5-diene-1,4-dione



60%; Yellow solid; ¹H NMR (400 MHz, CDCl₃) δ:1.78 (s, 3H), 2.00 (s, 6H), 2.12 (s, 3H), 2.31 (s, 3H), 6.67 (s,1H), 6.92 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ: 13.1, 15.9, 19.7, 21.1, 128.2, 129.7, 133.4, 135.0, 137.9, 142.8, 144.1, 145.6, 186.4, 188.2; m.p. 58–60 °C (EtOAc); ESI-HRMS: *m/z* calcd for C₁₇H₁₉O₂ [M+H]⁺: 255.1385; found: 255.1382.



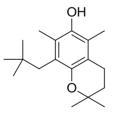
General procedure to synthesize hydroquinones from benzoquinones^{S2}

To a solution of benzoquinone (2.2 mmol) in acetic acid (6.0 ml), zinc powder (4 eq, 589 mg) was added. The mixture was then sonicated for 30 minutes. Upon consumption of quinone, the reaction mixture was filtered and the residue was washed with ethyl acetate. The combined filtrates were concentrated *in vacuo*, affording the pure reduction product in quantitative yield.

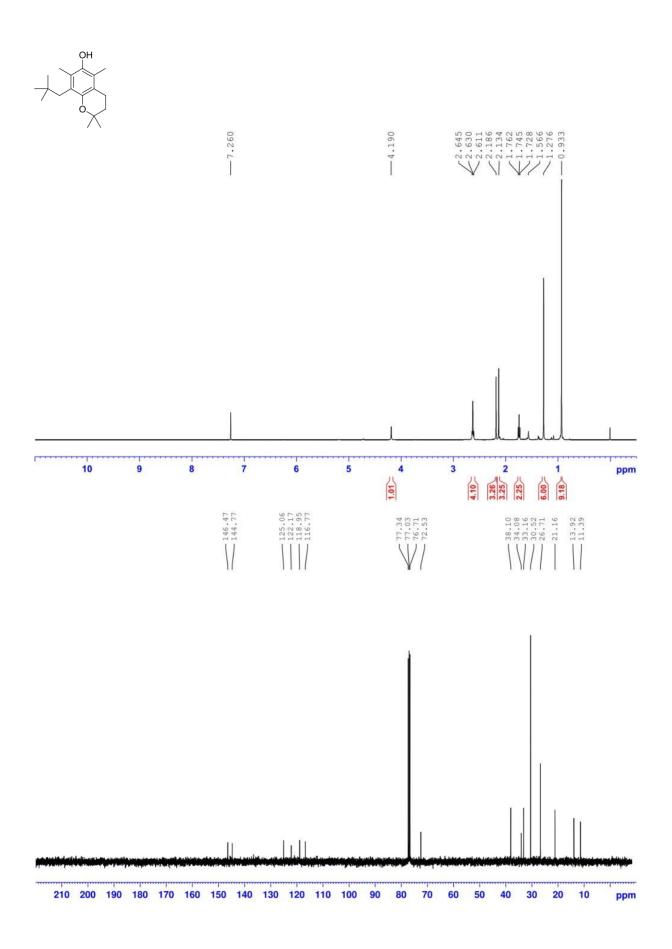
General procedure to synthesize different substituted 3,4-dihydro-2H-1-benzopyran-6-ol^{S3}

A solution of hydroquinone and 2-methylbut-3-en-2-ol in 1:1 mole ratio in minimum amount of trifluoroacetic acid was stirred at room temperature under inert atmosphere, and completion of reaction was monitored by thin-layer chromatography. The solvent was removed under reduced pressure, and the residue was dissolved in hexane. The organic layer was neutralized with saturated sodium bicarbonate, then washed with saturated sodium chloride, dried over anhydrous NaSO₄, and concentrated *in vacuo*. Purification was performed by silica gel chromatography to yield pure product in (56–72%).

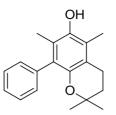
8-(2,2-dimethylpropyl)-2,2,5,7-tetramethyl-3,4-dihydro-2H-1-benzopyran-6-ol



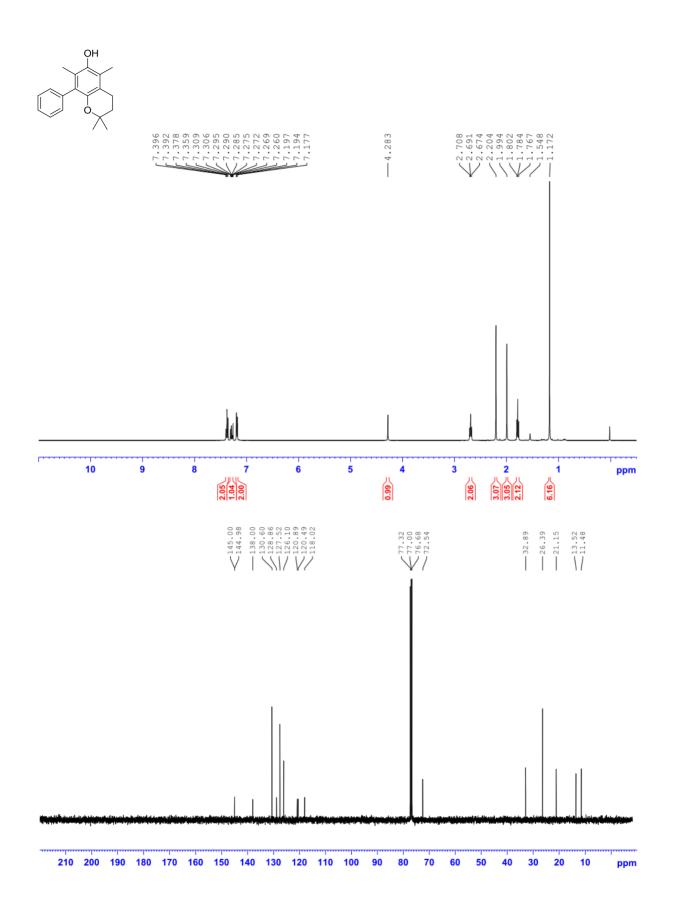
68%; White solid; ¹H NMR (400MHz, CDCl₃) δ: 0.93 (s, 9H), 1.28 (s, 6H), 1.74 (t, 2H, J = 6.8 Hz), 2.13 (s, 3H), 2.19 (s, 3H), 2.63 (s, 2H), 2.63 (t, 2H, J = 7.5 Hz), 4.19 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 11.4, 13.9, 21.2, 26.7, 30.5, 33.2, 34.1, 38.1, 72.5, 116.8, 118.9, 122.2, 125.1, 144.8, 146.5; m.p. 84–86 °C (EtOAc); ESI-HRMS: m/z calcd for C₁₈H₂₉O₂ [M+H]⁺: 277.2168; found: 277.2168.



