

ELECTRONIC SUPPLEMENTARY INFORMATION

A novel pyrimidine tetrad contributing to stabilize tetramolecular G-quadruplex structures

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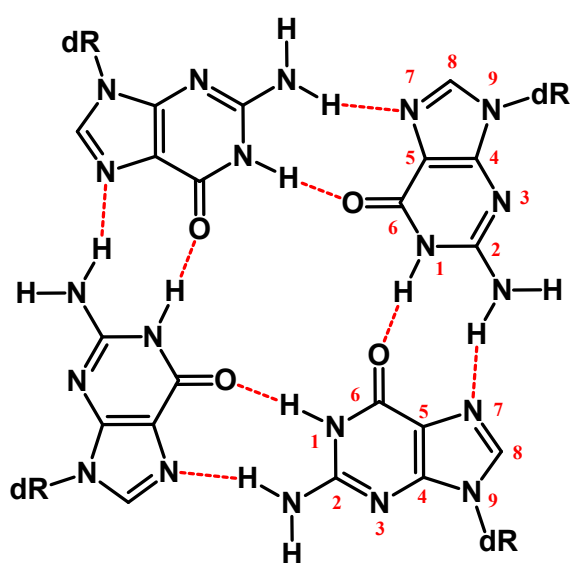
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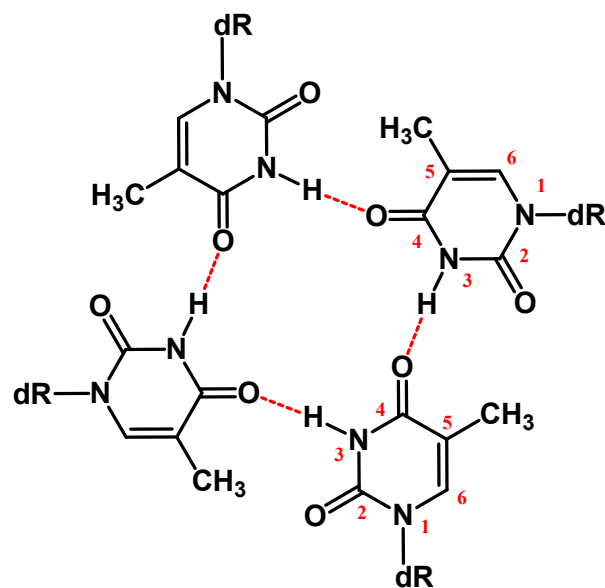
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- HYDROGEN BONDING PATTERNS (G-TETRAD AND T-TETRAD)
- CD SPECTRA
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- ³¹P-NMR SPECTRUM (**AM**)
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- MOLECULAR MODELS (**BR** AND **TH**)
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G-tetrad



T-tetrad

Figure S1: Hydrogen bond patterns of a G-tetrad and a T-tetrad.

