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Supporting Information

Synthesis, characterization and photophysical studies of selfassembled azo biphenyl urea derivatives

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Scheme 1. Schematic representation for photo-isomerization reaction of 7(7a/7b) upon UV and visible light irradiation.



Scheme 2. Optimized structure for *trans-cis* isomerization 7(7a/7b).



Scheme 3. Possible sheet like self-assembly structure of 7a and 7b through hydrogen bonding.



Fig. S1 Normalized absorption (a) and emission (b) spectra of DPOU (7a) in different solvents.



Fig. S2 Normalized absorption (a) and emission (b) spectra of DPBHU (7b) in different solvents.

Solvents	$\lambda_{abs}(nm)$	$\lambda_{ex}(nm)$	$\lambda_{emi}(nm)$	Stoke's shift (cm ⁻¹)
THF	368	368	406, 431	3972
CHCl ₃	365	365	410, 434	4355
DMF	358	358	409, 434	4891
DMSO	360	360	413, 438	4947
ACN	354	354	411, 436	5313
МеОН	356	356	415, 440	5363

Table 1. Absorption and emission values of 7a (DPOU) in different solvents.

Table 2. Absorption and emission values of **7b** (DPBHU) in different solvents.

Solvents	$\lambda_{abs}(nm)$	$\lambda_{ex}(nm)$	$\lambda_{emi}(nm)$	Stoke's shift (cm ⁻¹)
THF	372	372	412, 434	3867
CHCl ₃	367	367	412, 434	4207
DMF	365	365	410, 432	4249
DMSO	365	365	411, 435	4409
ACN	364	364	413, 438	4641
МеОН	364	364	415, 439	4719



Fig. S3 Absorbance spectra of Compound 7a (a) and 7b (b) in different THF: H₂O fractions.



Fig. S4 Peak intensity plot of 7a and 7b in different THF: H₂O fractions.



Fig. S5 Selected area electron diffraction pattern of compounds 7a and 7b (from left to right).



Fig. S6 AFM image of self-assembled **7a** (**a**) and **7b** (**b**) in THF-water mixture (40%) respectively, Samples were prepared by drop casting the solution (THF-water (40%)) of the compound onto a freshly cleaved, glass surface.



Fig. S7 FT-IR spectrum of compound 7 taken in before (A) and after (B) light irradiation at 365 nm.



Fig. S8 DFT-computed molecular frontiers orbitals of 7a.

Table 3. TD-DFT electronic transitions of all the compounds.

Compounds	wavelength (nm)	Oscillator strength (f)	Electronic transitions
Trans (7a)	443.17 nm	0.0014	HOMO-1 - LUMO
	374.57 nm	1.6996	HOMO - LUMO
	274.39 nm	0.0053	HOMO - LUMO
Cis (7a)	466.28 nm	0.1449	HOMO-1 - LUMO
	355.40 nm	0.3474	HOMO - LUMO
	260.60 nm	0.1014	HOMO - LUMO
Trans (7b)	454.00 nm	0.0011	HOMO-1 - LUMO
	367.99 nm	1.6999	HOMO - LUMO
	261.00 nm	0.0406	HOMO - LUMO
Cis (7b)	474.35 nm	0.1306	HOMO-1 - LUMO
	363.17 nm	0.2631	HOMO - LUMO
	257.11 nm	0.1852	HOMO - LUMO



Fig. S10 ¹³C NMR Spectrum of Compound 2.

-1.3715 7.9454 7.9454 7.8879 Ň 3 4.00 ~ 24 07-8.5 8.0 5.0 4.5 f1 (ppm) 4.0 3.5 3.0 2.5 2.0 1.5 1.0 0.5 9.0 7.5 7.0 6.5 6.0 5.5 0.0

Fig. S11 ¹H NMR Spectrum of Compound 3.



Fig. S12 ¹³C NMR Spectrum of Compound 3.



Fig. S13 ¹H NMR Spectrum of Compound 5.



Fig. S14 ¹³C NMR Spectrum of Compound 5.



Fig. S15 ¹H NMR Spectrum of Compound 6a.



Fig. S16¹³C NMR Spectrum of Compound 6a.



Fig. S17 ESI - Mass Spectrum of Compound 6a.



Fig. S18 ¹H NMR Spectrum of Compound 6b.



Fig. S19 ¹³C NMR Spectrum of Compound 6b.



Fig. S20 ESI - Mass Spectrum of Compound 6b.





Fig. S22 ¹³C NMR Spectrum of Compound 7a.



Fig. S23 ESI - Mass Spectrum of Compound 7a.



Fig. S24 ¹H NMR Spectrum of Compound 7b.



Fig. S25 ¹³C NMR Spectrum of Compound 7b.



Fig. S26 ESI - Mass Spectrum of Compound 7b.