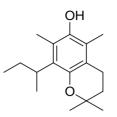
2,2,5,7-tetramethyl-8-phenyl-3,4-dihydro-2H-1-benzopyran-6-ol



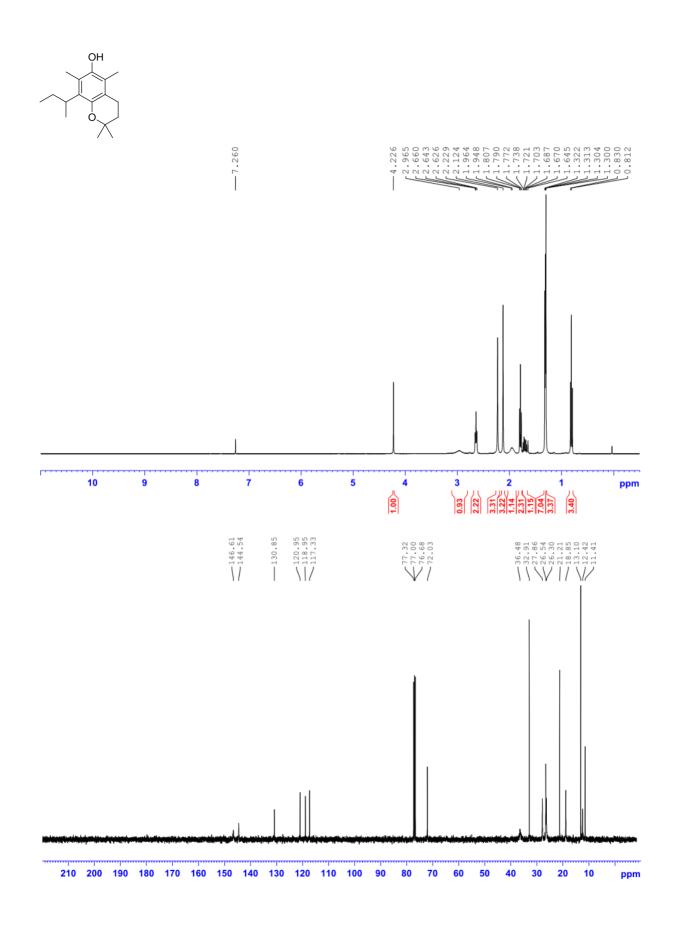
60%; White solid; ¹H NMR (400 MHz, CDCl₃) δ : 1.17 (s, 6H), 1.78 (t, 2H, *J* = 7.2 Hz), 1.99 (s, 3H), 2.20 (s, 3H), 2.69 (t, 2H, *J* = 6.8 Hz), 4.28 (s, 1H), 7.18–7.19 (d, 2H, *J* = 6.8 Hz), 7.20–7.31 (m, 1H), 7.36–7.40 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ : 11.5, 13.5, 21.2, 26.4, 32.9, 72.5, 118.0, 120.5, 120.9, 126.1, 127.5, 128.9, 130.6, 138.0, 144.9, 145.0; m.p. 88–90 °C (EtOAc); ESI-HRMS: *m*/*z* calcd for C₁₉H₂₃O₂ [M+H]⁺: 283.1698; found: 283.1696.



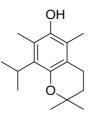
8-(butan-2-yl)-2,2,5,7-tetramethyl-3,4-dihydro-2H-1-benzopyran-6-ol