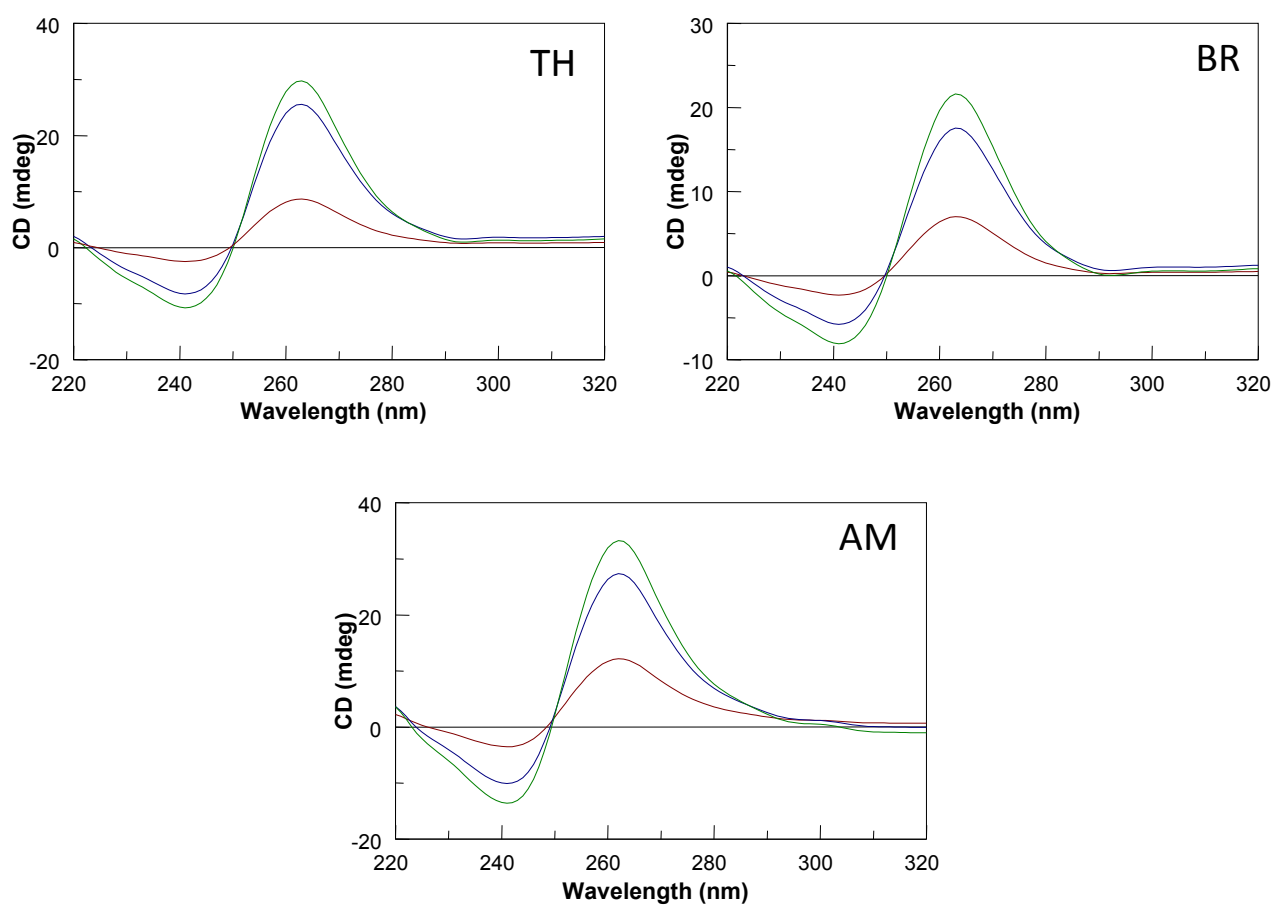


Figure S2. CD spectra at 20°C of investigated G-quadruplexes in potassium buffer solutions (20 mM $\text{KH}_2\text{PO}_4/\text{K}_2\text{HPO}_4$, pH 7.0) at different KCl concentration: 10 mM (red), 100 mM (blue), 300 mM (green). ODN strand concentration: 100 μM .

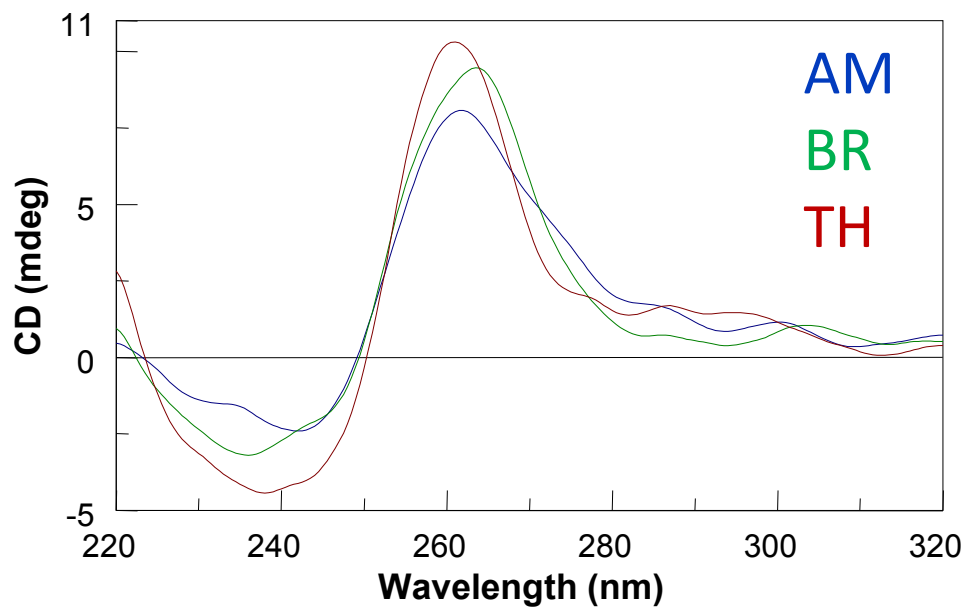


Figure S3. CD spectra at 20°C of investigated G-quadruplexes at 100 μ M ODN strand concentration in a buffer solution 20 mM $\text{NaH}_2\text{PO}_4/\text{Na}_2\text{HPO}_4$, 100 mM NaCl (pH 7.0).