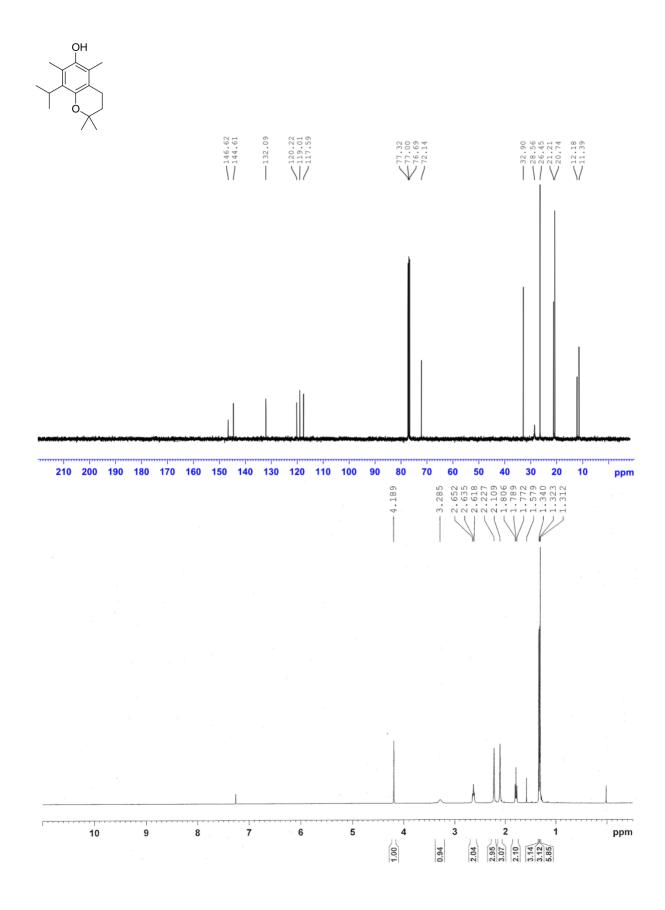
70%; White solid; ¹H NMR (400 MHz, CDCl₃) δ : 0.79 (t, 3H, *J* = 7.6 Hz), 1.28 (s, 6H), 1.29 (d, 3H, *J* = 2.4 Hz),1.64–1.71 (m, 1H), 1.77 (t, 2H, *J* = 7.2Hz), 1.93–1.94 (m, 1H), 2.10 (s, 3H), 2.21 (s, 3H), 2.62 (t, 3H, *J* = 6.8 Hz), 2.94 (s, 1H), 4.18 (s, 1H) ; ¹³C NMR (100 MHz, CDCl₃) δ : 11.4, 12.4, 13.1, 18.8, 21.2, 26.3, 26.5, 27.9, 32.9, 36.5, 72.0, 117.3, 118.9, 120.9, 130.9, 144.5, 146.6; m.p. 47–49 °C (EtOAc); ESI-HRMS: *m/z* calcd for C₁₇H₂₇O₂ [M+H]⁺: 263.2011; found: 263.2009.



2,2,5,7-tetramethyl-8-(propan-2-yl)-3,4-dihydro-2H-1-benzopyran-6-ol

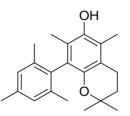


72%; White solid; ¹H NMR (400 MHz, CDCl₃) δ : 1.31 (s, 6H), 1.32 (s, 3H), 1.34 (s, 3H), 1.79 (t, 2H, J = 6.8 Hz), 2.11 (s, 3H), 2.23 (s, 3H), 2.64 (t, 2H, J = 6.8 Hz), 3.29 (s, 1H), 4.19 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ : 11.4, 12.2, 20.7, 21.2, 26.4, 28.6, 32.9, 72.1, 117.6, 119.0, 120.2, 132.1, 144.6, 146.6; m.p. 66–69 °C (EtOAc); ESI-HRMS: m/z calcd for C₁₆H₂₅O₂ [M+H]⁺: 249.1855; found: 249.1858.



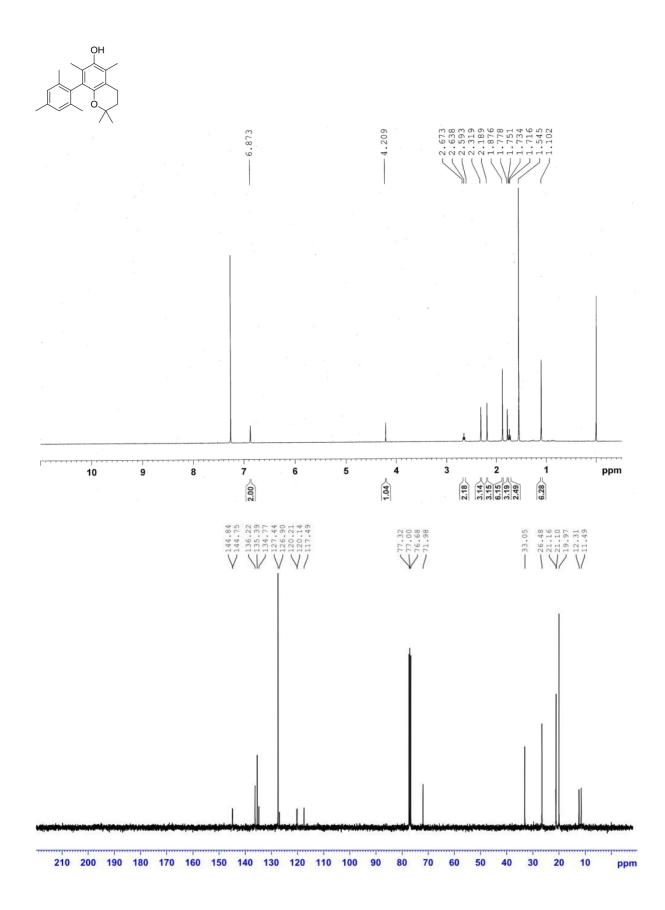
S23

2,2,5,7-tetramethyl-8-(2,4,6-trimethylphenyl)-3,4-dihydro-2H-1-benzopyran-6-ol¹⁷

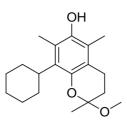


Trimethylphenyl dimethyl hydroquinone (1.2 mmol, 307 mg) and *p*-toluenesulfonic acid (TsOH·H₂O) (0.15 eq, 0.18 mmol, 34.7 mg) were added to a two-necked flask, then dry toluene (5 ml) was added under nitrogen. After stirring for five minutes, 2-methyl-but-3-en-2-ol (1.2 eq, 0.15 ml) was injected dropwise into the flask by syringe. The reaction was then stirred for 48 hours at 100 °C. After cooling, ethyl acetate was added to dilute the solution. The organic layer was washed with sodium bicarbonate and water, and dried with MgSO₄. After the solvent was removed, the residue was separated by column chromatography (hexane:EtOAc = 99:1) to give the white solid in 12% yield.

12%; White solid; ¹H NMR (400 MHz, CDCl₃) δ : 1.10 (s, 6H), 1.73 (t, 2H, J = 7.2 Hz), 1.78 (s, 3H), 1.88 (s, 6H), 2.19 (s, 3H), 2.32 (s, 3H), 2.64 (t, 2H, J = 7.21 Hz), 4.21 (s, 1H), 6.87 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ : 11.5, 12.3, 20.0, 21.1, 21.2, 26.5, 33.0, 72.0, 117.5, 120.1, 120.2, 126.9, 127.4, 134.8, 135.4, 136.2, 144.7, 144.8; m.p. 84–86 °C (EtOAc); ESI-HRMS: m/z calcd for C₂₂H₂₈O₂Na [M+Na]⁺: 347.1987; found: 347.1992.



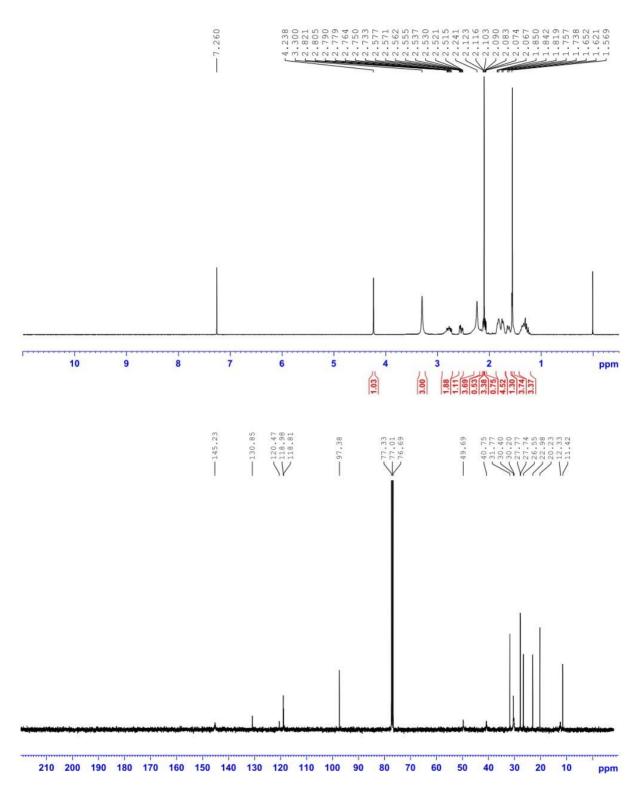
8-cyclohexyl-2-methoxy-2,5,7-trimethyl-3,4-dihydro-2H-1-benzopyran-6-ol¹⁷



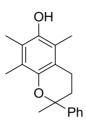
Cyclohexyldimethylhydroquinone (441 mg, 2.0 mmol) was added to a round-bottomed flask, and methanol (10 ml) was injected under nitrogen. After methyl vinyl ketone (0.32 ml, 2.0 mmol) was added, conc. sulfuric acid (2 drops) was added quickly. The solution was stirred for five minutes and trimethyl orthoformate (0.33 ml, 3 mmol) was added, then the solution was stirred for 48 hours at room temperature. After the solvent was removed, ethyl acetate and water were added. The organic layer was washed with sodium bicarbonate and brine, dried with Na₂SO₄. The residue was purified by chromatography (hexane:EtOAc = 4:1) to give the white solid product (50%).

50%; White solid; ¹H NMR (400 MHz, CDCl₃) δ: 1.28–1.37 (m, 3H), 1.56 (s, 3H), 1.56–1.57 (m, 1H), 1.62–1.65 (m, 1H), 1.74–1.85 (m, 5H), 2.07–2.12 (m, 2H), 2.10 (s, 3H), 2.24 (s, 3H), 2.52–2.58 (m, 1H), 2.73–2.82 (m, 2H), 3.30 (s, 3H), 4.24 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 11.4, 12.3, 20.2, 23.0, 26.6, 27.7, 27.8, 30.2, 30.4, 31.8, 40.7, 49.7, 97.4, 118.8, 119.0, 120.5, 130.9, 145.3; m.p. 136–138 °C (EtOAc); ESI-HRMS: *m/z* calcd for C₁₉H₂₃O₂ [M+H]⁺: 283.1698; found: 283.1696.