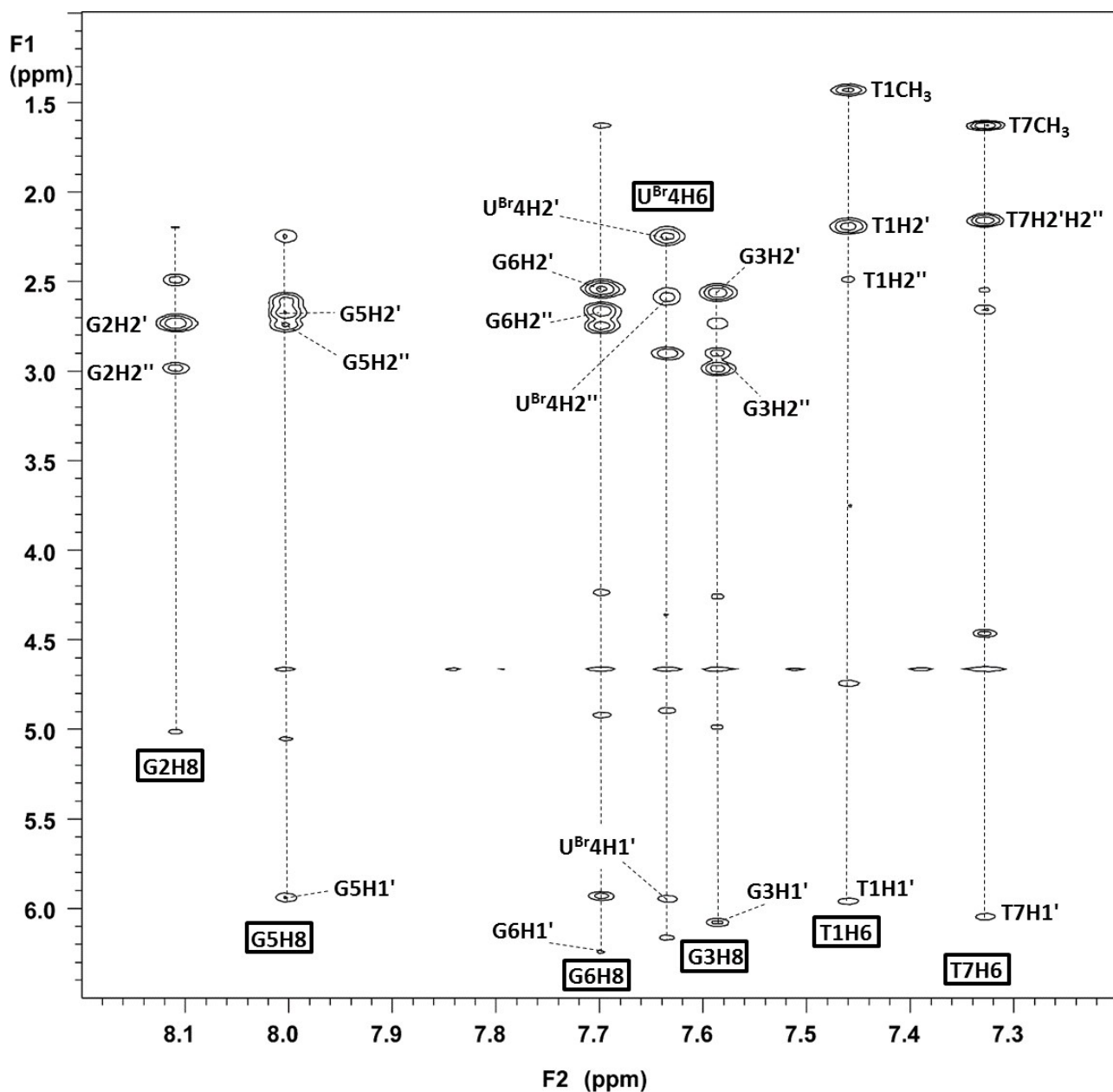


Figure S4: Expanded region of 2D NOESY spectrum of G-quadruplex **BR** (500 MHz; 25°C; strand concentration 1.5 mM; 20 mM KH₂PO₄/K₂HPO₄, 100 mM KCl and 0.2 mM EDTA, pH 7.0 in H₂O/D₂O 9:1; total volume 0.6 ml; mixing time 180 ms) correlating bases H8/H6 and sugar protons.

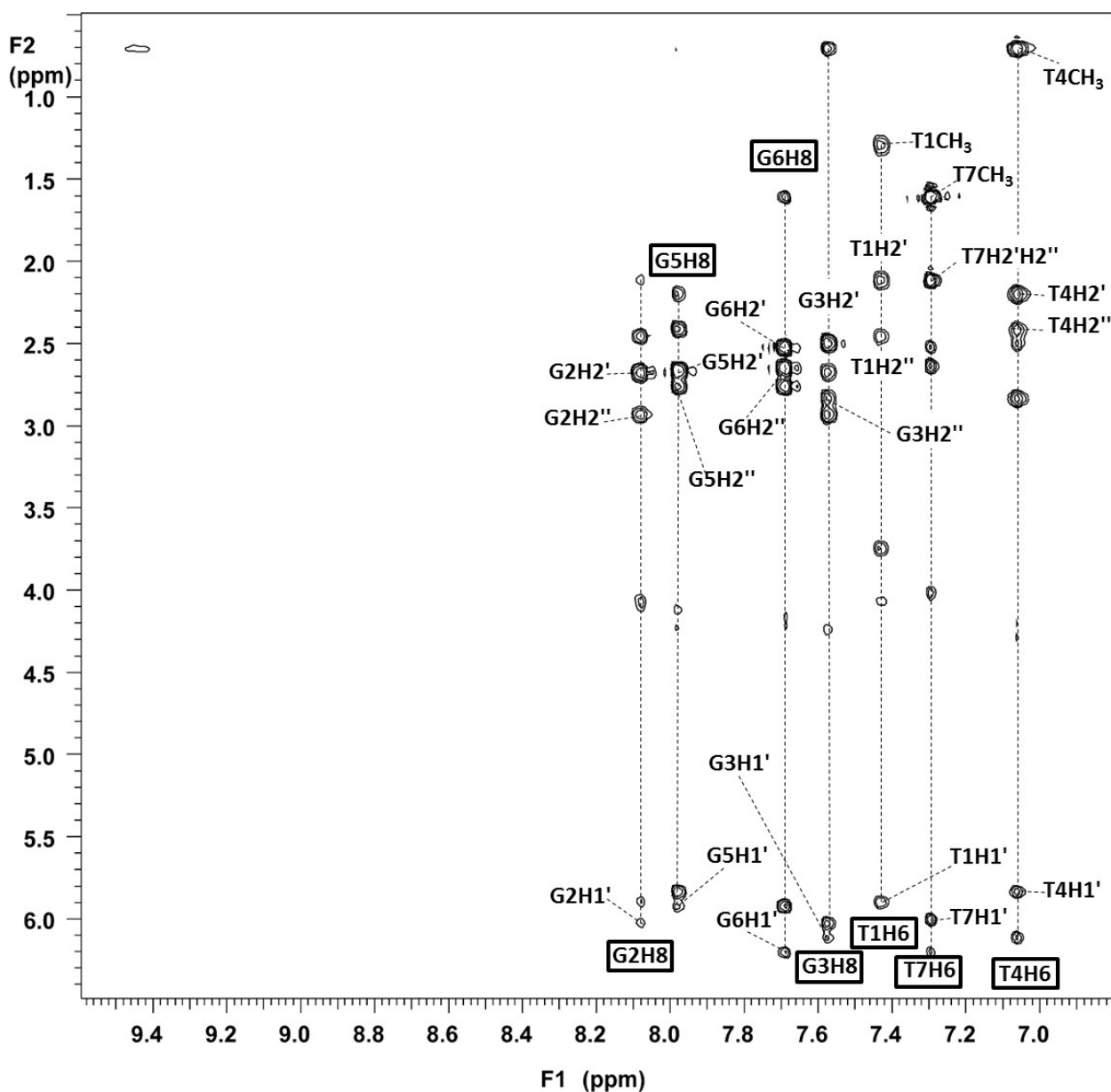


Figure S5: Expanded region of 2D NOESY spectrum of G-quadruplex **TH** (500 MHz; 25°C; strand concentration 1.5 mM; 20 mM KH₂PO₄/K₂HPO₄, 100 mM KCl and 0.2 mM EDTA, pH 7.0 in H₂O/D₂O 9:1; total volume 0.6 ml; mixing time 180 ms) correlating bases H8/H6 and sugar protons.