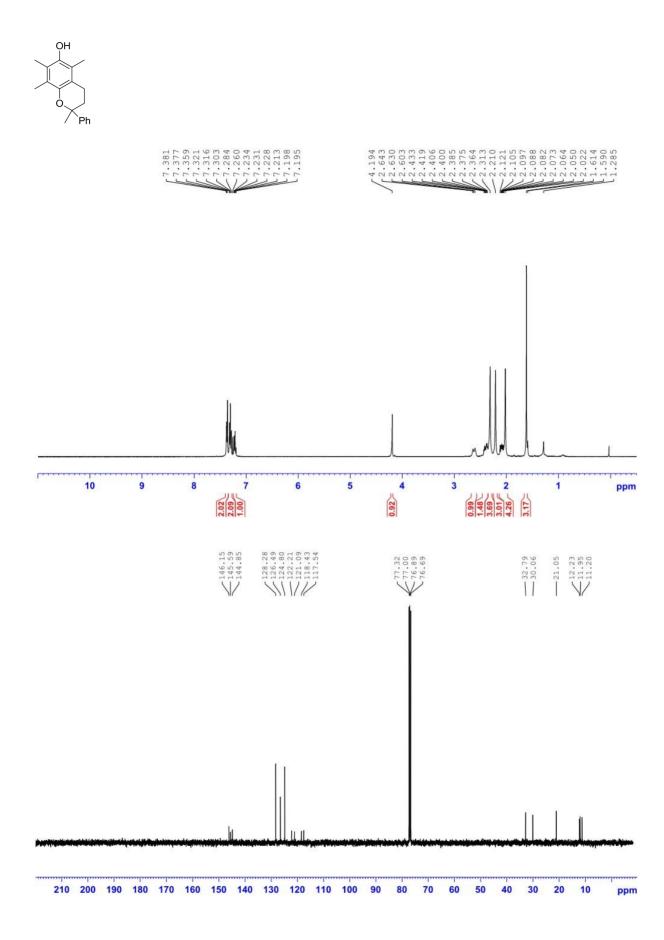


2,5,7,8-tetramethyl-2-phenyl-3,4-dihydro-2H-1-benzopyran-6-ol¹⁷

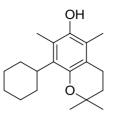


Trimethyl hydroquinone (1.2 mmol, 183 mg) and *p*-toluenesulfonic acid (TsOH·H₂O) (0.152 eq, 0.18 mmol, 34.7 mg) were added to a two-necked flask, and dry toluene (5 mml) was added under nitrogen. After stirring for five minutes, 2-phenyl-but-3-en-2-ol (1.2 eq, 213 mg) was injected dropwise into the flask by syringe. Then it was stirred for 18 hours at 100 °C. After cooling, ethyl acetate was added to dilute the solution, and the organic layer was washed with sodium bicarbonate and water, and dried with MgSO₄. The solvent was removed, the residue was separated by column chromatography (hexane:EtOAc = 49:1) to give the white solid in 45% yield.

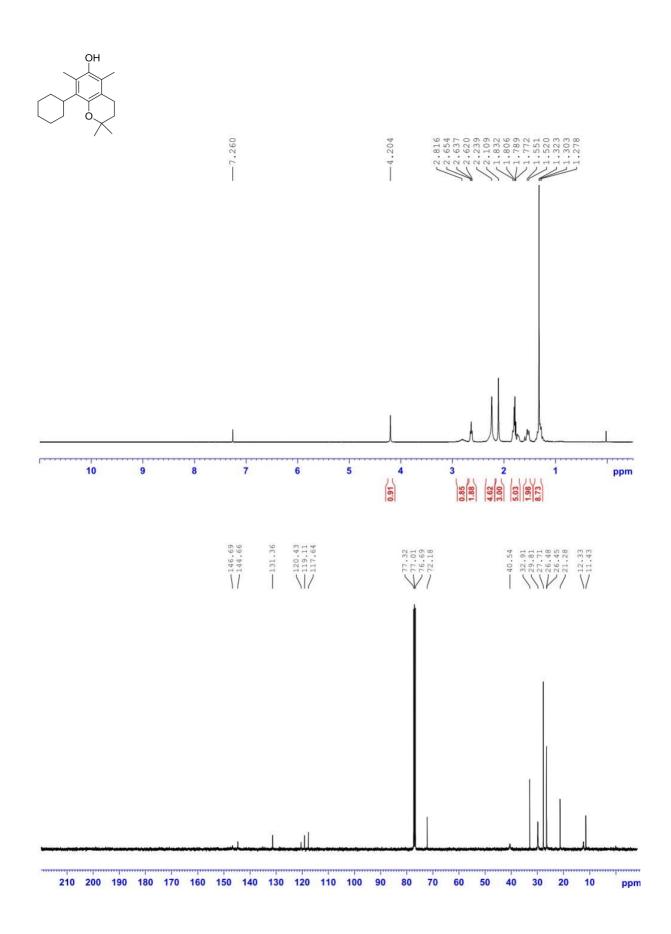
45%; White solid; ¹H NMR (400 MHz, CDCl₃) δ 1.61 (s, 3H), 2.02 (s, 3H), 2.05–2.12 (m, 1H), 2.21 (s, 3H), 2.31 (s, 3H), 2.36–2.43 (m, 2H), 2.60–2.64 (m, 1H), 4.19 (s, 1H), 7.20–7.23 (m, 1H), 7.28–7.32 (m, 2H), 7.36–7.38 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ: 11.2, 11.9, 12.2, 21.0, 30.1, 32.8, 76.9, 117.5, 118.4, 121.1, 122.2, 124.8, 126.5, 128.3, 144.8, 145.6, 146.2; m.p. 77–79 °C (EtOAc); ESI-HRMS: *m/z* calcd for C₁₉H₂₉O₃ [M+H]⁺: 305.2117; found: 305.2114.



8-cyclohexyl-2,2,5,7-tetramethyl-3,4-dihydro-2H-1-benzopyran-6-ol



56%; White solid; ¹H NMR (400 MHz, CDCl₃) δ 1.28–1.32 (m, 9H), 1.52–1.55 (m, 2H), 1.77–1.83 (m, 5H), 2.11(s, 3H), 2.24 (brs, 5H), 2.64 (t, 2H, *J* = 6.8 Hz), 2.82 (brs, 1H), 4.20 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ: 11.4, 12.3, 21.3, 26.4, 26.5, 27.7, 29.8, 32.9, 40.5, 72.2, 117.6, 119.1, 120.4, 131.4, 144.7, 146.7; m.p. 85–88 °C (EtOAc); ESI-HRMS: *m*/*z* calcd for C₁₉H₂₉O₂ [M+H]⁺: 289.2168; found: 289.2163.



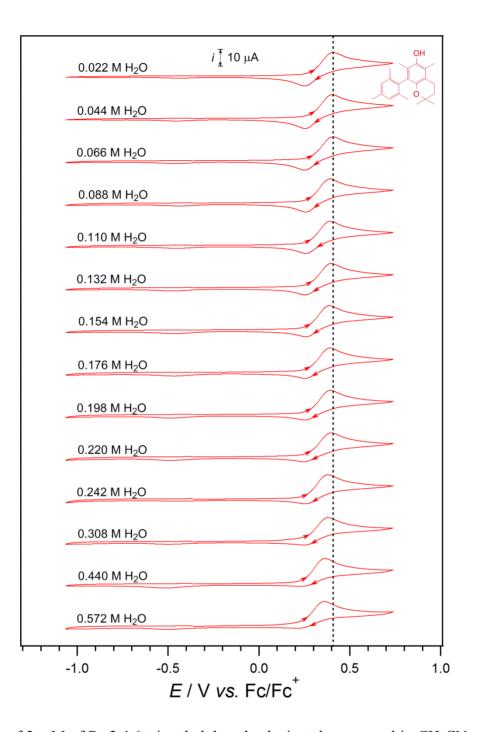


Figure S1. CVs of 2 mM of R₃ 2,4,6-trimethylphenyl substituted compound in CH₃CN containing 0.2 M Bu₄NPF₆ at a 1 mm diameter circular planar Pt electrode at 22 ± 2 °C at a scan rate of 0.1 V s⁻¹ at the specified concentration of water.

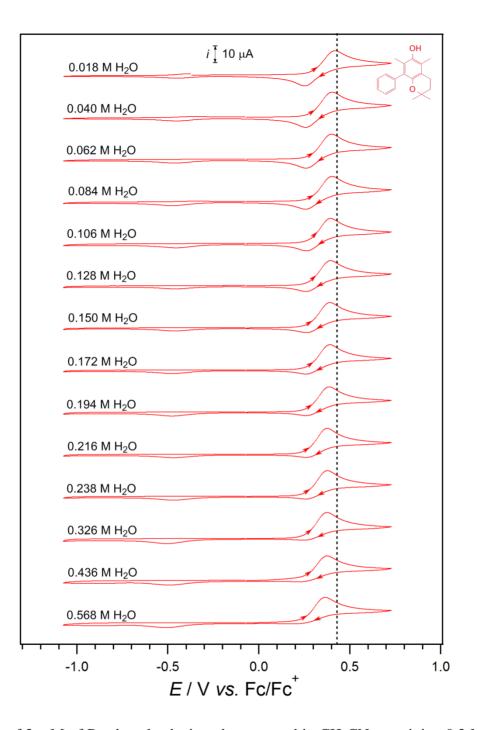


Figure S2. CVs of 2 mM of R_3 phenyl substituted compound in CH₃CN containing 0.2 M Bu₄NPF₆ at a 1 mm diameter circular planar Pt electrode at 22 ± 2 °C at a scan rate of 0.1 V s⁻¹ at the specified concentration of water.

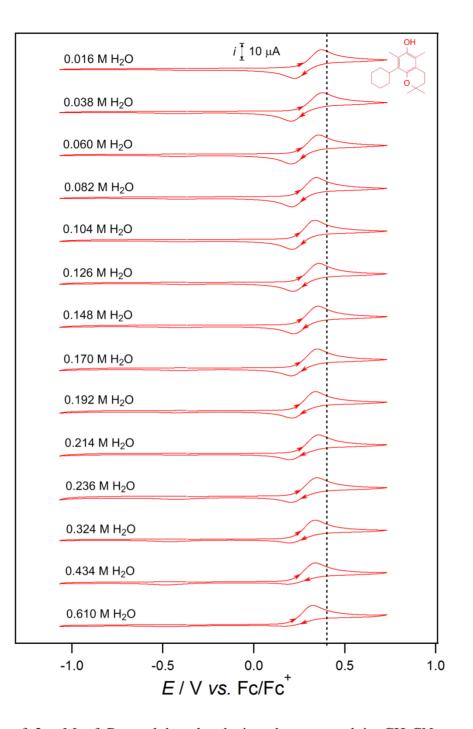


Figure S3. CVs of 2 mM of R_3 cyclohexyl substituted compound in CH₃CN containing 0.2 M Bu₄NPF₆ at a 1 mm diameter circular planar Pt electrode at 22 ± 2 °C at a scan rate of 0.1 V s⁻¹ at the specified concentration of water.

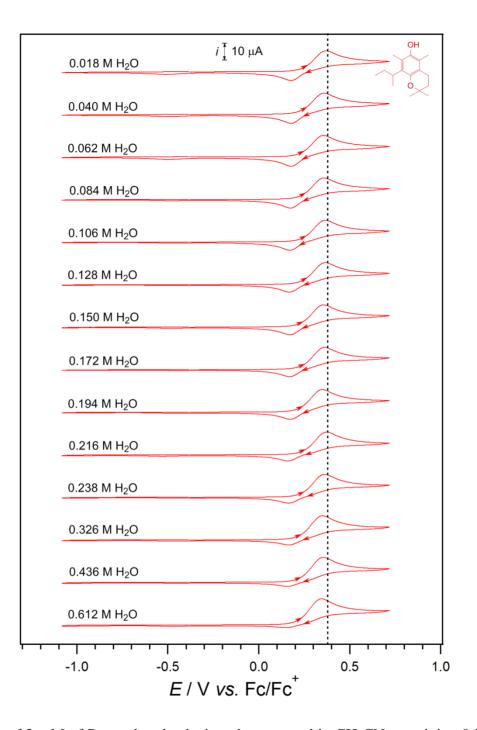


Figure S4. CVs of 2 mM of R_3 *sec*-butyl substituted compound in CH₃CN containing 0.2 M Bu₄NPF₆ at a 1 mm diameter circular planar Pt electrode at 22 ± 2 °C at a scan rate of 0.1 V s⁻¹ at the specified concentration of water.