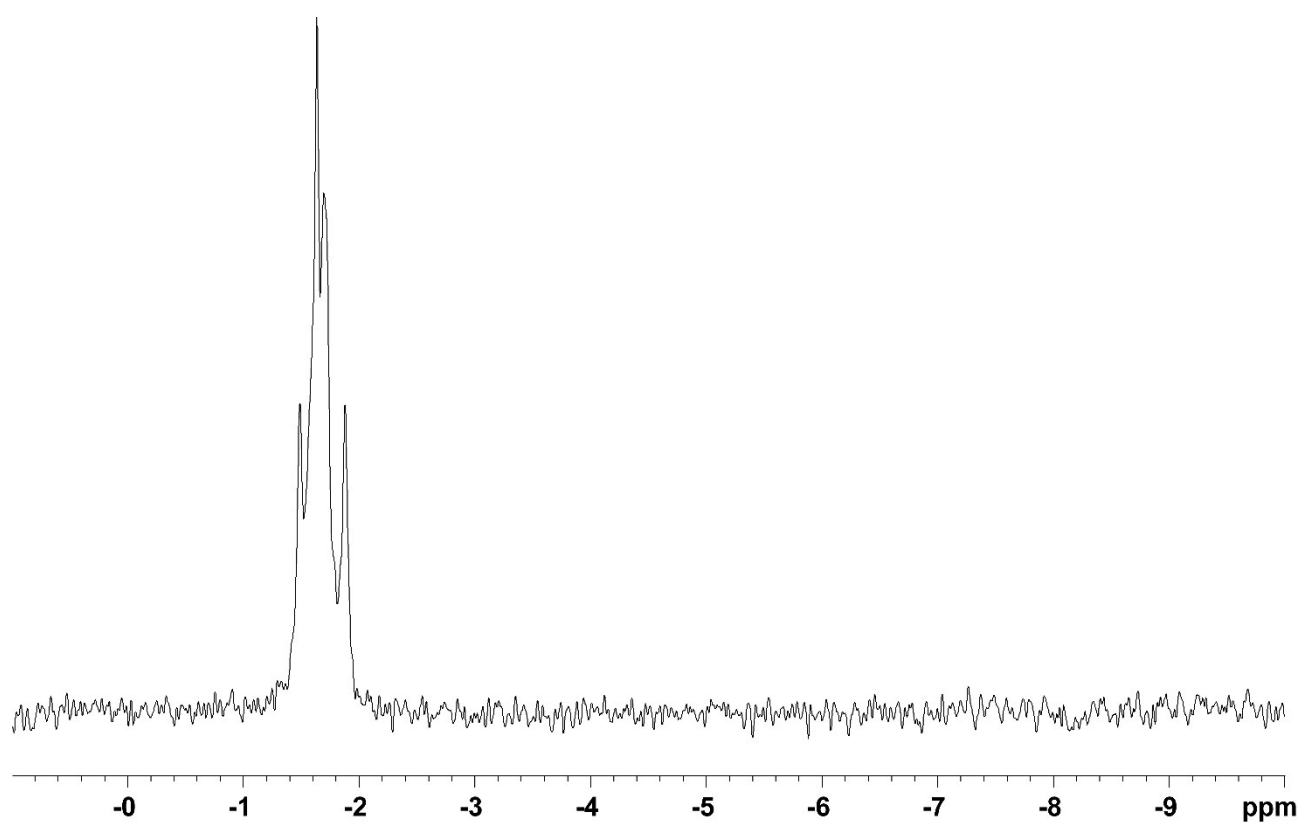


Figure S6: Proton decoupled ^{31}P -NMR spectrum of **AM** (500 MHz, $T = 25^\circ\text{C}$, 20 mM KH_2PO_4 , 100 mM KCl, 0.2 mM EDTA, pH=7).

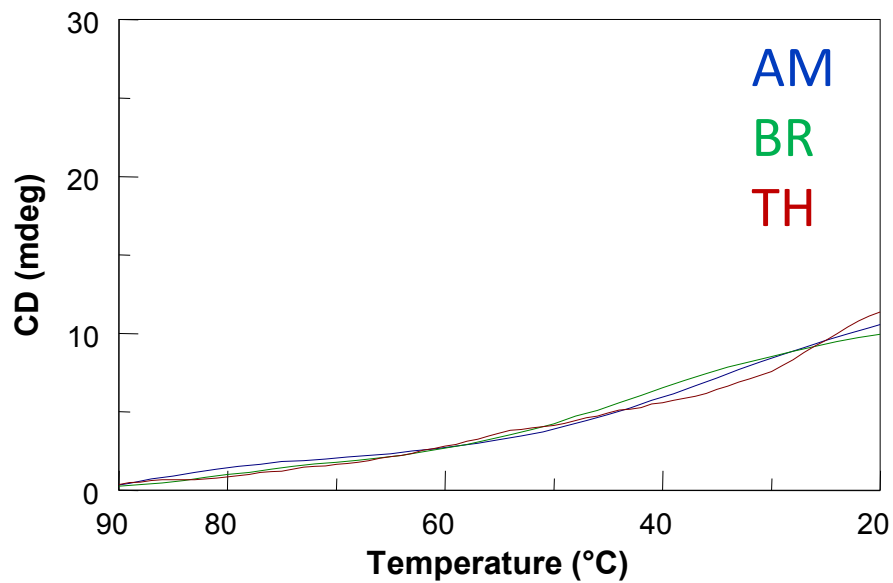


Figure S7. CD annealing profiles registered as a function of temperature from 90°C to 20°C for all investigated G-quadruplexes at their maximum Cotton effect wavelengths. CD data were recorded in a 0.1 cm pathlength cuvette with a scan rate of 10°C/h at 100 μ M ODN strand concentration in a buffer solution 20 mM $\text{KH}_2\text{PO}_4/\text{K}_2\text{HPO}_4$, 100 mM KCl (pH 7.0).

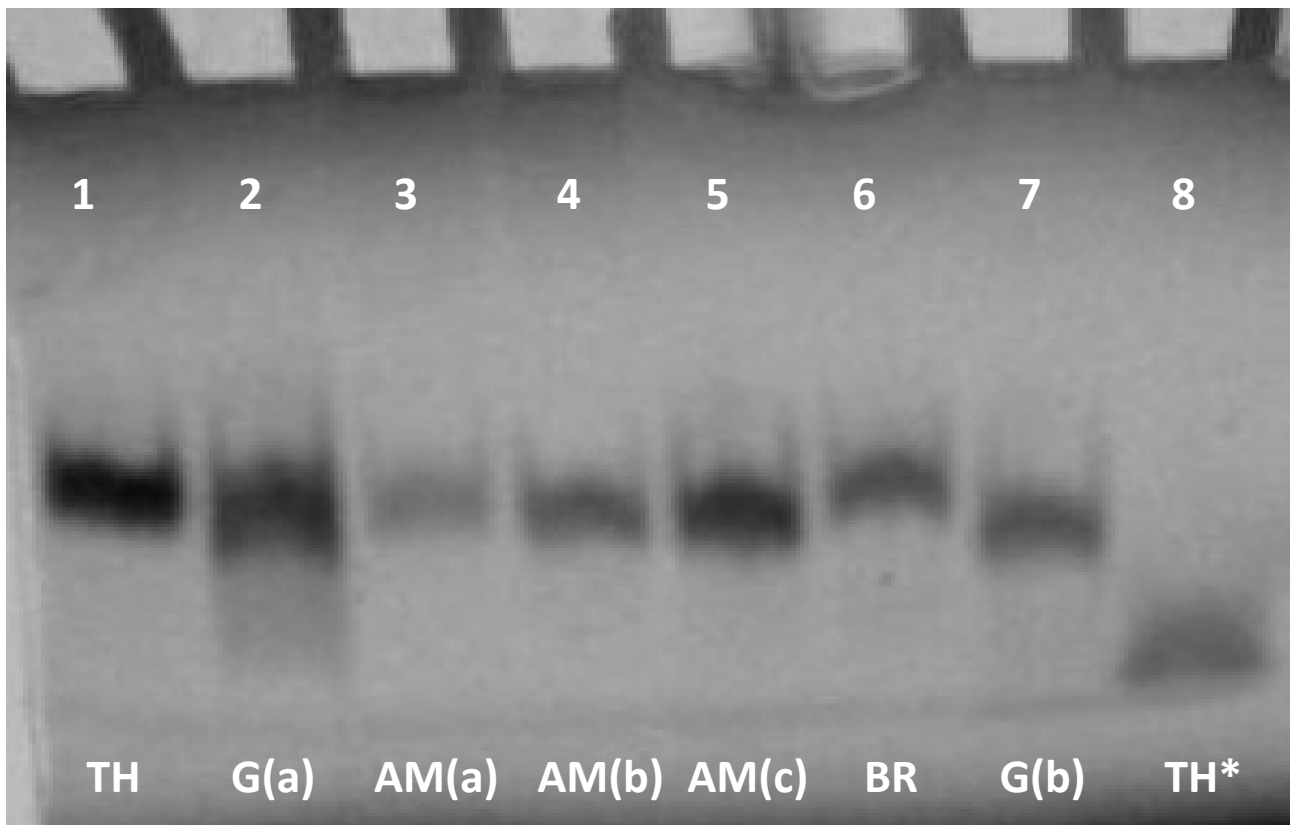


Figure S8: Non-denaturing polyacrylamide gel electrophoresis of the investigated G-quadruplex structures and $[d(TG_5T)]_4$ as a reference. Lane 1: **TH**, $[ODN] \approx 1.5$ mM; lanes 2 and 7 $[d(TG_5T)]_4$, $[ODN] \approx 1.5$ mM G(a) and 1 mM G(b), respectively; lanes 3, 4 and 5: **AM**, $[ODN] \approx 0.5$ mM AM(a), 1 mM AM(b) and 1.5 mM AM(c), respectively; lane 6: **BR**, $[ODN] \approx 1$ mM; lane 8 (single strand reference): **TH** denatured by pre-treatment with LiOH (40 mM, 5' at 80°C) followed by neutralization with 40 mM HCl and immediate loading on the gel, $[ODN] \approx 1$ mM.