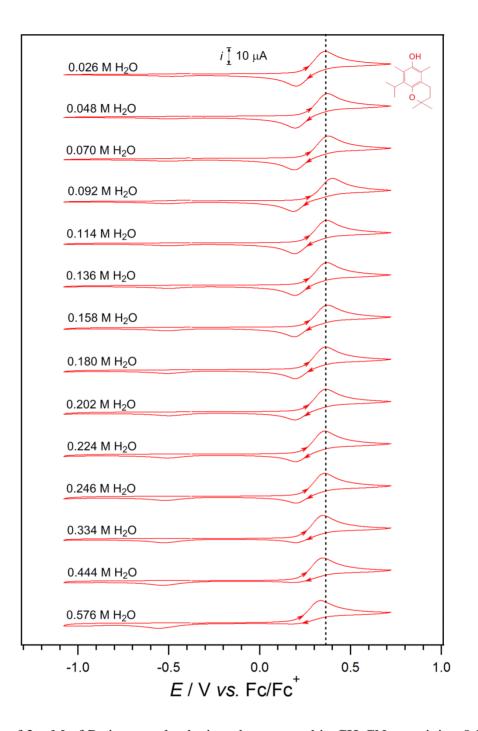


Figure S5. CVs of 2 mM of R_3 isopropyl substituted compound in CH₃CN containing 0.2 M Bu₄NPF₆ at a 1 mm diameter circular planar Pt electrode at 22 ± 2 °C at a scan rate of 0.1 V s⁻¹ at the specified concentration of water.

3. Computational Studies

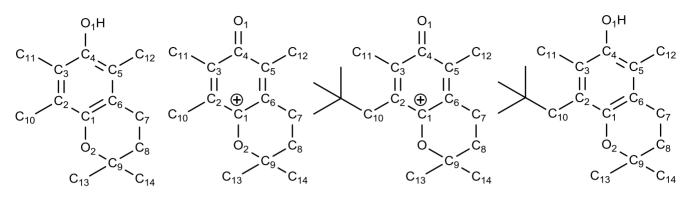


Figure S6. Numbering system for atoms in density functional theory calculations.

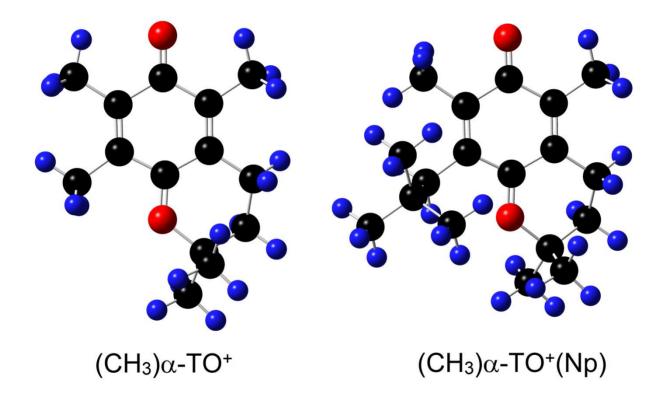


Figure S7. Density functional theory B3LYP/6-311+G(2df,p) optimized structures of the diamagnetic cations

Bond	Bond Length / Å								
	(СН3)а-ТОН		(CH ₃)-aTOH(Np)		(CH ₃)α-TO ⁺		(CH ₃)-αTO ⁺ (Np)		
	Gas	CH ₃ CN	Gas	CH ₃ CN	Gas	CH ₃ CN	Gas	CH ₃ CN	
C ₁ -O ₂	1.37726	1.37746	1.37630	1.37558	1.28499	1.28052	1.28403	1.27935	
$C_4 - O_1$	1.37950	1.38135	1.37993	1.38234	1.21056	1.21491	1.21134	1.21538	
C ₉ –O ₂	1.44333	1.45069	1.44571	1.45147	1.51975	1.51976	1.52089	1.52460	
$C_1 - C_2$	1.40232	1.40282	1.40930	1.41091	1.45676	1.45890	1.46491	1.46753	
C ₂ –C ₃	1.39760	1.39948	1.40324	1.40422	1.35464	1.35148	1.35913	1.35583	
C ₃ -C ₄	1.39692	1.39730	1.40221	1.40312	1.50042	1.49782	1.49939	1.49805	
C ₄ -C ₅	1.39474	1.39624	1.38990	1.39086	1.50521	1.50201	1.50044	1.49670	
C5-C6	1.40389	1.40410	1.40134	1.40253	1.35427	1.35206	1.35237	1.35051	
$C_1 - C_6$	1.39720	1.39889	1.39610	1.39681	1.44826	1.44769	1.45303	1.45178	
C ₆ -C ₇	1.51383	1.51392	1.51107	1.51130	1.50726	1.50394	1.50667	1.50287	
C7-C8	1.52690	1.52659	1.52543	1.52532	1.52690	1.52583	1.52580	1.52526	
C ₈ -C ₉	1.52985	1.52859	1.52859	1.52766	1.52085	1.51982	1.51950	1.51894	
C2-C10	1.50699	1.50782	1.51689	1.51738	1.50134	1.49916	1.51159	1.51024	
C ₃ -C ₁₁	1.50731	1.50829	1.51100	1.51121	1.49382	1.49495	1.49044	1.49161	
C ₅ -C ₁₂	1.51047	1.51051	1.50725	1.50803	1.48971	1.49004	1.49015	1.49074	
C ₉ -C ₁₃	1.53243	1.53134	1.53301	1.53203	1.52293	1.52237	1.52287	1.52153	
C ₉ -C ₁₄	1.52422	1.52365	1.52484	1.52431	1.51771	1.51747	1.51845	1.51795	
Energy / hartrees	-696.226614	-696.233328	-853.523478	-853.529546	-695.358120	-695.424351	-852.656778	-852.721333	
ZPE / hartrees	0.310284	0.309934	0.423576	0.422933	0.299181	0.298922	0.412232	0.412188	
Total Energy / hartrees	-695.919774	-695.926834	-853.104604	-853.111307	-695.062260	-695.128747	-852.249122	-852.313720	

Table S1. Calculated (B3LYP/6-311+G(2df,p)) bond lengths and energies (gaseous and solution phase) for the α -tocopherol model compound [(CH₃) α -TOH)] and its neopentyl derivative [(CH₃)- α TOH(Np)] and their associated diamagnetic cationic forms, (CH₃) α -TO+ and (CH₃)- α TO⁺(Np).