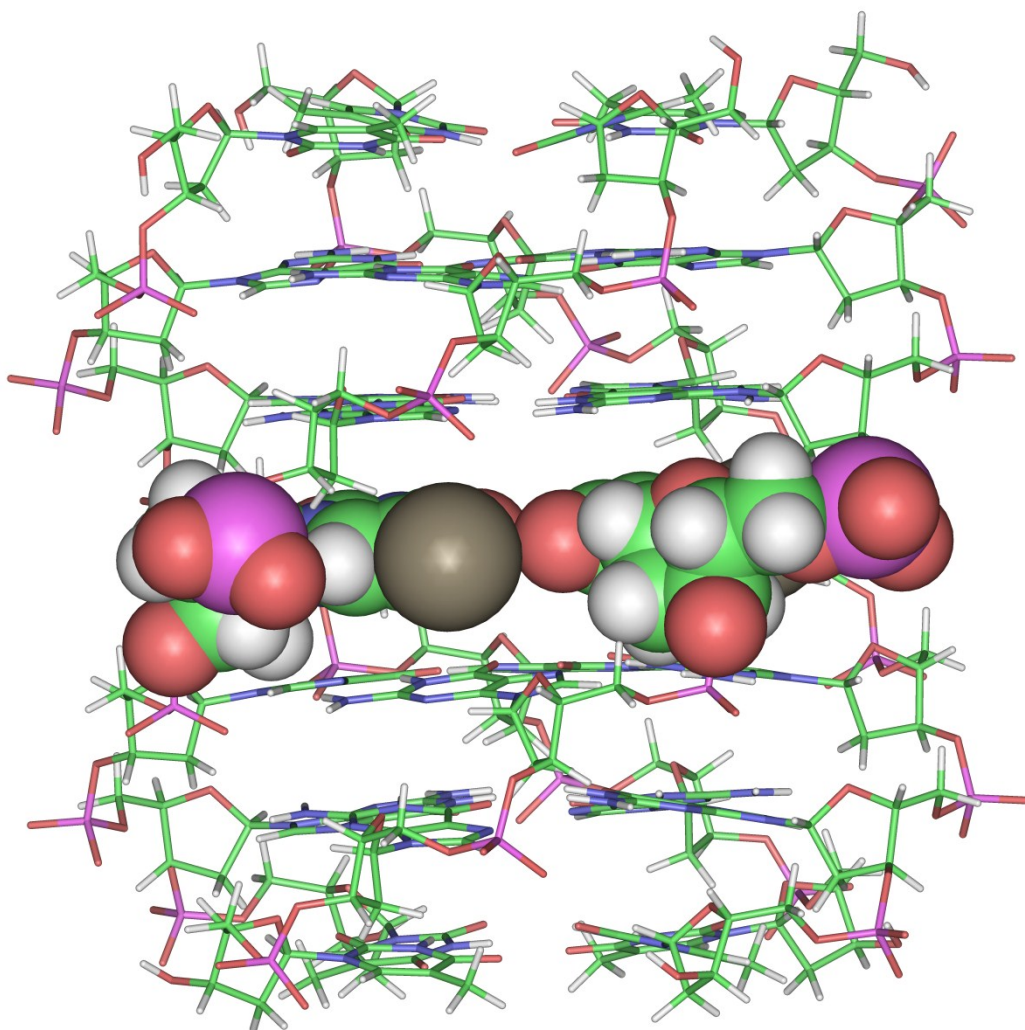


Figure S9: Side view of stick model of G-quadruplex **BR**. The U^{Br}-tetrad is reported in CPK. Heavy atoms are shown with different colors (carbons, green; nitrogens, blue; oxygens, red; hydrogens, white; phosphorus, purple; bromine, dark grey).

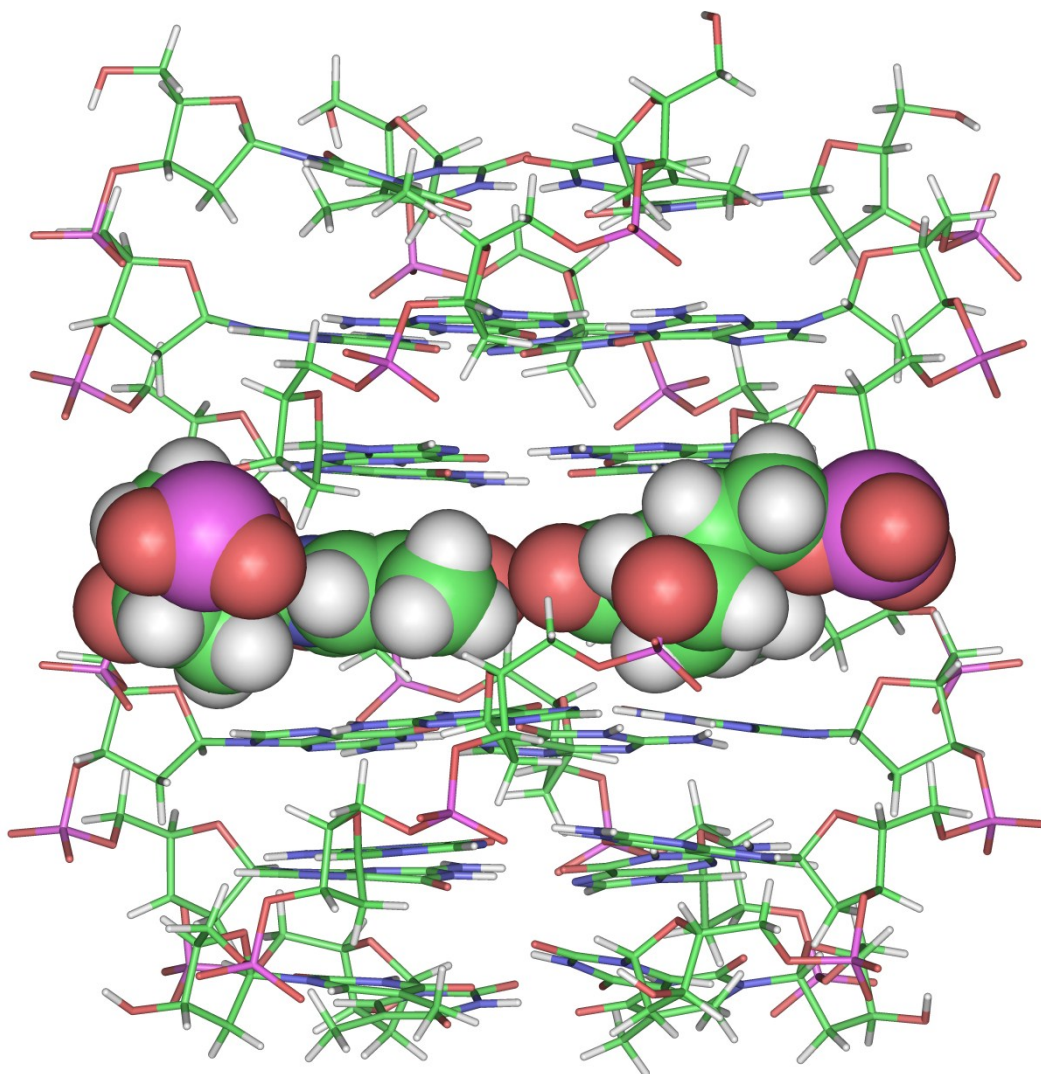


Figure S10: Side view of stick model of G-quadruplex **TH**. The T-tetrad is reported in CPK. Heavy atoms are shown with different colors (carbons, green; nitrogens, blue; oxygens, red; hydrogens, white; phosphorus, purple).

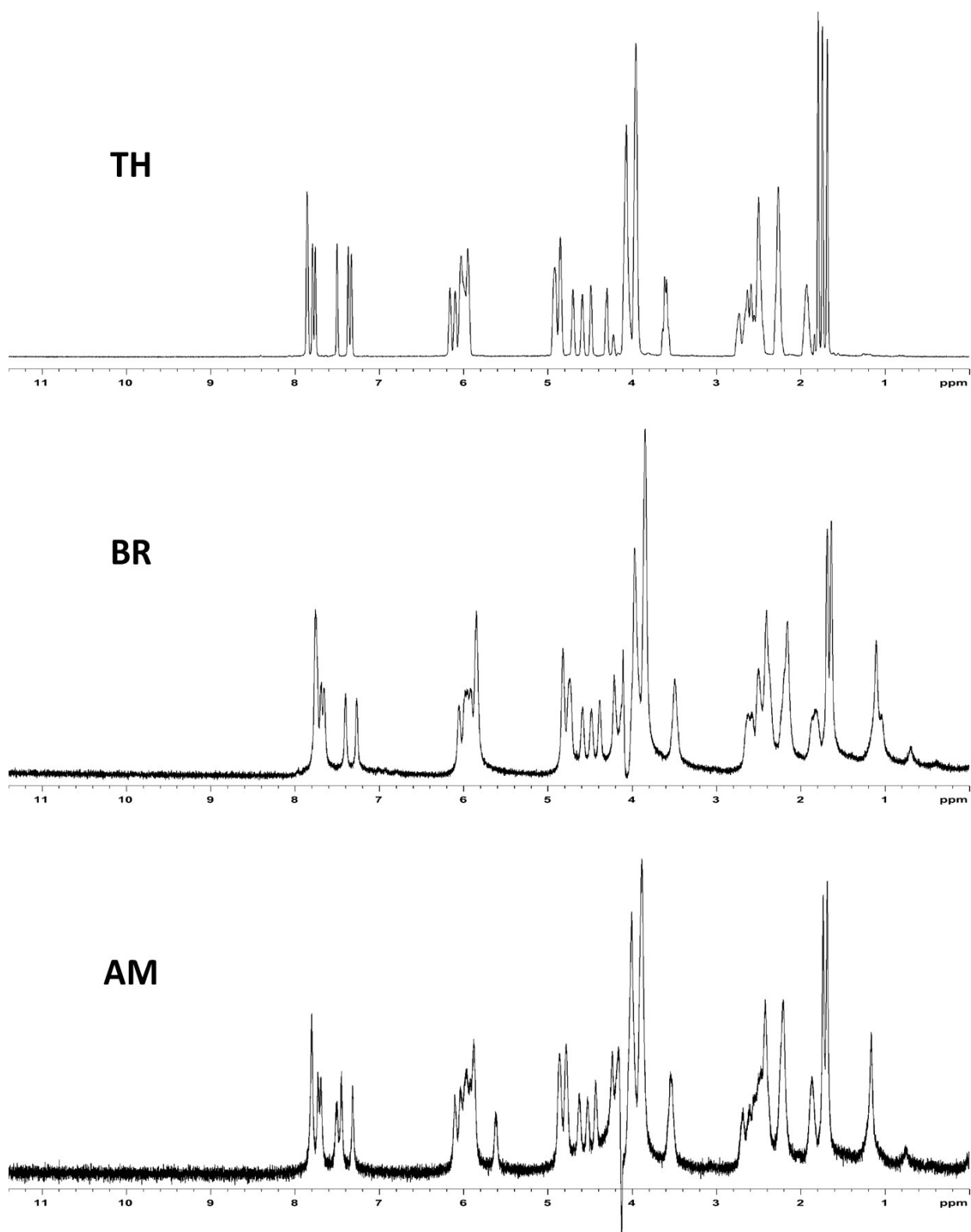


Figure S11: High-resolution ¹H-NMR spectra (500 MHz, D₂O, 80°C, no salt) of **TH**, **BR** and **AM**.