Table S2. Calculated (B3LYP/6-31G(d)) Mulliken (default) charges (gaseous and solution phase) for the α -tocopherol model compound [(CH₃) α -TOH)] and its neopentyl derivative [(CH₃)- α TOH(Np)] and their associated diamagnetic cationic forms, (CH₃) α -TO+ and (CH₃)- α TO⁺(Np).

Bond	Charge								
	(CH ₃)α-TOH		(CH ₃)-aTOH(Np)		$(CH_3)\alpha$ -TO ⁺		$(CH_3)-\alpha TO^+(Np)$		
	Gas	CH ₃ CN	Gas	CH ₃ CN	Gas	CH ₃ CN	Gas	CH ₃ CN	
Н	0.411	0.411	0.412	0.412	-	-	-	-	
O ₁	-0.665	-0.666	-0.665	-0.667	-0.385	-0.388	-0.387	-0.390	
O_2	-0.574	-0.576	-0.578	-0.579	-0.455	-0.452	-0.457	-0.455	
C ₁	0.270	0.272	0.275	0.276	0.376	0.376	0.372	0.372	
C_2	0.049	0.049	0.041	0.041	0.060	0.059	0.082	0.083	
C ₃	0.098	0.099	0.102	0.102	0.086	0.086	0.049	0.048	
C_4	0.205	0.205	0.206	0.206	0.390	0.389	0.394	0.392	
C ₅	0.095	0.095	0.084	0.084	0.062	0.063	0.059	0.061	
C ₆	0.081	0.080	0.075	0.074	0.093	0.094	0.094	0.094	
C ₇	-0.371	-0.372	-0.362	-0.362	-0.370	-0.371	-0.370	-0.369	
C_8	-0.275	-0.275	-0.278	-0.278	-0.308	-0.308	-0.309	-0.310	
C ₉	0.310	0.309	0.313	0.312	0.248	0.246	0.251	0.248	
C ₁₀	-0.532	-0.531	-0.351	-0.352	-0.522	-0.521	-0.359	-0.360	
C ₁₁	-0.535	-0.536	-0.591	-0.590	-0.557	-0.557	-0.524	-0.522	
C ₁₂	-0.592	-0.591	-0.534	-0.534	-0.551	-0.550	-0.550	-0.551	
C ₁₃	-0.457	-0.457	-0.460	-0.459	-0.465	-0.463	-0.466	-0.465	
C ₁₄	-0.456	-0.456	-0.458	-0.459	-0.463	-0.461	-0.463	-0.460	
Residual Atoms	2.937	2.941	2.766	2.768	3.748	3.723	3.585	3.583	
Total	-0.001	0.001	-0.003	-0.005	0.987	0.965	1.001	0.999	

Table S3. Calculated (B3LYP/6-31G(d)) charges from natural bond orbital (NBO) analysis (gaseous and solution phase) for the α -tocopherol model compound [(CH₃) α -TOH)] and its neopentyl derivative [(CH₃)- α TOH(Np)] and their associated diamagnetic cationic forms, (CH₃) α -TO+ and (CH₃)- α TO⁺(Np).

Bond	Charge								
	(CH ₃)α-TOH		(CH ₃)-aTOH(Np)		$(CH_3)\alpha$ -TO ⁺		(CH_3) - $\alpha TO^+(Np)$		
	Gas	CH ₃ CN	Gas	CH ₃ CN	Gas	CH ₃ CN	Gas	CH ₃ CN	
Н	0.411	0.490	0.491	0.491	-	-	-	-	
O ₁	-0.665	-0.707	-0.707	-0.708	-0.427	-0.427	-0.434	-0.434	
O_2	-0.574	-0.563	-0.563	-0.564	-0.431	-0.430	-0.431	-0.430	
C ₁	0.270	0.304	0.311	0.311	0.559	0.561	0.566	0.568	
C_2	0.049	-0.049	-0.057	-0.057	-0.065	-0.064	-0.062	-0.061	
C ₃	0.098	-0.062	-0.086	-0.086	0.033	0.031	0.035	0.032	
C_4	0.205	0.321	0.319	0.319	0.516	0.514	0.514	0.512	
C ₅	0.095	-0.090	-0.058	-0.058	0.064	0.063	0.061	0.060	
C ₆	0.081	-0.088	-0.086	-0.086	-0.103	-0.103	-0.103	-0.102	
C ₇	-0.371	-0.472	-0.470	-0.470	-0.487	-0.486	-0.485	-0.484	
C ₈	-0.275	-0.474	-0.475	-0.475	-0.493	-0.493	-0.492	-0.493	
C ₉	0.310	0.287	0.289	0.288	0.283	0.283	0.284	0.285	
C ₁₀	-0.532	-0.692	-0.468	-0.468	-0.706	-0.706	-0.484	-0.484	
C ₁₁	-0.535	-0.693	-0.707	-0.707	-0.726	-0.725	-0.713	-0.712	
C ₁₂	-0.592	-0.709	-0.693	-0.693	-0.727	-0.726	-0.727	-0.726	
C ₁₃	-0.457	-0.696	-0.696	-0.696	-0.704	-0.702	-0.702	-0.700	
C ₁₄	-0.456	-0.679	-0.679	-0.679	-0.697	-0.696	-0.696	-0.695	
Residual Atoms	2.937	4.571	4.328	4.342	5.110	5.104	4.873	4.859	
Total	-0.001	-0.001	-0.007	0.004	0.999	0.998	1.004	0.995	

4. Additional References

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