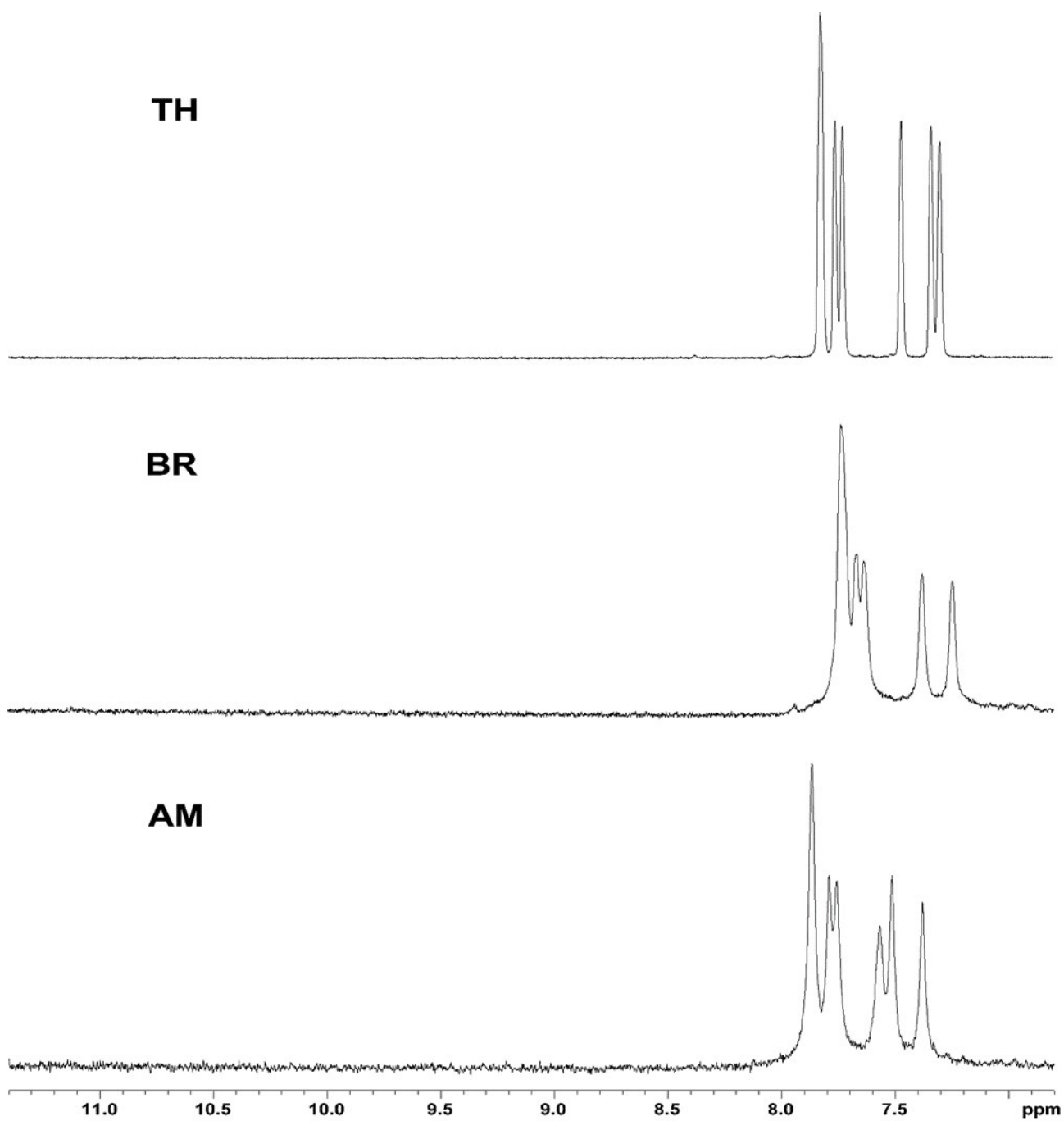


Figure S12: Imino and aromatic proton regions of the high-resolution ¹H-NMR spectra (500 MHz, D₂O, 80°C, no salt) of **TH**, **BR** and **AM**.

5'-T₁G₂G₃U^{NH₂}G₅G₆T₇-3' (AM)								
	H8/H6	H1'	H2'/H2''	H3'	H4'	H5'/H5''	CH ₃ /NH ₂	NH
T ₁	7.31	6.04	2.18	4.49	4.23	4.04	1.66	-
G ₂	7.64	6.03	2.75/3.00	5.01	4.27	N.D.	-	11.53
G ₃	7.58	6.18	2.45/2.87	4.98	4.24	N.D.	-	11.28
U ^{NH₂} ₄	7.21	5.82	2.13/2.45	4.84	4.38	N.D.	4.48	10.75
G ₅	8.13	6.03	2.77/2.89	5.09	4.27	4.13	-	11.22
G ₆	7.64	6.20	2.57/2.63	4.11	N.D.	N.D.	-	11.04
T ₇	7.31	6.04	2.18	4.23	4.04	4.04	1.66	-

5'-T₁G₂G₃U^{Br}₄G₅G₆T₇-3' (BR)								
	H8/H6	H1'	H2'/H2''	H3'	H4'	H5'/H5''	CH ₃	NH
T ₁	7.46	5.96	2.20/2.50	4.75	N.D.	4.10/3.76	1.44	-
G ₂	8.11	6.08	2.73/3.00	5.02	4.41	4.12/4.08	-	11.66
G ₃	7.59	6.16	2.57/2.90	4.99	4.50	4.27/4.01	-	11.28
U ^{Br} ₄	7.64	5.95	2.26/2.60	4.90	4.36	4.25	-	-
G ₅	8.00	5.93	2.67/2.75	5.06	4.41	4.25/4.13	-	11.25
G ₆	7.70	6.24	2.55/2.66	4.92	4.49	4.24	-	11.00
T ₇	7.33	6.04	2.17	4.47	4.22	4.06/3.64	1.64	-

5'-T₁G₂G₃T₄G₅G₆T₇-3' (TH)								
	H8/H6	H1'	H2'/H2''	H3'	H4'	H5'/H5''	CH ₃	NH
T ₁	7.43	5.90	2.13/2.46	4.71	4.08	3.76	1.30	-
G ₂	8.08	6.03	2.68/2.93	5.00	4.40	4.12/4.07	-	11.56
G ₃	7.57	6.11	2.50/2.86	4.99	4.47	4.24/4.08	-	11.23
T ₄	7.06	5.84	2.21/2.43	4.89	4.30	4.20/4.11	0.72	9.44
G ₅	7.98	5.94	2.67/2.78	5.04	4.43	4.24/4.13	-	11.23
G ₆	7.69	6.21	2.54/2.65	4.93	4.47	4.20/4.03	-	10.99
T ₇	7.30	6.01	2.13	4.46	4.20	4.05/4.01	1.61	-

Table S1: Proton chemical shifts assignment for G-quadruplex structures formed by ODNs **AM**, **BR** and **TH** (500 MHz, T = 25°C) in 20 mM $\text{KH}_2\text{PO}_4/\text{K}_2\text{HPO}_4$, 100 mM KCl and 0.2 mM EDTA (pH 7.0).